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Double annulation of *ortho*- and *peri*-C–H bonds of fused (hetero)arenes to unusual oxepino-pyridines†

Majji Shankar,^a Raja K. Rit,^a Somratan Sau,^a Kallol Mukherjee,^a Vincent Gandon^{*bc} and Akhila K. Sahoo^{*a}

Direct difunctionalization of chemically distinct *ortho*- and *peri*-C–H bonds of fused hetero(arenes) is illustrated through an unusual one-pot domino $\{[4 + 2] \& [5 + 2]\}$ double annulation with alkynes for the first time. This process is viable under Ru(II)-catalysis using a sulfoximine directing group and builds four bonds $[(C-C)-(C-N)]$ and $[(C-C)-(C-O)]$ in a single operation. Such synthetic manifestation offers access to uncommon $[6,7]$ -fused oxepino-pyridine skeletons. DFT calculations provide mechanistic insight into this double annulation of naphthoic acid derivatives with alkynes and corroborate the participation of a ruthena-oxabicyclooctene intermediate, which is responsible for the rare 7-membered ring formation.

Introduction

Diversity oriented synthesis provides efficient access to complex molecular architectures that are present in natural products, pharmaceuticals, agrochemicals, and advanced-materials.¹ This approach has sustained the development of novel therapeutic agents or probes for molecular biology, based on the resilient interaction of heterocycles with biological systems.^{2,3} Continuous efforts have therefore been directed towards the conception of straightforward synthetic methods for the construction of complex heteroarenes.³ In this regard, transition-metal (TM) catalyzed annulations of C–H bonds of (hetero)arenes with alkynes have proven invaluable.^{4,5} In particular, the TM-catalyzed direct functionalization or annulation of the *ortho*-C(2)–H bond of fused (hetero)arenes with alkynes are successful with acid/amide directing groups (DGs) *via* 5/7-membered metallacycle (Fig. 1A-I).⁵ With –OH, –NHR', and –SR'' DGs, the reactivity is shifted towards the *peri*-C(8)–H bond through 5/7-membered metallacycle (Fig. 1A-II).⁶ On the other hand, the activation of the *peri*-C(8)–H bond of fused (hetero)arene carboxylic acid derivatives [e.g. 1-naphthoic acid] is much more challenging and underdeveloped, due probably to the

involvement of a strained $[6,6,6]$ -fused metallacycle (Fig. 2A).⁷ Insertion of an alkyne would not even funnel such C–H activation step, as it would lead to an even more strained $[6,6,8]$ -fused metallacycle (Fig. 2A). Thus, the molecular rigidity and conformational strain have hampered the development of such annulations at the *peri*-C(8)–H bond to form 7-membered fused compounds (Fig. 2A).^{8,9}

Recent domino one-pot double annulation of *o/o'*-C–H bonds of (hetero)arenes with alkynes have led to $[6,6]$ -fused heteroaryls.^{10,11} Although important issues of regio- and chemoselectivity, cumbersome mixtures due to incomplete conversion, catalytic viability, *etc.*, could be addressed,¹² such domino double C–H annulations were not extended to the formation of $[6,7]$ -fused heteroarenes. To make such synthetic

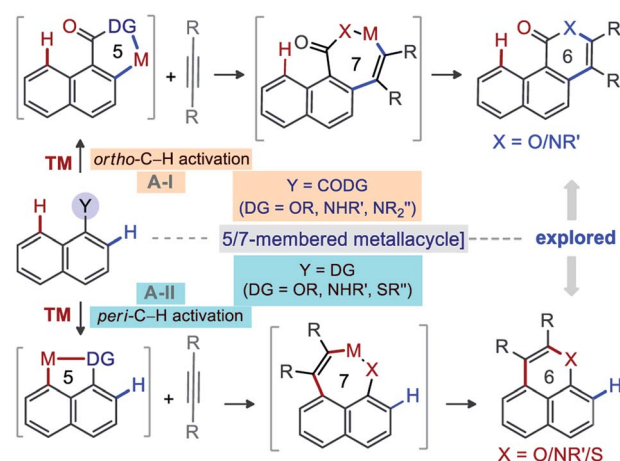


Fig. 1 Background: Annulation of *ortho*-C(2)–H & *peri*-C(8)–H bond of 1-naphthalene derivatives with alkynes.

^aSchool of Chemistry, University of Hyderabad, Hyderabad-500046, India. E-mail: akhilchemistry12@gmail.com; akssc@uohyd.ac.in

^bInstitut de Chimie Moléculaire et des Matériaux d'Orsay, CNRS UMR 8182, Université Paris-Sud, Université Paris-Saclay, Bâtiment 420, 91405 Orsay Cedex, France. E-mail: vincent.gandon@universite-paris-saclay.fr

^cLaboratoire de Chimie Moléculaire (LCM), CNRS UMR 9168, Ecole Polytechnique, Institut Polytechnique de Paris, route de Saclay, 91128 Palaiseau Cedex, France

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Fig. 2 Multiple annulation of (hetero)arenes.

plan feasible, we hypothesized a Ru-catalyzed double annulation of 1-naphthoic acid derivatives with alkynes.

We believed the reaction would be initiated by N-aided C(2)-H activation and annulation with the alkyne to first form an angularly [6,6,6]-fused benzo[*h*]isoquinolinol. As *peri*-C-H bonds of fused-arenes are susceptible to electrophilic substitution, we anticipated an O-directed ruthenation of the proximal *peri*-C(8)-H bond to provide **Int-Z** (Fig. 2B). Finally, second alkyne incorporation to **Int-Z** and reductive elimination would build the unusual [6,7]-fused oxepino-pyridine motif (Fig. 2B). This one-pot domino double annulation uses the methylphenyl sulfoximine (MPS)-DG.^{12b} Thus, the sequential activation of *ortho*- and *peri*-C-H bonds and annulation results in the formation of N- and O-enabled 6- and 7-membered rings on fused (hetero)arenes by generating four bonds (C-C & C-N and C-C & C-O) in a single operation (Fig. 2B).

