

PAPER

View Article Online
View Journal | View Issue

Cite this: *Nanoscale Adv.*, 2021, 3, 6157

Cooperative transport by flocking phototactic micromotors†

Jianhua Zhang,^{ab} Fangzhi Mou,^{ab} *^a Zhen Wu,^a Jiaqi Song,^b Joshua E. Kauffman,^b Ayusman Sen ^b and Jianguo Guan ^a

Cargo delivery by micro/nanomotors provides enormous opportunities for micromanipulation, environmental cleaning, drug delivery, etc. However, due to the limited driving force, it is usually difficult for single micro/nanomotors to transport cargoes much larger or heavier than themselves. Here, we demonstrate that flocking phototactic TiO₂ micromotors can cooperatively transport multiple and different types of large cargoes based on light-responsive diffusiophoresis. Utilizing spontaneous diffusiophoretic attraction, flocking TiO₂ micromotors can load large cargoes. Under UV light navigation, flocking TiO₂ micromotors cooperatively carry and transport cargoes *via* collective diffusiophoretic repulsion in open space or complex microenvironments. After reaching the destination, the carried cargoes can also be unloaded from the flock and be deployed at a predetermined destination by disassembling or reversing the flock. This study may pave the way for developing intelligent swarming micro/nanorobots for cooperative targeting micromanipulation and advancing their applications in drug delivery and microengineering.

Received 22nd August 2021
Accepted 31st August 2021

DOI: 10.1039/d1na00641j

rsc.li/nanoscale-advances

Introduction

Controllable capture and transport cargoes such as drug molecules, proteins, capsules, and cells in microenvironments is a pivotal nanotechnology that is extensively studied.^{1–6} Synthetic micro/nanomotors can extract various energies from the surrounding environment, and convert them into mechanical output allowing for autonomous motions.^{7–15} Thus, micro/nanomotors are endowed with the undeniable capability to serve as micro/nanocarriers to realize micromanipulation by establishing physical/chemical interactions between the motors and cargoes.^{16–18} Up to now, cargo delivery *via* single micro/nanomotors has been investigated extensively by introducing diffusiophoretic, electrostatic, and magnetic motor–cargo interactions through chemical-powering^{19–21} and external-field-driven strategies.^{22–27} However, the magnitude of the generated driving forces of single motors would be seriously hindered by their dimensions, thus larger or heavier cargoes are difficult to transport. Inspired by the ant army carrying food cooperatively, the driving force would be much enhanced by grouping the motors.^{28–32} Therefore, growing endeavors have been made to construct intelligent micro/nanorobot swarms to cooperatively

transport large cargoes.³³ The key to achieving cooperative transport is to align the forces of active individuals in swarms in the same direction. For example, reconfigurable magnetic motor swarms can regulate their forces effectively in elaborated magnetic fields, thereby manipulating passive cargoes by utilizing hydrodynamic flow.^{34,35} Chemically/photochemically powered micro/nanomotors can respond to signaling chemicals secreted by their neighbors and behave as motor groups utilizing local diffusiophoretic interactions.^{36–42} Nevertheless, aligning the forces of these individuals to effectively manipulate multiple and different types of large cargoes is still an insurmountable challenge.

In this study, we demonstrate the cooperative transport of multiple and different types of large cargoes by TiO₂ micro-motor flocks (TiO₂-MFs) under UV navigation. Specifically, isotropic TiO₂ micromotors with rich hydroxyl groups can autonomously form clusters based on electrolyte diffusiophoretic attraction, which is also capable of capturing heterogeneous passive large cargoes. Meanwhile, the non-electrolyte diffusiophoretic repulsive force resulting from the photo-degrading H₂O₂ by the flocking TiO₂ motors can be aligned toward a determined direction against the UV light. Thus, TiO₂-MFs, even loaded with multiple cargoes, can exhibit negative phototaxis with intense collective expansion. Even though TiO₂-MFs slow down gradually with increasing loads of cargoes, they can transport up to nine large cargoes. In addition, the carried cargoes can also be unloaded from the flock and be deployed at a designated destination by disassembling the flock into highly dispersed micromotors under intense long-time UV irradiation

^aState Key Laboratory of Advanced Technology for Materials Synthesis and Processing, International School of Materials Science and Engineering, Wuhan University of Technology, 122 Luoshi Road, Wuhan 430070, China. E-mail: moufz@whut.edu.cn

^bDepartment of Chemistry, The Pennsylvania State University, University Park, PA 16802, USA

† Electronic supplementary information (ESI) available. See DOI: 10.1039/d1na00641j



or by reversing the flock away from cargoes. TiO_2 -MFs show broad applicability and excellent feasibility for loading and ferrying different types of large cargoes (SiO_2 microspheres, amino-polystyrene microspheres, carboxyl-polystyrene microspheres and perfluorooctane droplets) benefitting from the universal light-switchable motor-cargo interactions (diffusiophoretic attraction and repulsion). The cooperative transport of large cargoes by TiO_2 -MFs in microchannels has also been investigated, manifesting the possibility of realizing cargo transport in microenvironments with complex landscapes. The as-proposed cooperative transport strategy may open a window for the development of general transport systems based on swarming micro/nanorobots to deliver different functional micro/nanoobjects, such as drug capsules, micro-droplets, micromachinery parts for drug delivery, micromanipulation, and microengineering.

