

Cite this: *Chem. Sci.*, 2020, **11**, 7158

All publication charges for this article have been paid for by the Royal Society of Chemistry

Small molecule-mediated co-assembly of amyloid- β oligomers reduces neurotoxicity through promoting non-fibrillar aggregation†

Hao Liu,^a Chengyuan Qian,^a Tao Yang,^b Yanqing Wang,^d Jian Luo,^a Changli Zhang,^e Xiaohui Wang,^{ib} *ac Xiaoyong Wang,^{ib} b and Zijian Guo,^{ib} *c

Amyloid- β (A β) oligomers, particularly low molecular weight (LMW) oligomers, rather than fibrils, contribute very significantly to the onset and progression of Alzheimer's Disease (AD). However, due to the inherent heterogeneity and metastability of oligomers, most of the conventional anti-oligomer therapies have indirectly modulated oligomers' toxicity through manipulating A β self-assembly to reduce oligomer levels, which are prone to suffering from the risk of regenerating toxic oligomers from the products of modulation. To circumvent this disadvantage, we demonstrate, for the first time, rational design of rigid pincer-like scaffold-based small molecules with blood-brain barrier permeability that specifically co-assemble with LMW A β oligomers through directly binding to the exposed hydrophobic regions of oligomers to form non-fibrillar, degradable, non-toxic co-aggregates. As a proof of concept, treatment with a europium complex (EC) in such a structural mode can rescue A β -mediated dysfunction in *C. elegans* models of AD *in vivo*. This small molecule-mediated oligomer co-assembly strategy offers an efficient approach for AD treatment.

Received 21st January 2020

Accepted 19th June 2020

DOI: 10.1039/d0sc00392a

rsc.li/chemical-science

Introduction

The 'toxic A β oligomer' hypothesis has attracted more and more attention as a way of overcoming the dilemma of amyloid- β (A β) plaque-directed therapeutics for Alzheimer's Disease (AD), in which all phase III clinical trials have failed, even though the levels of A β plaques in the brain can be significantly reduced.^{1,2} The hypothesis asserts that A β oligomers rather than fibrils might be the central player in AD pathogenesis and have been recognized as the most toxic species with strong correlation with disease symptoms.^{3,4} Importantly, the oligomeric species can be produced in the earliest stages of AD and even appear as early as ~10–15 years before the clinical symptoms.⁵ Therefore, A β oligomers seem to be the most promising targets for disease-modifying therapies of AD particularly at the early presymptomatic stage. However, the oligomers are heterogeneous and metastable in nature, which has

seriously restricted the development of anti-oligomer therapies. To date, the majority of therapeutic approaches have focused on indirect modulation of oligomers' toxicity through immunotherapy⁶ and manipulating A β self-assembly,^{7,8} including inhibition of oligomerization,^{9,10} acceleration of fibril formation,^{11,12} redirection of A β into off-pathway aggregates,^{13–16} and disaggregation of oligomers.¹⁷ However, the generated A β monomers or fibrils in such treatments might function as reservoirs for oligomers, which probably lead to recurring neurotoxicity.

Multicomponent co-assembly of peptides is an emerging supramolecular strategy, which presents an opportunity to irreversibly alter the properties of the self-assemblies of individual peptides *via* co-assembly.¹⁸ Considering the causative link between the structure of oligomers and their toxicity,^{19,20} this strategy would be effective to reduce oligomers' toxicity through forming co-aggregates that are structurally distinct from A β self-aggregates. Recently, a few polymers and peptides have plausibly succeeded in co-assembling with A β into non-toxic co-aggregates, depending on their abundant binding sites for protein interactions.^{21–27} However, the deficient oligomer specificity limits their inhibitory effect on oligomers' toxicity, which hinders their further utilization *in vivo*. Alternatively, only very few small molecules have exhibited high specificity for oligomers derived from A β peptide.^{28–31} Nevertheless, none have enabled co-assembly with A β oligomers to form non-toxic co-aggregates, probably due to the lack of principle-based rational design of multicomponent co-assemblies of peptides. Despite being challenging, small molecule-mediated co-assembly with A β

^aCollege of Chemistry and Molecular Engineering, Nanjing Tech University, Nanjing 211816, P. R. China. E-mail: wangxhui@njtech.edu.cn

^bKey Laboratory of Pharmaceutical Biotechnology, School of Life Sciences, Nanjing University, Nanjing, 210093, P. R. China

^cState Key Laboratory of Coordination Chemistry, Nanjing University, Nanjing, 210093, P. R. China. E-mail: zguo@nju.edu.cn

^dInstitute of Environmental Toxicology and Environmental Ecology, Yancheng Teachers University, Yancheng 224007, P. R. China

^eSchool of Environmental Science, Nanjing Xiaozhuang College, Nanjing 211171, P. R. China

† Electronic supplementary information (ESI) available. See DOI: 10.1039/d0sc00392a

oligomers would be reasonably attainable to diminish oligomer-mediated toxicity, given the many advantages of small molecules, such as cheap and facile preparation, controllable oligomer specificity, tunable biocompatibility and blood–brain barrier (BBB) permeability.⁸

Here we propose a rationally designed small molecule-mediated A β co-assembly that co-assembles A β oligomers and small molecular probes into non-toxic co-aggregates based on intermolecular interactions. Under the guidance of this concept, a pincer-like europium(III) complex (EC) with rigid configuration was constructed as a representative (Fig. 1A). The EC shows excellent oligomer specificity through binding to the exposed hydrophobic regions of oligomers as the oligomer specific binding sites that are responsible for both toxicity and aggregation propensity of oligomers. Based on the specific recognition and directional interactions, the EC is able to accelerate and promote A β aggregation into non-fibrillar EC–A β co-aggregates, which are almost non-toxic in neuroblastoma cells and degradable in microglial cells through upregulating autophagy upon phagocytosis. Strikingly, the EC can also ameliorate A β -mediated toxicity in *Caenorhabditis elegans* (*C. elegans*) models of AD through co-assembly *in vivo*. Our work for the first time establishes a small molecule-mediated A β oligomer co-assembly that can irreversibly diminish oligomers' toxicity as a potential therapeutic approach for AD treatment.

Results and discussion

Design of the EC with oligomer specificity

Both specific recognition and directional interactions are important parameters for small molecules to achieve co-

assembly with peptides.³² First, given that the exposure of hydrophobic regions to solvents has been recognized as a unique structural feature of A β oligomers in contrast to other counterparts, the specific recognition of small molecules for oligomers can be identified through testing their binding affinity to such solvent-exposed hydrophobic patches.^{28,33} 6-Iodo-2-(4-dimethylamino)phenylimidazo[1,2-*a*]pyridine (IMPY), a single-photon emission computed tomography imaging probe for amyloid plaques,³⁴ has exhibited favorable pharmacokinetic properties with good BBB permeability *in vivo*.³⁵ Inspired by these findings, an aromatic 4-(imidazo[1,2-*a*]pyridin-2-yl) aniline (IPA) group was selected as the targeting group in view of the binding affinity of its scaffold to the central hydrophobic regions of A β . On the other hand, rigid configuration would be necessary for the small molecules to consolidate the directional interactions with mutable A β oligomer and to stabilize the co-assembled structure. For this purpose, a diethylene-triaminepentaacetate (DTPA)-based Eu(III) chelating center was chosen as a bridge to link two IPA groups to form a rigid pincer-like complex, the EC (Fig. 1A). The chelating center with multiple coordinate bonds can endow the EC with high stability and rigidity. The optimized structure of the EC based on quantum mechanical calculations at the B3LYP/6-31G* level shows that two IPA groups as pincer arms are firmly restricted by the chelating center (Fig. 1A). Another reason for the selection of the Eu(III) center is that the luminescence of Eu(III) can be used to verify the oligomer specificity of the EC. Finally, the A β toxicity could be significantly suppressed by the proposed co-assembly system because of the coverage of the hydrophobic motifs of oligomers.^{33,36} The synthetic route and characterization of the EC are described in Scheme S1 and Fig. S1 (ESI).†

