



Cite this: *Nanoscale*, 2021, **13**, 659

Received 3rd November 2020,
Accepted 18th December 2020

DOI: 10.1039/d0nr07844a

rsc.li/nanoscale

Single molecule electronic devices with carbon-based materials: status and opportunity

Shima Ghasemi and Kasper Moth-Poulsen *

The field of single molecule electronics has progressed remarkably in the past decades by allowing for more versatile molecular functions and improving device fabrication techniques. In particular, electrodes made from carbon-based materials such as graphene and carbon nanotubes (CNTs) may enable parallel fabrication of multiple single molecule devices. In this perspective, we review the recent progress in the field of single molecule electronics, with a focus on devices that utilizes carbon-based electrodes. The paper is structured in three main sections: (i) controlling the molecule/graphene electrode interface using covalent and non-covalent approaches, (ii) using CNTs as electrodes for fabricating single molecule devices, and (iii) a discussion of possible future directions employing new or emerging 2D materials.

Introduction

Semiconductor technology is ubiquitous in our modern, connected society. In recent years, the use of on-line devices powering the internet of things (IoT) has massively expanded. This transformation is powered by impressive developments in the fabrication and miniaturization of silicon-based systems. Yet, to realize the rapid miniaturization trend as described by Moore's law will likely require new technologies.¹ These technologies should not only be quantum sized, but also consume little power, have a low environmental ramification and must be competitive in terms of cost. In this context, single-molecule electronics have been explored for several decades as one possible avenue for electronic devices. For example, single molecule devices with advanced functions such as switching,^{2–4} rectification,^{5,6} transistors^{7,8} and molecular diodes⁹ have been developed. However, parallel fabrication and reproducibility remain key issues for single-molecule electronic devices.

This could be resolved using carbon-based materials. In particular, graphene significantly reduces the number of possible contact geometries and adds compelling electronic functions to single-molecule electronic devices. 2D materials may soon also enable parallel fabrication of single-molecule electronic devices for the first time. In this review, we highlight and discuss the status and opportunities of carbon-based materials used in single-molecule electronics devices focusing on the progress in the recent 3–5 years (2016–2020). This review concentrates on devices with a single or a small number of active molecules. The broader field of electronic

devices based on self-assembled monolayers (SAMs) is not within the scope of this review.^{10–12,13}

Molecule-electrode interface

When working with single molecule electronics devices, the gold coupled organic molecules with thiol based “alligator clips” have been the preferred electrodes. However, the high atomic mobility and low chemical stability of metallic contacts could induce the destruction of molecular junctions upon oxidation or electromigration. Furthermore, small variations in the positioning of a single molecule onto the electrode may radically affect the electronic coupling between the molecule and electrode. Further, the positioning of a gate close to the molecule in the metallic nanogap is difficult. These challenges could limit practical device function.^{14–17} Therefore, controlling the fabrication of molecular junctions in the electrode–molecule interface has been challenging. Using electrodes made of graphene could be a viable solution for the formation of reliable molecular junctions. Arrangement of sp^2 -hybridized carbon atoms in graphene provides the desired compatibility with organic molecules. The 2D structure of graphene reduces the number of possible contact geometries compared to 3D structures, which potentially leads to a more reproducible single molecule device. The binding of anchoring groups to graphene electrodes can be achieved by either π – π interaction or covalent bonding. Anchoring groups play a crucial role for improving the mechanical stability and electronic coupling of the molecule–electrode interface when they are attached to the molecular backbone. Other factors such as inserting spacers between the molecular backbone and anchoring groups could optimize the molecule–electrode interface efficiently. With this

Department of Chemistry and Chemical Engineering, Chalmers University of Technology, SE-412-96 Göteborg, Sweden. E-mail: kasper.moth-poulsen@chalmers.se



review, we present how different anchoring groups can control charge transport properties, switches and facilitate dynamic investigations based on the graphene–molecule–graphene single-molecule junctions (GMG-SMJJs).

Non-covalent bonding between molecule and graphene electrode

Interaction of different anchoring groups with electrodes can directly influence the charge transport behavior of single-molecule devices. Any change of the molecular electrode interface, either electronically or geometrically could significantly change the electrical conductance (G) of single molecules. In order to control the molecule–electrode interface through π – π stacking with a number of different anchoring groups (Fig. 1, structure a–i) has been investigated. In particular, El Abbassi *et al.* obtained a mechanically stable device by anchoring a small number of molecules onto the SiO_2 substrate between two graphene electrodes through a silanization reaction.¹⁸ The molecule was comprised of three main parts: a silane group, a π -conjugated head group and a non-conjugated alkane chain (Fig. 2a). The silane group facilitated the chemical anchoring of the molecule to the substrate *via* covalent bonding. This covalent anchoring of the molecule to the substrate overcame the problem of ill-defined edge terminations in graphene that can occur after nanofabrication or preparation of the contact electrodes. Additionally, covalent bonds between the molecule and the substrate led to stable graphene–molecule junctions. The π -conjugated head group biphenyl *N*-carbazole (BPC; Fig. 1a) ensured coupling to graphene through π – π interaction. The non-conjugated alkane chain decoupled the mechanical anchoring from the electronic head group. To reveal the transport characteristics at the molecule–electrode interface in the device, density functional theory (DFT) calculations have been performed. DFT analysis indicated that the frontier orbitals of



Fig. 2 (a) Molecular structure. (b) Schematic of the molecular junction including molecules bridged *via* different π – π stacking into graphene. Reprinted with permission from El Abbassi *et al.*, *Nat. Nanotechnol.*, 2019, **14**, 957. Copyright 2019 Nature Publishing Group.

the BPC molecule were mostly localized on the head group. This shows that the electron transport occurs mainly through the head group. They also found the evidence for the formation of transport channels between the head groups of two neighboring molecules through π – π stacking in the graphene nanogap (Fig. 2b). Apart from the BPC molecule, they did electrical characterization of devices with other headgroups, *i.e.*, molecule C with a carbazole (Fig. 1b) and molecule M without a π -conjugated head group. Current–voltage (I – V) curves showed a maximum current of about 10 picoampere (pA) at a bias voltage (V_b) of 2 V for the junction using molecule M. I – V curves for the junctions using molecule C illustrated efficient charge transport over the molecular junction through π – π stacking. The alteration in I – V shapes and current levels were explained by the weak electronic interaction between the



Fig. 1 Examples of non-covalent anchoring groups used in graphene-based devices. (a) Biphenyl *N*-carbazole, (b) Carbazole, (c) Fullerene, (d) Pyrene, (e) Amine, (f) Pyrene-1, (g) Pyrene-2, (h) Tetrabenzofluorene, (i) Anthracene.



neighboring head groups, which were arranged in various molecular conformations.

Tan *et al.* realized a different approach for investigation of charge transport through a class of junctions using the mechanically controlled break junction (MCBJ) technique.¹⁹ They fabricated single-fullerene (Fig. 1c) junctions with a fullerene bridged between two graphene electrodes *via* van der Waals interactions. A one-dimensional (1D) conductance histogram of graphene/single-C₆₀, C₇₀, C₇₆, C₉₀/graphene junctions indicated that the conductance improved >250% from $2.0 \times 10^{-5} G_0$ to $5.1 \times 10^{-5} G_0$ when the size of the fullerene was changed from C₆₀ to C₉₀. Furthermore, they observed that the conductance was enhanced after the fullerene was doped with a nitrogen atom. For example, the conductance of graphene/single-(C₅₉N)₂/graphene increased by 150% reaching $2.1 \times 10^{-5} G_0$ compared to the conductance of graphene/single-C₁₂₀/graphene junctions. However, when breaking the conjugation of the π -system of fullerene, the conductance decreased. The conductance of low-conjugated graphene/single-C₅₀H₁₀/graphene junctions decreased by about one order of magnitude compared to the conductance of conjugated single-C₆₀.