Experimental section

Materials

Silica, carboxyl-polystyrene, and amino-polystyrene microparticles, as large cargoes, were purchased from Sigma-Aldrich (Shanghai, China), and perfluorooctane was purchased from Aladdin Chemistry Co., Ltd. (Shanghai, China). Tetrabutyl titanate (TBT) was purchased from Sinopharm Chemical Reagent (Shanghai, China).

Synthesis of TiO_2 micromotors

The constituent isotropic TiO_2 micromotors used in this work were prepared following previous reports.^{26,43} Typically, 25 mL ethanol was mixed with 100 μL sodium chloride aqueous solution (0.1 M) and 425 μL TBT at ambient temperature. After magnetic stirring for 18 min, the solution was allowed to stand for 24 h. The resulting precipitate was separated, washed three

times with alcohol and deionized water, and then dried at 60 $^\circ\text{C}$ for 12 h to obtain amorphous TiO_2 microspheres. Finally, anatase TiO_2 micromotors were obtained *via* calcinating amorphous TiO_2 at 300 $^\circ\text{C}$ for 2 h.

Cooperative transport

A 50 μL suspension of the TiO_2 micromotors (1 mg mL^{-1}) was dropped onto a glass slide (Citotest 1A5107), followed by adding 50 μL H_2O_2 fuel solution (0.75 wt%) and 50 μL cargo microparticle suspension (0.03 mg mL^{-1}). The dispersed TiO_2 micromotors usually spontaneously gather with the cargo microparticles and form into clusters within five minutes. Four UV-LED light sources (SZ Lamplic Technology, China) with a wavelength of 365 nm and adjustable intensity (0–500 mW cm^{-2}) were set above the glass substrate with an incident angle of 45 $^\circ$, as demonstrated in the experimental setup in Fig. S2.† The on/off and incident direction of the UV light were adjusted to investigate the phototaxis of TiO_2 -MFs and their capability for cooperatively transporting large cargoes. The experimental results were observed and recorded at room temperature using an inverted optical microscope (Leica DMI 3000 M, Germany). Videos were analyzed by using ImageJ and Video Spot Tracker V08.01 software. The velocity of the flocks was determined by calculating the displacement of centroid of flocks per second under light irradiation. Over four TiO_2 -MFs were analyzed to obtain the statistic result. The error bar given in the plot is the standard error of the mean.

Numerical simulation

Numerical simulations were performed using the diffusion module of COMSOL Multiphysics software. The simulation model was built up by immersing a 10 μm SiO_2 microsphere surrounded by 65 TiO_2 micromotors (1.2 μm in diameter (d)) in

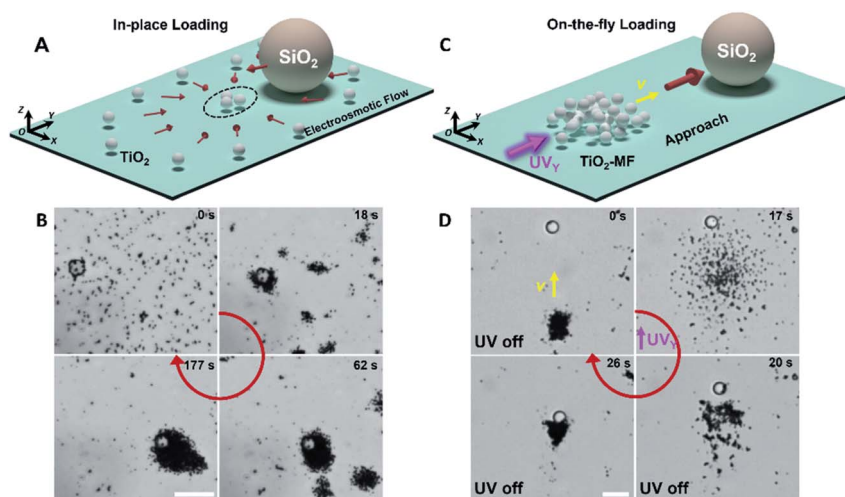


Fig. 1 Two modes of large-cargo loading by TiO_2 -MFs. (A and B) The schematic diagram (A) and optical microscope snapshots (B) depicting the in-place large-cargo loading utilizing spontaneous swarming of TiO_2 micromotors (d : 1.2 μm) together with a large SiO_2 microsphere (d : 10 μm). (C and D) The schematic diagram (C) and optical microscope snapshots (D) depicting the on-the-fly large-cargo loading by steering a TiO_2 -MF to approach a large SiO_2 microsphere (10 μm). The yellow arrow indicates the motion direction of the flock. H_2O_2 concentration: 0.25 wt%. Scale bars in (B) and (D): 20 μm .



the bottom of a 0.01 mm^2 square space which was filled with H_2O_2 aqueous solution (0.25 wt%). The flux (J_{O_2}) of oxygen molecules from the UV illuminated surface of the TiO_2 micromotors was set as $4.13 \times 10^{-4}\text{ mol m}^{-2}\text{ s}^{-1}$ according to the previous report.⁴³ The diffusion coefficient of O_2 molecules in water was set to be $1.97 \times 10^{-9}\text{ m}^2\text{ s}^{-1}$. The distribution of O_2 concentration was simulated at the steady state, and the average interparticle distance (L) was derived from the experimental results of the collective expansion of TiO_2 -MFs under UV irradiation.

