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Total synthesis of (–)-penicimutanin a and related congeners†‡

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The first total synthesis of penicimutanin A (**1**) was achieved within 10 steps (LLS). Key innovations in this synthesis consist of (1) a highly efficient electro-oxidative dearomatization; (2) an unprecedented bisoxirane-directed intermolecular aldol reaction from the sterically hindered face of the ketone and (3) the diastereoselective one-step Meerwein–Eschenmoser–Claisen rearrangement enabling the construction of vicinal quaternary stereocenters. Related family members *e.g.* penicimutanolone (**3**) and penicimutatin (**5**) have also been synthesized alongside, elucidating their absolute configurations, hence the absolute configuration of **1**.

Introduction

Fungi have been exploited as rich sources to generate structurally appealing molecules featuring synthetically challenging motifs and consecutive stereocenters, which often display promising bioactivities towards life-threatening diseases.¹ In particular, for deep sea-derived fungal strains, chemical mutagenesis is commonly used to activate “muted genes” for the discovery of new natural products.² Penicimutanin A (**1**) and B (**2**) are two representative secondary metabolites reported by Cui and coworkers in 2014 (Fig. 1), isolated from a marine-derived fungal strain identified as *Penicillium purpurogenum* G59 through chemical mutagenesis.³ An interesting observation is that, in addition to **1**, penicimutanolone (**3**) and penicimutatin (**5**) together with vicinal quaternary center-bearing pyrroloindoline-containing^{4c–g} fructigenine A (**4**)^{4a,b} were also isolated alongside, indicating their biosynthetic interrelationship. More importantly, it implied that chemical mutagenesis may have activated a “linchpin gene” of the marine-derived fungus which endogenously joined penicimutanolone (**3**) and fructigenine A (**4**) together, in contrast to the terrestrial-derived fungal strain where only analogues of **3** were isolated.^{5,6} These secondary metabolites and their congeners were found to possess a range

of biological activities from anti-fungal as a histidine kinase inhibitor to anti-inflammatory as well as anti-tumor effects.^{3–5} However, the scarcity of these compounds from natural sources greatly hampered their further biological profiling. What's more, the absolute configuration of penicimutanin A (**1**) and penicimutanolone (**3**) remained undetermined.

It is clear that to address the aforementioned structural uncertainty issues, practical and reliable chemical synthesis of these natural products would be necessary. However, to the best of our knowledge, total synthesis of penicimutanin A and its congeners (*e.g.* **2**, **3** and **5**) has not been reported.

The challenge for their synthesis could be three-fold. First, these natural products are characterized by a fully oxygenated cyclohexane ring which contains an acid/base-sensitive *cis*-

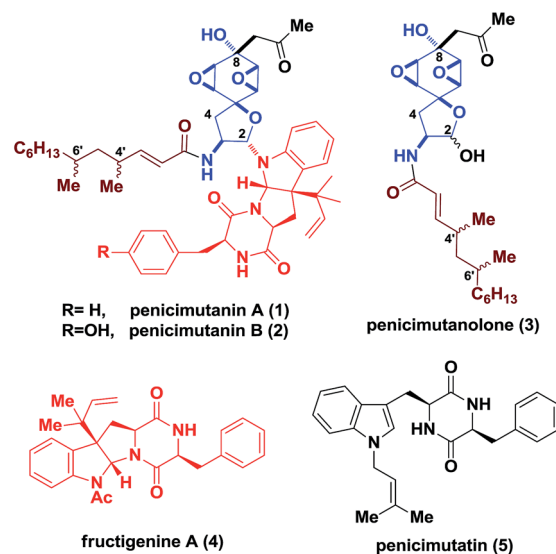


Fig. 1 Penicimutanin A and the related congeners.

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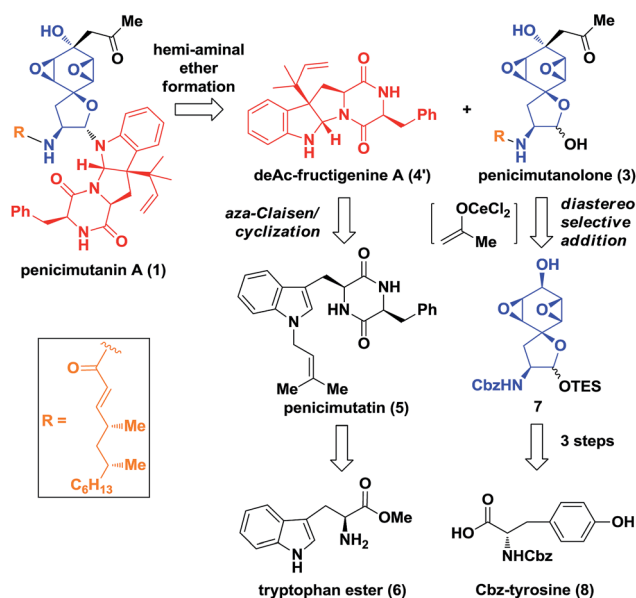
† This article is dedicated to Professor Zhen Yang on the occasion of his 60th birthday.

