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A study on the effect of four thermoplastic elastomers on the properties of double-base propellants

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To improve the low-temperature mechanical properties of double-base propellants, glycidyl azide polymer-energetic thermoplastic elastomer (GAP-ETPE), polyether *block* copolyimide-thermoplastic elastomer (PEBA-TPE), 1,2-polybutadiene thermoplastic elastomer (PB-TPE), and ethylene oxide-tetrahydrofuran copolyester-thermoplastic elastomer (PET-TPE) were used to modify them. The effects of these four thermoplastic elastomers (TPEs) on the energy properties of double-base propellants were studied via theoretical calculations. The mechanical properties of double-base propellants at high, low, and room temperature were compared, and their thermal properties were analyzed. It was found that GAP-ETPE had a significant effect on improving the low-temperature mechanical properties of double-base propellants, with the maximum tensile strength and maximum elongation at a low temperature for the double-base propellant reaching 43.30 MPa and 6.24%, respectively.

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1 Introduction

A double-base propellant is a kind of solid propellant that is widely used in free-loading missiles and rocket engines.¹

In general, the main components of a double-base propellant are nitrocellulose (NC) and nitroglycerin (NG), along with ballistic stabilizers, combustion catalysts, and other additives.^{2,3} Nitrocellulose is the main structural material of double-base propellants,^{4,5} and nitroglycerin acts as the plasticizer for nitrocellulose. Double-base propellants show viscoelasticity above their glass transition temperature.^{6–8} However, double-base propellants are in the glassy state at a lower temperature, which would become brittle and easy to crack. As a result, the mechanical properties of double-base propellants are very poor at low temperatures. Therefore, the application range of double-base propellants is limited.

There are two main methods to improve the low-temperature mechanical properties of solid propellants: addition of polymer materials with good low-temperature mechanical properties (mostly a thermoplastic elastomer) and low-temperature plasticizer, respectively. The first is the most effective way. Thermoplastic elastomers (TPEs) are also called artificial rubber or synthetic rubber. TPEs have numerous characteristics of convenient processing and wide processing methods of ordinary plastics as well as various excellent properties such as

resistant to oil and aging, and high elasticity of traditional cross-linked vulcanized rubber.⁹ TPE as a solid propellant binder was proposed in the late 1970s, and there had been a lot of research so far.^{10–12}

Guo¹³ used varied proportions of GAP-ETPE/NC as binders to improve the thermal and mechanical properties of a nitramine propellant. The study found that after introducing GAP-ETPE into the binder system, the thermal stability of the propellant greatly improved. When the ratio of GAP-ETPE/NC in the binder system was 1 : 2, the mechanical properties of the nitramine propellant was relatively excellent, and the maximum impact strength increased from 3.58 kJ m^{−2} to 7.593.58 kJ m^{−2} compared with the formula without GAP-ETPE.

Bozic¹⁴ researched the introduction of a polyether *block* copolyimide-thermoplastic elastomer (PEBA-TPE) into solid propellants, which showed that the energy was higher than that of active HTPB-TPE/AP propellants, and it had appropriate mechanical properties as well as excellent processing performance and aging characteristics at a low cost. The properties of 1,2-polybutadiene are related to its crystallinity. When the crystallinity is 10–40%, 1,2-butadiene became a thermoplastic elastomer, namely 1,2-polybutadiene thermoplastic elastomer (PB-TPE). PB-TPE is a homopolymer with a crystalline phase as a hard segment and an amorphous phase as a soft segment. The application of PB-TPE to solid propellants have been reported.^{15,16} The mechanical properties and processability have also been improved.

Chen¹⁷ synthesized an ethylene oxide-tetrahydrofuran copolyester-thermoplastic elastomer (PET-TPE) with PET and polyethylene glycol (PEG) mixed polyether as the soft segment.

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When the molecular weight and content of PEG were controlled to 4000 g mol⁻¹ and 6% of soft segments, respectively, the property of PET-TPE was optimal. The propellant prepared with the thermoplastic elastomer had good processing temperature and excellent mechanical properties. The normal (20 °C), high (50 °C), and low (-40 °C) temperature tensile strengths were 1.14 MPa, 0.51 MPa, and 9.94 MPa, respectively. The combustion properties showed that the maximum impulse reaches up to 265–270 s which was slightly higher than that of the HTPB propellant. Also, the combustion of the propellant was easy to control.