Results and discussion

The constituent isotropic TiO_2 micromotors ($1.2\text{ }\mu\text{m}$ in d) of TiO_2 -MFs were fabricated referring to the previous work.²⁶ According to the reported mechanism, the TiO_2 micromotors with rich hydroxyl groups secrete H^+ and OH^- ions in aqueous media, and thus can spontaneously gather into clusters due to the diffusiophoretic motor-motor attraction without external energy input.⁴³ It is noted that this diffusiophoretic attraction can also compel adjacent passive particles to assemble with the TiO_2 micromotors. Fig. 1A shows the schematic diagram depicting the spontaneous in-place loading of a large SiO_2 microsphere ($10\text{ }\mu\text{m}$) during the gathering of the dispersed TiO_2 micromotors. As shown in the corresponding experimental microscope snapshots in Fig. 1B, the TiO_2 micromotors aggregate gradually to form small clusters, and then these small clusters further merge into large clusters along with time progression (ESI Video 1†). Remarkably, multiple large SiO_2 microspheres could be loaded by the swarming TiO_2 micromotors (Fig. S1†).

Like other photochemical TiO_2 -based micromotors, the gathered TiO_2 micromotors can be powered by photocatalytically decomposing surrounding H_2O_2 fuels.^{44–50} Under sidewise UV irradiation, the gathered TiO_2 micromotors exhibit negative phototaxis as a flock (TiO_2 -MF) in the medium with the H_2O_2 fuel,^{43,44} and thus can also be driven to approach and load a large cargo on the fly. The on-the-fly cargo loading is illustrated in Fig. 1C and D (taken from ESI Video 2†) shows the corresponding experimental results. Under the navigation of sidewise UV light (UV_Y in Fig. 1D), the TiO_2 -MF moved toward a SiO_2 microsphere utilizing its dilatational negative phototaxis based on the dominant nonelectrolyte diffusiophoresis (0–17 s in Fig. 1D), and finally loaded it into the flock *via* the electrolyte diffusiophoretic attraction after removing the UV_Y (20–26 s in Fig. 1D).

After being loaded by a TiO_2 -MF, the large cargo can then be transported because it also feels the electrolyte/nonelectrolyte gradients generated by active TiO_2 micromotors, suggesting that it will move and gather with the flocking TiO_2 micromotors (in resemblance to a failed constituent in the flock) when the UV light is on and off, respectively. In other words, the active transport of passive cargoes can be achieved by operating TiO_2 -MFs *via* UV light. Fig. 2A and B depict the schematic diagram and experimental snapshots of the cooperative transport of a large SiO_2 microsphere ($d: 10\text{ }\mu\text{m}$) by a TiO_2 -MF along the pre-designed path (red dashed lines in Fig. 2B) to a predetermined destination (ESI Video 3†). In detail, when turning on the UV_X

