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Efficient polymerase chain reaction assisted by metal–organic frameworks†

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As a powerful tool for obtaining sufficient DNA from rare DNA resources, polymerase chain reaction (PCR) has been widely used in various fields, and the optimization of PCR is still in progress due to the dissatisfactory specificity, sensitivity and efficiency. Although many nanomaterials have been proven to be capable of optimizing PCR, their underlying mechanisms are still unclear. So far, the scientifically compelling and functionally evolving metal–organic framework (MOF) materials with high specific surface area, tunable pore sizes, alterable surface charges and favourable thermal conductivity have not been used for PCR optimization. In this study, UiO-66 and ZIF-8 were used to optimize error-prone two round PCR. The results demonstrated that UiO-66 and ZIF-8 not only enhanced the sensitivity and efficiency of the first round PCR, but also increased the specificity and efficiency of the second round PCR. Moreover, they could widen the annealing temperature range of the second round PCR. The interaction of DNA and Taq polymerase with MOFs may be the main reason. This work provided a candidate enhancer for PCR, deepened our understanding on the enhancement mechanisms of nano-PCR, and explored a new application field for MOFs.

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Introduction

As the gold standard of *in vitro* nucleic acid amplification, polymerase chain reaction (PCR) has been extensively used in many fields related to medical diagnostics and molecular biology.¹ However, due to the lack of correction mechanisms similar to *in vivo* amplification, nonspecific amplification and primer mismatch might occur occasionally in the PCR, resulting in unsatisfactory specificity, sensitivity and efficiency. Until now, many sophisticated methods have been developed to make up for these drawbacks, such as the development of PCR machines with rapid heating–cooling responses,² adopting unique PCR strategies,³ optimization of reaction conditions and adding additives to the PCR mixture.⁴ In the last two decades, nanomaterials as effective enhancers of PCR have attracted more attention (termed nano-PCR). So far, the reported nanomaterials contain gold nanoparticles, graphene oxide, quantum dots, ZnO, TiO₂ and nanocomposites.⁵ However, most of them

could only optimize one or two characteristics of PCR including specificity, sensitivity and efficiency, and few materials could enhance these three characteristics together. Besides, the limited optimization effect, complex preparation methods and high price would limit their widespread application. Therefore, looking for new materials with comprehensive optimization effects and simple preparation methods with low prices will be highly desirable.

Metal–organic frameworks (MOFs) are special organic–inorganic hybrid porous solids with extraordinarily high surface areas, tunable pore sizes, adjustable internal surface properties and extraordinary degree of variable structures. These features endow them with potential applications in gas or liquid adsorption/storage, drug delivery, polymerization, catalysis and biosensors.⁶ As far as we know, the combination of MOFs and DNA has been frequently studied.⁷ However, MOFs have not yet been used in the improvement of PCR.

In this study, two kinds of MOFs with exceptional chemical robustness and thermal stability, ZIF-8 and UiO-66, were used to optimize PCR. They were synthesized according to the relative references,^{8,9} and their characterization results are shown in Fig. S1, S2 and Table S1† respectively. The stock solutions (10 g L^{−1}) were stable aqueous solutions formed by sonication and autoclaving. The two-round error-prone PCR system with λ-DNA (48 502 bp) as template was employed as the model, and a single 283-bp band will be amplified after 35 cycles of the first round PCR, while many nonspecific bands will be accumulated after the second round 35-cycle PCR (Fig. S3†). The effect of MOFs on

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PCR is directly proportional to the densitometric value (DV) of PCR products on agarose gel. Specifically, the sensitivity of PCR could be reflected by the minimum template concentration that produces visual target bands, and the efficiency and specificity of PCR can be reflected by the DV of target bands ($DV_{\text{target band}}$) and the proportion of target bands in the total bands ($DV_{\text{target band}}/DV_{\text{total band}}$), respectively. The outline of this study is illustrated in Table 1.

