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2, 7064Optical and electronic properties of transparent
conducting Ta:TiO₂ thin and ultra-thin films:
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The development of low-dimensional transparent conducting systems is nowadays gaining interest in view of novel optoelectronic applications. In this paper, we investigate the evolution of optical and electronic properties of Ta-doped TiO₂ films when their thickness is decreased down to 5 nm and as a function of Ta doping (5–10 at%), and we correlate the observed behavior with the structural properties, showing a high degree of tunability. Ta:TiO₂ polycrystalline anatase films are synthesized via pulsed laser deposition, followed by vacuum annealing. For films of thickness 50–200 nm, the electrical resistivity is $\sim 8 \times 10^{-4}$ – $1 \times 10^{-3} \Omega \text{ cm}$ and the charge carrier density increases with the doping content while the mobility decreases. Below a thickness of 20 nm, the electrical properties partially deteriorate, but still conductive ultra-thin films can be obtained down to 5 nm. The optical response changes with Ta addition, i.e. the absorption band in the UV range blue-shifts, according to the Moss–Burstein effect, while absorption in the IR range increases because of free carriers. Finally, we provide estimates of the effective mass and the plasma energy in the IR range. The fine tunability of the optoelectrical properties of Ta:TiO₂ films makes them suitable as transparent conductive components for devices and for photonic or plasmonic applications in the visible and IR ranges.

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Introduction

Transparent conducting oxides (TCOs) have been attracting much attention because of their ability to combine good electrical properties and transparency in the visible region.¹

These peculiar properties enable TCOs to be employed as electrodes in optoelectronic devices, such as solar cells,² photodetectors,³ flexible displays,⁴ and light-emitting diodes.⁵ The most employed TCO for consumer products is Sn-doped indium oxide (ITO) because of its high optical transparency in the visible range (i.e. >80%) and very low resistivity ($1\text{--}2 \times 10^{-4} \Omega \text{ cm}$);⁶ however, the limited availability of In has been driving the research toward alternative materials, including Al-doped ZnO (AZO) and TiO₂-based TCOs.^{1,7–9} TiO₂ is widely investigated in the literature for energy conversion and photocatalytic applications because of its excellent activity, low cost, non-toxicity and chemical stability;¹⁰ it is commonly employed as the photoanode in dye sensitized solar cells thanks to its favourable electronic band alignment with the active component of the cell, and is a promising photo-absorbing material for photovoltaics and photocatalysis.^{11–13}

The conductivity of TiO₂ can be improved by n-type doping, which involves substitution of dopant atoms whose excess electrons are transferred to the conduction band of the matrix.¹⁴ Among the TiO₂ allotropes, anatase is the most attractive because of its band gap of 3.2 eV and the low conduction-band effective mass, around 1 m_e (although the

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actual value depends on the direction due to the anisotropy of the structure),^{15,16} while the energy gap and the effective mass of the rutile phase are about 3.0 eV and 20 m_e , respectively.¹⁰ Nb- and Ta-doped anatase have been studied as TiO₂-based TCOs, in particular Ta is reported to have both higher solubility and lower effective mass than Nb, which favors the mobility.¹⁶ Experimentally, both Nb- and Ta-doped anatase films epitaxially grown *via* pulsed laser deposition (PLD) showed very similar resistivity around $2\text{--}3 \times 10^{-4} \Omega \text{ cm}$,^{17,18} suggesting almost identical electrical conduction mechanisms, with a resistivity comparable to that of commercial polycrystalline ITO. However, the resistivity of Nb and Ta:TiO₂ polycrystalline films is typically higher than that of epitaxial films and is strongly dependent on the film structural properties, which implies a correlation with the synthesis conditions. Moreover, the electrical properties were demonstrated to be dependent on the oxygen stoichiometry and, in particular, the optimal values were obtained for a specific amount of oxygen defects.¹⁹ Indeed, the calculated formation energy of Nb_{Ti} and Ta_{Ti} is very low under Ti-rich/O-poor conditions,¹⁴ while oxygen-rich conditions promote electron acceptor defects, namely cationic vacancies and interstitial oxygen atoms, which reduce the conductivity.^{15,20,21}

Mazzolini *et al.* performed a study on the structural, electrical, and optical properties of Ta:TiO₂ with 5 at% Ta nominal content, while the oxygen content of the films was varied during the synthesis.¹⁵ In particular, Ta:TiO₂ films with thickness around 200 nm were obtained *via* PLD, which allows one to control the oxygen partial pressure during film deposition by means of gas mass flow, enabling in turn the tuning of oxygen stoichiometry in the deposited films. After deposition, thermal treatment in a vacuum is required in order to obtain films with the anatase

crystalline phase. The study revealed a dependency of the electrical and optical properties on the oxygen partial pressure during deposition and optimal conditions were identified.

Starting from this knowledge, in this work the structural, optical and electrical properties of thin and ultra-thin Ta-doped TiO₂ films are investigated as a function of thickness (from 200 down to 10 nm) and Ta content (5 and 10 at%) and compared with those of bare TiO₂. Such a systematic study is presently lacking, despite being essential for a better comprehension of the material behaviour in view of possible applications where size-dependent effects can become relevant. Indeed, the thickness can play a key role for TCOs in optoelectronic devices, where the highest possible transparency in the visible range should be combined with the desired electrical resistivity and charge carrier density. In addition, thin and ultra-thin TCO films have been proposed for their implementation in sub-wavelength nano-optics structures,²² replacing metals, which present a percolation threshold when deposited on silicon or glass substrates and have poor compatibility with microfabrication foundry processes, while extending the applicable wavelength range toward the mid- and near-infrared range.^{23–25}

In detail, homogeneous and compact polycrystalline Ta:TiO₂ thin films in the anatase phase were deposited on silicon and glass substrates with different Ta contents and thicknesses, down to ultra-thin layers. The electrical characterization demonstrates marginal changes for films in the 50–200 nm thickness range. The electrical resistivity is between 8×10^{-4} and $1 \times 10^{-3} \Omega \text{ cm}$ for both Ta contents, while the charge carrier density increases almost linearly with the Ta concentration and concurrently the Hall mobility decreases. For films thinner than 20 nm, the electrical properties start to

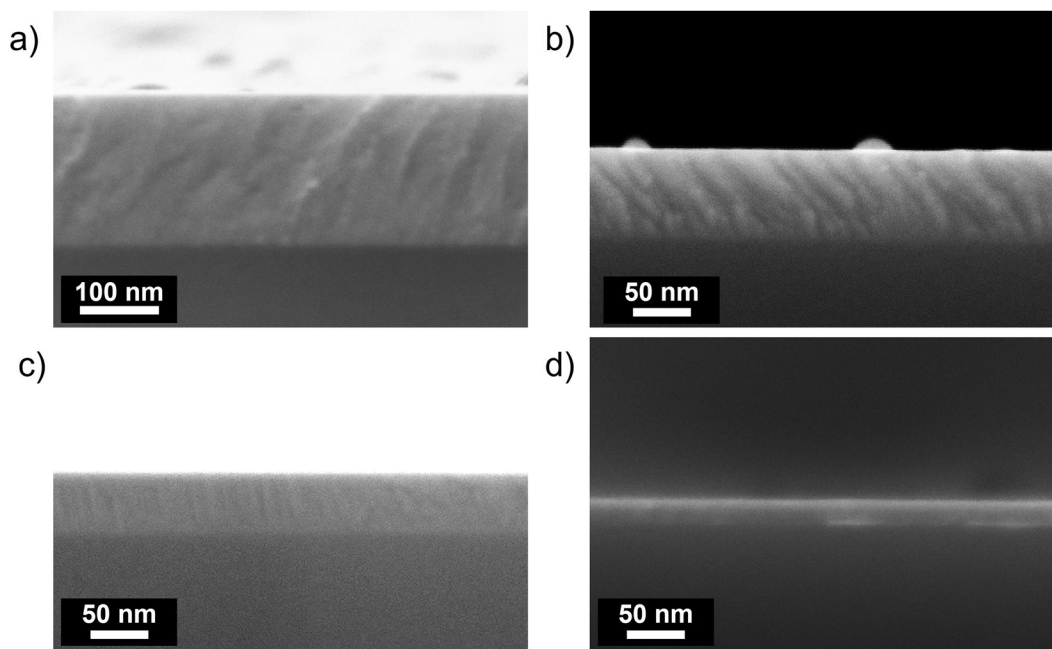


Fig. 1 SEM cross-sectional micrographs of vacuum-annealed Ta(5%):TiO₂ films deposited on a Si substrate with different thicknesses (a) 200, (b) 100, (c) 50 and (d) 20 nm.



deteriorate but still conductive 10 nm-thick films were obtained. The optical properties are extracted from ellipsometric measurements, combined with transmittance and reflectance spectra, and the effect of Ta-doping is explored in the near-IR to the UV ranges. Finally, the material properties of interest were calculated starting from experimental results in terms of optical band gap, effective mass, and plasma energy in view of potential optoelectronic applications.

Results

Morphology and structure

Ta:TiO₂ thin films with nominal contents of Ta = 5 and 10 at% (named Ta(5%):TiO₂ and Ta(10%):TiO₂, respectively) and bare TiO₂ films were deposited *via* PLD at a background pressure of 1 Pa O₂, followed by vacuum thermal treatment. Such a background oxygen pressure during deposition was carefully chosen to optimize the electrical properties of the resulting Ta:TiO₂ films (Fig. S1, ESI[†]), corresponding to the lowest resistivity and highest carrier mobility. Fig. 1 shows the SEM cross-sectional images of vacuum-annealed Ta(5%):TiO₂ thin films with different thicknesses, *i.e.* from 200 down to 20 nm. Both Ta(10%):TiO₂ and bare TiO₂ thin films present a compact morphology. The estimation of the amount of tantalum in Ta:TiO₂ *via* EDS is about 4.5 and 9 at% for films with nominal Ta contents of 5 and 10%, respectively, as expected for PLD depositions that preserve the cationic ratio of the ablated material.