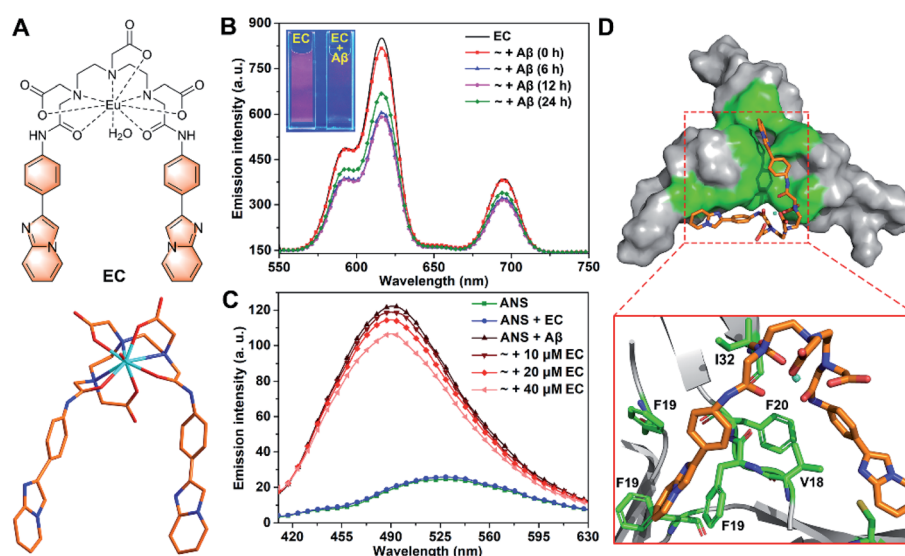


Fig. 1 (A) Structural formula of the EC (top) and optimized structure of the EC at the B3LYP/6-31G* level (bottom). (B) Time-resolved luminescence spectra of the EC (20 μ M, λ_{ex} = 300 nm) in the absence and presence of A β 40 species (20 μ M) with different pre-incubation times (0, 6, 12, and 24 h) in buffer (20 mM Tris–HCl, 150 mM NaCl, 4% v/v DMSO, pH 7.4). Inset shows the luminescence images of EC solutions with or without A β oligomers under a UV lamp at 254 nm. (C) Fluorescence spectra of ANS (20 μ M, λ_{ex} = 380 nm) in the absence and presence of A β 40 oligomers (20 μ M) with or without 10, 20, and 40 μ M EC in buffer (20 mM Tris–HCl, 150 mM NaCl, 4% v/v DMSO, pH 7.4). (D) Lowest energy docking model of the EC with A β trimer (PDB ID: 5HOX) in surface (top) and cartoon (bottom) representations. The amino acid residues interacting with the EC are highlighted in green.



Because Lipinski's rules do not necessarily apply to metal-lodrugs, the BBB permeability of the EC was evaluated from the quantity of Eu in the mouse brain by inductively coupled plasma mass spectrometry (ICP-MS) after i.v. injection. The amount of Eu rapidly increased to its peak level at 10 min after dosing, followed by a relatively slow decrease (Fig. S2†), suggesting that the EC is able to cross the BBB.

Next, the luminescence responses of the EC for A β species were tested, which would provide useful evidence for the selectivity of the EC for oligomers. Different A β aggregates were prepared *via* pre-incubation for corresponding times according to the aggregation states recognized by the growth kinetics curve in the Thioflavin T (ThT) fluorescence assay using the dye ThT as a fluorescence probe for fibrils (Fig. S3†).^{37,38} Similar excitation and absorption spectra of the EC consolidate the energy transfer from IPA groups as antennae to Eu³⁺ ions (Fig. S4†).³⁹ Consequently, three characteristic emission bands of europium were observed at 592, 615, and 695 nm upon excitation of the EC at 300 nm by time-resolved luminescence measurements, which can be assigned to the transition from the ⁵D₀ excited state to the ⁷F₁, ⁷F₂, and ⁷F₄ ground states of Eu³⁺, respectively (Fig. 1B).⁴⁰ The emission intensity of the EC remarkably decreased in the presence of abundant low molecular weight (LMW) oligomers with pre-incubation for 6 and 12 h, while the luminescence of the EC partially recovered in the presence of A β species with pre-incubation for 24 h, in which the amount of LMW oligomers decreased with conversion into high molecular weight (HMW) oligomers and protofibrils. In contrast, a slight decrease of EC luminescence was observed in the presence of A β without any pre-incubation (0 h), which contains few LMW oligomers (*vide infra*). The results indicate that the luminescence response of the EC depends on the amount of LMW oligomers. Interestingly, such a luminescence change of the EC was clearly visible under a UV lamp (254 nm), where bright red emission was almost quenched by LMW oligomers (inset of Fig. 1B). Furthermore, the stronger luminescence of the EC in D₂O than in H₂O both in the presence and absence of A β oligomers indicates that the coordinated water still binds to Eu³⁺ (Fig. S5†),^{39,41} thereby excluding the possibility of coordination between Eu³⁺ and A β . In the presence of LMW oligomers, non-covalent interactions between IPA groups and hydrophobic regions of A β may take place to favor the photo-electron transfer from the amino acid residues to the IPA groups,⁴² thereby preventing the energy transfer from the antenna to Eu³⁺ and eventually quenching EC luminescence. Therefore, IPA-involved high affinity binding to LMW oligomers would dominantly contribute to the selectivity of the EC for LMW oligomers.

To validate the binding mode of the EC with A β oligomers, an 8-anilino-naphthalene-1-sulfonate (ANS) fluorescence competition assay was carried out. ANS has been shown to bind to solvent-exposed hydrophobic patches of A β oligomers with remarkable fluorescence enhancement and a blue shift of its maximum emission wavelength.¹⁹ As shown in Fig. 1C, the oligomer-induced strong fluorescence of ANS decreased with addition of an increasing amount of the EC, indicating that the EC may displace ANS to bind with the hydrophobic regions of

oligomers, because the EC *per se* hardly affects the fluorescence of ANS. To further understand the structural features of the EC for the selective binding to LMW A β oligomers, docking studies were performed using a well-known A β 17-36 trimer as a working model (PDB ID: 5HOX).⁴³ In the trimer, a large hydrophobic surface consists of hydrophobic residues L17, F19, and V36 of three β -hairpins packing against the residues A21, I32, and L34 of the adjacent chains, which should be the oligomer specific binding motif, although it may not completely represent the structure of full-length A β oligomers. In the lowest energy conformation of the adduct, the EC adopting a slightly twisted geometry is positioned in the hydrophobic domain mainly through π - π stacking interactions between the planar IPA groups and the side chains of F19 residues (Fig. 1D), while the DTPA group also forms intermolecular CH \cdots O interaction with the I32 side chain upon binding. Moreover, the EC displayed the lowest binding free energy ($\Delta G = -43.39$ kJ mol⁻¹) and inhibition constant ($K_i = 24.85$ nM) for the trimer than for both the monomer and fibrils through thermodynamics analysis (Table S1†), indicating the specificity of the EC for the oligomeric conformation of A β , which could be dominantly attributed to the pincer-like configuration decorated with hydrophobic motif-binding ligands.

The EC dramatically promotes A β aggregation

On the basis of the LMW oligomer specificity of the EC, we next investigated the effect of the EC on A β aggregation. In this work, the ThT fluorescence assay is unable to monitor the aggregation of A β in the presence of the EC, since the EC can significantly enhance the ThT emission intensity with a red shift of the emission wavelength in both the presence and absence of A β (Fig. S6†). Alternatively, because there is no absorption of the EC itself at 405 nm (Fig. S4†), the turbidity assay was employed to measure the absorbance/turbidity at 405 nm (A_{405}) of the A β solution, which can reflect the degree of A β aggregation.^{44,45} Moreover, the time course of A β 40 self-aggregation by the turbidity assay was similar to that of the ThT fluorescence assay, adopting a sigmoidal mode that consists of a lag phase, an elongation phase, and a plateau phase, as reported in previous studies (Fig. S3†).^{37,38} Accordingly, the time-dependent turbidities of the A β 40 solution in the presence of different concentrations of the EC were determined. As shown in Fig. 2A, the resulting kinetic absorption curves were obtained from data fitting with a sigmoidal equation (eqn (S1)†),⁴⁶ while the time at half completion of the aggregation process (t_{50}) and the elongation rate constant (k) were calculated (Table S2†). Compared with A β 40 self-aggregation, the EC induced a slight increase of turbidity of the solution by ~ 2 -fold and a minor change in the values of t_{50} and k after incubation at concentrations of 5 and 10 μ M. Noticeably, significant enhancement of turbidity with a relatively short t_{50} (4.94 h) and fast aggregation rate ($k = 0.66$ h⁻¹) was observed after addition of 20 μ M EC. Moreover, 40 μ M EC had a similar effect on the turbidity with 20 μ M EC, but longer t_{50} (10.95) and slower aggregation rate ($k = 0.44$ h⁻¹). These results clearly indicate that the EC can not only accelerate the aggregation of A β 40 through shortening the lag phase and elongation