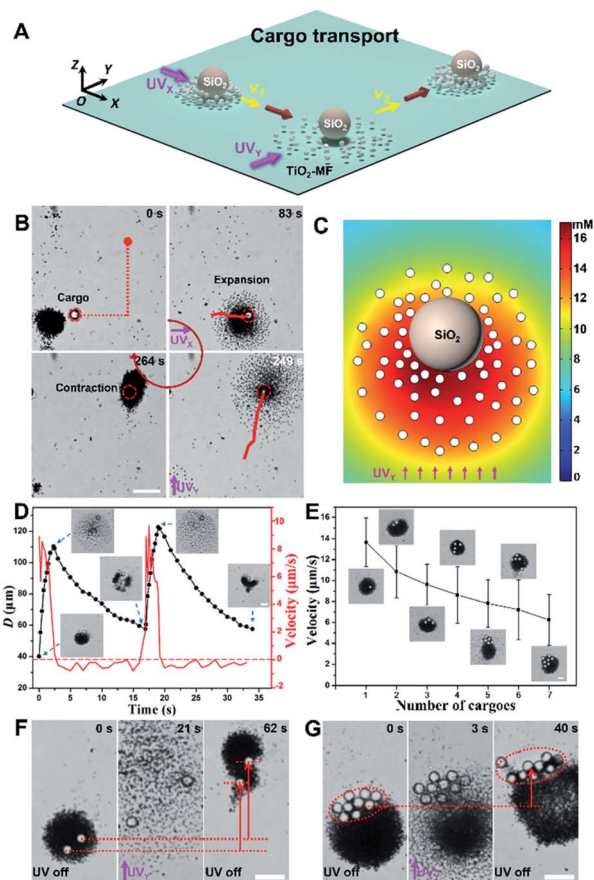


Fig. 2 Cooperative large-cargo transport by flocking phototactic TiO_2 micromotors. (A) Schematic diagram of light-controlled directional cooperative transport of large cargoes by operating the TiO_2 -MF. (B) A SiO_2 microparticle ($d: 10\text{ }\mu\text{m}$) was delivered by a TiO_2 -MF along a pre-designed path under the navigation of pulsed UV irradiation. Images are taken from ESI Video 3.† Scale bar: $10\text{ }\mu\text{m}$. (C) Simulated O_2 concentration around a passive SiO_2 microsphere surrounded by 65 TiO_2 micromotors in H_2O_2 aqueous solution (0.25 wt%) under UV irradiation (500 mW cm^{-2}). The interparticle distance was derived from the experimental results on transport. (D) The size (D) variation of a typical flock and the velocity variation of the loaded SiO_2 microsphere versus time under the pulsed UV irradiation. The insets depict the states of the flock and the cargo at various moments. Scale bar: $10\text{ }\mu\text{m}$. (E) Average motion velocity of typical TiO_2 -MFs (with similar D) carrying different numbers of cargoes (SiO_2 microspheres with a d of $10\text{ }\mu\text{m}$) under UV light illumination. The insets illustrate the microscopic images of typical flocks loading with different numbers of large cargoes. The error bar of each data point given in the plot is the standard error of the mean of over four different TiO_2 -MFs. Scale bar: $10\text{ }\mu\text{m}$. (F and G) The optical microscope snapshots of cooperative transport of two (F) and nine (G) large SiO_2 microspheres by TiO_2 -MFs with UV manipulation. Images are taken from ESI Video 4.† Scale bar: $20\text{ }\mu\text{m}$. UV light intensity: 500 mW cm^{-2} . H_2O_2 concentration: 0.25 wt%.

(UV light is toward the $+X$ axis direction), the large SiO_2 microsphere was pushed expansively *via* nonelectrolyte diffusiophoretic repulsion of the TiO_2 -MF. Once removing the light, the cargo was dragged back to form a resting flock with TiO_2 micromotors again at a new central point and showed a net phototactic displacement. When the UV_Y (UV light is toward the



+Y axis direction) was then turned on, the flock exhibited an immediate turning toward the +Y direction. The O_2 distribution around a large SiO_2 microsphere surrounded by 65 TiO_2 micromotors in the transport process was simulated by using the diffusion module of COMSOL Multiphysics software. As illustrated in Fig. 2C, upon UV_Y irradiation, a distinct O_2 concentration gradient forms across the large cargo which can lead to the non-electrolyte diffusiophoresis to propel the cargo. Fig. 2D shows the diameter (D) variation of a typical TiO_2 -MF and the velocity variation of the loaded cargo (the large SiO_2 microsphere) *versus* time under two pulsed UV_Y irradiation, where the instantaneous velocity vector subtly reflects the corresponding pushing and dragging processes discussed above. The cargo moved forward (positive velocity of red curve in Fig. 2D) with the expanding TiO_2 -MF (insets and black curve in Fig. 2D) under UV illumination, and then slightly moved backward (negative velocity of the red curve in Fig. 2D) when the TiO_2 -MF contracted back to a tight resting flock in the absence of UV light (insets and black curve in Fig. 2D). The agreement of these two curves in time proves the synchronized motions of the TiO_2 -MF and the cargo.

More interestingly, the TiO_2 -MF can also transport multiple large cargoes due to its strong driving force (Fig. 2E). Insets in Fig. 2E are the microscope snapshots showing TiO_2 -MFs with similar D (*ca.* 50 μm) loaded with different numbers of large cargoes (SiO_2 microspheres). Under the same light and fuel conditions, these TiO_2 -MFs moved slower when carrying more SiO_2 microspheres. Specifically, the average motion velocity of TiO_2 -MFs decreased from 13.7 to 6.2 $\mu m s^{-1}$ when the number of the carried SiO_2 microspheres increased from one to seven. The decreasing velocity can be rationalized by the greater

burden if more passive cargoes were carried while the powering parts (TiO_2 micromotors) were unchanged. The slight difference in the size of TiO_2 -MFs was believed to have a negligible influence on the velocity of these flocks as they had a similar velocity (Fig. S3†). The light-controlled cooperative transport of multiple large cargoes by TiO_2 -MFs is further visually illustrated in Fig. 2F and G (taken from ESI Video 4†), and the displacements in the +Y direction were clearly observed whether the transported cargo number was two or nine (red arrows in Fig. 2E and F).

Besides, TiO_2 -MFs can cooperatively transport different types of heavy cargoes, such as amino- and carboxyl-polystyrene microspheres, and perfluorooctane droplets (Fig. 3). Similar to the spontaneous loading process of the large SiO_2 microsphere (Fig. 1A and B), three different types of large cargoes, including an amino-polystyrene microsphere (d : 10 μm), a carboxyl-polystyrene microsphere (d : 10 μm) and perfluorooctane droplets (average d : 5 μm) can be loaded into the TiO_2 -MF because of the electrolyte diffusiophoretic attraction, and then exhibited synchronized negative phototaxis with TiO_2 -MFs under pulsed UV_Y light (ESI Video 5†). Fig. 3A, C and E depict the schematic diagrams of transporting corresponding cargoes. The red dashed circles in Fig. 3B, D and F indicate the real-time positions of the cargo particles. These results prove the good adaptability of TiO_2 -MFs, which guarantees the ability to be used in different application scenarios.

