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Transition metal-catalyzed alkene isomerization as an enabling technology in tandem, sequential and domino processes

Daniele Fiorito,  † Simone Scaringi† and Clément Mazet  *

One-pot reactions elaborated around transition metal-catalyzed isomerization of alkenes not only offer the inherent advantages of atom-, step- and redox-economy but also enable the preparation of value-added products that would be difficult to access by conventional methods. In this Review, we cover seminal and recent examples of tandem, sequential and domino processes, which incorporate the most attractive features of olefin isomerization.

1. Introduction

A countless number of catalytic reactions are based on the ability of transition metals to isomerize or migrate carbon-carbon double bonds.¹ The scope of applications of these processes spans across multiple areas such as food, fragrance, materials, medicinal chemistry, pharmaceutical and petrochemical. A strategy based on C=C bond isomerization provides the distinct advantages to (i) control the position and geometry of the unsaturation, (ii) install new functionalities at a different position of the starting alkene if it is

judiciously combined with a second chemical transformation. The latter reduces the number of steps, functional group manipulations and protection/deprotection events.²

The economic and environmental benefits associated with the development of multistep ‘one-pot’ catalytic reactions has raised strong interest in academia and industry.³ The field emulates from Nature’s ability to separate different reactive sites within an enzyme, or different enzymes within a cell to accomplish a series of orthogonal transformations in a given sequence without the need to isolate reactive intermediates.⁴ In this context, numerous research groups have associated olefin isomerization with (at least) one additional chemical transformation using transition metal catalysts, organocatalysts, enzymatic catalysts or any combination thereof.⁵ In such ventures, a

Organic Chemistry Department, University of Geneva, 30 quai Ernest Ansermet, Geneva 1211, Switzerland. E-mail: clement.mazet@unige.ch

† These authors contributed equally to this work.



Daniele Fiorito

mobility fellowship to pursue postdoctoral studies in the group of Prof. Varinder K. Aggarwal at the University of Bristol (UK).

Daniele Fiorito studied Chemistry at the University of Milano (Italy) where he received his MSc in 2015 working under the supervision of Prof. Cesare Gennari. He completed his PhD in chemistry at the University of Geneva (Switzerland) under the direction of Prof. Clément Mazet working on the development of several transition metal-catalyzed selective reactions (2019). He was recently awarded a Swiss National Science Foundation early postdoc



Simone Scaringi

doctoral studies under the direction of Prof. Clément Mazet. He is currently working on transition metal-catalyzed selective isomerizations of olefins for remote functionalizations.

Simone Scaringi was born in Milano (Italy) in 1995. He received his BSc (2017) and MSc (2019) from the University of Pavia (Italy) under the supervision of Prof. Stefano Protti. During his MSc studies, he spent 3 months in the group of Prof. Géraldine Masson at the Institut de Chimie des Substances Naturelles (Gif-sur-Yvette, France). In September 2019, he moved to the University of Geneva (Switzerland) to pursue his



major task for the synthetic chemist is to meet the requirement of temporal resolution and compatibility of all reaction components to a single set of experimental conditions. Importantly, the elaboration of multicatalytic systems is not achieved at the expense of molecular complexity; rather it provides access to novel chemical space by exploiting the reactivity of a transiently generated (sensitive) functional group.

This Review aims at providing the reader with an overview of the most significant advances in the field of 'one-pot' transformations articulated around transition metal-catalyzed alkene isomerization. To highlight the inherent benefits of a 'single-vessel' approach, our contribution is organized according to the nature of the function transiently generated after the first catalytic transformation. It is further sub-divided according to the nature of the coupled catalytic process at play. For sake of clarity, we follow the generally accepted classification set by Fogg and dos Santos in 2004 and discuss examples of tandem catalysis, sequential catalysis as well as domino reactions (Fig. 1).⁶ *Tandem catalysis* encompasses one-pot reactions, which occur *via* two consecutive mechanistically distinct and non-interfering processes.⁷ Within this category, one can further distinguish auto-tandem catalysis which employs a single catalyst from assisted-tandem catalysis in which the initial catalyst is modified by an external intervention at a given time point to initiate the second transformation.⁸ A final subset is orthogonal tandem catalysis, which operates with two catalysts present at the beginning of the reaction.⁹ By contrast, *sequential catalysis* is based on the use of several distinct precatalysts added sequentially, in the same flask, at different time points of the reaction in order to trigger a series of successive transformations. Finally, *domino reactions* based on alkene isomerization are one-pot processes in which the migrating alkene can be either trapped intermolecularly with an external reagent or engaged intramolecularly in a skeletal

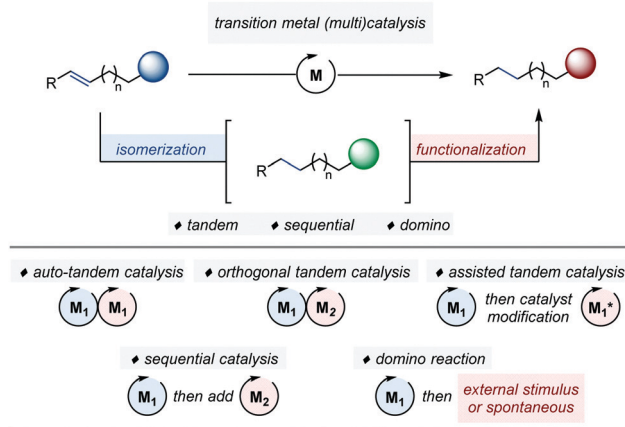


Fig. 1 Transition metal-catalyzed alkene isomerization in tandem, sequential and domino processes.

rearrangement.¹⁰ Independently of any categorization, the identification of an appropriate thermodynamic driving force for olefin isomerization is an important strategic aspect for the development of one-pot processes relying on this elementary transformation.¹

In the present digest, we arbitrarily decided to place our focus on systems which either rely on a single metal precursor or which combine two distinct transition metal catalysts.¹¹ Arguably two of the most common associations falling within this categorization are isomerization/metathesis and isomerization/hydroformylation processes. While the former has been summarized recently and will not be covered in the present conspectus,¹² an update of the most significant contributions published since it has been last reviewed in 2015 is presented in Section 2.1 for the latter.¹³ Of important note, a vast body of the literature covers catalytic isomerizations of alkenes terminated by a functionalization event which operate through a single catalytic cycle (*i.e.* isomerizing hydrofunctionalization). For this very reason, these systems cannot be defined as tandem, domino or sequential processes and, therefore, fall beyond the scope of this Review.^{11a,14,15}

2. Relocated alkenes as transient intermediates

The formation of thermodynamically more stable alkenes is a commonly employed tactic in olefin isomerization. This can be achieved either by formation of a more substituted double bond or by conjugating the migrating alkene with a neighboring unsaturated functional group (carbonyls, arenes, alkenes...). In the following sections, emphasis has been placed on one-pot processes where a transiently relocated alkene is engaged in a subsequent transformation by means of an auto-tandem, a domino or a sequential approach. The net result is the functionalization of a previously unactivated C(sp³), which does not rely on a directing group strategy.



Clément Mazet

Clément Mazet earned his MSc from the University of Strasbourg (France) where he was introduced to research in organometallic chemistry by the late John A Osborn. He received his PhD from the same institution in 2002 working with Lutz H. Gade. After postdoctoral stays with Andreas Pfaltz (University of Basel, Switzerland) and Eric N. Jacobsen (Harvard University, USA), he began his independent career at the University of Geneva (Switzerland). His interests include mechanistic and synthetic chemistry with emphasis on all aspects of selective catalysis. He received the Zasshikai Lectureship Award from the University of Tokyo (2012) and the Werner Prize from the Swiss Chemical Society (2013).



2.1 Tandem processes

As part of a study on Ru-catalyzed addition of aldehydes to conjugated 1,3-dienes to access β,γ -unsaturated ketones, the Ryu group disclosed an isolated example of an auto-tandem isomerization/addition process starting from a non-conjugated 1,5-diene (Fig. 2).¹⁶ Based on preliminary investigations, the authors proposed that a coordinatively unsaturated [Ru-H] *in situ* generated from [RuHCl(CO)(PPh₃)₃] (**Ru1**) is likely to chemoselectively react with the terminal alkene and re-conjugate the system *via* successive migratory insertion/ β -H elimination sequences (*i.e.* 2-C chain walking *via* 1,2-H shift).¹⁷ Subsequent formation of a Ru- π -allyl intermediate (**A2**) by reinsertion of the Ru-hydride across the conjugated diene **A1** is supported by NMR spectroscopy.¹⁸ This intermediate was proposed to react with aldehydes *via* a Zimmerman-Traxler 6-membered transition state that minimizes detrimental steric interactions. The resulting alkoxy-ruthenium **A3** was proposed to deliver the targeted carbonyl compound (**3**) upon β -H elimination.

A remote and seemingly unactivated C(sp³) can become the site for C-C bond formation by exploiting its transient electrophilic character. Goossen and co-workers developed an auto-tandem system in which a Rh catalyst isomerizes ω -unsaturated esters (**4**) into conjugated esters (**B**).¹⁹ The *in situ* generated Michael acceptors, in turn, undergo a Rh-catalyzed conjugate addition of C- or N-centered nucleophiles (Fig. 3).²⁰ The authors showed that the rhodium catalyst initially establishes a dynamic equilibrium of positional and geometrical alkene isomers and, once the C=C bond is conjugated with the ester group, it is continuously removed from the equilibrium by a rhodium-catalyzed 1,4-addition of either tetraarylborates, primary or secondary amines. The combination of [Rh(acac)(cod)] and a bulky bidentate phosphite (1.5 mol% each) is uniquely effective at catalyzing each individual event. In the case of N-nucleophiles, even though the aza-Michael addition proceeds thermally, the rhodium catalyst led to markedly improved yields.

The conversion of biobased fatty acids to hydrocarbons is a highly sought-after goal to provide alternative sources of raw materials to the petrochemical industry. In this context, the catalytic decarboxylation of mixtures of fatty acids has been identified as a particularly attractive strategy.²¹ In 2014, Doll and co-workers developed an auto-tandem isomerization/decarboxylation, which gives access to mixtures of alkenes from

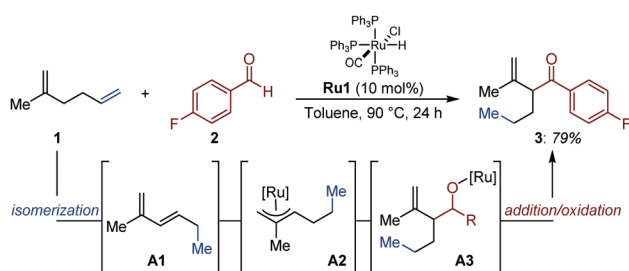


Fig. 2 Ru-Catalyzed auto-tandem isomerization/aldehyde addition.

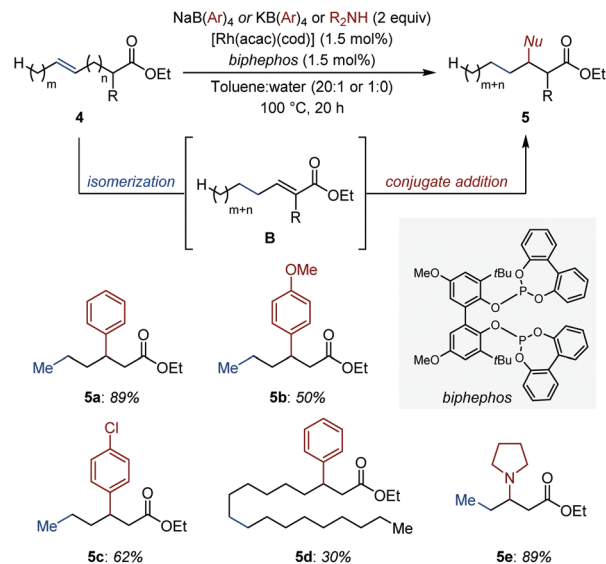


Fig. 3 Rh-Catalyzed auto-tandem isomerization/conjugate addition of ω -unsaturated esters.

unsaturated fatty acids using [Ru₃(CO)₁₂] as precatalyst and reached promising turn-over frequencies.²² The advantage of this system over existing protocols is that the catalyst dynamically isomerizes the alkene towards the terminal carbonyl moiety and subsequently decarboxylates the most reactive α,β - or β,γ -unsaturated acid derivatives (**C1** and **C2**). Moreover, it maintains efficiency at slightly lower temperatures and it offers a better mass-balance compared to reagent-assisted decarboxylation methods. This protocol is also scalable to multi-gram quantity (Fig. 4).