Results and discussion

In most cases, one round conventional PCR with a suitable copy template was sufficient to obtain the desired amount of DNA, while fewer copy templates often result in random amplification, false negative or failure to obtain enough products. Therefore, improving the sensitivity of the first round PCR is a critical issue. Our study found that UiO-66 and ZIF-8 could improve the sensitivity and efficiency of the first round PCR in the absence of MOFs with good stability (Fig. 1A and S4†). And no band could be detected when the concentration of template was less than $0.5 \mu\text{g L}^{-1}$ (Fig. 1A, left lane 1–3). By contrast, in the presence of UiO-66 (Fig. 1A, middle lane 1–7) or ZIF-8 (Fig. 1A, right lane 1–7), obvious target bands can still be detected even at a template concentration as low as $0.125 \mu\text{g L}^{-1}$ (Fig. 1A). Importantly, UiO-66 showed much better PCR efficiency than ZIF-8 at special and lower template concentrations from 0.125 to $0.5 \mu\text{g L}^{-1}$ (Fig. 1B). Moreover, the normalized analysis of $DV_{\text{target band}}$ showed that UiO-66 and ZIF-8 could increase the PCR efficiency about 39- and 23-fold compared to conventional PCR, respectively (Fig. 1C). In all, this experiment suggested that UiO-66 and ZIF-8 could improve the sensitivity and efficiency of the first round PCR.

The effect of UiO-66 and ZIF-8 on the second round PCR is shown in Fig. 2, in which the 100-fold diluted product of the first round PCR was used as template. In the absence of UiO-66 or ZIF-8, nonspecific bands accumulated as manifested by smear bands (Fig. 2A lane 1, Fig. 2B lane 1 and Fig. S2† lane 1), while gradually diminishing smear bands were observed with the increase of UiO-66 or ZIF-8 concentration from 1 to 20 mg L^{-1} (Fig. 2A lane 2–4 and Fig. 2B lane 2–5). And a single predominant target band was observed at a concentration of 20 mg L^{-1} (Fig. 2A lane 4 and Fig. 2B lane 5). Specifically, the metal ions, specific ligand and excess addition of UiO-66 or ZIF-8 can significantly inhibit PCR as manifested by the attenuation or disappearance of target bands (Fig. S5,† Fig. 2A lane 5–6 and

Fig. 1 UiO-66 and ZIF-8 improve the sensitivity and efficiency of the first round PCR. (A) Gel electrophoresis of PCR products: left lane 1–7, without MOFs; middle lane 1–7, with UiO-66; right lane 1–7, with ZIF-8. The final concentration of λ -DNA in lane 1–7 was 0.125, 0.25, 0.5, 2, 4 and $20 \mu\text{g L}^{-1}$ respectively. M refers to the DL2000 marker; the white frame indicates the location of the target band in all lanes. (B) The densitometric value (DV) of the target band calculated using Image J software. (C) Relative ratio of $DV_{\text{target band}}$ to conventional PCR.

Fig. 2B lane 6–7). Furthermore, the analysis of DV showed that the specificity and efficiency of PCR increased first and then decreased with the increase of concentrations of UiO-66 or ZIF-8 (Fig. 2C and D). This phenomenon happened in the first round PCR (Fig. S6†). When the optimal concentration of 20 mg L^{-1} MOFs was used, UiO-66 and ZIF-8 could increase the specificity of PCR by 29 times and 8 times and increase the efficiency of PCR by 24 times and 2 times, respectively (Fig. 2C and D). It is worth noting that the performance of UiO-66 is much better than that of ZIF-8. In summary, these experiments confirmed that both UiO-66 and ZIF-8 can improve the specificity and efficiency of the second round PCR at appropriate concentrations.