Once arriving at a predesigned destination, large cargoes can be unloaded from TiO_2 -MFs by modulating UV light. Two modes of cargo unloading can be realized depending on the different relative positions of the cargoes and TiO_2 -MFs, including disassembling and reversing the flock, as shown in

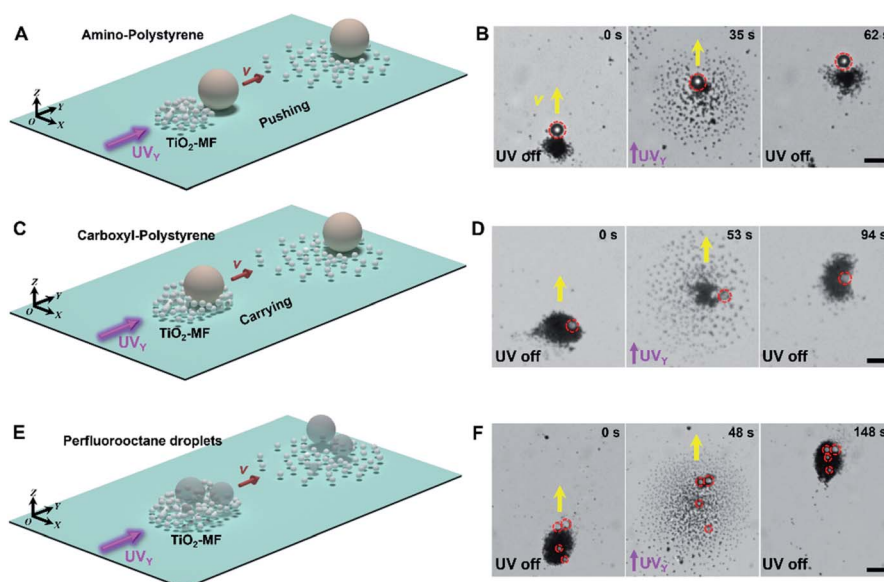


Fig. 3 Cooperative transport of different types of large cargoes by flocking phototactic TiO_2 micromotors. The schematic diagrams and optical microscope snapshots depicting the cooperative transport of an amino-polystyrene microsphere ((A) and (B), d : 10 μm), a carboxyl-polystyrene microsphere ((C) and (D), d : 10 μm) and perfluorooctane droplets ((E) and (F), the average d is 5 μm) by TiO_2 -MFs. The red dashed circles show the positions of the corresponding cargo particles. Yellow arrows indicate the motion direction of the flocks. Images are taken from ESI Video 5.† Scale bars: 10 μm . UV light intensity: 500 $mW cm^{-2}$. H_2O_2 concentration: 0.25 wt%.



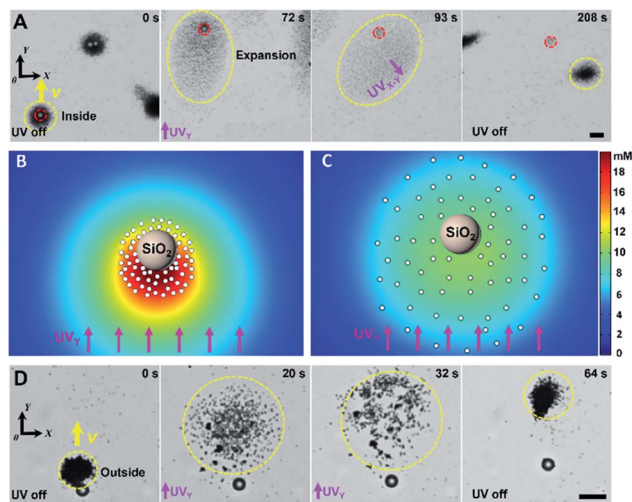


Fig. 4 Light-controlled large-cargo unloading from TiO₂-MFs. (A) Time-lapse optical microscopy images illustrating the unloading of a SiO₂ microsphere that was inside of a TiO₂-MF by disassembling the flock, taken from ESI Video 6.† (B and C) Simulated O₂ concentration around a passive SiO₂ microsphere surrounded by 65 TiO₂ micromotors in H₂O₂ aqueous solution (0.25 wt%) under UV irradiation (500 mW cm⁻²) with different *L* of 2.5 (B) and 6 μm (C). The different *L* is because of the collective expansion based on nonelectrolyte diffusiophoresis upon UV illumination. (D) Time-lapse optical microscopy images illustrating the unloading of an amino-polystyrene microsphere that was outside of a TiO₂-MF by reversing the flock away from the microsphere, taken from ESI Video 7.† Yellow and red dashed circles represent objects of TiO₂-MFs and cargoes, respectively. Yellow arrows indicate the motion direction of the flocks. Scale bars: 10 μm. UV light intensity: 500 mW cm⁻², H₂O₂ concentration: 0.25 wt%.