In order to solve the problem of double-base propellant's brittleness at a lower temperature, four TPEs (PET-TPE, GAP-ETPE, PEBA-TPE, and PB-TPE) were selected as modifiers of the double-base propellant. In this study, the change in the theoretical properties of the propellant was compared by theoretical calculations, and the change in mechanics and α transition temperature were measured by a static tensile mechanical property test and dynamic thermomechanical analysis, respectively. The influence of TPEs on the thermal decomposition behavior of a double-base propellant was analyzed by a thermogravimetry test. Combined with the influence of the double-base propellant on mechanics and energy, the best type of thermoplastic elastomer was determined.

2 Experimental

2.1 Materials

Tetrahydrofuran (THF), AR, was purchased from Beijing Chemical Reagent Factory. Deionized water and PET-thermoplastic polyurethane elastomer (PET-TPE), $\overline{M}_n = 82760$ g mol⁻¹, were prepared in the laboratory.¹⁸ GAP-energetic thermoplastic polyurethane elastomer (GAP-ETPE), $\overline{M}_n = 31490$ g mol⁻¹, was synthesized in the laboratory.¹⁹ Polyether *block* copolyimide elastomer (PEBA-TPE), $\overline{M}_n = 69000$ g mol⁻¹, was purchased from Arkema, France. 1,2-Polybutadiene thermoplastic elastomer (PB-TPE); $\overline{M}_n = 72000$ g mol⁻¹, was purchased from JSR, Japan. A compound of nitrocellulose and nitroglycerin (NG: 47.5%, NC: 51.5%, dimethyl diphenyl urea: 1.0%), NC (nitrogen content: 12.6%), dinitrotoluene (DNT), titanium dioxide (TiO₂); adipic acid copper (d-Cu), purchased from Shanxi North Xing'an Chemical Industry Co., Ltd, China, were of technical grade.

2.2 Characterization and analysis

(1) Theoretical calculation

NASA "propellant property computer" software calculated the specific impulse (I_s), oxygen balance (OB), combustion temperature (T_c), explosion heat (Q), and density (ρ) of each formula, and the test pressure was 7 MPa.

(2) Static tensile mechanical property test

The static tensile mechanical property test was performed on an AGS-J, Shimadzu, Japan. Test method is referenced ASTM D638-2003. The stretch rate was 10 mm min⁻¹, and the test temperature was -40 °C, 20 °C, and 50 °C.

(3) Dynamic thermomechanical analysis (DMA)

The dynamic thermomechanical analysis was performed on a DMA/SDTA861e, METTLER-TOLEDO, Switzerland. The sample size was 4 × 4 mm, the frequency was 1 Hz, the amplitude was 5 μ m, the heating rate was 2 K min⁻¹, and the range of temperature was -100 to 120 °C.

(4) Thermogravimetric analysis (TG)

The thermogravimetric analysis was performed on a TGA/DSC1SF/417-2, METTLER-TOLEDO, Switzerland. The sample mass was less than 1 mg, the test temperature was 30–600 °C, the heating rate was 10 K min⁻¹, and the N₂ flow rate was 40 mL min⁻¹.

2.3 Preparation of a thermoplastic elastomer-modified double-base propellant

2.3.1 Formulas of the elastomer-modified double-base propellant. A propellant formula containing 52.3% NC, 39.2% NG, and 8.5% additives was regarded as the control composition, and four propellant compositions containing 8.5% additives with the partial replacement of NC and NG with four elastomers were formulated. The specific propellant formulas are given in Table 1. The original formula and formulas of GAP-ETPE, PEBA-TPE, PB-TPE, and PET-TPE-modified propellants are recorded as DB-0, DB-1, DB-2, DB-3, and DB-4, respectively. Table 1 lists the mass percentages of raw materials, corresponding to different double-base propellants.

In the formulas, dinitrotoluene (DNT) could increase the energy of the propellant and act as a cosolvent; TiO₂ acts as a ballistic stabilizer; copper adipate (d-Cu) as a catalyst; dimethyl diphenyl urea (C2) plays a stabilizing role.

2.3.2 Preparation of the thermoplastic elastomer-modified double-base propellant. TPEs were dissolved in THF to prepare a solution, and then the mixture was added into a polytetrafluoroethylene beaker with other raw materials of the propellant, immersed in deionized water. Further, the polytetrafluoroethylene beaker was heated in a water bath at 50 °C, and stirred at 500 rpm for 2 h to evenly disperse it into a gray-green suspension that was filtered for about 5 min. Next, the open mill was turned on, the pitch of the twin rolls was set to 1 mm, and the temperature was set to 100 °C to perform preheating. Finally, the mixture was calendered in an open mill at 100 °C to obtain a propellant sample. The preparation procedure of the DB-0 propellant was same as the above except that TPE was added after dissolved. The specific preparation process is shown in Fig. 1.

