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# Characterising and communicating the potential hazard posed by potentially toxic elements in indoor dusts from schools across Lagos, Nigeria†

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Ambient and indoor air pollution results in an estimated 7 million premature deaths globally each year, representing a major contemporary public health challenge, but one poorly quantified from a toxicological and source perspective. Indoor exposure represents possibly the greatest potential overall exposure, yet our indoor environments are still poorly understood, modelled and characterized. In rapidly growing cities, such as Lagos, Nigeria, environmental monitoring can play an important role in establishing baseline data, monitoring urban pollution trends and in environmental education. Classroom dust samples were collected from 40 locations from across the twenty local government areas (LGAs) of Lagos, in June 2019. The aim of the study was to assess the potential hazard posed by PTE in indoor dusts and to develop a suitable risk communication strategy to inform and educate the public, promoting environmental health literacy. Concentrations of total PTE in indoor dusts were assessed using Energy Dispersive X-Ray Fluorescence (ED-XRF) spectrometry. Oral bioaccessibility determinations using the unified BARGE method, and analysis by Inductively Coupled Plasma Optical Emission Spectrometry (ICP-OES) were also performed on the dust samples to determine the fraction available for absorption in the gastrointestinal tract. Results showed that the indoor dust samples were largely uncontaminated, with only few exceptions (2 samples). Enrichment factor pollution trend for the total PTE concentrations was in the order of Pb > Zn > U > Cr > Cu > Ba > Mn > V > As > Cd > Ni > Al. Source apportionment studies using factor analysis suggests concentrations of Al, As, Fe, Mn, Ni, and U may be influenced largely by lithogenic factors, while Cd, Cu and Pb originated principally from anthropogenic sources. Chromium, V and Zn appear to originate from mixed sources of both lithogenic and anthropogenic origin. Our oral bioaccessibility determinations indicate that the assumption of 100% bioavailability based on pseudototal or total concentrations would overestimate the hazard potential of PTE in these indoor dusts. Zinc was the most bioaccessible PTE (mean of 88%), with Mn (57%), Pb (48%), Ba (48%), Al (41%), As (37%), Cu (36%), Ni (28%), Cr (10%) and Fe (7%) the least bioaccessible. Human health risk assessment, for both children and adults using the bioaccessible fraction, showed values to be within acceptable risk levels.

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## Environmental significance

Exposure to particulate matter can initiate or enhance disease in humans, yet the nature of the hazard that indoor dust presents remains poorly characterized from a toxicological and a source perspective. Our research provides key information about the spatial distributions and concentrations of contaminants in indoor dusts from school environments in Lagos, Nigeria, providing geochemical baseline information and source characterization of 12 Potentially Toxic Elements (PTE). A range of contamination assessment criteria were employed, as well as a modified approach incorporating oral bioaccessibility measurements into human exposure risk models. The online 'dust atlas' and ancillary supporting information enables participants to interactively view the levels of PTE in indoor dusts across all participating regions and the range of practical actions to improve indoor environmental quality.

## 1. Introduction

Urbanization and industrialization have contributed to increased pollution levels in cities, including the contamination of environment media such as soil, dusts, and particulate matter (PM) with potentially toxic elements, PTE.<sup>1</sup> Particulate matter can pose serious risks to human health and an

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increasing number of associations are now being reported between air particulates and a broad number of disease outcomes.<sup>2</sup> Ambient and indoor air pollution results in an estimated 7 million premature deaths globally each year,<sup>3</sup> representing a major contemporary public health challenge. Health impacts include increased incidences of strokes,<sup>4</sup> Alzheimer's,<sup>5</sup> respiratory disease,<sup>6</sup> cardiovascular disease<sup>7</sup> and cancer.<sup>8</sup>

Dusts contain harmful agents, including metals/metalloids, organic substances, microbes and allergens. Indoor dust is a crucial, often ignored, part of the daily exposure to PM given the amount of time people spend indoors (in homes, offices and schools); estimated to be about 88% of the day for adults and 71–79% of the day for children.<sup>9</sup> Potential sources of dust in the indoor environment include ingress of ambient particles, cooking, smoking, cleaning, wear and tear, and soils transported on clothes and footwear. However, incidental or deliberate ingestion of indoor dust can be an important non-dietary pathway by which PTE get into human body, especially children. Involuntary inhalation of resuspended dust particles in the indoor environment coupled with dermal contact are other PTE exposure pathways to humans.<sup>10</sup>

It is increasingly recognized that the human health risk associated with PM is not solely related to the physical abundance of the PM, or total or pseudototal PTE concentration, but with the biologically available portion of the contaminant.<sup>11</sup> Bioaccessibility is a surrogate *in vitro* estimation of bioavailability. Oral bioaccessibility of a contaminant is the fraction extractable in the human gastro-intestinal (GI) system.<sup>12</sup> Indeed, a significant portion of the PM inhaled also reaches the GI system through particle clearance mechanisms in the human pulmonary system.<sup>13</sup> In addition to lung clearance mechanisms, indoor dust has a direct oral ingestion pathway to the GI system through hand-to-mouth activity. When these fractions reach the GI tract the PM reacts with gastric juices and the PTE released represent the orally available (bioaccessible) fraction.<sup>14</sup>

The 2019 United Nations world population statistics revealed that more than half of the world population growth will occur in Africa between now and the year 2050.<sup>15</sup> Environmental degradation, with accompanying health impacts, may be one of the resultant effects of the projected population growth, and associated urbanization and industrialization. Several studies which include oral bioaccessibility and risk assessment of PTE in indoor dusts have been reported from a range of countries globally,<sup>16–21</sup> however previous studies on PM in Nigeria have largely focused on the outdoor environment, with soil and street dusts the prime research focus.<sup>22,23</sup> Recent relevant PM concerns for the Nigerian community include release and resuspension of mining and smelting sources, power generation emissions, traffic, and major fire incidents.<sup>24</sup> Lagos is Nigeria's most populous city and seventh fastest growing city in the world.<sup>25</sup> Lagos, like many other urban centres across Nigeria, has experienced an unprecedented increase in urban migration, but one amplified by the fact that it serves as the commercial nerve centre of Nigeria.<sup>10</sup> Lagos State Government estimated the population of Lagos as 17.5 million during a parallel census conducted in 2006 with over 12 million people living in the

urban areas. A more recent account reported its population as 21 million, making it the largest city in Africa.<sup>26</sup> In comparison to most developed countries, developing countries typically lack suitable public and environmental health standards and thus residential, community and educational indoor environments can be a source of unacceptable exposure to PM and to PTE in these indoor dusts. This problem can be exacerbated due to the lack of, or poor implementation of, environmental sanitation policies. The aim of this current study is to (i) assess the potential hazard posed by PTE in indoor dusts from schools across Lagos and, (ii) to develop and implement a risk communication strategy to help inform and educate the public around the potential hazards of PM, promoting environmental health literacy to reduce exposure. Specific objectives include: (1) to assess the level of PTE in indoor dusts collected from nursery and primary schools in all 20 local government areas of Lagos state, (2) application of an oral bioaccessibility procedure to indoor dusts (250  $\mu\text{m}$  fraction) to determine the fraction of PTE available for absorption, (3) assessment of the human health risk of PTE in classroom dust to children and adults in Lagos environs, and (4) dissemination of the research findings to the local communities in an accessible format. To widen the reach of our environmental health literacy message the focus of our indoor dust collection was on schools.

