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## Correction: Anion exchange-induced single-molecule dispersion of cobalt porphyrins in a cationic porous organic polymer for enhanced electrochemical CO<sub>2</sub> reduction *via* secondary-coordination sphere interactions

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Correction for 'Anion exchange-induced single-molecule dispersion of cobalt porphyrins in a cationic porous organic polymer for enhanced electrochemical CO<sub>2</sub> reduction *via* secondary-coordination sphere interactions' by Jia-Kang Tang *et al.*, *J. Mater. Chem. A*, 2020, DOI: 10.1039/d0ta07068h.

The authors regret an error in Fig. 2a in the published article. In Fig. 2a, the TEM image of POP-Py(0) was presented as the same as that of POP-Py(0)/CoTCPP. The error was due to the same image being repetitively imported in the graphics software in the revised version. The authors confirm that the error has no effect on the conclusions of this paper. Furthermore, the authors state that the raw data of the TEM image shown in Fig. 2 are available from the first author (J. T.) and/or the corresponding authors (D. Z. and L. Z.) upon request. The corrected version of Fig. 2 is shown below.

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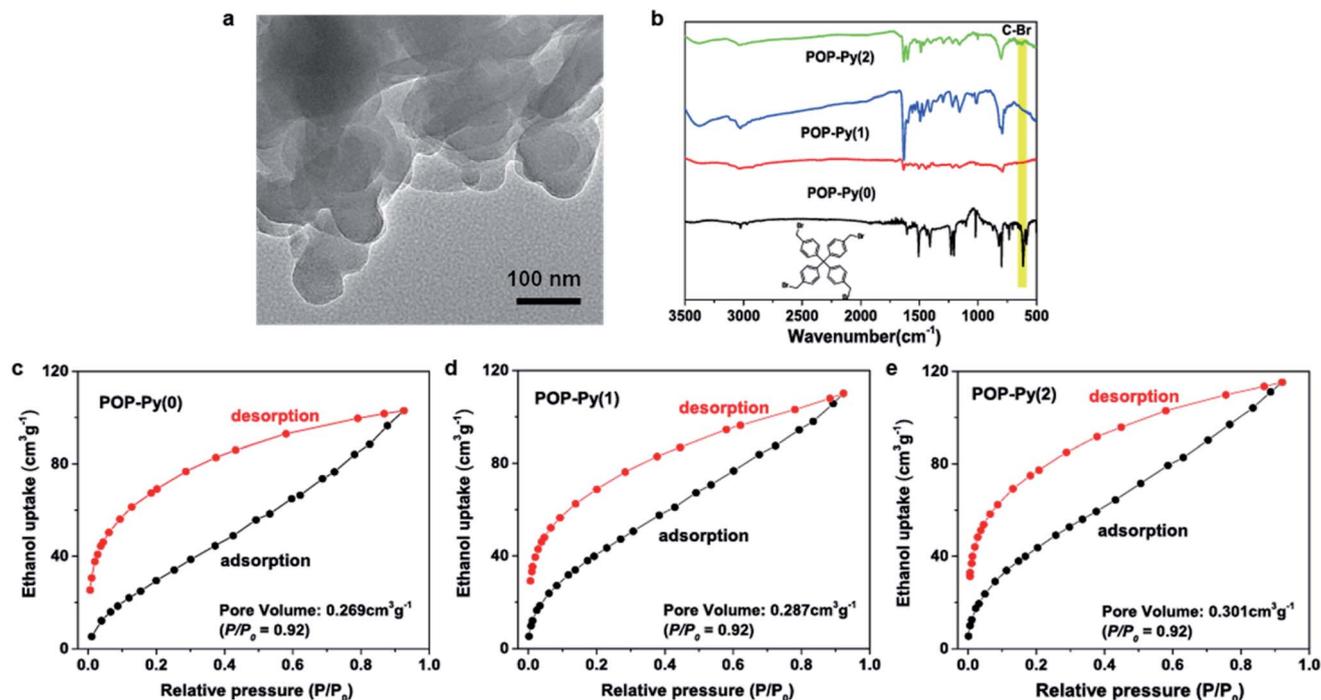


Fig. 2 The structural and porosity characterization of three cationic POP-Py(*n*). (a) A representative TEM image of POP-Py(0). (b) FT-IR spectra comparison of the TBM precursor (inset) and POP-Py(*n*). The C–Br stretching vibration at  $\sim 615\text{ cm}^{-1}$  is highlighted. (c–e) Ethanol adsorption/desorption isotherms of POP-Py(0) (c), POP-Py(1) (d) and POP-Py(2) (e) at  $25\text{ }^{\circ}\text{C}$ . The pore volume of each POP-Py(*n*) is calculated at  $P/P_0 = 0.92$ .

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

