



Cite this: *Chem. Soc. Rev.*, 2021, 50, 39

Received 20th July 2020

DOI: 10.1039/d0cs00344a

rsc.li/chem-soc-rev

Photocatalytic methods for amino acid modification

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Amino acid modification plays an important role across several fields, including synthetic organic chemistry, materials science, targeted drug delivery and the probing of biological function. Although a myriad of methods now exist for the modification of peptides or proteins, many of these target a handful of the most reactive proteinogenic amino acids. Photocatalysis has recently emerged as a mild approach for amino acid modification, generating a sizable toolbox of reactions capable of modifying almost all of the canonical amino acids. These reactions are characterised by their mild, physiologically compatible conditions, greatly enhancing their usefulness for amino acid modification. This review aims to introduce the field of photocatalytic amino acid modification and discusses the most recent advances.

Key learning points

- (1) Amino acids are involved across a wide range of fields, from organic synthesis to biology to materials science.
- (2) Side chains of amino acids provide unique properties to peptides or proteins.
- (3) Side chains of amino acids can be selectively targeted to enable modification of the amino acid, peptide or protein.
- (4) Photocatalysis uses mild conditions to facilitate modification, making it appropriate for use on biological systems.
- (5) Appropriate selection of photocatalyst, based on redox potentials, can enable selective modification of most amino acids.

1. Introduction

α -Amino acids are the building blocks of peptides and proteins synthesised by living cells.¹ Containing a primary amine and a carboxylic acid, separated by a single carbon atom, amino acids are primed for amide, or peptide, bond formation (Fig. 1). Nature provides twenty common amino acids, with side chains displaying polar, aromatic or aliphatic groups. Within proteins, these groups provide interactions for stabilising the three-dimensional, tertiary, structure. On the surface, these groups provide unique environments for recognition of other biomolecules. Within proteins with catalytic function, enzymes, nearby side chains set the steric and electronic requirements for the binding of substrate molecules and can contribute to catalysis.

Researchers typically produce peptides or proteins for study in one of two ways. Peptides and short proteins may be chemically synthesised by solid-phase peptide synthesis (SPPS). This method can easily facilitate the introduction of

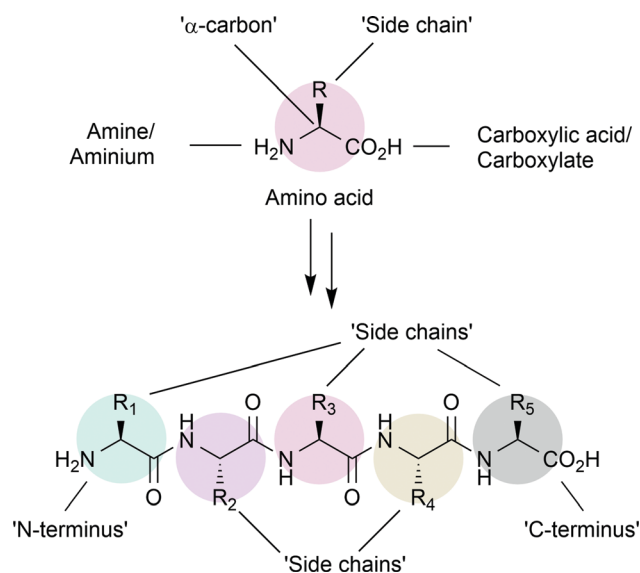


Fig. 1 Basics of amino acid nomenclature. Amino acids, and peptides, contain an amino (N) and a carboxy (C) terminus. Each individual amino acid also has a side chain, of which there are 20 commonly found in nature, and which proffer specific properties to peptides.

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non-canonical amino acids (ncAAs), and while this was previously limited to peptides of less than 50 AA, recent developments have drastically increased that limit.² Alternatively, larger peptides and proteins may be produced by *in vitro* cellular expression, by introduction of the encoding DNA sequence into the expression host. While large proteins are more easily produced using this method, introduction of ncAAs *in vitro* requires the use of codon-suppression technology and addition of the ncAA to the cell suspension.³

Amino acid modification is relevant across multiple fields (Fig. 2). In nature itself, many amino acid residues have post-translational modifications (PTMs); phosphate groups, acetyl groups, sulfonate groups and methyl groups are all known to be added to amino acid side chains after the translation of an mRNA strand into a peptide to transiently control their activity and increase the structural diversity available from just 20 available building blocks.⁴ Mimicking these modifications can help researchers study their effects. Beyond simple PTMs, a range of other chemical moieties have been artificially added to peptides or proteins.⁵ For instance, addition of chemical

moieties (e.g. fluorophores) to proteins of interest can aid in the elucidation of cellular processes (e.g. subcellular protein localisation). Alternatively, addition of a cytotoxic or modulatory drug to a targeting protein (e.g. a tumour-specific antibody) can produce a biotherapeutic with superior efficacy and tolerability compared to chemotherapy.⁶ As readily available chiral reagents, α -amino acids are used extensively in synthetic organic chemistry. Therefore, methods for the modification of the common 20 amino acids enhances their usefulness as building blocks. Beyond this, biocatalytic transformations are now widespread in organic synthesis. While separation of an enzyme from a reaction mixture can be challenging, modification of the enzyme such that it is attached to a solid support (at a residue which is not significant for enzyme activity), purification can be greatly simplified. Finally, the use of biomolecules in the development of materials is garnering increased interest.⁷

The modification of an amino acid greatly expands its utility in chemical synthesis. Many synthetic strategies rely on naturally occurring chirality to control the stereoselectivity of



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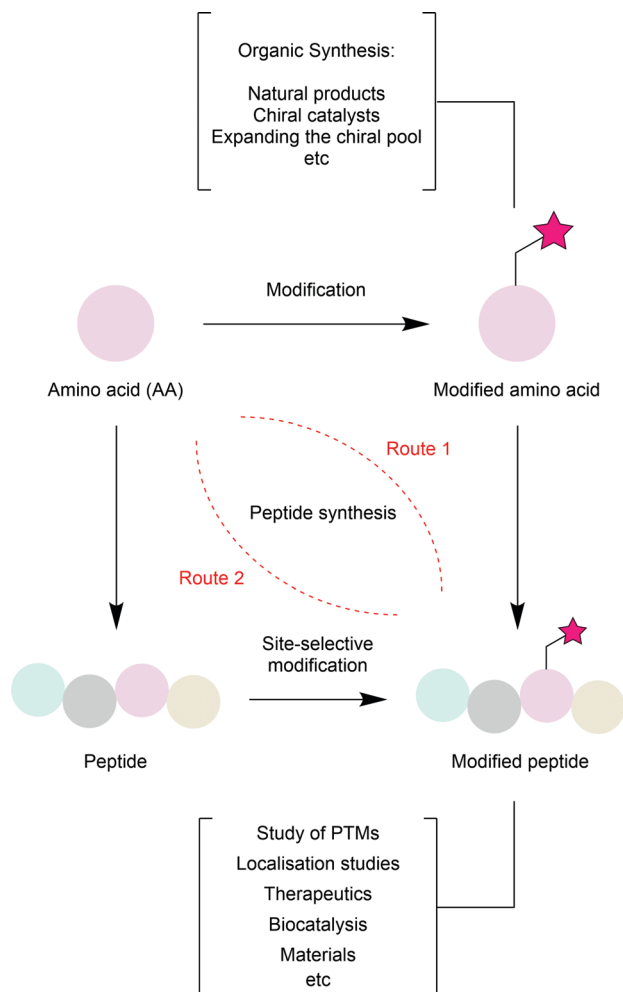


Fig. 2 Routes to, and applications of, amino acid and peptide modification. Amino acids and their derivatives can be used in chemical synthesis as chiral reagents or building blocks to control or install stereochemistry. Amino acids can also be used to generate modified peptides *via* two distinct routes. Modification can be either made to an individual amino acid which is then incorporated during peptide synthesis (route 1) or achieved site-selectively after peptide synthesis (route 2). Modification of peptides can aid discovery across a wide variety of fields.

reactions, either using chiral catalysts (reagent control), or through the influence of chiral centres within the substrate (substrate control). α -Amino acids constitute a significant portion of the so-called 'chiral pool' used by researchers to this end. There are 20 commonly available amino acids, which can provide unique functionality within a synthesis, either through incorporation into the substrate, or for the generation of catalysts. Methods for modifying an amino acid without epimerisation of the chiral centre would enable the generation of a far wider selection of reagents, thereby deepening the chiral pool. These alternative amino acids can also be used in peptide or protein synthesis as ncAAs.

The modification of an amino acid within a peptide or protein through the addition of a chemical moiety is commonly known as bioconjugation. Bioconjugation is also used more generally to describe the chemical addition of a moiety to any

biological molecule. It has long been established that these added moieties can have a profound impact on the behaviour of the biomolecule of interest.

Any bioconjugation reaction must consider the implications of the chemistry on the substrate. Peptides and proteins are typically folded into an intricate secondary and often tertiary structure, formed in their native cellular environment. This can be disrupted by reagents (such as ureas), pH fluctuation, or heat, leading to a breakdown of structure and consequently function.¹ Therefore, bioconjugation protocols must utilise exceptionally mild reaction conditions to maintain the structure of the biomolecule. Additionally, many peptides or proteins use particular side chains or groups of side chains to carry out their function. The purpose of the modification must therefore be carefully considered; to block or inhibit function through modification, an amino acid side chain which is required for activity should be targeted. Conversely, if retained function is desired, modification distal to the functional site or domain should be undertaken. An 'ideal' bioconjugation strategy can therefore be said to involve:

- Site-selectivity (chemo- and regioselective)
- Compatibility with physiological conditions (*ca.* 37 °C, aqueous solvent, neutral pH)
- Fast reaction kinetics
- Minimal, non-toxic by-products

Using traditional chemistry, the vast majority of bioconjugation tools have made use of the nucleophilicity of a handful of amino acid side chains, particularly lysine or cysteine, to install the modification.⁸ An alternative strategy which has been widely used in protein or peptide modification is the incorporation of non-canonical amino acids with side chains that contain bioorthogonally reactive functional groups (*e.g.* azides, ketones).⁹ These groups are then amenable to selective modification with the desired payload containing a complementary reactive handle. While effective, these strategies require genetic code expansion for ribosomal installation in a growing peptide chain and/or chemical synthesis of the complex ncAAs. In recent years, transition metal catalysis has also been applied to the modification of amino acids, peptides and proteins.¹⁰

Photocatalysis has recently emerged as a mild method for amino acid modification.¹¹ Prominent in the attractive attributes of photocatalysis are the mild conditions often required and the potential modification of residues with otherwise inert side chains. Through photocatalysis, radical generation is simplified greatly – the need for high energy reagents or initiators, which are commonly required in traditional radical processes, is eliminated. Furthermore, visible-light-mediated methods for radical generation can typically be achieved at ambient temperature. For amino acid modification, these mild conditions reduce the chance of epimerisation of the α -carbon. While some photoredox-catalysed methods are only suitable for individual amino acid modification, the mild conditions can also allow for application of the chemistry to peptides, or even proteins.

This tutorial review provides an introduction to photoredox catalysis for amino acid modification, both individually and



within larger peptide sequences. The underlying principles of photoredox catalysis are described, followed by implications for amino acid modification. For each amino acid, a brief history of its photocatalytic modification is provided, along with the most recent advances. Of particular note is the recent development of C–H functionalisation methods, which have enabled modification of amino acids for which there are currently limited alternatives. For discussions of other amino acid modification strategies please refer to other reviews of the field.^{8,12–16}

2. Photoredox catalysis

The advent of visible-light photosensitisers has re-energised research into light-mediated transformations. Light has been used in chemistry for decades, for example to promote bond cleavage (*e.g.* initiators such as azobisisobutyronitrile (AIBN)) or enable thermally-prohibited transformations by promoting alkenes to their excited states (*e.g.* cycloadditions/pericyclic reactions). However, with the discovery of molecules which interact with visible light came the realisation that they can act as mediators for a plethora of chemical transformations. Using visible light, a photosensitiser (or photocatalyst) and a substrate with an appropriately matched redox potential, radical species can be generated under remarkably mild conditions, and subsequently harnessed to achieve chemical transformations.^{17,18}

The photocatalytic cycle for most photochemical processes discussed here involves electron transfer between excited

catalyst and substrate, although energy transfer is also possible.¹⁹ Photoredox catalysis begins with the electronic excitation of an organic or inorganic compound (PC), through absorption of light (Fig. 3). These excited state molecules (PC*) then undergo quenching, typically through single electron transfer (SET), resulting in oxidation (blue) or reduction (red) of a substrate. The substrate is consequently activated for bond cleavage, atom abstraction, or nucleophilic or electrophilic attack. After quenching, the oxidised or reduced catalyst (PC^{ox} or PC^{red}) regains, or loses, an electron to return to the starting ground state catalyst (PC).