Equipping the next generation, and more broadly the wider public, with better information about dust-related hazards in their local environment has the potential to improve the health and wellbeing of communities by empowering citizens to take ownership of response strategies.

## 2. Materials and methods

### 2.1 Sampling

Dust samples were collected from 40 locations from across the twenty local government areas (LGAs) of Lagos. The sampling campaign was conducted in June 2019 and two schools were sampled in each of the LGAs. Classroom dusts were collected using a combination of short sweeping brush and/or long handled broom to sweep together dust from across the room to be sampled. Floor dust from five to ten classrooms (according to both the size of the school and the volume of dust available) across an individual school were bulked together to obtain a composite sample for that school. All samples were stored in polythene bags, labelled and taken to the Lagos State Environmental Protection Agency (LASEPA) laboratory for further processing. Samples were air-dried, refuse and small stones were removed, and samples stored prior to transportation to Northumbria University, Newcastle upon Tyne, England UK for subsequent processing.

### 2.2 Sample preparation and analysis

Samples were sieved through 250  $\mu\text{m}$  plastic mesh and approximately 4.0 g of the resultant PM (weighed to 4 decimal places) was made into pellets with a Cereox binder (Fluxana, Borschelstr, Germany) in ratio 4 : 1 for total metal determination analysis using an Energy Dispersive X-Ray Fluorescence



(ED-XRF) spectrometer (manufactured by Spectro Xepos Kleve, Germany). The equipment uses an adaptive excitation, advance tube design and high count throughput detection system which allows for low limits of detection for a wide range of elements.

The *in vitro* oral bioaccessibility extraction protocol employed in this work was based on the unified BARGE method (UBM).<sup>27</sup> This *in vitro* extraction method has been widely employed and validated for oral bioaccessibility measurements in soil and dusts.<sup>28–30</sup> The simulated digestive fluids were prepared and stored in the refrigerator 1 day before extraction commences.<sup>31</sup> Approximately 0.6 g (weighed to 4 decimal places) of dust sample was weighed into individual polycarbonate extraction tubes and mixed with 9.0 ml of saliva. After a quick manual shaking (10 seconds), 13.5 ml of gastric solution was pipetted and added. The pH of the solution was adjusted to  $1.20 \pm 0.05$  with HCl (37%). The tubes were shaken using an end-over-end rotator inserted into a thermostatic water bath (@ 37 °C) for 1 hour and then centrifuged at 4500 g for 15 minutes. The supernatant was thereafter carefully pipetted and was acidified using HNO<sub>3</sub> and stored in a refrigerator prior to inductively coupled plasma optical emission spectrometry (ICP-OES) using a PerkinElmer Optima 8000 (configured with software Winlab 32 for ICP, version 5.5), Waltham Massachusetts USA.

Determination of the pseudo-total metal concentration was undertaken using a microwave digestion technique. Approximately 0.50 g (weighed to 4 decimal places) of dust was weighed into Teflon digestion vessels and 12.0 ml of *aqua-regia* mixture was added. The sample was heated in a temperature-controlled microwave digestion system (CEM MARS 6™, Buckingham UK). After digestion, any violent chemical reactions were allowed to subside as the vessels cooled to near room temperature. Digests were filtered and diluted using deionized water and stored in a refrigerator prior to ICP-OES analysis.

### 2.3 Quality control

The potential of ED-XRF to quantify a wide range of elements is well known, but the number actually determined is dependent upon the calibration standards used, and the elements they are

quantified for. For the ED-XRF analysis, calibration for major, minor and trace elements was performed by measuring a series of fourteen certified reference materials (CRMs: GBW 7308, 7310, 7313, 7401, 7402, 7403, 7404, 7405, 7406, 7407, 7411, 7603, SRM 2710A, BCR 146R; further details of each CRM are provided in ESI Table 1†). By comparing certified concentrations with our determined ED-XRF concentrations we calculated individual elemental correction factors which were then applied to 'calibrate' the dust data. Good agreement between certified and determined values was obtained for all of our PTEs (regression  $R^2$  values all exceeded 93%), and individual element correction factors ranged from 0.812 for Ba to 1.15 for Cd.

For improved short-term precision (repeatability), each pellet is rotated during acquisition of the spectra in the Xepos, and as part of our quality control procedures we additionally ran selected sample pellets three times to monitor the % relative standard deviation (% RSD). Where sample concentrations were above 1.0 mg kg<sup>-1</sup>, % RSDs ranged from <1% to 10% for all of the elements of interest. Instrument limits of detection (LOD, determined for a pure quartz sample) are: Al 27 mg kg<sup>-1</sup>; As 0.1 mg kg<sup>-1</sup>; Ba 2.0 mg kg<sup>-1</sup>; Cd 0.2 mg kg<sup>-1</sup>; Cr 0.2 mg kg<sup>-1</sup>; Cu 0.5 mg kg<sup>-1</sup>; Fe 1.2 mg kg<sup>-1</sup>; Mn 0.2 mg kg<sup>-1</sup>; Ni 1.0 mg kg<sup>-1</sup>; Pb 0.2 mg kg<sup>-1</sup>; V 0.3 mg kg<sup>-1</sup>; Zn 0.2 mg kg<sup>-1</sup>, and U 0.2 mg kg<sup>-1</sup>, respectively, with all determinations made using the instruments TurboQuant II software (Spectro Analytical Instruments, XRF-113-AB).<sup>32</sup>

All sample extractions were run in triplicate with procedural blanks (*i.e.* reagents only, used for blank subtraction and to monitor for any contamination introduced during the laboratory processing) included in each batch. For ICP-OES analysis of the extracts, calibration curves were prepared for the PTE under investigation, using aqueous standards across the calibration range from 0 to 100 mg L<sup>-1</sup>. The ICP-OES six-point calibration curves produced linear graphs over the determined ranges, with regression coefficient ( $R^2$ ) data of at least 0.998 for all PTE analysed. An internal standard solution (indium) was used to monitor and correct for instrument drift and 2% nitric acid was used to achieve appropriate dilutions. Instrument drift was additionally monitored by analysing calibration standards after

**Table 1** Recovery of BGS 102 (UBM extraction and ICP-OES analysis;  $n$  = number of samples analyzed; mg kg<sup>-1</sup>)