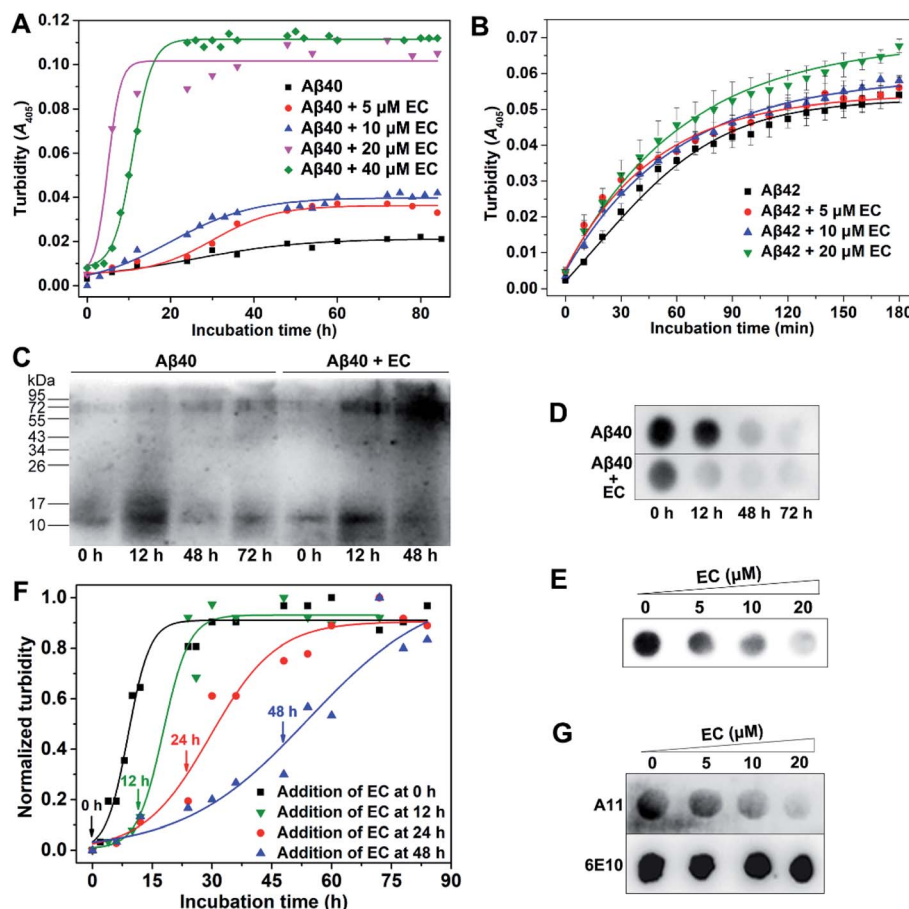


Fig. 2 (A) Representative profile of the aggregation kinetics of A β 40 (20 μ M) in the presence of 0, 5, 10, 20, or 40 μ M EC in buffer (20 mM Tris–HCl, 150 mM NaCl, 4% v/v DMSO, pH 7.4) at 37 $^{\circ}$ C by the turbidity assay. (B) The aggregation kinetics of A β 42 (10 μ M) in the presence of 0, 5, 10, or 20 μ M EC in buffer (20 mM Tris–HCl, 150 mM NaCl, 4% v/v DMSO, pH 7.4) at 37 $^{\circ}$ C by the turbidity assay. Error bars indicate \pm s.d. (n = 5 independent experiments). (C) Western blot analysis of the time-dependent effect of the EC (20 μ M) on A β 40 (20 μ M) aggregation using anti-A β antibody 6E10. (D) Dot blot analysis of the time-dependent effect of the EC (20 μ M) on soluble A β 40 species using 6E10 antibody. (E) Dot blot analysis of the concentration-dependent effect of the EC (20 μ M) on soluble A β 40 species after incubation for 24 h using 6E10 antibody. (F) Representative profile of turbidity-monitored aggregation kinetics of A β 40 (20 μ M) when the EC was added at different time points indicated by the arrows. (G) Detection of A β oligomers by dot blot probed with oligomer-specific antibody A11. Total A β 40 levels were detected by 6E10 antibody.

rate, but also boost the level of A β 40 aggregates. Besides A β 40, the effect of the EC on A β 42 aggregation was also measured by the turbidity assay (Fig. 2B). Due to the additional two C-terminal residues, A β 42 can enter into the plateau phase within a short time more quickly than A β 40. The EC showed a weaker promotive effect on the aggregation of A β 42 than of A β 40, probably owing to the ultra-fast self-aggregation kinetics of A β 42. Nevertheless, the EC-induced acceleration at the early stage of A β 42 aggregation can be obviously observed, similar to that of A β 40. Moreover, as the potential binding site of the EC, the central hydrophobic stretches are responsible for the aggregation in both A β 40 and A β 42.⁴⁷ Thus, it is suggested that the EC may interact with A β 42 in a similar mechanism to that with A β 40. Accordingly, A β 40 was used in most of the experiments in this work and is cheaper and seems more favorable for clarifying the mechanism of action of the EC under our experimental conditions.

To confirm the promotion of the EC on A β aggregation, sodium dodecyl sulfate polyacrylamide gel electrophoresis

(SDS-PAGE) followed by western blotting using anti-A β antibody 6E10 was employed in a time-dependent manner. As shown in Fig. 2C, in the absence of the EC, few LMW oligomers, notably dimers and trimers (10–17 kDa), were observed at the start of A β aggregation, accompanied by a remarkable increase at the early stage of the lag phase within 12 h due to the propensity for A β self-aggregation, which obviously decreased after 48 h of incubation in the elongation phase. Meanwhile, the larger aggregates (>70 kDa) of A β increased. In contrast, a higher amount of large aggregates was observed in the presence of the EC (20 μ M) as early as 12 h, whereas the LMW oligomers almost disappeared at 48 h. This result is consistent with those of turbidity assays. A similar result was also obtained from the western blot analysis of the concentration-dependent effect of the EC on A β aggregation (Fig. S7[†]). Dot blot assays using 6E10 were further performed. The total amounts of A β in whole solutions for different incubation times are almost identical (Fig. S8[†]). In contrast, the soluble A β species in the supernatant of the



incubating solutions dramatically decreased at 12 h in the presence of the EC (20 μ M), compared with that of the A β self-aggregation solution (Fig. 2D), indicating that the EC induced more aggressive aggregation of A β into insoluble species. Identical to the results of turbidity and western blot assays, the EC exhibited the most efficient promotion of A β aggregation at 20 μ M with the lowest level of soluble A β aggregates in the supernatant (Fig. 2E).

The EC preferentially targets soluble LMW A β oligomers

We next examined whether the EC targets oligomers to promote A β aggregation. First, the promotion effect of the EC on A β aggregation upon addition of the complex at different time points after the start of A β aggregation was measured. As shown in Fig. 2F, the EC exhibited similar acceleratory efficacy when added at initiation (0 h) and in the middle of the lag phase (12 h) of A β aggregation. In contrast, the efficacy gradually decreased when added at the late stage of the lag phase (24 h) and elongation phase (48 h) of A β aggregation, which was confirmed by t_{50} , k , and the lag time (t_{lag}) values (Table S3†). LMW oligomers already existed at the start of A β aggregation and obviously increased within 12 h (Fig. 1B and 2C). Instead, HMW oligomers and insoluble large aggregates including protofibrils and fibrils can be recognized as the dominant constituents at 24 h and 48 h of A β aggregation, respectively

(Fig. 2C and S3†). Given its binding selectivity, it is reasonable that the EC can still preferentially target LMW oligomers in the dynamic process of A β aggregation. As a consequence, the lag phase of A β aggregation was significantly shortened with reduced t_{lag} , thereby accelerating the A β aggregation.

The binding behavior of the EC with oligomers during A β aggregation was further confirmed by the dot blot assay using A β oligomer-specific antibody A11 that selectively recognizes soluble A β oligomers.⁴⁸ 6E10 was also used to verify equal sample loading. As shown in Fig. 2G, A11-sensitive oligomers gradually decreased with the increasing concentration of the EC and were almost diminished in the presence of 20 μ M EC upon incubation for 24 h. In contrast, no change in the signal intensity was observed for the samples stained with 6E10. These results support the view that the EC can target oligomers and promote the conversion of oligomers into large conformers, which cannot be recognized by the A11 antibody.

The EC rapidly converts oligomers into non-fibrillar aggregates

To further study the morphology of EC-induced A β aggregates, transmission electron microscopy (TEM) analysis was carried out. The images of A β samples with or without the EC after incubation for 24 h at 37 $^{\circ}$ C are shown in Fig. 3A. Consistent with the result of ThT-monitored aggregation kinetics (Fig. S3†), short and slender