Results and discussion

This one-pot [4 + 2] & [5 + 2] annulation was developed under Ru-catalysis using *N*-[1-naphthoyl]methylphenyl sulfoximine (**1a**) and 4-octyne (**2a**). The optimization studies are detailed in Table 1.¹³ The oxepino-pyridine **3aa** was detected in 8% yield using {[RuCl₂(*p*-cymene)]₂ (5.0 mol%), AgSbF₆ (20 mol%), NaOAc (1.0 equiv.)} as catalytic system, in ClCH₂CH₂Cl (DCE) at 120 °C for 24 h (entry 1). The cleavage of the sulfoximine motif presumably helps the formation of **3aa**.^{11d} In general, metal acetates facilitate Ru-mediated C-H activation through CMD (concerted metalation deprotonation), and also act as oxidant in the regeneration of the active catalyst.⁴ Accordingly, the double annulation was slightly improved when the reaction was conducted in the presence of the redox active bases Mn(OAc)₂, AgOAc, and Zn(OAc)₂·2H₂O (entries 2–4), while Cu(OAc)₂·H₂O was found more promising as it delivered **3aa** in 35% yield

Table 1 Optimization of reaction conditions^a

Entry	Additive 1 (20 mol%)	Additive 2 (1.0 equiv.)	Solvent	Yield 3aa ^b (%)
1	AgSbF ₆	NaOAc	DCE	8
2	"	Mn(OAc) ₂	DCE	12
3	"	AgOAc	DCE	15
4	"	Zn(OAc) ₂ ·2H ₂ O	DCE	11
5	"	Cu(OAc) ₂ ·H ₂ O	DCE	35
6	KPF ₆	"	DCE	<5 ^c
7	NaPF ₆	"	DCE	6
8	AgBF ₄	"	DCE	30
9	AgSbF ₆	"	MeCN	<5 ^c
10	"	"	Toluene	7
11	"	"	TCE	22
12	"	"	1,4-Dioxane	41
13 ^d	AgSbF ₆	Cu(OAc) ₂ ·H ₂ O	1,4-Dioxane	68
14 ^e	AgSbF ₆	Cu(OAc) ₂ ·H ₂ O	1,4-Dioxane	77
15	AgSbF ₆	—	1,4-Dioxane	<5 ^c
16	—	Cu(OAc) ₂ ·H ₂ O	1,4-Dioxane	<5 ^c



^a Conditions: **1a** (0.3 mmol), **2a** (0.9 mmol), [RuCl₂(*p*-cymene)]₂ (5.0 mol%), additive-1 (20 mol%), additive-2 (0.3 mmol), solvent (2.0 mL) at 120 °C. ^b Isolated yield. ^c ¹H NMR conversion. ^d [RuCl₂(*p*-cymene)]₂ (10 mol%), AgSbF₆ (40 mol%) was used. ^e **2a** (1.2 mmol), [RuCl₂(*p*-cymene)]₂ (10 mol%), AgSbF₆ (40 mol%), Cu(OAc)₂·H₂O (1.5 equiv.) was used. DCE = ClCH₂CH₂Cl, TCE = 1,1,2,2-tetrachloroethane.

(entry 5). Additives such as KPF₆, NaPF₆, or AgBF₄ instead of AgSbF₆ were not beneficial (entries 6–8). The reaction efficiency was low when conducted in MeCN, toluene or TCE (entries 9–11). The domino diannulation in 1,4-dioxane provided **3aa** in 41% yield (entry 12). The yield of **3aa** was significantly improved to 68% when 10 mol% of Ru-catalyst and 40 mol% of AgSbF₆ were used (entry 13). Finally, the catalytic conditions comprising [Ru(*p*-cymene)Cl₂]₂ (10 mol%), AgSbF₆ (40 mol%), and Cu(OAc)₂·H₂O (1.5 equiv.) in 1,4-dioxane at 120 °C for 24 h were found optimum (entry 14), producing **3aa** in 77% yield. Control experiments revealed that the silver salt and the acetate base were crucial (entries 15 and 16).^{14d}

To validate the role of DGs in this one-pot domino {[4 + 2] & [5 + 2]} double annulation strategy, various DG-enabled 1-naphthyl bearing amides (**I–VI**) were subjected to the annulation with **2a** under the optimized conditions (bottom of Table 1). The substrates having NH-Me (**I**) and NH-tosyl (**II**) DGs proved unreactive, whereas, simple 1-naphthylamide (**III**) underwent this domino annulations with **2a** producing **3aa** in poor yield.⁶ The N-oxidizable group protected amides [**IV** (with N-O bond), **V**, and **VI** (with N-N bond)] provided **3aa** in 15%,