	Al	As	Ba	Cd	Cr	Cu	Fe	Mn	Ni	Pb	V	Zn
$n = 5$ Determined mean	2500 ± 198	5.5 ± 0.51	118 ± 6.73	3.54 ± 10.6	36.7 ± 2.58	5.82 ± 1.41	545 ± 299	3470 ± 268	13.1 ± 1.34	17.1 ± 1.53	1.0 ± 1.24	46.2 ± 3.49
$n = 50$ Reported data <sup>33</sup>	2568 ± 245	3.9 ± 0.36	72.2 ± 6.02	0.24 ± 0.03	36.7 ± 2.48	8.5 ± 1.03	854 ± 237	2979 ± 294	13.0 ± 1.27	15.2 ± 2.97	6.1 ± 0.91	41.5 ± 4.45
% recovery	97	140	163	NIL	100	68	64	117	101	112	16	111

**Table 2** Recovery of BCR 143R (*aqua-regia* digestion and ICP-OES analysis; mg kg<sup>-1</sup>)

	Cd	Cr	Mn	Ni	Pb	Zn
Determined mean ( $n = 6$ )	65.1 ± 4	458 ± 69	783 ± 120	272 ± 13	160 ± 7	1060 ± 120
BCR 143R certificate values	72.0 ± 1.8	426 ± 12	858 ± 11	296 ± 4	174	1063 ± 16
% Recovery	90	109	91	92	92	100



every tenth sample measurement. BGS 102 guidance material was used to determine the accuracy of the determinations in the UBM procedure<sup>33</sup> (Table 1), and BCR 143R for the *aqua-regia* digests (Table 2).

## 2.4 Pollution indices

A wide array of methods for determining metal enrichment have been reported in the literature.<sup>34–36</sup> All come with their own advantages and disadvantages associated with their implementation. Given these statistical approaches have typically been applied to media and settings other than indoor PM we took a pragmatic approach and selected enrichment factors as one of the more commonly applied approaches.<sup>35</sup>

Enrichment factor (EnF) of PTE in the indoor dust samples were calculated using eqn (1):

$$\text{EnF} = \frac{X_a \text{ (test element in sample)} / X_b \text{ (reference element in sample)}}{B_a \text{ (background value for test element)} / B_b \text{ (background value for reference element)}} \quad (1)$$

where  $X_a$  is the concentration of the test PTE measured in the sample,  $X_b$ , the concentration of reference element in the sample,  $B_a$  is the background value for the test element in crustal rock and  $B_b$  is the background value for reference element in crustal rock. In keeping with several other studies,<sup>37–39</sup> Fe was chosen as the reference element since it is one of the most abundant elements in the earth's crust and its geochemistry is similar to many of the PTE determined in this study. Furthermore, its standard deviation of successive measurements in the indoor dusts measurements were generally good (ESI; Table 2†). As background levels of PTE are yet to be established in Nigeria, the crustal shale values of PTE published by Turekian and Wedepohl was used.<sup>40</sup>

Zhang and Liu<sup>41</sup> noted that EnF values between 0.5 and 1.5 indicate PTE is entirely from natural origins whereas EnF values greater than 1.5 suggests source/s are more likely to be anthropogenic. Five contamination categories have been proposed to indicate the degree of pollution: no or minimal enrichment (EnF < 2); moderate enrichment (EnF = 2–5); significant enrichment (EnF = 5–20); very high enrichment (EnF = 20–40); extremely high enrichment (EnF > 40).

## 2.5 Health risk assessment

The difficulty of obtaining standardised, representative indoor dust samples is well acknowledged in the literature, and dust metal concentrations (as an amount of metal per given quantity of dust;  $\text{mg kg}^{-1}$ ) are not the same measurements as dust metal loadings (quantity of dust on the area sampled;  $\mu\text{g m}^{-2}$ ).<sup>20,42</sup> The dust samples collected in this study were bulk, composite, floor sweepings.

Whilst dust metal loadings are important for assessing potential exposure to indoor dust, the resultant concentration data provides us with a useful indicator of the presence of different sources of metals across the Lagos schools investigated. This concentration data has also been shown to be of use for comparing indoor dust with outdoor sources such as street

dust and soil.<sup>43,44</sup> Where only metal concentration data exist (without loadings data), the approach commonly adopted by other studies is based on the health risk assessment method developed by United States Environmental Protection Agency.<sup>45–49</sup>

Human exposure to PTE can occur *via* three main routes; (a) inhalation of particulates through the nose and mouth (b) dermal absorption adhered to exposed skin (c) through ingestion of particles. For this study, we computed the carcinogenic and non-carcinogenic risks of PTE exposure to two age groups; children and adults.

The average daily dose values for the three main exposure routes were calculated as follows:

$$D_{\text{ing}} = C \times \left( \frac{\text{IngR} \times \text{EF} \times \text{ED}}{\text{BW} \times \text{AT}} \right) \times 10^{-6} \quad (2)$$

$$D_{\text{inh}} = C \times \left( \frac{\text{InhR} \times \text{EF} \times \text{ED}}{\text{PEF} \times \text{BW} \times \text{AT}} \right) \quad (3)$$

$$D_{\text{dermal}} = C \times \left( \frac{\text{SL} \times \text{SA} \times \text{ABS} \times \text{EF} \times \text{ED}}{\text{BW} \times \text{AT}} \right) \times 10^{-6} \quad (4)$$

where  $D$  (mg per kg per day) is the dose *via* ingestion ( $D_{\text{ing}}$ ), inhalation ( $D_{\text{inh}}$ ) and dermal contact ( $D_{\text{dermal}}$ ). For this study, ingestion rate (IngR) was modelled as 200 mg per day for children and 100 mg per day for adults;<sup>50</sup> an inhalation rate (InhR) of 7.6  $\text{m}^3$  for children and 20  $\text{m}^3$  for adults; an Exposure Frequency (EF) of 180 days per year (in Nigeria, an average of 12 weeks per term (@ 3 terms per year, 5 days a week = 36 weeks/180 days), which equates to a typical annual term time for both primary and secondary school pupils/students; Exposure Duration (ED) of 6 years for children and 24 years for adult (children in Nigeria compulsorily spend 6 years each for both primary and secondary school respectively and a teacher spends maximum of 35 years in active service or retires before 60 years old, therefore the ED value of 6 and 24 years was chosen as a conservative and average realistic figure for both children and teachers). Averaging time (AT) is estimated differently for non-carcinogens and carcinogens. The carcinogenic intake is a cumulative dose over a lifetime (*e.g.* 70 years) whilst the non-carcinogenic intake is averaged over the period of exposure; AT for non-carcinogens is ED  $\times$  365 days and for carcinogens 55.2 years  $\times$  365 days (WHO, 2018 data for life expectancy in Nigeria).<sup>51</sup> Skin surface area (SA) is 2800  $\text{cm}^2$  for children and 3300  $\text{cm}^2$  for adults. Skin adherence (SL) factor is 0.2  $\text{mg cm}^{-2}$  for children and 0.7  $\text{mg cm}^{-2}$  for adults; dermal absorption factor (ABS) (0.03 for arsenic and 0.001 for Cd; no values for other PTE, therefore 1.0 was used). Particle emission factor (PEF) is  $1.36 \times 10^9 \text{ m}^3 \text{ kg}^{-1}$ .<sup>10,49</sup>