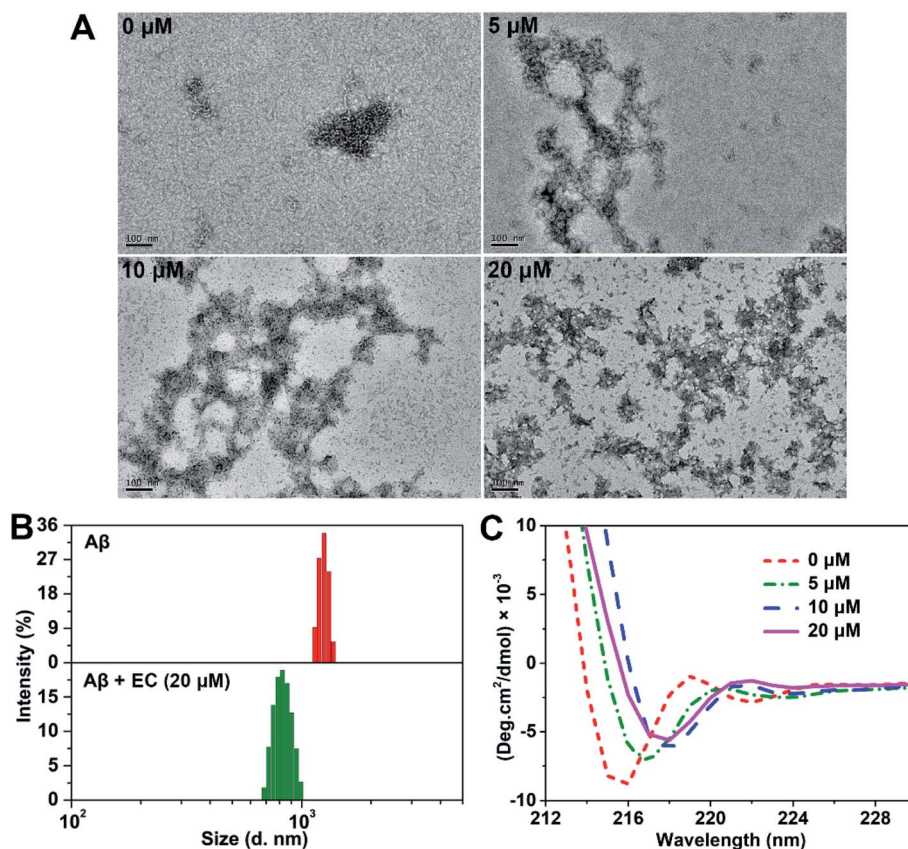


Fig. 3 (A) Morphological analysis of A β 40 aggregates in the presence of 0, 5, 10, or 20 μ M EC by TEM (scale bar = 100 nm). (B) DLS analysis of A β 40 fibrils and EC-induced A β aggregates. (C) CD spectra of A β 40 (20 μ M) in the absence or presence of different concentrations of the EC (20 μ M) after incubation for 24 h. All samples were incubated in a buffer (20 mM Tris-HCl, 150 mM NaCl, 4% v/v DMSO, pH 7.4) at 37 $^{\circ}$ C.



string-like aggregates including HMW oligomers and protofibrils were seen in the solution of A β 40 in the absence of the EC, which gradually decoupled in the samples with an increasing concentration of the EC. Instead, more and more amorphous aggregates can be observed after incubating A β 40 with the EC. While 20 μ M EC totally remodeled the fibrillary species into massive smear-like disordered aggregates. Similar morphological changes can be obtained upon co-incubation of the EC with fresh A β or 12 h-incubated A β for 72 or 60 h, respectively (Fig. S9†). The effect of the EC (20 μ M) on the size distribution of A β 40 was further tested by dynamic light scattering (DLS). Both samples with and without the EC were pre-incubated to their plateau phases in order to obtain the EC-generated aggregates and mature fibrils as dominant species, respectively. As shown in Fig. 3B, the mean hydrodynamic radius of the EC-treated A β sample was around 380 nm, which was much smaller than that of A β fibrils (\sim 600 nm). These observations suggest that the EC not only promotes A β aggregation but also converts LMW oligomers into non-fibrillar aggregates instead of fibrils.

Circular dichroism (CD) spectroscopy was used to evaluate the influence of EC-induced conversion of A β aggregates on the characteristic β -sheet conformation.⁴⁹ The negative band (\sim 195 nm) of the random coil structure for the 0 h sample did not appear, probably resulting from the influence of the formed LMW oligomers under our experimental conditions during data collection. The typical β -sheet structure characterized by one single minimum at \sim 216 nm,⁵⁰ however, was observed after incubation of A β for 12 h, which would belong to the oligomers produced in the middle of the lag phase. A significant decrease of the β -sheet structure can be seen in the presence of the EC (20 μ M) after 12 h of co-incubation (Fig. S10†). Moreover, EC treatment can also reduce the negative band (\sim 216 nm) with red shifts in a concentration-dependent manner (Fig. 3C), which was substantially consistent with the results of the above EC-promoted A β aggregation experiments. Taking into account all the observations, we deduce that EC-mediated promotion of A β aggregation can lead to the formation of non-fibrillar aggregates with weak β -sheet conformation, which are structurally distinct from A β fibrils.

Molecular mechanism of EC-mediated A β co-aggregation

We then investigated the microscopic molecular mechanism of EC-mediated promotion of A β aggregation. A growing body of kinetic research has revealed that A β aggregation is a supra-molecular self-assembly process undergoing multiple mechanisms substantially divided into three categories: nucleation and fragmentation processes, growth processes, and dissociation processes.^{51,52} In comparison, the nucleation and fragmentation processes are determinative factors for the aggregation levels and rate, which can be classified into primary pathways and secondary pathways based on their dependencies on the concentrations of free monomers and of existing aggregates, respectively.⁵³ The addition of preformed seeds can accelerate A β aggregation *via* secondary nucleation pathways.^{9,12,13,25} Accordingly, a pre-seeded assay was performed through monitoring turbidity. As shown in Fig. 4A, addition of preformed fibers as seeds (10%, v/v) to an unseeded A β 40 solution accelerated the aggregation by shortening the t_{50} from 5.40 ± 0.82 to 3.80 ± 0.47 h, confirming the dominant role of the secondary process in A β 40 self-fibrillation. Since the EC can promote A β aggregation to form non-fibrillar aggregates through targeting LMW oligomers, EC-treated samples are presumed unable to seed the fibrillization of unfolded monomers. However, to our surprise, EC-induced non-fibrillar aggregates can also promote A β aggregation to an even higher extent than EC-untreated seeds, yielding a lower t_{50} of 2.35 ± 0.16 h. The result implies that EC-promoted A β aggregation may also mainly adopt a secondary nucleation pathway for formation of non-fibrillar aggregates. To verify this speculation, aggregation kinetics depending on the initial A β 40 concentration with or without the EC were further investigated (Fig. 4B and S11†). The initial concentration dependence of t_{50} can be described by the scaling exponent, γ , from the power law, $t_{50} \sim [\text{A}\beta]^\gamma$, which is commonly used to predict the dominant nucleation mechanism.^{54,55} In this regard, a completely monomer-independent secondary nucleation pathway is characterized by a value of $\gamma = -0.5$, whereas a fully monomer-dependent process yields an exponent of $\gamma = -1.5$.^{38,51} A linear relationship was obtained when the power law was plotted as a double

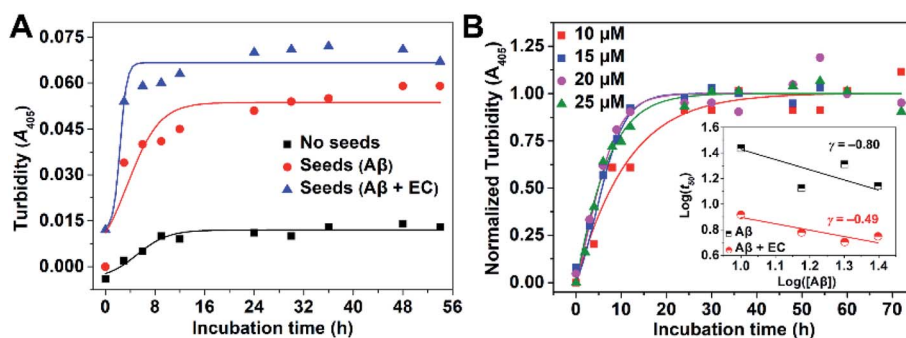


Fig. 4 (A) Representative profile of seed-catalyzed aggregation kinetics of A β 40 (20 μ M) in the absence and presence of preformed seeds (10%, v/v) with or without the EC by the turbidity assay. (B) Representative profile of turbidity-monitored aggregation kinetics of A β 40 with different initial concentrations (10, 15, 20, or 25 μ M) in buffer (20 mM Tris-HCl, 150 mM NaCl, 4% v/v DMSO, pH 7.4) at 37 $^{\circ}$ C. The inset shows the logarithm of the time at half completion of the aggregation (t_{50}) as a function of logarithm of initial A β concentration. The scaling exponent can be obtained by linear fitting.



logarithmic plot, in which the slope is equal to γ (inset of Fig. 4B). In the case of A β 40 itself with a concentration interval of 10–25 μM , the scaling exponent was found to be -0.80 , suggesting that the nucleation process would depend on both monomers and aggregates, which is indicative of surface-catalyzed secondary nucleation on existing aggregates.^{53,56} In contrast, the presence of the EC resulted in an exponent of $\gamma = -0.49$, strongly implying the saturated secondary nucleation mechanism in accordance with the poor dependence on monomers. The results suggest that the assembly of EC-involved hetero-oligomers may dominate the further aggregation of A β .