The charge transport in a series of C₆₀-graphene junctions has been investigated by Leitherer *et al.* using DFT and Landauer transport theory.²⁰ As building blocks of the molecular junctions, they used a molecular bridge between two C₆₀ molecules connected by two styrene groups covalently attached to the [2,2] paracyclophane moiety in the center of C₆₀. The two fullerene anchor groups of this structure were bound to graphene electrodes through π - π interaction.²¹ They found out two structural conformers with either zigzag- or armchair-terminated graphene nanocontacts (Fig. 3a and b) and investi-

gated how electronic transport is influenced by graphene edge terminations. Interestingly, the edge-induced transport contributions were only observed in conformer A with zigzag-terminated graphene electrodes. This phenomenon was explained by the differences in the interface geometry and the resulting molecule-graphene contact in conformer A. As it is shown in Fig. 3, for conformer B dimer/bridge and a hexagon hole geometry corresponded to the left- and right-hand side respectively. The binding energy of the hexagon/hole and pentagon/hole orientation, corresponded to conformer A, was higher than that of the dimer/bridge geometry, which resulted the lower molecule-electrode coupling in conformer B. The different coupling strengths in the pentagon and hexagon geometries was corroborated by earlier experimental work by Danilov *et al.*²² who studied voltage-induced switching in single C₆₀ molecules located in the nano gap between two silver electrodes.

Limburg *et al.* explored five derivatives of zinc porphyrin in single-molecule junctions to find out the effect of various non-covalent anchor groups on charge transport properties and the junction formation probability (JFP).²³ The five derivatives were constituted of an identical redox-active zinc porphyrin core and different flat π -aromatic systems as anchor groups. They chose one reference molecule without anchor groups. Tetrabenzofluorene (Fig. 1h) anchors showed different charge transport behaviour compared to the other anchors as they are not π -conjugated with the porphyrin. The JFP is highly depended on the anchor group. The larger anchor groups indicated higher binding energy which corresponded with higher JFPs of molecular devices. Specially, in this experiment, they showed that larger anchor groups had a significant positive effect on JFP. For example, 1,3,8-tridodecyloxyppyrene and 2,5,8,11,14-pentadecylhexa-*peri* hexabenzocoronene had the highest JFPs values of about 36% and 38%, respectively. They also showed that additional alkyl tails on the π -systems played the role for better interaction of the π -system to the graphene. The large binding energies and high JFP of larger anchor groups make them as a potential candidate for fabrication of devices based on molecular anchoring to graphene *via* π -stacking.

To investigate electron transport through spin-crossover (SCO) molecules at the single molecular level, Burzuri *et al.* bridged the molecule [Fe(L)₂](BF₄)₂·CH₃CN·H₂O (Fe-SCO) between two few-layer graphene (FLG) electrodes.²⁴ The coordination of two 2,6-bispyrazolylpyridine (bpp) ligands (L) with Fe (II) was through an octahedral (O_h) symmetry distorted to a S₄ symmetry. Two pyrene (Fig. 1d) groups anchored these ligands *via* π - π stacking to graphene electrodes. After deposition of the Fe-SCO molecules, they calculated the current (*I*)-voltage (*V*) characteristics at cryogenic temperatures for three different junctions containing an Fe-SCO molecule. When a molecular junction was formed, the current was increased sharply around two orders of magnitude compared with the empty junction. In these three junctions, they observed a reproducible conductance bi-stability between a large-gap (LG) and a small-gap (SG) state. This observation confirmed that the spin-crossover switch was active in these single-molecule junctions.



Fig. 3 Interface geometry of (a) conformer A and (b) conformer B. Reprinted with permission from Leitherer *et al.*, *Nanoscale*, 2017, 9, 7217. Copyright 2017 Royal Society of Chemistry.



DFT analysis indicated that the SCO transition induced a considerable change in the energy spectrum of the molecular orbitals led to conductance bistability. It has been shown that small changes of the ligand distance in the molecular junctions could induce switching between high-spin (HS) and low-spin (LS) states at the single-molecule level at any temperature. Lau *et al.* studied the charge transport properties across a graphene–fullerene single-molecule transistor using a C₆₀ bisadduct functionalized with pyrene as an anchoring group. Two pyrene anchor groups were bound to graphene nanoelectrodes *via* π – π interactions.²⁵ DFT calculations revealed that the most stable configuration was the *cis*-3 isomer of relaxed graphene–fullerene–graphene structure without steric hindrance. The iso-surfaces of the frontier orbitals for *cis*-3 and *cis*-2 indicated that the *cis*-3 isomer showed better π – π interaction with the graphene electrodes. It should be noted that the transport in pristine C₆₀ was dominated by the lowest unoccupied molecular orbital (LUMO), whereas functionalization of C₆₀ with pyrene anchoring groups modified the electron delocalization and energy levels, which led to transport dominated by the highest occupied molecular orbital (HOMO). Due to the stability of the device which showed weak vibron-relaxation, they investigated redox-dependent electron–vibron coupling and avalanche transport.

To identify the influence of different anchoring groups on the transport through graphene-based molecular junctions, Sadeghi *et al.* theoretically determined the electronic properties of zinc-porphyrin monomers, dimers, and trimers attached to graphene electrodes through either π – π or covalent anchoring.²⁶ The authors' model device structure was comprised of monomer, dimer and, trimer zinc-porphyrin wires coupled to graphene electrodes (Fig. 4a) through either covalent C–C bond attachment to the edge of the graphene ribbons (Fig. 4b), amine anchors (Fig. 4c), π – π stacking through pyrene with two different moieties (Fig. 4d and e) or a tetrabenzofluorene (TBF) anchoring group (Fig. 4f). Direct C–C coupling was found to have the highest conductance, followed by π – π stacking with the pyrene anchors, direct coupling *via* amine anchors, and π – π stacking with TBF. The direct C–C bonds were coupled to the electrodes more strongly than other anchoring groups because of the great width of the HOMO resonance in this junction.²⁷ Depending on the connectivity of the pyrene anchor groups to the porphyrin wire, pyrene anchors could induce either destructive quantum interference or different degrees of constructive quantum interference. However, the coupling of TBF to both the graphene and the porphyrin wires were weak as the twisted structure of the TBF anchors provided a nonuniform π – π overlap between π -orbitals of TBF and the graphene electrodes. Furthermore, the disturbance in the conjugation of the π -aromatic system at the cyclopentadiene connection point led to a reduction in the conductance and electronic coupling. The lower conductance of amine anchors was explained by presence of a carbonyl group, which broke the conjugation of the π -aromatic system.

Prins *et al.* developed a method for the formation of nanometer-separated FLG electrodes by using the feedback-con-



Fig. 4 (a) The porphyrin wires with one, two, and three zinc porphyrin units ($n = 1, 2, 3$) are contacted to the graphene by pyrene anchors. Other anchor groups studied include covalent anchoring to the edge of graphene electrodes *via* (b) a direct carbon–carbon bond and (c) amine anchors or through π – π interactions with (d) pyrene-1 anchors, (e) pyrene-2 anchors, and (f) tetrabenzofluorene (TBF) anchors. Reprinted with permission from Sadeghi *et al.*, *Nano Lett.*, 2017, 17, 4611. Copyright 2017 American Chemical Society.

trolled electroburning technique.²⁸ The high stability of these electrodes was demonstrated by only small variations in the tunneling characteristics after several weeks of storage under vacuum condition. In order to show the long-term stability and gate independent transport across the gap, an anthracene-functionalized curcuminoid molecule (1,7-(di-9-anthracene)-1,6-heptadiene-3,5-dione, 9Accm) was placed in the graphene nanogap. The anthracene groups (Fig. 1i) interacted *via* π – π interactions with the FLG electrodes and the high π -electron density of 9 Accm facilitated the charge transport. The difference in I – V characteristics before and after molecule deposition could be explained by the presence of molecules in the gap influencing its stability. Before deposition, the conductance was gate-independent and after deposition, the conductance of the device improved and gate-dependent transport at room temperature was observed, indicative of molecular device formation.