$C$  (exposure point concentration,  $\text{mg kg}^{-1}$ ) in eqn (2)–(4) is an estimate of reasonable maximum exposure, but in this current study is replaced with bioaccessible concentration values ( $\text{mg kg}^{-1}$ ) which represents the fraction that is available for absorption in the gastro intestinal tract of humans. For non-carcinogenic risk estimation each PTE average daily dose was divided with corresponding reference dose (RfD) to yield a hazard quotient (HQ) as shown in eqn (5). The hazard index





(HI) was estimated to be equal to the sum of the HQs calculated. The HI index evaluated the overall potential for non-carcinogenic risks to man from the three exposure routes considered (eqn (6)). If the HI value is greater than 1, there is a probability of non-carcinogenic risks occurrence in a lifetime. Carcinogenic risk is the probability of an individual developing any type of cancer during a lifetime exposure. Risk was evaluated by multiplying the average daily dose values against oral slope factor (SF) (eqn (7)).

$$HQ = \frac{(D) \text{ average daily dose}}{(RfD) \text{ reference dose}} \quad (5)$$

$$HI = \sum HQ_i \quad (6)$$

$$CR = D \times SF \quad (7)$$

Only As, Cr, Ni and Pb were considered since oral slope factors are only available for these elements. Calculated cancer risks above  $1 \times 10^{-4}$  were considered unacceptable and risks below  $1 \times 10^{-6}$  were considered not to have the capability of triggering any health effect, while values between  $10^{-6}$  to  $10^{-4}$  were considered to be within acceptable limits.<sup>23</sup>

## 2.6 Statistical analyses

Microsoft office excel package 2016 was used for the computation of PTE concentration measurements and simple descriptive statistical analysis such as standard deviation, mean, and human health risk calculations. Factor analysis (FA) was carried out to predict the possible sources of each PTE studied. Factor analysis was conducted using the principal component extraction method thereby reducing the dimensionality of the data set and then describing the variability among observed correlated variables. The components of the FA were transformed using varimax rotation and Kaiser normalization after analysis.<sup>52</sup> Minitab software version 18.0 was used for the aforementioned exploratory and multivariate statistical analysis.

## 2.7 Public focused dissemination activity

As part of our strategy to communicate the potential hazards associated with indoor dust, a short, bespoke report has been prepared for each school, setting that school's dust results within the wider context of the indoor dust data available for Lagos (the reporting template is included in ESI†), as well as key strategies on how to reduce personal exposure. To further promote public awareness of indoor environmental quality issues, and wider access to these data, we worked with Apperception-Group, LLC, and colleagues at Indiana University-Purdue University, USA (Gabe Filippelli), and Macquarie University, Australia (Mark Taylor), to develop an online resource, accessible at <https://mapmyenvironment.com>.<sup>53</sup> Data visualisations are "double-jittered" to ensure individual sample locations are not specifically identified. Liu *et al.*<sup>54</sup> found that an interactive portal was important to the success of citizen science projects, especially one that evolves and provides users with access to dynamic information. The MapMy Environment site continuously updates to display global data on environmental quality indicators, alongside plain language information and videos to enhance the environmental health literacy of global citizen.

## 3. Results and discussion

### 3.1 Total dust metal concentrations

The summary statistics of PTE concentration in the indoor dusts (determined by ED-XRF) are presented in Table 3, with the full dataset presented in ESI Tables 2 and 3.† Apart from Pb, no guidance or regulatory limits exist for the concentration of these PTE in indoor dusts. To provide some context for the range of total PTE concentrations observed we have included selected soil guideline values (for soil in residential areas) for three countries that highlight the range of international values reported (Table 3). Indeed, the outdoor environment typically plays a significant role in the indoor environment and research by Oomen and Lijzen (2004)<sup>55</sup> and Van Holderbeke *et al.*

Table 3 Total concentrations of PTE in the indoor dusts (mg kg<sup>-1</sup>) determined by ED-XRF

	Al	As	Ba	Cd	Cr	Cu	Fe	Mn	Ni	Pb	U	V	Zn
Minimum	13 200	1.4	97.4	<LOD	39.8	9.6	10 000	106	10.5	14.9	2.2	15.9	66.9
Mean	32 000	3.3	254	0.5	130	28.1	24 500	368	20.9	47.4	6.1	52.4	208
Median	30 800	3.2	212	0.4	76.9	23.0	21 100	339	20.1	43.7	5.5	49.4	163
Max	60 700	4.8	1560	1.3	606	110	42 400	764	40.3	114	15.0	103	622
SD	10 300	0.96	233	0.29	114	17.6	8530	141	6.34	20.1	2.76	22.4	129
Canada SGV <sup>a</sup>		12		10	0.4	63			50	140			200
Norway SGV <sup>b</sup>		12		12.2	75	150			90	90			300
CLEA, UK <sup>c</sup>		40		150						310			
Earth's crust <sup>d</sup>	80 500	1.7	650	0.13	83	47	46 500	1000	58	16		90	83
Continental Shale <sup>e</sup>	80 000	13	580	0.3	45	45	47 200	850	68	20	3.7	130	95

<sup>a</sup> [https://www.esdat.net/environmentalstandards/canada/soil/rev\\_soil\\_summary\\_tbl\\_7.0\\_e.pdf](https://www.esdat.net/environmentalstandards/canada/soil/rev_soil_summary_tbl_7.0_e.pdf). <sup>b</sup> <https://www.miljodirektoratet.no/globalassets/publikasjoner/klif2/publikasjoner/andre/1691/ta1691.pdf>. <sup>c</sup> Contaminated Land: Applications in Real Environments (CLAIRE) consortium, SP1010 Final Project Report (Revision 2). Development of Category 4 Screening Levels for Assessment of Land Affected by Contamination. DEFRA R&P Project Report, 2014a. <sup>d</sup> V. Alekseenko, A. Alekseenko, The abundances of chemical elements in urban soils, *Journal of Geochemical Exploration*, 2014, volume 147, part B, pages 245–249. <sup>e</sup> <https://pubs.geoscienceworld.org/gsa/gsabulletin/article/72/2/175/5262/Distribution-of-the-Elements-in-Some-Major-Units>.



(2008),<sup>56</sup> amongst others, report that the contribution of exterior soil to interior house dust ranges from 8% to >80% for Pb. The USEPA's recent review of dust Pb hazard standards lowered the floor Pb loading threshold from 40 to 10  $\mu\text{g}$  per square foot.<sup>57</sup> This threshold of 10  $\mu\text{g}$  per square foot is a combination of dust mass and Pb concentration in the dust, so does not readily translate into a Pb concentration in bulk sampled vacuum dust. However, for comparison purposes, initial work by Pat Rasmussen (personal communication), based on extensive investigations on house dusts<sup>20,43,44</sup> suggests a threshold of 10  $\mu\text{g}$  per square foot approximately equates to a median Pb concentration of 800 ppm.

Aluminium, Ba and V concentrations were compared against the typical background soil concentration. None of these three PTE (Al, Ba and V) were elevated when compared to these typical soil concentrations/crustal abundances. This suggests, levels of Al, Ba and V may largely be from natural sources, as anthropogenic contributions are not pronounced considering comparisons with respective background values.

