


 Cite this: *Chem. Commun.*, 2021, 57, 1137

 Received 23rd November 2020,  
Accepted 18th December 2020

DOI: 10.1039/d0cc07675a

rsc.li/chemcomm

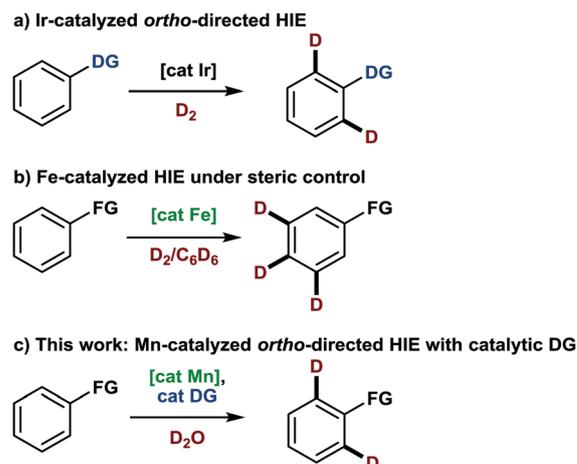
# Manganese-catalyzed selective C–H activation and deuteration by means of a catalytic transient directing group strategy†

 Sara Kopf, , Helfried Neumann \* and Matthias Beller \*

**A novel manganese-catalyzed C–H activation methodology for selective hydrogen isotope exchange of benzaldehydes is presented. Using D<sub>2</sub>O as a cheap and convenient source of deuterium, the reaction proceeds with excellent functional group tolerance. High *ortho*-selectivity is achieved in the presence of catalytic amounts of specific amines, which *in situ* form a transient directing group.**

Not only since the first deuterated drug was accepted by the U.S. Food and Drug Administration (FDA) in 2017, an increasing interest in the development of methodologies for the selective late-stage incorporation of deuterium into organic molecules has manifested itself within the field of organic synthesis.<sup>1</sup> As a result, the current toolbox for labeling reactions comprises a number of established techniques for diverse substrates. As an example, deuteration of arenes predominantly relies on iridium-catalyzed hydrogen isotope exchange (HIE) (Scheme 1a). Here, the most important representative, the NHC-ligated Kerr catalyst, affects deuteration under deuterium gas atmosphere *ortho* to a number of pre-installed directing groups, including N-heterocycles, carbonyl groups and sulfur-containing functional groups.<sup>2</sup> Most recently, base metal-catalyzed methodologies have expanded the field by presenting new selectivity concepts based on steric and/or electronic control. In this context, the deuteration of the least sterically hindered positions in arenes and heteroarenes has been achieved under catalysis with iron pincer complexes (Scheme 1b),<sup>3</sup> whereas a diamine nickel complex preferentially enabled HIE on electron-deficient positions of pyridine derivatives.<sup>4</sup> However, the precision of a directed approach has so far been unmet by alternative strategies. In this respect, the development of metal catalyzed directed HIE methodologies,<sup>5</sup> combining inexpensive and available metals with predictable selectivity, is desirable.

Among the different earth-abundant 3d metals, manganese is a most promising candidate, having proven its efficacy for catalytic directed C(sp<sup>2</sup>)-H functionalization reactions by enabling a plethora of transformations.<sup>6</sup> Moreover, although manganese-catalyzed deuteration of arenes has only been scarcely investigated yet,<sup>7</sup> some mechanistic investigations already hint at the potential success of such an endeavor.<sup>8</sup> However, despite these advances, manganese-catalyzed C–H activation is still limited to substrates comprising nitrogen-containing directing groups<sup>9</sup> and thus often demands additional synthetic steps. For instance, a recently disclosed allylation reaction of arenes required the pre-formation of an imine as a directing group, which subsequently needed to be hydrolyzed in a separate step.<sup>10</sup> A methodology that allows this imine formation to occur *in situ* with a catalytic amount of amine and thus reduces time and waste would certainly be more appealing. Interestingly, such a transient directing group strategy has been utilized in noble metal catalysis,<sup>11</sup> but has thus far not been applied to either manganese-catalyzed C–H activation or directed HIE. We believed that both fields could be merged thus report herein on the first