Table 1 The formulas of double-base propellants (wt%)

Propellant formula	NC	NG	Elastomer	DNT	TiO ₂	d-Cu	C2
DB-0	52.3	39.2	0	2.2	1.5	3	1.8
DB-1	50.3	37.2	4	2.2	1.5	3	1.8
DB-2	50.3	37.2	4	2.2	1.5	3	1.8
DB-3	50.3	37.2	4	2.2	1.5	3	1.8
DB-4	50.3	37.2	4	2.2	1.5	3	1.8





Fig. 1 Preparation process flow chart of the double-base propellant.

Table 2 Theoretical calculation of the properties of double-base propellants

Parameter	DB-0	DB-1	DB-2	DB-3	DB-4
$\rho/\text{g cm}^{-3}$	1.620	1.599	1.583	1.565	1.580
I_s/s	237.9	234.7	232.6	223.5	225.8
OB	0.645	0.623	0.594	0.569	0.598
T_c/K	2656.9	2590.0	2516.5	2346.2	2431.3
$Q/\text{kJ kg}^{-1}$	3758	3659	3569	3306	3441

3 Results and discussion

3.1 Theoretical calculation of the properties of the thermoplastic elastomer-modified double-base propellant

According to the molecular formula, density, and enthalpy of formation of each thermoplastic elastomer, the theoretical values of density (ρ), specific impulse (I_s), oxygen balance (OB), combustion temperature (T_c), and explosion heat (Q) for double-base propellant formulas were calculated by the NASA "prospect property computer" software, as shown in Table 2.

It can be seen from Table 2 that the theoretical values of ρ , I_s , OB, T_c , and Q of double-base propellants was reduced with the addition of four thermoplastic elastomers. In terms of density, GAP-ETPE had the smallest impact on the propellant, whereas PB-TPE had the largest impact. In terms of energy, GAP-ETPE was the only energetic thermoplastic material among four TPEs, so the theoretical specific impulse of the double-base propellant decreased relatively less with the addition of GAP-ETPE, and the energy level of the propellant was not greatly reduced. The other three TPEs caused the specific impulse, combustion temperature, and explosion heat of double-base propellants to be reduced to a certain extent. From another aspect, the introduction of TPEs reduced the combustion temperature, which improved the ablation resistance of the propellant engine. Thus, the above analysis shows that it is not suitable to add too much elastomer.

3.2 Static mechanical properties of the thermoplastic elastomer-modified double-base propellant

The static mechanical results of the original formula and the four thermoplastic elastomer-modified double-base propellants at room temperature, low temperature, and high temperature are shown in Fig. 2, 3, and 4, respectively.

It can be seen from Fig. 2–4 that the mechanical properties of DB-2 had decreased at room temperature and high temperature, but the maximum tensile strength was 33.03 MPa at a low temperature, which was slightly improved compared with DB-0. For DB-3, the mechanical properties of the propellant decreased at a low temperature and room temperature except for the small increase at a high temperature, where the maximum tensile strength reached 2.74 MPa. For DB-4, the addition of PET-TPE improved the mechanical properties at room temperature and the maximum tensile strength at a high temperature.

The maximum tensile strength of DB-1 at a low temperature increased from 28.42 MPa to 43.30 MPa, and the maximum elongation increased from 5.17% to 6.24% as compared to DB-0. The mechanical properties of double-base propellants were improved by adding GAP-ETPE for the following reasons: on the