Optical microscopy allowed for the surface investigation of films with thickness in the range of 50–200 nm, while thinner films did not provide any sizeable optical contrast. The average lateral size of the polycrystalline domains was measured to be 15–20 μm for Ta(5%):TiO₂ and 25–30 μm for Ta(10%):TiO₂ films. This difference can be ascribed to the different laser fluence employed during film depositions. The surface images of Ta(5%):TiO₂ and Ta(10%):TiO₂ 200 nm-thick films captured by using polarized-light microscopy are shown in Fig. S2 (ESI[†]).

The average roughness of 200 nm-thick films of Ta:TiO₂ and TiO₂ (both vacuum- and air-annealed) was estimated by means of spectroscopic ellipsometry by employing the model described in the Experimental section, resulting in the range of 3 to 5 nm for all samples. These values are in agreement with the root mean square roughness measured by atomic force microscopy on selected samples (not shown).

The structural characterization of the films was performed *via* XRD. Fig. 2 shows the resulting diffractograms as a function of Ta content for vacuum-annealed films with a nominal thickness of 200 nm, compared to vacuum- and air-annealed TiO₂ films. An initial investigation of the XRD patterns highlights that all the analysed films reveal only the presence of a crystalline anatase phase, while no rutile phase is observed. In addition, in both the tantalum-doped films neither the presence of tantalum oxide (Ta₂O₅) phases nor metallic tantalum aggregates are detected, suggesting the formation of homogeneous solid solutions.²⁶

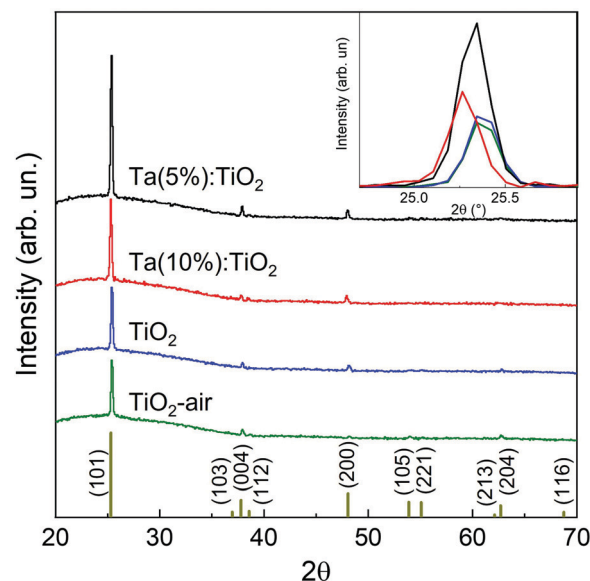


Fig. 2 XRD diffractograms of air- and vacuum-annealed TiO₂, Ta(5%):TiO₂ and Ta(10%):TiO₂ films with a nominal thickness of 200 nm. The inset shows the magnification of the (101) peak. The characteristic peaks of the anatase phase are highlighted at the bottom.

In all the analysed films the more intense diffraction peak is the one associated with the (101) surfaces. This behaviour can be explained since it is known that anatase thin films grown on amorphous substrates preferably form [101]-oriented polycrystalline systems, since these possess the lowest surface energy.^{27,28} Interestingly, in both the Ta-doped films, the relative intensity of the (101) peak is higher than those in the TiO₂ samples, suggesting that in the Ta-doped systems the doping could induce a preferential direction towards the lowest surface energy, namely the [101].

The atomic radius of Ta is slightly larger than the atomic radius of Ti, 145 and 140 pm, respectively, and, although the percentage of Ta is relatively low, this could introduce lattice distortions in the samples, which lead to a shift of the measured XRD peaks, as shown for the (101) peak in the inset of Fig. 2. A whole profile fitting performed on the films allows one to calculate the lattice parameters of the tetragonal cell of the TiO₂ anatase phase, demonstrating a variation of the length for the *a* and *c* axes in relation to the Ta content. Specifically, in the Ta-doped films both the *a* and *c* axes are longer than in the pristine TiO₂ films and their length increases with the Ta content (see Table 1).

The process of doping also affects the crystallite size along the direction normal to the film surface, which can be calculated by using the Pawley method on diffractograms as described in the Experimental section, and in fact both the Ta-doped films show a slightly larger mean domain size than the undoped TiO₂ films. In particular, Ta(5%):TiO₂ films display a larger domain size equal to 76 nm, while in Ta(10%):TiO₂ this is reduced to 68 nm. Interestingly, the different annealing procedures, namely in a vacuum or air, on undoped TiO₂ films do not produce a marked influence on the mean crystallite domain size that



Table 1 Lattice parameters (*a* and *c*), cell volume and mean crystalline domain size (τ) of 200 nm-thick films of air- and vacuum-annealed TiO₂, Ta(5%):TiO₂ and Ta(10%):TiO₂

Sample	<i>a</i> [Å]	<i>c</i> [Å]	Volume [Å ³]	τ [nm]
TiO ₂ -air	3.634	9.590	126.680	53
TiO ₂	3.783	9.483	135.735	56
Ta(5%):TiO ₂	3.793	9.498	136.646	76
Ta(10%):TiO ₂	3.793	9.542	137.299	68

remains almost unchanged (*i.e.* 53 nm and 56 nm for air- and vacuum-annealed TiO₂, respectively, see Table 1), while the lattice parameters and the cell volume show more notable differences, as already observed in the literature.¹⁵

Raman spectroscopy has been carried out for all Ta:TiO₂ films down to 10 nm thickness. Fig. 3a shows the Raman spectra of vacuum-annealed Ta(5%):TiO₂ thin films; all spectra display the Raman peaks of the anatase phase, while no presence of Ta₂O₅ or other TiO₂ polymorphs is detected, confirming the XRD results. In particular, the superimposing of E_g(1), B_{1g}(1), and B_{1g}(2) peaks on the A_{1g} and E_g(3) peaks of the anatase phase is evident around their nominal values, *i.e.* 144, 399, 519, 513, and 638 cm⁻¹ for the thickest films (*i.e.* 75–200 nm).^{29,30} On the other hand, for thicknesses in the 10–50 nm range, only the E_g(1) and E_g(3) peaks are evident, probably because of the poor signal-to-noise ratio.³¹ The Raman spectra of Ta(10%):TiO₂ films show the same trend as that described for the 5% Ta content (see Fig. S3, ESI†).

The Ta content also affects the position of the E_g(1) peak, as demonstrated in Fig. 3b; for 200 nm films, the E_g(1) peak shifts

from 144 cm⁻¹ to 152 and to 155 cm⁻¹ for Ta(5%):TiO₂ and Ta(10%):TiO₂, respectively, while the peak FWHM broadens from about 10 to 13–14 cm⁻¹. This trend has been already observed in the literature and is associated with the correlation between the E_g(1) peak position and the measured charge carrier density, independently of the type and amount of extrinsic doping, or the specific deposition and annealing conditions.³² Fig. 3c highlights the evolution of the E_g(1) peak position as a function of film thickness. At a fixed Ta content, the E_g(1) energy does not significantly depend on the thickness, at least down to 20 nm (in agreement with the measured charge carrier density, see the next section). On the other hand, for 10 nm-thick films the E_g(1) energy shifts for both Ta contents, namely to 147 and 150 cm⁻¹, respectively, while the peak FWHM increases up to 17–18 cm⁻¹. This behaviour can be associated with a larger film defectivity, size confinement as well as a smaller charge carrier density.²⁹

Electrical properties

Fig. 4 reports the resistivity (ρ), charge carrier density (N_e), and Hall mobility (μ) measured for Ta:TiO₂ specimens as a function of thickness (from 10 to 200 nm) and Ta content. Measurements on bare TiO₂ films (thicknesses of 50, 100 and 200 nm), synthesized and annealed in a vacuum under the same conditions as that of the Ta:TiO₂ films, are reported as reference (blue triangles). Notably, stoichiometric TiO₂ is expected to be insulating, as confirmed from the measurements on PLD TiO₂

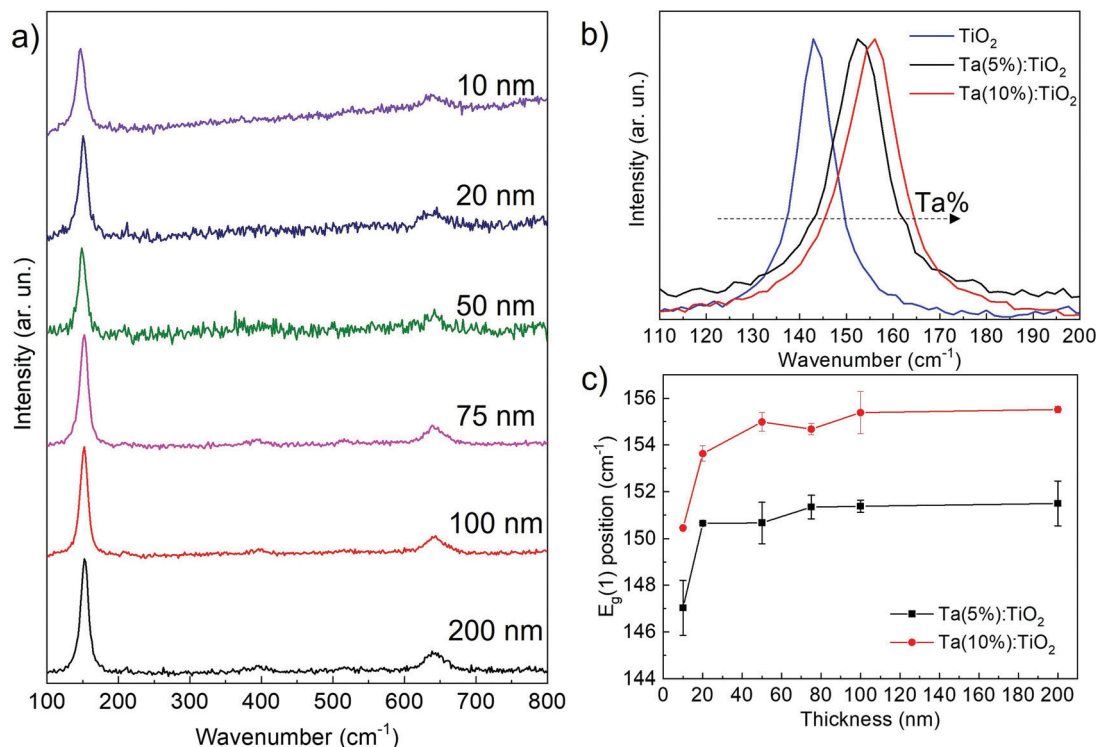


Fig. 3 (a) Raman spectra of Ta(5%):TiO₂ films with different thicknesses, *i.e.* from 10 up to 200 nm; (b) magnification of the E_g(1) peak of Raman spectra of films with different Ta content (5, 10 at% and bare TiO₂) with a thickness of 200 nm; (c) centre energy of the fitted E_g(1) peak as a function of the film thickness for different Ta contents.