Formation of a styrenyl intermediate by conjugative isomerization of a terminal alkene to a distant arene is a robust strategy to relocate an alkene with high positional fidelity. Liu, Chen and co-workers developed an auto-tandem Pd-catalyzed isomerization/Wacker oxidation of terminal alkenes (**8**) using *tert*-butyl hydrogen peroxide as oxidant, which operates at room temperature within short reaction times (Fig. 5).²³

The isomerization catalyst – obtained by addition of 1 equivalent of TsOH to [Pd(OAc)₂] (5 mol%) – converts allyl(hetero)aryls into the thermodynamically more stable vinyl(hetero)arenes (**D1**) by migrating the alkene over 1C atom.²⁴ The *in situ* generated alkene was proposed to produce a peroxy-palladacycle

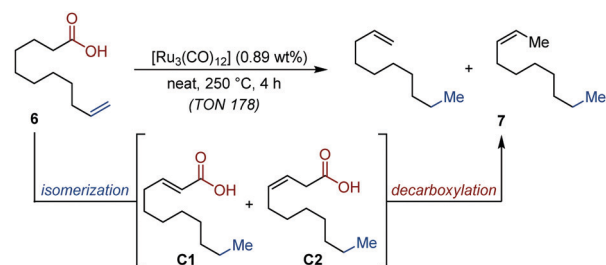


Fig. 4 Ru-Catalyzed auto-tandem isomerization/decarboxylation of 10-undecenoic acid into decenes.



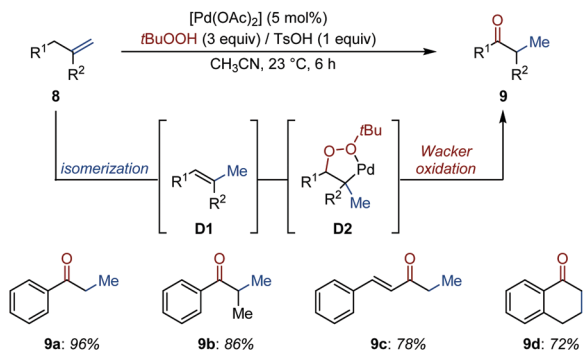


Fig. 5 Pd-Catalyzed auto-tandem isomerization/Wacker oxidation of terminal alkenes.

(D2), which degrades into the desired carbonyl compound *via* heterolytic O–O bond cleavage.²⁵ The method was applied to a dozen of examples and the products were usually isolated in high to very high yield.

Stoltz and co-workers showed that synthetically useful carboxylic acids are accessible by an assisted-tandem 1C isomerization/oxidation of terminal olefins using Grubbs second-generation (**Ru2**) as transition metal precursor (Fig. 6).²⁶ Their strategy relies on the formation of thermodynamically more stable internal olefins followed by oxidative cleavage of the repositioned unsaturation. Using Nishida's protocol (*vide infra*), the initial Ru-alkylidene complex **Ru2** is converted *in situ* into a [Ru–H] isomerization catalyst **Ru3**.²⁷ By a simple solvent switch once the C=C bond migration is complete, the metal hydride is converted *in situ* with NaIO₄ into an ill-defined ruthenium-oxo complex **Ru4** (presumably resembling [RuO₄])

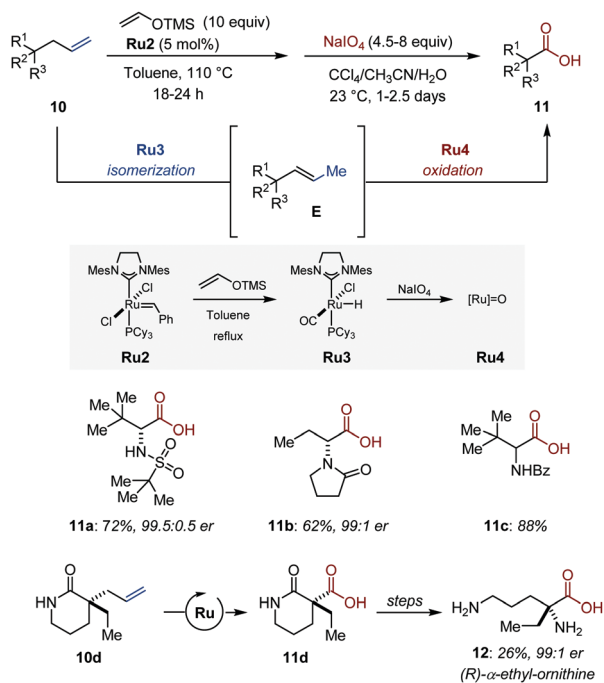


Fig. 6 Ru-Catalyzed assisted-tandem isomerization/oxidation of terminal olefins.

for the subsequent oxidation of the vinyl intermediate into carboxylic acids. The overall transformation is a formal two-carbon scission, which provides expedient access to enantio-enriched tri- and tetrasubstituted α -amino acids. Among other examples, the utility of the method was further established with the synthesis of (*R*)- α -ethyl-ornithine (**12**) in an elegant sequence that combines the Ru-catalyzed tandem isomerization/oxidation with a Curtius rearrangement followed by a simple hydrolysis.

Metal-catalyzed isomerization/hydroformylation processes have been the subject of numerous investigations.¹³ These systems typically rely on orthogonal-tandem catalysis, where an alkene isomerization catalyst produces statistical mixtures of internal and terminal olefins. A second catalyst reacts with the thermodynamically less stable terminal alkenes, and constantly shifts the equilibrium towards what appears to be the contra-thermodynamic product. Recently, Vorholt showed that excellent linear/branched selectivity (**14**, *rr* 10.1:1) could be achieved with methyl oleate **13** by associating the Pd-catalyst employed by Gooßen for orthogonal-tandem isomerization/metathesis with the well-established [(biphephos)Rh] hydroformylation catalyst (Fig. 7A).^{28,29} Key to the success of this tandem process is the ability of the Rh-catalyst to react exclusively with terminal olefins under the reaction conditions.³⁰

In 2018, the Beller and Pereira groups extended a [Ru/Rh] bimetallic system developed by Nozaki and co-workers five

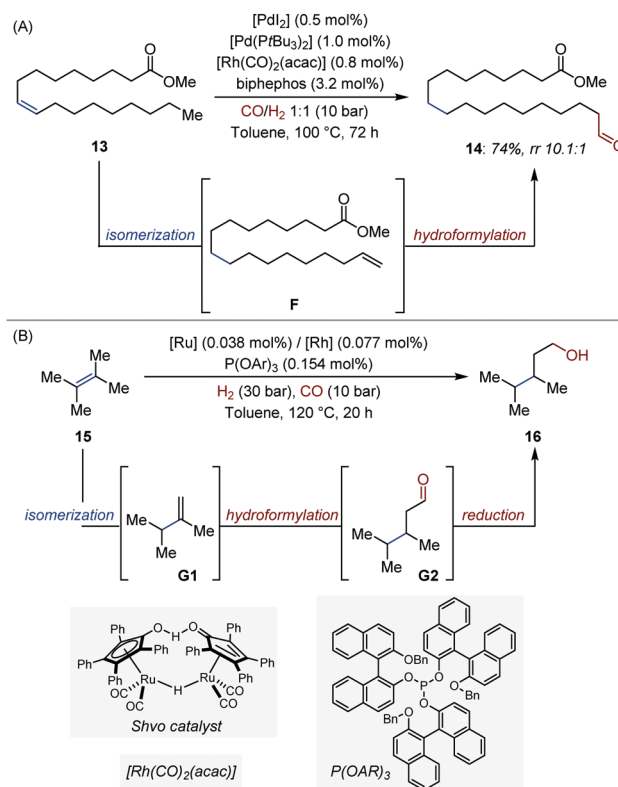


Fig. 7 (A) Orthogonal-tandem isomerization/hydroformylation of methyl oleate. (B) Orthogonal-tandem isomerization/hydroformylation of 2,3-dimethylbut-2-ene.



years earlier to tetrasubstituted olefins, a particularly challenging sub-class of alkenes (Fig. 7B).^{31,32} The system consists of a succession of three non-interfering catalytic cycles (isomerization, hydroformylation, reduction). Similar cooperative effects as in the original report were observed between the Rh and Ru catalysts. High yield, chemo- and regio-selectivity were obtained in many cases, including for stereochemically complex steroid derivatives.

2.2 Domino processes

Upon isomerization, an alkene can be strategically positioned to engage in a subsequent pericyclic transformation. In this context, diastereomerically pure annulenes (**18**) and cycloheptenones (**19**) were prepared through an expedient Ir-catalyzed isomerization/Cope rearrangement by the Marek group, starting from readily accessible dialkenyl cyclopropanes or alkenyl silylated cyclopropanols (**17**) (Fig. 8).³³ While the isomerization was found to proceed at room temperature, the pericyclic rearrangement was thermally triggered by heating up the system at 85 °C. These domino procedures appeared to be highly functional group tolerant and afforded densely functionalized bicyclic compounds in moderate to good yield. The authors proposed that the high level of diastereoselectivity obtained originated from excellent control of the double-bond geometry during Ir-catalyzed chain walking, a process sustainable over up to 10 carbon atoms (**18b**).³⁴

2.3 Sequential processes

A series of remote functionalizations based on sequential multimetallic catalytic reactions has been reported by Mazet and co-workers.³⁵ The reactions are triggered by hydrometalation of α,β -unsaturated esters using [Pd-H] or [Ru-H] isomerization catalysts and are driven by the formation of thermodynamically more stable (hetero)vinyl arenes (Fig. 9). Intermediates such as **I** were generated with high levels of stereoselectivity and were engaged in a subsequent functionalization reaction in the same reaction vessel. Several [Pd/Cu] catalytic systems were developed.

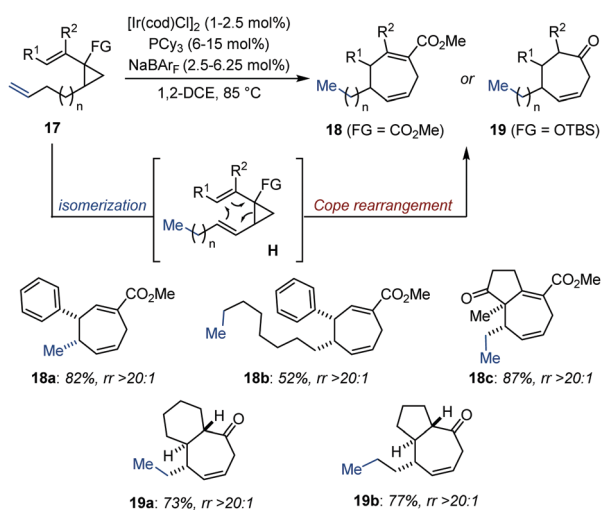


Fig. 8 Ir-Catalyzed domino olefin migration/Cope rearrangement.

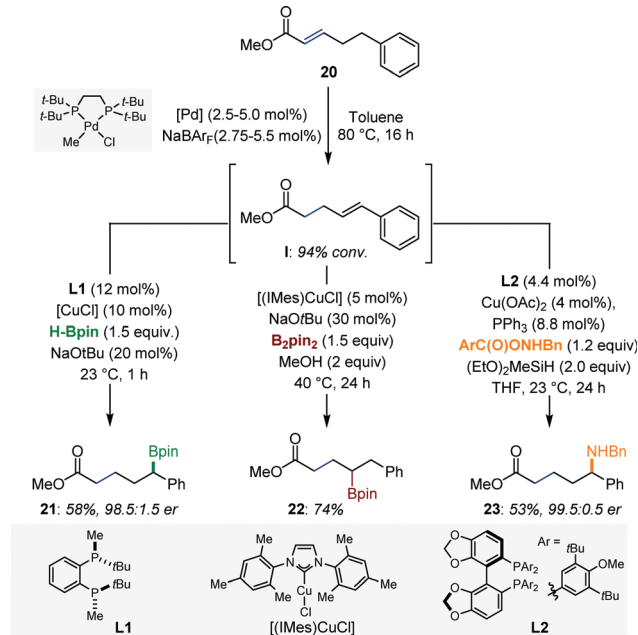


Fig. 9 Remote functionalization of α,β -unsaturated esters by sequential multimetallic catalysis.