Cadmium measurements ranged from below the LOD (0.1  $\text{mg kg}^{-1}$ ) to 1.3  $\text{mg kg}^{-1}$ , with a mean concentration of 0.5  $\text{mg kg}^{-1}$ . None of the Cd measurements exceeded the Norwegian and Canadian soil guideline values (SGVs) but nearly all of the sample measurements were higher than the typical soil background value. The highest Cd measurement in a previous study of residential soils from Lagos and Ibadan (Nigeria) was 2.8  $\text{mg kg}^{-1}$ , with an average concentration of 0.94  $\text{mg kg}^{-1}$ ,<sup>58</sup> and indicated a similar range of Cd measurements to this current study.

Iron and Mn are major components of soil minerals and have been noted to be abundant elements in the earth crust.<sup>59</sup> Our Fe and Mn concentrations were within typical soil concentration ranges, with all measurements for both of these PTE falling below the earth's crust and continental shale values respectively. Our dust samples were similarly not contaminated with Cu and Ni, with concentrations below typical soil concentrations and the reported SGVs (Table 3).

Zinc, Pb and Cr displayed a rather different behaviour among the PTE. Nearly all of the Zn concentration measurements in the dust samples exceeded the background values

(99%) and about 33% were higher in value than Canadian SGV; some multiple times higher. Several samples did exhibit slightly elevated Cr and Pb concentrations when compared to the typical background concentrations (earth crust and continental shale values), but only Cr had dust samples which exceeded the reported SGVs.

To provide further context for the data, PTE in indoor house dusts from selected studies are reported (Table 4). Direct comparisons however are again somewhat limited as these studies focus on house dust, and differences in sample preparation and analysis also exist. However, our Lagos indoor school dust samples typically had PTE concentrations that fell within the mean/geomean data reported in these other studies (Table 4). Our observed As concentrations ranged between 1.4–4.8  $\text{mg kg}^{-1}$ ; comparable to concentrations measured in classrooms and offices from Ogun state, Nigeria by Olujimi *et al.*<sup>10</sup> and below representative countries soil guideline values (Table 3). Our Cr concentrations were also largely comparable with those data reported in Table 4. An average concentration of Cr in cement typically ranges from 179–257  $\text{mg kg}^{-1}$ .<sup>60</sup> The aging concrete (cement) floors within the classrooms may have contributed to Cr concentrations measured in the samples. Whilst the majority of PTE we investigated in this study did not indicate elevated concentrations when compared to our selected SGVs (with the exception of Zn and Cr), many were higher than those measured in other Nigerian cities. For instance, Cd, Cu, Fe, Mn and Pb measurements in Ilorin city,<sup>61</sup> Makurdi,<sup>62</sup> and Kano city<sup>63</sup> (other cities in Nigeria) were all lower than concentrations measured in this current study. Lagos is the commercial nerve centre of the country and over sixty percent of all industrial activities in Nigeria are domiciled in Lagos. The huge industrialization and urbanization in Lagos may have contributed to the higher PTE load in our indoor dusts in comparison to these other Nigerian cities. Indeed, in a recent study in the Niger delta area of Nigeria,<sup>64</sup> classroom dusts collected from schools along major roads with high traffic density showed elevated PTE concentrations. In keeping with this study, several of our sampled schools were located in close proximity to busy markets (*e.g.* Ojuwoye and Balogun markets in Mushin, and Lagos Island LGA), and so elevated PTE might be

Table 4 Previous studies on PTE concentrations in indoor dusts

	As	Cd	Cr	Cu	Mn	Ni	Pb	Zn
Ogun state, Nigeria <sup>10</sup>	0.96–5.33	82.9–8595	17.7–105	2.37–250	81.7–600	2.37–58.4	3.80–127	15.2–480
Niger-delta Nigeria <sup>45</sup>		0.00–22.8	0.00–1.12	1.33–153	0.31–2.60	0.00–34.6	0.06–1.63	5.00–149
Lagos, Nigeria <sup>24</sup>		0.01–8.28	3.45–17.4				0.06–82.5	
Xian, China <sup>1</sup>	6.0–38.3		88.9–751	33.2–164	398–824	20.5–104	87.5–593	184–1838
Sri Serdang, Malaysia <sup>49</sup>		0.4–2.65		6.44–86.5			0.17–68.3	
Canada <sup>20</sup>	0.1–153	0.5–223	0.5–2930	24–4880		17.3–2300	14.2–7800	144–6630
Port Harcourt, Nigeria <sup>65</sup>	0.07–1.53	0.53–3.52	19–45.2				10.3–25.4	
Kumasi, Ghana <sup>66</sup>		BDL	0.00–33.2			0.00–41.9		
Guangzhou, China <sup>67</sup>		3.30–155		355–2745		17.7–281	178–3394	331–3928
Istanbul, Turkey <sup>68</sup>		0.4–20	2.8–460	62–1800	8.0–1300	120–2600	3.0–300	210–2800
Japan <sup>69</sup>		0.18–5.62	14.8–285	66.8–7720	38.6–679		11.9–3730	188–5920
Stratoni, Greece <sup>70</sup>		3–28			71–8390		390–6920	
This study	1.4–4.8	0–1.3	39.8–606	9.6–110	106–764	10.5–40.3	14.9–114	66.9–622



expected in our indoor dust samples attributed to traffic sources. The high influx of vehicles, emissions from small and medium scale enterprises, alternative power supply (use of gasoline or diesel electricity generators) and artisan workshops around the vicinities of our sampled schools may all have contributed to the PTE load in the dust samples. Activities in the local vicinity will influence the PTE load in indoor dusts. Potentially toxic elements find their way into buildings through a range of mechanisms, such as backtracking of particles on people and animals, as airborne dust particles (such as through open windows), as well as from numerous indoor sources, such as building materials, furnishings and consumer products. Darus *et al.*<sup>71</sup> noted that indoor dust or particle load can be associated with street dust or soil and are transported *via* wind blow through building openings and by the movement of occupants in and out of the building. On average, the As, Cd, Cu, Ni, Pb and Zn contents obtained in this study on Lagos were lower, and in some instances multiple times lower, than ones reported for Xian, China;<sup>1</sup> Guangzhou, China;<sup>67</sup> Istanbul Turkey;<sup>68</sup> Japan,<sup>69</sup> Straton Greece;<sup>70</sup> (see Table 4). The higher PTE levels in indoor dust from these cities, compared to that of this study (Lagos), are likely connected to the long histories of industrialization and urbanization in these countries. In contrast, the industrialization of Lagos only started in the post-independence era (1960 to 70 s).<sup>72</sup> Of particular concern is the apparent build-up of PTE in the Lagos environment over a relatively short period of time, as observed when we compare our data (on samples from 2019) with those of a previous study across Lagos<sup>24</sup> from 2012 (although we acknowledge that these were on a different set of 20 public primary schools). Making such data publicly available can raise awareness of environmental issues and drive appropriate mitigation strategies.