A more reproducible electrode–molecule contact than for traditional anchoring groups covalently binding to the metal electrode has been achieved by designing a new curcuminoid-based molecule (9ALCcmoid) with a well-defined geometry of aromatic anchoring groups such as pyrene and anthracene.²⁹ 9ALCcmoid was comprised of a conjugated linear chain of 11 C atoms which included a β -diketone group at the center of the backbone and two anthracene groups situated on both sides of the chain. The molecule–electrode interface was controlled by π – π stacking of the anthracene groups to the FLG electrodes. I – V curves revealed that the current was reduced during controlled electroburning of a typical device in each step. Modelling of the structure indicated that the



9ALCccmoid molecule was positioned on top of the vicinity of each graphene electrode with one anthracene group. An energy minimum observed with a negligible relative horizontal shift of the graphene electrodes could be the reason for the non-symmetric final structure.

The charge transport through a stable single-molecule single-electron transistor (SET) was investigated by Mol *et al.*³⁰ They used a molecular backbone of a zinc-porphyrin moiety connecting to 'butterfly' TBF anchoring groups with a spacer, which facilitated the binding *via* π - π -stacking to defect free graphene nano-electrodes. DFT analysis revealed no sign of steric hindrance to adsorption as well as a planar geometry of the molecular wire across the graphene nanogap. A wide range of molecular wires can possibly be used for the π - π anchoring to the highly stable graphene electrodes.

Covalent bonding between molecule and graphene electrodes

Apart from studying the charge transport properties within single molecule electronics, there are several reports using GMG-SMJ platform for investigation of dynamic processes at the single-molecule level such as host-guest interactions, hydrogen bond interactions, and intermolecular chemical reactions. In these studies, they controlled the molecule-electrode interface by covalent coupling of a wide range of organic molecules have been used with graphene electrodes.

Using of single-molecule junctions for studying reaction mechanism at the single-event level has been recently reported. Gu *et al.* developed a method to use single-molecule junctions (SMJs) as an electrical platform to measure a reversible unimolecular nucleophilic substitution (S_N1) reaction at the single-molecule level.³¹ Stable graphene-molecule-graphene single-molecule junctions (GMG-SMJs) were formed *via* covalent attachment of a single molecular wire comprised of a functional center of 9-phenyl-9-fluoreneol into graphene nanogapped electrodes (Fig. 5). To determine the acid-catalyzed S_N1 reaction mechanism, the authors applied 9-phenyl-9-fluoreneol as a reactant in the mixture of acetic acid (HAc) and trifluoroacetic acid (TFA). They found that the reaction proceeded in two steps. First, when the leaving group departed from the reactant, the carbocation formed an intermediate. Then, a nucleophile attached to the carbocation intermediate to form the final product. The acidic environment was assumed to stabilize the carbocation intermediate. For further investigation of reaction mechanism, competitive reactions between different nucleophiles were performed. In consistent with this, they studied the role of bromide anion in solution as another nucleophile. After monitoring reactions between the carbocation and different nucleophiles, the authors found different rate constants for the heterolytic dissociation process of the acetate/bromide forms (k_d) and the combination process of the carbocation with acetic acid and bromide (k_c). After comparing the reaction condition without the addition of the bromide anion, the authors confirmed that carbocation was an inevitable



Fig. 5 Schematic of a graphene-molecule-graphene single-molecule junction which indicates the formation dynamics of a 9-phenyl-9-fluorenyl cation in an S_N1 reaction. Reprinted with permission from Gu *et al.*, *Nano Lett.*, 2018, 18, 4156. Copyright 2018 American Chemical Society.

intermediate in the transition between the bromide and acetate forms, indicative of a classical S_N1 mechanism. The authors were able to distinguish the carbocation electrically from the reactant and product because the conductance increased when the hybridization was changed from sp^3 to sp^2 .

To investigate the dynamic process of basic chemical reactions, Guan *et al.* also took advantage of GMG-SMJs.³² The authors investigated the reaction dynamics between carbonyl groups on GMG-SMJs and NH_2OH in solution using a micro-channel, in which the solution was confined to a small channel area. They were able to observe a base-catalyzed nucleophilic addition reaction between the carbonyl group on the molecule and NH_2OH in solution. After the loss of an H_2O molecule from the intermediate, oxime was formed as a final product. The junctions were constructed *via* covalent bonding of a single molecule with 9-fluorenone in the center to graphene electrodes. The authors suggested that the noticeable stability of the GMG-SMJs was owing to the formation of amide linkages by covalent coupling of amino groups to graphene electrodes. To gain more insight into the formation of the GMG-SMJs, I - V curves of the devices were recorded. The current was observed to decrease after etching by oxygen plasma. However, the recovery of the current after molecular assembly indicated successful device fabrication. Computationally calculated reaction energy profiles indicated the transformation of a reactant state (RS) to a shallow intermediate (IS'), and subsequently, to a stable intermediate state (IS) through proton transfer in water. Although a few microseconds were required for the reversible reaction from RS to IS, the reverse reaction from the final product state (PS) to IS took thousands of hours. Hence, high- and low-conductance states with dwell times of a few microseconds corresponded to the reversible reaction between the RS and IS.

To investigate hydrogen bond dynamics at the molecular level, Zhou *et al.* constructed hydrogen bond-bridged SMJs (HBB-SMJs) using promising strategy in which a ureido pyrimidine-dione (UPy) monomer molecule was covalently coupled between graphene point contacts.³³ A UPy molecule can easily



dimerize in a weak polar solvent such as diphenyl ether. In this study, A UPy dimer played the role as a good conducting channel and facilitated the formation of a strong donor–donor–acceptor–acceptor array of intermolecular quadruple hydrogen bonding. However, changing to strong polar solvents such as water, diphenyl ether could lead to on–off switching of the conductance of HBB-SMJs. This result indicated the successful fabrication of the devices. After studying the obtained current–time ($I-t$) curves in 1,1,2,2-tetrachloroethane (TeCA) at 293 K, they observed a series of large-amplitude two-state fluctuations with considerable dependency to temperature and solvent dependence. It has been shown that compared to the ‘high’ state conductance, the conductance of the ‘low’ state was much smaller which indicated a weaker hydrogen bonding system. The average lifetimes for the high-conductance and low-conductance states were reported $\tau_{\text{high}} = \sim 0.660 \pm 0.042$ ms and $\tau_{\text{low}} = \sim 0.581 \pm 0.042$ ms respectively. In order to clarify the dynamic mechanism, they studied the experiments in two solvents with three different temperatures. They observed two large-amplitude peaks in TeCA. However, a multimodal distribution (at least four obvious peaks) was detected in diphenyl ether suggesting different mechanisms of intermolecular interactions through hydrogen bonding in different solvents. Both experimental and theoretical results revealed that a multimodal distribution could arise from the stochastic rearrangement of the hydrogen bond structure mainly through intermolecular proton transfer and lactam–lactim tautomerism. This work shows a promising strategy for investigation of hydrogen bond dynamics at the single molecule level.

