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## CORRECTION

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## Correction: Flat-shaped carbon—graphene microcomposites as electrodes for high energy supercapacitors

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Correction for 'Flat-shaped carbon-graphene microcomposites as electrodes for high energy supercapacitors' by Gelines Moreno-Fernández et al., J. Mater. Chem. A, 2019, 7, 14646–14655.

The authors regret a mistake in Table. 1 of the published article. The correct version of the Table 1 is shown below.

The authors also wish to amend the following sentences in the original version of the manuscript:

- The sentence "The t-plot method was used to calculate the external surface area ( $S_{\text{EXT}}$ ) in the relative pressure range of 0.07–0.25." included in the physicochemical characterization (page 14648) section should be removed.
  - The last sentence of page 14648, where it is written "The increment of  $S_{\rm EXT}$ ..." should say "The increment of  $V_{\rm meso}$ ..."
- The text on line 7, page 14649 which reads "The isotherm of the activated material also presents a marked micropore contribution but, as evidenced by  $V_{\rm meso}$  and  $S_{\rm EXT}$  values (Table 1) some narrow mesoporosity is generated during activation." should instead read as follows: "The isotherm of the activated material also presents a marked micropore contribution but, as evidenced by  $V_{\rm meso}$  values (Table 1) some narrow mesoporosity is generated during activation".

The authors acknowledge Dr Teresa Centeno, INCAR-CSIC (Spain) for detecting these errors.

Table 1 Total pore volume ( $V_{T}$ ), mesopore volume ( $V_{meso}$ ), micropore volume ( $V_{DR}$ ), BET specific surface area ( $S_{BET}$ ), DFT specific surface area ( $S_{DR}$ ), total micropore surface area ( $S_{DR}$ ), narrow micropore volume ( $V_{DR}$ ), and narrow micropore specific surface area ( $S_{DR}$ )

	$N_2^{\ a}$						$\mathrm{CO_2}^b$	
	$V_{\rm T}$ (cm <sup>3</sup> g <sup>-1</sup> )	$V_{\rm meso}$ (cm <sup>3</sup> g <sup>-1</sup> )	$V_{\mathrm{DR}}^{c} \left(\mathrm{cm}^{3} \mathrm{g}^{-1}\right)$	$S_{\mathrm{BET}} \left( \mathrm{m}^2 \mathrm{g}^{-1} \right)$	$S_{\mathrm{DFT}}^{}} \left( \mathrm{m}^2 \; \mathrm{g}^{-1} \right)$	$S_{\mathrm{DR}}^{e} \left(\mathrm{m}^{2} \mathrm{g}^{-1}\right)$	$V_{\mathrm{DR}}^{c} \left(\mathrm{cm}^{3} \mathrm{g}^{-1}\right)$	$S_{\mathrm{DR}}^{e} \left(\mathrm{m}^{2} \mathrm{g}^{-1}\right)$
ResFa	0.37	0.13	0.24	598	720	551	0.19	632
ResFaGO	0.50	0.11	0.39	948	1150	880	0.22	840
ResFaGO-A	1.12	0.21	0.91	1961	1991	1373	0.47	1122

<sup>&</sup>lt;sup>a</sup> Data obtained from N₂ adsorption–desorption isotherms at −196 °C. <sup>b</sup> Data obtained from CO₂ adsorption isotherms at 0 °C. <sup>c</sup> Data obtained from Dubinin–Radusckevich. <sup>d</sup> Data obtained from the 2D-NLDFT. <sup>e</sup> Data obtained from Dubinin–Radusckevich.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

<sup>°</sup>CIC EnergiGUNE, Parque Tecnológico de Álava, 01510 Miñano, Álava, Spain. E-mail: dcarriazo@cicenergigune.com; Tel: +34 94 529 71 08

<sup>&</sup>lt;sup>b</sup>Universidad Del País Vasco, UPV/EHU, 48080 Bilbao, Spain

cIKERBASQUE, Basque Foundation for Science, 48013 Bilbao, Spain