‡ Electronic supplementary information (ESI) available. CCDC 1843076 and 1950968. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c9sc05252f

§ Haiyong and Yan contributed equally to this work.

Results and discussion

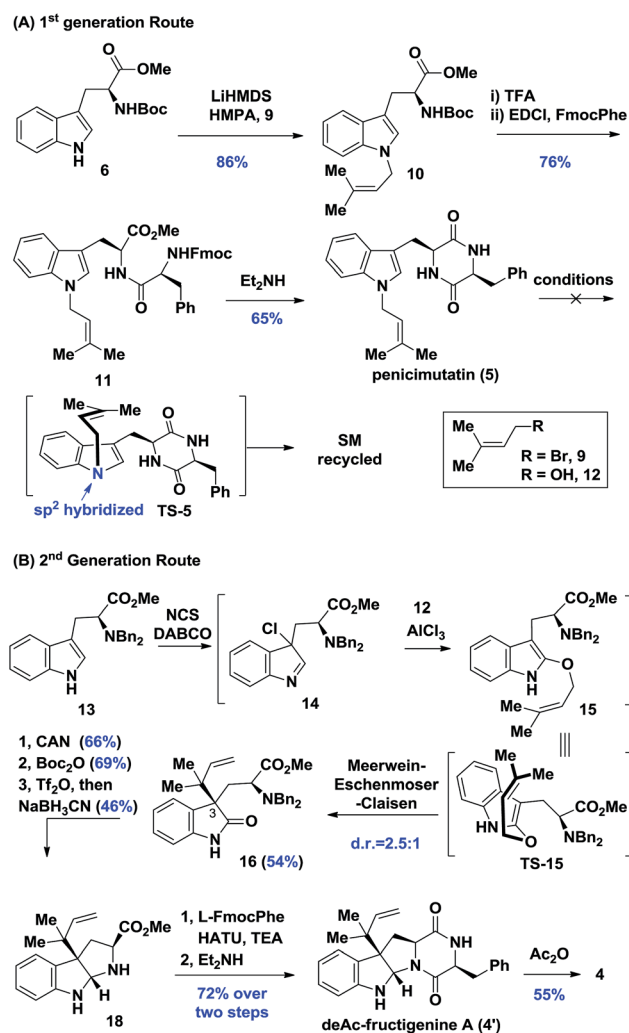
Biosynthetically, it was hypothesized that penicimutanin A (**1**) could be disconnected into deAc-fructigenin A (**4'**) and penicimutanolone (**3**) through the hemi-aminal ether formation ("linchpin" step). The south fragment **4'** could be accessed through a biomimetic aza-Claisen rearrangement followed by indoline formation from the natural congener **5**, which itself was an *N*-prenylated diketopiperazine compound and could be reached from commercially available (*S*)-tryptophan ester **6**. On the other hand, the northern fragment **3** could be easily accessed through a side chain attaching after a diastereoselective aldol reaction from bisoxirane **7**, which might be obtained through an oxidative dearomatization from the commercially available (*S*)-Cbz-tyrosine **8**. This design features a biomimetic, convergent strategy, in the hope of obtaining four natural products from common starting materials (Scheme 1).



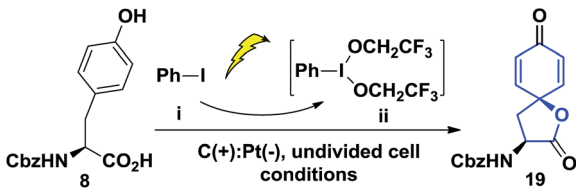
Scheme 1 Retro-synthesis.

Our synthesis of deacetyl fructigenine A (**4'**) started with selective prenylation of commercially available tryptophan ester **6** in 86% yield with isoprenyl bromide **9**. The desired prenylation product **10** was condensed with Fmoc-L-tyrosine after Boc removal, affording compound **11** in 76% overall yield. Upon Fmoc deprotection under standard conditions, the diketopiperazine-containing penicimutatin **5** was achieved with 65% yield (Scheme 2A). The intended biomimetic aza-Claisen was proved unviable after extensive condition screening (see the ESI,[†] for details). Our rationale is that the intrinsic sp² hybridized nitrogen atom of the indole framework inhibited the formation of the chair transition state **TS-5**, in which the N-atom was twisted into a transient sp³ hybridized form (**TS-5** in Scheme 2A).