To study the capability of particular GMG-SMJs to discover thermodynamic and kinetic parameters of a host–guest complex, Wen *et al.* constructed a conjugated molecular wire featuring bis-*p*-phenylene [34] crown-10 (BPP34C10) terminated with two amine handles namely, BPP34C10DAM that was covalently attached to carboxylic acid–terminated graphene point contacts using a 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide (EDCI) coupling protocol.³⁴ With this interface, they were able to transduce the physical pseudorotaxane (de)formation processes between the electron-rich crown ether and a dicationic guest into real-time electrical signals. To gain more insight to the influence of host–guest interactions on the conductivity of molecular junctions, they started the complexation by introducing methyl viologen (MV^{2+}), which widely used as a guest for the BPP34C10 host, into the GMG-SMJ system. After immersion in a Me_2SO solution containing 1 mM $MV \cdot 2PF_6$ for 12 hours. The $I-t$ curves of resulting devices ($MV^{2+} \subset BPP34C10$ -SMJ) indicated two distinct ‘high’ and ‘low’ states. With the knowledge of a change in conductance in the solid state after MV^{2+} addition, the high and low conductance states could be attributed to the association and dissociation processes between the BPP34C10 host and MV^{2+} guest molecules at the device–liquid interface. For further investigation, they calculated the transmission spectra of both the BPP34C10 host and the $MV^{2+} \subset BPP34C10$ pseudorotaxane in the SMJ by using an equilibrium Green’s function technique based on DFT. The theoretical

result confirmed that electron (LUMO) transport was predominant for both BPP34C10 and the $MV^{2+} \subset BPP34C10$ complex in the graphene-based system. However, the LUMO of the $MV^{2+} \subset BPP34C10$ pseudorotaxane SMJ was more delocalized, indicative of a better conductive channel. This study can provide this opportunity for investigation of stochastic and dynamic nature of molecular recognition processes in both chemical and biological system.

In a similar approach, Zhou *et al.* observed the dynamic process of stochastic supramolecular interactions at the single-molecule level using unique GMG-SMJ.³⁵ The junctions were fabricated through covalent coupling of a conjugated molecular wire with a macrocyclic side group (permethylated cyclodextrin (PM- α CD) to graphene electrodes (Fig. 6). Each monomer served as a bridge to connect the amino moiety of the conjugated molecular wire to the carboxylic groups at the edge of the graphene point contacts using EDCI coupling in pyridine. To investigate the dynamic mechanism, they immersed GMG-SMJs in a water solution containing three types of dodecamethylene units namely dodecanedioic acid, 1,12-dodecanediamine, and 12-aminododecanoic acid with different charges (negative, positive, or amphoteric on both chain ends, respectively) as the shuttle component. Large-amplitude two-level or three-level conductance states have been observed in the current distribution histograms. Both experimental and theoretical results indicated that the pseudorotaxane movement process between PM- α CD and three alkylene chains with different charges led to three distinct reproducible large-amplitude signal oscillations within the time scale of a few milliseconds.

In order to design novel single molecule transistors, Xu *et al.* covalently attached individual conductive molecules including 1,4-diaminobenzene (1Ph) and 4,4'-diaminobiphenyl (2Ph) to the oxidized edges of electroburnt graphene nanogaps.³⁶ They used reaction chemistry to bind individual conductive molecules to the oxidized edges of electroburnt gra-



Fig. 6 Schematic of the device structure of GMG-SMJs formed by individual molecular shuttles. Reprinted with permission from Zhou *et al.*, *Small Methods*, 2019, 3, 1900464. Copyright 2019 Wiley-VCH Verlag GmbH & Co. KGaA.



phene nanogaps. The shape and size of the graphene nanogaps provided a pathway to optimize the coupling chemistry. This led to high connection yields with ultrashort covalent single-molecule bridges (up to 40%). Therefore, extremely short molecules (as short as 0.6 nm in length) could find a way to connect into the electrical circuit chemically. The reconnection was observed after increasing the conductance across the nanogap using gate-dependent single-electron tunnelling properties. They realized that the nature of the molecule influenced the yield of reconnection.

An approach by Xin *et al.* presents a new avenue for engineering high-performance electronic devices based on GMG-SMJs with an ionic liquid as a gate dielectric modulating the charge transport efficiently.³⁷ To gain more insight into the effectiveness of ionic liquid gating, three different phenyl-based molecular structures with various molecular lengths including biphenyl, triphenyl and hexaphenyl molecules were covalently coupled to graphene electrodes through amide linkages (Fig. 7). The conductance of the single-molecule junctions decreased when the length of the molecular backbone was increased. These three devices indicated gate-controlled conductance behaviour. It has been shown that the molecular length influenced the gate efficiency. When the gate voltage was changed from -1.5 V to 1.5 V, both triphenyl and hexaphenyl molecular junctions showed ambipolar charge transport characteristics. This phenomenon explained by changing of

the dominant conducting orbital from HOMO to LUMO when the gate voltage was changed from negative to positive.

Sun *et al.*³⁸ fabricated reliable junctions through a chemical reaction between the amino-terminated hexaphenyl aromatic molecule 4,4'-diaminohexaphenyl (DAHP) and the carboxyl groups at the edges of single-layer graphene nanoelectrodes. Hexaphenyl molecules are covalently linked to the graphene electrodes through amide linkages. They recorded I - V curves of graphene-hexaphenyl-graphene (GHG) junctions at various points after fabrication. The I - V curves before and after DAHP molecule connection were compared. When one or few molecules have been interlinked the nano-gaps, a distinct current increase was observed. In order to determine of stability of single-molecule devices, they recorded I - V curves of the same graphene-hexaphenyl-graphene (GHG) junctions at various points after fabrication. The authors did not observe any remarkable changes in the I - V curves of the GHG junctions measured immediately after fabrication, and two and four weeks after fabrication. It has been shown that more than 52% of the devices could remain stable for up to four weeks at room temperature.

Several papers have reported to date on the design and fabrication of molecular photoswitches incorporated into graphene devices. For example, Jia *et al.* observed reproducible conductance switching activated by external light on a new platform of graphene-molecule junctions (Fig. 7a).³⁹ The authors used three photochromic diarylethene derivatives with different substituents that could transition reversibly between two distinct isomers with open/closed conformations (Fig. 7b). In order to better understand the molecular energy levels at the molecule-electrode contact, they modified the interface between the graphene electrodes and the functional photo-switch by introducing methylene spacer groups. Introducing the methylene groups disturbed the π -electron delocalization between the terminal amine group and the functional centre on each side of the molecules, which led to a drop in the strong electronic coupling between the diarylethene molecules and the electrodes. The energy levels of the diarylethene derivative substituted with methylene groups were lower than those of the other two derivatives. This finding indicates that in single-molecule junctions, the on/off conductance ratios depend on the side groups outside the switching core.

In addition, Jia *et al.* reported the formation of stable molecular junctions by incorporating three methylene (CH₂) groups covalently bound to graphene electrodes on either side of the molecular backbone (Fig. 8).⁴⁰ Theoretical analysis revealed that introduction of methylene groups as molecular bridge, resulted in reduced electrode-molecule coupling which avoided the quenching of the excited molecular state by electrodes. This changed the transport regime from the Landauer regime without the CH₂ groups to the Coulomb blockade regime. Furthermore, they demonstrated that the molecular junctions could perform reversible conductance photo switching in a robust way. However, the stochastic switching of the closed junction was only observed between two conductive states for the closed form between 160 to 240 K.



Fig. 7 (a) Switching of graphene-diarylethene junctions. (b) Molecular structures of three photochromic diarylethene. Reprinted with permission from Jia *et al.*, *Angew. Chem., Int. Ed.*, 2013, 52, 8666. Copyright 2013 Wiley-VCH Verlag GmbH & Co. KGaA.