In developing a photocatalytic protocol, several key properties of the catalyst must be considered: redox potential, excited state lifetime, absorption maximum, and solubility. If oxidation of the substrate is desired, a photocatalyst which has an excited state reduction potential above that of the substrate is required. Conversely, if reduction is desired, the reverse is true. Photocatalysts have been discovered or developed, with a wide range of excited state and ground state redox potentials (most within *ca.* ± 2 V *vs.* the standard hydrogen electrode, SHE), thus enabling selection of the most appropriate catalyst for a chosen substrate. Significantly, the lifetime of the excited state of the catalyst before spontaneous return to the ground state varies greatly across catalysts. Catalysts with longer lifetimes allow for a greater probability of reaction, as the probability of the reactive catalyst species associating with substrate molecules increases with increased excited state lifetime. Transition metal-based catalysts tend to have longer excited state lifetimes

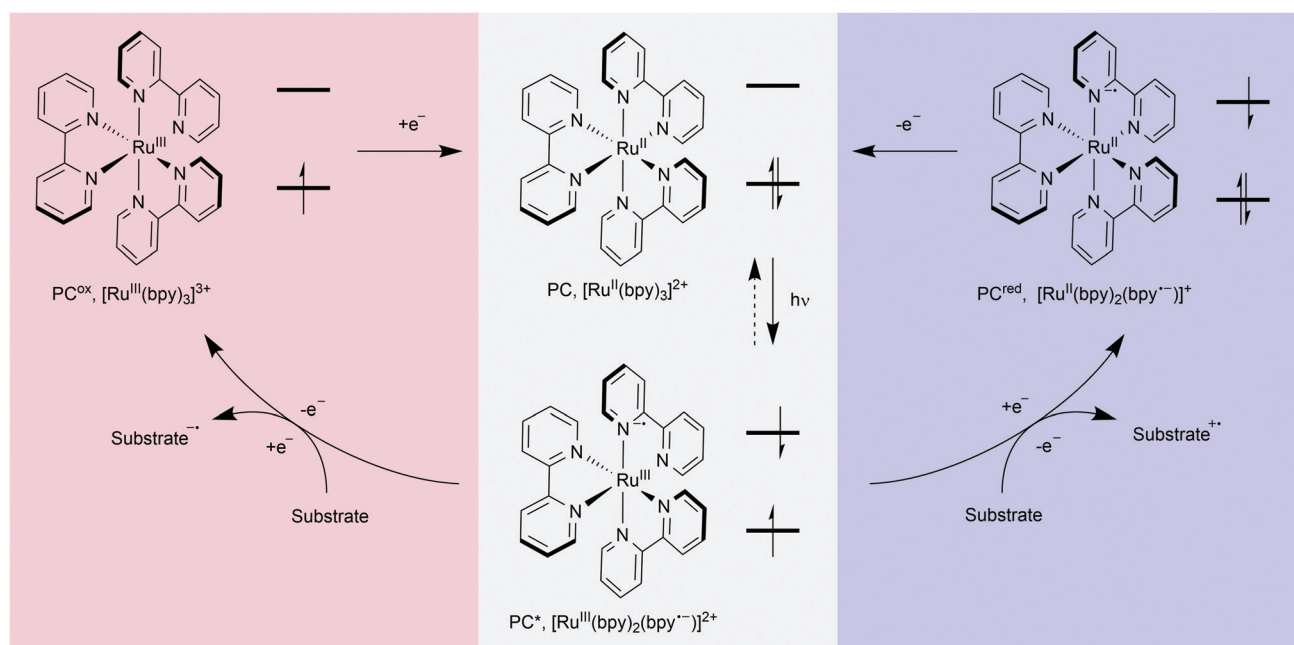


Fig. 3 Visible-light-mediated activation of a photoredox catalyst: upon absorption of light of an appropriate wavelength, the photocatalyst (PC) becomes excited (usually denoted by an asterisk), with one of its electrons promoted to a higher energy orbital. This excited state is activated for electron (or energy) transfer. Substrates with a higher redox potential can accept an electron from the excited catalyst (red); substrates with a lower redox potential can lose an electron to the excited catalyst (blue). In this way, organic radicals are formed on the substrates, which can then undergo further reaction. Subsequently, electrons are returned to, or taken from, the intermediate catalyst species to regenerate the ground state (GS) catalyst and close the catalytic cycle. Bpy = 2,2'-bipyridyl.



than purely organic photocatalysts (μs compared to ns). Finally, the wavelength at which the catalyst absorbs is also important. Matching the light source with the absorption maximum of the catalyst will greatly increase the proportion of catalyst molecules in the excited state, thereby increasing the rate of reaction. Most commonly used photocatalysts absorb in the blue region of the visible spectrum, although new catalysts are being developed which can absorb longer wavelength red light.²⁰ Redox potentials, excited state lifetimes, and absorption maxima for commonly used photocatalysts are readily available.^{17,18}

Practically, photocatalysis is simple to perform. After careful selection of an appropriate catalyst, a light source must be acquired. Despite calls for standardisation in recent years,²¹ most published reports detail the use of 'shop-bought' LED light sources. Provided the absorption maximum of your catalyst lies within the emitted wavelength of your source, any light should provide results. However, while household iridescent lightbulbs can provide the required photons, they often also emit large quantities of heat. LEDs, on the other hand, offer irradiation without excessive heating. Despite this, over time, minimal heating from the light source can significantly raise the reaction temperature. For this reason, a fan is often used to maintain constant reaction conditions.

3. Photocatalytic amino acid modification

Many side chains of amino acids, particularly those with heteroatoms, have appropriate oxidation potentials for quenching excited state photocatalysts. With careful control of oxidation potentials *via* selection of an appropriate catalyst, selective oxidation or reduction of, and therefore radical generation at, amino acid residues can be performed. These radicals can then be coupled, generally with electrophiles or other radicals, to install the desired modification.

Recent years have witnessed a significant increase in the number of photocatalytic methods being reported. While the majority of these do not focus solely on amino acid modification, several document investigations on amino acid or peptide substrates to showcase the utility of the methodology. In the following sections, advances aimed specifically at amino acid or peptide modification, and those that demonstrate applicability to these, will be discussed. To aid the reader in understanding the application of each method, a colour code will be used. Procedures used in modification of individual amino acids will be marked red, those demonstrated on peptide substrates will be marked blue, and those demonstrated on protein substrates will be marked green.

Polar amino acid side chains

Polar amino acids (see Fig. 4), most notably cysteine and lysine, have received great interest from the scientific community in the development of ionic methods for amino acid modification.⁸

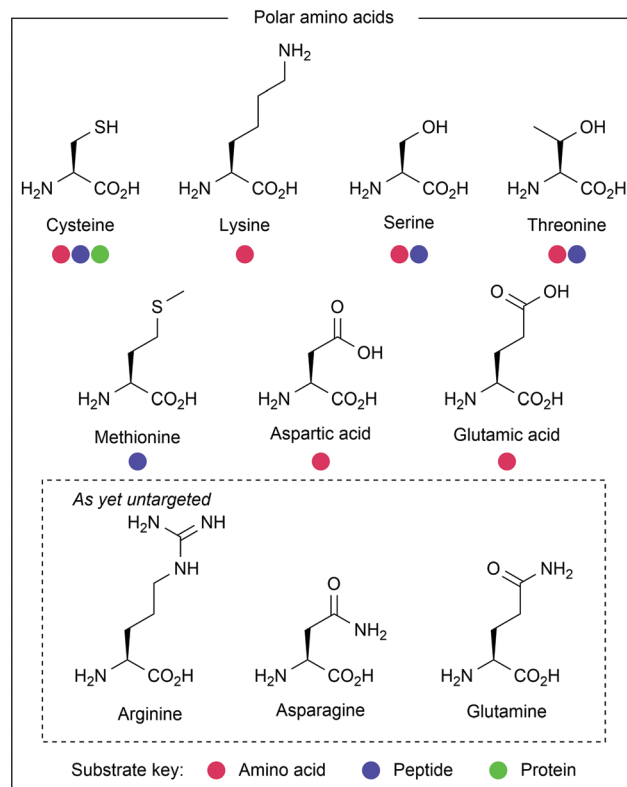


Fig. 4 Amino acids containing polar side chains, or whose photochemical activity involves the presence of heteroatoms.

However, they are also amenable to modification *via* radical processes, and have thus been targeted through photocatalysis.

Cysteine. Cysteine residues are commonly modified using ionic pathways, by addition to electrophiles such as maleimides or other conjugate acceptors.⁸ However, photocatalytic methods have been developed which expand the range of transformations that can be made.

Thiol-ene reactions, commonly achieved using chemical radical initiators or ultraviolet (UV) light, can also be achieved using visible light photocatalysis. In 2017, Wang and co-workers used allyl alcohols and amides as reagents for cysteine and thio-sugar modification (Fig. 5).²² An excited acridinium catalyst was used as the necessary oxidising agent, generating the thiyl radical after deprotonation. The alkene could then be attacked, and the product generated after a final hydrogen atom abstraction, possibly from unreacted thiol, thereby generating additional thiyl radical. Testing the reaction scope with benzyl mercaptan, allyl esters, amides, alcohols and silanes were shown to be competent coupling partners after irradiation for 6 hours (76–87%). Although few examples using cysteine were presented, those couplings were also achieved in high yields (75–90%), attaching a sugar, nucleic acid, or peptide to the residue. Of particular interest was the treatment of cysteine with the allyl ester of aspartic acid, which gave the cross-linked amino acids in high yield (85%), suggesting the potential utility of this methodology for peptide macrocyclisation.

In 2018, Molander and co-workers developed a method for cysteine arylation using a ruthenium photocatalyst, nickel



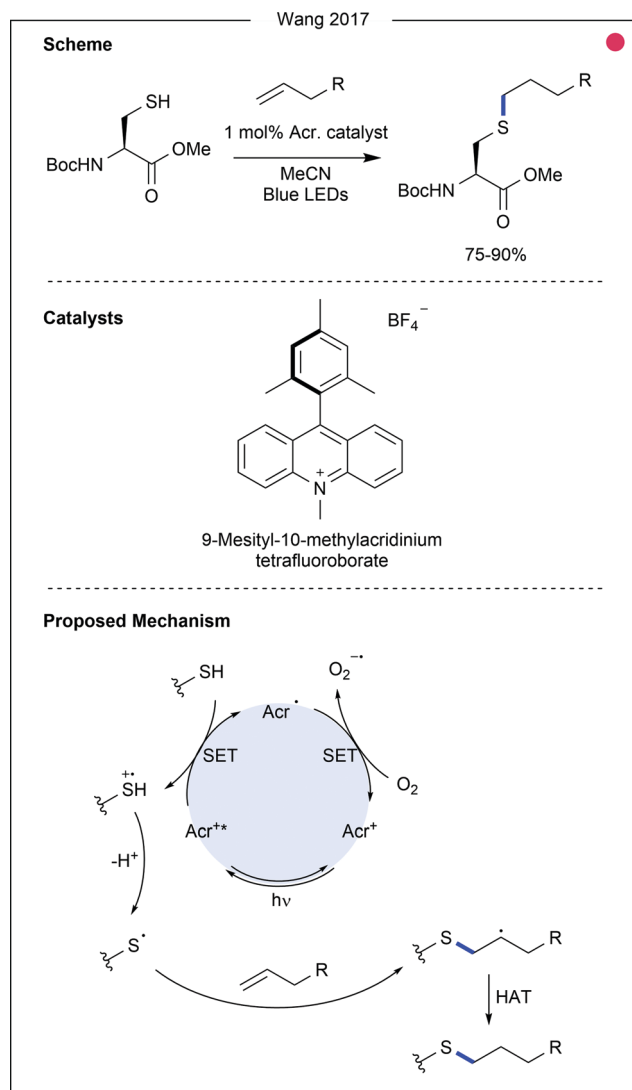


Fig. 5 Thiol-ene reaction of cysteine. Using an acridinium catalyst to generate the thiyl radical, cysteine undergoes alkene addition. The resultant secondary radical persists for long enough for hydrogen atom transfer (HAT) to occur to generate the modification product. Acr. = 9-mesityl-10-methylacridinium.

co-catalyst and a silicate-based hydrogen atom transfer reagent (Fig. 6).²³ After excitation of the photocatalyst, oxidation of the silicate leads to the homolytic cleavage of the C-Si bond to produce a carbon-centred radical. This was able to abstract the proton from the cysteine thiol, generating a thiyl radical. Trapping of the thiyl radical with the nickel(0) species was then followed by oxidative addition of an aryl halide. Finally, reductive elimination from the nickel(III) complex produced the C(sp²)-S bond. The aryl substrate scope and amino acid compatibility was explored using glutathione (GSH) as substrate. Substituents on the aryl bromide greatly influenced the success of the reaction, as yields ranged from 14% (with a *para*-alcohol) to 83% (using 4-bromobenzyl boronic acid). Aryl bromides containing nitrile, halide, amide, sulfonamide, ether, alcohol, and carboxylate groups were all successful coupling partners.