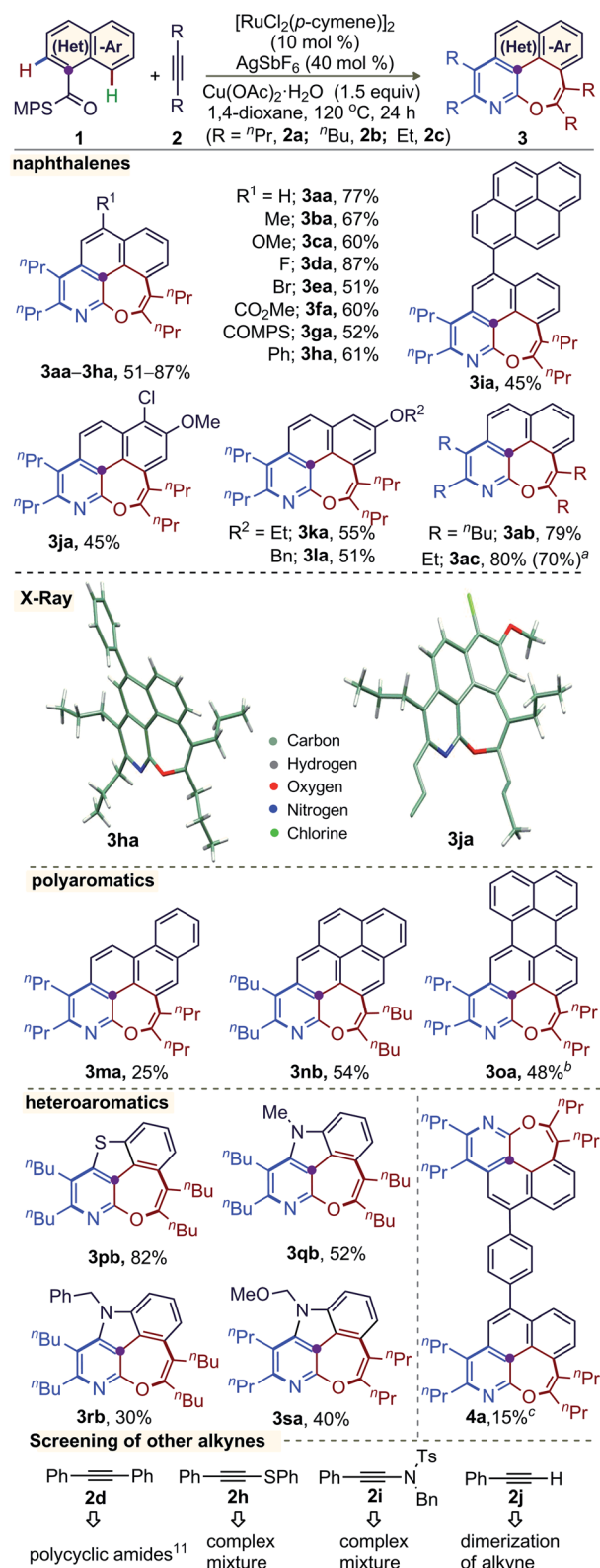


7%, and 26% yield, respectively. Thus, the MPS-DG was found most effective for the construction of the [6,7]-fused oxepino-pyridine skeleton.¹³

The generality of this annulation among fused (hetero)arenes exhibiting *peri*-C–H bonds and unactivated alkynes was explored under the optimized catalytic conditions (Scheme 1). The annulation of naphthalene derivatives **1a–l**, bearing either electron-donating (Me, OMe, OEt), labile halo (F, Cl, Br), electron-withdrawing (CO₂Me, COMPS), arene (Ph, pyrene), and OBN substituents at position 4, 5, or 6, with **2a**, was successful in producing the respective 6,7-fused oxepino-pyridine **3aa–la** in 45–87% yield. The tolerance of modifiable functionalities (*i.e.* F, Cl, Br, CO₂Me, COMPS) offers the possibility of further functionalization. The core structure of **3ha** and **3ja** were elucidated by X-ray crystallographic analysis.^{14,15} Likewise, this double-annulation of **1a** with the other internal alkynes **5-decyne (2b)** and **3-hexyne (2c)** delivered **3ab** (79%) and **3ac** (80%), respectively. Moreover, the gram scale synthesis of **3ac** (1.15 g) with recovery of PhSOMe (0.44 g) showed the robustness of the catalytic system and the transformable nature of the MPS group.^{5g} Polyarene bearing scaffolds, for example: phenanthrene (**1m**), pyrene (**1n**), and perylene (**1o**), delivered **3ma**, **3nb** and **3oa**, albeit in moderate yield.

Importantly, benzothiophene derivative **1p** smoothly reacted with **2b** to afford **3pb** in 82% yield. Indole-3-carboxylic acid derivatives **1q–s** were used in this double annulation with **2b** and **2a**. The respective complex heteroarenes **3qb**, **3rb**, and **3sa** were reliably accessed. The common N-protecting groups benzyl and MOM did not prevent the reaction. The yields are moderate in these cases, but the construction of these molecular scaffolds with three heteroatoms (*i.e.* S–N–O, N–N–O) in a 5,6,7-fused system is remarkable. Notably, the current synthetic plan was successful in making 8 bonds (4 C–C, 2 C–N, and 2 C–O) in a single operation; thus, an extended π -conjugated system **4a** with two oxepino-pyridine motifs was made. The reaction of **1a** with diphenylacetylene provided polycyclic amides through linear diannulation.^{11,14} On the other hand, the reaction of a thioalkyne or an ynamide with **1a** produced complex mixtures (Scheme 1). Lastly, the terminal alkyne phenylacetylene underwent dimerization under the optimized oxidative condition.

The site-specific introduction of a novel functionality on an unreactive site of a complex motif has tremendous significance to the field of complex molecule synthesis and is often termed as late stage functionalization (LSF).¹⁶ In particular, LSF through C–H functionalization is very useful in drug discovery and draws significant attention from the scientific community. Accordingly, a range of biologically relevant motifs moulded with MPS-bearing naphthalene-1-carboxylic acid (**5a–g**) were synthesized and were independently subjected to the optimized reaction conditions with **2a** and **2c** (Scheme 2). Thus, the desired oxepino-pyridines **6aa**– β -citronellol, **6bc**–camphorsultam, **6ca**–(–)-boreneol, **6cc**–cholesterol, **6fc**–estrone, and **6gc**–lithocholic acid were constructed without any structural (chemical and stereochemical) changes of the complex architecture.¹⁴ The poor-to-moderate synthetic yields are due to low conversions. Isolation of unreacted precursors justifies the mass balance of the transformation.