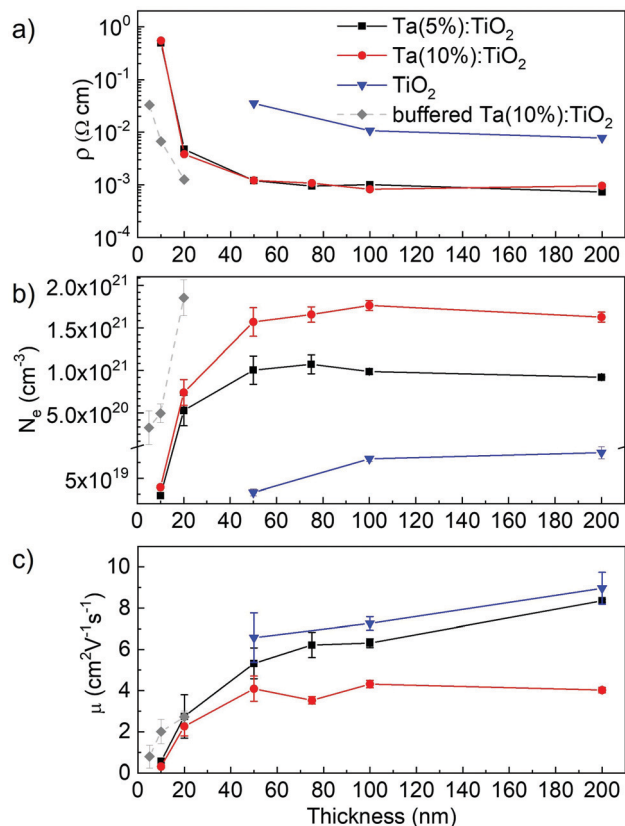


Fig. 4 (a) Resistivity (ρ), (b) charge carrier density (N_e) and (c) Hall mobility (μ) as a function of film thickness (from 10 to 200 nm) and Ta content: Ta(5%):TiO₂ (black squares), Ta(10%):TiO₂ (red circles), bare TiO₂ (blue triangles) and Ta(10%):TiO₂ ultra-thin films deposited on a buffer layer of insulating 50 nm-thick TiO₂ (grey rhombus).

films annealed in air (for which the resistivity was too high for the accessible range of our experimental setup). In this case, TiO₂ films obtained by PLD at 1 Pa O₂ after vacuum annealing treatment display a resistivity of about 10^{-2} Ω cm and a charge carrier density of $5\text{--}7 \times 10^{19}$ cm^{-3} , in agreement with the work of Mazzolini *et al.*¹⁵ This behaviour can be associated with a TiO₂ anatase phase with oxygen vacancies induced by the atmospheric conditions during deposition and annealing, which represent an effective doubly negative charged donor state.^{15,33}

For Ta:TiO₂ films, the resistivity is up to one order of magnitude lower than that of bare TiO₂ for both Ta contents, *i.e.* down to 10^{-3} Ω cm, because of conduction electrons from the Ta active dopants (see Fig. 4a). Focusing on Ta(5%):TiO₂ films, the resistivity is almost constant for thicknesses ranging from 200 down to 50 nm, with a slight increase from 8×10^{-4} up to 1×10^{-3} Ω cm, while a more evident increment occurs for 20 nm-thick films, where the resistivity reaches 3×10^{-3} Ω cm. The same trend is found for Ta(10%):TiO₂, where resistivity is around 1×10^{-3} Ω cm for films in the 50–200 nm thickness range, while it increases to about 2×10^{-3} Ω cm for 20 nm-thick films.

As for resistivity, the charge carrier density does not significantly vary with thickness down to 50 nm for both doping contents (Fig. 4b). The values of ρ and n of Ta(5%):TiO₂ films

are in agreement with the previous work of Mazzolini *et al.*, where polycrystalline Ta:TiO₂ 150 nm-thick films with the same Ta content were synthesized under the optimal oxygen sub-stoichiometric conditions.¹⁵ However, the resistivity of this work is slightly higher (the lowest one is around 8×10^{-4} Ω cm instead of 5×10^{-4} Ω cm) likely due to a slight decrease of the Hall mobility (see below). In addition, the same order of magnitude of resistivity was found for polycrystalline Nb:TiO₂ thin films with thickness above 100 nm.^{8,34–36}

On the other hand, the stability of the electrical properties down to a thickness of 50 nm is a novel and interesting result. Indeed, in a previous work, Liu *et al.* synthesized thin films of Nb and Ta co-doped TiO₂ (nominal content of 3 at% for both Nb and Ta) and the effect of the thickness was studied in the range of 50–500 nm.³¹ They found an increase of the resistivity by more than one order of magnitude when reducing the film thickness from 500 to 200 nm (*i.e.* from about 9×10^{-4} to 1×10^{-2} Ω cm), and even worse with a further reduction in thickness. In addition, even if the doping content was kept constant, the charge carrier density varied with the thickness, *i.e.* 2.3×10^{20} , 3.1×10^{20} and 1.1×10^{21} cm^{-3} for thicknesses 50, 200 and 500 nm, respectively, and this behavior was associated with crystalline defects.

In our films, the carrier density N_e varies with the Ta content in the 50–200 nm thickness range. Indeed, the N_e of the Ta(10%):TiO₂ films is almost twice that of Ta(5%):TiO₂, with average values in the 50–200 nm thickness range of about 1.65×10^{21} and 9.95×10^{20} cm^{-3} , respectively. For 20 nm-thick films, the charge carrier density is slightly smaller than in thicker films for both Ta contents, but Ta(10%):TiO₂ films still show a higher value than Ta(5%):TiO₂ (7.4×10^{20} cm^{-3} against 5.3×10^{20} cm^{-3} , respectively). These results indicate that the charge carrier density increases proportionally with the Ta content under the condition of high dopant activation efficiency (calculated values of 74% and 68% for Ta(5%):TiO₂ and Ta(10%):TiO₂, respectively, with respect to the measured EDS concentrations) as observed for thicker films as well.^{18,32} In addition, the measured charge carrier density is in good agreement with the observed energy shift of the E_g(1) Raman peak, which increases with the Ta content and decreases for films thinner than 20 nm.

For 200 nm-thick films, the Ta content significantly affects the Hall mobility (Fig. 4c). Indeed, while the mobility μ measured for Ta(5%):TiO₂ is almost the same as that for bare TiO₂ (*i.e.* around $8 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$), the Hall mobility of Ta(10%):TiO₂ is reduced by a factor of about 2. This behaviour is mostly related to the higher concentration of ionized impurities in the crystal lattice, which affects electron scattering as well as the higher probability of carrier-carrier and carrier-defect interaction at larger values of N_e .^{8,32} Moreover, the Hall mobility for Ta(5%):TiO₂ decreases gradually from about 8 down to about $5 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$ for thicknesses varying from 200 down to 50 nm, while it reaches a value of about $1.5 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$ for 20 nm-thick films. On the other hand, the mobility remains almost constant and around $4 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$ for Ta(10%):TiO₂ films down to 50 nm and decreases to about $2 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$ for 20 nm-thick



films. This trend can be related to the reduction of the characteristic grain size as well as to the presence of defects such as grain boundaries or dislocations affecting the mobility.³⁷ Indeed, the mean crystalline domain size calculated from XRD measurements (76 nm and 68 nm for Ta(5%):TiO₂ and Ta(10%):TiO₂, respectively) confirms that the higher Hall mobility of Ta(5%):TiO₂ is also accompanied by larger grains; moreover, a significant decrease of μ for decreasing thicknesses is expected for thicknesses much smaller than the calculated mean domain size, such as in 20 nm-thick films, where crystalline grains cannot develop completely.

On the other hand, 10 nm-thick films display a radically different electrical behaviour, almost independent of the Ta content. Indeed, for both Ta(5%):TiO₂ and Ta(10%):TiO₂ 10 nm-thick films, the resistivity measured is about 0.5 Ω cm, *i.e.* more than two orders of magnitude higher than in thicker Ta:TiO₂ films (see Fig. 4a). Hall effect measurements on these films were less reliable and reproducible, probably because of the higher resistance of the film as well as the difficulty to measure such a thin film with our setup. The measured charge carrier density was 2.3 and 3.5×10^{19} cm⁻³ for 5 and 10 at% Ta contents, respectively, namely around 2 orders of magnitude smaller than in thicker films (Fig. 4b). At the same time, the Hall mobility decreases by one order of magnitude with $\mu = 0.56$ cm² V⁻¹ s⁻¹ for Ta(5%):TiO₂ and $\mu = 0.31$ cm² V⁻¹ s⁻¹ for Ta(10%):TiO₂ films (Fig. 4c). This behavior can be explained with the presence of defects at the film surface and at the interface with the substrate, causing electron trapping and scattering that become dominant in ultra-thin films with a thickness below 20 nm, as already observed in previous studies on ultra-thin films of AZO,^{37–39} together with the smaller domain size as discussed before.