Specifically, an isomerization/enantioselective α -borylation, an isomerization/ β -borylation and an isomerization/enantioselective α -amination provided access to multifunctional compounds with excellent levels of regio- and/or enantio-control. The authors showed that the combination of existing protocols is not necessarily trivial, in particular in the context of enantioselective catalysis.

Finally, difunctionalization can occur simultaneously at the benzylic and homobenzylic positions by combining the deconjugative isomerization with Sharpless asymmetric dihydroxylation (SAD) using either [Pd/Os] or [Ru/Os] couples. Starting from a simple α,β -unsaturated ester, the isomerization/dihydroxylation sequence followed by *in situ* lactonization delivered a naturally occurring γ -butyrolactone in enantiopure form.

3. Enol ethers as transient intermediates

Enol ethers are synthetically useful entities but their inherent reactivity poses several challenges for their preparation and isolation.³⁶ Catalytic approaches, which consist in generating enol ethers from stable precursors prior to engaging them *in situ* in a subsequent catalytic reaction, are therefore attractive.

3.1 Tandem processes

In 1998, Crivello and Rajaraman reported a Co-catalyzed auto-tandem isomerization/polymerization of allyl ethers (Fig. 10).³⁷ They found that in presence of a silane the precatalyst $[\text{Co}_2(\text{CO})_8]$ disproportionates into $[\text{HCo}(\text{CO})_4]$ and $[\text{R}_3\text{SiCo}(\text{CO})_4]$. Upon loss of carbon monoxide, the former readily produces $[\text{HCo}(\text{CO})_3]$, which catalyzes the isomerization of allyl ethers into a mixture of



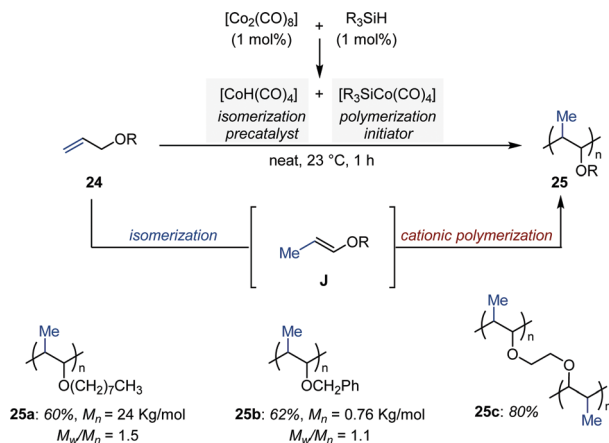


Fig. 10 Co-Catalyzed auto-tandem isomerization/polymerization of allyl ethers.

cis- and *trans*-1-propenyl ethers. These were subsequently polymerized by the silyl-cobalt complex $[R_3SiCo(CO)_4]$ at room temperature. Multifunctional allylic ethers were also polymerized by this method to produce cross-linked insoluble matrices. Finally, because polymerization was inhibited by Lewis bases, a cationic polymerization mechanism was proposed to be operating.

By exploiting the propensity of enol ethers to undergo intramolecular rearrangements when brought in close proximity of another reactive unsaturation, Tanaka and co-workers identified a cationic bisphosphine–rhodium complex ($[(dppf)Rh(cod)]BF_4$) for the isomerization/propargyl Claisen rearrangement of allyl propargyl ethers (26) to allenyl aldehydes (27) (Fig. 11).³⁸ The reaction proceeds *via* enol ethers **K** generated by a Rh-catalyzed 1-carbon olefin isomerization. The system

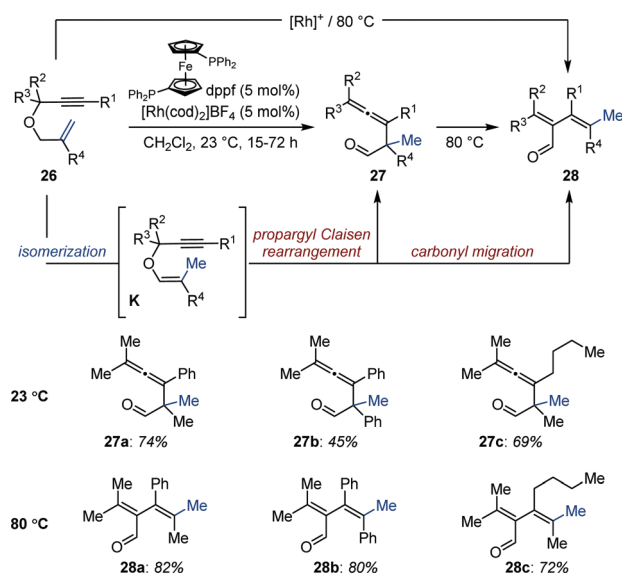


Fig. 11 Thermally-controlled Rh-catalyzed isomerization/propargyl Claisen rearrangement and isomerization/propargyl Claisen rearrangement/carbonyl migration reactions. Allenyl aldehydes were isolated as alcohol after reduction with sodium borohydride in methanol.

operates at room temperature and the products are isolated in good yield after reduction using sodium borohydride in methanol. The authors found that carbonyl migration occurred when the reactions were conducted at 80 °C to afford the corresponding dienals in high yield (28). However when primary and secondary propargyl ethers were investigated under these conditions the system afforded allenes (27) resulting from a propargyl-Claisen rearrangement. A sequence consisting in aldehyde C–H activation, hydorrhodation of the allene, carbonyl migration and reductive elimination was proposed to account for the formation of dienals 28. The intermediacy of enol ethers **K** and aldehydes 27 was confirmed by monitoring the reaction by 1H NMR spectroscopy.

As a direct continuation of this study and inspired by seminal work from Eilbracht,³⁹ Tanaka disclosed a Rh-catalyzed olefin isomerization/allyl Claisen rearrangement/intramolecular hydroacylation of diallyl ethers (29) (Fig. 12).⁴⁰ The authors propose that chemoselective isomerization of the 1,1-disubstituted alkene in 29 produces alkenyl ethers **L1**, which subsequently undergo allyl Claisen rearrangement. Rhodium-catalyzed intramolecular hydroacylation of the pendant allyl ether moiety in **L2** delivers substituted cyclopentanones (30). Of note, both **L1** and **L2** were observed by conducting the reaction at room temperature. Isotopically labelled substrates provided additional evidences in support of this mechanistic hypothesis. Overall, the catalytic system tolerates a variety of functional group and is compatible with heteroaryl-containing bis-allyl ethers. Cyclopentanones with two contiguous quaternary centers are accessible using this method as established with the synthesis of a naturally occurring terpenoid *rac*- α -cuparenone 30d.

3.2 Domino processes

In 1977, Reuter and Salomon showed that non-symmetrical bis-allyl ethers (31) were competent substrates for the Ru-catalyzed isomerization/Claisen rearrangement (ICR) to produce γ,δ -unsaturated aldehydes and ketones (33) (Fig. 13A).⁴¹ Metal-catalyzed isomerization of the less substituted alkene occurs

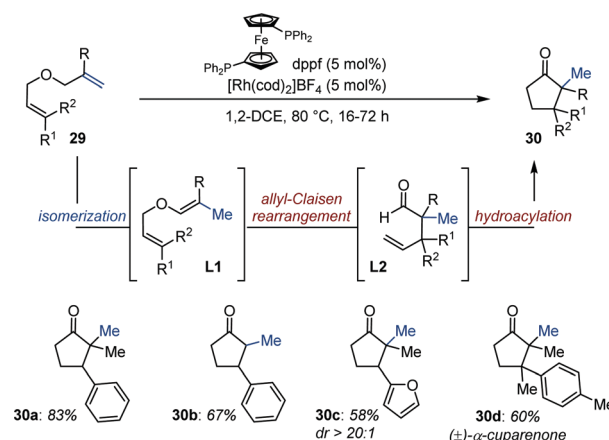


Fig. 12 Rh-Catalyzed auto-tandem isomerization/Claisen rearrangement/hydroacylation of bis-allyl ethers.



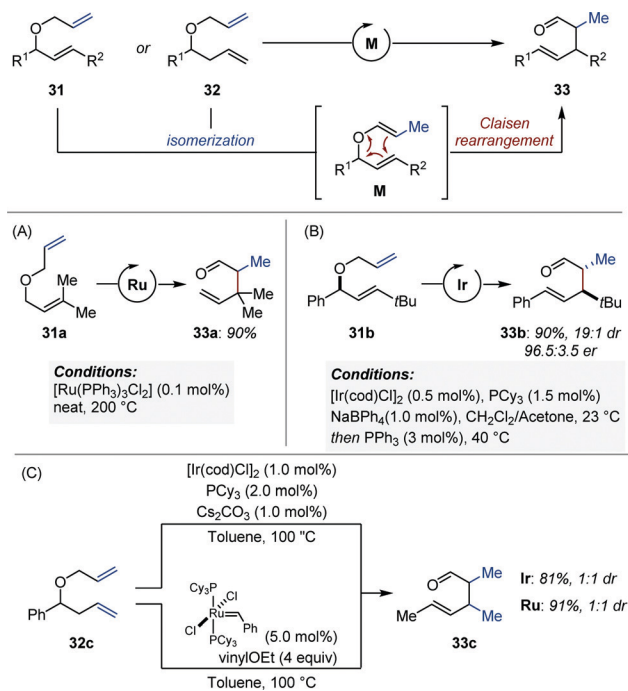


Fig. 13 Metal-catalyzed isomerization/Claisen rearrangement of non-symmetrical bis-allyl ethers.

preferentially and poises the C=C bond in a position favorable for a subsequent thermal Claisen rearrangement (M). The scope of this method was later extended by the Dixneuf group with the use of *in situ* generated [(NHC)Ru] complexes.⁴² Cadierno showed that similar reactions could be performed in refluxing water employing a Ru-arene precatalyst.⁴³ Between 2003 and 2007, Nelson developed a series of elegant and practical protocols using iridium catalysis and optically active bis-allyl ethers (31) to afford *syn*-2,3-dialkyl-4-pentalen derivatives (33) with high diastereoselectivity and enantiospecificity (Fig. 13B).⁴⁴ While the transition metal catalyzed isomerization occurs readily at room temperature, the [3,3] sigmatropic rearrangement requires temperatures that never exceed 40 °C. This contrasts markedly with the ruthenium-based catalytic systems for which temperatures ranging from 100 to 200 °C are necessary.^{41–43} Of important note, addition of triphenylphosphine (3 mol%) was hypothesized to attenuate the Lewis acid character of iridium in order to avoid epimerization of the aldehyde α -stereocenter. The origin of the high levels of diastereocontrol were attributed to the highly *E*-selective isomerization, which occurred exclusively at the less substituted allyl ether. The same group published a similar method for the construction of vicinal quaternary and tertiary stereocenters starting from 1,1-disubstituted allylic ethers, and later showed that *anti*-2,3-disubstituted-pentalenals were accessible by means of a simple Pd-catalyzed Claisen rearrangement protocol.^{44d,45}

The efficiency of the iridium-catalyzed ICR protocol developed by Nelson was recognized by the Trost and Chen groups, who employed it in the total syntheses of communiol A and (+)-antroquinol respectively.⁴⁶ Of note, in addition to ruthenium and iridium, Stork, and Echavarren reported

examples of domino isomerization/Claisen rearrangement using palladium catalysts.⁴⁷

Ishii and Schmidt independently achieved the domino metal-catalyzed isomerization/Claisen rearrangement of unsymmetrical allylic/homoallylic ethers (32) using iridium and ruthenium catalysts respectively (Fig. 13C).^{48,49} Although the products are identical to those obtained from bis-allyl ethers (*vide supra*), these reactions involve a double C=C bond migration prior to the thermal [3,3] sigmatropic rearrangement. The absence of diastereoselectivity in either case can be attributed to the lack of stereocontrol upon olefin isomerization.