Scheme 1 Synthesis of 6,7-oxepino[2,3-*b*]pyridine. Reactions were carried out with **1** (0.3 mmol) and **2** (1.2 mmol). ^aGram scale: **1a** (1.54 g, 5.0 mmol); PhS(OMe) (63%) was isolated. ^bReactions were carried out in DCE. ^c**2a** (1.8 mmol).



Scheme 2 Double annulation of MPS-bearing naphthalene-1-carboxylic acid moulded in biologically relevant motifs. Reactions were carried out with **5** (0.3 mmol), **2** (1.2 mmol), $[\text{RuCl}_2(p\text{-cymene})]_2$ (10 mol%), AgSbF_6 (40 mol%), 1,4-dioxane (2.0 mL) at 120 °C for 24 h. ^aIsolation of unreacted precursors (20–55%).

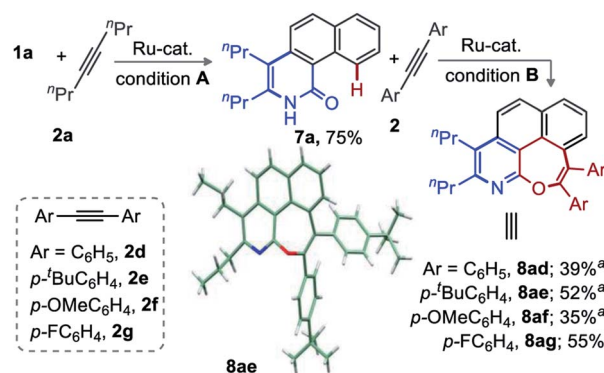
Encouraged by the broad range of oxepino-pyridines derivatives obtained (Schemes 1 and 2), the title reaction was next envisaged with two different alkynes. However, the difference in reactivity, regio- and chemoselectivity with different alkynes led to unexploitable annulation mixtures.¹² To make this challenging unsymmetrical transformation viable, a two-step annulation sequence was tested. Accordingly, benzo[*h*]isoquinolinone **7a** (0.5 mmol, 75%) was accessed from **1a** and **2a** when the reaction was carried out in presence of AcOH under Ru-catalysis (Scheme 3, Conditions A). Presumably the acid suppresses the second annulation through proto-demetalation.¹¹ Next, the annulation of **7a** with 1,2-diaryl alkynes (**2d–g**) led to the respective [6,7]-fused oxepino-pyridines (**8ad–ag**) in moderate yields (Scheme 3). The structure of **8ae** was unambiguously confirmed by X-ray crystallography.^{14,15} A deuterium scrambling study and competition experiments were then performed to gain some mechanistic insight into this annulation (Scheme 4).

Exposing **1a** to the optimized conditions in presence of $\text{CD}_3\text{CO}_2\text{D}$ (2.5 equiv.) resulted in D-incorporation at C2 (65%) and C8 (62%) positions (eqn (1)). Similarly, 55% of deuterium incorporation occurred at C8 in an identical experiment with **7a** (eqn (2)). Therefore, activation of both the *ortho*- and *peri*-C–H bonds of MPS-enabled-1-naphthylamide is reversible. The competitive annulation of an equimolar mixture of **1c** and **1f** with **2a** led to a 2 : 1 ratio of **3ca** and **3fa**; thus, an electron-rich arene reacts faster than an electron-poor one (eqn (3)).

In general, the π -conjugated polyfused heteroarenes show interesting photophysical properties. Thus, the absorption and

emission spectra of oxepino-pyridines **3nb**, **3oa**, **3pb**, **3qb**, **3sa**, **4a**, and **8ae** were measured in dichloromethane (1×10^{-5}).¹⁴ Of note, compounds **3nb** and **3ob** show emission maxima at 436–512 nm with broad bandwidths and weak intensities.¹⁴

The mechanism of the title reaction has been studied computationally, employing the Gaussian 09 software package.¹⁷ Following a recent report, optimizations were carried



Scheme 3 Unsymmetrical double-annulation of arenes with different alkynes. Conditions A: **1** (0.5 mmol), **2a** (1.0 mmol), $[\text{RuCl}_2(p\text{-cymene})]_2$ (5.0 mol%), AgSbF_6 (20 mol%), AcOH (4.0 mmol), DCE (2.5 mL) at 120 °C for 20 h. Conditions B: **7a** (0.3 mmol), **2** (0.45 mmol), $[\text{RuCl}_2(p\text{-cymene})]_2$ (7.5 mol%), AgSbF_6 (30 mol%), $\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$ (0.3 mmol), KH_2PO_4 (0.6 mmol), 1,4-dioxane (2.0 mL) at 120 °C for 20 h. ^aIsolation of unreacted mono-annulation product (30–45%).