### 3.2 Contamination assessment

To quantify levels of PTE enrichment in the indoor dusts investigated, enrichment factors were calculated (Fig. 1). The degree of enrichment was estimated using Fe as the reference element and its average continental shale value as the background level. In the current study, the mean EnF pollution trend for the PTE studied was in the order of Pb (4.3) > Zn (3.6) > U

(2.9) > Cr (2.2) > Cu (1.1) > Ba (0.84) > Mn (0.78) > V (0.7) > As (0.5) > Cd (0.3) > Ni (0.1) > Al (<0.001). Aluminium, As, Ba, Cu, Mn, Ni and V EnF values were below the limit of contamination of 1.5 suggesting no enrichment or minimal enrichment in a few cases. Exceptions included Ba with an EnF of 6.30 in sample A1. Ba and its compound have been noted to have a variety of uses including getters in electronic tubes, colorants in paints, rodenticides and X-ray contrast medium.<sup>73</sup> Although there were no obvious sources for this enrichment in the indoor dust from the Amowo-Odofin area of Lagos.

Moderate enrichment of Cu was found in samples A2 (3.31), A19 (1.51) and A39 (2.02). Anthropogenic Cu sources have been linked to car components, tyre abrasion, corrosion of galvanized metal parts in cars, lubricants, engine wear, brake dust and thrust bearings.<sup>74</sup> So, it is not surprising to see enrichment of Cu in a number of samples taken from our urban schools, it is perhaps more surprising that we do not see this moderate enrichment more widespread across the sample set.

The Pb, U and Zn EnF values all indicated moderate to significant enrichment, with the majority of the Cr samples (63%) also indicating either moderate or significant enrichment. Lead and Zn enrichment in indoor dusts have been linked to biomass burning, vehicular traffic and industrial activities<sup>18</sup> whilst releases from mill tailings, combustion of coal and other fuels, mining of phosphate ores, use of phosphate fertilizers and emissions from nuclear plants have variously been blamed for the anthropogenic loading of U in soils.<sup>75</sup> Although we can rule out a nuclear source, no nuclear power plants exist in Lagos, the combustion of coal and other fuels are likely a key contributor to the widespread enrichment of Pb, Zn and U in our indoor school dusts.

### 3.3 Source apportionment

A wide number of methods for determining source apportionment of total elemental concentrations data have been reported in the literature<sup>43,55</sup> and here we employ factor analysis (FA). Such source identification tools allow for the prediction or determination of the contribution of possible pollution sources to total amount of PTE. Results of varimax rotated factor loadings, communalities, Eigen values and percentage variance of PTE in the dust samples are presented in Table 5. Results showed PTE groupings in three components which accounted for 75% of the total data variation.

Factor 1 was dominated by Al, As, Fe, Mn, Ni, V and U accounting for 39% of the total variance. This suggests concentrations of these elements may be influenced by a common factor or source, which given the crustal abundance of many of these elements is likely to be the local lithology of the study area. Indeed, previous studies have observed this PTE grouping and their association with natural sources.<sup>18</sup>

The second component explains 24% of the total variance and was negatively loaded on Cd, Cu and Pb. The close association between these elements in urban soils and dusts have been observed and identified as 'typical urban metals'.<sup>76</sup> Indeed, these PTE are typically associated with releases from leaded gasoline, braking, engine wear and other traffic-related

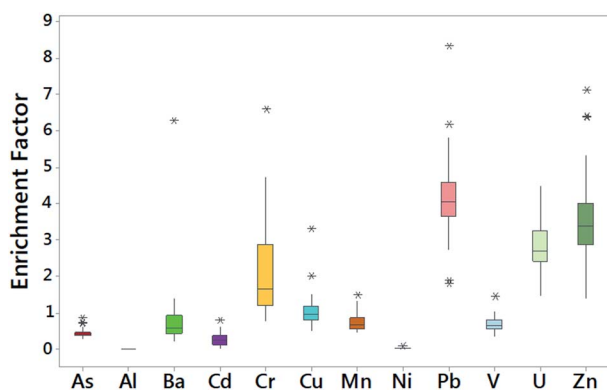


Fig. 1 Box plots of PTE enrichment factors in the indoor dusts samples.



**Table 5** Factor analysis (rotated factor loadings and communalities). Variables which contribute >0.700 to the respective factor loadings have been highlighted in bold

Variable	Factor 1	Factor 2	Factor 3	Communality
As	<b>0.748</b>	−0.283	−0.151	0.662
Al	<b>0.832</b>	0.068	0.161	0.722
Ba	0.080	−0.261	−0.544	0.370
Cd	−0.095	<b>−0.872</b>	−0.104	0.780
Cr	0.122	−0.271	<b>0.858</b>	0.825
Cu	0.252	<b>−0.832</b>	−0.009	0.756
Fe	<b>0.786</b>	−0.481	0.154	0.873
Mn	<b>0.737</b>	−0.012	−0.073	0.548
Ni	<b>0.855</b>	−0.327	0.199	0.877
Pb	0.468	<b>−0.793</b>	0.155	0.871
V	<b>0.739</b>	−0.093	0.630	0.952
U	<b>0.780</b>	−0.252	0.020	0.672
Zn	0.661	−0.657	0.097	0.877
Variance	5.0467	3.1490	1.5910	9.7868
% Var.	38.8	24.2	12.2	75.3

activities,<sup>68,69</sup> and have been reported to have their sources from pesticides, coal and biomass burning, sewage sludge, and industrial emissions.<sup>18</sup> Factor 2 thus suggests PTE originating predominantly from anthropogenic sources.

The third component explains 12% of the total variance and is primarily loaded on Cr, with a slightly lower loading on V (<0.700 but >0.600). As factor 1 has been associated with natural origins for the different PTE, and factor 2 as predominantly anthropogenic sources, factor 3 may suggest a more mixed source. Here, we should also note that Zn occurs, at moderate loadings on both factors 1 and 2 (0.661 on factor 1 and −0.657 on factor 2; Table 5). It is likely that Cr, V and Zn are indicating a mixed origin, rather than dominated by either natural or anthropogenic sources. Geochemical association of Cr with Fe and Mn hydroxides and silicate minerals may suggest a natural source, while urban sources of Cr include cigarette smoke and ash, abrasion of stainless steel, cooking pans, yellow/red/green chromium-bearing paints as well as chrome plated fixtures and fittings.<sup>77</sup> Urban anthropogenic releases of V are typically small in comparison to lithogenic releases, with oxides and hydroxides of Fe identified as a common primary source of V, but the increased presence/use of V has been reported within urban areas.<sup>78</sup> Zinc occurs naturally in the earth's crust with trace levels found in water, air, soil and food, but contamination

often occurs in relation to industrial processes, and industrial wastes. Compounds containing Zn are commonly found in association with anthropogenic activity, as Zn is used in wood preservatives, fabric dyes, in paints, and is typically associated with the construction and the automobile industry.<sup>18</sup>

Based on the findings of our multivariate analyses we allocated each PTE to one of three groups, identifying the dominant source for that PTE in indoor school dusts across Lagos:

A predominantly lithogenically derived group: Al, As, Fe, Mn, Ni, U.