Fig. 8 Schematic of a graphene-diarylethene-graphene junction that highlights the expansion of the molecular bridge by methylene groups. Reprinted with permission from Jia *et al.*, *Science*, 2016, **352**, 1443. Copyright 2016 AAAS.

A different type of photoswitch based on a GMG-SMJ was introduced by Meng *et al.* using an azobenzene derivative, a common photochromic material with two states (*cis/trans*).⁴¹ Individual 2'-*p*-tolyl diazenyl-1,1':4,4'-terphenyl-4,4''-dicarboxylic acid (TTDA) molecules, were covalently attached to graphene electrodes through amide bond formation. Both the *cis* and *trans* forms of the azobenzene unit affected the electronic structure of the whole molecule. Isomerization between the states could be induced under irradiation of different wavelengths of light (UV/visible) causing different resistance display by different isomers. Both theoretical and experimental analysis confirmed that an electric field could modulate not only the energy difference between the *trans* and *cis* forms, but also the energy barrier of conformational changes.

Wu *et al.* theoretically determined the switching properties in a class of porphyrin molecules with pendant dipoles.⁴² Device fabrication involved covalent coupling of porphyrin to graphene electrodes. The porphyrin moiety was gated by a third electrode. This study revealed that by altering the number of C–C triple bonds (used as spacer units between porphyrin and graphene) from one to four, the room temperature on/off conductance ratio changed from 100 to 200. When the temperature was reduced to 100 K, this switching ratio enhanced approximately to 2200. These findings indicate that porphyrin molecular junctions can be tuned systematically by adjusting the number of spacer units.

Xin *et al.* investigated the stereoelectronic effect of biphenyl molecules on its electrical conductivity using robust single-molecule junctions. For this, they covalently coupled a specifically designed hexaphenyl aromatic molecule *via* amide bonds to graphene electrodes.⁴³ Both theoretical and temperature-dependent experimental analysis revealed that both π - π overlap differences between the phenyl rings and two dihedral angles between the phenyl rings might result in strongly conjugated (high conductance) and weakly conjugated (low conductance) states (Fig. 9). They realized that the occurrence of stochastic switching was highly dependent on the variation in



Fig. 9 Schematic of conformational transition of terphenyl units between strongly conjugated and weakly-conjugated states and corresponding hexaphenyl aromatic chain single-molecule junctions. Reprinted with permission from Xin *et al.*, *Nano Lett.*, 2017, **17**, 856. Copyright 2017 American Chemical Society.

temperature. *I*-*V* analysis indicated that stochastic switching occurred between two conductance states when the temperature was increased to 120 K. It should be noted that stochastic switching at positive biases is stronger than that at negative biases. They concluded that is the asymmetric switching could be explained by the asymmetry in the two sides of graphene contacts or other unknown reasons.

Graphene nanoribbons (GNRs) have also been used as electrodes for fabricating molecular devices. Unique electronic structures and transport characteristics in GNRs make them potential candidates for investigating different physical properties, such as rectification behavior negative differential resistance (NDR), switching and spin filter effect.

To investigate the conductance behavior and rectifying performance of a series of molecular diodes, Song *et al.* used chemical doping in an armchair edged GNR. They constructed a series of molecular diodes based on an azulene-like dipole molecule capped with chains of carbon atoms between two types of symmetrically arranged electrodes, (i) undoped zGNRs and (ii) doped armchair graphene nanoribbons (aGNRs).⁴⁴ Dopants included the elements in groups III and V, boron, aluminum, nitrogen, and phosphorus located in the center positions. In addition, two carbon atoms per two carbon unit cells in an aGNRs electrode were replaced by two B or N atoms. After applying doping to the aGNR electrodes, they observed considerable rectifying behaviors with rectification ratios up to the order of 10^5 . Compared to phosphorus dopants, the rectifying performance of aGNR electrode doped with boron and aluminum was enhanced.

Li *et al.* investigated the spin-dependent electron transport through a junction comprised of a single pyridine-linked (PDL) molecule with a π - σ - π building block coupled between two zigzag-edged graphene nanoribbon (zGNR) electrodes.⁴⁵ For this study, they considered three different molecules including 4,4'-bipyridine, 4,4'-vinylenedipyridine, and 4,4'-ethylenedipyridine. DFT calculations combined with the nonequilibrium Green's function (NEGF) formalism studies indicated that the performance of different magnetic configurations in the zGNR electrodes affected the spin-charge transport modulation. They observed the spin-filtering within the parallel (P)



spin configuration and dual spin-filtering effect with antiparallel (AP) spin configuration. It is worth noting that for the junctions including a 4,4'-bipyridine molecule, the rectification ratio (RR) and the NDR effect with peak to valley ratio (PVR) could reach up to 10^4 and 328, respectively.

Wang *et al.* investigated the influence of intramolecular proton transfer on the current of single-molecule devices based on keto and enol tautomers.⁴⁶ They used 6,11-dioxo-5,6,11,12 tetrahydrobenzo[*b*]phenazine-1,4,7,10-tetracarbonitrile (DO-THBbPA-TCN) molecule attached covalently between two zigzag graphene nanoribbon (zGNR) electrodes (Fig. 10).⁴⁷ The interconversion between the keto and enol forms occurred through intermolecular proton transfer. They found that compared to keto molecular devices, the switching effect was enhanced for the enol molecular device with an on/off ratio of up to 3.4×10^2 .

To understand bidirectional conductance switching, Han *et al.* theoretically studied the molecule–electrode interface of reversible molecular photoswitches based on diarylethene molecules.⁴⁸ The molecular junctions were designed and built based on the ring-closed and ring-open forms of diarylethene molecules that were sandwiched between two monolayer graphene nanoribbons (GNRs) and two Au (111) electrodes. The photoswitching circle of the open and closed isomers was activated in both directions using irradiation with visible light. They found that the electrical conductivity and switching behavior depends on the electronic structure of molecule in both the GNRs and Au electrode. Theoretical predications indicated that the on/off current ratio of the devices could be as large as 10^3 .

To perform two-terminal transport measurements, Li *et al.* designed and fabricated atomically precise single-molecule devices based on covalent bonding of a magnetic porphyrin with GNR electrodes (Fig. 11).⁴⁹ They used scanning tunneling microscopy to contact the end of a linear GNR–porphyrin–GNR hybrid system *via* on-surface synthesis (OSS). Transport investigation revealed the spin-excitation spectral fingerprint



Fig. 10 Schematic of investigated molecular junctions, where the keto and enol molecules are covalently fused into the zGNR electrodes. The left and right electrodes are denoted by the red and blue rectangles, respectively. Reprinted with permission from Wang *et al.*, *Nano Lett.*, 2018, **31**, 17650. Copyright 2018 American Chemical Society.



Fig. 11 Illustration of the target experimental geometry. The inset shows a typical conductance–distance plot simultaneously recorded during tip approach and retraction. Reprinted with permission from Li *et al.*, *Nano Lett.*, 2019, **19**, 3288. Copyright 2019 American Chemical Society.

was dependent on the connecting patterns between GNR and single molecule (SM).

Carbon nanotubes

Carbon nanotubes (CNTs) have been used both as electric contacts at junctions and for controlling the geometry at the molecule–electrode interface.^{50,51} In several papers, the use of CNTs as electrodes in single molecule electronics has been reported. For example, Zhu *et al.*⁵² introduced a strategy for the fabrication of solution-processable molecular transport junctions (MTJs) covalently coupled to metallic CNT electrodes using molecular construction. DFT calculation indicated that linear junctions with two molecules embedded between two CNTs were energetically less favorable than junctions with one bridging molecule. A second molecule in the junction induced a significant strain, which increased the formation energy by $\sim 68 \text{ kJ mol}^{-1}$ (Fig. 12). This method of fabricating MTJs could enable in future the design of different solution-processable nanoelectronics devices.