The electrical properties of ultra-thin Ta:TiO₂ films can be enhanced by exploiting an appropriate buffer layer that improves the wettability and the interface quality and reduces the high defectivity promoted by growing the films on lattice-mismatched substrates (*e.g.* glass and silicon).^{39–43} In this case, we investigated the effect of a 50 nm-buffer layer of polycrystalline stoichiometric TiO₂ (air-annealed) in the anatase phase on the growth of Ta(10%):TiO₂ films with 5, 10 and 20 nm thicknesses. The TiO₂ buffer layer alone is insulating and the measured electrical properties of the buffered films are reported in Fig. 5 (grey diamonds), showing an improvement for both 10 and 20 nm thick films. In particular, the buffered 20 nm-thick film displays a decrease in resistivity and an increase in charge carrier density up to values similar to the unbuffered Ta(10%):TiO₂ films with 50–200 nm thickness, while the buffered 10 nm-thick films show improvement in all the measured electrical properties to values similar to those of the unbuffered 20 nm-thick films. Finally, since the buffer layer favours a better wettability of the growing film onto the substrate, we were able to deposit conductive Ta(10%):TiO₂ films with a thickness as low as 5 nm, which is unaffordable by depositing directly on glass, even though the electrical properties show a deterioration. These results highlight the possibility of designing devices with ultra-thin conductive films, replacing metals which present a percolation threshold.²²

Optical properties

Ellipsometric measurements were performed on Ta:TiO₂ films with 200 nm nominal thickness, grown on Si substrates covered by a few-nm thick native SiO₂ layer as described in the Experimental section. The corresponding ψ and Δ measurements are shown in Fig. S5 of the ESI† Fig. 5a and b report the real (ϵ_1) and the imaginary (ϵ_2) part of the dielectric functions extracted by means of the optical model described in the Experimental section. The results of Ta:TiO₂ films with Ta contents 5 and 10 at% are compared to those of bare TiO₂ with the same nominal thickness, both air-annealed and vacuum-annealed. The corresponding refractive index and extinction coefficient are reported in Fig. S5 (ESI†).

For the real part (ϵ_1 , Fig. 5a), air-annealed TiO₂ shows two humps in the high-energy region (3.5–4.5 eV), while vacuum-annealed TiO₂ shows a unique band in the same range. In the visible range, ϵ_1 decreases with decreasing energy for both air- and vacuum-annealed TiO₂ with almost the same slope. For both samples, ϵ_1 values are in accordance with those reported in the literature for the anatase phase over the investigated spectral range, with a maximum around 12.5 at 3.8–3.9 eV and a value between 6 and 9 in the visible range (1.7–3.3 eV).^{44–47} In the near-IR range, ϵ_1 is flat for air-annealed TiO₂, while its value increases for decreasing energy for vacuum-annealed TiO₂. On the other hand, Ta doping leads to blue-shifted and narrowed features in the UV range, while the maximum values of ϵ_1 are close to those of bare TiO₂. In the visible range, the ϵ_1 curves of Ta:TiO₂ films almost overlap and decrease as in bare TiO₂ films but at lower values (this behaviour has been already reported by Manole *et al.* for the Nb-doped anatase phase),⁴⁸ while in the near-IR range ϵ_1 decreases, as expected by the free-carrier contribution that sets the plasma frequency in the mid-IR spectral range.

In Fig. 5b, ϵ_2 of air-annealed TiO₂ exhibits a band in the UV range associated with optical band gap absorption with a peak around 4.5 eV and a shoulder around 3.8 eV. This shape is associated with the anisotropy of the crystal structure in the anatase phase, which results in the optical properties being strongly dependent on the polarization direction of the incident light beam. Indeed, the shoulder and the peak are associated with the perpendicular and parallel components (with respect to the *c*-axis) of the dielectric tensor.^{44,45,49} Similar to ϵ_1 , also ϵ_2 shows a single feature in the UV range for vacuum-annealed TiO₂ at the same energy as that of air-annealed TiO₂, but with a slightly smaller value. The effect of Ta doping is the shift of the absorption band in the UV range towards higher energy values, suggesting a larger optical band gap with respect to bare TiO₂ films, due to the Moss–Burstein effect (see next section).^{50,51} In the visible range, all ϵ_2 curves are close to zero, suggesting no absorption as expected for TiO₂ and TiO₂-based TCOs. In the lower-energy region of the visible range (below 2 eV) and in the near-IR range, ϵ_2 of air-annealed TiO₂ curve is close to zero, while for vacuum-annealed TiO₂ it increases to a maximum around 0.84 eV. The different behaviour of vacuum-annealed TiO₂ with respect to air-annealed TiO₂ in this range can be associated with the oxygen vacancy concentration in the



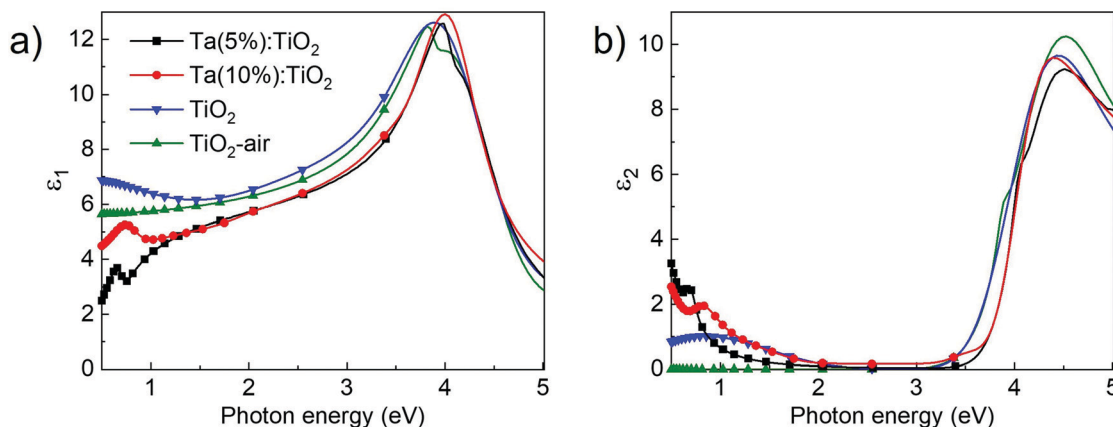


Fig. 5 (a) Real and (b) imaginary parts of the dielectric constant of Ta:TiO₂ films with different doping levels (5 and 10 at%) compared with those of bare vacuum-annealed (TiO₂) and air-annealed (TiO₂-air).

anatase structure (as already discussed in the previous section).^{52,53} On the other hand, Ta:TiO₂ films show a larger increase of ϵ_2 in this range, which is again consistent with a prominent plasma response in the mid-IR region.

The behavior of the real and imaginary parts of the dielectric constant of both Ta:TiO₂ films is in accordance with previous studies where the refractive index and the extinction coefficient in the UV and visible ranges of Ta:TiO₂, Nb:TiO₂ and Ta and Nb co-doped thin films are reported.^{19,48,54,55} The optical properties in the near-IR range show an absorptive behavior associated with the tail of free carrier (plasma) oscillations that play a role below the screened plasma frequency range. This trend was observed in the work of Neubert *et al.*, who performed ellipsometry on polycrystalline anatase Ta:TiO₂ films synthesized by means of DC magnetron sputtering (Ta = 2 at%, thickness 300–400 nm),¹⁹ and in the theoretical work of Muhammady *et al.*, regarding the calculation of the dielectric function of Ta:TiO₂ (Ta = 6.25 at%) in anatase phase by employing the Drude-Lorentz model.⁵⁶ In both cases this behavior was associated with the free carrier contribution, as already observed for other TCOs in thin film forms (*e.g.* ITO, AZO and Ga-doped ZnO), for which the occurrence of plasmonic excitations has been verified.^{57,58} Finally, the onset of the metallic behavior is expected to blue-shift with charge carrier density,⁵⁹ as observed on comparing Ta(5%) and Ta(10%):TiO₂.

Notably, both Ta:TiO₂ films show a peak in the near-IR range in ϵ_2 curves at 0.68 and 0.82 eV for Ta contents of 5 and 10 at%, respectively, that correspond to dispersive features in the ϵ_1 curve in the same range. The explanation of the weak features in the near-IR range is not straightforward. A tentative hypothesis is the establishment of a strong interaction between plasma oscillation of free carriers and phonons, with a formation of longitudinal optical phonon-plasmon coupled modes. This phenomenon is typical for polar semiconductors and it occurs in the IR dielectric response,^{60,61} which can be described by a harmonic oscillator function with a Lorentzian line shape.⁶² Although Ta:TiO₂ is not a polar material, the large carrier density due to Ta doping as well as the large degree of

lattice defects could induce an increase in the ionic character of the Ti-O covalent bond.

The optical constants extracted from ellipsometry were directly applied to the modelling of normal-incidence transmission spectra, by allowing only for small adjustments in the oscillator parameters to account for the different soda-lime substrate compared to the silicon one. In Fig. 6, experimental data from the transmission measurements are shown (squares) along with the fit data (continuous lines), which are the outcomes of the modelling. The respective dielectric constants, refractive index and extinction coefficients that came out of the modelling of this system are shown in Fig. S6 (ESI[†]). Indeed, transmittance spectra confirm the optical behaviour observed with the previous analysis of the dielectric function. For all the spectra, the transmittance goes to zero in the UV range due to interband absorption, while it increases in the visible and near-IR ranges. The oscillations in this range are due to thin-film interference and are related to the film optical path length.^{8,63} Vacuum-annealed TiO₂ displays lower transmittance intensity than air-annealed TiO₂ in the energy range from visible to near-IR associated with absorption by oxygen vacancies, while both Ta:TiO₂ films show a stronger decrease in the transmittance in the near-IR range due to free carrier absorption related to the response of the conduction plasma. Furthermore, the effect of increasing the Ta content is observed as a blue-shift of the absorption edge in the UV range, because of the Moss-Burstein effect, as previously explained.