As we know, the annealing temperature is important for efficiency and specificity of PCR. Generally speaking, a low annealing temperature facilitates the efficiency of PCR and a high annealing temperature reduces the specificity of PCR.¹⁰ Therefore, much more sophisticated optimization of annealing temperature was further investigated. Like several other nanomaterials,¹¹ UiO-66 and ZIF-8 exhibited reliable PCR results at

Table 1 The outline of this study

	First round PCR (283bp)		Second round PCR (283bp)	
	PCR	MOF-PCR (UiO-66 or ZIF-8)	PCR	MOF-PCR (UiO-66 or ZIF-8)
Sensitivity	$0.5 \mu\text{g L}^{-1}$	$0.125 \mu\text{g L}^{-1}$	Inapplicable	Inapplicable
Specificity	Inapplicable	Inapplicable	1	29/8
Efficiency	1	39/23	1	24/2
Annealing temperature	Unverified	Unverified	Narrow	Wide

Fig. 2 UiO-66 and ZIF-8 improve the specificity and efficiency of the second round PCR. (A and B) show gel electrophoresis of the PCR products; the final concentration of UiO-66 in lane 1–6 was 0, 1, 10, 20, 100 and 1000 mg L⁻¹ and the final concentration of ZIF-8 in lane 1–7 was 0, 0.1, 1, 10, 20, 100 and 1000 mg L⁻¹, respectively. The yellow frame indicates the area used to calculate the densitometric value of total bands in each lane, and the white frame indicates the location of the target band in all lanes. M refers to the DL2000 marker. Lane 7 for (A) and lane 8 for (B) are negative controls with no template. (C) The relative ratio of the target band to conventional PCR. (D) The relative ratio of (DV_{target band}/DV_{total bands}) to conventional PCR.

low annealing temperatures (Fig. 3). When non-optimal annealing temperatures (25–40 °C) were employed in PCR, conventional PCR produced a predominant band corresponding to 200-bp with numerous non-specific bands and few target bands. However, in the presence of UiO-66/ZIF-8 (20 mg L⁻¹), most non-specific bands disappeared, and target bands were amplified as the main band with an obviously higher intensity than the conventional PCR. More strikingly, the predominant band corresponding to 200-bp produced in conventional PCR

totally disappeared in the presence of ZIF-8 (Fig. 3B). The relative DV analysis showed that both UiO-66 and ZIF-8 could improve the specificity of PCR at all tested non-optimal annealing temperatures, and this effect became more obvious with the increase of the annealing temperatures (Fig. 3C). Furthermore, ZIF-8 had a better effect than UiO-66 (Fig. 3C). In addition, at almost all the test temperatures, UiO-66 and ZIF-8 showed more improved PCR efficiency than conventional PCR. With the increase of annealing temperatures, the improvement extent of the PCR efficiency also slightly increased, which was especially evident in ZIF-8 (Fig. 3D). Moreover, the four different kinds of PCR experiments showed similar results (Fig. S7†). In short, UiO-66 and ZIF-8, especially ZIF-8, could widen the ideal range of annealing temperature from 40–65 °C to below 40 °C. It not only could obviate the time-consuming screening process, but also could improve the efficiency at much lower annealing temperatures.

Although some studies related to nano-PCR attributed the widened annealing temperature to the superior thermal conductivity of nanomaterials,¹² both UiO-66 and ZIF-8 are thermally inert MOFs, and their thermal conductivity is about 0.11 and 0.165 W m⁻¹ K⁻¹ respectively,¹³ which are far lower than the thermal conductivity of reported nano-PCR materials.¹⁴ Nevertheless, PCR was performed with two different cycles, in which denaturation, annealing and extension times were reduced. It showed that smear bands amplified in the second round PCR (Fig. 4A, lane 1 and lane 4) were only slightly reduced in the presence of UiO-66 (Fig. 4A, lane 2 and lane 5), but they were obviously reduced in the presence of ZIF-8 (Fig. 4A, lane 3 and 6). Besides, the intensity of target bands obviously increased in the presence of MOFs. These results suggested that the reduction of reaction time weakened rather than blocking the effect of UiO-66 and ZIF-8 on the specificity and efficiency of PCR. It means that the thermal conductivity of MOFs may contribute to the improvement of PCR. The better effect of ZIF-8 on PCR than UiO-66 might be explained by the higher thermal conductivity of ZIF-8.