Given the above results, a one-pot diastereoselective Meerwein–Eschenmoser–Claisen (MEC) rearrangement⁸ was developed to construct C3 reverse prenylated⁹ oxindoles from non-protected C3-ester indole and isoprenyl alcohol through an AlCl₃ catalytic protocol. The L-tryptophan derivative **13** was subjected to our AlCl₃-mediated conditions. To our delight, the oxindole **16** was achieved in a satisfactory 54% yield with



Scheme 2 Synthesis of fructigenine A (**4**) and penicimutatin (**5**).

Table 1 Selected electrooxidation optimization^a


Entry ^a	Conditions	Time	Yield ^b (%)
1	LiClO ₄ (1 eq.), I = 2.1 mA, PhI (1 eq.)	8 h	38
2	LiClO ₄ (10 eq.), I = 6.1 mA, PhI (4 eq.)	4 h	64
3	LiClO ₄ (10 eq.), I = 6.1 mA, PhI (4 eq.)	6 h	65
4	LiClO ₄ (13 eq.), I = 10 mA, PhI (5 eq.)	6 h	65
5	LiClO ₄ (15.6 eq.), I = 10 mA, PhI (6 eq.)	3 h	65
6 ^c	LiClO ₄ (13 eq.), I = 10 mA, PhI (5 eq.)	4 + 6 h	82
7 ^d	LiClO ₄ (15.6 eq.), I = 10 mA, PhI (6 eq.)	5 + 5 h	85

^a All reactions were run with LiClO₄ as the electrolyte, PhI as the additive and a C(+)/Pt(−) anode in an undivided cell on a 0.31 mmol scale using 2,2,2-trifluoroethanol (TFE) as the solvent at a concentration of 0.04 M at ambient temperature for indicated hours with a constant electrical current unless otherwise noted. ^b Isolated yields. ^c The electrical current was stopped after 4 h but stirring was continued for 6 h. ^d The electrical current was stopped after 5 h but stirring was continued for 5 h.

synthetically acceptable stereocontrol (d.r. = 2.5 : 1). Compound **16** possessed the key reversely prenylated vicinal quaternary oxindole motif, which was cyclized through reductive amination conditions after Bn-deprotection (66% yield) and Boc protection (69% yield) to achieve hexahydropyrroloindoline **18** in 46% yield. Compound **18** was coupled with L-Fmoc-phenylalanine and cyclized upon Fmoc removal, achieving the deAc-fructigenine **4'** in 72% overall

yield. The known natural product fructigenine **4** (reported^{3a} as $[\alpha]_D^{20} -129.3^\circ$ (c 0.30, MeOH)) was also successfully obtained through acylation in 55% yield and spectroscopic data of **4** ($[\alpha]_D^{20} -128.0^\circ$ (c 0.35, MeOH)) were in good agreement with those previously reported (see the ESI† for other spectroscopic data).

Our synthesis of the penicimutanolone (**3**) started with oxidative dearomatization of the commercially available (*S*)-Cbz-tyrosine (**8**).¹⁰ The recent development and innate sustainability of electrochemistry¹¹ prompted us to develop the electro-oxidative dearomatization of **8** (Table 1) in a synthetically useful fashion. The idea is to use an *in situ* formed milder hypervalent iodine oxidant **ii** from PhI (**i**) (Table 1).¹² We first tried the electro-oxidation with 1 equivalent of LiClO₄ at a constant electrical current of 2.1 mA in TFE. Gratifyingly, the desired spirocycle **19** was isolated in 38% yield. We further investigated the electrical current (entries 2 and 3, Table 1), electrolyte (entries 2–5) and loading of **i** (entry 2–5) and concluded that the reaction can be optimized to 65% isolated yield when using optimal amounts of the aforementioned factors. Serendipitously, an electrical power cut happened during another trial, which left the reaction standing still overnight after only several hours of electrical oxidation. However the reaction yield was improved to 82% (entry 6, Table 1). A careful adjustment of other co-factors and stirring time enabled us to obtain the oxidative dearomatization reaction in a satisfactory 85% isolated yield.