Fig. 7a and c show the optical transmittance spectra as a function of film thickness, from 10 nm up to 200 nm, of Ta(5%):TiO₂ and Ta(10%):TiO₂ films. Thickness reduction leads to an increase in transmittance in the visible and near-IR ranges because of the reduced absorption, according to the Lambert-Beer law, while interference fringes change according to the optical path length.⁶³ This modulation of fringe position is a feature that can be exploited in applications in which TCO films are employed as transparent electrodes and a fine tuning of the film transparency at specific wavelengths is required. The effect of thickness in the UV range is more appreciable looking at the absorption coefficient (α), reported in Fig. 7b and d as a function of energy, as evaluated from the Lambert-Beer law (see



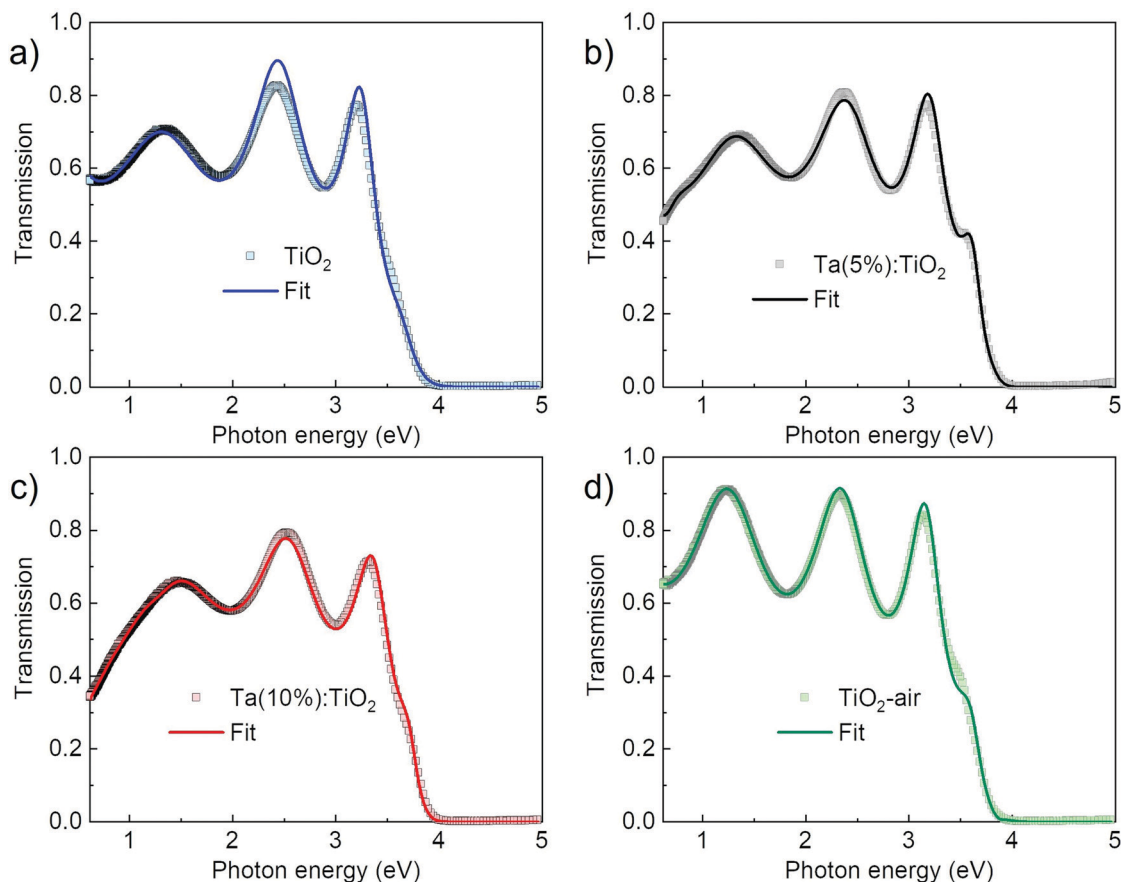


Fig. 6 Transmission measurements of vacuum-annealed (a) TiO_2 , (b) $\text{Ta}(5\%):\text{TiO}_2$, (c) $\text{Ta}(10\%):\text{TiO}_2$ and (d) air-annealed TiO_2 films of 200 nm thickness, grown on soda-lime glass substrates. Squares represent the experimental data while lines represent the theoretical fit, as extracted from ellipsometry measurements.

the Experimental section). The absorption coefficient shows the onset of the absorption band, which does not depend on the thickness down to 50 nm, while it redshifts for 10 and 20 nm-thick films.

Discussion

The optical gap (E_{gap}) of $\text{Ta}:\text{TiO}_2$ thin films was calculated by means of a Tauc plot of the absorption coefficient (see the Experimental section) and the results are reported in Fig. 8 as a function of Ta content and thickness. As a reference, the results of 20, 50 and 200 nm-thick vacuum-annealed TiO_2 films are also reported and the calculated band gap is about 3.29 eV, in line with the band gap of the anatase phase, namely 3.2–3.4 eV.^{10,46} A very similar value of E_{gap} was obtained for air-annealed TiO_2 films.

For the thickest $\text{Ta}:\text{TiO}_2$ films, *i.e.* in the 50–200 nm range, E_{gap} increases with the Ta content as a consequence of the Moss-Burstein effect, where free carriers partially fill the conduction band forcing higher energy optical transitions.⁵⁰ Indeed, the measured charge carrier density of $\text{Ta}(10\%):\text{TiO}_2$ is almost twice that of $\text{Ta}(5\%):\text{TiO}_2$. Congruently, the average band gap calculated for 50–200 nm TiO_2 films doped with 5

and 10 at% of Ta was 3.48 and 3.56 eV, respectively. The value of E_{gap} of $\text{Ta}(5\%):\text{TiO}_2$ films agrees with the work of Mazzolini *et al.*, in which a gap of 3.46 eV was found for the optimal synthesis conditions.¹⁵ In addition, the blue-shift of the E_{gap} in this energy range with doping content has been already observed in other studies on $\text{Ta}:\text{TiO}_2$ and $\text{Nb}:\text{TiO}_2$ thin films when considering a similar amount of dopant atoms.^{36,54,64,65} Specifically, the Moss-Burstein effect responsible for the E_{gap} blue-shift occurs because of the increase of charge carrier density, and it is typical for TCOs. Indeed, in the work of Bin-Bin *et al.*, $\text{Ta}:\text{TiO}_2$ thin films were deposited by PLD with a fixed Ta content (4 at%) and the E_{gap} was found to vary with the post-deposition thermal treatment because of the different amounts of measured charge carrier density.⁶⁶ Even in this case, for the same measured value of carrier density, the E_{gap} agrees with the ones of this work. Finally, Liu *et al.* found an increase of E_{gap} from 3.53 to 3.78 eV with the thickness from 50 to 500 nm for their Ta and Nb co-doped TiO_2 films (already described in the Electrical properties section) because of the increase of charge carrier density,³¹ while in our study the charge carrier density and, as a consequence, the E_{gap} of the films of thickness 50–200 nm is almost constant. On the other hand, E_{gap} calculated for 10–20 nm-thick films decreases down to the value expected for an undoped film, in agreement



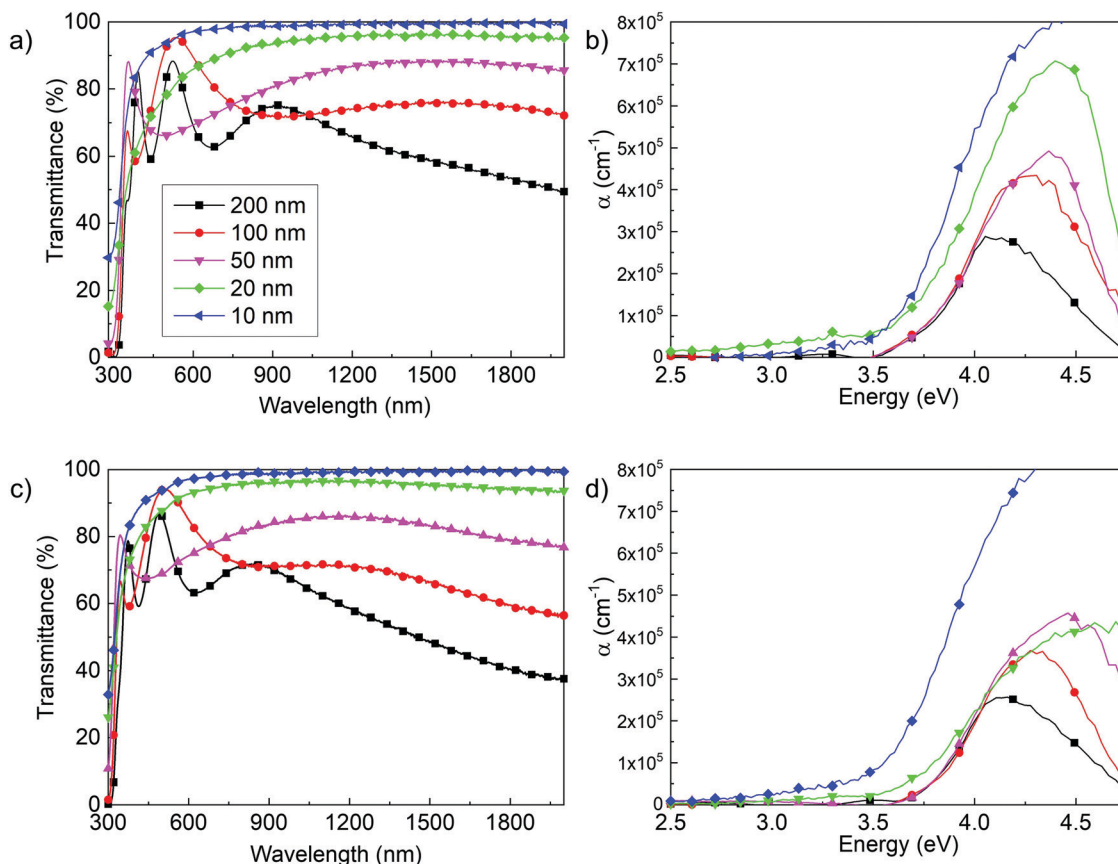


Fig. 7 Optical transmittance and absorption coefficient (α) as a function of thickness (10–200 nm) of (a and b) Ta(5%):TiO₂ and (c and d) Ta(10%):TiO₂ films; the legend of all images is in figure (a).