the experimental results in Fig. 4A and B. For a large cargo that was carried inside of a TiO₂-MF, it can be unloaded by disassembling the flock. Specifically, when a SiO₂ microsphere (as the large cargo, 10 μm) was carried to the destination under the navigation of the pulsed UV_Y irradiation (0–72 s in Fig. 4A), an extra UV light (UV_X) was turned on. Under the superimposed continuous UV_X and UV_Y irradiation (or the continuous UV_{X-Y}

irradiation in a direction with an angle of 45° between the +*X* and –*Y* axis), the TiO₂-MF was disassembled into scattered micromotors. With the prolonging irradiation time, the scattered TiO₂ micromotors moved away from the large cargo (72–93 s in Fig. 4A). Thus, a distinct displacement difference was induced between the TiO₂-MF and cargo (93 s in Fig. 4A), and the large cargo was then unloaded because the electrolyte diffusiophoretic attraction became too weak due to the large flock-cargo distance after the UV_{X-Y} irradiation was off (208 s in Fig. 4A). The yellow and red dashed circles in Fig. 4A mark the real-time positions of the TiO₂-MF and the cargo, respectively, which also record the corresponding unloading process. To decipher the cargo unloading by disassembling the TiO₂-MF, we simulated the distribution of the photogenerated O₂ molecules across the large cargo surrounded by a cluster of TiO₂ micromotors (1.2 μm) with different average motor-motor distances (*L*). With increasing *L* from 2.5 (Fig. 4B) to 6 μm (Fig. 4C), corresponding to the different scattered states of the TiO₂ micromotors under normal UV_Y irradiation (72 s in Fig. 4A) and under continuous UV_{X-Y} irradiation (93 s in Fig. 4A), the O₂ concentration gradient across the SiO₂ microsphere along the +*Y* direction decreased sharply, weakening its nonelectrolyte diffusiophoresis (or the subjected diffusiophoretic pushing force from the micromotors). As a result, the SiO₂ cargo is left behind by the scattered TiO₂ micromotors and finally unloaded from the TiO₂-MF. On the other hand, in the scenario that the large cargo is outside of the TiO₂-MF, it would be easier for the cargo to be unloaded. As shown in Fig. 4D, with the UV irradiation pointing in the +*Y* direction, the TiO₂-MF would move backward away from the carried cargo (an amino-polystyrene microsphere, 10 μm), and then the cargo was unloaded from the flock successfully.

The transport capability of TiO₂-MFs in a confined space was also tested here, as demonstrated in Fig. 5 and ESI Video 8.† When the TiO₂ micromotors and three large cargoes (SiO₂ microspheres, 10 μm) were transferred into a glass microchannel in a rotated S-shape, they formed two flocks at different corners, and the three cargoes (circled by red, green, and yellow circles, respectively) were located at different positions. In the

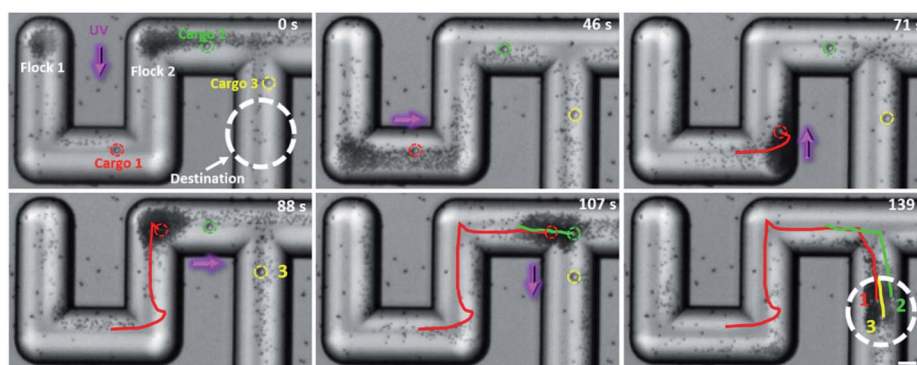


Fig. 5 Time-lapse optical microscopy images illustrating the cooperative transport of three large cargoes by TiO₂-MFs under the navigation of UV light in a microchannel. The cargoes (SiO₂ microspheres, 10 μm) are marked in different colors. The white dashed circles represent the target area. The optical microscope snapshots are taken from ESI Video 8.† Scale bar: 20 μm. UV light intensity: 500 mW cm⁻², H₂O₂ concentration: 0.25 wt%.

presence of UV_Y (−Y direction), the two TiO₂-MFs initiated negative phototaxis toward the same wall and merged into a large flock (0–46 s). In the meantime, cargo 1 was loaded by the TiO₂-MF. After turning the UV light to +X direction, the merged TiO₂-MF coercing cargo 1 moved toward the right corner. By a similar approach, the TiO₂-MF was guided to pass through the “S” microchannel and deliver the cargoes to the pre-designed destination within 139 s. The red, green and yellow lines in Fig. 5 depict the trajectories of cargoes 1, 2, and 3, respectively. The above results reveal the feasibility of cooperative transport in complex landscapes by the flocking TiO₂ micromotors.