To gain deeper insight into EC-mediated assembly processes of A β , we performed computational simulations of the binding of the EC with two A β trimers *via* molecular docking. As shown in Fig. S12 (ESI),[†] the EC is anchored in the middle of two trimers by noncovalent interactions with the hydrophobic domain of the trimers, including π - π stacking interactions between IPA groups and the side chains of F19 residues as well as intermolecular CH \cdots O interactions between DTPA group and L17, L34, and V36. In addition, lower ΔG ($-45.98 \text{ kJ mol}^{-1}$) and K_i (8.75 nM) can be obtained for this triadduct than for the above-mentioned biadduct of the EC with A β trimer (Table S1[†]). The results imply that the EC may be able to co-assemble with the A β trimer in alternate binding mode. The rigid scaffold of

the EC may play a critical role in maintaining the stability of EC-A β co-aggregates, which was verified by the effect of the ligand of the EC, L-3H, on A β aggregation. The result demonstrated that L-3H with a flexible configuration can hardly influence A β aggregation (Fig. S13[†]), suggesting the indispensability of the rigidity of small molecules for co-assembly with A β oligomers.

Taken together, we proposed a mechanistic model as depicted in Fig. 5 for EC-mediated co-assembly of LMW oligomers. The π - π stacking interactions and hydrogen bonding between the EC and the exposed hydrophobic motifs of oligomers trigger the bicomponent co-assembly in the alternate binding mode to disrupt A β self-assembly and rapidly convert A β into EC-containing non-fibrillar architectures. Given the seed-like feature of the peculiar co-aggregates, the secondary nucleation mechanism may govern the co-assembly, that is, the preformed co-aggregates can act as seeds to catalyze formation of secondary EC-involved hetero-oligomers, ultimately promoting A β aggregation with greater degrees than A β self-assembly through accelerating secondary nucleation and elongation processes.

EC-mediated co-assembly of A β oligomers reduces cellular toxicity

To evaluate whether EC-mediated co-assembly of A β oligomers influences A β -induced toxicity, 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay was performed to determine the cell viability upon co-incubation of SH-SY5Y neuroblastoma cells with A β or/and different concentrations of the EC. More than 88% cell survival was still observed for cells treated with an EC concentration of 40 μM (Fig. 6A), indicating the minimal toxicity of the compound *per se* under such conditions. In contrast, the incubation of cells with A β oligomers resulted in an obvious reduction of cell viability to *ca.* 69%. The neurotoxicity, however, was significantly suppressed in the presence of the EC in a dose-dependent manner (Fig. 6B), which is positively related to the capacity of the EC for promoting A β aggregation at different concentrations. More than 96% cell survival can be observed for cells treated with up to 20 μM EC (Fig. 6B, $^{**}P < 0.01$), suggesting that the co-assembly with

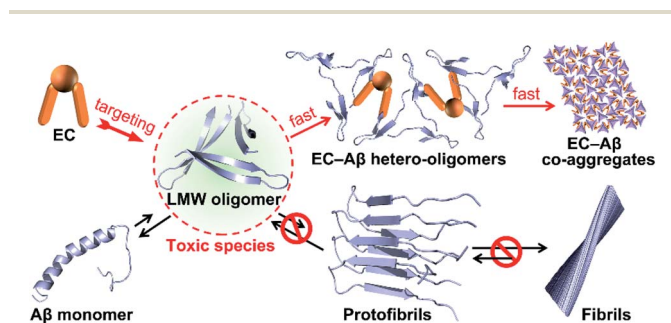


Fig. 5 Proposed mechanism of EC-mediated co-assembly of A β oligomers. The co-assembly blocks A β self-aggregation and redirects A β fibrils into non-fibrillar EC-A β co-aggregates.

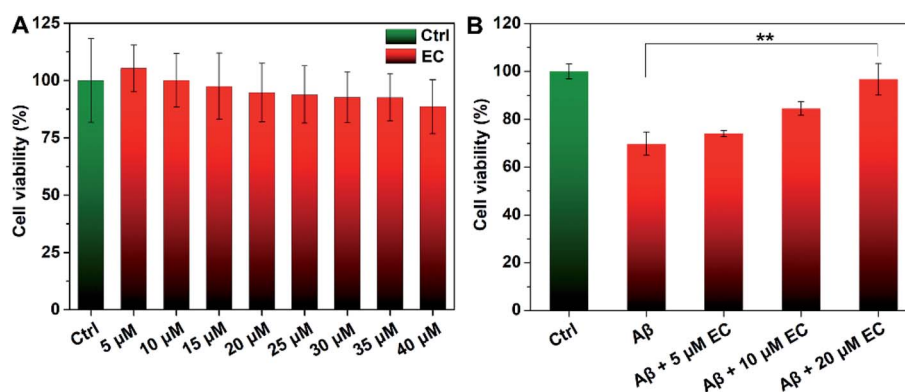


Fig. 6 Viability (%) of SH-SY5Y cells upon incubation with different concentrations of the EC in the absence (A) or presence (B) of A β (40 μM). The data was normalized and calculated as a percentage of untreated cells only containing 4% DMSO as a control. Error bars indicate \pm s.d. ($n = 5$ independent experiments). $^{**}P < 0.01$ (Student's *t*-test).



coverage of hydrophobic regions of oligomers by the EC can form EC-A β co-aggregates with subtle toxicity, thereby diminishing the oligomers' toxicity in the cell culture medium.

Phagocytosis of EC-mediated co-aggregates by microglia cells

Microglia-dependent phagocytosis is one of the major mechanisms of A β clearance.^{57,58} Previous studies suggested that the alteration of A β conformation would facilitate the removal of A β by glial cells.⁵⁹⁻⁶¹ Given the unique structure of EC-A β co-aggregates, the phagocytosis of the preformed aggregates in murine microglial cells (BV-2) was investigated by confocal laser scanning microscopy (CLSM). Immunofluorescence analysis of BV-2 cells with anti-A β antibody 6E10 showed that a much higher

level of A β uptake was observed from the cells incubated with co-aggregates for 4 h, compared to those incubated with A β fibrils, in which the fluorescence signals were mainly located outside the cells (Fig. 7A). By contrast, only an extremely faint fluorescence signal can be found around EC-treated cells in the absence of A β (Fig. S14†). The statistic results of fluorescent intensity in the cytosolic region also exhibited better internalization of EC-A β co-aggregates (Fig. 7B, *** P < 0.001), suggesting that the co-aggregates are more easily phagocytosed by microglial cells than A β fibrils, which was consistent with the result of dot blot analysis of cell lysates (Fig. 7C). Despite being controversial, autophagy has been plausibly recognized as the crucial pathway to degrade and clear the phagocytosed A β .⁵⁷ In order to evaluate

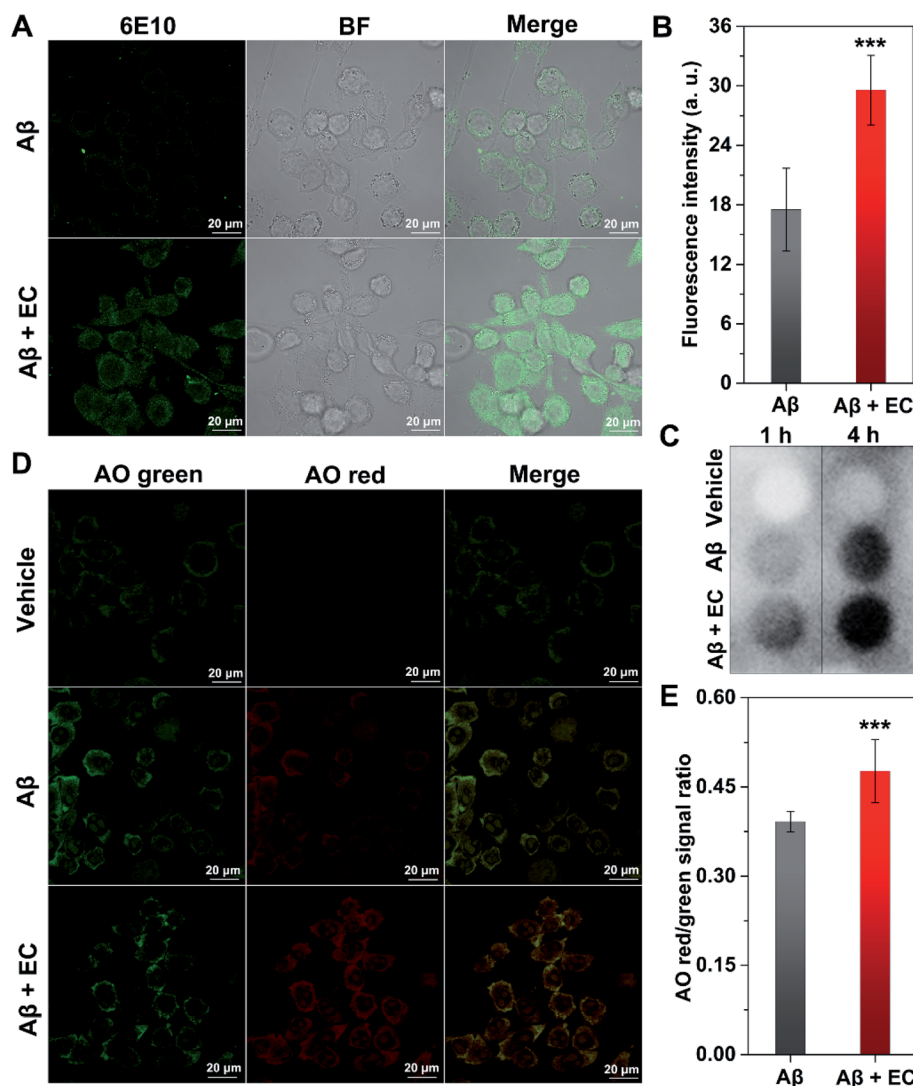
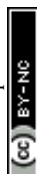