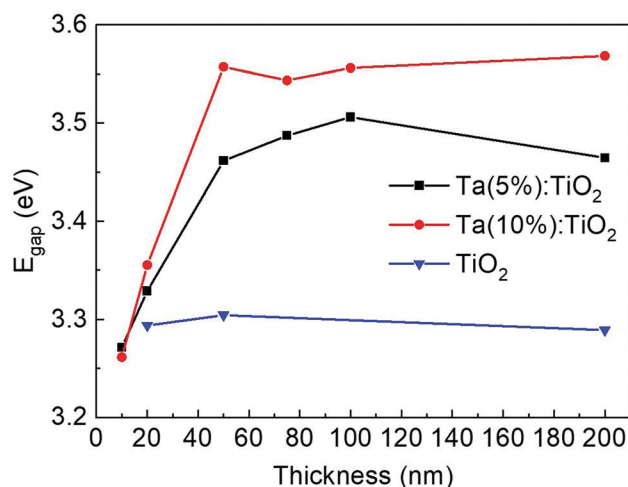


Fig. 8 Optical band gap as a function of thickness (10–200 nm) and Ta content (bare TiO₂, 5 and 10 at%) calculated via the Tauc plot.

with the trend of the charge carrier density which starts to decrease for 20 nm films, down to a value typical of undoped TiO₂ for 10 nm-thick films.

To support this analysis, the optical band gap was calculated from ellipsometric measurements as well. The Tauc plot with α

extracted from ellipsometry for 200 nm-thick films gives results in agreement with the previous findings, *i.e.* 3.28 eV for bare TiO₂, as well as 3.46 and 3.53 eV for Ta(5%):TiO₂ and Ta(10%):TiO₂, respectively.

Another important property for a TCO is the effective mass (m^*) of electrons in the conduction band. For the calculation of m^* we used the Drude model, which describes the motion of free electrons moving against a fixed background of positive ions, while the overall lattice potential effect is taken into account by m^* :

$$\epsilon_{n\text{Drude}} = 1 - \frac{\omega_p^2}{\omega^2 + i\gamma\omega} \quad (1)$$

$$\omega_p = \sqrt{\frac{N_e q^2}{m^* \epsilon_0}} \quad (2)$$

where ω is the photon frequency, ω_p is the unscreened plasma frequency, which can be extracted from the Drude term in the ellipsometric fitting, γ is the damping parameter, ϵ_0 is the vacuum permittivity, q is the electron charge, and m^* is the effective mass of the charge carriers. By knowing N_e from the electrical characterization, *i.e.* the charge carrier density measured electrically, the effective mass can be calculated by combining the above equations. For Ta(5%):TiO₂, such



estimates provide a value for $m^* = 0.82 m_e$, while for Ta(10%):TiO₂, $m^* = 1.89 m_e$. Theoretical calculations for both Nb- and Ta-doped TiO₂ and experimental work on Nb-doped TiO₂ indicate an anisotropic effective mass, related to anisotropy of the anatase tetragonal crystal structure.^{16,67,68} In particular, theoretical calculations report an orthogonal effective mass (m_x^*) and a parallel one (m_z^*) with respect to the tetragonal axis, which are equal to $0.4\text{--}0.6 m_e$ and $>3.5 m_e$, respectively. Moreover, the calculations predict an increase of m_z^* with the charge carrier concentration.¹⁶ Taking into account that the Ta:TiO₂ films reported in this study are polycrystalline, our m^* results are consistent with the reported calculated m_x^* and m_z^* values and, at the same time, m^* shows an increase with the Ta content.

An independent estimate of m^* can also be obtained from the optical bandgap increase as a function of the charge carrier density due to the Moss–Burstein effect. In a simplified parabolic-band picture, the shift in the optical band gap follows the equation:⁶⁹

$$\Delta E_{\text{gap}} = \frac{\hbar^2}{2m^*} (3\pi^2 N_e)^{2/3} \quad (3)$$

from which we can evaluate the effective mass by knowing N_e from the electrical measurements. The ΔE_{gap} is calculated by subtracting the optical gap of bare TiO₂ (3.29 eV) from that of Ta:TiO₂ films (in Fig. 8). Since electrical measurements and optical band gap calculations do not show a significant dependence on thickness in the 50–200 nm range, the effective mass also varies marginally with thickness. Indeed, m^* values of $1.7\text{--}2.2 m_e$ and $1.8\text{--}2.1 m_e$ are obtained for Ta(5%):TiO₂ films and Ta(10%):TiO₂ films, respectively, notably almost the same for both Ta contents. These results are smaller than the effective mass reported in the literature in a previous work on Ta:TiO₂ films that employed the same calculation method (*i.e.* $2.9 m_e$).¹⁵ However, both calculation procedures (eqn (2) and (3)) should be considered only as a qualitative description, since the effect of non-parabolicity of the bands, which could be accounted for by many-body type interactions, is not considered in the model and the extrapolation of the mid-IR plasma frequency from near-IR ellipsometric data is subject to a certain degree of uncertainty.^{70–72} For 20 nm-thick films, larger values of m^* are obtained but the result is affected by the unreliable evaluation of ΔE_{gap} due to the transparency of the film, while for 10 nm films, m^* cannot be estimated at all.

Finally, we provide an estimate for the screened plasma energy (E_p) by revisiting the ellipsometric fitting in the IR range and introducing a real-valued offset, indicated as ε_∞ , to take the contribution of the bound electrons of the semiconductor into account. The value of ε_∞ (equal to about 5.9 and 6.0 for Ta(5%):TiO₂ and Ta(10%):TiO₂, respectively, in good agreement with the literature),^{73,74} allows calculating the screened plasma energy as $E_p = \frac{\hbar\omega}{\sqrt{\varepsilon_\infty}}$. The estimated E_p value is about 0.2 eV for both Ta(5%):TiO₂ and Ta(10%):TiO₂. It should be noted that this estimate is obtained by extrapolating the ellipsometry model out of the measured range and, thus, must be

considered only as a rough indication of the expected plasma energy values. However, these findings can be crosschecked against reflectance and transmittance spectra collected in the 0.1–1.49 eV range by means of Fourier Transform infrared spectroscopy (FTIR, see Fig. S7 of the ESI†). For both doping levels, a reflectivity increase, which is a typical signature of the onset of the plasmonic behavior, is observed in the 0.2–0.3 eV range, in good agreement with the ellipsometric extrapolation. No reflectance decrease is instead observed for bare TiO₂ films. These E_p values in the near- and mid-IR ranges are expected for a TCO with the charge carrier density of the order of 10^{21} cm^{-3} .^{59,75} Unfortunately, no straightforward and physically meaningful combination of Drude and Lorentz contributions allows fitting the FTIR data, a fact that reveals a possibly higher level of complexity and will be the subject of future studies.

Conclusions

In this work, Ta-doped TiO₂ thin films were investigated as a function of Ta content (5 and 10 at%) and thickness (in the 10–200 nm range), focusing on their structural, electrical, and optical properties, while comparing the results with those of bare TiO₂. Homogeneous and compact Ta:TiO₂ films were synthesized *via* pulsed laser deposition followed by a vacuum thermal treatment, obtaining polycrystalline films in the anatase phase with preferential growth along the (101) direction.

The effect of the Ta content on the electrical properties is found to depend on the thickness. For films in the 50–200 nm range, the measured resistivity is between 8×10^{-4} and $1 \times 10^{-3} \Omega \text{ cm}$ for both Ta contents, while the charge carrier density N_e increases proportionally to the Ta content, suggesting a condition of high dopant activation efficiency. On the other hand, the Hall mobility decreases with the Ta content due to a higher concentration of ionized impurities in the crystal lattice, which affects the electron scattering at higher values of N_e . The electrical properties deteriorate for films thinner than 20 nm almost independently of the Ta content, but still conductive films are obtained with 10 nm thickness (with a resistivity of about $0.5 \Omega \text{ cm}$). In addition, the electrical properties of ultra-thin films demonstrated an enhancement by employing a 50 nm-thick buffer layer of polycrystalline and stoichiometric TiO₂. In this case, conductive films were obtained down to a thickness of 5 nm. This result highlights the possibility of designing devices with ultra-thin conductive films, which is a thickness difficult to achieve with metals.

Optical characterization shows that the Ta doping changes the optical response as a consequence of the increased charge carrier density. Indeed, the optical absorption in the UV range blue-shifts with the Ta content, according to the Moss–Burstein effect, while the absorption in the IR range increases because of free carriers.

The estimated optical band gaps increase with the Ta content for films of 50–200 nm thickness, while it starts to decrease for films of 20 nm thickness, down to a value typical for undoped TiO₂ films with 10 nm thickness.



Notably, both the electrical and the optical properties demonstrated no relevant changes in the thickness range from 200 nm down to 50 nm.

The effective mass of Ta:TiO₂ films was estimated as well, and the results are in accordance with theoretical calculations from the literature for a Ta-doped anatase phase. Moreover, the predicted increment of the effective mass with the Ta content is confirmed. Finally, the plasma energy was calculated to be in the near/mid-IR range, as expected for a TCO with a charge carrier density of the order of 10²¹ cm⁻³.

The overall results, especially the fine tunability of the electrical and optical properties by means of the Ta content and the investigation of the thickness dependence of the film properties, show that Ta:TiO₂ films could be suitable as transparent conductive components in optoelectronic applications as well as for plasmonic applications in the IR range.