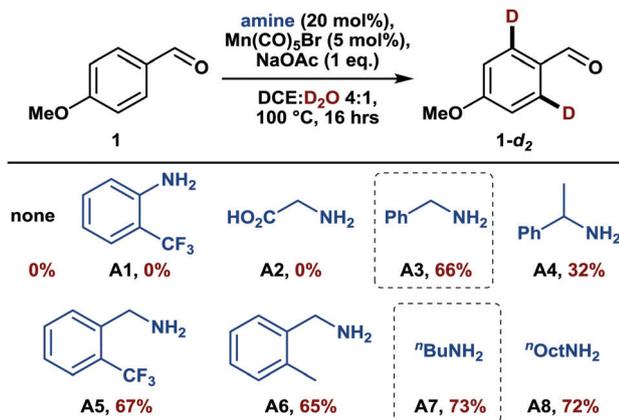


Scheme 1 Selectivity control of deuteration methodologies.

 Leibniz-Institut für Katalyse, Albert-Einstein-Straße 29a, Rostock 18059, Germany.  
E-mail: matthias.beller@catalysis.de

† Electronic supplementary information (ESI) available. See DOI: 10.1039/d0cc07675a





**Scheme 2** Mn-catalyzed *ortho*-deuteration of *p*-anisaldehyde in the presence of catalytic amine additives for the *in situ* formation of transient directing groups. Scale: 0.5 mmol **1**. Deuterium incorporation was determined by  $^1\text{H}$  NMR through the decrease of the doublet at 7.81 ppm while referencing on the protons in *meta* position.

manganese-catalyzed *ortho*-selective C–H deuteration of aromatic aldehydes through transient formation of a catalytic directing group (Scheme 1c). We started our investigations by studying the deuteration of *para*-anisaldehyde under typical conditions for base-assisted manganese-catalyzed C–H activation. More specifically, we used 5 mol%<sup>12</sup> manganese pentacarbonyl bromide along with  $\text{D}_2\text{O}$  as the cheapest source of deuterium in the presence of catalytic amounts of different amines (Scheme 2).

Firstly, in the absence of amine additives, no deuteration was observed, confirming that the aldehyde moiety is not a feasible directing group for such C–H activation. Furthermore, established transient directing groups such as anilines and glycine<sup>11</sup> could not afford any deuteration either. Surprisingly, the addition of 20 mol% of benzylamine to the reaction mixture enabled selective deuteration in the *ortho* positions of the aldehyde with a deuterium incorporation of 66% (**A3**). The use of sterically more hindered 1-phenylethylamine led to a drop in deuterium incorporation whereas substituents on the aromatic ring did not have a major influence on the reactivity (**A5** and **A6**). A slightly increased deuteration was observed with aliphatic amines such as *n*-butylamine (**A7**) so that both benzylic and aliphatic amines can be considered efficient catalytic additives for transient directing group formation.

Notably, the deuterium incorporation can be simply increased by re-submitting the isolated, partially deuterated compound to the reaction conditions (Table 1, entry 1). Interestingly, the reaction does not benefit from elevated temperature (120 °C) but works equally well at 80 °C (Table 1, entries 2–4). Finally, in the absence of a manganese catalyst, no deuteration took place (Table 1, entry 5).

Having gained insights into the influence of various parameters on the deuteration efficiency, we moved on to explore the substrate scope of this novel deuteration methodology. For this purpose, we conducted reactions at a temperature of 100 °C to ensure that less reactive substrates are also deuterated efficiently. To our delight, a variety of benzaldehyde derivatives

**Table 1** Mn-catalyzed *ortho*-deuteration of *p*-anisaldehyde under various conditions

Entry	Variation	D [%]
1	<b>1-d<sub>2</sub></b> (73% D) instead of <b>1</b>	88
2	120 °C	79
3	80 °C	76
4	60 °C	49
5	No [Mn]	0

Scale: 0.5 mmol **1**. Deuterium incorporation was determined by  $^1\text{H}$  NMR through the decrease of the doublet at 7.81 ppm while referencing on the protons in *meta* position.