A predominantly anthropogenically derived group: Cd, Cu and Pb.

A highly mixed origin group: Cr, V and Zn.

### 3.4 Oral bioaccessibility

A subset of indoor dust samples ( $n = 8$ ) were investigated for oral bioaccessibility (ESI Table 3†). The subset of selected samples for oral bioaccessibility measurements were chosen based on values obtained from enrichment factor calculations. Samples with high EnF values (higher concentration values) were selected for this purpose.

Bioaccessibility in this study is expressed as metal solubilized in the UBM stomach phase relative to the pseudototal concentration and expressed on percentage basis (see eqn (8)):

$$\% \text{ BAF} = \frac{\text{UBM extracted in stomach phase}}{\text{Pseudototal PTE concentration in indoor dust}} \quad (8)$$

Vanadium, Cd were below detection in the UBM extract and are not considered further. Of the remaining PTE investigated, Fe, Cr and Ni were the least bioaccessible PTE, with mean% BAF of 7%, 10% and 28% respectively. This indicated only a small fraction of these PTE would be available in the case of accidental ingestion.

Copper (36%), As (37%) and Al (41%) indicated higher mean bioaccessible concentrations in the indoor dusts, while Pb (48%), Ba (48%) and Mn (57%) were higher again, with mean bioaccessible concentrations of close to 50% or above. Zinc was the most bioaccessible PTE (mean of 88%), with bioaccessibility of around 100% in half of the indoor dust samples studied, suggesting that the simulated UBM gastric solution was as efficient in solubilizing Zn as the *aqua-regia*. Assuming 100%

**Table 6** Health risk assessment of PTE in indoor dusts

	HQ <sub>ing</sub>	HQ <sub>inh</sub>	HQ <sub>der</sub>	ΣHI children	HQ <sub>ing</sub>	HQ <sub>inh</sub>	HQ <sub>der</sub>	ΣHI adult	CR children	CR adult
Al	$1.70 \times 10^{-2}$	$7.24 \times 10^{-6}$	$5.76 \times 10^{-2}$	$7.50 \times 10^{-2}$	$2.20 \times 10^{-3}$	$2.17 \times 10^{-6}$	$3.64 \times 10^{-3}$	$5.85 \times 10^{-3}$		
As	$3.97 \times 10^{-2}$	$5.21 \times 10^{-10}$	$3.96 \times 10^{-3}$	$4.37 \times 10^{-2}$	$5.05 \times 10^{-3}$	$1.56 \times 10^{-10}$	$2.50 \times 10^{-4}$	$5.31 \times 10^{-3}$	$1.68 \times 10^{-6}$	$8.19 \times 10^{-7}$
Cd	$6.92 \times 10^{-2}$	$4.53 \times 10^{-9}$	$1.08 \times 10^{-1}$	$1.77 \times 10^{-1}$	$8.79 \times 10^{-3}$	$1.36 \times 10^{-9}$	$5.81 \times 10^{-4}$	$9.37 \times 10^{-3}$		
Cr	$8.22 \times 10^{-3}$	$1.61 \times 10^{-10}$	$1.09 \times 10^5$	$1.10 \times 10^3$	$1.04 \times 10^{-3}$	$4.84 \times 10^{-11}$	$6.90 \times 10^{-2}$	$7.00 \times 10^{-2}$	$4.57 \times 10^{-6}$	$1.42 \times 10^{-6}$
Cu	$2.36 \times 10^{-3}$		$7.86 \times 10^{-3}$	$1.02 \times 10^{-2}$	$3.01 \times 10^{-4}$		$4.97 \times 10^{-4}$	$7.97 \times 10^{-4}$		
Mn	$7.72 \times 10^{-3}$	$1.42 \times 10^{-8}$	$2.56 \times 10^{-2}$	$3.33 \times 10^{-2}$	$9.82 \times 10^{-4}$	$4.25 \times 10^{-9}$	$1.62 \times 10^{-3}$	$2.60 \times 10^{-3}$		
Ni	$1.71 \times 10^{-3}$	$2.09 \times 10^{-7}$	$1.42 \times 10^{-1}$	$1.43 \times 10^{-1}$	$2.18 \times 10^{-4}$	$6.26 \times 10^{-8}$	$8.98 \times 10^{-3}$	$9.19 \times 10^{-3}$	$5.86 \times 10^{-6}$	$1.82 \times 10^{-6}$
Pb	$4.68 \times 10^{-4}$		$1.33 \times 10^{-1}$	$1.33 \times 10^{-1}$	$5.95 \times 10^{-5}$		$8.42 \times 10^{-3}$	$8.48 \times 10^{-3}$	$4.42 \times 10^{-7}$	$1.37 \times 10^{-7}$
Zn	$5.65 \times 10^{-3}$		$1.87 \times 10^{-2}$	$2.44 \times 10^{-2}$	$7.19 \times 10^{-4}$		$1.18 \times 10^{-3}$	$1.90 \times 10^{-3}$		





bioavailability based on pseudototal or total concentrations would overestimate the risk of potential of these PTE.

Our observed trend in indoor dust bioaccessibility was thus, in ascending order,  $\text{Fe} < \text{Cr} < \text{Ni} < \text{Cu} < \text{As} < \text{Al} < \text{Ba} < \text{Pb} < \text{Mn} < \text{Zn}$ . Such a trend is largely comparable to the order for mean values of bioaccessibility of the elements observed in urban street dusts from a number of studies (*e.g.* street dust from Nanjing, China:  $\text{Cd} > \text{Zn} > \text{Pb} \approx \text{Mn} > \text{As} > \text{Cu} > \text{Ni} > \text{V} > \text{Cr} \approx \text{Fe}^{80}$ ). Indeed, mean Pb bioaccessibility is commonly reported to be around 50% across urban soils and street dusts (*e.g.* 47% bioaccessibility in urban street dusts Nanjing,<sup>79</sup> China, 53% in urban street dusts of Newcastle upon Tyne, UK<sup>80</sup>) with typically low Cr bioaccessibility reported (*e.g.* 4% in some urban soils<sup>81</sup>). Studies on indoor dusts have similarly reported high bioaccessibility of Zn, Mn and Al.<sup>19</sup> About 80 to 90% of Zn content were bioaccessible in household dusts of Estarreja, Portugal.<sup>18</sup>

Whilst we acknowledge the challenges in comparing bioaccessibility extraction test results due to the array of extraction protocols employed, differences in sample composition, environmental media (rural, urban, street dust, indoor dust, sediments *etc.*), geological profile of study areas *etc.* there do appear to be common trends in PTE bioaccessibility reported in the literature that are also evidenced in our school dusts.