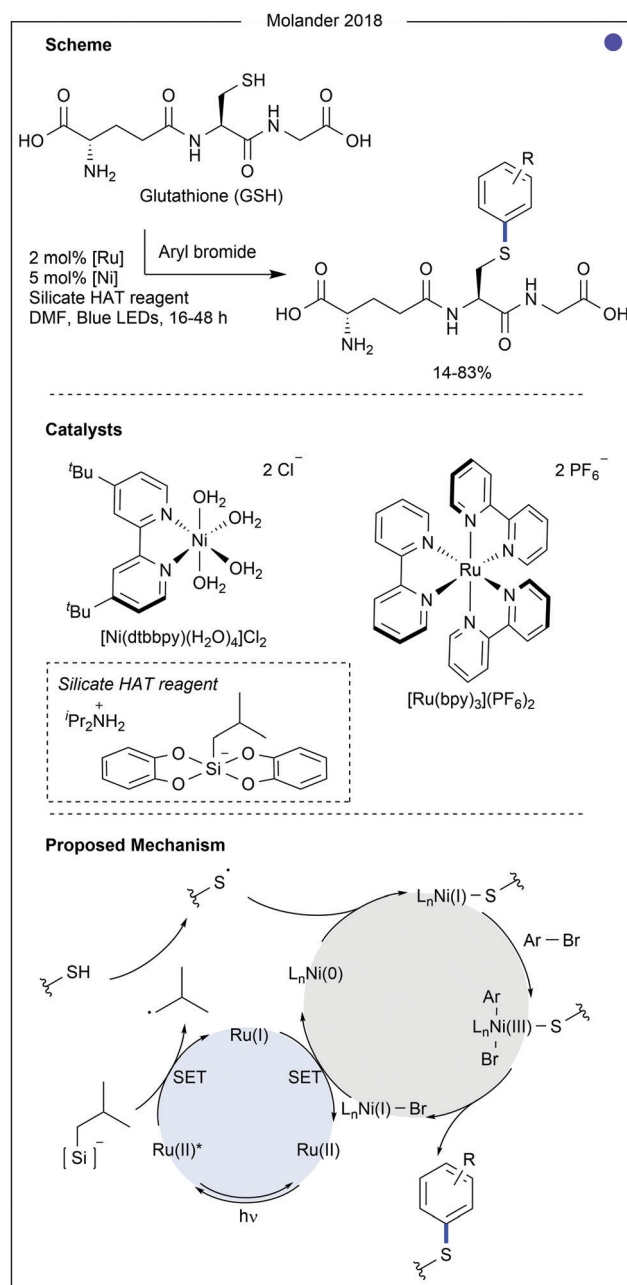


Fig. 6 Cysteine arylation using dual nickel-ruthenium catalysis. A silicate reagent was proposed to generate the thiyl radical required in the nickel catalytic cycle through hydrogen atom transfer (HAT). Dtbbpy = 4,4'-di-*tert*-butyl-2,2'-bipyridyl. Bpy = 2,2'-bipyridyl.

Disappointingly, a pendant azide group was not tolerated on the aryl bromide. Aryl bromide-containing chemical linkers, biologic probes and drug molecules were all used for the modification, producing GS-aryl conjugates in moderate to good yields (35–83%). The method was also successfully applied to the modification of a polypeptide (9 amino acids (AAs)) using 4-bromobenzonitrile (complete conversion was observed). Carboxylate functionality and aromatic amino acids were shown not to interfere with the reaction, whereas the presence of lysine, proline and arginine significantly hindered reactivity.



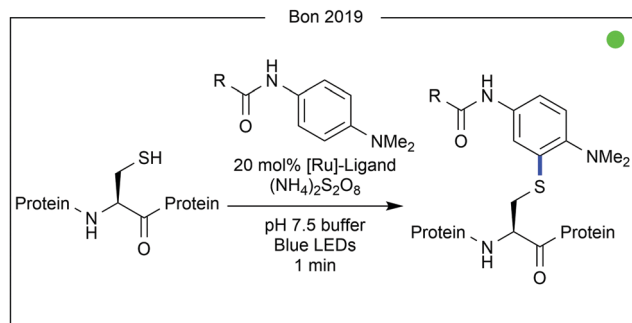


Fig. 7 Ligand directed modification of cysteine on a protein surface. Using methodology similar to that used on tyrosine by Nakamura (*vide infra*), Wilson and co-workers were able to selectively label cysteine on the MCL-1 protein.

Bon, Wilson and co-workers recently demonstrated a ligand-directed approach to the selective labelling of a native cysteine-containing protein, and current anticancer target, human MCL-1 (Fig. 7).²⁴ By appending a ruthenium photocatalyst to an oligopeptide (19 AAs) known to bind the target protein, the group were able to selectively oxidise and therefore modify a proximal cysteine residue (Cys286) on MCL-1 after irradiation with visible light. The thiyl radical formed after cysteine oxidation was trapped with aniline-based reagents similar to those used in previous work by Nakamura and co-workers for photocatalytic tyrosine modification (*vide infra*). While significant conversion was observed, increased irradiation time led to the generation of by-products, presumed to be due to oxidation of the MCL-1 protein by the photocatalyst. Attachment of a fluorophore or a biotin tag was achieved, the latter of which enabled purification of the target protein conjugate. Finally, competition experiments were performed using a mixture of MCL-1 and an analogous protein, BCL-x_i to which the ligand peptide did not bind. Only limited labelling of the control protein was observed.

Lysine. Lysine residues are traditionally targeted using chemistry based on their inherent nucleophilicity. However, Rovis and co-workers used the stability of radicals α to sulfonamides to incite reactivity at the ϵ -carbon of a protected lysine residue (Fig. 8).²⁵ Using a trifluoromethylsulfonamide (triflamide) protecting group for the side-chain amine, addition of *tert*-butyl acrylate at the carbon α to the ϵ -amine was demonstrated. After excitation of an iridium catalyst, single electron oxidation of quinuclidine forms an intermediate capable of performing hydrogen atom transfer (HAT) with the α -carbon, thereby producing the radical anion of the sulfonamide. The lifetime of the radical is sufficient to perform radical conjugate addition to the *tert*-butyl acrylate, which is followed by SET and proton transfer to close the catalytic cycle. The electrophile substrate scope, assessed through reaction with *N*-triflyl propylamine, included α,β -unsaturated esters, ketones, phosphonates, sulfones, nitriles, amides and aromatics (32–77%). Although the primary goal of this work was to enable generic α -amino radical addition to electrophiles, the application of the technique to a lysine residue, albeit with moderate yield (58%), represents an exciting new route for lysine modification.

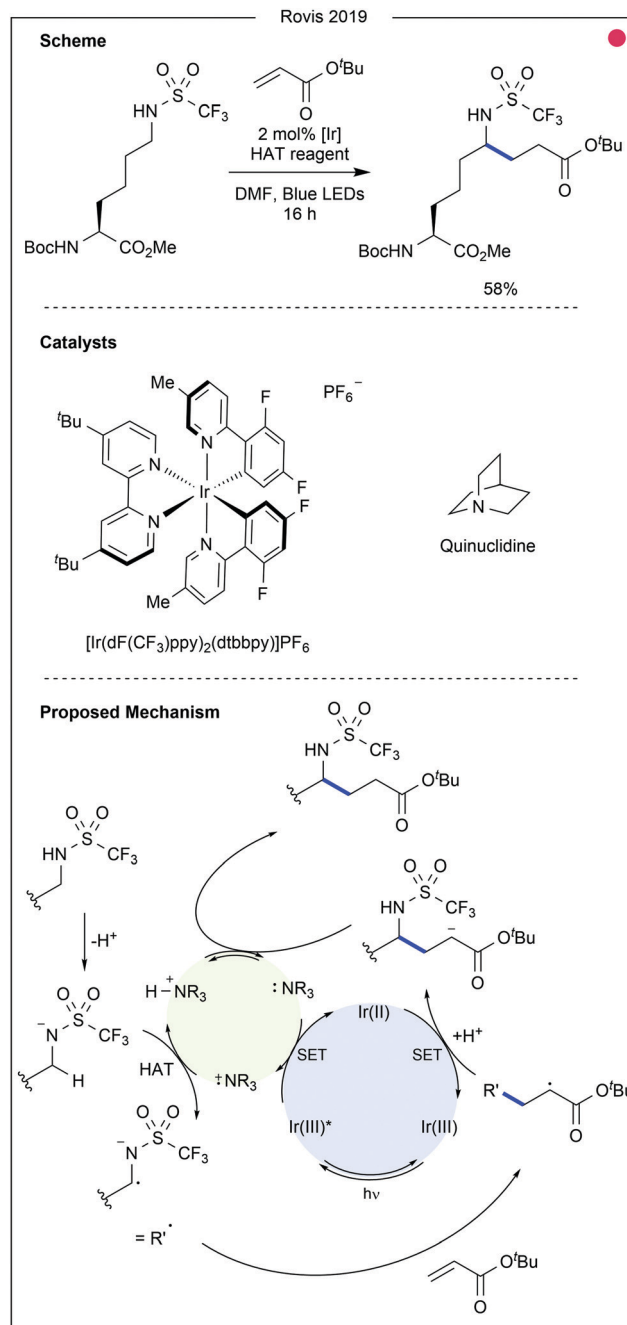


Fig. 8 Triflamide-enabled photocatalytic modification of lysine residues. The use of a triflamide to acidify the side chain N–H encourages hydrogen abstraction from the adjacent carbon atom. Conjugate addition then occurs, and the resultant radical is reduced to an anion through single electron transfer (SET) from the photocatalyst. Protonation then provides the final product. Quinuclidine acts as an intermediary; after oxidation by the excited photocatalyst, it becomes primed for hydrogen atom transfer from the substrate. [Ir] = [Ir(dF(CF₃)ppy)₂(dtbbpy)]PF₆.

Serine. Serine and threonine are rarely used as sites for bioconjugation, as their nucleophilicity is significantly lower than the amino or thiol side chains of lysine or cysteine, respectively. However, building on work by the MacMillan group on cross-coupling of aryl bromides with alcohols *via*



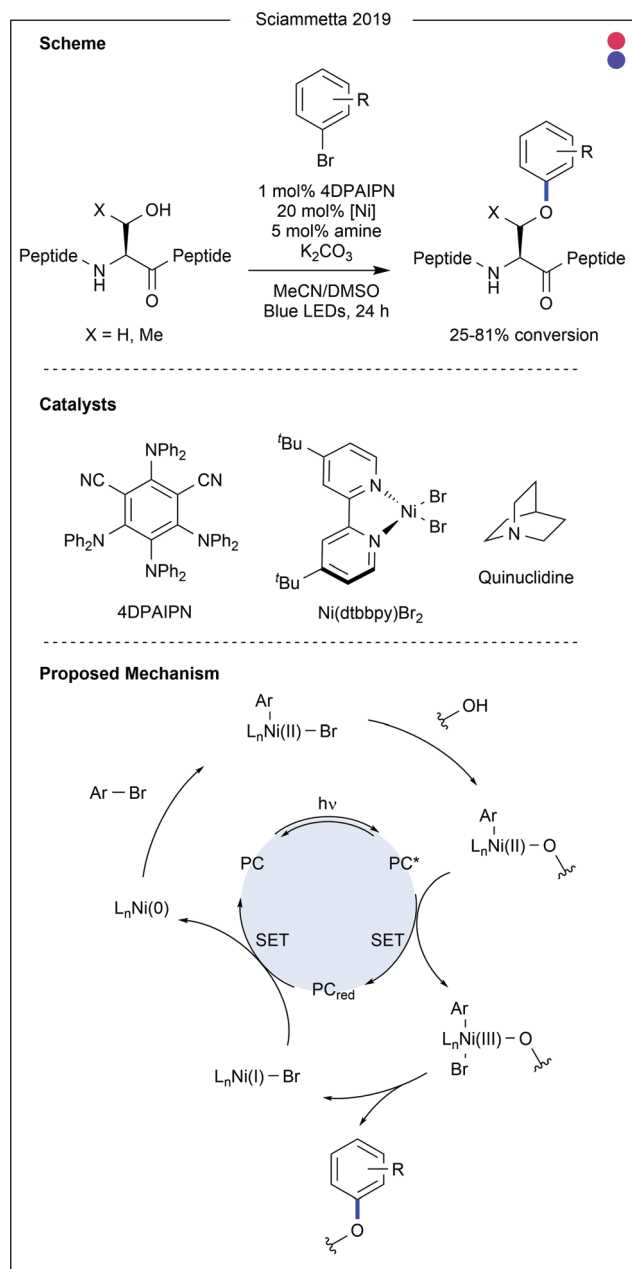


Fig. 9 Serine cross-coupling with aryl halides using dual organophotocatalysis and nickel catalysis. Using a photocatalyst to modify the electronic state of nickel during the catalytic cycle enables efficient cross coupling of alcohols such as serine and threonine with aryl bromides. PC = photocatalyst (4DPAIPN).

iridium–nickel dual catalysis,²⁶ Sciamentta and co-workers demonstrated the successful arylation of serine and threonine residues (Fig. 9).²⁷ The method involves a traditional transition metal catalysis process, whereby after oxidative addition of the aryl bromide to the nickel centre, ligation by the alcohol occurs. While nickel(II) complexes undergo slow reductive elimination, after excitation using visible light, a photosensitiser can oxidise the nickel centre to a nickel(III) species, from which reductive elimination can occur rapidly. Reduction of the nickel(I) species by the ground state reduced photocatalyst regenerates both

catalysts. It was postulated that a *para*-bromobenzoyl group was sufficiently electron-deficient that fast oxidative addition could occur at the nickel centre. This could be easily generated by benzoylation of the N-terminus of a peptide immediately after solid-phase peptide synthesis, removing the need for specialist unnatural amino acids to be generated. A variety of aryl bromide-capped peptides were used to form a range of inter- and intra-molecularly coupled products (25–81% conversion). *N*-Aryl bromide capped peptides containing phenylalanine, tryptophan, methionine and protected glutamic acid, glutamine and lysine were all suitable coupling partners for C- and N-terminally protected serine residues. Intramolecular coupling, otherwise known as macrocyclisation, of serine-containing peptides was significantly enhanced by the presence of a proline–proline, or β -hairpin, motif in the peptide sequence. β -Hairpin motifs force a peptide to fold back on itself, bringing the serine and aryl bromide into close proximity if strategically positioned. Unfortunately, water was observed to reduce the yield of the reactions, resulting instead in the nickel-mediated degradation of the aryl bromide to the corresponding phenol.