3.3 Sequential processes

While tandem and domino reactions involving enol ethers occur preferentially intramolecularly, intermolecular transformations have been achieved by sequential multi-catalysis. Romano and Mazet identified conditions for the one-pot multi-metallic conversion of allyl methylethers (34) in alkenes (35) with high degree of stereocontrol (Fig. 14).⁵⁰ In the first step, a cationic iridium precatalyst is activated by molecular hydrogen and isomerizes 34 in vinyl methylethers (N) at room temperature. Two complementary [(NHC)Ni] catalysts were identified for the subsequent cross-coupling between the *in situ*-generated C–O electrophiles and a variety of Grignard reagents at elevated temperatures.⁵¹ The method is compatible with sensitive functional groups and a several olefinic substitution patterns. The newly generated alkenes were systematically obtained with high to very high levels of stereocontrol (*E/Z*: 4.5:1 to >25:1). Control experiments showed that the Ir-catalyzed process is stereoconvergent in nature and that the stereochemistry set during isomerization is preserved upon cross-coupling. Finally, the one-pot isomerization/cross-coupling sequence was conducted with a prochiral substrate and a chiral iridium precatalyst (**Ir2**), originally developed for the isomerization of allylic alcohols into aldehydes.⁵² The final olefinic products

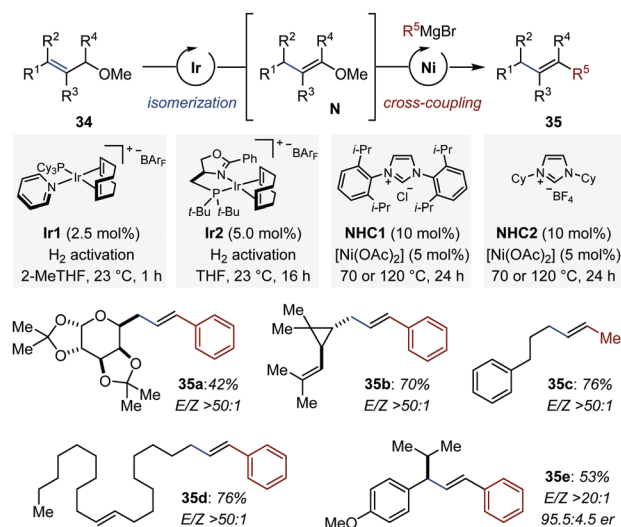


Fig. 14 One-pot sequential Ir-catalyzed isomerization/Ni-catalyzed cross coupling to convert allyl methylethers in substituted alkenes.



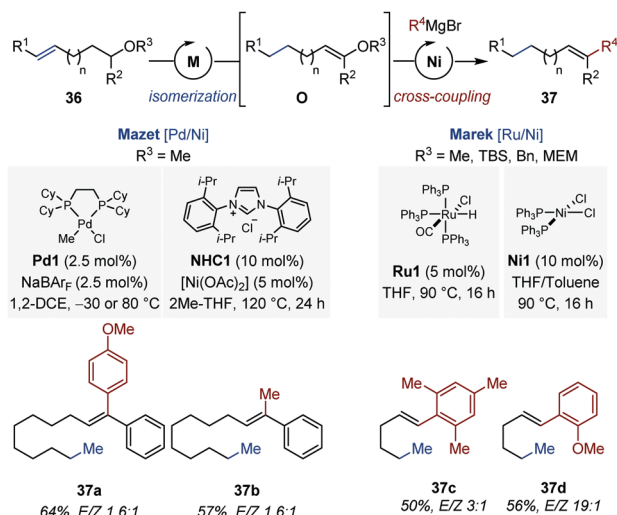


Fig. 15 One-pot [Pd/Ni]- and [Ru/Ni]-catalyzed remote isomerization/cross coupling sequences for the synthesis of (poly)substituted alkenes.

were isolated in moderate yield but with particularly high levels of stereo- and enantiocontrol (35e).⁵⁰

The inability of the iridium catalysts to displace C=C bonds over more than one carbon atom prompted the same group to evaluate alternative systems for the development of a related long-range isomerization/cross-coupling sequence (Fig. 15).

Thus, a [Pd/Ni] catalytic sequence was identified where the palladium catalysts enabled isomerization of the remote C=C bond in several alkenyl methyl ethers (36).^{50,53} Solvent exchange between the isomerization step and the cross-coupling reaction was necessary. The final di- and trisubstituted alkenes were obtained in good yield but with low level of stereocontrol (originating from the isomerization step). Concomitantly, the Marek group reported a complementary [Ru/Ni] catalytic combination for the remote isomerization/cross-coupling of alkenyl benzyl-, silyl-, methoxyethoxy-methyl ethers leading to 1,2-disubstituted alkenes (37).⁵⁴ While the isomerization was performed in refluxing THF, addition of toluene was found to be beneficial for the cross-coupling reaction (1:1 v/v). Only C(sp²) Grignard reagents were disclosed, and higher levels of stereocontrol were achieved in comparison with the [Pd/Ni] system (1:1 to 19:1).

Subsequently, the same group disclosed a related sequential catalytic isomerization/Suzuki cross-coupling strategy in which the vinyl boronic ester intermediates generated *via* isomerization act as a transmetalating agent rather than electrophilic coupling partner.⁵⁵ Overall, 1,2-disubstituted styrenes were prepared with similar efficiency and stereocontrol than with the sequential alkenyl ether isomerization/Kumada cross-coupling approach.

4. Enamines as transient intermediates

4.1 Tandem processes

Enamines are typically prepared by Schiff-base condensation between a carbonyl compound and a secondary amine. Nonetheless,

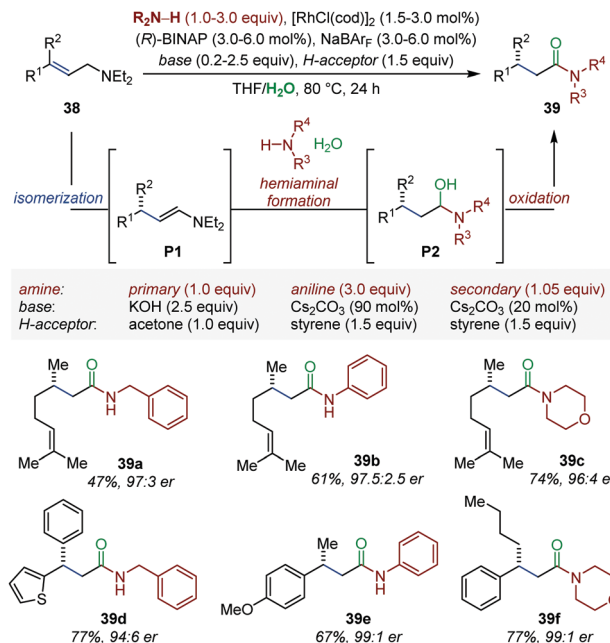


Fig. 16 Enantioselective auto-tandem Rh-catalyzed isomerization/oxidation of allylic amines to β -chiral amides.

they can also be accessed through transition metal-catalyzed isomerization of allylic amines. In this context, the Hull laboratory developed an elegant enantioselective auto-tandem process consisting of Rh-catalyzed isomerization of allyl amines (38), followed by enamine exchange and subsequent Rh-catalyzed oxidation to chiral β -branched amides (39) (Fig. 16).⁵⁶ While the first step takes inspiration from the Rh-catalyzed enantioselective isomerization optimized by the Takasago Company for the production of menthol,⁵⁷ the exchange process is based on the Rh-catalyzed conversion of allylic alcohols into amides under oxidative conditions identified in their own laboratory.⁵⁸ In the present system, under basic reaction conditions the *in situ* generated enamines (P1) is converted to a hemiaminal (P2) in presence of an external amine nucleophile and water. Rh-catalyzed oxidation of P2 into amides 39 is proposed to occur next. A sacrificial hydrogen acceptor is used regenerate the initial Rh(i) complex. Overall, this approach enables the installation of a stereocenter and an amide functionality in a single step under mild conditions from readily available starting material. Excellent diastereo- and enantioselectivity were obtained using a broad set of allylic amines and amine nucleophiles such as primary amines, secondary amines or even anilines. In the following years, a similar protocol was documented using primary, secondary or tertiary alcohols as external nucleophiles giving access to β -branched esters in excellent yield and enantioselectivity starting from simple allylic amines.⁵⁹ Finally, an assisted tandem Rh-catalyzed isomerization/reductive amination of allylic amines was also reported.⁶⁰ The rhodium isomerization catalyst was converted into a hydrogen transfer catalyst by addition of formic acid at the final stage of the tandem process. Overall, this method delivers enantioenriched β -branched amines.



4.2 Sequential processes

While in the auto-tandem approaches devised by the Hull laboratory, enamines are employed in intermolecular reactions, Arisawa and Nishida showed that sequential processes offer the possibility to exploit transient enamines for intramolecular heterocycle synthesis. In 2006, they established that a well-defined ruthenium hydride **Ru3** is quantitatively generated upon treatment of Grubbs second-generation catalyst **Ru2** with vinyloxytrimethylsilane (Fig. 6).²⁷ In their seminal study, an isomerization/metathesis sequence using both ruthenium complexes for the preparation of a handful of protected indoles was developed.

Later, Arisawa reported that commercially available ruthenium hydride **Ru1** could be associated with Hoveyda–Grubbs metathesis catalyst **Ru5** in a sequence consisting of a one-carbon isomerization/ring-closing metathesis/oxidation/1,3-dipolar addition using benzoquinone to deliver isoindolo[1,2-*a*]isoquinoline **41** (Fig. 17).⁶¹ This particular example highlights the versatility of their sequential isomerization/metathesis approach for heterocycle synthesis starting from simple *N*-allyl anilines.

5. Carbonyls as transient intermediates

The conversion of allylic alcohols into carbonyl compounds is a cornerstone of transition metal-catalyzed olefin isomerization.^{52,62} Not only is the process thermodynamically favorable, but rapid increase in molecular complexity can be achieved from readily available starting materials. Several research groups have also combined this isomerization in one-pot processes by capitalizing on the ability of the carbonyl intermediate either to undergo α -functionalization or to direct C–H activation. Examples where the carbonyl intermediates are engaged *in situ* in reductive or oxidative transformations have also been reported.

5.1 Tandem processes

Jun and co-workers reported a Rh-catalyzed auto-tandem isomerization/hydroacylation yielding non-symmetric aliphatic ketones (**43**) from simple allylic alcohols (**42**) and alkenes (Fig. 18).⁶³ In the first step, aldehydes **R1** are produced upon

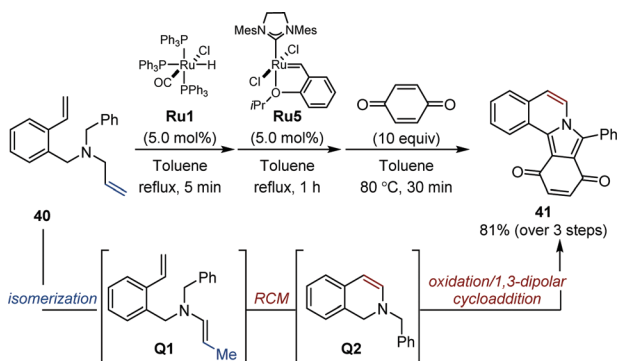


Fig. 17 Sequential Ru-catalyzed isomerization/ring-closing metathesis/oxidation/1,3-dipolar cycloaddition.

isomerization of **42**. Schiff-base condensation with 2-amino-4-picoline in presence of benzoic acid gives aldimine **R2** *in situ*. Subsequent chelation-assisted hydroiminoacylation of the appropriate alkene followed by hydrolysis of the resulting ketamine affords the targeted ketone. Noticeably, this strategy (i) prevents aldol condensation, (ii) reduces catalyst deactivation upon decarbonylation, two processes often observed in hydroacylation employing aliphatic aldehydes. The reaction is critically influenced by the alkene substitution pattern. For instance, α,β -unsaturated ketones were observed as by-products in the isomerization/hydroacylation of trisubstituted allylic alcohols.⁶⁴

Bartoszewicz and Martín-Matute documented a Ru-catalyzed auto-tandem isomerization/C–H activation starting from secondary allylic alcohols.⁶⁵ In a likely mechanistic scenario, the ketone intermediate (**S1**) generated upon isomerization directs the subsequent *ortho*-C–H activation (Fig. 19). Migratory insertion of the [Ru–H] (**S2**) across triethoxyvinylsilane, followed by reductive elimination affords **45**. While sodium formate acts as both a base and a reducing agent, reproducibility issues were circumvented by addition of a catalytic amount of a bulky, electron-rich trialkylphosphine. The amount of double C–H activation product varied between *ca.* 1:1 to 1:0 depending upon the structure of the allylic alcohol (**44**). One example using a homoallylic alcohol was provided, indicating olefin migration can be sustained over several carbon atoms (**45c**).