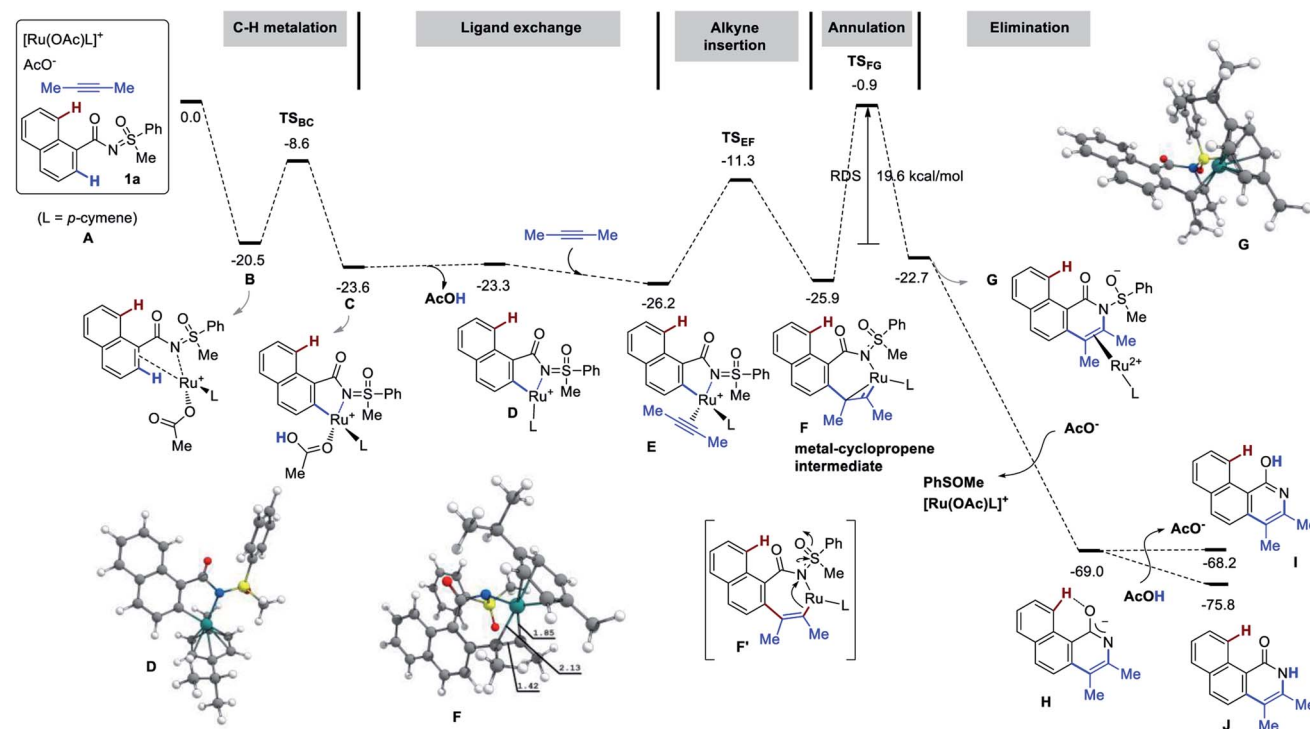


Scheme 4 Deuterium scrambling and competition studies.

out with the M06 functional, the 6-31G(d,p) basis set for all main group elements, and the LANL2DZ+f (ECP)¹⁸ basis set for Ru. Single point calculations were conducted at the M06/6-311++G(d,p)-SDD+f(ECP) level of theory. Solvation energies were obtained at the single point level using SMD approach for 1,4-dioxane. The discussed values are solvent-corrected Gibbs free energies at 393.15 K in kcal mol^{−1} (ΔG_{393}). The molecular system **A** [**1a**, 2-butyne (2.0 equiv.), $[\text{Ru}(\text{OAc})\text{L}]^+$ ($\text{L} = p\text{-cymene}$), AcO^-] was used as a reference for the free energies (Fig. 3). Thus, **A** contains two acetates to ensure two deprotonation of

1a. The complexation of the putative active species $[\text{Ru}(\text{OAc})(p\text{-cymene})]^+$ with **1a** at first provides **B** with a release of 20.5 kcal mol^{−1}. Next, C–H metalation occurs through **TS_{BC}** lying 11.9 kcal mol^{−1} above **B** to provide metallacycle **C** (−23.6 kcal mol^{−1}). Elimination of acetic acid and insertion of 2-butyne delivers the alkyne-complex **E** (more stable than **C** by 2.6 kcal mol^{−1}). Alkyne insertion does not yield the proposed metal-alkenyl complex **F'**, but rather its valence isomer **F**, which is a metallacyclopentene as witnessed by the distortion of the 7-membered ring and by the short Ru–C distance of 1.85 Å. The formation of **F** is slightly endergonic by 0.3 kcal mol^{−1} that requires 14.9 kcal mol^{−1} of free energy of activation (**TS_{EF}**). Then, intramolecular nucleophilic addition to the N=S bond gives the annulation intermediate **G** (see arrows in **F'**). The conversion of **F** to **G** is the rate-determining step with a barrier 25.0 kcal mol^{−1} (19.6 kcal mol^{−1} from **B**), which is consistent with the temperature of the reaction (120 °C). Although the resulting complex **G** is less stable than **F** by 3.2 kcal mol^{−1}, the acetate aided dissociation of $[\text{Ru}(\text{OAc})\text{L}]^+$ promotes spontaneous elimination of PhSOMe from the free ligand to give **H**, located as low as −69.0 kcal mol^{−1} on the energy surface. The liberation of PhSOMe, the conjugation of the anion, and the strong H-bond in **H** assist the loss of the sulfur moiety.

Finally, protonation of **H** by AcOH produces pyridine **I** or the pyridone species **J**. In line with the experimental observations, **J** is significantly more stable. The mechanistic insight directed towards the second annulation for the construction of pyridine-fused 7-membered oxepine ring is depicted in Fig. 4. The complexation of **H** (at −69.0 kcal mol^{−1}) with $[\text{Ru}(\text{OAc})\text{L}]^+$ is exergonic by 56.4 kcal mol^{−1} and yields **K** at −125.4 kcal mol^{−1}. Intermediate **K** shows a H-bond between the acetate ligand and

Fig. 3 Free energy profile (ΔG_{393} , kcal mol^{−1}), part 1 (first annulation).