Methionine. In 2017, MacMillan and co-workers reported a procedure for coupling alkyl bromides *via* activation of the C–H bonds α to heteroatoms (Fig. 10).²⁸ By using iridium photocatalysis coupled with an amine hydrogen atom transfer reagent, abstraction of C–H hydrogen α to either sulfur, nitrogen or oxygen could be achieved. A nickel catalyst was used to ligate both the resultant radical and the alkyl bromide through oxidative addition, followed by reductive elimination to provide the desired product and complete the catalytic cycle. Among the many methyl- and methylene-reactive substrates were a di- and a tri-peptide sequence containing methionine (52% and 59% yields for reaction with 4-bromobutyronitrile, respectively). In these substrates, regioselectivity for the methyl, rather than the methylene, carbon adjacent to the sulfur atom was observed, as well as general chemoselectivity over the peptide backbone, containing multiple C–H bonds α to nitrogen. While only a single alkyl bromide was coupled to methionine, a wider selection of alkyl bromides, containing ester, ether, halide, phosphonate and N-heterocyclic functional groups, were demonstrated to be adequate coupling partners on non-amino acid substrates (41–83% yield).

While photocatalytic modification of native methionine within proteins remains limited, recent work from Gaunt and co-workers demonstrated site-selective addition of a reactive diazo handle to methionine, upon which photochemical modification was performed (Fig. 11).²⁹ Using a C₄-modified Hantzsch ester and a ruthenium photocatalyst, good conversion of the diazo group to a benzyl moiety was achieved (86%).

Carboxylates. After oxidation of a carboxylate, collapse to release CO₂ and generate an alkyl radical can occur under photocatalytic conditions. However, primary radicals formed from the decarboxylation of aspartic acid or glutamic acid are highly unstable (Fig. 12). Thus, decarboxylation of the native residue is often not observed. A common method for overcoming this issue in organic synthesis is through conversion of



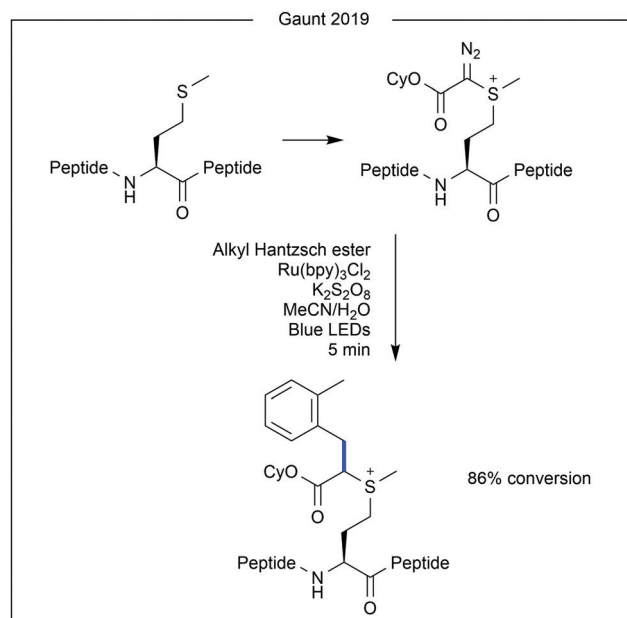


Fig. 10 Recent photocatalytic modification of methionine using iridium/nickel/HAT triple catalysis. Quinuclidine acts as a mediator between the iridium photocatalyst and the methionine side chain. After oxidation by the excited photocatalyst, the amine is capable of hydrogen atom transfer (HAT) from the carbons adjacent to the sulfur atom. Ligation to the nickel centre, followed by an oxidative addition, produces a nickel(III) complex primed for reductive elimination to generate the product. [Ir] = [Ir(dFCF₃ppy)₂(dtbbpy)]PF₆. Dtbbpy = di-*tert*-butyl-bipyridyl. Phe = phenylalanine. Ala = alanine.

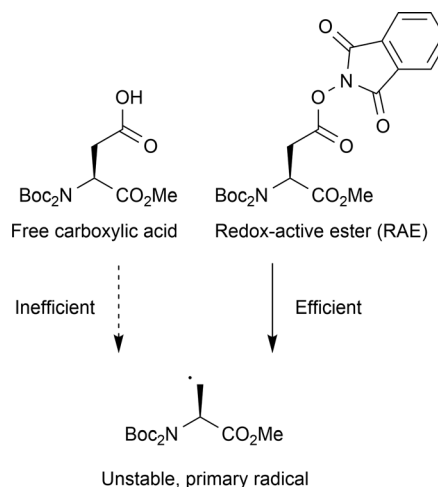


Fig. 12 Primary radical generation on amino acids containing side chain carboxylates. Due to the relative instability of the resulting primary radical, photocatalytic decarboxylation is inefficient from the free carboxylic acid. Instead, the side chain must be activated for decarboxylation using a redox-active ester (RAE), which undergoes simple O–N homolytic fission after single electron reduction. This generates a carboxylate radical which will rapidly decarboxylate to provide the desired primary carbon radical.

was performed with a variety of α,β -unsaturated carbonyls (ketones, esters and amides, 52–92% yields), and alkynyl sulfones (62–85% yields), bearing aromatics with ether, halide and nitrile functional groups. These applications, whilst demonstrating significant utility for the modification of these two residues, involve additional steps to form the redox-active esters, thereby reducing the simplicity of the approach.

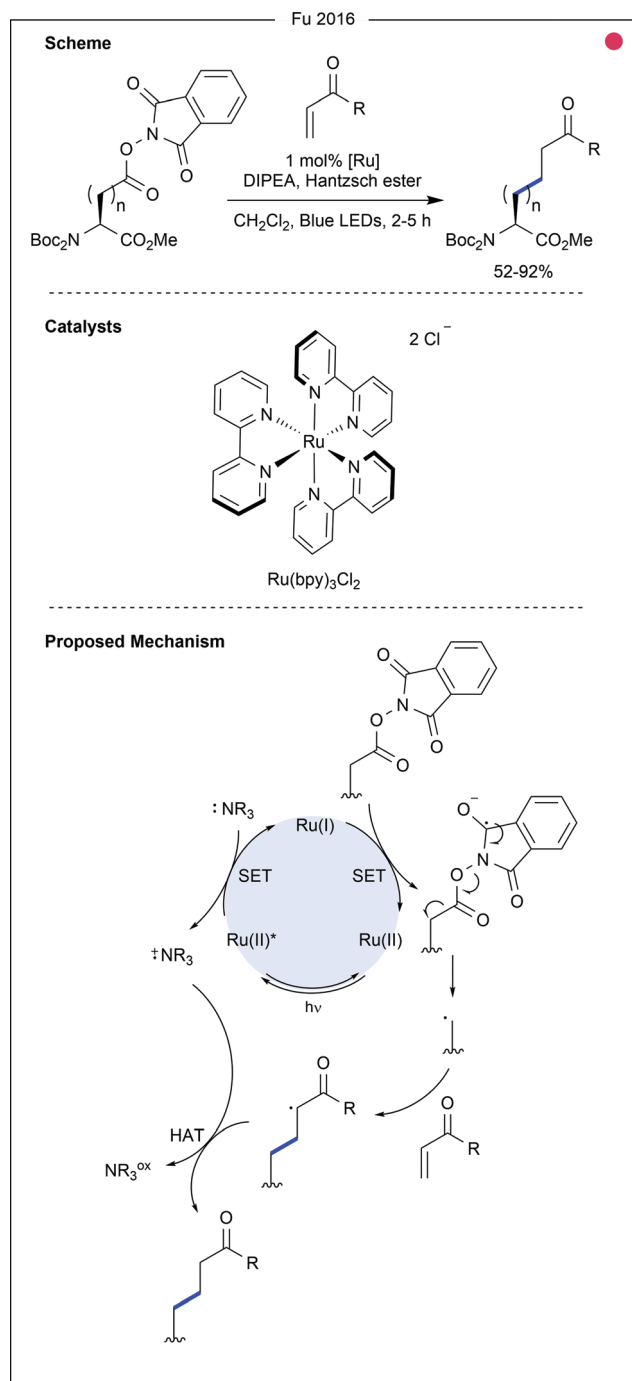


Fig. 13 Alkylation of aspartic acid ($n = 1$) and glutamic acid ($n = 2$) redox-active esters (RAEs). Single electron reduction of the RAE provides an alkyl radical which can undergo conjugate addition. Stoichiometric amine acts to afford catalyst turnover and subsequently enables hydrogen atom transfer (HAT) to generate the final product. Bpy = 2,2'-bipyridyl.

Aromatic amino acid side chains

The aromatic amino acids (see Fig. 14) offer great potential for photocatalysis as benzyl and heteroaryl rings have accessible redox potentials. In the context of bioconjugation, due to similarity across the aromatic amino acids, many traditional modification approaches fail to differentiate them. However,

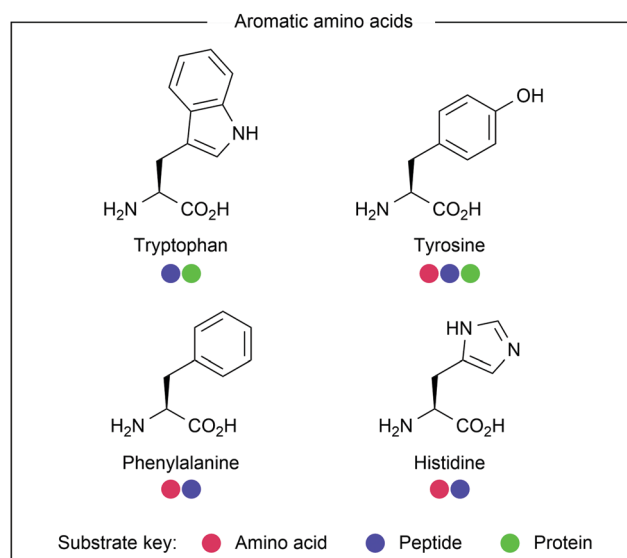


Fig. 14 Amino acids bearing aromatic side chain groups.

due to electronic differences, photocatalysis has been harnessed to enable residue-selective modification.