The most well-known application of MOFs is their adsorption/storage ability of ions, nucleic acids, proteins and other substances.¹⁵ And the conjugated π -electron system of MOFs allows their binding of ssDNA through π - π stacking.¹⁶ Using this property, UiO-66 and ZIF-8 have been used in biosensors to test some special single stranded DNA (ssDNA).^{8,17} Therefore, it is reasonable to speculate that UiO-66 and ZIF-8 selectively attract or adsorb ssDNA rather than double stranded DNA (dsDNA) to improve the efficiency of PCR. To confirm this speculation, we compared the relative fluorescence intensity of different solutions containing UiO-66 (or ZIF-8) and a fluorescence labelled nucleic acid probe (FP) in the absence or presence of a complementary probe (CP). For the sequences of the FP and CP, refer to Table S2.† As shown in Fig. 4B, the solutions containing the FP and CP have obviously higher fluorescence intensity than the FP alone, which means that the attraction or adsorption to ssDNA contribute to the optimization effect of UiO-66 and ZIF-8.

Another advantage of MOFs is their potential as an enzyme immobilization platform.¹⁸ It could be inferred that DNA

Fig. 3 UiO-66 and ZIF-8 changed the effect of annealing temperature on the second round PCR. M, DL2000 marker; lane 1, 3, 5 and 7 was conventional PCR at annealing temperatures of 25, 30, 35 and 40 °C, while lanes 2, 4, 6, and 8 were UiO-66 (A) or ZIF-8 (B) assisted PCR at the corresponding annealing temperature. The yellow frame indicates the area used to count the total bands in each lane, and the white frame indicates the location of the target band in all lanes. (C) Relative ratio of DV_{target band} to conventional PCR. (D) Relative ratio of (DV_{target band}/DV_{total bands}) to conventional PCR.

Fig. 4 Mechanism speculation of improved PCR by UiO-66 and ZIF-8. M, DL2000 marker. (A) Gel electrophoresis of PCR products produced by two different cycle parameters (cycle 1 and cycle 2) in the absence or presence of UiO-66 or ZIF-8. (B) UiO-66 and ZIF-8 selectively bind with ssDNA rather than dsDNA. After the addition of complementary ssDNA (CP, 150 nM), fluorescence recovery was observed in PA (50 nM) @ UiO-66 or PA (50 nM) @ ZIF-8 dissolved in 1× PCR buffer (pH 7.2). (C) PCR was carried out with different Taq polymerases and 0.1 μg L⁻¹ λ-DNA. Left lanes 1–4, conventional PCR; middle lanes 1–4, UiO-66 assisted PCR; right lane 1–7, ZIF-8 assisted PCR. The final concentrations of Taq polymerase from lane 1 to 4 are 0.25 U, 0.5 U, 1 U and 2 U per 10 μL respectively. In (A and C), the white frame indicates the location of the target band in all lanes. (D) The possible interaction among templates, Taq polymerase and UiO-66 (or ZIF-8) during PCR.

polymerase is immobilized by UiO-66 or ZIF-8 materials, and the efficiency of DNA polymerase is improved greatly, leading to the improved PCR effect. To test this possibility, we performed the first round PCR with a lower template concentration (0.2 μg L⁻¹) that could cause amplification failure in conventional PCR. It could not be reversed by increasing the concentration of Taq polymerase (lane 2–4) (Fig. 4C lane 1). However, the addition of UiO-66 (20 mg L⁻¹) reversed the amplification failure, and with the increase of the Taq polymerase concentration, the intensity of target bands and some non-specific bands increased (Fig. 4C lane 5–8), while ZIF-8 (20 mg L⁻¹) did not show the same effect as UiO-66, and only some non-specific bands were observed with the highest concentration of Taq polymerase (Fig. 4C lane 9–12). This difference may be caused by their different pore sizes and total pore volumes, just as shown in the inset of Fig. S1, S2 and Table S1.† This suggests that ZIF-8 contains microporous particles, while UiO-66 contains tridimensional mesoporous particles. Besides, UiO-66 has a larger total pore volume than ZIF-8. Maybe these characters make UiO-66 much more suitable for enzyme immobilization than ZIF-8.