Zoppi *et al.* conducted a theoretical study to investigate charge transport phenomena in molecular junctions with corannulene ($\text{C}_{20}\text{H}_{10}$) covalently coupled between two CNTs.⁵³ They investigated the influence of various design factors, such as spacer linking composition, the constitution of the active molecule element, the sensor capabilities of the active element, and the response of the nanojunction to an external electric field, on the electron transport through the CNT–[spacer]–*f*– $\text{C}_{20}\text{H}_{10}$ –[spacer]–CNT junctions. Similarly, Sandonas *et al.*⁵⁴ developed a comprehensive computational study indi-





Fig. 12 Schematic of a PPD (*p*-phenylenediamine)-linked molecular junction and reactions of formation for linear junctions with one and two bringing molecules. Reprinted with permission from Zhu *et al.*, *J. Am. Chem. Soc.*, 2016, **9**, 2905. Copyright 2016 American Chemical Society.

cating that heterogeneous molecular junctions, including molecular wires embedded between two different CNTs, could play the role of a selective phonon filter. They designed nanojunctions containing two colinear (6,6)-nanotubes (NTs) joined by a central molecular structure as a selective molecular-scale phonon filter.

Significant studies have been investigated in connection with biomolecular interactions at the single-molecule level for different biotechnology applications.⁵⁵ Vernick *et al.* fabricated point-functionalized carbon nanotube transistor to measure the kinetics and thermodynamics of DNA (Deoxyribonucleic acid) hybridization for finding potential applications in

genomic identification.⁵⁶ They covalently attached an amine-modified oligonucleotide probe to a single defect site, generated on the single-walled carbon nanotubes (SWCNT) sidewall. In this study, the CNT played the role as the channel of a field-effect transistor.

In order to measure the conductivity a single DNA duplex coupled with a carbon electrode, Guo *et al.* integrated two amine-modified DNAs to the SWCNTs electrodes through covalent bonds.⁵⁷ They observed modification of DNA sequences with amines on either the 5' terminus or both the 3' and 5' termini led to a resistance on the order of 1 M Ω . Likewise, Sorgenfrei *et al.* covalently attached the probe DNA, terminated with an amine group and a three-carbon linker at the 5' end to the point defect in a carbon nanotube.⁵⁸ It has been shown that this novel single molecule label-free bioanalytical system could probe molecular kinetics at microsecond timescales.

Freeley *et al.* conducted a study to investigate the covalent coupling of single proteins to the terminal ends of individual SWCNTs using an in-solution methodology.⁵⁹ They designed two green fluorescent protein (sfGFP) variants to introduce an azide CNT-anchoring handle. The two GFP mutants (short axis, GFP^{SA} and long axis, GFP^{LA}) were constructed to form ssDNA/SWCNT-GFP hybrids with specific protein orientations. Fluorescence investigation revealed the evidence of site-specific coupling between the SWCNTs and the proteins for only the SWCNT-GFP^{SA} nanohybrids. Recently, Thomas *et al.* reported the covalent coupling of four different fluorescent proteins to CNT side walls through phenyl azide photochemistry (Fig. 13).⁶⁰ Depending on the position of the protein on the CNT side wall, GFP function switched from GFP^{LA} and GFP^{SA}, which resulted in a change of the fluorescence wavelength.

Choi *et al.* developed an approach to monitor single-molecule lysozyme dynamics by attaching the molecule of interest to a field-effect transistor (FET) device using a noncovalent immobilization based on pyrene linkers.⁶¹ The binding of pyrenes to SWNTs through π - π stacking might pave the pathway for dilute anchor points for further derivatization of the surface. In this study, a thiol from a single cysteine variant of T4 lysozyme (S90C) was covalently coupled to a pyrene male-



Fig. 13 Direct protein attachment to CNTs via phenyl azide photochemistry. Scheme of photochemical insertion of a protein via genetically encoded phenyl azide chemistry. Reprinted with permission from Thomas *et al.*, *Bioconjugate Chem.*, 2019, **3**, 584.





Fig. 14 Image of $\text{DySc}_2\text{N}@C_{80}$ (left) and $\text{DySc}_2\text{N}@C_{80}$ encapsulated in an SWCNT (right). Reprinted with permission from Nikanishi *et al.*, *J. Am. Chem. Soc.*, 2018, **35**, 10955. Copyright 2018 American Chemical Society.

imide anchor site. Recently, Nakanishi *et al.* encapsulated single-molecule magnets (SMMs) based on the endohedral metallofullerene $\text{DySc}_2\text{N}@C_{80}$ into the internal 1D nanospace of single-walled carbon nanotubes (Fig. 14).⁶² This succeeded because of the strong π - π interactions of SWCNTs with encapsulated fullerenes which are commonly called “peapods” and have shown to be able to modulate electrical properties in SWCNTs.

Conclusions and outlook

As discussed above, different methods have been investigated for fabrication of reliable single molecule electronics. However, there are still several challenges when it comes to rationale design and engineering of stable carbon-based molecular junctions.

One of the key issues is device stability which arises due to the oxidation of metallic electrodes in air and electromigration. This results in poor reproducibility of the formation of metal–molecule–metal junctions. The strong sp^2 C–C covalent bonds in graphene can increase the possibility of stable nanogaps formation. It should be mentioned here that considerable device stability of carbon-based molecular junctions formed *via* covalent amide linkage has been observed up to 4 weeks at ambient temperature.

Another formidable challenge is controlling the molecule–electrode interface. Compared to metallic electrode, the use of (sp^2 -) carbon-based materials allows a wide range of possibility to anchor different molecules. The π -stacking of large π -conjugated anchor groups can mitigate the differences in the level broadening (Γ) by averaging the interaction over a larger interfacial area and also leads to improvements in device yield.²³ The modification in the molecule–electrode coupling changes the level broadening (Γ) and subsequently the conductance through the junctions. However, there are still some questions regarding how different parameters such Γ is modulated under the influence of the edge chemistry of the graphene. Manipulating the structure of molecular bridges by adding more side groups and spacers might be another potential solution. The modification of the interface between the graphene and electrodes by incorporating spacer groups, the on/off conductance ratios for a series of photoswitches were shown to be changed significantly. This finding can serve as a

promising future for reliable fabrication of photoswitch based on a GMG-SMJ.

To achieve reproducible atomic size fabrication of nanogaps, single-molecule electronics can be a potential alternative to Si-based microelectronics in the future. Recently, works on dynamic process at the single molecule level including host–guest interactions, hydrogen bond interactions and studying of physical, chemical, and biological processes using the GMG-SMJ platform have been reported. Therefore, this field of research is growing to introduce brand new functions which can lead to new applications. Recently there are several reports regarding the interaction of 2D materials with biomolecules such as DNA.^{63,64} DNA sequencing could be assessed through the formation of nanogap by the lateral heterostructure of graphene and hexagonal boron nitride (h-BN) classified as 2D materials.⁶⁵ The passivation of the edge of graphene electrode with h-BN could be a viable solution for addressing the practical challenges when it comes to high reactivity and controlling the engineering of graphene edge structure. In this context, a recent development that may create new avenues for single molecule electronics is transition metal dichalcogenide materials that can be manipulated with atomic precision to lead point contacts.⁶⁶

Graphene nanostructure have been widely used as single-molecule biosensors with focus on DNA sequencing applications. One notable issue which should be taken into account is noise characteristics of graphene structures with sizes on the order of a few nanometres. Recently, Fried *et al.*, investigated the noise characteristics of a graphene quantum dot at room temperature in liquid environments.⁶⁷ Unlike reported studies,⁶⁸ they observed that the noise was significantly independent of the liquid environment and the substrate. In this study, the noise was attributed to charge trapping and detrapping at the disordered graphene edges. In order to improve the potential of graphene nanodevices in applications such as single molecule biosensing, the high noise level and presence of two-level fluctuations should be carefully addressed. This issue serve as another challenge in this field as some studies on the noise characteristics of graphene structures have been reported at cryogenic temperatures and under high vacuum.^{68,69}

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

The authors would like to acknowledge the Chalmers Foundation for financial support.