In the period 2007–2009, Cadierno, Gimeno and co-workers developed a series of Ru-catalyzed isomerization/transfer hydrogenation systems of primary and secondary allylic alcohols (Fig. 20).⁶⁶ The tandem process consists first in the isomerization of the allylic alcohol (**46**), followed by transfer hydrogenation of the resulting carbonyl (**T**). The system operates under basic conditions using iso-propanol as hydrogen donor and $[(\eta^6\text{-C}_6\text{Me}_6)\text{RuCl}(\mu\text{-Cl})_2]$ **Ru6** as precatalyst and is applicable to a broad array of substrates, including derivatives with a trisubstituted olefin. The products are obtained in quasi-quantitative yield within relatively short reaction times. The overall transformation offers an attractive alternative to alkene

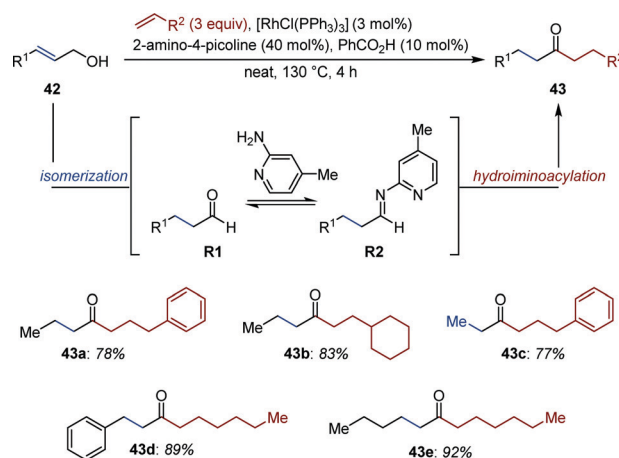


Fig. 18 Rh-Catalyzed auto-tandem allylic alcohol isomerization/chelation-assisted hydroacylation.



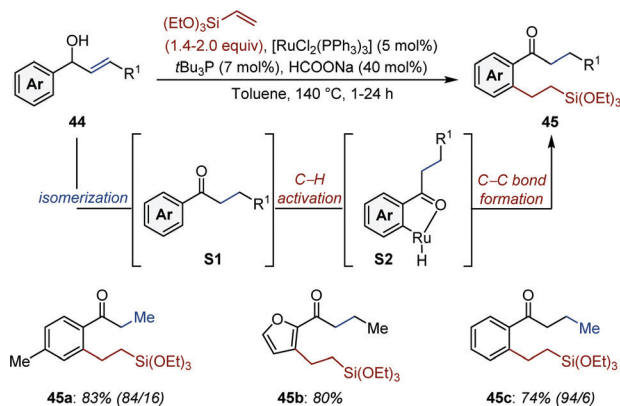


Fig. 19 Ru-Catalyzed auto-tandem isomerization/C-H activation of secondary allylic alcohols. In parenthesis is given the ratio of the products of mono- and bis-functionalization.

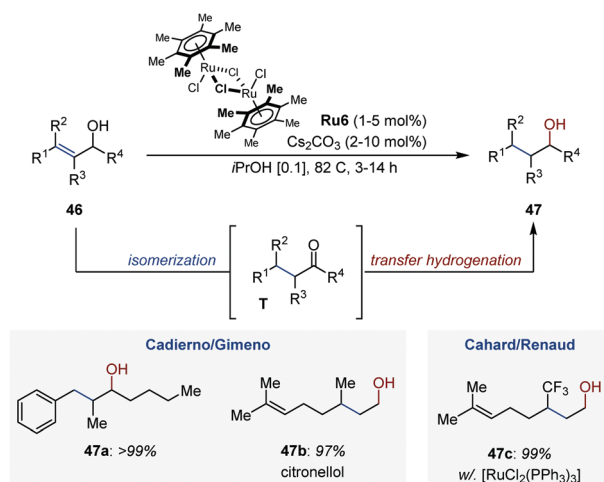


Fig. 20 Ru-Catalyzed auto-tandem isomerization/transfer hydrogenation of allylic alcohols.

hydrogenation because it circumvents the use of molecular hydrogen. It is also highly chemoselective as illustrated with the isomerization/transfer hydrogenation of geraniol leading exclusively to citronellol (**47b**). In the same contribution, the authors showed that the reaction could be carried out in water with similar efficiency when sodium formate was employed as hydrogen source. They later found that glycerol could act as both solvent and hydrogen donor.⁶⁷ In a remarkable study, Bizet *et al.* disclosed an analogous tandem isomerization/transfer hydrogenation of γ -trifluoromethylated allylic alcohols using $[(\text{PPh}_3)_3\text{RuCl}_2]$ as catalyst. The saturated alcohols were obtained in excellent yield and could be accessed from β -trifluoromethylated enones with similar performances.⁶⁸

In 2012, Sowa disclosed the first version of the Ru-catalyzed enantioselective isomerization/transfer hydrogenation of various allylic alcohols (Fig. 21).⁶⁹ The highest enantioselectivities were obtained for geraniol and nerol for which (*S*)-Tol-Binap and (*S,S*)-iPrDuphos were found to be the most effective chiral ligands. In analogy with related hydrogenation reactions, the

system was found to be stereospecific with (*S*)-Tol-Binap and stereoconvergent with (*S,S*)-iPrDuphos (Fig. 21).⁷⁰ On the basis of deuterium labelling experiments, the authors proposed that the isomerization proceeds *via* a 1,3-H shift mechanism and is the enantio-determining step of the tandem process. The same group and the Adolffson lab concurrently reported a related tandem isomerization/asymmetric transfer hydrogenation of secondary allylic alcohols using either diamine or α -amino acid hydroxyamide chiral ligands.⁷¹

Bruneau established that saturated amines could be accessed from allylic alcohols *via* Ru-catalyzed auto-tandem isomerization/reductive amination of primary allylic alcohols using arene-ruthenium complex **Ru7** at elevated temperatures (Fig. 22).⁷² Under the optimized reaction conditions, secondary amines were reacted in excellent yields, while primary anilines suffered from parasitic dialkylation. The reaction proceeds in presence of formic acid, which serves as reducing agent. Under the optimized reaction conditions, a broad range of primary and secondary amines was isolated in moderate to good yield. Of particular note, aniline **49e** could be generated in 71% yield from a mixture of five different monoterpenoids. Beyond the synthetic interest of working with complex mixtures, this result indicates that hydrogen (auto)-transfer occurs from different alcohols and aldehydes and converges toward the formation of a single product. A related process was developed shortly after for allylic and homoallylic alcohols using a light-activated Fe catalyst and polymethylhydrosiloxane as reducing agent.⁷³

As part of their program directed to the synthesis of valuable amides from simple starting materials, Wu and Hull disclosed a Rh-catalyzed tandem oxidative amidation of allylic alcohols using a wide array of primary and secondary amines with acetone or styrene as hydrogen acceptor.⁵⁸ The system served as a basis for the enantioselective tandem oxidative amidation of allylic amines presented in Section 4.1. Given the mechanistic similarities, it will not be discussed in further details in this section.

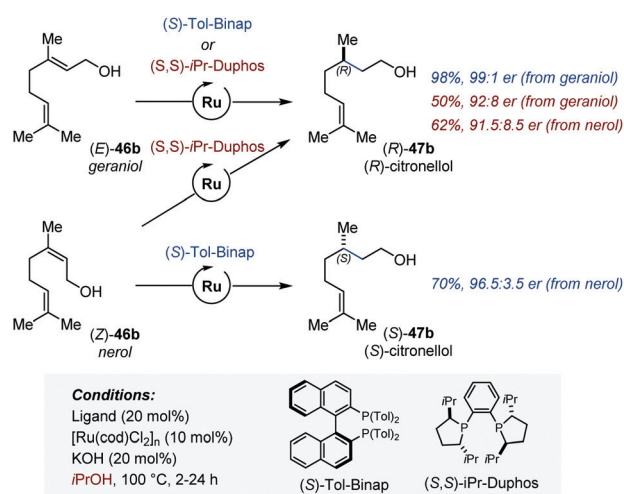


Fig. 21 Ru-Catalyzed enantioselective auto-tandem isomerization/transfer hydrogenation of allylic alcohols.



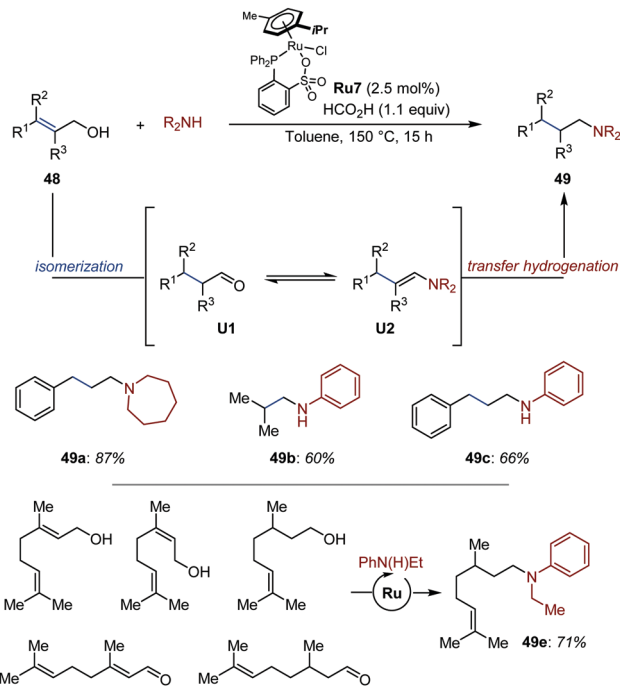


Fig. 22 Ru-Catalyzed auto-tandem reductive amination of allylic alcohols.

Based on a related strategy, Li and co-workers developed a Ru-catalyzed isomerization/acceptorless dehydrogenation of allylic alcohols for the synthesis of quinazolinones (**51**) using *o*-aminobenzamide nucleophiles (Fig. 23).⁷⁴ The system is general and requires only low loading of the commercially available [*p*-cymene]RuCl₂₂ complex under microwave irradiations (130 °C). A plausible mechanism has been proposed based on a series of control experiments. After initial Ru-catalyzed isomerization, condensation of *o*-aminobenzamide to the resulting aldehydes (**V1**) generates 2,3-dihydroquinazolinones (**V2**). Observation of gas evolution and NMR studies provided evidences for a Ru-catalyzed acceptorless dehydrogenation producing quinazolinones (**51**).

6. Transition metal enolates as transient intermediates

While the formation of main group enolates typically requires enolization of a carbonyl compound under basic conditions, transition metal enolates are usually prepared by exchange with preformed alkali enolates. Alternatively, transition metal enolates can be generated by oxidative addition into α -halo carbonyls, by decomposition of α -azo carbonyls or by conjugate reduction of α,β -unsaturated carbonyls.⁷⁵ A different approach consists in generating transient transition metal enolates by olefin isomerization before engaging them in a subsequent transformation. Examples of this strategy are discussed below.