Another possible explanation may lie in their different zeta potentials; ZIF-8 (about 15 mV) is more positively charged than UiO-66 (about -5 mV) (Fig. S8†), which means that they prefer to bind with negatively charged nucleic acids. The SEM and TEM morphology and elemental analysis of MOFs before and after the addition of a high concentration of Taq polymerase or template also supports this explanation (Fig. S9–S16†). As

revealed in the SEM results, UiO-66 showed significant morphological change after adding a high concentration of the Taq enzyme, while ZIF-8 had indistinct morphological changes after adding a high concentration of λ-DNA (Fig. S9†). Moreover, the TEM results indicated that the adsorption capacity of UiO-66 was stronger than that of ZIF-8 (Fig. S10–S15†). The results showed that both surface charges and pore sizes have certain roles in the MOFs' enhancement effect on PCR, and pores may play a more important role for UiO-66 rather than ZIF-8. And the relationship among templates, Taq polymerase and UiO-66 (or ZIF-8) during the extension process of PCR is illustrated in Fig. 4D.

Conclusions

In summary, our work preliminarily demonstrated that for the error-prone two round PCR system, UiO-66 and ZIF-8 could remarkably improve the sensitivity and efficiency of the first round PCR and they also could remarkably enhance the specificity and efficiency of the second round PCR, even at a very low annealing temperature. The possible mechanism might be attributable to their selective adsorption to ssDNA and the ability of immobilizing Taq polymerase. This work provides a new insight into the understanding of the nano-PCR effects and their enhancement mechanism for PCR. It also opened up a new field for the application of MOFs. MOFs could provide a variety of materials with remarkable geometrical shapes and

chemical properties of the internal surface of nanoporous materials. Further investigation into the effect of MOFs on PCR should be focused on the influence of shape, particle size, hydrophilicity, colloidal stability, surface charge, pore sizes and other factors on PCR.

Conflicts of interest

The authors declare no conflict of interest.