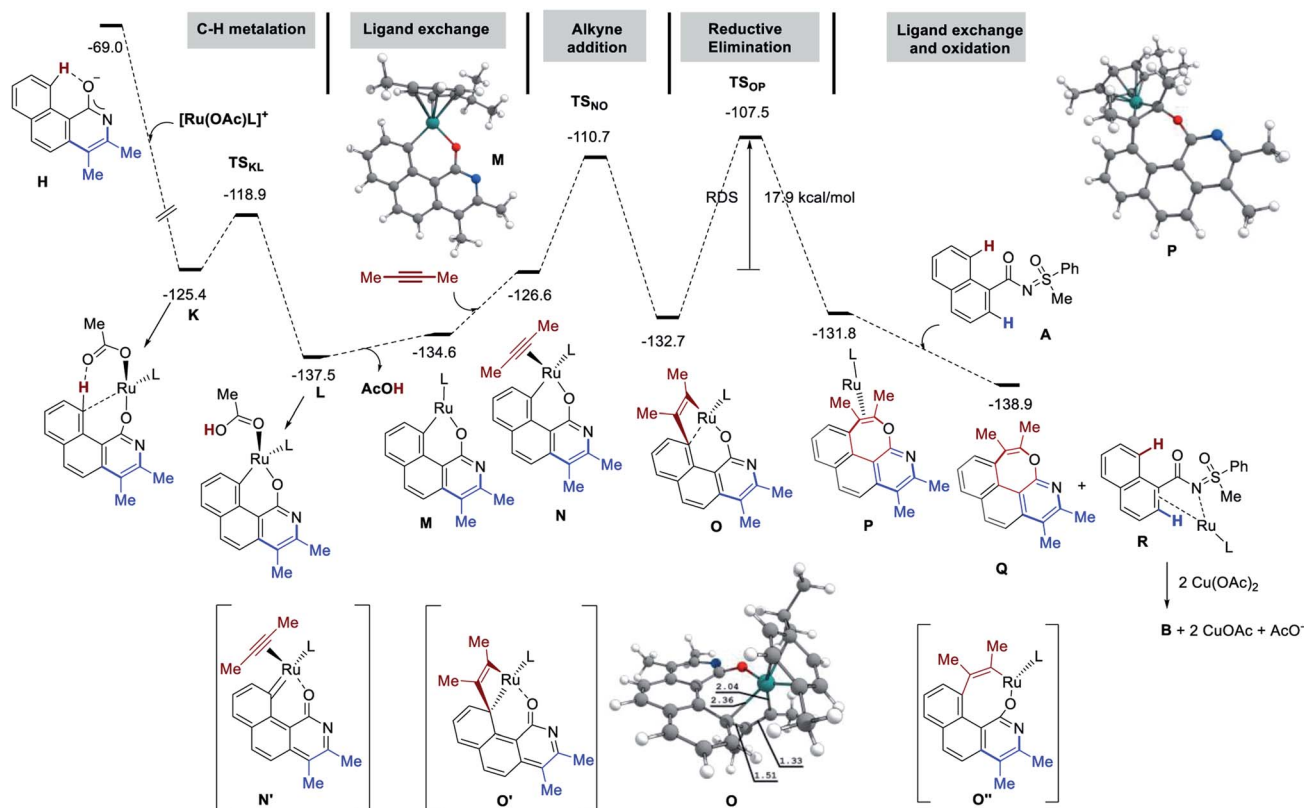


Fig. 4 Free energy profile (ΔG_{393} , kcal mol $^{-1}$), part 2 (second annulation).

the *peri*-H of the naphthalene moiety. The Ru–C bond is short (2.36 Å), due to the coordination of Ru to the *ipso*-carbon and makes the *peri*-H acidic. The C–H metalation of the pre-organized complex **K** provides **L** (at –137.5 kcal mol $^{-1}$ on the energy surface). This step requires 6.5 kcal mol $^{-1}$ free energy of activation (TS_{KL}). Next, the substitution of acetic acid with second alkyne equivalent is endergonic by 10.9 kcal mol $^{-1}$ to afford **N** (–126.6 kcal mol $^{-1}$). Of particular interest, the formation of 7-membered ring does not arise from the reductive elimination of a simple 8-membered metallacycle (**O'**). Instead, at the expense of 15.9 kcal mol $^{-1}$ of free energy of activation, the ruthena-oxabicyclooctene complex **O**, located at –132.7 kcal mol $^{-1}$, is achieved from **N** via TS_{NO} . Among the Lewis depiction of **O** and **O'**, the structure **O** is supported by the Ru–C^{*ipso*} distance of 2.35 Å and other geometrical parameters. Its formation can be understood as an intramolecular [2 + 2] cycloaddition between the alkyne and a Ru=C bond as shown in **N'** (a fictive valence isomer of **N**). This process eventually avoids the participation of a highly strained phenanthrene-containing 8-membered ring (**O''**). Then, the reductive elimination of **O** demands 25.2 kcal mol $^{-1}$ free energy of activation to give **P**. This process is slightly endergonic and is the rate-determining step of this second annulation process. The transfer of the RuL moiety from **P** to the precursor **1a** produces the desired [6,7]-fused oxepino-pyridine skeleton **Q** and chelate **R**. This step is exergonic by 7.9 kcal mol $^{-1}$. Finally, as it is generally accepted, one can then propose that complex **R** transforms into **B** by Cu(OAc) $_2$ mediated oxidation. Based on

the experimental observations and insightful computational data, the mechanism of this double annulation is sketched in Fig. 5.⁴

The active Ru-catalyst {generated from [Ru(*p*-cymene)Cl $_2$] $_2$, AgSbF $_6$, and AcO $^-$ } first coordinates to MPS and activates the C(2)–H bond of **1a** to form **I** (**D** in Fig. 3). The coordination of alkyne to **I** and its migratory insertion leads to **II** (**F** in Fig. 3).