Notes and references

- 1 M. Y. Li, S. K. Su, H. S. P. Wong and L. J. Li, *Nature*, 2019, **567**, 169–170.



- 2 I. Franco, C. B. George, G. C. Solomon, G. C. Schatz and M. A. Ratner, *J. Am. Chem. Soc.*, 2011, **133**, 2242–2249.
- 3 Z. Donhauser, B. Mantooth, K. Kelly, L. Bumm, J. Monnell, J. J. Stapleton, D. Price, A. Rawlett, D. Allara and J. Tour, *Science*, 2001, **292**, 2303–2307.
- 4 J. L. Zhang, J. Q. Zhong, J. D. Lin, W. P. Hu, K. Wu, G. Q. Xu, A. T. Wee and W. Chen, *Chem. Soc. Rev.*, 2015, **44**, 2998–3022.
- 5 B. Capozzi, J. Xia, O. Adak, E. J. Dell, Z.-F. Liu, J. C. Taylor, J. B. Neaton, L. M. Campos and L. Venkataraman, *Nat. Nanotechnol.*, 2015, **10**, 522–527.
- 6 I. Díez-Pérez, J. Hihath, Y. Lee, L. Yu, L. Adamska, M. A. Kozhushner, I. I. Oleynik and N. Tao, *Nat. Chem.*, 2009, **1**, 635–641.
- 7 J. Park, A. N. Pasupathy, J. I. Goldsmith, C. Chang, Y. Yaish, J. R. Petta, M. Rinkoski, J. P. Sethna, H. D. Abruña and P. L. McEuen, *Nature*, 2002, **417**, 722–725.
- 8 J. Hwang, M. Pototschnig, R. Lettow, G. Zumofen, A. Renn, S. Götzinger and V. Sandoghdar, *Nature*, 2009, **460**, 76–80.
- 9 R. M. Metzger, *Chem. Rev.*, 2015, **115**, 5056–5115.
- 10 P. Song, S. Guerin, S. J. R. Tan, H. V. Annadata, X. Yu, M. Scully, Y. M. Han, M. Roemer, K. P. Loh and D. Thompson, *Adv. Mater.*, 2018, **30**, 1706322.
- 11 Y. Ai, A. Kovalchuk, X. Qiu, Y. Zhang, S. Kumar, X. Wang, M. Kühnel, K. Nørgaard and R. C. Chiechi, *Nano Lett.*, 2018, **18**, 7552–7559.
- 12 L. Cao, L. Yuan, M. Yang, N. Nerngchamnong, D. Thompson, X. Yu, D.-C. Qi and C. A. Nijhuis, *Nanoscale Adv.*, 2019, **1**, 1991–2002.
- 13 L. Sun, Y. A. Diaz-Fernandez, T. A. Gschneidner, F. Westerlund, S. Lara-Avila and K. Moth-Poulsen, *Chem. Soc. Rev.*, 2014, **43**, 7378–7411.
- 14 A. Bergvall, K. Berland, P. Hyldgaard, S. Kubatkin and T. Löfwander, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2011, **84**, 155451.
- 15 Y. Xue and M. A. Ratner, *Int. J. Quantum Chem.*, 2005, **102**, 911–924.
- 16 J. Hihath and N. Tao, *Semicond. Sci. Technol.*, 2014, **29**, 054007.
- 17 C. Jia and X. Guo, *Chem. Soc. Rev.*, 2013, **42**, 5642–5660.
- 18 M. El Abbassi, S. Sangtarash, X. Liu, M. L. Perrin, O. Braun, C. Lambert, H. S. J. van der Zant, S. Yitzchaik, S. Decurtins and S.-X. Liu, *Nat. Nanotechnol.*, 2019, **14**, 957–961.
- 19 Z. Tan, D. Zhang, H.-R. Tian, Q. Wu, S. Hou, J. Pi, H. Sadeghi, Z. Tang, Y. Yang and J. Liu, *Nat. Commun.*, 2019, **10**, 1–7.
- 20 S. Leitherer, P. B. Coto, K. Ullmann, H. B. Weber and M. Thoss, *Nanoscale*, 2017, **9**, 7217–7226.
- 21 K. Ullmann, P. B. Coto, S. Leitherer, A. Molina-Ontoria, N. Martín, M. Thoss and H. B. Weber, *Nano Lett.*, 2015, **15**, 3512–3518.
- 22 A. V. Danilov, P. Hedegård, D. S. Golubev, T. Bjørnholm and S. E. Kubatkin, *Nano Lett.*, 2008, **8**, 2393–2398.
- 23 B. Limburg, J. O. Thomas, G. Holloway, H. Sadeghi, S. Sangtarash, I. C. Y. Hou, J. Cremers, A. Narita, K. Müllen and C. J. Lambert, *Adv. Funct. Mater.*, 2018, **28**, 1803629.
- 24 E. Burzurí, A. García-Fuente, V. García-Suárez, K. S. Kumar, M. Ruben, J. Ferrer and H. S. van der Zant, *Nanoscale*, 2018, **10**, 7905–7911.
- 25 C. S. Lau, H. Sadeghi, G. Rogers, S. Sangtarash, P. Dallas, K. Porfyarakis, J. Warner, C. J. Lambert, G. A. D. Briggs and J. A. Mol, *Nano Lett.*, 2016, **16**, 170–176.
- 26 H. Sadeghi, S. Sangtarash and C. Lambert, *Nano Lett.*, 2017, **17**, 4611–4618.
- 27 C. Lambert, *Chem. Soc. Rev.*, 2015, **44**, 875–888.
- 28 F. Prins, A. Barreiro, J. W. Ruitenbergh, J. S. Seldenthuis, N. Aliaga-Alcalde, L. M. Vandersypen and H. S. van der Zant, *Nano Lett.*, 2011, **11**, 4607–4611.
- 29 E. Burzurí, J. O. Island, R. Díaz-Torres, A. Fursina, A. González-Campo, O. Roubeau, S. J. Teat, N. Aliaga-Alcalde, E. Ruiz and H. S. van der Zant, *ACS Nano*, 2016, **10**, 2521–2527.
- 30 J. A. Mol, C. S. Lau, W. J. Lewis, H. Sadeghi, C. Roche, A. Cnossen, J. H. Warner, C. J. Lambert, H. L. Anderson and G. A. D. Briggs, *Nanoscale*, 2015, **7**, 13181–13185.
- 31 C. Gu, C. Hu, Y. Wei, D. Lin, C. Jia, M. Li, D. Su, J. Guan, A. Xia and L. Xie, *Nano Lett.*, 2018, **18**, 4156–4162.
- 32 J. Guan, C. Jia, Y. Li, Z. Liu, J. Wang, Z. Yang, C. Gu, D. Su, K. N. Houk and D. Zhang, *Sci. Adv.*, 2018, **4**, eaar2177.
- 33 C. Zhou, X. Li, Z. Gong, C. Jia, Y. Lin, C. Gu, G. He, Y. Zhong, J. Yang and X. Guo, *Nat. Commun.*, 2018, **9**, 1–9.
- 34 H. Wen, W. Li, J. Chen, G. He, L. Li, M. A. Olson, A. C.-H. Sue, J. F. Stoddart and X. Guo, *Sci. Adv.*, 2016, **2**, e1601113.
- 35 C. Zhou, X. Li, H. Masai, Z. Liu, Y. Lin, T. Tamaki, J. Terao, J. Yang and X. Guo, *Small Methods*, 2019, **3**, 1900464.
- 36 Q. Xu, G. Scuri, C. Mathewson, P. Kim, C. Nuckolls and D. Bouilly, *Nano Lett.*, 2017, **17**, 5335–5341.
- 37 N. Xin, X. Li, C. Jia, Y. Gong, M. Li, S. Wang, G. Zhang, J. Yang and X. Guo, *Angew. Chem.*, 2018, **130**, 14222–14227.
- 38 H. Sun, Z. Jiang, N. Xin, X. Guo, S. Hou and J. Liao, *ChemPhysChem*, 2018, **19**, 2258–2265.
- 39 C. Jia, J. Wang, C. Yao, Y. Cao, Y. Zhong, Z. Liu, Z. Liu and X. Guo, *Angew. Chem., Int. Ed.*, 2013, **52**, 8666–8670.
- 40 C. Jia, A. Migliore, N. Xin, S. Huang, J. Wang, Q. Yang, S. Wang, H. Chen, D. Wang and B. Feng, *Science*, 2016, **352**, 1443–1445.
- 41 L. Meng, N. Xin, C. Hu, J. Wang, B. Gui, J. Shi, C. Wang, C. Shen, G. Zhang and H. Guo, *Nat. Commun.*, 2019, **10**, 1–8.
- 42 Q. Wu, S. Hou, H. Sadeghi and C. J. Lambert, *Nanoscale*, 2018, **10**, 6524–6530.
- 43 N. Xin, J. Wang, C. Jia, Z. Liu, X. Zhang, C. Yu, M. Li, S. Wang, Y. Gong and H. Sun, *Nano Lett.*, 2017, **17**, 856–861.
- 44 Y. Song, Z. Xie, Y. Ma, Z.-L. Li and C.-K. Wang, *J. Phys. Chem. C*, 2014, **118**, 18713–18720.
- 45 X. Li, L. Cao, H.-L. Li, H. Wan and G. Zhou, *J. Phys. Chem. C*, 2016, **120**, 3010–3018.
- 46 Z.-Q. Wang, M.-Z. Wei, M.-M. Dong, G.-C. Hu, Z.-L. Li, C.-K. Wang and G.-P. Zhang, *J. Phys. Chem. C*, 2018, **122**, 17650–17659.