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Notes and references

- (a) Y. L. Jung, C. Jung, H. Parab, D. Y. Cho and H. G. Park, *ChemBioChem*, 2011, **12**, 1387–1390; (b) R. K. Saiki, D. H. Gelfand, S. Stoffel, S. J. Scharf, R. Higuchi, G. T. Horn, K. B. Mullis and H. A. Erlich, *Science*, 1988, **239**, 487–491; (c) R. K. Saiki, S. Scharf, F. Faloona, K. B. Mullis, G. T. Horn, H. A. Erlich and N. Arnheim, *Science*, 1985, **230**, 1350–1354; (d) Y. Wang, C. Liu, X. Zhang, W. Yang, F. Wu, G. Zou, X. Wen and X. Zhou, *Chem. Sci.*, 2018, **9**, 3723–3728.
- (a) J. Berg, V. Nagl, G. Mühlbauer and H. Stekel, *J. Clin. Virol.*, 2001, **20**, 71–75; (b) M. S. Ibrahim, R. S. Loftis, P. B. Jahrling, E. A. Henchal, V. W. Weedn, M. A. Northrup and P. Belgrader, *Anal. Chem.*, 1998, **70**, 2013–2017.
- (a) J. H. Van, M. M. Ehlers, W. Z. Van and W. O. Grabow, *Water Res.*, 2003, **37**, 3704–3708; (b) N. Paul, J. Shum and T. Le, *Hot start PCR*, Humana Press, Totowa, NJ, 2010, vol. 630, pp. 301–318; (c) R. H. Don, P. T. Cox, B. J. Wainwright, K. Baker and J. S. Mattick, *Nucleic Acids Res.*, 1991, **19**, 4008; (d) A. Roloff and O. Seitz, *Chem. Sci.*, 2013, **4**, 432–436.
- (a) W. Henke, K. Herdel, K. Jung, D. Schnorr and S. A. Loening, *Nucleic Acids Res.*, 1997, **25**, 3957–3958; (b) M. Musso, R. Boccardi, S. Parodi, R. Ravazzolo and I. Ceccherini, *J. Mol. Diagn.*, 2006, **8**, 544–550; (c) H. A. Erlich, D. Gelfand and J. J. Sninsky, *Science*, 1991, **252**, 1643–1651; (d) R. T. D'Aquila, L. J. Bechtel, J. J. Videler, J. J. Eron, P. Gorczyca and L. C. Kaplan, *Nucleic Acids Res.*, 1991, **19**, 3749.
- (a) H. Li, J. Huang, J. Lv, H. An, X. Zhang, Z. Zhang, C. Fan and J. Hu, *Angew. Chem., Int. Ed.*, 2005, **44**, 5100–5103; (b) M. Li, Y. Lin, C. Wu and H. Liu, *Nucleic Acids Res.*, 2005, **33**, e184; (c) J. Jia, L. Sun, N. Hu, G. Huang and J. Weng, *Small*, 2012, **8**, 2011–2015; (d) Y. Liang, F. Luo, Y. Lin, Q. Zhou and G. Jiang, *Carbon*, 2009, **47**, 1457–1465; (e) X. Cao, J. Chen, S. Wen, C. Peng, M. Shen and X. Shi, *Nanoscale*, 2011, **3**, 1741–1747; (f) F. Sang, Z. Zhang, L. Yuan and D. Liu, *Analyst*, 2018, **143**, 1259–1267; (g) A. Khaliq, P. J. Sonawane, B. K. Sasi, B. S. Sahu, T. Pradeep, S. K. Das and N. R. Mahapatra, *Nanotechnology*, 2010, **21**, 255704; (h) X. Cao, X. Shi, W. Yang, X. Zhang, C. Fan and J. Hu, *Analyst*, 2009, **134**, 87–92.
- (a) G. Maurin, C. Serre, A. Cooper and G. Ferey, *Chem. Soc. Rev.*, 2017, **46**, 3104–3107; (b) J. Navarro-Sánchez, N. Almora-Barrios, B. Lerma-Berlanga, J. J. Ruiz-Pernía, V. A. Lorenz-Fonfria, I. Tuñón and C. Martí-Gastaldo, *Chem. Sci.*, 2019, **10**, 4082–4088; (c) W. Wang, S. Wu, J. Wang, Z. Li, H. Cui, S. Lin, J. Zhu and Q. Chen, *Chem. Sci.*, 2019, **10**, 4476–4485.
- (a) S. Wang, C. M. McGuirk, M. B. Ross, S. Wang, P. Chen, H. Xing, Y. Liu and C. A. Mirkin, *J. Am. Chem. Soc.*, 2017, **139**, 9827–9830; (b) W. Morris, W. E. Briley, E. Auyeung, M. D. Cabezas and C. A. Mirkin, *J. Am. Chem. Soc.*, 2014, **136**, 7261–7264; (c) J. S. Kahn, L. Freage, N. Enkin, M. A. A. Garcia and I. Willner, *Adv. Mater.*, 2017, **29**, 1602782; (d) B. V. Schmidt and . Macromol, *Rapid Commun. Mass Spectrom.*, 2019, 1900333; (e) Y. Liu, W. Hou, L. Xia, C. Cui, S. Wan, Y. Jiang, Y. Yang, Q. Wu, L. Qiu and W. Tan, *Chem. Sci.*, 2018, **9**, 7505–7509.
- Y. Pan, S. Zhan and F. Xia, *Anal. Biochem.*, 2018, **546**, 5–9.
- J. H. Cavka, S. Jakobsen, U. Olsbye, N. Guillou, C. Lamberti, S. Bordiga and K. P. Lillerud, *J. Am. Chem. Soc.*, 2008, **130**, 13850–13851.
- W. Rychlik, W. J. Spencer and R. E. Rhoads, *Nucleic Acids Res.*, 1990, **18**, 6409–6412.
- (a) H. Li, J. Huang, J. Lv, H. An, X. Zhang, Z. Zhang, C. Fan and J. Hu, *Angew. Chem.*, 2005, **44**, 5100–5103; (b) H. R. Kim, A. Baek, I. J. Lee and D. E. Kim, *ACS Appl. Mater. Interfaces*, 2016, **8**, 33521–33528; (c) L. Wang, Y. Zhu, Y. Jiang, R. Qiao, S. Zhu, W. Chen and C. Xu, *J. Phys. Chem. B*, 2009, **113**, 7637–7641.
- (a) M. Li, Y. C. Lin, C. C. Wu and H. S. Liu, *Nucleic Acids Res.*, 2005, **33**, e184; (b) Z. Xun, X. Zhao and Y. Guan, *Nanotechnology*, 2013, **24**, 355504.
- (a) H. Babaei, A. J. McGaughey and C. E. Wilmer, *Chem. Sci.*, 2017, **8**, 583–589; (b) B. Cui, C. O. Audu, Y. Liao, S. T. Nguyen, O. K. Farha, J. T. Hupp and M. Grayson, *ACS Appl. Mater. Interfaces*, 2017, **9**, 28139–28143; (c) J. Huang, X. Xia, X. Hu, S. Li and K. Liu, *Int. J. Heat Mass Transfer*, 2019, **138**, 11–16.
- (a) A. A. Balandin, S. Ghosh, W. Bao, I. Calizo, D. Teweldebrhan, F. Miao and C. N. Lau, *Nano Lett.*, 2008, **8**, 902–907; (b) A. Li, B. Zhou, C. S. Alves, B. Xu, R. Guo, X. Shi and X. Cao, *ACS Appl. Mater. Interfaces*, 2016, **8**, 25808–25817; M. Fujii, X. Zhang, H. Xie, H. Ago, K. Takahashi, T. Ikuta and T. Shimizu, *Phys. Rev. Lett.*, 2005, **95**, 065502.
- (a) M. Eddaoudi, J. Kim, N. Rosi, D. Vodak, J. Wachter, M. O'Keeffe and O. M. Yaghi, *Science*, 2002, **295**, 469–472; (b) O. M. Yaghi, M. O'Keeffe, N. W. Ockwig, H. K. Chae, M. Eddaoudi and J. Kim, *Nature*, 2003, **423**, 705–714; (c) G. Ferey, *Chem. Soc. Rev.*, 2008, **37**, 191–214.