6.1 Tandem processes

Building on the observation that alkenyl alcohols undergo isomerization/aldol dimerization under Ru-catalysis,⁷⁶ Ryu

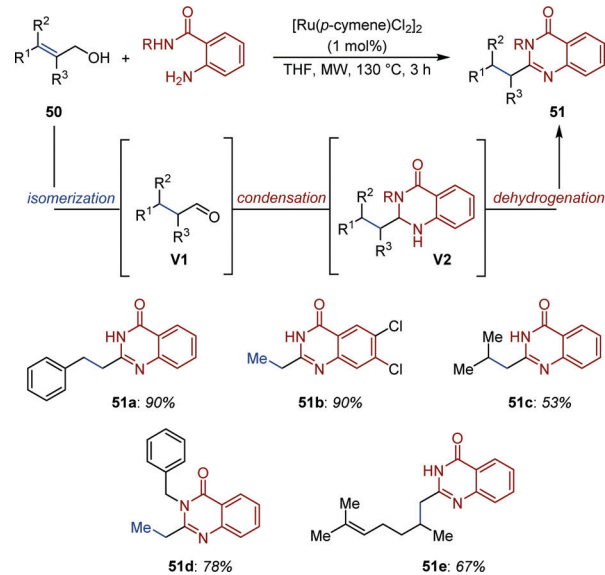


Fig. 23 Synthesis of quinazolinones via Ru-catalyzed auto-tandem redox isomerization/acceptorless dehydrogenation.

and co-workers followed a related C–C bond formation strategy for the construction of 2-alkyl substituted 1,3-diketones (**53**) starting from non-conjugated enones (**52**) and aldehydes (Fig. 24).⁷⁷ Initially, the ruthenium hydride **Ru1** re-conjugates the remote C=C bond with the carbonyl group (**W1**). Subsequent hydro-ruthenation generates Ru-enolate **W2**, which undergoes aldol condensation with the aldehyde to produce β -keto alkoxy-ruthenium intermediate **W3**, which upon β -H elimination delivers the product.¹⁶ This sequence is supported by preliminary mechanistic experiments performed with isotopically labelled aldehydes. The method was applied to a handful of aliphatic and aromatic aldehydes using hex-5-en-2-one as model substrate. The expected 1,3-diketones were isolated in good to excellent yield.

A joint effort from the Lautens and Taylor groups led to the development of an elegant Rh-catalyzed auto-tandem isomerization/allylation of unsymmetrical diallyl carbonates for the synthesis of aldehydes possessing a α -quaternary center (Fig. 25).⁷⁸ The protocol displays high chemoselectivity, as

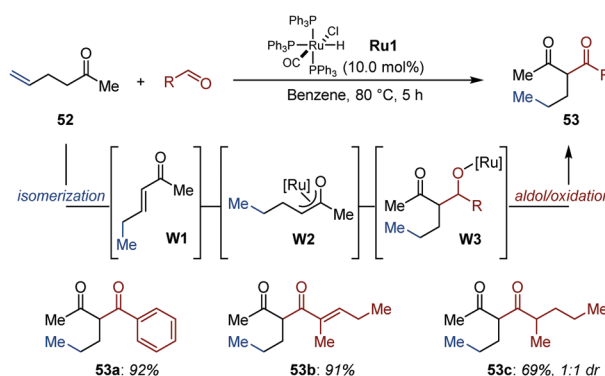


Fig. 24 Auto-tandem Ru-catalyzed isomerization/aldol addition/oxidation of hex-5-en-2-one with aldehydes.



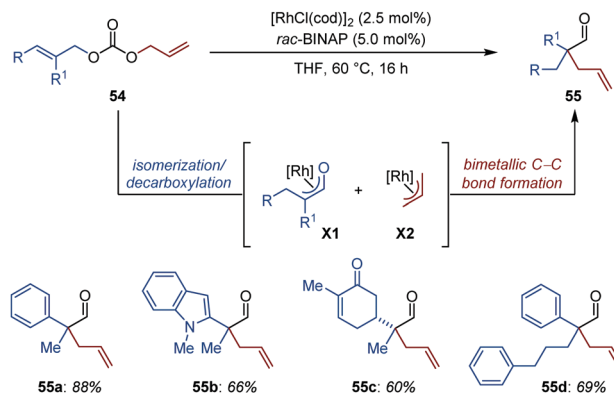


Fig. 25 Rh-Catalyzed auto-tandem isomerization/allylation of diallyl carbonates.

ketones, Michael acceptors and heteroaromatics are all well-tolerated. The reaction is also compatible with substrates possessing an enolizable position (55c). Competition and cross-over experiments together with reaction progress kinetic analysis (RPKA) shed light on the operating mechanism of this intriguing catalytic process. The concomitant formation of a nucleophilic Rh-enolate (X1) and an electrophilic Rh-allyl species (X2) is proposed to precede bimetallic C-C bond formation to produce 55. While formation of X2 is believed to proceed *via* an oxidative addition/decarboxylation sequence, X1 is presumably generated *via* isomerization of the most hindered C=C bond. Moreover, the reaction was found to be second order in rhodium, which is consistent with the C-C bond formation step operating through a rate-limiting bimetallic pathway. Of note, controlled experiments ruled out the possibility of a Claisen rearrangement being involved in product formation.

6.2 Domino processes

Pioneering studies from Bosnich and Motherwell established the potential of enolates *in situ* generated by metal-catalyzed isomerization of allylic alcohols to engage in subsequent C-C bond-forming domino aldol or Mannich processes.⁷⁹ Quite remarkably, with this approach formation of the metal enolate does not require the presence of a carbonyl group or of a carbonyl group flanked with a reactive α -functionality. The reader is invited to consult the exhaustive review written by Martín-Matute to appreciate the quantity and quality of the work accomplished in this area.⁸⁰

In recent years, the field has witnessed the elaboration of related domino reactions in which sources of electrophilic halides have been employed as quenching agents of the transient enolate (Fig. 26). The selective synthesis of monosubstituted α -iodo- (57a), α -bromo- (57b), α -chloro- (57c) and α -fluoroketones (57d) has been accomplished under mild and base-free reaction conditions using cyclopentadienyl-iridium complex Ir1 and aqueous solvent systems.⁸¹ Acyloins (57e) have been prepared following the same approach using an oxoammonium tetrafluoroborate salt derived from TEMPO.⁸² Finally, an ingenious variant based on an isomerization/umpolung functionalization strategy has been published by

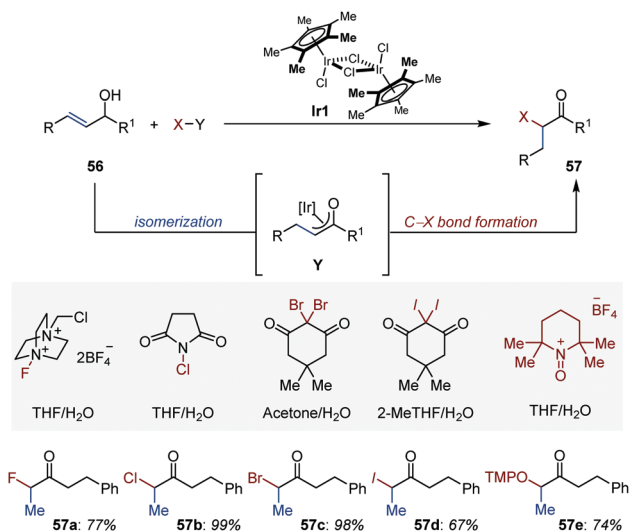


Fig. 26 Ir-Catalyzed domino isomerization/C-X and C-O bond formation processes.

the Martín-Matute group.⁸³ The combination of the appropriate iodine(III) reagent in methanol provided access to the targeted α -alkoxy ketones.

6.3 Sequential processes

The development of iterative enantioselective cross-aldol reactions between two distinct aldehydes is particularly challenging

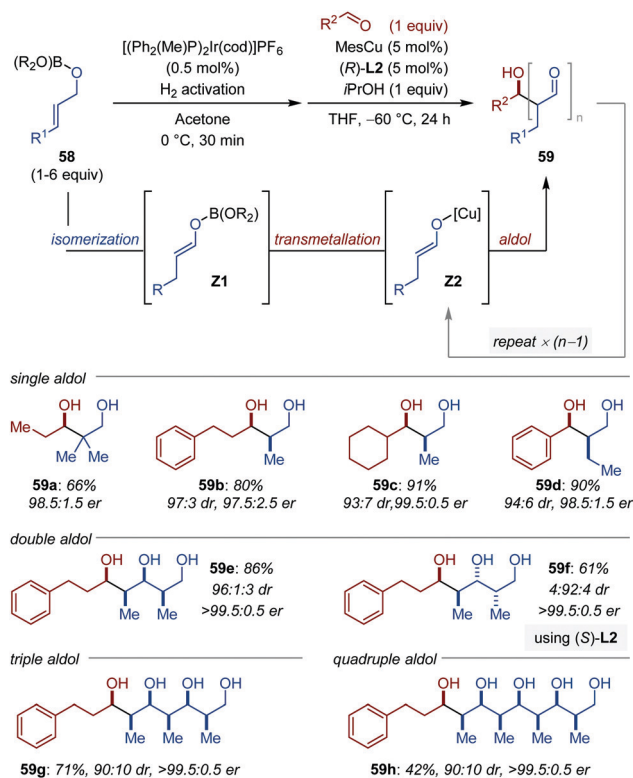


Fig. 27 Sequential [Ir/Cu]-catalyzed isomerization/iterative aldol reaction. L2 = DTBM-SEGPHOS (see Fig. 9). Products were isolated after reduction.



because it necessitates to differentiate their role as donor and acceptor in order to avoid synthesis of 1,3-polyols of random composition.⁸⁴ In a remarkable study, Lin *et al.* described the Cu-catalyzed aldol reaction between aldehydes (acceptor) and boron enolates (donor, (**Z1**)) *in situ* generated by Ir-catalyzed isomerization of the allylic oxyboronates (**58**) (Fig. 27).⁸⁵ The proof of concept was validated for the single aldol reaction. Under the optimized conditions, the iridium catalyst is activated with molecular hydrogen at 0 °C in acetone and the aldol reaction is accomplished with 5 mol% of a [(*R*)-(DTBM-SEGPHOS)CuMes] catalyst in the presence of 1 equivalent of isopropanol in THF at -60 °C. Aliphatic, aryl, and heteroaryl aldehydes all gave the cross-aldol products in moderate to excellent yields with consistently high diastereo- and enantioselectivity (after reduction (**59a-d**)). By adjustment of the nature and stoichiometry of the boronate, the alcohol and the nature of the amine additive, the aldol process can be repeated by using the aldol products as acceptor in the subsequent aldol addition. One-pot diastereo- and enantioselective double, triple- and quadruple aldol reactions were successfully achieved following this strategy (**59e-h**). Moreover, the stereodivergency of the process was demonstrated by simply using the chiral bisphosphine ligand of opposite handedness (**59e-f**).

7. Conclusions

One-pot processes based on transition metal-catalyzed isomerization of olefins enable chemists to convert commodity chemicals into more complex molecular architectures with maximum efficiency, be it in terms of atom-, step-, and redox-economy. Over the last decades, a large number of novel transformations have been developed using either tandem, domino or sequential approaches. Adjusting and controlling the reactivity of a distal position by catalytic migration of an alkene offers new perspectives for retrosynthetic analysis. For instance, remote and apparently inert sp³ hybridized carbon atoms can now be functionalized by judicious design of isomerization/functionalization processes. In the same vein, ethers and tertiary amines can be converted into far more reactive entities upon transition metal-catalyzed olefin isomerization (*i.e.* enol ethers and enamines) and subsequently engaged in one-pot transformations that would not be conceivable with traditional approaches. Electrophilic carbonyls or nucleophilic enolates can also be generated by alkene isomerization and used *in situ* for the functionalization of previously inaccessible carbon atoms. In addition to providing access to novel reactivity patterns, recent reports have highlighted the possibility to exert exquisite relative and absolute stereocontrol in complex systems based on remote functionalization strategies or iterative sequences. In spite of these remarkable achievements, there is certainly much ingenious one-pot catalytic reactions that can be developed and novel reactivity that can be unveiled.⁸⁶ For instance, the latest reports on zirconium-catalyzed olefin isomerization augurs well for the development of tandem approaches exploiting early transition metal catalysis.⁸⁷ Moreover, the intermediacy of alkyl-boron or

silicon compounds, generated by well-established isomerizing hydrofunctionalization of alkenes, could open new synthetic opportunities, if coupled with a sequential catalytic functionalization of the C-B and C-Si bond.⁸⁸ If one looks beyond the field of organic synthesis, extension of tandem or sequential catalysis to chain-walking polymerization processes represent an exciting venue for polymer chemistry.⁸⁹ We hope this Review will gather interest and inspire the design of novel one-pot processes based on transition metal-catalyzed olefin isomerization.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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Notes and references