- 47 D. Weckbecker, P. Coto and M. Thoss, *Nano Lett.*, 2017, **17**, 3341–3346.
- 48 L. Han, X. Zuo, H. Li, Y. Li, C. Fang and D. Liu, *J. Phys. Chem. C*, 2019, **123**, 2736–2745.
- 49 J. Li, N. Friedrich, N. Merino, D. G. de Oteyza, D. Peña, D. Jacob and J. I. Pascual, *Nano Lett.*, 2019, **19**, 3288–3294.
- 50 X. Guo, J. P. Small, J. E. Klare, Y. Wang, M. S. Purewal, I. W. Tam, B. H. Hong, R. Caldwell, L. Huang and S. O'Brien, *Science*, 2006, **311**, 356–359.
- 51 A. K. Feldman, M. L. Steigerwald, X. Guo and C. Nuckolls, *Acc. Chem. Res.*, 2008, **41**, 1731–1741.
- 52 J. Zhu, J. McMorrow, R. Crespo-Otero, G. Ao, M. Zheng, W. P. Gillin and M. Palma, *J. Am. Chem. Soc.*, 2016, **138**, 2905–2908.
- 53 L. Zoppi, A. Ferretti and K. K. Baldrige, *J. Chem. Theory Comput.*, 2015, **11**, 4900–4910.
- 54 L. Medrano Sandonas, Á. Rodríguez Méndez, R. Gutierrez, J. M. Ugalde, V. Mujica and G. Cuniberti, *J. Phys. Chem. C*, 2019, **123**, 9680–9687.
- 55 S. Sorgenfrei, C.-Y. Chiu, M. Johnston, C. Nuckolls and K. L. Shepard, *Nano Lett.*, 2011, **11**, 3739–3743.
- 56 S. Vernick, S. M. Trocchia, S. B. Warren, E. F. Young, D. Bouilly, R. L. Gonzalez, C. Nuckolls and K. L. Shepard, *Nat. Commun.*, 2017, **8**, 1–9.
- 57 X. Guo, A. A. Gorodetsky, J. Hone, J. K. Barton and C. Nuckolls, *Nat. Nanotechnol.*, 2008, **3**, 163–167.
- 58 S. Sorgenfrei, C.-Y. Chiu, R. L. Gonzalez Jr., Y.-J. Yu, P. Kim, C. Nuckolls and K. L. Shepard, *Nat. Nanotechnol.*, 2011, **6**, 126.
- 59 M. Freeley, H. L. Worthy, R. Ahmed, B. Bowen, D. Watkins, J. E. Macdonald, M. Zheng, D. D. Jones and M. Palma, *J. Am. Chem. Soc.*, 2017, **139**, 17834–17840.
- 60 S. K. Thomas, W. D. Jamieson, R. E. Gwyther, B. J. Bowen, A. Beachey, H. L. Worthy, J. E. Macdonald, M. Elliott, O. K. Castell and D. D. Jones, *Bioconjugate Chem.*, 2019, **31**, 584–594.
- 61 Y. Choi, I. S. Moody, P. C. Sims, S. R. Hunt, B. L. Corso, I. Perez, G. A. Weiss and P. G. Collins, *Science*, 2012, **335**, 319–324.
- 62 R. Nakanishi, J. Satoh, K. Katoh, H. Zhang, B. K. Breedlove, M. Nishijima, Y. Nakanishi, H. Omachi, H. Shinohara and M. Yamashita, *J. Am. Chem. Soc.*, 2018, **140**, 10955–10959.
- 63 Q. Li, H. Liu, Y. Tian, J. Guo, G. Chen and J. Y. Lee, *J. Phys. Chem. C*, 2020, **124**, 10823–10831.
- 64 R. L. Kumawat and B. Pathak, *J. Phys. Chem. C*, 2019, **123**, 22377–22383.
- 65 V. Shukla, N. K. Jena, A. Grigoriev and R. Ahuja, *ACS Appl. Mater. Interfaces*, 2017, **9**, 39945–39952.
- 66 B. Munkhbat, A. B. Yankovich, D. G. Baranov, R. Verre, E. Olsson and T. O. Shegai, *Nat. Commun.*, 2020, **11**, 1–8.
- 67 J. P. Fried, X. Bian, J. L. Swett, I. I. Kravchenko, G. A. D. Briggs and J. A. Mol, *Nanoscale*, 2020, **12**, 871–876.
- 68 X.-X. Song, H.-O. Li, J. You, T.-Y. Han, G. Cao, T. Tu, M. Xiao, G.-C. Guo, H.-W. Jiang and G.-P. Guo, *Sci. Rep.*, 2015, **5**, 8142.
- 69 P. Puczkarski, Q. Wu, H. Sadeghi, S. Hou, A. Karimi, Y. Sheng, J. H. Warner, C. J. Lambert, G. A. D. Briggs and J. A. Mol, *ACS Nano*, 2018, **12**, 9451–9460.