With enough amounts of **19** in hand, we set off to install the key bisoxirane moiety (Scheme 3). Due to **19**'s instability, a selective NaBH₄ reduction followed by TES protection were carried out to release the strain of spirocyclic lactone, yielding dienone **20** in 45% yield over two steps. A diastereoselective Luche-reduction was performed, providing β-OH alcohol (d.r. = 3 : 1), which allowed the *m*CPBA mediated

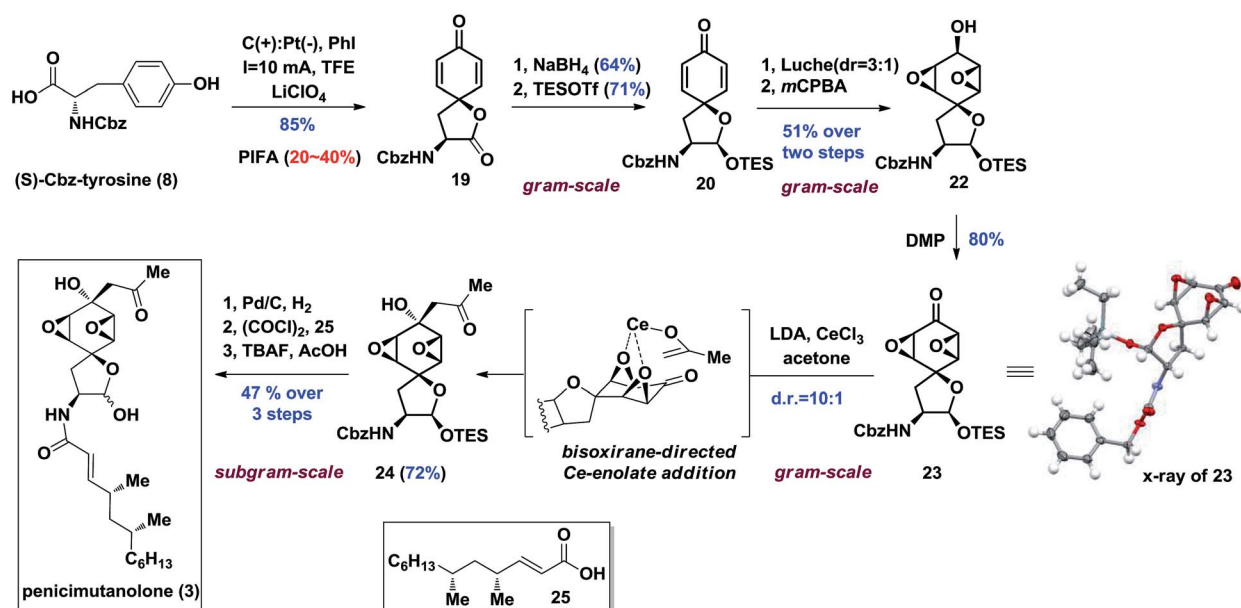
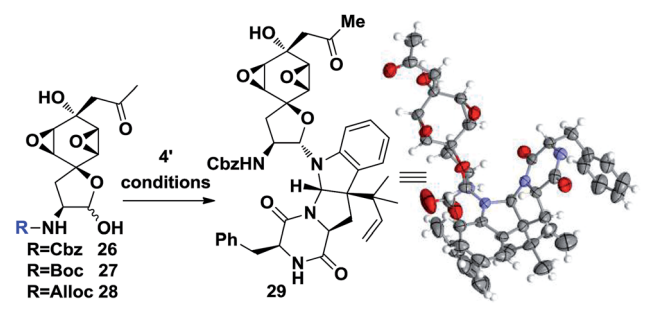
Scheme 3 Synthesis of penicimutanolone **3**.

Table 2 Selected electrooxidation optimization^a


Entry ^a	Substrate	Conditions	Yield ^b (%)
1	26	THF, rt, 4 h	0
2	26	TMSI, 0 °C, TEA, 4' DCM	0
3	26	TMSCl, KBr, 0 °C, TEA, 4' DCM	0
4	27	TMSBr, 0 °C, TEA, 4' DCM	0
5	28	TMSBr, 0 °C, TEA, 4' DCM	0
6	26	TMSCl, NaI, -20 °C, DMAP, 4' DCM	8
7 ^c	26	PTSA, MS, 4' MeCN/AcOH = 10/1	35 (55)

^a All reactions were run with 1 equivalent of a Lewis acid and 2 equivalents of a base on a 0.1 mmol scale with DCM (1 mL) at 0 °C and stirring overnight unless otherwise noted. ^b Isolated yields.

^c PTSA was used in 1 equivalent and 4 Å MS was added, MeCN/AcOH = 10/1 mixture (1 mL) was used instead of DCM, and the reaction was stirred at 60 °C for 72 hours.