- (a) E. Larionov, H. Li and C. Mazet, *Chem. Commun.*, 2014, **50**, 9816–9826; (b) A. Vasseur, J. Bruffaerts and I. Marek, *Nat. Chem.*, 2016, **8**, 209–219; (c) H. Sommer, F. Juliá-Hernández, R. Martin and I. Marek, *ACS Cent. Sci.*, 2018, **4**, 153–165; (d) T. Kochi, S. Kanno and F. Kakiuchi, *Tetrahedron Lett.*, 2019, **60**, 150938.
- (a) B. M. Trost, *Science*, 1991, **254**, 1471–1477; (b) B. M. Trost, *Angew. Chem., Int. Ed. Engl.*, 1995, **34**, 259–281; (c) P. A. Wender, V. A. Verma, T. J. Paxton and T. H. Pillow, *Acc. Chem. Res.*, 2008, **41**, 40–49; (d) N. Z. Burns, P. S. Baran and R. W. Hoffmann, *Angew. Chem., Int. Ed.*, 2009, **48**, 2854–2867; (e) J. J. Molloy, T. Morack and R. Gilmour, *Angew. Chem., Int. Ed.*, 2019, **58**, 13654–13664.
- Y. Hayashi, *Chem. Sci.*, 2016, **7**, 866–880.
- (a) S. F. Mayer, W. Kroutil and K. Faber, *Chem. Soc. Rev.*, 2001, **30**, 332–339; (b) I. Oroz-Guinea and E. García-Junceda, *Curr. Opin. Chem. Biol.*, 2013, **17**, 236–249.
- (a) O. J. Pamies and E. Bäckvall, *Chem. Rev.*, 2003, **103**, 3247–3262; (b) Z. Shao and H. Zhang, *Chem. Soc. Rev.*, 2009, **38**, 2745–2755; (c) M. Rueping, R. M. Koenigs and I. Atodiresei, *Chem. – Eur. J.*, 2010, **16**, 9350–9365; (d) L. M. Ambrosini and T. H. Lambert, *ChemCatChem*, 2010, **2**, 1373–1380; (e) A. Galván, F. J. Fañanás and F. Rodríguez, *Eur. J. Inorg. Chem.*, 2016, 1306–1313.
- D. E. Fogg and E. N. Dos Santos, *Coord. Chem. Rev.*, 2004, **248**, 2365–2379.
- (a) J. M. Lee, Y. Na, H. Han and S. Chang, *Chem. Soc. Rev.*, 2004, **33**, 302–312; (b) J. C. Wasilke, S. J. Obrey, R. T. Baker and G. C. Bazan, *Chem. Rev.*, 2005, **105**, 1001–1020; (c) C. Robert and C. M. Thomas, *Chem. Soc. Rev.*, 2013, **42**, 9392–9402; (d) A. Behr, A. J. Vorholt, K. A. Ostrowski and T. Seidensticker, *Green Chem.*, 2014, **16**, 982–1006; (e) H. Pellissier, *Adv. Synth. Catal.*, 2020, **362**, 2289–2325.



- 8 (a) N. Shindoh, Y. Takemoto and K. Takasu, *Chem. – Eur. J.*, 2009, **15**, 12168–12179; (b) J. E. Camp, *Eur. J. Org. Chem.*, 2017, 425–433.
- 9 (a) T. L. Lohr and T. J. Marks, *Nat. Chem.*, 2015, **7**, 477–482; (b) M. M. Lorion, K. Maindan, A. R. Kapdiand and L. Ackermann, *Chem. Soc. Rev.*, 2017, **46**, 7399–7420.
- 10 (a) L. F. Tietze, *Chem. Rev.*, 1996, **96**, 115–136; (b) C. Grondal, M. Jeanty and D. Enders, *Nat. Chem.*, 2010, **2**, 167–178; (c) S. A. Snyder, Applications of Domino Transformations in Organic Synthesis, *Science of Synthesis*, Georg Thieme Verlag, Stuttgart, 2016, vol. 1–2; (d) H. Pellissier, *Adv. Synth. Catal.*, 2016, **358**, 2194–2259; (e) H. Pellissier, *Adv. Synth. Catal.*, 2019, **361**, 1733–1755.
- 11 For reviews on tandem and sequential processes that combine transition metal-catalyzed alkene isomerization and organic catalysis, see: (a) P. Kathe and I. Fleischer, *ChemCatChem*, 2019, **11**, 3343–3354; (b) P. S. Wang, D. F. Chen and L. Z. Gong, *Top. Curr. Chem.*, 2020, 378.
- 12 J. Pollini, W. M. Pankau and L. J. Gooßen, *Chem. – Eur. J.*, 2019, **25**, 7416–7425.
- 13 (a) M. Vilches-Herrera, L. Domke and A. Börner, *ACS Catal.*, 2014, **4**, 1706–1724; (b) V. Goldbach, P. Roesle and S. Mecking, *ACS Catal.*, 2015, **5**, 5951–5972.
- 14 For reviews, see: (a) V. Goldbach, P. Roesle and S. Mecking, *ACS Catal.*, 2015, **5**, 5951–5972; (b) D. Janssen-Müller, B. Sahoo, S. Z. Sun and R. Martin, *Isr. J. Chem.*, 2020, **60**, 195–206. For selected examples, see: (c) J. V. Obligation and P. J. Chirik, *J. Am. Chem. Soc.*, 2013, **135**, 19107–19110; (d) P. Roesle, L. Caporaso, M. Schmitte, V. Goldbach, L. Cavallo and S. Mecking, *J. Am. Chem. Soc.*, 2014, **136**, 16871–16881; (e) F. Juliá-Hernández, T. Moragas, J. Cornella and R. Martin, *Nature*, 2017, **545**, 84–88; (f) F. Chen, K. Chen, Y. Zhang, Y. He, Y.-M. Wang and S. Zhu, *J. Am. Chem. Soc.*, 2017, **139**, 13929–13935; (g) X. Chen, Z. Cheng, J. Guo and Z. Lu, *Nat. Commun.*, 2018, **9**, 3939; (h) A. J. Borah and Z. Shi, *J. Am. Chem. Soc.*, 2018, **140**, 6062–6066; (i) R. Sang, P. Kucmierczyk, K. Dong, R. Franke, H. Neumann, R. Jackstell and M. Beller, *J. Am. Chem. Soc.*, 2018, **140**, 5217–5223; (j) Y. Liu, K. Dong, M. Beller and S. Mecking, *ACS Catal.*, 2018, **8**, 9232–9237; (k) T. Kochi, K. Ichinose, M. Shigekane, T. Hamasaki and F. Kakiuchi, *Angew. Chem., Int. Ed.*, 2019, **58**, 5261–5265; (l) P. M. Kathe and I. Fleischer, *Org. Lett.*, 2019, **21**, 2213–2217; (m) J. Yang, J. Liu, H. Neumann, R. Franke, R. Jackstell and M. Beller, *Science*, 2019, **366**, 1514–1517; (n) J. Li, S. Qu and W. Zhao, *Angew. Chem., Int. Ed.*, 2020, **59**, 2360–2364.
- 15 Contributions for which a tandem catalytic process has been invoked but the mechanistic scenario not fully ascertained have been excluded from our selection. See: (a) W.-C. Lee, C.-H. Wang, Y.-H. Lin, W.-C. Shih and T.-G. Ong, *Org. Lett.*, 2013, **15**, 5358–5361; (b) T. Yamakawa and N. Yoshikai, *Chem. – Asian J.*, 2014, **9**, 1242–1246; (c) Y. Ebe, M. Onoda, T. Nishimura and H. Yorimitsu, *Angew. Chem., Int. Ed.*, 2017, **56**, 5607–5611; (d) H. S. Li, G. Guo, R. Z. Zhang and F. Li, *Org. Lett.*, 2018, **20**, 5040–5043.
- 16 S. Omura, T. Fukuyama, J. Horiguchi, Y. Murakami and I. Ryu, *J. Am. Chem. Soc.*, 2008, **130**, 14094–14095.
- 17 H. Wakamatsu, M. Nishida, N. Adachi and M. Mori, *J. Org. Chem.*, 2000, **65**, 3966–3970.
- 18 F. Shibahara, J. F. Bower and M. J. Krische, *J. Am. Chem. Soc.*, 2008, **130**, 6338–6339.
- 19 (a) D. M. Ohlmann, L. J. Gooßen and M. Dierker, *Chem. – Eur. J.*, 2011, **17**, 9508–9519; (b) L. J. Gooßen, D. M. Ohlmann and M. Dierker, *US Pat.*, 8877958 B2, 2014.
- 20 T. Hayashi and K. Yamasaki, *Chem. Rev.*, 2003, **103**, 2829–2844.
- 21 X. Zhang, F. Jordan and M. Szostak, *Org. Chem. Front.*, 2018, **5**, 2515–2521.
- 22 R. E. Murray, E. L. Walter and K. M. Doll, *ACS Catal.*, 2014, **4**, 3517–3520.
- 23 J. Zhao, L. Liu, S. Xiang, Q. Liu and H. Chen, *Org. Biomol. Chem.*, 2015, **13**, 5613–5616.
- 24 M. Hassam, A. Taher, G. E. Arnott, I. R. Green and W. A. L. van Otterlo, *Chem. Rev.*, 2015, **115**, 5462–5569.
- 25 B. W. Michel, L. D. Steffens and M. S. Sigman, *J. Am. Chem. Soc.*, 2011, **133**, 8317–8325.
- 26 M. Liniger, Y. Liu and B. M. Stoltz, *J. Am. Chem. Soc.*, 2017, **139**, 13944–13949.
- 27 M. Arisawa, Y. Terada, K. Takahashi, M. Nakagawa and A. Nishida, *J. Org. Chem.*, 2006, **71**, 4255–4261.
- 28 T. Gaide, J. Bianga, K. Schlipköter, A. Behr and A. J. Vorholt, *ACS Catal.*, 2017, **7**, 4163–4171.
- 29 (a) D. M. Ohlmann, N. Tschauder, J.-P. Stockis, K. Gooßen, M. Dierker and L. J. Gooßen, *J. Am. Chem. Soc.*, 2012, **134**, 13716–13729; (b) S. De, N. Sivendran, B. Maity, N. Pirkkl, D. Koley and L. J. Gooßen, *ACS Catal.*, 2020, **10**, 4517–4533.
- 30 (a) E. Billig, A. G. Abatjoglou and D. R. Bryant, *US Pat.*, 4668651, 1987 and 4769498, 1988; (b) G. D. Cuny and S. L. Buchwald, *J. Am. Chem. Soc.*, 1993, **115**, 2066–2068; (c) A. van Rooy, K. Goubitz, J. Fraanje, P. C. J. Kamer, P. W. N. M. van Leeuwen, N. Veldman and A. L. Spek, *Organometallics*, 1996, **15**, 835–847.
- 31 F. M. S. Rodrigues, P. K. Kucmierczyk, M. Pineiro, R. Jackstell, R. Franke, M. M. Pereira and M. Beller, *ChemSusChem*, 2018, **11**, 2310–2314.
- 32 Y. Yuki, K. Takahashi, Y. Tanaka and K. Nozaki, *J. Am. Chem. Soc.*, 2013, **135**, 17393–17400.
- 33 H. Sommer, T. Weissbrod and I. Marek, *ACS Catal.*, 2019, **9**, 2400–2406.
- 34 I. Massad and I. Marek, *ACS Catal.*, 2020, **10**, 5793–5804.
- 35 C. Romano, D. Fiorito and C. Mazet, *J. Am. Chem. Soc.*, 2019, **141**, 16983–16990.
- 36 V. Milata, S. Rádl and S. Voltrová, in *32.5.3 Product Subclass 3: Enol Ethers. Science of Synthesis*, ed. J. Mulzer, Thieme, Stuttgart, 2008, vol. 32, p. 589.
- 37 J. V. Crivello and S. K. Rajaraman, *Macromol. Symp.*, 1998, **132**, 37–43.
- 38 K. Tanaka, E. Okazaki and Y. Shibata, *J. Am. Chem. Soc.*, 2009, **131**, 10822–10823.
- 39 P. Eilbracht, A. Gersmeir, D. Lennartz and T. Huber, *Synthesis*, 1995, 330–334.