Tryptophan. Researchers at Merck have reported a chemo-selective photocatalytic modification of tryptophan (Fig. 15).³¹ Irradiation of an iridium photocatalyst led to oxidation of the indole nitrogen, greatly acidifying the benzylic methylene protons. Treatment with a phosphate base thus formed a stable benzylic, secondary radical which could react with a small range of Michael acceptors, followed by reduction and protonation. α,β -unsaturated esters, amides, ketones and sulfones were demonstrated to be compatible electrophiles in this system (59–96% conversion on a tripeptide substrate). An α,β -unsaturated amide bearing a pendant alkyne was also compatible with the chemistry, providing a potential handle for further modification with azides. The substrate scope and selectivity of the approach was then assessed on short peptides (3–6 AAs, 47–96% conversion to the modified tryptophan residue). Neither tyrosine nor phenylalanine showed any β -modification under the same conditions. However, when histidine was present in a peptide, competing reaction of the nucleophilic imidazole nitrogen with the electrophile was observed, reducing conversion at the tryptophan residue (17–33% of singly modified product). This histidine reactivity appeared to be protein-dependent, as it was not observed during modification of a larger peptide, glucagon (28 AAs, 45% conversion to solely tryptophan-modified product). Further evidence for protein-dependent selectivity was observed with modification of a lysine side chain amine in the native gastrointestinal peptide hormone GLP-1 (residues 7–36, 29 AAs). Amino acid nucleophilicities can be modulated by nearby residues, and peptide structure heavily influences side chain accessibility, both of which could be cause for the variable selectivity observed using this protocol. Additionally, where the C-terminus was left unprotected, some C-terminal modification of peptides was observed due to decarboxylation. However, the



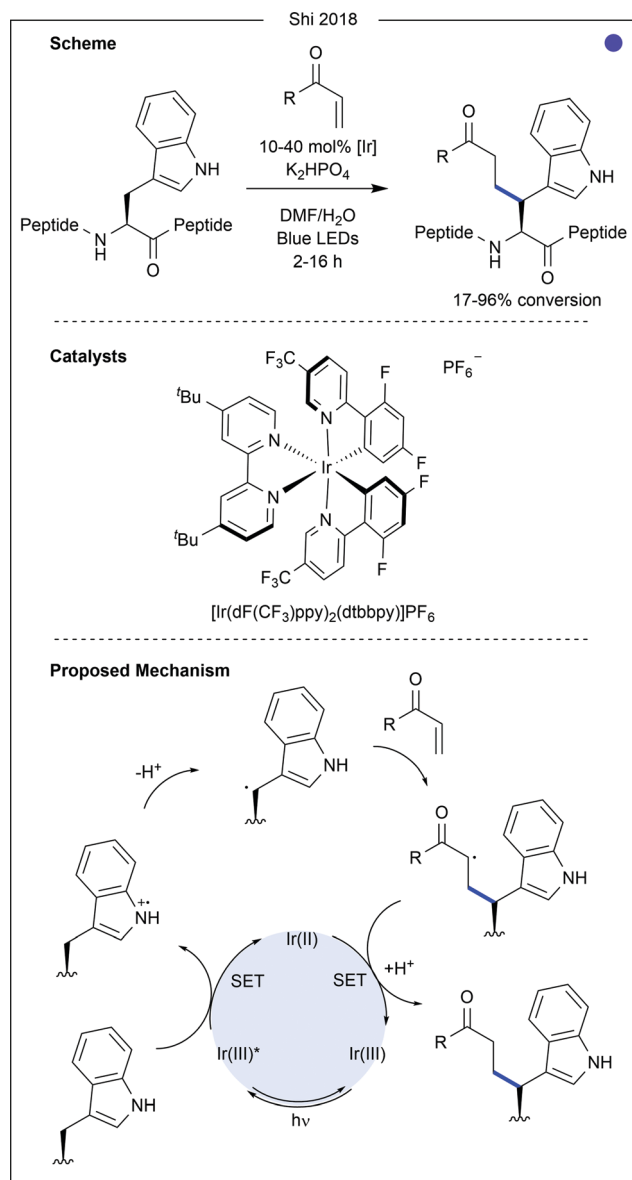


Fig. 15 Photocatalytic modification of tryptophan at the methylene position. Single electron oxidation of the tryptophan nitrogen activates the methylene carbon for deprotonation. The resultant radical species, a stable secondary benzylic radical, can undergo conjugate addition. Single electron reduction, again by the photocatalyst itself, coupled with protonation generates the final product. [Ir] = [Ir(dF(CF₃)ppy)₂(dtbbpy)]PF₆.

demonstration of tryptophan β -C(sp³)-H reactivity encourages further research in this area.

Contrastingly, Taylor and co-workers recently used photocatalysis to access modification of the C₂-position of tryptophan (Fig. 16).³² By employing *N*-carbamoylpyridinium salts, the resulting charge transfer complex between the pyridinium and tryptophan indole could be excited by irradiation with ultraviolet (UV-B) light. Single electron transfer from tryptophan to the pyridinium leads to homolytic cleavage of the N-N bond, generating a radical intermediate. As this intermediate is generated proximal to the oxidised tryptophan residue, combination of the two to produce the product is favourable. A range

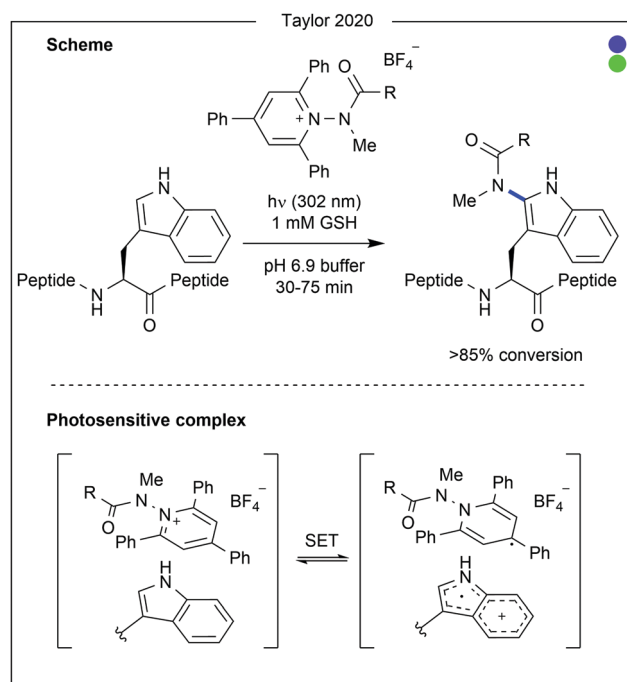


Fig. 16 Photo-induced modification of tryptophan at the C₂-position. The photosensitive complex undergoes single electron transfer when irradiated by ultraviolet light, leading to scission of the N-N bond in the Katritzky-type salt. Radical combination then generates the final product. GSH = glutathione.

of carbamoyl groups containing pendant functional groups such as ethers and carbamates, tags such as biotin, or reactive groups for further modification such as an alkyne could be transferred to tryptophan in this way. After demonstrating high reactivity on a short peptide (8 AAs, >90% conversion for all salts used), the chemistry was applied to larger peptide and protein substrates (1–14 kDa) achieving high (>85%) conversion in all cases. Notably, this protocol does not require any organic co-solvent, proceeding in aqueous buffer at near-neutral pH, and is fast, only requiring irradiation for 30–75 minutes to achieve high conversion.

Tyrosine. Tyrosine has been the target of a multitude of conjugation methods through ionic or cycloaddition pathways. A number of proteins contain tyrosine-tyrosine cross-links, formed through radical coupling of oxidised tyrosine residues. Indeed, work from the 1990s used a photocatalyst to artificially create these cross-links.³³ Modern methods of tyrosine modification using a photocatalyst have used a similar approach. Nakamura and Sato demonstrated that specific tyrosine residues located close to ligand binding sites could be selectively oxidised and labelled using a ligand-photocatalyst construct (Fig. 17).³⁴ By linking a ruthenium photocatalyst to a known ligand of the target protein, the photocatalyst was brought close to the protein surface. Through irradiation with visible light, the ruthenium catalyst preferentially oxidised proximal tyrosine residues. Addition of an aniline-based tyrosyl radical trapping (TRT) reagent produced modified tyrosine residues. The technique, which used biological pH (7.4) and short irradiation times (15 min),



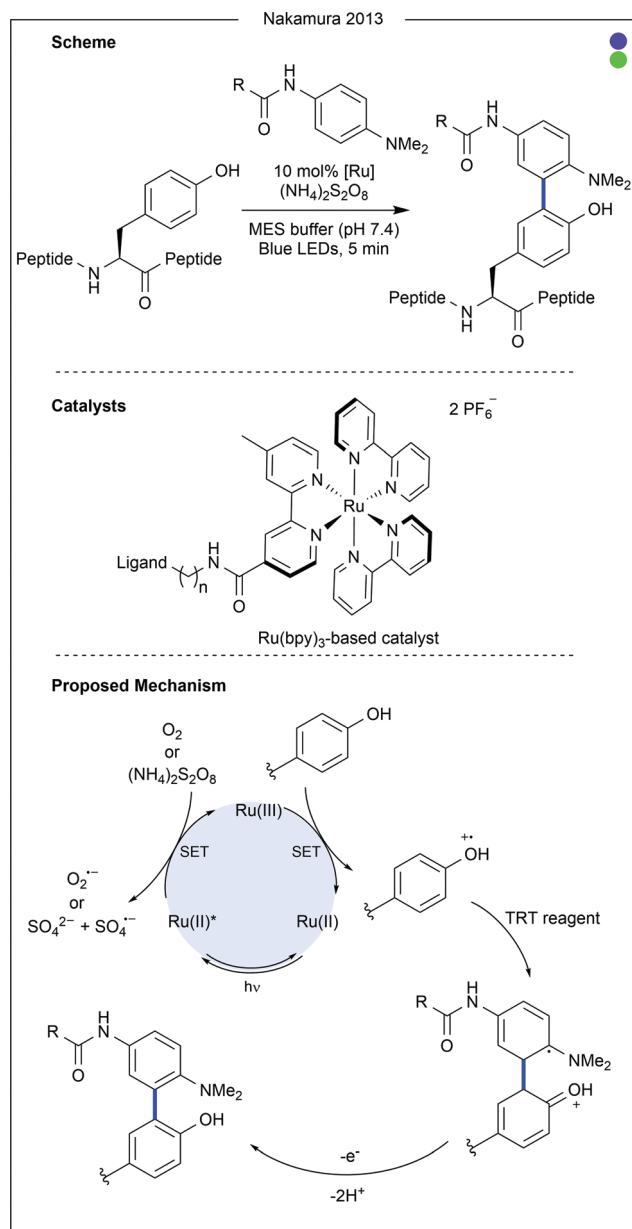


Fig. 17 Ligand-directed photocatalytic modification of tyrosine. By attaching a ruthenium photocatalyst to a known ligand, selective modification of proximal tyrosine residues on the target protein was conducted. After irradiation, the excited ruthenium catalyst was proposed to reduce either oxygen or ammonium persulfate. The resultant ruthenium(III) species could then oxidise nearby tyrosine residues. A substituted aniline, a tyrosyl radical trapping (TRT) reagent, could then attack the tyrosyl radical, which after further oxidation led to the bi-aryl cross-coupled product. MES = 2-morpholinoethanesulfonic acid. Bpy = 2,2'-bipyridyl.

was used to successfully distinguish between two proteins in solution, bovine serum albumin (BSA) and carbonic anhydrase (CA), selectively modifying the latter when benzene sulfonamide was used as the ligand. Using an undirected ruthenium photocatalyst ($\text{Ru}(\text{bpy})_3\text{Cl}_2$), both proteins underwent modification. The technique was also able to selectively modify CA in mouse erythrocyte lysate, and in intact cells. Both fluorophore- and

biotin-containing TRTs were used, demonstrating the varied utility of the approach.

Tyrosine-derived ethers have also been recently modified through photocatalytic oxidation of the tyrosine aryl ring. Work by Nicewicz and co-workers on novel $\text{S}_{\text{N}}\text{Ar}$ chemistry highlighted methoxybenzenes as suitable reagents following initial oxidation by an excited acridinium photocatalyst (Fig. 18).³⁵ The work demonstrated unique *ipso*-reactivity, whereby methanol was released after attack of a nucleophilic N-heterocycle. Among the substrate scope was tyrosine methyl ether, which was reacted with imidazole in moderate yield (38%). Tyrosine methyl ether was also reacted with a C- and N-terminally protected histidine residue to produce the coupled product, albeit in low yield (17%). This protocol makes use of unique tyrosine reactivity to enable the generation of unnatural amino acids. While initial results also

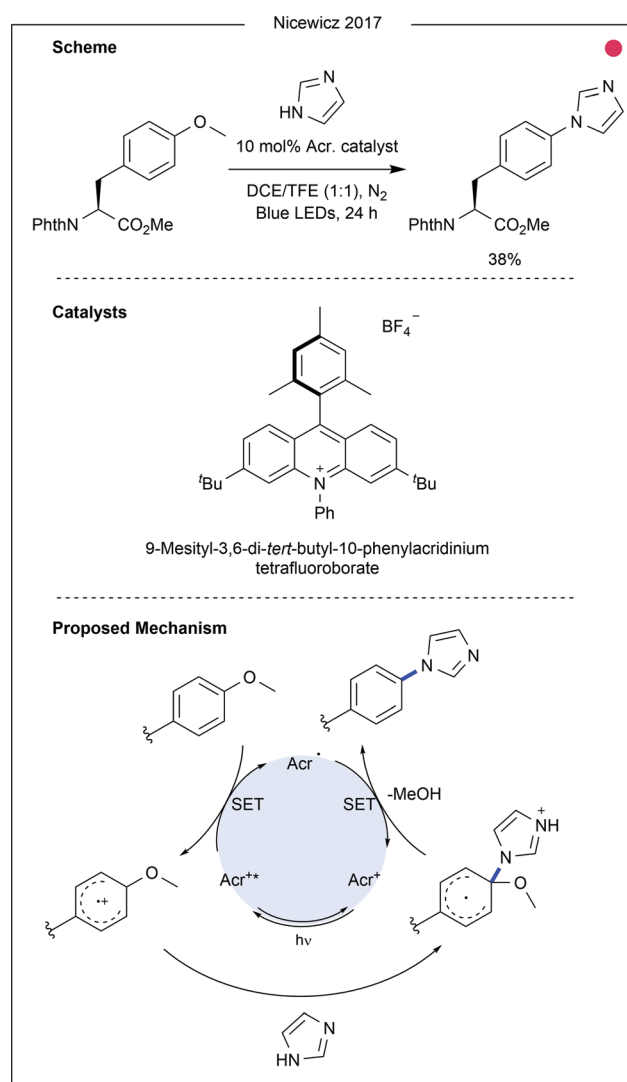


Fig. 18 Tyrosine methyl ether activated for $\text{S}_{\text{N}}\text{Ar}$ reactivity through the use of an acridinium photocatalyst. After oxidation of the tyrosine aryl ring by the photocatalyst, imidazole was shown to attack. Reduction occurred with loss of methanol to generate the product. Phth = phthaloyl. Acr. = 9-mesityl-3,6-di-tert-butyl-10-phenylacridinium tetrafluoroborate. DCE = dichloroethane. TFE = trifluoroethanol.



suggest that the protocol could be used for amino acid cross-coupling or peptide macrocyclisation, further optimisation would be required to make this a viable tool in this respect.