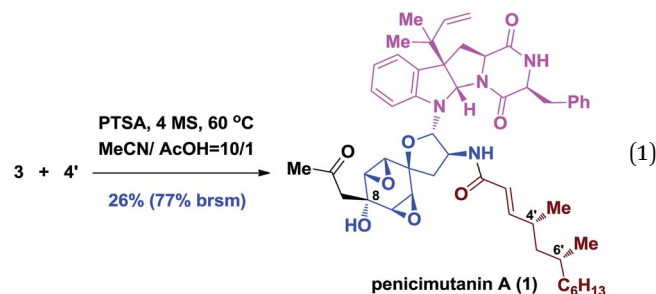
epoxidation to proceed smoothly affording **22** as a single diastereoisomer in 51% overall yield. The key reaction precursor **23** was obtained in 80% yield upon DMP oxidation. Crystallographic analysis showed unambiguous structural evidence of compound **23**.¹³

It was hypothesized that introduction of an acetyl group at the C8 position could be delivered from the more sterically hindered β -face of **23**, thus affording the desired stereoisomer by utilizing the bisoxirane moiety as a directing group (Scheme 3). With this assumption, the proposed bisoxirane-directed 1,2-addition was attempted by *in situ* generation of the oxophilic cerium-enolate instead of its traditional lithium counterpart. Gratifyingly, the transformation was realized in a 72% yield, with a resultant diastereoselective (d.r. = 10 : 1) formation of the desired α -OH isomer **24**. The efficacy of this challenging diastereoselective 1,2-addition step was attributed to a better chelation effect of the cerium-enolate as well as its reduced basicity. With the core skeleton in hand, side chain **25** (see the ESI[†] for preparation and characterization of **25**) was coupled with compound **24** over a three-step sequence, affording penicimutanolone **3** in an overall 47% yield. The spectroscopic data of synthetic C4',C6'-(*R,R*)-penicimutanolone ($[\alpha]_D^{25} -1.0^\circ$ (c 0.88, MeOH)) correspond to penicimutanolone (**3**) ($[\alpha]_D^{25} -3.5^\circ$ (c 1.0, MeOH)) very well thus confirming the absolute configuration of C4' and C6' of the natural product.

At this point, the linchpin¹⁴ step conditions were investigated. Different substrates and conditions were screened (Table 2). Simply mixing the desilylation fragment **26** and south fragment **4'** didn't provide the product **29** (entry 1, Table

2). A Lewis acid promoted *in situ* formation of an electrophilic oxonium intermediate didn't yield any product either, while slight decomposition of **26** was observed (entries 2 and 3). It was hypothesized that the benzyl group of Cbz might block one side of the oxonium thus inhibiting its reactivity. Different substrates **27** and **28** were prepared and tested, but no fruitful results were obtained (entries 3 and 4). To our delight, 8% of desired **29** could be isolated when TMSCl and DMAP were used and the efficiency was further improved to 35% (55% brsm) yield when, unexpectedly, acidic conditions were employed (entry 6). In addition, an X-ray crystallographic analysis of **28** undisputedly verified all the stereochemistry as well as absolute configurations corresponding to natural product **1** (Table 2).

Although the coupling efficiency is still low, we nevertheless tried it on the real substrates of penicimutanolone **3** and deAc-fructigenine **4'**. As shown in eqn (1), natural product penicimutanin A (**1**) was successfully achieved under modified conditions in 26% (77% brsm) yield demonstrating our biomimetic hypothesis. The spectroscopic data of synthetic C4',C6'-(*R,R*)-penicimutanin A ($[\alpha]_D^{25} -91.1^\circ$ (c 0.25, MeOH)) correspond to the natural sample of penicimutanin A ($[\alpha]_D^{25} -99.7^\circ$ (c 0.5, MeOH)) very well thus confirming its absolute configuration (see Table S1 in the ESI[†] for a full comparison of NMR data).



Conclusion

In summary, we have completed the first total synthesis and determined the absolute configurations of penicimutanin A (**1**) and its related congeners penicimutanolone (**3**), fructigenine A (**4**) and penicimutatin (**5**) in a highly convergent and biomimetic fashion from the commercially available (*S*)-tryptophan and (*S*)-tyrosine derivatives for the first time. The highly concise total synthesis of **3** (10 step LLS) was enabled by an improved and highly efficient electro-oxidative dearomatization as well as an unprecedented diastereoselective 1,2-addition directed by the bisoxirane moiety. A modified one-pot diastereoselective MEC rearrangement was developed to reach fructigenine A (**4**) in only 6 steps. Further investigation of this methodology in natural product synthesis and biological profiling of penicimutans is currently on-going in our group.

Conflicts of interest

There are no conflicts to declare.



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