Experimental section

PLD was performed in a vacuum chamber equipped with mass flow controllers to tune the partial gas pressure. Ablation was performed with a ns-pulsed laser (Nd:YAG, 2nd harmonic, $\lambda = 532$ nm) with pulse duration in the 5–7 ns range and a repetition rate of 10 Hz. The laser pulses were focused on the target material through a viewport and the target was mounted on a roto-translational manipulator ensuring a uniform ablation. Ta:TiO₂ thin films with a nominal content of Ta = 5, 10 at% were grown by ablating 2 inch targets of Ta₂O₅:TiO₂ with a molar ratio equal to 0.025:0.975 and 0.05:0.95, respectively (99.99% pure, provider “Testbourne Ltd”), while pure TiO₂ films were obtained from a 99.9% pure TiO₂ target (provider “Kurt J. Lesker”). We name the doped films Ta(5%):TiO₂ and Ta(10%):TiO₂, according to their nominal content of Ta. The laser fluence (F) on the target was set at 2.27 J cm⁻² for the deposition of TiO₂ and Ta(5%):TiO₂, while $F = 2.73$ J cm⁻² was employed for Ta(10%):TiO₂. Depositions were performed at room temperature and in a pure O₂ background at a pressure of 1 Pa. The employed substrates were Si(100) and soda-lime glass cleaned in an ultrasonic bath with isopropanol and mounted on a rotating sample holder at a fixed target-to-substrate distance of 50 mm. Thin films with different nominal thicknesses, from 10 to 200 nm, were deposited by varying the deposition duration and the actual thickness values were evaluated by means of a field emission scanning electron microscope (FEG-SEM, Zeiss SUPRA 40) on samples grown on silicon. The deposition rate is related to the laser fluence and was estimated around 11 nm min⁻¹ for TiO₂ and Ta(5%):TiO₂ and 18 nm min⁻¹ for Ta(10%):TiO₂. The FEG-SEM microscope is equipped with an Oxford Instruments Si(Li) detector for energy-dispersive X-ray spectroscopy (EDS), which was employed in this work to estimate the tantalum atomic percentage (at%) with an accelerating voltage of 10 kV.

All the as-deposited TiO₂-based thin films were amorphous, therefore post deposition annealing treatments in a vacuum atmosphere ($p < 4 \times 10^{-5}$ Pa) in a homemade furnace at 550 °C

(10 °C min⁻¹ ramp, 1 h dwell) were employed to obtain polycrystalline anatase films, according to previous studies.⁷⁶

The thin film surfaces were investigated by means of optical microscopy (Leitz orthoplan-pol). Structural characterization was carried out by X-ray diffraction (XRD) and Raman spectroscopy. XRD patterns were collected using a Bruker D8 Advance X-ray diffractometer at 293 K (Cu K α 1 radiation – 1.5406 Å). The measurements were carried out in a Bragg–Brentano geometry with a step-scan technique and a 2θ range of 10–70°. Data were acquired by using a Lynx Eye detector in continuous scanning mode with a step size of 0.038° and a time step of 0.15 s. The mean crystallite size domain (τ) along the film growth direction and the lattice parameters of the samples were determined by using the full profile fitting Pawley method.⁷⁷ The Pawley method was carried out using the program-suite TOPAS from Bruker. Raman spectra of the samples were collected using a Renishaw InVia micro Raman spectrometer with an argon ion laser (514 nm), with a laser power of 0.13 mW and a spectral resolution of about 3 cm⁻¹.

Electrical characterization and Hall effect measurements were performed on TiO₂ and Ta:TiO₂ films deposited on soda-lime glass in a four-point probe configuration with a Keithley K2400 source/measure unit as a current generator (from 100 nA to 10 mA), an Agilent 34970A voltage meter, and a 0.57 T Ecopia permanent magnet.

Optical transmittance (T) and reflectance (R) spectra in the spectral range of 250–2000 nm were evaluated on samples deposited on soda-lime glass with a UV-vis-NIR PerkinElmer Lambda 1050 spectrophotometer with a 150 mm diameter integrating sphere. All the acquired spectra were normalized with respect to the glass substrate contribution. The absorption coefficient (α) was evaluated from the Lambert–Beer law $\alpha = -1/d \ln[T/(1 - R)]$, where d is the film thickness. The optical energy gap (E_{gap}) was calculated by means of a Tauc plot, where $(\alpha h\nu)^n$ versus $h\nu$ was plotted in the proximity of the absorption onset of the films in the UV range and the exponent n was chosen equal to 0.5 because of the indirect band gap of anatase; E_{gap} was extrapolated with the intercept in the energy axis using a linear fit.

Fourier transform infrared spectroscopy (FTIR) was performed both in transmission and reflection geometries with a commercial instrument (INVENIO-R by Bruker), employing a halogen lamp emitting in the near-IR range and a global source emitting in the mid-IR spectral range, together with a mercury cadmium telluride (MCT) detector to acquire spectra in the 0.1–1.49 eV spectral range. The beam incidence on the sample is almost normal (about 10° incidence angle) and the diameter of the beam in the focus is about 1 mm. For these measurements, films with a thickness of 670 nm (in order to have enough transmittance contrast in the measured range) were deposited on CaF₂ substrates.

Spectroscopic ellipsometry (SE) was performed by means of a J. A. Woollam V-VASE ellipsometer (0.5–5.05 eV energy range, incidence angle of 60°). The analysis involves evaluating the ellipsometric angles $\psi(\lambda)$ and $\Delta(\lambda)$, defined as $\frac{r_p}{r_s} = \tan \Psi \cdot e^{i\Delta}$,



where $r_{p(s)}$ is the Fresnel reflection coefficient for p(s)-polarized radiation. The measurement of amplitude and phase-related quantities adds a degree of precision, useful for deducing the dielectric function of materials, even with complex nanoscale morphology.⁷⁸ From the optical point of view, the system was modelled as a stack of dielectric layers, with each of them being characterized by its thickness and complex dielectric function. The optical response of the system was calculated assuming Fresnel boundary conditions at the interface between the layers. Bottom to top, the model included: (i) a semi-infinite Si substrate, (ii) a native oxide layer, (iii) the Ta:TiO₂ film and (iv) a roughness layer (the latter modelled as a Bruggemann effective-medium layer composed of 50% of Ta:TiO₂ and 50% of voids). For the modelling of the optical properties of Ta:TiO₂ films, we employed a combination of Lorentz, Lorentz–Gaussian and so-called PSEMI oscillators, along with a Drude-type contribution to take the doping-induced free carriers into account. PSEMI oscillators are parameterized functions widely employed to model the optical response of crystalline semiconductors.^{79,80} The oscillator parameters of the undoped TiO₂, and the Ta:TiO₂, layers, as well as the thickness of all the optical layers, were carefully fitted in order to achieve the best agreement between the experimental data and the simulated SE spectra. The thickness of the SiO₂ native oxide was assessed by SE measurements on a bare substrate and was found to be around 2 nm.

Author contributions

Beatrice R. Bricchi: writing – original draft, writing – review & editing, data curation, formal analysis, investigation and conceptualization. Maria Sygletou and Giancarlo Terraneo: formal analysis, investigation and writing – review & editing. Luca Ornago: formal analysis, investigation, conceptualization and writing – review & editing. Cristina Mancarella, Lorenzo Stasi, Francesco Rusconi, Erika Moggi and Matteo Ghidelli: investigation and writing – review & editing. Andrea Li Bassi, Francesco Bisio and Paolo Biagioni: supervision, conceptualization writing – review & editing.

Conflicts of interest

There are no conflicts to declare.

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References

- 1 S. C. Dixon, D. O. Scanlon, C. J. Carmalt and I. P. Parkin, *J. Mater. Chem. C*, 2016, **4**, 6946–6961.
- 2 D. H. Kim, S. Lee, J. H. Park, J. H. Noh, I. J. Park, W. M. Seong and K. S. Hong, *Sol. Energy Mater. Sol. Cells*, 2021, **96**, 276–280.
- 3 S. Dias, S. Chirakkara, N. Patel and S. B. Krupanidhi, *J. Mater. Sci.: Mater. Electron.*, 2018, **29**, 2131–2139.
- 4 T. Hanada, T. Negishi, I. Shiroishi and T. Shiro, *Thin Solid Films*, 2010, **518**, 3089–3092.
- 5 S. H. Park, J. B. Park and P. K. Song, *Curr. Appl. Phys.*, 2010, **10**, S488–S490.
- 6 D. S. Ginley and J. D. Perkins, *Handbook of Transparent Conductors*, Springer, Boston, MA, 2011.
- 7 F. Wang, M. Z. Wu, Y. Y. Wang, Y. M. Yu, X. M. Wu and L. J. Zhuge, *Vacuum*, 2013, **89**, 127–131.
- 8 G. Wan, S. Wang, X. Zhang, M. Huang, Y. Zhang, W. Duan and L. Yi, *Appl. Surf. Sci.*, 2015, **357**, 622–625.
- 9 R. L. Moss, E. Tzimas, H. Kara, P. Willis and J. Kooroshy, Assessing Rare Metals as Supply-Chain Bottlenecks in Low-Carbon Energy Technologies, JRC Scientific and Technical Reports, EUR 24884, 2011.
- 10 H. Tang, K. Prasad, R. Sanjinés, P. E. Schmid and F. Lévy, *J. Appl. Phys.*, 1994, **75**, 2042–2047.
- 11 A. Fujishima, T. N. Rao and D. A. Tryk, *J. Photochem. Photobiol., C*, 2000, **1**, 1–21.
- 12 T. Bak, J. Nowotny, M. Rekas and C. C. Sorrell, *Int. J. Hydrogen Energy*, 2002, **27**, 991–1022.
- 13 A. Hagfeldt, G. Boschloo, L. Sun, L. Kloo and H. Pettersson, *Chem. Rev.*, 2010, **110**, 6595–6663.
- 14 J. Osorio-Guillén, S. Lany and A. Zunger, *Phys. Rev. Lett.*, 2008, **100**, 036601.
- 15 P. Mazzolini, P. Gondoni, V. Russo, D. Chrastina, C. S. Casari and A. Li, Bassi, *J. Phys. Chem. C*, 2015, **119**, 6988–6997.
- 16 H. Anh Huy, B. Aradi, T. Frauenheim and P. Deák, *J. Appl. Phys.*, 2012, **112**, 016103.
- 17 Y. Furubayashi, T. Hitosugi, Y. Yamamoto, K. Inaba, G. Kinoda, Y. Hirose, T. Shimada and T. Hasegawa, *Appl. Phys. Lett.*, 2005, **86**, 252101.
- 18 T. Hitosugi, Y. Furubayashi, A. Ueda, K. Itabashi, K. Inaba, Y. Hirose, G. Kinoda, Y. Yamamoto, T. Shimada and T. Hasegawa, *Jpn. J. Appl. Phys.*, 2005, **44**, L1063.
- 19 M. Neubert, S. Cornelius, J. Fiedler, T. Gebel, H. Liepack, A. Kolitsch and M. Vinnichenko, *J. Appl. Phys.*, 2013, **114**, 083707.
- 20 D. C. Qi, A. R. Barman, L. Debbichi, S. Dhar, I. Santoso, T. C. Asmara, H. Omer, K. Yang, P. Krüger, A. T. Wee and T. Venkatesan, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2013, **87**, 245201.
- 21 H. Kamisaka, T. Hitosugi, T. Suenaga, T. Hasegawa and K. Yamashita, *J. Chem. Phys.*, 2009, **131**, 034702.
- 22 S. A. Maier, *Curr. Nanosci.*, 2005, **1.1**, 17–22.
- 23 A. V. Kildishev, A. Boltasseva and V. M. Shalaev, *Science*, 2013, **339**, 6125.
- 24 G. V. Naik, V. M. Shalaev and A. Boltasseva, *Adv. Mater.*, 2013, **25**, 3264–3294.
- 25 Z. Wang, C. Chen, K. Wu, H. Chong and H. Ye, *Phys. Status Solidi A*, 2019, **216**, 1700794.
- 26 A. V. Chichagov, Information-calculating system on crystal structure data of minerals (MINCRYST), *Mater. Sci. Forum*, 1994, 193–198.