Conclusions

This work has demonstrated that phototactic TiO₂-MFs can cooperatively manipulate multiple and different types of large cargoes under UV navigation. The swarming and collective phototaxis of TiO₂ micromotors are produced from the spontaneous electrolyte diffusiophoretic attraction and photo-induced nonelectrolyte diffusiophoretic repulsion, which can also be tuned to load, transport, and release large cargoes when controlling the on/off, illumination time and incident direction of UV light. Thus, under UV navigation, multiple and different types of large cargoes can be cooperatively transported along a predefined path and be deployed at a predetermined destination in an open space or complex microenvironments. This strategy of cooperative transport may advance the applications of swarming micro/nanomotors in drug delivery, micromanipulation and microengineering.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

Fangzhi Mou and Jianguo Guan are grateful for the financial support from the National Natural Science Foundation of China (21875175, 52073222 and 51521001) and the Natural Science Foundation of Hubei Province (2019CFA048). Jiaqi Song, Joshua E. Kauffman and Ayusman Sen acknowledge financial support by the Department of Energy, Office of Basic Energy Sciences (DOE-DE-SC0020964). Jianhua Zhang acknowledges the Scholarship Fund from the China Scholarship Council (CSC).

References

- 1 S. Sanchez, A. A. Solovev, S. Schulze and O. G. Schmidt, *Chem. Commun.*, 2011, **47**, 698.
- 2 S. Campuzano, J. Orozco, D. Kagan, M. Guix, W. Gao, S. Sattayasamitsathit, J. C. Claussen, A. Merkoci and J. Wang, *Nano Lett.*, 2012, **12**, 396.
- 3 A. Llopis-Lorente, A. Garcia-Fernandez, N. Murillo-Cremaes, A. C. Hortelao, T. Patino, R. Villalonga, F. Sancenon, R. Martinez-Manez and S. Sanchez, *ACS Nano*, 2019, **13**, 12171.
- 4 Y. Yang and M. A. Bevan, *Sci. Adv.*, 2020, **6**, eaay7679.
- 5 D. Ha, S. Seo, K. Lee and T. Kim, *ACS Nano*, 2019, **13**, 12939.
- 6 S. Tang, F. Zhang, H. Gong, F. Wei, J. Zhuang, E. Karshalev, B. E.-F. d. Ávila, C. Huang, Z. Zhou, Z. Li, L. Yin, H. Dong, R. H. Fang, X. Zhang, L. Zhang and J. Wang, *Sci. Robot.*, 2020, **5**, eaba6317.
- 7 Y. Tu, F. Peng, A. A. Andre, Y. Men, M. Srinivas and D. A. Wilson, *ACS Nano*, 2017, **11**, 1957.
- 8 H. Xu, M. Medina-Sanchez, V. Magdanz, L. Schwarz, F. Hebenstreit and O. G. Schmidt, *ACS Nano*, 2018, **12**, 327.
- 9 M. Luo, Y. Feng, T. Wang and J. Guan, *Adv. Funct. Mater.*, 2018, **28**, 1706100.
- 10 I. C. Yasa, A. F. Tabak, O. Yasa, H. Ceylan and M. Sitti, *Adv. Funct. Mater.*, 2019, **25**, 1808992.
- 11 Z. Wu, J. Troll, H.-H. Jeong, Q. Wei, M. Stang, F. Ziemssen, Z. Wang, M. Dong, S. Schnichels, T. Qiu and P. Fischer, *Sci. Adv.*, 2018, **4**, eaat4388.
- 12 S. K. Srivastava, M. Medina-Sanchez, B. Koch and O. G. Schmidt, *Adv. Mater.*, 2016, **28**, 832.
- 13 J. Li, B. E.-F. d. Ávila, W. Gao, L. Zhang and J. Wang, *Sci. Robot.*, 2017, **2**, eaam6431.
- 14 K. Kim, J. Guo, Z. Liang and D. Fan, *Adv. Funct. Mater.*, 2018, **28**, 1705867.
- 15 Y. Xing, S. Tang, X. Du, T. Xu and X. Zhang, *Nano Res.*, 2020, **14**, 654.
- 16 A. F. Demirors, F. Eichenseher, M. J. Loessner and A. R. Studart, *Nat. Commun.*, 2017, **8**, 1872.
- 17 W. Gao and J. Wang, *Nanoscale*, 2014, **6**, 10486.
- 18 J. Palacci, S. Sacanna, A. Vatchinsky, P. M. Chaikin and D. J. Pine, *J. Am. Chem. Soc.*, 2013, **135**, 15978.
- 19 D. Kagan, R. Laocharoensuk, M. Zimmerman, C. Clawson, S. Balasubramanian, D. Kong, D. Bishop, S. Sattayasamitsathit, L. Zhang and J. Wang, *Small*, 2010, **6**, 2741.
- 20 S. Sundararajan, S. Sengupta, M. E. Ibele and A. Sen, *Small*, 2010, **6**, 1479.
- 21 C. Chen, X. Chang, P. Angsantikul, J. Li, B. E.-F. d. Ávila, E. Karshalev, W. Liu, F. Mou, S. He, R. Castillo, Y. Liang, J. Guan, L. Zhang and J. Wang, *Adv. Biosyst.*, 2018, **2**, 1700160.
- 22 W. Gao, D. Kagan, O. S. Pak, C. Clawson, S. Campuzano, E. Chuluun-Erdene, E. Shipton, E. E. Fullerton, L. F. Zhang, E. Lauga and J. Wang, *Small*, 2012, **8**, 460.
- 23 A. M. Boymelgreen, T. Balli, T. Miloh and G. Yossifon, *Nat. Commun.*, 2018, **9**, 760.
- 24 F. Qiu, R. Mhanna, L. Zhang, Y. Ding, S. Fujita and B. J. Nelson, *Sens. Actuators, B*, 2014, **196**, 676.
- 25 D. Kagan, M. J. Benchimol, J. C. Claussen, E. Chuluun-Erdene, S. Esener and J. Wang, *Angew. Chem., Int. Ed.*, 2012, **51**, 7519.
- 26 C. R. Chen, F. Mou, L. Xu, S. Wang, J. Guan, Z. Feng, Q. Wang, L. Kong, W. Li, J. Wang and Q. Zhang, *Adv. Mater.*, 2017, **29**, 1603374.
- 27 H. Xu, M. Medina-Sanchez and O. G. Schmidt, *Angew. Chem., Int. Ed.*, 2020, **59**, 15029.
- 28 A. Gelblum, I. Pinkoviezky, E. Fonio, A. Ghosh, N. Gov and O. Feinerman, *Nat. Commun.*, 2015, **6**, 7729.