Fig. 7 (A) Immunofluorescence imaging of BV-2 cells treated with A β (10 μ M) in the absence and presence of the EC (10 μ M) for 4 h. (B) The fluorescence intensity in the internal regions of cells of A was quantified with ZEN microscope software. More than 20 cells were counted under each condition and data were analyzed by Student's t -test (*** P < 0.001). (C) Dot blot probed with 6E10 antibody to evaluate the uptake of preformed A β aggregates in the absence and presence of the EC in BV-2 cells. (D) Acridine Orange (AO) staining of BV-2 cells treated with A β (10 μ M) in the absence and presence of the EC (10 μ M) for 4 h. Green fluorescence was observed at Ex 488 nm, Em 500 to 560 nm; red fluorescence was acquired with Ex 488 nm, Em 600 to 700 nm using a Zeiss confocal microscope. (E) Quantification of the red/green intensity ratio in the cytosolic region of cells of (D). The red/green signal ratio was calculated using ZEN microscope software. More than 30 cells were counted under each condition and data were analyzed by Student's t -test (*** P < 0.001).



the autophagy effect of A β aggregates on BV-2 cells, Acridine Orange (AO), a pH-sensitive fluorescence probe for autophagy,⁶² was employed to stain the cells, followed by imaging using CLSM. AO is a cell-permeable green fluorophore that can accumulate in acidic vesicular organelles (AVOs) and convert into its protonated type with red fluorescence.⁶³ Thus, the red-to-green fluorescence intensity ratio of AO can be used to quantify the volume of AVOs, which increases during the cellular autophagy. As shown in Fig. 7D, the brightest red fluorescence was observed in the co-aggregate-treated cells after incubation for 4 h, compared to that of DMSO-treated and A β 40 fibril-treated cells. The quantification of fluorescence in the region of whole cells also revealed the higher red/green signal ratio of co-aggregate-treated cells than of A β 40 fibril-treated cells (Fig. 7E, *** P < 0.001). Based on the above findings, EC-mediated A β co-aggregates might be able to recover the phagocytosis of microglia, subsequently upregulating the autophagy of the cells to facilitate their degradation and clearance.

The EC rescues A β -mediated dysfunction in *C. elegans* models of AD

We further examined the effects of EC-mediated co-assembly of toxic A β 42 oligomers in a transgenic *Caenorhabditis elegans* (*C.*

elegans) model of AD *in vivo*, in which the human A β gene was expressed.⁶⁴ *C. elegans* strain CL4176 can express A β 42 in body wall muscle cells, which accumulates into a toxic deposit and results in progressive paralysis of the worms.⁶⁵ Accordingly, we exposed the CL4176 worms to different concentrations (20 and 40 μ M) of the EC and measured their rate of paralysis, which is indicative of the effect of the EC on A β -associated fitness of the worms. Given the preferential interactions between the EC and LMW A β oligomers, worms at the L2 larval stage of the life cycle where A β aggregates have not formed yet⁵³ were selected with the temperature increased from 15 $^{\circ}$ C to 25 $^{\circ}$ C to induce A β 42 expression. In contrast to untreated worms, significant delays of the A β 42-induced paralysis rate to a similar extent were observed in the CL4176 worms fed with 20 and 40 μ M EC (Fig. 8A), which is consistent with the results of the concentration-dependent promotive effect of the EC on A β aggregation. Conversely, no paralysis can be found in the wild-type worm model CL802 that does not express A β in both the absence and presence of the EC (Fig. 8A). The effect of the EC on the motility of CL4176 is visually shown in Fig. 8B and Movies S1–S3 (ESI).† After living at 25 $^{\circ}$ C for 44 h, most of the untreated worms were paralyzed in the straight line shape, but only moved the head *in situ*. In contrast, some EC-treated worms can still

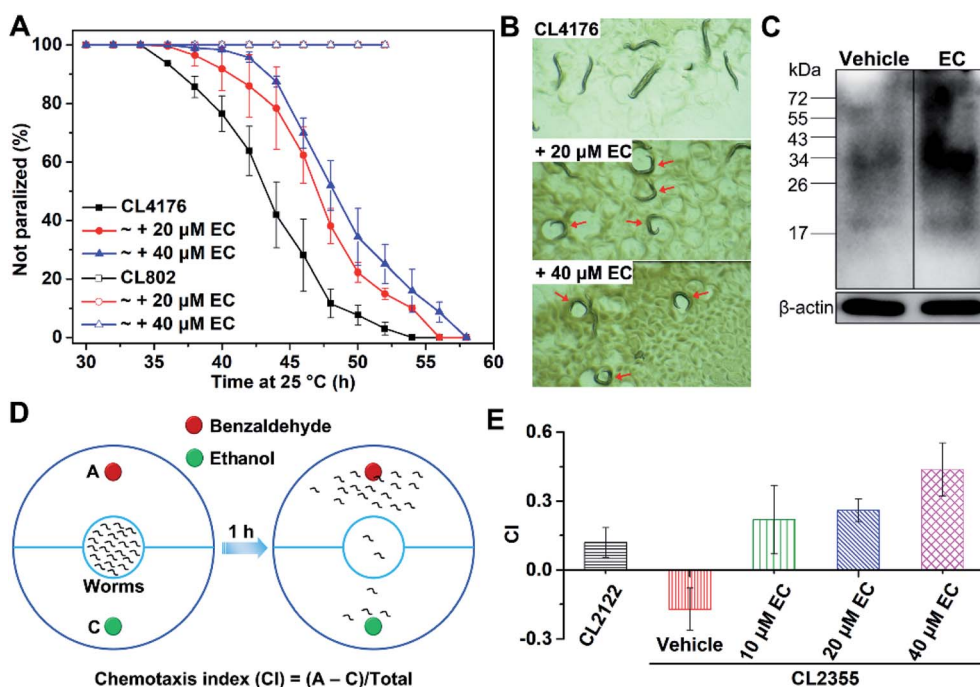


Fig. 8 (A) Measurements of the effect of different concentrations of the EC on the A β 42-induced paralysis in the transgenic *C. elegans* strain CL4176. The wild-type worm model CL802 that does not express A β was used as a control strain. (B) The motion modalities of *C. elegans* CL4176 upon treatment with DMSO (4% v/v) or different concentrations of the EC after living at 25 $^{\circ}$ C for 44 h. (C) Western blot analysis of A β extracted from *C. elegans* CL4176 in the absence and presence of the EC using 6E10 antibody. β -Actin was used as the internal reference to ensure equal protein loading. (D) Schematic diagram of the chemotaxis behavior assay. Culture plates were divided into two semicircles. Each semicircle was labeled either as an attractant (A) or control odorant (C). Worms were placed in the inner circle at the center of the plate and counted on each semicircle after 1 h. Individuals that did not cross out of the inner circle were not counted. A chemotaxis index (CI) was calculated using the equation $CI = (A - C) / \text{total}$ (total of scored worms) to indicate the chemotaxis behavior of worms. (E) Measurements of the effect of different concentrations of the EC on A β toxicity-induced defects in neuron-controlled chemotaxis behavior of transgenic *C. elegans* strain CL2355. The wild-type worm model CL2122 that does not express A β was used as a control strain. Error bars in (A) and (E) indicate mean \pm s.d. (n = 3 independent experiments).



sinusoidally move in the characteristic “C” curve just like the movements of wild-type *C. elegans* in both the presence and absence of the EC (Fig. S15, Movies S4 and S5†).⁶⁶ These results indicated that the EC is able to effectively rescue Aβ42-induced behavioral dysfunction of *C. elegans* *in vivo*.

We further investigated the level of Aβ aggregates in EC-treated and untreated worms by western blot assay using anti-Aβ antibody 6E10. Compared with untreated *C. elegans*, a higher amount of large aggregates (>34 kDa) was observed from the EC-treated worms, indicating that the EC is also able to promote Aβ aggregation *via* co-assembly *in vivo* (Fig. 8C). These findings would suggest that the EC-delayed paralysis rate of transgenic *C. elegans* is mainly attributed to the inhibitory effect of co-assembly between the EC and Aβ42 on Aβ-induced toxicity through promoting the conversion of toxic Aβ42 oligomers into non-toxic co-aggregates.