Phenylalanine. Leonori and co-workers have recently reported the development of $C(sp^2)$ -N cross-coupling of secondary amines with largely unfunctionalised aromatics (Fig. 19).³⁶ *In situ*

conversion of various secondary amines into the corresponding *N*-chloroamines using *N*-chlorosuccinimide (NCS) was followed by reduction by $Ru(bpy)_3$ under visible light irradiation to form secondary aminium radicals. These were observed to attack a range of aromatic compounds to form an aryl radical, which after single-electron oxidation by the ground-state photocatalyst lost a proton to regain aromaticity. This general procedure, using two equivalents of the aromatic coupling partner, was demonstrated upon a wide range of substrates in moderate to high yields (21–99%, average yield 66%). Together, the aryl and amine substrate scopes contained reagents bearing alkyl, aryl and halide groups, alcohols, protected amines, esters and amides. The technique was then explored for the late-stage functionalisation of complex and biologically relevant molecules. Among these, *N*-acetyl phenylalanine ethyl ester was successfully reacted to form both an azetidine- and an azide-containing product (98% and 65% respective yields). These reactions were also performed on a tetrapeptide (63% and 53%), demonstrating the potential applicability of the chemistry to biological targets. Further work is mandated to assess the selectivity of the chemistry over other native residues, in particular tyrosine and tryptophan. Irrespective of this, the protocol has opened up phenylalanine as a ‘blank slate’ for the generation of unnatural aromatic amino acids.

Histidine. The nucleophilicity of histidine’s imidazole is difficult to exploit in bioconjugation due to competition from other, more nucleophilic residues such as cysteine or lysine. By using a photoredox strategy, Wang, Chen and co-workers reversed the reactivity of the imidazole ring by enabling Minisci-like attack (Fig. 20).³⁷ It is believed that homolytic cleavage of the C–C bond in a Hantzsch ester, dihydropyridine (DHP) alkylating reagent occurs under visible light irradiation. The alkyl radical formed can then attack a protonated histidine imidazole ring. SET to a DHP intermediate can then return the alkylated histidine residue in its protonated form. Intriguingly, the DHP reagent acts as both alkyl-radical-source and terminal oxidant in the process. It is proposed that after homolytic cleavage of the C–C bond, the resulting DHP radical is oxidised by a further molecule of unreacted DHP (present in excess). It was noted that air must be excluded from the reaction, as unwanted oxidation of other amino acids in the peptide upon irradiation with visible light was observed under aerobic conditions. Both secondary and tertiary alkyl groups were transferable from the DHP reagents, including those functionalised with ethers, alcohols, acids, azides, alkynes, esters or amides. In contrast, primary alkyl groups could not be conjugated. The protocol involved multiple resubmissions of the substrate to the reaction conditions to afford sufficient conversion, but the methodology could be successfully applied to a wide range of peptides (up to 76 AAs) in varying isolated yields (20–69%). Importantly, modification was selective over all other aromatic amino acid side chains, and only reduced cysteine was incompatible with the conditions. Subsequent ‘click’ chemistry was performed on a peptide containing an azide-modified histidine to install a fluorophore payload,

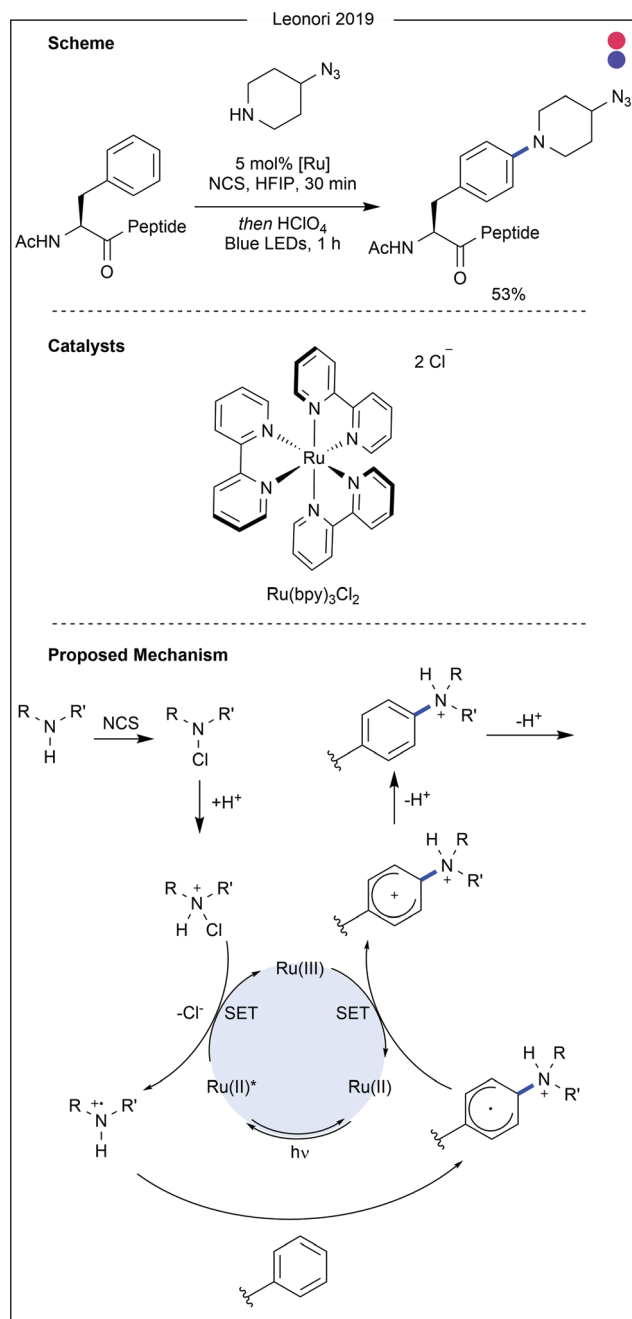


Fig. 19 Recent photocatalytic modification of phenylalanine. An excited ruthenium catalyst was shown to reduce *N*-chloroammonium reagents to produce oxidised secondary amines. These species were sufficiently reactive to undergo radical attack from the aryl side chain of phenylalanine. After single electron oxidation by the photocatalyst, double deprotonation formed the aminated amino acid in moderate to high yields. Bpy = 2,2'-bipyridyl. NCS = *N*-chlorosuccinimide. HFIP = hexafluoroisopropanol.



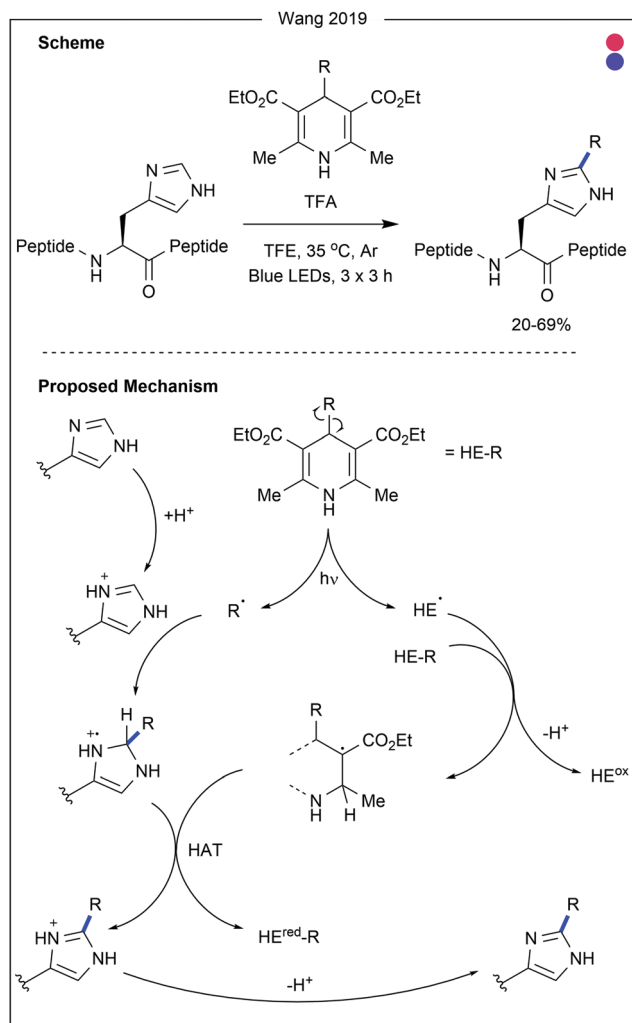


Fig. 20 Photomediated histidine modification using alkylated Hantzsch esters (HE-R). After photomediated homolytic cleavage of the HE-R bond, the alkyl radical can undergo Minisci-type addition to protonated histidine residues. Hydrogen atom transfer (HAT) to a later Hantzsch ester derivative generates the final product. TFA = trifluoroacetic acid. TFE = trifluoroethanol.

demonstrating the utility of the approach. The C–H functionalisation protocol conserves the N-unsubstituted nature of native histidine residues, thereby limiting the impact of the modification on their biological function.

Aliphatic amino acid side chains

Aliphatic residues (see Fig. 21) have remained underexploited in the realm of amino acid and peptide modification, as their inherently inert side chains preclude modification using traditional methods. Palladium-catalysed C–H activation chemistry has been used to target these residues, but many procedures require biologically-incompatible conditions, most notably exceedingly high temperatures. In contrast, the milder conditions used in photocatalytic protocols are more suitable for bioconjugation purposes.

Leucine and isoleucine. Recently, Leonori and co-workers presented a novel method for the functionalisation of C–H

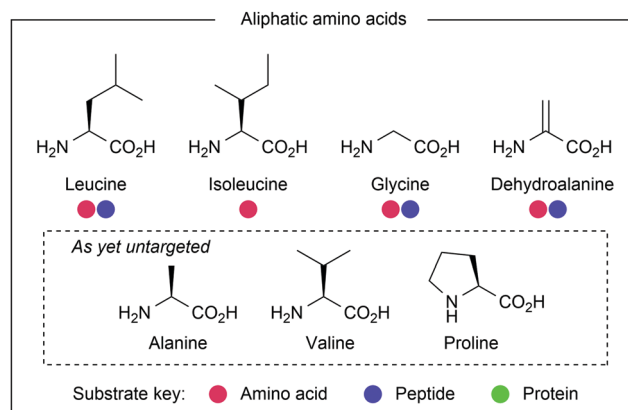


Fig. 21 Amino acids with aliphatic side chains.

bonds distal from amides and protected amines (Fig. 22).³⁸ This protocol uses an *N*-oxy species as a precursor to aminyl or amidyl radicals. It is proposed that the N–O bond is cleaved upon treatment with the excited photocatalyst, generating an N-centred amide radical. This is capable of undergoing 1,5 hydrogen atom transfer (HAT) with the δ -carbon to the amide nitrogen. This nucleophilic C-centred radical was trapped with several SOMO-philic, including Selectfluor, *N*-chlorosuccinimide, a sulfur-based electrophile and 2-iodoxybenzoic acid (IBX) reagents. Application of the method was demonstrated on Boc-, Phth- and Cbz-protected leucine and isoleucine, which had been synthesised with the precursor amide *N*-oxide on the C-terminus. The procedure was used successfully to install either a fluoride (using Selectfluor, 39–56%) or an alkyne (using an alkynyl IBX reagent, 47–65%) at the amino acid γ -carbon. It was noted that N-terminus protection was required as lower yield (using Selectfluor, 13%) was obtained when unprotected. The presence of the C-terminal *N*-oxy amide as a directing group removes potential selectivity issues, but consequently allows modification only of the C-terminus amino acid if applied to a peptide substrate, while leaving the peptide with a C-terminal *N*-methyl amide.