- 29 H. Wang and M. Pumera, *Chem. Soc. Rev.*, 2020, **49**, 3211.
- 30 B. Yigit, Y. Alapan and M. Sitti, *Soft Matter*, 2020, **16**, 1996.
- 31 Q. Wang and L. Zhang, *ACS Nano*, 2021, **15**, 149.
- 32 Q. Wang, K. F. Chan, K. Schweizer, X. Du, D. Jin, S. C. H. Yu, B. J. Nelson and L. Zhang, *Sci. Adv.*, 2021, **7**, eabe5914.
- 33 M. Rubenstein, A. Cornejo and R. Nagpal, *Science*, 2014, **345**, 795.
- 34 J. Yu, B. Wang, X. Du, Q. Wang and L. Zhang, *Nat. Commun.*, 2018, **9**, 3260.
- 35 H. Xie, M. Sun, X. Fan, Z. Lin, W. Chen, L. Wang, L. Dong and Q. He, *Sci. Robot.*, 2019, **4**, eaav8006.
- 36 Y. Hong, M. Diaz, U. M. Córdova-Figueroa and A. Sen, *Adv. Funct. Mater.*, 2010, **20**, 1568.
- 37 W. Duan, R. Liu and A. Sen, *J. Am. Chem. Soc.*, 2013, **135**, 1280.
- 38 J. Palacci, S. Sacanna, A. P. Steinberg, D. J. Pine and P. M. Chaikin, *Science*, 2013, **339**, 936.
- 39 J. Wang, Z. Xiong and J. Tang, *Adv. Intell. Syst. Comput.*, 2021, **3**, 2000170.
- 40 C. Wu, J. Dai, X. Li, L. Gao, J. Wang, J. Liu, J. Zheng, X. Zhan, J. Chen, X. Cheng, M. Yang and J. Tang, *Nat. Nanotechnol.*, 2021, **16**, 288.
- 41 F. Schmidt, B. Liebchen, H. Lowen and G. Volpe, *J. Chem. Phys.*, 2019, **150**, 094905.
- 42 P. Illien, R. Golestanian and A. Sen, *Chem. Soc. Rev.*, 2017, **46**, 5508.
- 43 F. Mou, J. Zhang, Z. Wu, S. Du, Z. Zhang, L. Xu and J. Guan, *iScience*, 2019, **19**, 415.
- 44 J. Zhang, J. Song, F. Mou, J. Guan and A. Sen, *Trends Chem.*, 2021, **3**, 387.
- 45 B. Dai, J. Wang, Z. Xiong, X. Zhan, W. Dai, C. C. Li, S. P. Feng and J. Tang, *Nat. Nanotechnol.*, 2016, **11**, 1087.
- 46 R. Dong, Q. Zhang, W. Gao, A. Pei and B. Ren, *ACS Nano*, 2016, **10**, 839.
- 47 B. Jang, A. Hong, H. E. Kang, C. Alcantara, S. Charreyron, F. Mushtaq, E. Pellicer, R. Buchel, J. Sort, S. S. Lee, B. J. Nelson and S. Pane, *ACS Nano*, 2017, **11**, 6146.
- 48 J. Zhang, F. Mou, Z. Wu, S. Tang, H. Xie, M. You, X. Liang, L. Xu and J. Guan, *ACS Appl. Mater. Interfaces*, 2019, **11**, 16639.
- 49 D. P. Singh, W. E. Uspal, M. N. Popescu, L. G. Wilson and P. Fischer, *Adv. Funct. Mater.*, 2018, **28**, 1706660.
- 50 L. Niese, L. Wang, S. Das and J. Simmchen, *Soft Matter*, 2020, **16**, 10585.