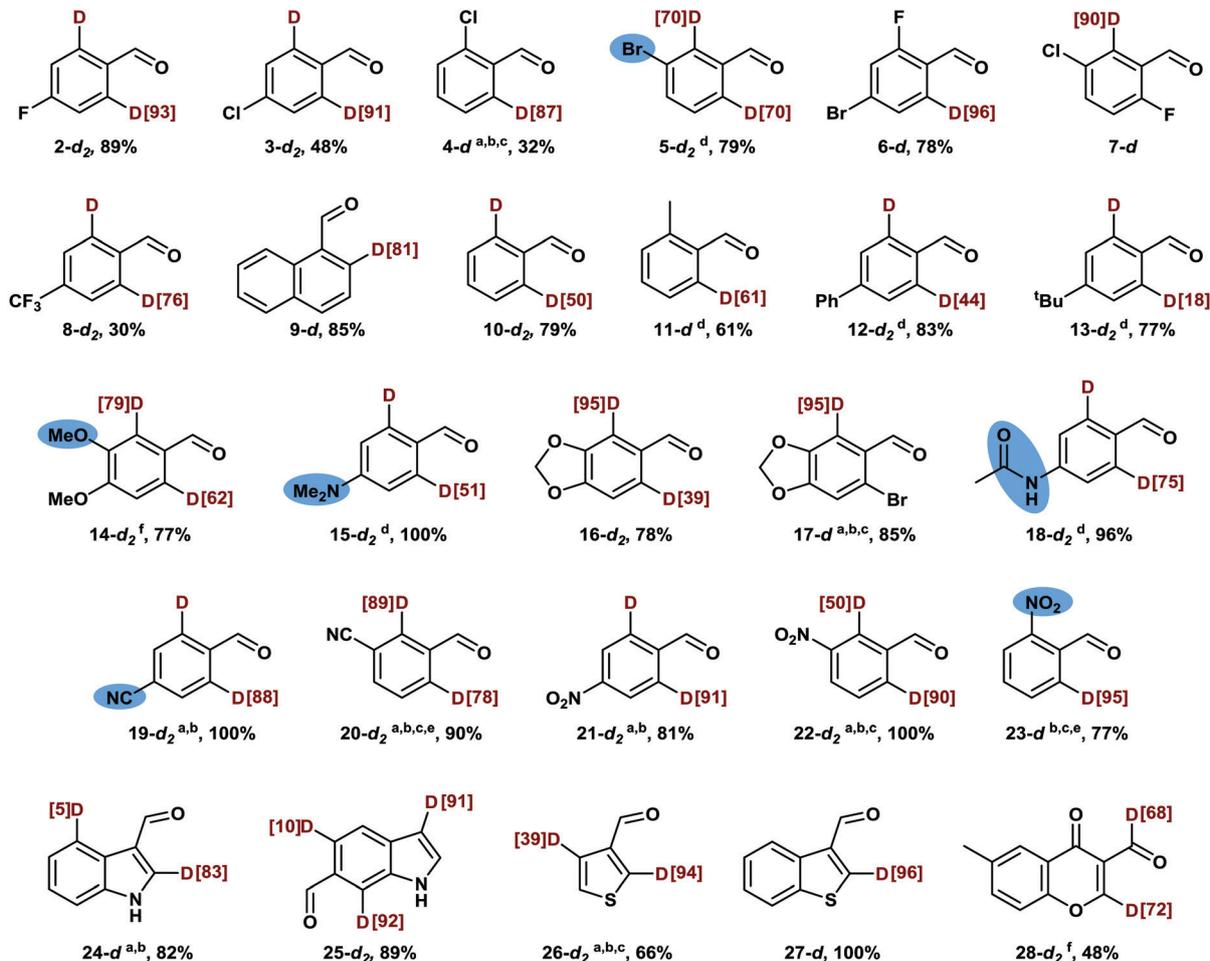
proved to be amenable substrates, being selectively deuterated in *ortho* position without any side product formation (Scheme 3). In this context, halogenated benzaldehydes proved to be particularly viable substrates and did not undergo competing deuterodehalogenation (2–7 and 17).<sup>13</sup> Only remaining amounts of the intermediary imine or the volatility of the substrates sometimes slightly impacted the recovery of the aldehyde. Further, substituents in *ortho*, *meta* and *para* positions as well as multi-substituted benzaldehydes are all well tolerated while guidance by Lewis-basic substituents is observed in the presence of two non-identical *ortho* positions, overriding potential steric effects (**14**, **16**, **20**).<sup>14</sup>