Glycine. Recently, Xu, Wang, Chang and co-workers demonstrated α -C(sp³)–H activation in N-terminally *para*-methoxyphenyl (PMP) or phenyl protected glycine residues *via* photocatalysis (Fig. 23).³⁹ Instead of using a separate photosensitiser, Katritzky salts were used, which can be formed from the corresponding amines, to generate an electron donor–acceptor (EDA) complex with the protected glycine N-terminus. Upon excitation of the complex, the peptide amine is oxidised by the Katritzky salt which, in turn, undergoes homolytic cleavage of the *N*-alkyl bond to produce a carbon-centred radical. Rearrangement of the peptide N-terminus radical onto the α -carbon then allows radical–radical combination with the locally generated alkyl radical. Only secondary alkyl radicals underwent recombination with the glycine residue, while benzylic radicals and primary radicals showed no reactivity. Katritzky salts with R groups containing alkyl, alcohol, protected amine and ether functionality were all adequate coupling partners. An ester- and an aryl fluoride-containing Katritzky salt were unsuccessful in



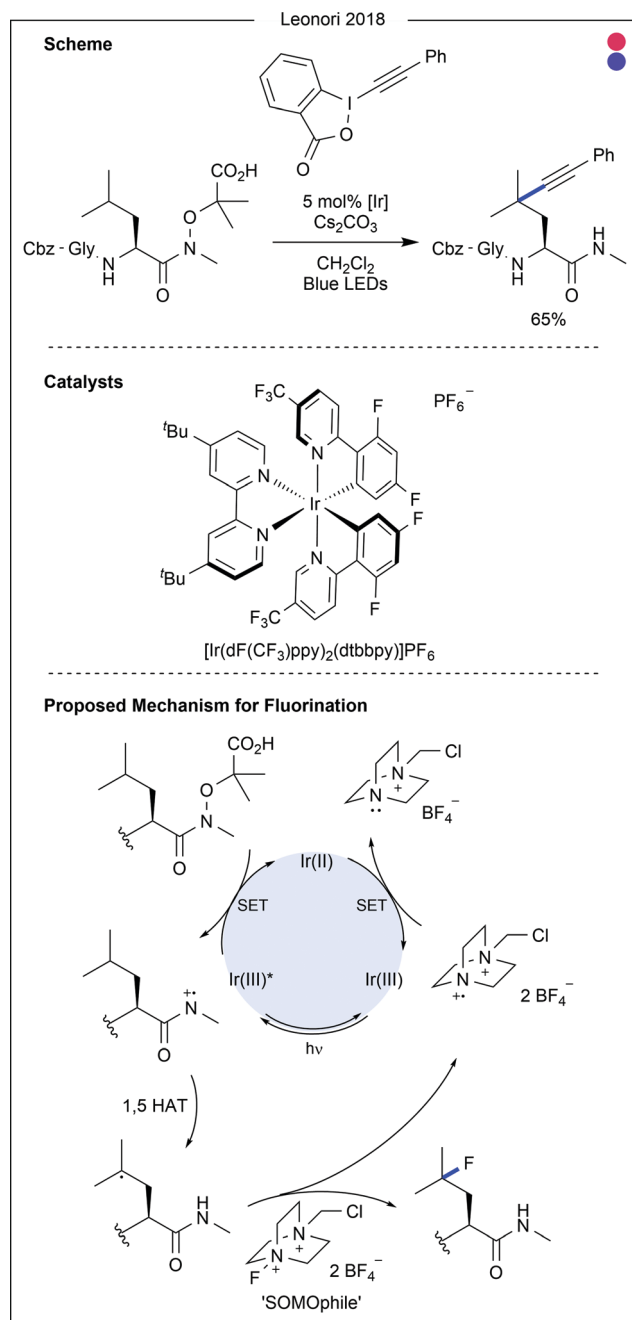


Fig. 22 Photocatalytic modification of leucine and isoleucine using 1,5 hydrogen atom transfer (HAT). Using a photolabile group, after single electron oxidation an amidyl radical could be formed. Via a 6-membered cyclic transition state, hydrogen atom transfer from the side chain to the amide nitrogen could be achieved, generating an alkyl radical. Using radical reaction partners such as alkynyl hypervalent iodine or *N*-fluoro species, modified amino acids could be generated. Gly = glycine. Dtbbpy = 4,4'-di-*tert*-butyl-2,2'-bipyridyl.

modifying the residue. For successful modifications, including addition of a steroid to the glycine residue, yields were generally high (60–90%). High yields were also maintained when the conditions were applied to peptides ranging from 2 to 6 amino acids, with selective modification of the N-terminally capped glycine residue. Aliphatic residues, methionine and protected

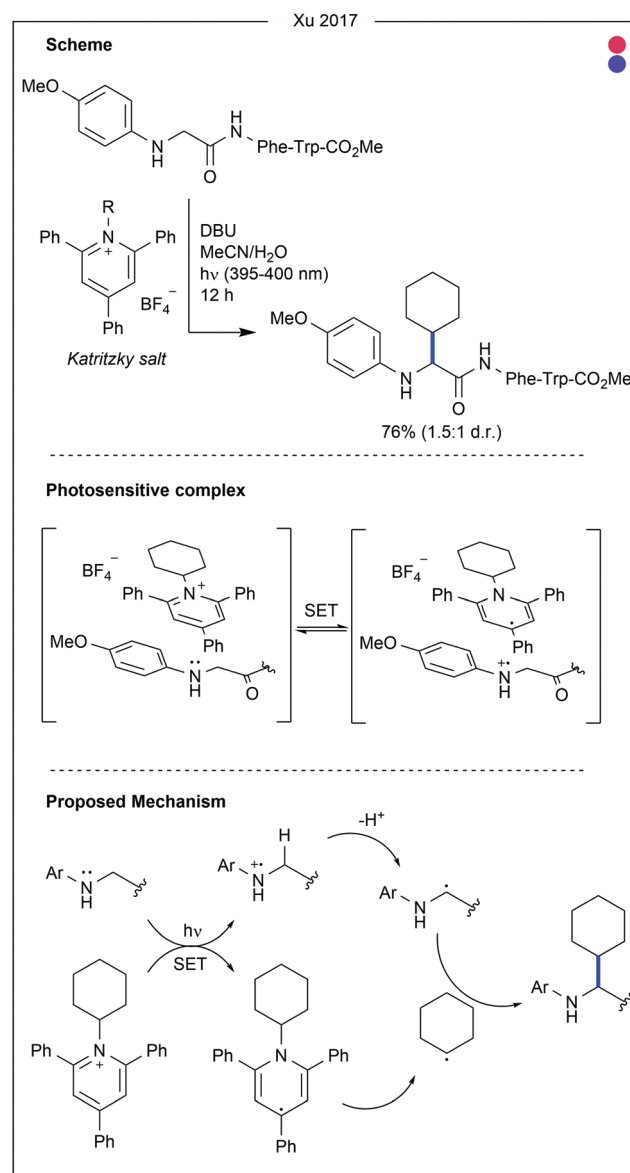


Fig. 23 Photo-enabled modification of glycine residues using Katritzky salts. An *N*-aryl glycine at the N-terminus of a short peptide sequence was shown to form an electron donor–acceptor (EDA) complex with Katritzky salts. Upon irradiation, single electron transfer produces both a glycine amino radical and a pyridinium radical. The pyridinium radical is able to collapse to generate the pyridine, with loss of an alkyl radical (R^\bullet), while deprotonation with base moves the glycine radical to the α -carbon. Radical–radical combination then affords the modified product. Phe = phenylalanine. Trp = tryptophan.

arginine, serine, threonine and aspartic acid were all compatible with the reaction conditions. Significantly, this protocol also avoided modification of an indole, found in tryptophan, a moiety which has been shown to form EDA complexes with Katritzky salts.⁴⁰ It was hypothesised that the process could be used to generate unnatural amino acids from glycine, which could then, after removal of the N-terminal protecting group, be used in peptide synthesis.

Dehydroalanine. An amino acid often found in naturally occurring antimicrobial peptides,⁴¹ dehydroalanine (Dha) has



the opposite reactivity to most other amino acids; it can be modified relatively selectively through the use of nucleophilic reagents. Due to its low natural abundance, several strategies have been developed that enable facile synthesis of Dha from other amino acids, most commonly cysteine.⁴¹

Dha has been used to generate unnatural amino acids, either independently or within a peptide chain (Fig. 24). Jui and co-workers used photocatalysis to generate a range of pyridyl radicals which performed radical conjugate addition into protected Dha, forming heteroaryl-containing amino acids after further reduction and protonation.⁴² Following single electron reduction of the halo-pyridine, mesolytic cleavage of the C–X bond occurs, releasing halide and forming a C(sp²)-centred radical. Hantzsch ester was used to provide terminal single electron reduction after conjugate addition. 2-, 3- and 4-halopyridines were successful coupling partners for *N*-Boc Dha methyl ester (34–97% yields). Alcohols, halides, amines and ethers were all compatible heteroaryl functional groups. Finally, Karady–Beckwith alkene, a chiral oxazolidinone derivative of Dha, was used to control the chirality at the α -carbon, inducing diastereoselective protonation of the enolate that is formed upon final reduction. Good yields (57–80%) were obtained using a small selection of heteroaromatic halides. Facile deprotection with hydrochloric acid yielded an example enantioenriched pyridylalanine product (97% ee).

A similar methodology was then reported using tertiary amines to generate α -amino radicals that also performed radical conjugate additions into Dha and the Karady–Beckwith oxazolidinone derivative.⁴³ Single electron oxidation of a tertiary amine produces an amino radical cation, greatly acidifying the α -protons. Deprotonation then generates the required α -amino radical. Use of aryl and alkyl tertiary amines, containing aldehydes, halides, alkenes, acids and esters, as well as complex bioactive tertiary amines, generated addition products in good yields (62–91%). Modification using a propargylamine derivative was less successful (25% yield). Modification of Dha was then attempted within short peptide sequences (3–9 AAs), providing the conjugate products in good isolated yields (41–86%).

More recently, Roelfes and Molander have independently reported using trifluoroborate salts as radical precursors for Dha modification.^{44,45} When exposed to photocatalytic conditions, C–B cleavage within the salt generated a carbon-centred radical which could undergo radical conjugate addition. A variety of unnatural amino acids were generated, including fluorinated variants.

Dha was also recently used as a substrate to demonstrate work from Leonori and co-workers on α -amino-radical-promoted halogen atom transfer (XAT).⁴⁶ The process, involving 4CzIPN as the photocatalyst, generates the α -radical of a tertiary amine through SET and deprotonation. This radical demonstrates efficient XAT from alkyl and aryl halides, producing the corresponding alkyl and aryl radicals. Assessing the substrate scope for the alkene coupling partners, the group generated 16 unnatural amino acids through coupling with Boc-protected Dha methyl ester, in moderate to excellent yields (22–99%). Halides containing amine, ether, alkyne, silane and