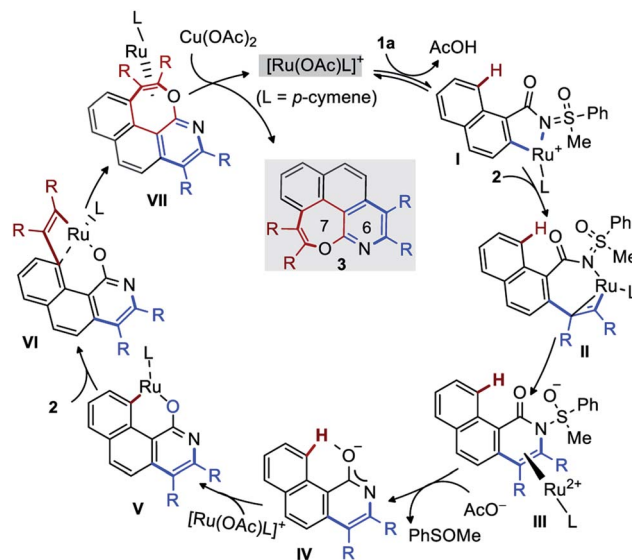


Fig. 5 Plausible catalytic cycle.

Next, the intramolecular nucleophilic addition to the N=S bond provides **III** (**G** in Fig. 3), which is the rate-determining step of the mono-annulation. The acetate-aided expulsion of [Ru(OAc)L]⁺ and elimination of PhSOMe leads to pyridone species **IV** (**H** in Fig. 3). Next, direct C(8)–H ruthenation of **IV** affords **V** (**M** in Fig. 4). Then, alkyne insertion into **V** generates the unusual ruthena-oxabicyclooctene complex **VI** (**O** in Fig. 4). The reductive elimination of **VI** gives **VII** (**P** in Fig. 4) and is the rate-determining step of the second annulation. Finally, Cu(OAc)₂ mediated transfer of RuL moiety to **1a** liberates the desired [6,7]-fused oxepino-pyridine skeleton.

Conclusion

In summary, we have developed an unprecedented Ru-catalyzed sulfoximine-directed one-pot domino $\{[4 + 2] \& [5 + 2]\}$ double annulation of 1-naphthoic acid derivatives with alkynes for the synthesis of unique [6,7]-fused oxepino-pyridine motifs. This transformation functionalizes both chemically distinct *ortho*- and *peri*-C–H bonds of fused-hetero(arenes) through double annulation, making four (C–C & C–N and C–C & C–O) bonds in a single operation. In addition, two-step unsymmetrical annulations with different alkynes are also shown. The detailed DFT calculations endorse the participation of metal-cyclopropene and ruthena-oxabicyclooctene intermediates. The construction of biologically relevant drugs anchored oxepino-pyridine scaffolds, broad scope, and gram scale synthesis make the transformation synthetically viable.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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Notes and references