Besides the paralysis phenotype reflecting the toxicity of Aβ aggregates to muscle cells, a behavior phenotype of the transgenic strain (CL2355) that express Aβ42 in neuronal cells has also been commonly used to investigate the neuroprotective effect of the candidates against Aβ-induced defects in neuron-controlled chemotaxis behavior.^{67,68} The chemotaxis behavior can be evaluated by measuring the chemotaxis index (CI), that is, the fraction of worms that are able to arrive at the location of the attractants.⁶⁸ To investigate the effect of the EC on the chemotaxis behavior, we chose benzaldehyde as an attractant and ethanol as a control (Fig. 8D). In comparison with the control strain CL2122 that does not express Aβ, the untreated CL2355 showed a quite low CI value, due to the Aβ-induced defect in chemotaxis behavior (Fig. 8E). However, EC feeding of the transgenic worms significantly reversed the reduced CI in a concentration-dependent manner. These results suggest that the EC can also improve the chemotaxis behavior of transgenic *C. elegans* through attenuating Aβ42-induced toxicity *in vivo*.

Conclusions

We have described a rational design of small molecules that irreversibly suppress oligomers' toxicity through co-assembly with Aβ oligomers. A pincer-like scaffold combining a rigid center as a linker and two pendants with high binding affinity to the hydrophobic regions of Aβ has been proposed, which is expected to directly bind to the exposed hydrophobic patches of Aβ oligomers that are responsible for toxicity and aggregation. As a proof of concept, a europium complex, EC, in such a structural mode demonstrated high affinity for LMW oligomers and can rapidly promote Aβ aggregation, but convert the peptides into non-fibrillar aggregates with weak β-sheet conformation, according to the spectrometry, kinetics studies, electrophoresis, and morphology analysis. Completely distinct from the reported off-pathway products that are totally seeding-incompetent, EC-mediated non-fibrillar aggregates exhibited strong ability to accelerate Aβ aggregation as seeds. The data imply that the EC would co-assemble with Aβ oligomers to form EC-Aβ co-aggregates, which are structurally distinct from Aβ self-aggregates. We have also found that the co-aggregates are almost non-toxic and prone to degradation by glial cells *via* up-

regulating autophagy. Importantly, the EC can effectively rescue Aβ-mediated dysfunction in *C. elegans* models of AD *in vivo* owing to the bicomponent co-assembly. Since Eu³⁺ does not participate in the interactions between the EC and Aβ oligomers, such a pincer-like scaffold would also be available to other hydrophobic motif-binding groups for constructing Aβ oligomer-specific small molecules but not limited to lanthanide complexes when linked by proper rigid groups. Thus, this ‘small molecule-mediated co-assembly of oligomers’ strategy representing a powerful anti-oligomer therapy would provide a new direction for design of small molecule-based drug-like leads in the treatment of AD. Further structural studies to elucidate the conformation of co-aggregates, particularly EC-Aβ42 co-assemblies, and the screening of this class of compounds on human Aβ-expressing rodent models are currently in progress.

Ethical statement

All animal experiments were conducted in accordance with the National Institutes of Health “Guide for the Care and Use of Laboratory Animals” and were approved by the Institutional Animal Care and Use Committee of the MARC.

Conflicts of interest

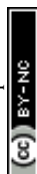
There are no conflicts to declare.

Acknowledgements

We appreciate the financial support from the National Natural Science Foundation of China (Grants: 21771105 and 21301090), the Natural Science Foundation of Jiangsu Province (Grant: BK20170103), the Natural Science Foundation of the Jiangsu Higher Education Institutions (Grant: 17KJB150018), and the Six Talent Peaks Project in Jiangsu Province (Grant: SWYY-043).

References

- 1 S. Makin, *Nature*, 2018, **559**, S4–S7.
- 2 I. Benilova, E. Karran and B. De Strooper, *Nat. Neurosci.*, 2012, **15**, 349–357.
- 3 C. Haass and D. J. Selkoe, *Nat. Rev. Mol. Cell Biol.*, 2007, **8**, 101–112.
- 4 B. Zott, M. M. Simon, W. Hong, F. Unger, H.-J. Chen-Engerer, M. P. Frosch, B. Sakmann, D. M. Walsh and A. Konnerth, *Science*, 2019, **365**, 559–565.
- 5 R. J. Perrin, A. M. Fagan and D. M. Holtzman, *Nature*, 2009, **461**, 916–922.
- 6 I. Klyubin, D. M. Walsh, C. A. Lemere, W. K. Cullen, G. M. Shankar, V. Betts, E. T. Spooner, L. Jiang, R. Anwyl and D. J. Selkoe, *Nat. Med.*, 2005, **11**, 556–561.
- 7 S. J. C. Lee, E. Nam, H. J. Lee, M. G. Savelieff and M. H. Lim, *Chem. Soc. Rev.*, 2017, **46**, 310–323.
- 8 S. Ayala, P. Genevaux, C. Hureau and P. Faller, *ACS Chem. Neurosci.*, 2019, **10**, 3366–3374.
- 9 S. Kumar, A. Henningknechtel, I. Chehade, M. Magzoub and A. D. Hamilton, *J. Am. Chem. Soc.*, 2017, **139**, 17098–17108.