Strongly electron-deficient benzaldehydes comprising nitro or cyano substituents (**19**–**23**) required slightly modified reaction conditions: due to the increased stability of the intermediary imine,<sup>15</sup> a switch from base to acid catalysis facilitated turnover of the catalytic amine, thus yielding the corresponding aldehydes with high deuterium incorporation even at a low catalyst loading of 2.5 mol%. Similarly, good results were achieved with heterocyclic aldehydes (**24**–**28**), demonstrating the broad applicability of this methodology. Only substrates containing heteroatoms in  $\alpha$ -position to the aldehyde moiety as well as pyridine derivatives appeared to form unreactive manganese complexes instead. Notably, chromone derivative **28** undergoes formyl deuteration additionally, presumably aided by coordination of the proximal keto group.

To test the functional group compatibility of this reaction beyond the groups already mentioned in the text and highlighted in Scheme 3, further deuteration experiments were conducted in the presence of equimolar amounts of a terminal olefin, a boronic acid, a sulfonamide, and phenol. All these additives were well-tolerated and did not affect the deuterium incorporation in the aldehyde substrate. On the other hand, hydroxy-substituted aldehydes as well as substrates comprising terminal alkynes were not deuterated.

From a synthetic point of view it is interesting that the reaction is selective for aldehydes, leaving aromatic ketones untouched despite their ability to form ketimines. Moreover, other carbonyl groups which had previously been employed as



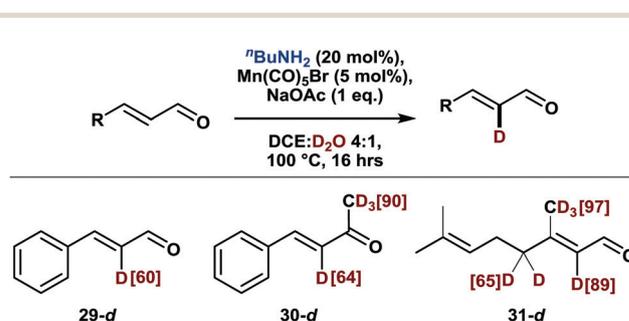


Scheme 3 Mn-catalyzed *o*-deuteration of various benzaldehyde derivatives. Scale: 0.5 mmol. Deuterium incorporation was determined by <sup>1</sup>H NMR by referencing on the formyl proton and/or protons in *m* or *p* positions and is indicated in square brackets (red). Percentage values below the structures refer to isolated yields. <sup>a</sup> *n*BnNH<sub>2</sub> instead of <sup>t</sup>BuNH<sub>2</sub>. <sup>b</sup> 50 mol% *p*-chlorobenzoic acid instead of 1 eq. NaOAc. <sup>c</sup> 2.5 mol% Mn(CO)<sub>5</sub>Br. <sup>d</sup> 72 h. <sup>e</sup> 8 h. <sup>f</sup> 10 mol% Mn(CO)<sub>5</sub>Br.

directing groups for manganese catalyzed C–H functionalization<sup>16</sup> did not induce *ortho*-deuteration under our conditions, whereas heterocyclic directing groups could potentially enable additional deuteration patterns in more complex substrates.

While a variety of methods exists for the labeling of arenes (*vide supra*), the deuteration of olefinic substrates is much less developed.<sup>17</sup> Notably, current methodologies for the labeling of  $\alpha,\beta$ -unsaturated carbonyl compounds either exchange all olefinic protons alike or are slightly selective for the  $\beta$  position.<sup>17c,e,f</sup> Likewise, a previously reported manganese-catalyzed olefinic C–H activation also led to exclusive functionalization in  $\beta$  position.<sup>18</sup> In this context, we explored the deuteration of  $\alpha,\beta$ -unsaturated aldehydes and ketones. Indeed, as shown in Scheme 4, labelling proceeded smoothly with high selectivity for the  $\alpha$ -position. While this work was in progress, an amine-mediated

$\alpha$ -selective deuteration of  $\alpha,\beta$ -unsaturated aldehydes and ketones was published.<sup>19</sup> Accordingly, substrates 29–31 can



Scheme 4  $\alpha$ -Deuteration of  $\alpha,\beta$ -unsaturated aldehydes and ketones. Scale: 0.5 mmol. Deuterium incorporation was determined by <sup>1</sup>H NMR.



be also deuterated in the absence of the manganese catalyst, whereby the amine acts as a nucleophilic catalyst rather than a transient directing group.