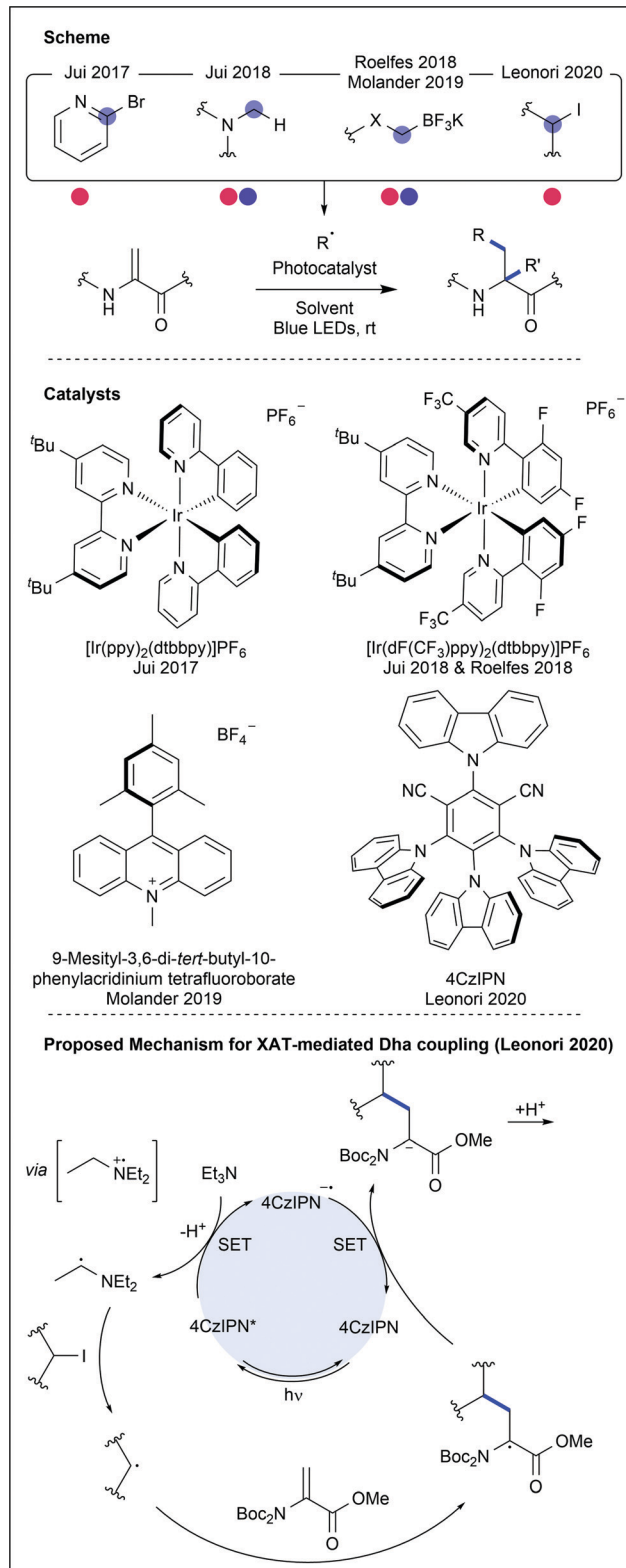


Fig. 24 Recent developments in the modification of dehydroalanine (Dha). Photocatalytic methods for Dha modification generally involve alkyl or aryl carbon radicals, which undergo conjugate addition to the electron-deficient alkene. A variety of catalysts have been used, due to differing methods of radical formation. Most recently, alkyl and aryl halides have been shown to be suitable radical precursors, thereby opening up a wide, commercially available pool of potential substrates.



boronic ester functionalities were all suitable coupling partners. The inclusion of both primary and secondary halides as coupling partners overcomes significant obstacles of other procedures, which often only enable the coupling of secondary carbon radicals

to amino acids. It is, however, important to note that no control over stereochemistry could be achieved.

Reactivity at terminal amino acid residues

The C-terminal carboxylate of amino acids and peptides presents an opportunity to apply the plethora of photocatalysed decarboxylation methods to amino acid modification. Indeed, many protocols have been demonstrated to work on amino acids or their derivatives.⁴⁷ However, modification of the C-terminal carboxylate of a lone amino acid precludes any subsequent products from being used in peptide synthesis. As these protocols would also not allow modification of a native peptide sequence, except at the C-terminus, most are beyond the scope of this review. There have, however, been a number of recent reports of C-terminal modification of peptides (Fig. 25).

MacMillan and co-workers described the selective C-terminal decarboxylation of native peptides using flavin-based photocatalysts.⁴⁸ Due to the significant difference in reduction potentials between the C-terminus carboxylate and those carboxylates found on amino acid side chains within the peptides, selective modification of the C-terminus could be achieved under oxidative conditions. C-Terminal decarboxylation results in a primary (if the amino acid is glycine) or secondary (if any other amino acid is used) α -amino carbon radical, which was trapped with α,β -unsaturated esters. Using a tetrapeptide containing a variable N-terminal residue, most polar and aliphatic amino acids were shown to be compatible with the reaction conditions. Tyrosine and histidine, however, significantly decreased conversion when present in the peptide. Eight larger peptides, ranging from 8–58 amino acids in length, were then successfully modified with moderate to good conversions (31–66%). While most of the substrates were trapped with an α,β -unsaturated diester, the potential for this chemistry to

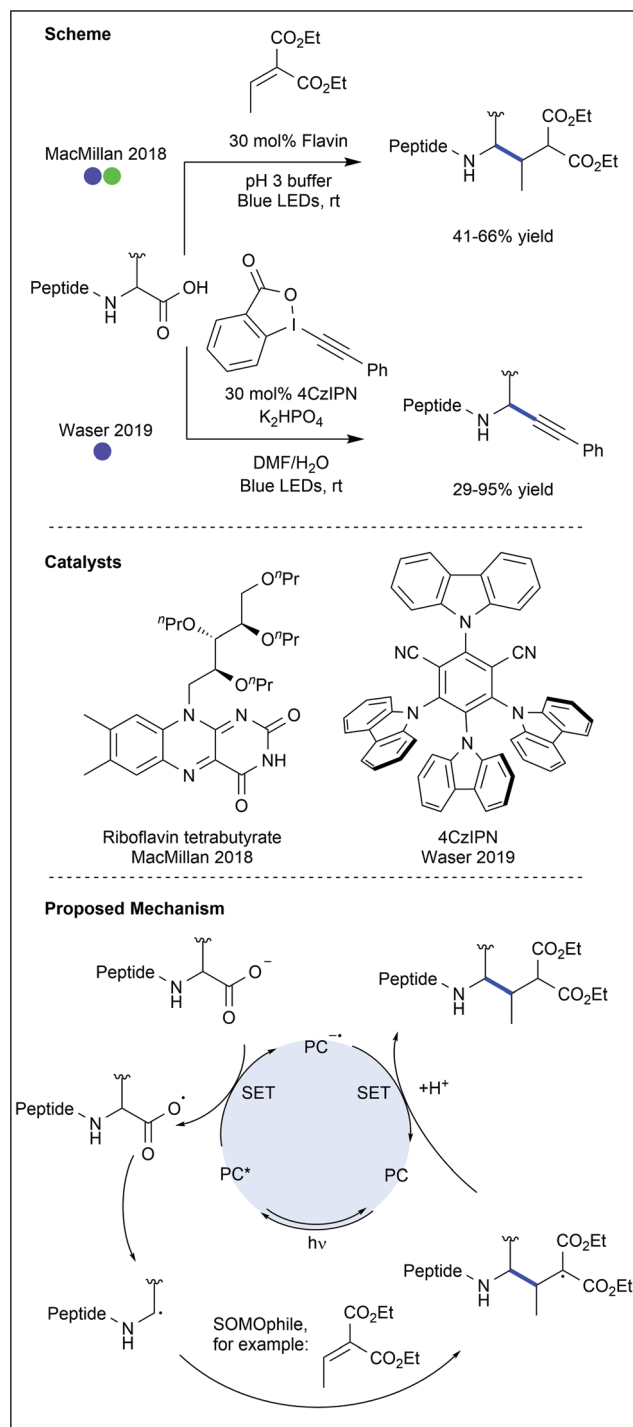


Fig. 25 Decarboxylative C-terminal modification of short peptides. Modification strategies showed selectivity for the C-terminal carboxylate over internal carboxylate residues (glutamic acid and aspartic acid). Decarboxylation generates the alkyl radical which can react with SOMOphiles, such as α,β -unsaturated carbonyls or hypervalent iodine reagents. PC = photocatalyst.

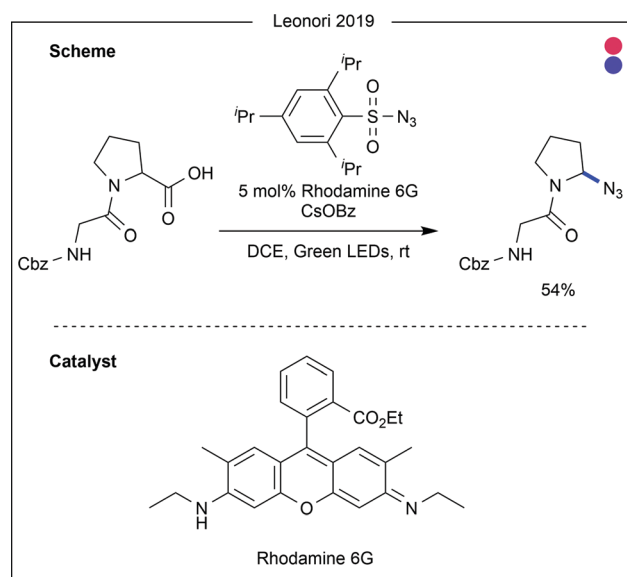


Fig. 26 Developments in decarboxylative azidation, shown to be applicable to proline and peptide substrates containing a C-terminal proline residue. DCE = dichloroethane.

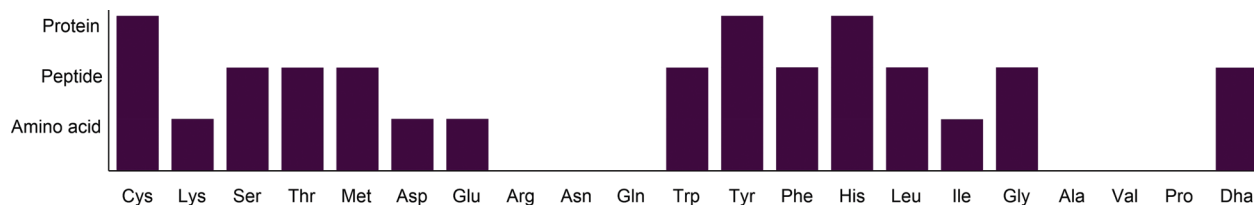


Fig. 27 Overview of the complexity of substrates to which photocatalytic amino acid modification has been applied. Proteins are defined here as peptides containing over 50 amino acids.

enable selective installation of an alkyne group on the A-chain C-terminus of human insulin was also demonstrated (41% conversion).

A decarboxylative alkynylation protocol has been developed by Waser and co-workers.⁴⁹ Using donor-acceptor cyanoarene (DACA) photocatalysts and ethynyl benziodoxolone (EBX) reagents, selective modification of the C-terminus was achieved on di-, tetra- and hexapeptides (17–95% yields). While amino acids such as cysteine, tyrosine and tryptophan led to reduced yields when incorporated in the peptide sequence, protecting groups, or the use of a less oxidising catalyst were shown to revive reactivity. Furthermore, selectivity for the C-terminal carboxylate was observed in the presence of glutamic acid and aspartic acid. While other C-terminal decarboxylative alkynylations have been developed, most require chemical pre-activation of the carboxylate. This procedure brings alkynylation in line with arylation, reduction, cyanation and Giese-type coupling, all of which have previously been applied to free C-terminal carboxylates (see references within ref. 49).

Recent advances in decarboxylation chemistry have enabled azidation of amino acid C-termini, as reported by Leonori and co-workers (Fig. 26).⁵⁰ Using a rhodamine-based photocatalyst, proline residues were decarboxylated to form the secondary alkyl radical, which was then trapped with an azide transfer reagent. The reaction proceeded in excellent yield when applied to *N*-Boc proline (87%), but the yields dropped when the procedure was attempted on dipeptides (54–68%). Azidation of proteins would allow for further modification through ‘click’ chemistry, but further progress on this azidation reaction is warranted for efficient protein modification.

4. Conclusions and outlook

Amino acid modification has been a key interest in recent years and has been applied across a wide variety of fields. Photocatalysis has offered new synthetic routes to amino acid modification. While lysine and cysteine have been widely targeted in bioconjugation through ionic methods, photocatalysis provides complementary methods to these, enabling modification with higher chemoselectivity. In addition to this, photocatalysis enables modification of amino acids for which there are either few or no other reported methods. An overview of the applicability of current photocatalytic methods for amino acid modification is given in Fig. 27.

One area in which photocatalysis is a great advance on previous methods is C–H functionalisation. Whereas most previous protocols require high temperature, which to a protein would cause significant decomposition, photocatalytic processes are rarely performed above room temperature, leading to greater stability of the substrates. This has notably been applied to aliphatic residues such as leucine, isoleucine and glycine, which have been modified with simple electrophiles under ambient conditions. The emergence of C–H functionalisation protocols compatible with peptide substrates and capable of modifying aliphatic side chains shows great promise for the future of amino acid modification.

Despite great advances in the field of amino acid modification, significant challenges remain. Many of the exciting new protocols have been shown to be less effective when applied to larger peptide substrates. While amino acid modification is itself a significant goal, for these to be valuable tools beyond single residue modification, progress must be made in adapting and improving the procedures. In addition, many of the procedures detailed in recent years involve organic solvents which are themselves not suitable for solvating larger peptides or proteins. For these protocols to be more widely used, developments should be made to enable the transformations in aqueous solvent, preferably buffered at physiological pH.

The number of articles detailing amino acids among their substrate scopes has risen exponentially over the last two decades. As amino acid modification remains a rich area for discovery, this is unlikely to fall soon. Photocatalysis shows great potential in expanding the toolbox of methods for bioconjugation and amino acid modification.

Conflicts of interest

There are no conflicts to declare.

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