- 16 (a) Y. J. Cui, Y. F. Yue, G. D. Qian and B. L. Chen, *Chem. Rev.*, 2012, **112**, 1126; (b) X. Zhu, H. Zheng, X. Wei, Z. Lin, L. Guo, B. Qiu and G. Chen, *Chem. Commun.*, 2013, **49**, 1276–1278.
- 17 (a) Y. Cui, Y. Yue, G. Qian and B. Chen, *Chem. Rev.*, 2012, **112**, 1126–1162; (b) H. T. Zhang, J. W. Zhang, G. Huang, Z. Y. Du and H. L. Jiang, *Chem. Commun.*, 2014, **50**, 12069–12072.
- 18 (a) X. Lian, Y. Fang, E. Joseph, Q. Wang, J. Li, S. Banerjee, C. Lollar, X. Wang and H. C. Zhou, *Chem. Soc. Rev.*, 2017, **46**, 3386–3401; (b) M. Ding, X. Cai and H. L. Jiang, *Chem. Sci.*, 2019, **10**, 10209–10230; (c) P. Li, J. A. Modica, A. J. Howarth, E. Vargas, P. Z. Moghadam, R. Q. Snurr, M. Mrksich, J. T. Hupp and O. K. Farha, *Chem*, 2016, **1**, 154–169; (d) J. Liang, F. Mazur, C. Tang, X. Ning, R. Chandrawati and K. Liang, *Chem. Sci.*, 2019, **10**, 7852–7858.