In conclusion, we describe a new base metal-catalyzed *ortho*-selective deuteration methodology using the most practical labelling reagent (D<sub>2</sub>O). This convenient alternative to current iridium-catalyzed directed hydrogen isotope exchange reactions shows broad scope of (hetero)aromatic aldehydes and versatile functional group compatibility. Moreover, we have introduced the transient directing group concept to the field of manganese-catalyzed C(sp<sup>2</sup>)-H activation which will undoubtedly improve the step economy of future transformations by alleviating the need for pre-installed directing groups.

We acknowledge the financial support from the European Union (Flow Chemistry for Isotope Exchange, FLIX, 862179), the BMBF and the State of Mecklenburg-Western Pomerania. We thank the analytical team of LIKAT for their kind support.

## Conflicts of interest

There are no conflicts to declare.

## Notes and references

- (a) J. Atzrodt, V. Derdau, W. J. Kerr and M. Reid, *Angew. Chem., Int. Ed.*, 2018, **57**, 1758; (b) J. Atzrodt, V. Derdau, W. J. Kerr and M. Reid, *Angew. Chem., Int. Ed.*, 2018, **57**, 3022.
- (a) J. A. Brown, A. R. Cochrane, R. Irvine, W. J. Kerr, B. Mondal, J. A. Parkinson, L. C. Paterson, M. Reid, T. Tuttle, S. Andersson and G. N. Nilsson, *Adv. Synth. Catal.*, 2014, **356**, 3551; (b) W. J. Kerr, M. Reid and T. Tuttle, *ACS Catal.*, 2015, **5**, 402; (c) M. Valero, T. Kruissink, J. Blass, R. Weck, S. Güssregen, A. T. Plowright and V. Derdau, *Angew. Chem., Int. Ed.*, 2020, **59**, 5626; (d) W. J. Kerr, G. J. Knox, M. Reid, T. Tuttle, J. Bergare and R. A. Bragg, *ACS Catal.*, 2020, **10**, 11120.
- (a) R. P. Yu, D. Hesk, N. Rivera, I. Pelczer and P. J. Chirik, *Nature*, 2016, **529**, 195; (b) J. Corpas, P. Viereck and P. J. Chirik, *ACS Catal.*, 2020, **10**, 8640; (c) S. Garhwal, A. Kaushansky, N. Fridman, L. J. W. Shimon and G. de Ruiter, *J. Am. Chem. Soc.*, 2020, **142**, 17131.
- (a) H. Yang, C. Zarate, W. N. Palmer, N. Rivera, D. Hesk and P. J. Chirik, *ACS Catal.*, 2018, **8**, 10210; (b) C. Zarate, H. Yang, M. J. Bezdek, D. Hesk and P. J. Chirik, *J. Am. Chem. Soc.*, 2019, **141**, 5034.
- J. Zhang, S. Zhang, T. Gogula and H. Zou, *ACS Catal.*, 2020, **10**, 7486.
- (a) W. Liu and L. Ackermann, *ACS Catal.*, 2016, **6**, 3743; (b) P. Gandeepan, T. Müller, D. Zell, G. Cera, S. Warratz and L. Ackermann, *Chem. Rev.*, 2019, **119**, 2192; (c) R. Cano, K. Mackey and G. P. McGlacken, *Catal. Sci. Technol.*, 2018, **8**, 1251.
- There is a report on manganese catalyzed aliphatic HIE, though: S. Kar, A. Goepfert, R. Sen, J. Kothandaraman and G. K. S. Prakash, *Green Chem.*, 2018, **20**, 2706.