- 40 R. Okamoto and K. Tanaka, *Org. Lett.*, 2013, **15**, 2112–2115.
- 41 J. M. Reuter and R. G. Salomon, *J. Org. Chem.*, 1977, **42**, 3360–3364.
- 42 (a) J. Le Nôtre, L. Brissieux, D. Sémeril, C. Bruneau and P. H. Dixneuf, *Chem. Commun.*, 2002, 1772–1773; (b) J. Le Nôtre, R. Touzani, O. Lavastre, C. Bruneau and P. H. Dixneuf, *Adv. Synth. Catal.*, 2005, **347**, 783–791.
- 43 B. Lastra-Barreira, J. Francos, P. Crochet and V. Cadierno, *Organometallics*, 2018, **37**, 3465–3474.
- 44 (a) S. G. Nelson, C. J. Bungard and K. Wang, *J. Am. Chem. Soc.*, 2003, **125**, 13000–13001; (b) B. D. Stevens and S. G. Nelson, *J. Org. Chem.*, 2005, **70**, 4375–4379; (c) S. G. Nelson and K. Wang, *J. Am. Chem. Soc.*, 2006, **128**, 4232–4233; (d) K. Wang, C. J. Bungard and S. G. Nelson, *Org. Lett.*, 2007, **9**, 2325–2328.
- 45 N. J. Kerrigan, C. J. Bungard and S. G. Nelson, *Tetrahedron*, 2008, **64**, 6863–6869.
- 46 (a) B. M. Trost and T. Zhang, *Org. Lett.*, 2006, **8**, 6007–6010; (b) R. S. Sulake and C. Chen, *Org. Lett.*, 2015, **17**, 1138–1141.
- 47 (a) G. Stork and K. S. Atwal, *Tetrahedron Lett.*, 1982, **23**, 2073–2076; (b) C. Nevado and A. M. Echavarren, *Tetrahedron*, 2004, **60**, 9735–9744.
- 48 T. Higashino, S. Sakaguchi and Y. Ishii, *Org. Lett.*, 2000, **2**, 4193–4195.
- 49 B. Schmidt, *Synlett*, 2004, 1541–1544.
- 50 C. Romano and C. Mazet, *J. Am. Chem. Soc.*, 2018, **140**, 4743–4750.
- 51 (a) B. M. Rosen, K. W. Quasdorf, D. A. Wilson, N. Zhang, A.-M. Resmerita, N. K. Garg and V. Percec, *Chem. Rev.*, 2011, **111**, 1346–1416; (b) J. Cornella, C. Zarate and R. Martin, *Chem. Soc. Rev.*, 2014, **43**, 8081–8097; (c) M. Tobisu and N. Chatani, *Acc. Chem. Res.*, 2015, **48**, 1717–1726.
- 52 H. Li and C. Mazet, *Acc. Chem. Res.*, 2016, **49**, 1232–1241.
- 53 For the original development of the Pd catalyst, see: (a) E. Larionov, L. Lin, L. Guénée and C. Mazet, *J. Am. Chem. Soc.*, 2014, **136**, 16882–16894; (b) L. Lin, C. Romano and C. Mazet, *J. Am. Chem. Soc.*, 2016, **138**, 10344–10350.
- 54 G. Ho, L. Judkele, J. Bruffaerts and I. Marek, *Angew. Chem., Int. Ed.*, 2018, **57**, 8012–8016.
- 55 G.-M. Ho, L. Segura and I. Marek, *Chem. Sci.*, 2020, **11**, 5944–5949.
- 56 Z. Wu, S. D. Laffoon, T. T. Nguyen, J. D. McAlpin and K. L. Hull, *Angew. Chem., Int. Ed.*, 2017, **56**, 1371–1375.
- 57 (a) K. Tani, T. Yamagata, S. Akutagawa, H. Kumobayashi, T. Taketomi, H. Takaya, A. Miyashita, R. Noyori and S. Otsuka, *J. Am. Chem. Soc.*, 1984, **106**, 5208–5217; (b) S. Inoue, H. Takaya, K. Tani, S. Otsuka, T. Sato and R. Noyori, *J. Am. Chem. Soc.*, 1990, **112**, 4897–4905; (c) M. Emura and H. Matsuda, *Chem. Biodiversity*, 2014, **11**, 1688–1699.
- 58 Z. Wu and K. L. Hull, *Chem. Sci.*, 2016, **7**, 969–975.
- 59 S. D. Laffoon, Z. Wu and K. L. Hull, *Chem. Commun.*, 2018, **54**, 7814–7817.
- 60 Z. Wu, S. D. Laffoon and K. L. Hull, *Nat. Commun.*, 2018, **9**, 1185.
- 61 Y. Fujii, T. Takehara, T. Suzuki, H. Fujioka, S. Shuto and M. Arisawa, *Adv. Synth. Catal.*, 2015, **357**, 4055–4062.
- 62 (a) R. C. van der Drift, E. Bouwman and E. J. Drent, *Organomet. Chem.*, 2002, **650**, 1–24; (b) R. Uma, C. Crévisy and R. Grée, *Chem. Rev.*, 2003, **103**, 27–51; (c) L. Mantilli and C. Mazet, *Chem. Lett.*, 2011, **40**, 341–344; (d) P. Lorenzo-Luis, A. Romerosa and M. Serrano-Ruiz, *ACS Catal.*, 2012, **2**, 1079–1086; (e) D. Cahard, S. Gaillard and J.-L. Renaud, *Tetrahedron Lett.*, 2015, **56**, 6159–6169.
- 63 D. Y. Lee, C. W. Moon and C. H. Jun, *J. Org. Chem.*, 2002, **67**, 3945–3948.
- 64 (a) M. C. Willis, *Chem. Rev.*, 2010, **110**, 725–748; (b) M. Fernández and M. C. Willis, Rhodium(I)-catalyzed hydroacylation, in *Rhodium catalysis in organic synthesis: Methods and reactions*, ed. K. Tanaka, Wiley-VCH, 2019.
- 65 A. Bartoszewicz and B. Martín-Matute, *Org. Lett.*, 2009, **11**, 1749–1752.
- 66 (a) V. Cadierno, J. Francos, J. Gimeno and N. Nebra, *Chem. Commun.*, 2007, 2536–2538; (b) V. Cadierno, P. Crochet, J. Francos, S. E. García-Garrido, J. Gimeno and N. Nebra, *Green Chem.*, 2009, **11**, 1992–2000.
- 67 A. E. Díaz-Álvarez, P. Crochet and V. Cadierno, *Catal. Commun.*, 2011, **13**, 91–96.
- 68 V. Bizet, X. Pannecoucke, J. L. Renaud and D. Cahard, *Adv. Synth. Catal.*, 2013, **355**, 1394–1402.
- 69 (a) R. Wu, M. G. Beauchamps, J. M. Laquidara and J. R. Sowa, *Angew. Chem., Int. Ed.*, 2012, **51**, 2106–2110. See also: (b) V. Bizet, X. Pannecoucke, J.-L. Renaud and D. Cahard, *Angew. Chem., Int. Ed.*, 2012, **51**, 6467–6470.
- 70 Y. Sun, C. LeBlond, J. Wang, D. G. Blackmond, J. Laquidara and J. R. Sowa Jr., *J. Am. Chem. Soc.*, 1995, **117**, 12647–12648.
- 71 (a) T. Slagbrand, H. Lundberg and H. Adolffson, *Chem. – Eur. J.*, 2014, **20**, 16102–16106; (b) C. O. Shoola, T. DelMastro, R. Wu and J. R. Sowa, *Eur. J. Org. Chem.*, 2015, 1670–1673.
- 72 Z. Sahli, B. Sundararaju, M. Achard and C. Bruneau, *Org. Lett.*, 2011, **13**, 3964–3967.
- 73 H. Li, M. Achard, C. Bruneau, J. B. Sortais and C. Darcel, *RSC Adv.*, 2014, **4**, 25892–25897.
- 74 W. Zhang, C. Meng, Y. Liu, Y. Tang and F. Li, *Adv. Synth. Catal.*, 2018, **360**, 3751–3759.
- 75 (a) The Chemistry of Metal Enolates, in *PATAI's Chemistry of Functional Groups*, ed. J. Zabicky, Wiley, 2018, vol. 2; (b) D. Vargová, I. Némethová, K. Plevová and R. Šebesta, *ACS Catal.*, 2019, **9**, 3104–3143.
- 76 T. Doi, T. Fukuyama, S. Minamino, G. Husson and I. Ryu, *Chem. Commun.*, 2006, 1875–1877.
- 77 T. Fukuyama, T. Doi, S. Minamino, S. Omura and I. Ryu, *Angew. Chem., Int. Ed.*, 2007, **46**, 5559–5561.
- 78 J. Masson-Makdissi, Y. J. Jang, L. Prieto, M. S. Taylor and M. Lautens, *ACS Catal.*, 2019, **9**, 11808–11812.
- 79 (a) S. H. Bergens and B. Bosnich, *J. Am. Chem. Soc.*, 1991, **113**, 958–967; (b) L. J. Gazzard, W. B. Motherwell and D. A. Sandham, *J. Chem. Soc., Perkin Trans. 1*, 1999, 979–993; (c) C. Schneider and M. Boomhoff, Aldol Reactions in Domino Processes, in *Domino Reactions*, ed. L. F. Tietze, Wiley-VCH, Weinheim, Germany, 2014, ch. 8, pp. 267–294.
- 80 N. Ahlsten, A. Bartoszewicz and B. Martín-Matute, *Dalton Trans.*, 2012, **41**, 1660–1670.



- 81 (a) N. Ahlsten, A. B. Gómez and B. Martín-Matute, *Angew. Chem., Int. Ed.*, 2013, **52**, 6273–6276; (b) A. B. Gómez, E. Erbing, M. Batuecas, A. Vázquez-Romero and B. Martín-Matute, *Chem. – Eur. J.*, 2014, **20**, 10703–10709; (c) A. Vázquez-Romero, A. B. Gómez and B. Martín-Matute, *ACS Catal.*, 2015, **5**, 708–714; (d) S. Martínez-Erro, A. Bermejo Gómez, A. Vázquez-Romero, E. Erbing and B. Martín-Matute, *Chem. Commun.*, 2017, **53**, 9842–9845; (e) A. Sanz-Marco, Š. Možina, S. Martínez-Erro, J. Iskra and B. Martín-Matute, *Adv. Synth. Catal.*, 2018, **360**, 3884–3888.
- 82 A. Sanz-Marco, S. Martínez-Erro and B. Martín-Matute, *Chem. – Eur. J.*, 2018, **24**, 11564–11567.
- 83 A. Sanz-Marco, S. Martínez-Erro, M. Pauze, E. Gómez-Bengoa and B. Martín-Matute, *Nat. Commun.*, 2019, **10**, 5244.
- 84 (a) B. Schetter and R. Mahrwald, *Angew. Chem., Int. Ed.*, 2006, **45**, 7506–7525; (b) P. B. Brady and H. Yamamoto, in *Modern Methods in Stereoselective Aldol Reactions*, ed. R. Mahrwald, Wiley-VCH, Weinheim, Germany, 2013, pp. 269–291; (c) J. Feng, Z. A. Kasun and M. J. Krische, *J. Am. Chem. Soc.*, 2016, **138**, 5467–5478.
- 85 L. Lin, K. Yamamoto, H. Mitsunuma, Y. Kanzaki, S. Matsunaga and M. Kanai, *J. Am. Chem. Soc.*, 2015, **137**, 15418–15421.
- 86 (a) X. Liu, B. Liu and Q. Liu, *Angew. Chem., Int. Ed.*, 2020, **59**, 6750–6755; (b) C. Yang, Y. Gao, S. Bai, C. Jiang and X. Qi, *J. Am. Chem. Soc.*, 2020, **142**, 11506–11513.
- 87 C. Matt, C. Kern and J. Streuff, *ACS Catal.*, 2020, **10**, 6409–6413.
- 88 S. Hanna, T. Wills, T. W. Butcher and J. F. Hartwig, *ACS Catal.*, 2020, **10**, 8736–8741.
- 89 (a) H. E. Basbug Alhan, G. R. Jones and E. Harth, *Angew. Chem., Int. Ed.*, 2020, **59**, 4743–4749; (b) G. R. Jones, H. E. Basbug Alhan, L. J. Karas, J. I. Wu and E. Harth, *Angew. Chem., Int. Ed.*, 2020, **59**, DOI: 10.1002/anie.202012400.