- 27 P. M. Oliver, G. W. Watson, E. T. Kelsey and S. C. Parker, *J. Mater. Chem.*, 1997, **7**, 563–568.
- 28 C. Yang, Y. Hirose, S. Nakao and T. Hasegawa, *J. Cryst. Growth*, 2013, **376**, 66–69.
- 29 A. Li Bassi, D. Cattaneo, V. Russo and C. E. Bottani, *J. Appl. Phys.*, 2005, **98**, 074305.
- 30 U. Balachandran and N. G. Eror, *J. Solid State Chem.*, 1982, **42**, 276–282.
- 31 Y. Liu, Y. D. Qiao and G. Yang, *Appl. Phys. A: Mater. Sci. Process.*, 2018, **124**, 530.
- 32 P. Mazzolini, V. Russo, C. S. Casari, T. Hitosugi, S. Nakao, T. Hasegawa and A. Li Bassi, *J. Phys. Chem. C*, 2016, **120**, 18878–18886.
- 33 P. Deák, B. Aradi and T. Frauenheim, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2012, **86**, 195206.
- 34 T. Hitosugi, A. Ueda, S. Nakao, N. Yamada, Y. Furubayashi, Y. Hirose, T. Shimada and T. Hasegawa, *Appl. Phys. Lett.*, 2007, **90**, 212106.
- 35 N. Yamada, T. Hitosugi, N. L. H. Hoang, Y. Furubayashi, Y. Hirose, S. Konuma, T. Shimada and T. Hasegawa, *Thin Solid Films*, 2008, **516**, 5754–5757.
- 36 Y. Sato, H. Akizuki, T. Kamiyama and Y. Shigesato, *Thin Solid Films*, 2008, **516**, 5758–5762.
- 37 B. Z. Dong, G. J. Fang, J. F. Wang, W. J. Guan and X. Z. Zhao, *J. Appl. Phys.*, 2007, **101**, 033713.
- 38 D. C. Look, K. D. Leedy, A. Kiefer, B. Clafflin, N. Itagaki, K. Matsushima and I. Suhariadi, *Opt. Eng.*, 2013, **52**, 033801.
- 39 D. C. Look, *Mater. Sci. Semicond. Process.*, 2017, **69**, 2–8.
- 40 S. Agarwal, M. S. Haseman, K. D. Leedy, D. J. Winarski, P. Saadatkhia, E. Doyle, L. Zhang, T. Dang, V. S. Vasilyev and F. A. Selim, *J. Electron. Mater.*, 2018, **47**, 2271–2276.
- 41 T. Okada, T. Kawashima, T. Mori and K. Washio, *Thin Solid Films*, 2020, **701**, 137954.
- 42 J. Nomoto, J. Oda, T. Miyata and T. Minami, *Thin Solid Films*, 2010, **519**, 1587–1593.
- 43 C. Lin, X. Li and C. Xu, *J. Mater. Sci.: Mater. Electron.*, 2019, **30**, 721–730.
- 44 G. E. Jellison, L. A. Boatner, J. D. Budai, B. S. Jeong and D. P. Norton, *J. Appl. Phys.*, 2003, **93**, 9537–9541.
- 45 S. Tanemura, L. Miao, P. Jin, K. Kaneko, A. Terai and N. Nabatova-Gabain, *Appl. Surf. Sci.*, 2003, **212–213**, 654–660.
- 46 N. Martin, C. Rousselot, D. Rondot, F. Palmino and R. Mercier, *Thin Solid Films*, 1997, **300**, 113–121.
- 47 L. Sun and P. Hou, *Thin Solid Films*, 2004, **455–456**, 525–529.
- 48 A. V. Manole, M. Dobromir, M. Gîrtan, R. Mallet, G. Rusu and D. Luca, *Ceram. Int.*, 2013, **39**, 4771–4776.
- 49 Y. R. Park and K. J. Kim, *Thin Solid Films*, 2005, **484**, 34–38.
- 50 E. Burstein, *Phys. Rev.*, 1954, **93**, 632–633.
- 51 W. Zhao, L. He, X. Feng, C. Luan and J. Ma, *CrystEngComm*, 2018, **20**, 5395–5401.
- 52 H. Poelman, H. Tomaszewski, D. Poelman, D. Depla and R. D. Gryse, *Surf. Interface Anal.*, 2004, **36**, 1167–1170.
- 53 A. Brudnik, H. Czernastek, K. Zakrzewska and M. Jachimowski, *Thin Solid Films*, 1991, **199**, 45–58.
- 54 C. M. Maghanga, G. A. Niklasson and C. G. Granqvist, *Thin Solid Films*, 2009, **518**, 1254–1258.
- 55 Y. Liu, Q. Peng, Y. Qiao and G. Yang, *J. Electron. Mater.*, 2018, **47**, 5334–5343.
- 56 S. Muhammadiyah, E. Nurfani, R. Kurniawan, I. M. Sutjahja, T. Winata and Y. Darma, *Mater. Res. Express*, 2017, **4**, 024002.
- 57 G. V. Naik, J. Kim and A. Boltasseva, *Opt. Mater. Express*, 2011, **1**, 1090–1099.
- 58 J. Kim, G. V. Naik, N. K. Emani, U. Guler and A. Boltasseva, *IEEE J. Sel. Top. Quantum Electron.*, 2013, **19**, 3, 4601907.
- 59 A. Agrawal, R. W. Johns and D. J. Milliron, *Annu. Rev. Mater. Res.*, 2017, **47**, 1–31.
- 60 B. B. Varga, *Phys. Rev.*, 1965, **137**, 1896–1902.
- 61 A. Mooradian and G. B. Wright, *Phys. Rev. Lett.*, 1966, **16**, 999–1001.
- 62 M. Schubert, *Infrared Ellipsometry on Semiconductor Layer Structures: Phonons, Plasmons, and Polaritons*, Springer Science & Business Media, 2004.
- 63 M. Bass, E. W. Van Stryland, D. R. Williams and W. L. Wolfe, *Handbook of Optics: Fundamentals, techniques, and design*, McGraw-Hill, 1995.
- 64 W. Zhao, L. He, X. Feng, H. Xiao, C. Luan and J. Ma, *Ceram. Int.*, 2018, **44**, 21114–21119.
- 65 E. Nurfani, I. M. Sutjahja, T. Winata, A. Rusydi and Y. Darma, *AIP Conf. Proc.*, 2015, **1677**, 1.
- 66 W. Bin-Bin, P. Feng-Ming and Y. Yu-E, *Chin. Phys. Lett.*, 2011, **28**, 118102.
- 67 H. A. Huy, B. Aradi, T. Frauenheim and P. Deák, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2011, **83**, 155201.
- 68 Y. Hirose, N. Yamada, S. Nakao, T. Hitosugi, T. Shimada and T. Hasegawa, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2009, **79**, 165108.
- 69 T. J. Coutts, D. L. Young and X. Li, *MRS Bull.*, 2000, **25**, 58–65.
- 70 A. Abdolazadeh Ziabari and S. M. Rozati, *Phys. B*, 2012, **407**, 4512–4517.
- 71 A. Jain, P. Sagar and R. M. Mehra, *Solid-State Electron.*, 2006, **50**, 1420–1424.
- 72 B. E. Sernelius, K. F. Berggren, Z. C. Jin, I. Hamberg and C. G. Granqvist, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1988, **37**, 10244–10248.
- 73 R. J. Gonzalez, R. Zallen and H. Berger, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1997, **55**, 7014–7017.
- 74 M. Eldlio, F. Che and M. Cada, *IAENG Transactions on Engineering Technologies*, 2014, pp. 41–49.
- 75 T. Taliencio and P. Biagioni, *Nanophotonics*, 2019, **8**, 949–990.
- 76 P. Mazzolini, T. Acartürk, D. Chrastina, U. Starke, C. S. Casari, G. Gregori and A. Li Bassi, *Adv. Electron. Mater.*, 2016, **2**, 1500316.
- 77 G. S. Pawley, *J. Appl. Crystallogr.*, 1981, **14**, 357–361.
- 78 C. Toccafondi, S. Uttiya, O. Cavalleri, G. Gemme, E. Barborini, F. Bisio and M. Canepa, *J. Phys. D: Appl. Phys.*, 2014, **47**, 48, 485301.
- 79 B. Johs, C. M. Herzinger, J. H. Dinan, A. Cornfeld and J. D. Benson, *Thin Solid Films*, 1998, **313–314**, 137–142.
- 80 M. Magnozzi, S. Terreni, L. Anghinolfi, S. Uttiya, M. M. Carnasciali, G. Gemme, M. Neri, M. Principe, I. Pinto, L.-C. Kuo, S. Chao and M. Canepa, *Opt. Mater.*, 2018, **75**, 94–101.