- (a) B. Zhou, P. Ma, H. Chen and C. Wang, *Chem. Commun.*, 2014, **50**, 14558; (b) S. Wu, Q. Yang, Q. Hu, Y. Wang, L. Chen, H. Zhang, L. Wu and J. Li, *Org. Chem. Front.*, 2018, **5**, 2852; (c) C. Zhu, R. Kuniyil and L. Ackermann, *Angew. Chem., Int. Ed.*, 2019, **58**, 5338.
- A few exceptions exist but need bimetallic C-H activation systems: (a) S. Sueki, Z. Wang and Y. Kuninobu, *Org. Lett.*, 2016, **18**, 304; (b) B. Zhou, Y. Hu, T. Liu and C. Wang, *Nat. Commun.*, 2017, **8**, 1169; (c) Y. Hu, B. Zhou, H. Chen and C. Wang, *Angew. Chem., Int. Ed.*, 2018, **57**, 12071; (d) X. Kong and B. Xu, *Org. Lett.*, 2018, **20**, 4495; (e) S. Ali, J. Huo and C. Wang, *Org. Lett.*, 2019, **21**, 6961.
- W. Liu, S. C. Richter, Y. Zhang and L. Ackermann, *Angew. Chem., Int. Ed.*, 2016, **55**, 7747.
- (a) P. Gandeepan and L. Ackermann, *Chemistry*, 2018, **4**, 199; (b) S. John-Campbell and J. A. Bull, *Org. Biomol. Chem.*, 2018, **16**, 4582; (c) B. Niu, K. Yang, B. Lawrence and H. Ge, *ChemSusChem*, 2019, **12**, 2955; (d) J. I. Higham and J. A. Bull, *Org. Biomol. Chem.*, 2020, **18**, 7291.
- Catalyst loadings of 10 mol% are most common for manganese-catalyzed C-H activation procedures. Cf. ref. 7, 8b, e.
- Recent examples of deuterodehalogenation: (a) Z.-Z. Zhou, J.-H. Zhao, X.-Y. Guo, X.-M. Chen and Y.-M. Liang, *Org. Chem. Front.*, 2019, **6**, 1649; (b) J. J. Gair, R. L. Grey, S. Giroux and M. A. Brodney, *Org. Lett.*, 2019, **21**, 2482; (c) M. Kuriyama, G. Yano, H. Kiba, T. Morimoto, K. Yamamoto, Y. Demizu and O. Onomura, *Org. Process Res. Dev.*, 2019, **23**, 1552.
- This effect has previously been observed in manganese catalyzed C-H activation: cf. ref. 9 and D. Zell, U. Dhawa, V. Müller, M. Bursch, S. Grimme and L. Ackermann, *ACS Catal.*, 2017, **7**, 4209.
- R. W. Layer, *Chem. Rev.*, 1963, **63**, 489.
- T. S. Amide, T. Yoshida, H. H. A. Mamari, L. Ilies and E. Nakamura, *Org. Lett.*, 2017, **19**, 5458; ester: ref. 8a.
- Examples for olefinic deuteration: (a) S. K. S. Tse, P. Xue, Z. Lin and G. Jia, *Adv. Synth. Catal.*, 2010, **352**, 1512; (b) A. Di Giuseppe, R. Castarlenas, J. J. Pérez-Torrente, F. L. Lahoz, V. Polo and L. A. Oro, *Angew. Chem., Int. Ed.*, 2011, **50**, 3938; (c) W. J. Kerr, R. J. Mudd, L. C. Paterson and J. A. Brown, *Chem. – Eur. J.*, 2014, **20**, 14604; (d) M. Hatano, T. Nishimura and H. Yorimitsu, *Org. Lett.*, 2016, **18**, 3674; (e) A. L. Garreau, H. Zhou and M. C. Young, *Org. Lett.*, 2019, **21**, 7044; (f) A. Bechtold and L. Ackermann, *ChemCatChem*, 2019, **11**, 435.
- B. Zhou, Y. Hu and C. Wang, *Angew. Chem., Int. Ed.*, 2015, **54**, 13659.
- V. G. Landge, K. K. Shrestha, A. J. Grant and M. C. Young, *Org. Lett.*, 2020, **22**, 9745.