- (a) M. D. Burke and S. L. Schreiber, *Angew. Chem., Int. Ed.*, 2004, **43**, 46; (b) W. R. J. D. Galloway, A. Isidro-Llobet and D. R. Spring, *Nat. Commun.*, 2010, **1**, 80; (c) S. Yi, B. V. Varun, Y. Choi and S. B. Park, *Front. Chem.*, 2018, **6**, 507.
- (a) K. Narita, K. Nakamura, Y. Abe and T. Katoh, *Eur. J. Org. Chem.*, 2011, 4985; (b) N. M. O'Boyle, I. Barrett, L. M. Greene, M. Carr, D. Fayne, B. Twamley, A. J. S. Knox, N. O. Keely, D. M. Zisterer and M. J. Meegan, *J. Med. Chem.*, 2018, **61**, 514.
- (a) S. K. Sinha, G. Zannoni and D. Maiti, *Asian J. Org. Chem.*, 2018, **7**, 1178; (b) D. J. Abrams, P. A. Provencher and E. J. Sorensen, *Chem. Soc. Rev.*, 2018, **47**, 8925.
- (a) T. Satoh and M. Miura, *Chem.–Eur. J.*, 2010, **16**, 11212; (b) J. Wencel-Delord, T. Dröge, F. Liu and F. Glorius, *Chem. Soc. Rev.*, 2011, **40**, 4740; (c) P. B. Arockiam, C. Bruneau and P. H. Dixneuf, *Chem. Rev.*, 2012, **112**, 5879; (d) L. Ackermann, *Acc. Chem. Res.*, 2014, **47**, 281; (e) M. Gulías and J. L. Mascareñas, *Angew. Chem., Int. Ed.*, 2016, **55**, 11000; (f) C. Sambiasi, D. Schönbauer, R. Blicke, T. Dao-Huy, G. Pototschnig, P. Schaaf, T. Wiesinger, M. F. Zia, J. Wencel-Delord, T. Besset, B. U. W. Maes and M. Schnürch, *Chem. Soc. Rev.*, 2018, **47**, 6603; (g) G. Duarah, P. P. Kaishap, T. Begum and S. Gogoi, *Adv. Synth. Catal.*, 2019, **361**, 654.
- (a) Y. Su, M. Zhao, K. Han, G. Song and X. Li, *Org. Lett.*, 2010, **12**, 5462; (b) L. Ackermann, A. V. Lygin and N. Hofmann, *Angew. Chem., Int. Ed.*, 2011, **50**, 6379; (c) B. Li, H. Xu, S. Feng and B. Wang, *Chem.–Eur. J.*, 2011, **17**, 12573; (d) L. Ackermann, J. Pospech, K. Graczyk and K. Rauch, *Org. Lett.*, 2012, **14**, 930; (e) H. Wang, C. Grohmann, C. Nimphius and F. Glorius, *J. Am. Chem. Soc.*, 2012, **134**, 19592; (f) W. Dong, L. Wang, K. Parthasarathy, F. Pan and C. Bolm, *Angew. Chem., Int. Ed.*, 2013, **52**, 11573; (g) M. R. Yadav, R. K. Rit, M. Shankar and A. K. Sahoo, *J. Org. Chem.*, 2014, **79**, 6123.
- (a) S. Mochida, M. Shimizu, K. Hirano, T. Satoh and M. Miura, *Chem.–Asian J.*, 2010, **5**, 847; (b) V. S. Thirunavukkarasu, M. Donati and L. Ackermann, *Org. Lett.*, 2012, **14**, 3416; (c) J. D. Dooley, S. Reddy Chidipudi and H. W. Lam, *J. Am. Chem. Soc.*, 2013, **135**, 10829; (d) X. Zhang, W. Si, M. Bao, N. Asao, Y. Yamamoto and T. Jin, *Org. Lett.*, 2014, **16**, 4830; (e) S. Reddy Chidipudi, D. J. Burns, I. Khan and H. W. Lam, *Angew. Chem., Int. Ed.*, 2015, **54**, 13975; (f) S. Moon, Y. Nishii and M. Miura, *Org. Lett.*, 2019, **21**, 233.
- (a) J. Yin, M. Tan, D. Wu, R. Jiang, C. Li and J. You, *Angew. Chem., Int. Ed.*, 2017, **56**, 13094; (b) J. Yin and J. You, *Angew. Chem., Int. Ed.*, 2019, **58**, 302.
- (a) A. Seoane, N. Casanova, N. Quiñones, J. L. Mascareñas and M. Gulías, *J. Am. Chem. Soc.*, 2014, **136**, 834; (b) B. Cendón, N. Casanova, C. Comanescu, R. Garcia-Fandino, A. Seoane, M. Gulías and J. L. Mascareñas, *Org. Lett.*, 2017, **19**, 1674.
- G. Favini, *J. Mol. Struct.*, 1983, **93**, 139.
- (a) X. Tan, B. Liu, X. Li, B. Li, S. Xu, H. Song and B. Wang, *J. Am. Chem. Soc.*, 2012, **134**, 16163; (b) J. Jayakumar, J. K. Parthasarathy, Y.-H. Chen, T.-H. Lee, S.-C. Chuang and C.-H. Cheng, *Angew. Chem., Int. Ed.*, 2014, **53**, 9889; (c) D. Ghorai and J. Choudhury, *ACS Catal.*, 2015, **5**, 2692; (d) M. Shankar, K. Ghosh, K. Mukherjee, R. K. Rit and A. K. Sahoo, *Org. Lett.*, 2018, **20**, 5144; (e) T. Guntreddi, M. Shankar, N. Kommu and A. K. Sahoo, *J. Org. Chem.*, 2019, **84**, 13033.
- (a) S. Mochida, N. Umeda, K. Hirano, T. Satoh and M. Miura, *Chem. Lett.*, 2010, **39**, 744; (b) G. Song, D. Chen, C.-L. Pan, R. H. Crabtree and X. Li, *J. Org. Chem.*, 2010, **75**, 7487; (c) M. Shankar, K. Ghosh, K. Mukherjee, R. K. Rit and A. K. Sahoo, *Org. Lett.*, 2016, **18**, 6416; (d) K. Ghosh,



- M. Shankar, R. K. Rit, G. Dubey, P. V. Bharatam and A. K. Sahoo, *J. Org. Chem.*, 2018, **83**, 9667.
- 12 (a) D. Sarkar, F. S. Melkonyan, A. V. Gulevich and V. Gevorgyan, *Angew. Chem., Int. Ed.*, 2013, **52**, 10800; (b) K. Ghosh, R. K. Rit, E. Ramesh and A. K. Sahoo, *Angew. Chem., Int. Ed.*, 2016, **55**, 7821; (c) A. Mandal, S. Dana, D. Chowdhury and M. Baidya, *Chem.-Asian J.*, 2019, **14**, 4074.
- 13 The release of sulfoxide from the sterically encumbered metallacyclic RuL-MPS moiety (Int-II or Int-III; Fig. 5) and the direct participation of Int-IV presumably helps the annulation.
- 14 See the ESI.†
- 15 CCDC 1979334 (**3ha**), CCDC 1979335 (**3ja**), and CCDC 1979336 (**8ae**) contain the supplementary crystallographic data for this paper.†
- 16 T. Cernak, K. D. Dykstra, S. Tyagarajan, P. Vachael and S. W. Krska, *Chem. Soc. Rev.*, 2016, **45**, 546.
- 17 M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery Jr, J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, O. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski and D. J. Fox, *Gaussian 09, revision D.01*, Gaussian, Inc., Wallingford CT, 2013.
- 18 (a) T. H. Dunning Jr and P. J. Hay, *Modern Theoretical Chemistry*, ed. H. F. Schaefer III, Plenum, New York, 1997, vol. 3; (b) P. J. Hay and W. R. Wadt, *J. Chem. Phys.*, 1985, **82**, 270; (c) W. R. Wadt and P. J. Hay, *J. Chem. Phys.*, 1985, **82**, 284; (d) P. J. Hay and W. R. Wadt, *J. Chem. Phys.*, 1985, **82**, 299; (e) D. Andrae, U. Häußermann, M. Dolg, H. Stoll and H. Preuß, *Theor. Chim. Acta*, 1990, **77**, 123.