- 10 Y. Jiang, X. Jiang, X. Shi, F. Yang, Y. Cao, X. Qin, Z. Hou, M. Xie, N. Liu, Q. Fang, F. Yin, W. Han and Z. Li, *iScience*, 2019, **17**, 87–100.
- 11 J. Bieschke, M. Herbst, T. Wiglenda, R. P. Friedrich, A. Boeddrich, F. Schiele, D. Kleckers, J. M. L. D. Amo, B. Gruning and Q. Wang, *Nat. Chem. Biol.*, 2012, **8**, 93–101.
- 12 R. Limbocker, S. Chia, F. S. Ruggeri, M. Perni, R. Cascella, G. T. Heller, G. Meisl, B. Mannini, J. Habchi and T. C. T. Michaels, *Nat. Commun.*, 2019, **10**, 225.
- 13 D. E. Ehrnhoefer, J. Bieschke, A. Boeddrich, M. Herbst, L. Masino, R. Lurz, S. Engemann, A. Pastore and E. E. Wanker, *Nat. Struct. Mol. Biol.*, 2008, **15**, 558–566.
- 14 J. Bieschke, J. Russ, R. P. Friedrich, D. E. Ehrnhoefer, H. J. Wobst, K. Neugebauer and E. E. Wanker, *Proc. Natl. Acad. Sci. U. S. A.*, 2010, **107**, 7710–7715.
- 15 A. R. A. Ladiwala, J. C. Lin, S. S. Bale, A. M. Marcelinocruz, M. Bhattacharya, J. S. Dordick and P. M. Tessier, *J. Biol. Chem.*, 2010, **285**, 24228–24237.
- 16 T. Arai, D. Sasaki, T. Araya, T. Sato, Y. Sohma and M. Kanai, *ChemBioChem*, 2014, **15**, 2577–2583.
- 17 A. R. A. Ladiwala, J. S. Dordick and P. M. Tessier, *J. Biol. Chem.*, 2011, **286**, 3209–3218.
- 18 D. M. Raymond and B. L. Nilsson, *Chem. Soc. Rev.*, 2018, **47**, 3659–3720.
- 19 S. Campioni, B. Mannini, M. Zampagni, A. Pensalfini, C. Parrini, E. Evangelisti, A. Relini, M. Stefani, C. M. Dobson and C. Cecchi, *Nat. Chem. Biol.*, 2010, **6**, 140–147.
- 20 A. R. A. Ladiwala, J. Litt, R. S. Kane, D. Aucoin, S. O. Smith, S. Ranjan, J. Davis, W. E. Van Nostrand and P. M. Tessier, *J. Biol. Chem.*, 2012, **287**, 24765–24773.
- 21 S. R. Chowdhury, M. Agarwal, N. Meher, B. Muthuraj and P. K. Iyer, *ACS Appl. Mater. Interfaces*, 2016, **8**, 13309–13319.
- 22 Y. Song, P. N. Cheng, L. Zhu, E. G. Moore and J. S. Moore, *J. Am. Chem. Soc.*, 2014, **136**, 5233–5236.
- 23 Z. Xu, S. Jia, W. Wang, Z. Yuan, B. J. Ravoo and D. Guo, *Nat. Chem.*, 2019, **11**, 86–93.
- 24 L. Zhu, Y. Song, P. N. Cheng and J. S. Moore, *J. Am. Chem. Soc.*, 2015, **137**, 8062–8068.
- 25 O. Szczepankiewicz, B. Linse, G. Meisl, E. Thulin, B. Frohm, C. S. Frigerio, M. T. Colvin, A. C. Jacavone, R. G. Griffin and T. P. J. Knowles, *J. Am. Chem. Soc.*, 2015, **137**, 14673–14685.
- 26 M. Suzuki, T. Takahashi, J. Sato, M. Mie, E. Kobatake and H. Mihara, *ChemBioChem*, 2010, **11**, 1525–1530.
- 27 E. A. Fradinger, B. H. Monien, B. Urbanc, A. Lomakin, M. Tan, H. Li, S. M. Spring, M. M. Condron, L. Cruz, C.-W. Xie, G. B. Benedek and G. Bitan, *Proc. Natl. Acad. Sci. U. S. A.*, 2008, **105**, 14175–14180.
- 28 C. L. Teoh, D. Su, S. Sahu, S. Yun, E. Drummond, F. Prelli, S. Lim, S. Cho, S. Ham and T. Wisniewski, *J. Am. Chem. Soc.*, 2015, **137**, 13503–13509.
- 29 G. Lv, A. Sun, P. Wei, N. Zhang, H. Lan and T. Yi, *Chem. Commun.*, 2016, **52**, 8865–8868.
- 30 Y. Li, D. Xu, A. Sun, S. Ho, C. Poon, H. Chan, O. T. W. Ng, K. K. L. Yung, H. Yan and H. Li, *Chem. Sci.*, 2017, **8**, 8279–8284.
- 31 P. J. Salveson, S. Haerianardakani, A. Thuy-Boun, S. Yoo, A. G. Kreutzer, B. Demeler and J. S. Nowick, *J. Am. Chem. Soc.*, 2018, **140**, 11745–11754.
- 32 B. O. Okesola and A. Mata, *Chem. Soc. Rev.*, 2018, **47**, 3721–3736.
- 33 M. Beeg, M. Stravalaci, M. Romeo, A. D. Carra, A. Cagnotto, A. Rossi, L. Diomedea, M. Salmona and M. Gobbi, *J. Biol. Chem.*, 2016, **291**, 6958–6966.
- 34 H. Fu, M. Cui, P. Tu, Z. Pan and B. Liu, *Chem. Commun.*, 2014, **50**, 11875–11878.
- 35 M. Kung, C. Hou, Z. Zhuang, B. Zhang, D. M. Skovronsky, J. Q. Trojanowski, V. M. Y. Lee and H. F. Kung, *Brain Res.*, 2002, **956**, 202–210.
- 36 B. Mannini, R. Cascella, M. Zampagni, M. Van Waardeverhagen, S. Meehan, C. Roodveldt, S. Campioni, M. Boninsegna, A. Penco and A. Relini, *Proc. Natl. Acad. Sci. U. S. A.*, 2012, **109**, 12479–12484.
- 37 M. G. Savelieff, S. Lee, Y. Liu and M. H. Lim, *ACS Chem. Biol.*, 2013, **8**, 856–865.
- 38 G. Meisl, J. B. Kirkegaard, P. Arosio, T. C. T. Michaels, M. Vendruscolo, C. M. Dobson, S. Linse and T. P. J. Knowles, *Nat. Protoc.*, 2016, **11**, 252–272.
- 39 X. Wang, H. Chang, J. Xie, B. Zhao, B. Liu, S. Xu, W. Pei, N. Ren, L. Huang and W. Huang, *Coord. Chem. Rev.*, 2014, **273**, 201–212.
- 40 K. Binnemans, *Coord. Chem. Rev.*, 2015, **295**, 1–45.
- 41 X. Wang, T. Yang, J. Luo, L. Yang and C. Yao, *Chem. Commun.*, 2015, **51**, 8185–8188.
- 42 E. A. Weitz, J. Y. Chang, A. H. Rosenfield and V. C. Pierre, *J. Am. Chem. Soc.*, 2012, **134**, 16099–16102.
- 43 A. G. Kreutzer, I. L. Hamza, R. K. Spencer and J. S. Nowick, *J. Am. Chem. Soc.*, 2016, **138**, 4634–4642.
- 44 G. V. De Ferrari, M. Canales, I. Shin, L. Weiner, I. Silman and N. C. Inestrosa, *Biochemistry*, 2001, **40**, 10447–10457.
- 45 M. Necula, R. Kaye, S. Milton and C. G. Glabe, *J. Biol. Chem.*, 2007, **282**, 10311–10324.
- 46 M. Necula, L. Breydo, S. Milton, R. Kaye, W. E. van der Veer, P. Tone and C. G. Glabe, *Biochemistry*, 2007, **46**, 8850–8860.
- 47 W. Kim and H. Hecht, *Proc. Natl. Acad. Sci. U. S. A.*, 2006, **103**, 15824–15829.
- 48 R. Kaye, E. Head, J. L. Thompson, T. M. McIntire, S. Milton, C. W. Cotman and C. G. Glabe, *Science*, 2003, **300**, 486–489.
- 49 N. Lorenzen, S. B. Nielsen, A. K. Buell, J. D. Kaspersen, P. Arosio, B. S. Vad, W. Paslawski, G. Christiansen, Z. Valnickovahansen and M. Andreassen, *J. Am. Chem. Soc.*, 2014, **136**, 3859–3868.
- 50 T. Yang, X. Wang, C. Zhang, X. Ma, K. Wang, Y. Wang, J. Luo, L. Yang, C. Yao and X. Wang, *Chem. Commun.*, 2016, **52**, 2245–2248.
- 51 S. I. A. Cohen, M. Vendruscolo, C. M. Dobson and T. P. J. Knowles, *J. Mol. Biol.*, 2012, **421**, 160–171.
- 52 J. Habchi, S. Chia, C. Galvagnion, T. C. T. Michaels, M. M. J. Bellaiche, F. S. Ruggeri, M. Sanguanini, I. Idini, J. R. Kumita and E. Sparr, *Nat. Chem.*, 2018, **10**, 673–683.
- 53 J. Habchi, P. Arosio, M. Perni, A. R. Costa, M. Yagiutsumi, P. D. Joshi, S. K. R. Chia, S. I. A. Cohen, M. B. Muller and S. Linse, *Sci. Adv.*, 2016, **2**, e1501244.



- 54 S. I. A. Cohen, S. Linse, L. M. Luheshi, E. Hellstrand, D. A. White, L. Rajah, D. E. Otzen, M. Vendruscolo, C. M. Dobson and T. P. J. Knowles, *Proc. Natl. Acad. Sci. U. S. A.*, 2013, **110**, 9758–9763.
- 55 G. Meisl, X. Yang, E. Hellstrand, B. Frohm, J. B. Kirkegaard, S. I. A. Cohen, C. M. Dobson, S. Linse and T. P. J. Knowles, *Proc. Natl. Acad. Sci. U. S. A.*, 2014, **111**, 9384–9389.
- 56 C. Dammers, M. Schwarten, A. K. Buell and D. Willbold, *Chem. Sci.*, 2017, **8**, 4996–5004.
- 57 R. A. Nixon, *Nat. Med.*, 2013, **19**, 983–997.
- 58 J. M. Tarasoffconway, R. O. Carare, R. S. Osorio, L. Glodzik, T. Butler, E. Fieremans, L. Axel, H. Rusinek, C. Nicholson and B. V. Zlokovic, *Nat. Rev. Neurol.*, 2015, **11**, 457–470.
- 59 Q. Luo, Y. Lin, P. Yang, Y. J. Wang, G. Qi, Z. Qiao, B. Li, K. Zhang, J. Zhang and L. Wang, *Nat. Commun.*, 2018, **9**, 1802.
- 60 Y. Zhao, J. Cai, Z. Liu, Y. Li, C. Zheng, Y. Zheng, Q. Chen, H. Chen, F. Ma and Y. An, *Nano Lett.*, 2019, **19**, 674–683.
- 61 K. Debnath, N. R. Jana and N. R. Jana, *ACS Biomater. Sci. Eng.*, 2019, **5**, 390–401.
- 62 D. J. Klionsky, F. C. Abdalla, H. Abeliovich, R. T. Abraham, A. Acevedoarozena, K. Adeli, L. Agholme, M. Agnello, P. Agostinis and J. A. Aguirreghiso, *Autophagy*, 2012, **8**, 445–544.
- 63 W. Fu, X. Li, X. Lu, L. Zhang, R. Li, N. Zhang, S. Liu, X. Yang, Y. Wang and Y. Zhao, *Cell Death Dis.*, 2017, **8**, e3086.
- 64 Y. Guan, Z. Du, N. Gao, Y. Cao, X. Wang, P. Scott, H. Song, J. Ren and X. Qu, *Sci. Adv.*, 2018, **4**, eaao6718.
- 65 C. D. Link, *Proc. Natl. Acad. Sci. U. S. A.*, 1995, **92**, 9368–9372.
- 66 C. D. Link, *Exp. Gerontol.*, 2006, **41**, 1007–1013.
- 67 E. F. Fang, Y. Hou, K. Palikaras, B. A. Adriaanse, J. S. Kerr, B. Yang, S. Lautrup, M. M. Hasanolive, D. Caponio and X. Dan, *Nat. Neurosci.*, 2019, **22**, 401–412.
- 68 R. Keowkase, N. Shoomarom, W. Bunargin, W. Sitthithaworn and N. Weerapreeyakul, *Biomed. Pharmacother.*, 2018, **107**, 656–664.