### 3.5 Implications to human health

Results obtained for human health risk assessment (both carcinogenic and non-carcinogenic risks) for this current study are presented in Table 6. For this current study, total or pseudototal concentrations ( $\text{mg kg}^{-1}$ ) were replaced with bioaccessible concentrations since it has been established (at least for Pb, As and Cd) that the latter tends to be a more realistic assessment of what is actually bioavailable in the blood stream of humans; thereby USEPA equation models were modified to accommodate this observation. Other studies have reported approximately 30% in the reduction of aggregated hazard quotients and cancer risks results obtained for assessment of human exposure to dusts when total concentrations were replaced by bioaccessible concentrations.<sup>80</sup> Exposures to PTE in indoor dusts studied were within acceptable limits for both carcinogenic and non-carcinogenic risks for both age groups considered (children and adult). Ingestion appeared to be most significant route of exposure for both categories of age groups, followed by dermal contact and inhalation in that order. Among the PTE studied, Cr was the largest contributor for non-carcinogenic risks for both children and adult followed by Ni, Pb, Al, As, Zn, Mn and Cu in that descending order. The hazard

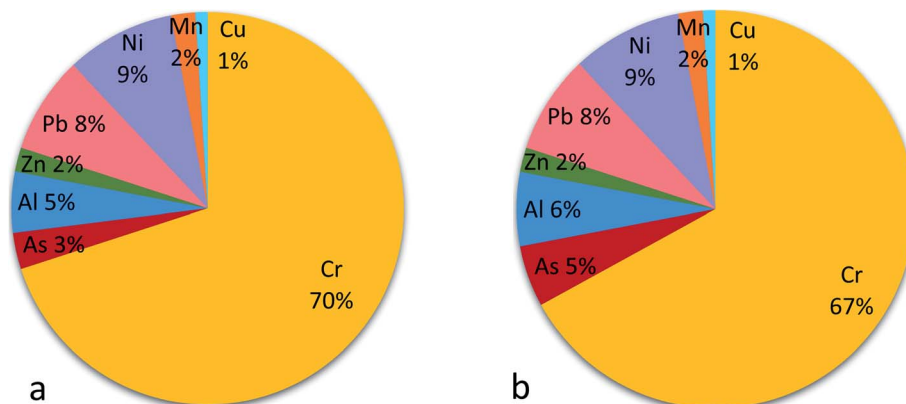


Fig. 2 Pie chart showing contributions of PTE to noncarcinogenic risks to children (a) and adult (b).

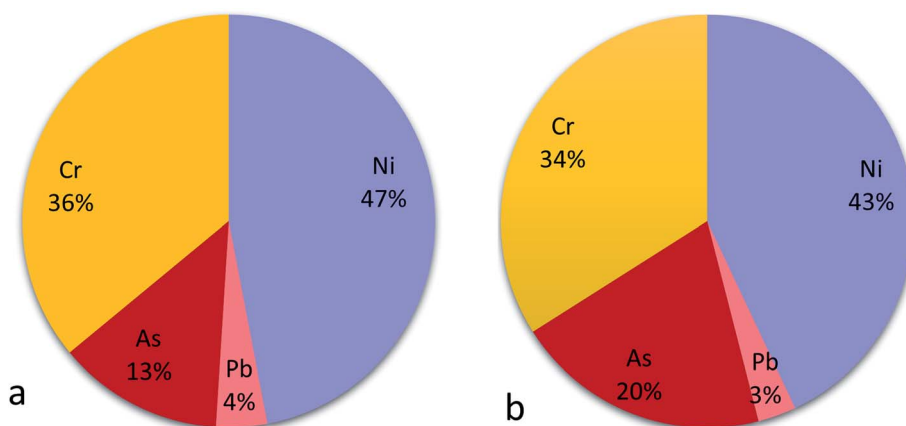


Fig. 3 Pie chart showing contributions of PTE to carcinogenic risks to children (a) and adult (b).



index (HI) which is summation of hazards quotients for ingestion, inhalation and dermal contact was less than 1 which suggests both children and adults are not at susceptible to non-carcinogenic risks. The only exception to this is HI index value for Cr in children which exceeded 1. Children prone to pica tendencies or geophagy may be at risk of Cr toxicity, however, speciation studies on the indoor dust samples may be required in future studies to ascertain the form in which Cr (either Cr III or Cr VI) exists in the dust samples. Results of this current study are consistent with other previous work on indoor dusts, urban soils, urban street dust.

Since not all PTE are a cancer risk, only As, Cr, Ni and Pb were additionally considered for carcinogenic risks assessment, using the slope factors for these PTE available in literature. Contributions of each PTE to possible carcinogenic and non-carcinogenic risks are presented in Fig. 2 and 3. Cr and Ni were the prominent among the PTE considered. Chromium was the largest contributor to non-cancer risks (70% in children and 67% in adult; Fig. 2), whilst Ni was the largest contributor to the total cancer risk (with 47% in children and 43% in adults; Fig. 3). The mean cancer risks were within acceptable limit of no probable carcinogenic effects ( $10^{-4}$  to  $10^{-6}$ ).

## 4. Conclusion

The average concentrations of PTE measured in our Lagos indoor dusts were higher than those reported from other Nigerian cities, including those of a previous study conducted in 2012,<sup>22</sup> suggesting PTE build up in the Lagos environment in recent years. Indeed, the pollution assessment tools we employed (EnF and factor analysis) suggest anthropogenic contributions dominate the enrichment of certain PTE in the indoor dusts (*i.e.* Cd, Cu, Pb, and to a lesser extent Zn). However, to set this within a broader context, then the determined PTE concentrations were lower than soil guideline values established across a range of countries (*e.g.* UK, Norway and Canada), and the calculated risks factors (cancer and non-cancer risks) were within acceptable limits for both children and adults. Our oral bioaccessibility determinations indicated that an assumption of 100% bioavailability, as commonly employed in risk assessment protocols based on pseudototal or total concentrations would overestimate the hazard potential of PTE in these indoor dusts. Whilst Zn was the most bioaccessible PTE (mean of 88%), all of the other PTE had oral bioaccessibility determinations significantly lower than 100% (ranging from 57% for Mn down to only 7% for Fe).

Human health risk assessment, for both children and adults using the bioaccessible fraction, showed values to be within acceptable risk levels, however children appeared to be more susceptible to non-carcinogenic and carcinogenic risks than adults since all hazard quotient values, HI and carcinogenic risks evaluation in children exceeded that of adults for all PTE.

To maintain the PTE content in Lagos indoor dusts as low as practicable it is important that environmental laws are strictly adhered to and enforced, in tandem with ongoing improvements in the regulatory environment. Increasing levels of environmental awareness potentially empowers citizens to take

actions to reduce exposure to indoor contaminants. An integral part of this project was public dissemination of our research findings to enhance the environmental health literacy of the staff and pupils within our participant schools.

As far as possible, 'plain english' was used to feedback results to the schools, with data made available on an interactive site at <https://mapmyenvironment.com>. Embedding citizen engagement is now acknowledged and accepted as an integral part of the research process as the potential exists for wider environmental benefits which result from a more environmentally aware society.

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## Conflicts of interest

There are no conflicts to declare.

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