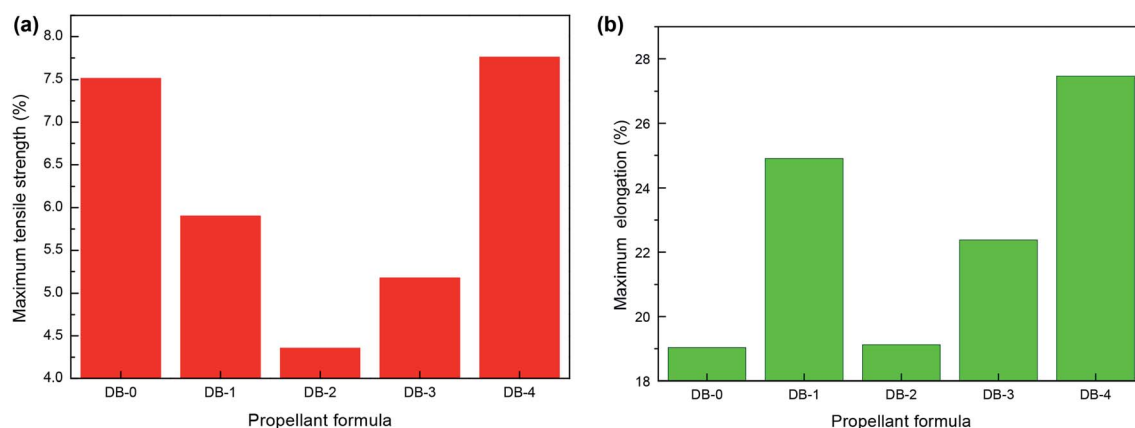


Fig. 2 The static mechanical properties of double-base propellants at room temperature (20 °C). (a) Maximum tensile strength. (b) Maximum elongation.



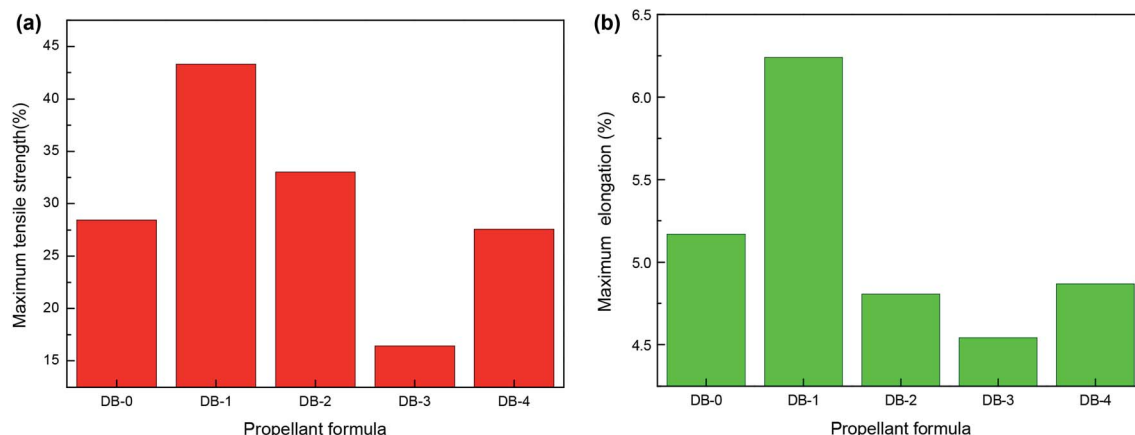


Fig. 3 The static mechanical properties of double-base propellants at low temperature ($-40\text{ }^{\circ}\text{C}$). (a) Maximum tensile strength. (b) Maximum elongation.

one hand, as an energetic binder, GAP-ETPE itself had good mechanical properties, which interacted with the double-based component, particularly the nitrocellulose molecular chain as a skeleton structure, through a secondary bond. In addition, since both nitrocellulose and GAP-ETPE have longer molecular chains, the entanglement would occur between the molecular chains, and the bonding effect occurred. On the other hand, GAP-ETPE played a role of “intramolecular plasticization” in semi-rigid cellulose macromolecules, the molecular chain spacing was widened, and thus the elongation of NC was increased.²⁰ However, at a high temperature ($50\text{ }^{\circ}\text{C}$), as the thermal movement of the molecular chains strengthened, the distance between the molecular chains increased, and the hydrogen bonding effect weakened, the adhesion of GAP-ETPE to the double-based components was reduced. Besides, the reduction of nitroglycerin was not conducive to plasticizing nitrocellulose and GAP-ETPE, so the maximum tensile strength of DB-1 at a high temperature was lower than that of DB-0.

Based on the mechanical properties at room temperature, low temperature, and high temperature as a reference, it can be

seen from Fig. 2–4 and analysis above that DB-1 has good mechanical properties at the room temperature, low temperature, and high temperature. To summarize, among the four thermoplastic elastomers, GAP-ETPE had the best effect on improving the mechanical properties of the double-base propellant.

3.3 Dynamic thermomechanical analysis of the thermoplastic elastomer-modified double-base propellant

Since the double-base propellant is a heterogeneous system, its relaxation process is complicated. Two relaxation processes could be obtained by the DMA test. The one at a lower temperature is called β relaxation, and the other at a higher temperature is called α relaxation. It is widely believed that polymer materials are in a rigid glassy state below the glass transition temperature (T_g). However, below T_g , a double-base propellant could still undergo a forced high-elastic deformation. Therefore, the double-base propellant still maintained a certain elasticity between α and β transition temperature. Some researchers have used β transition as the glass transition

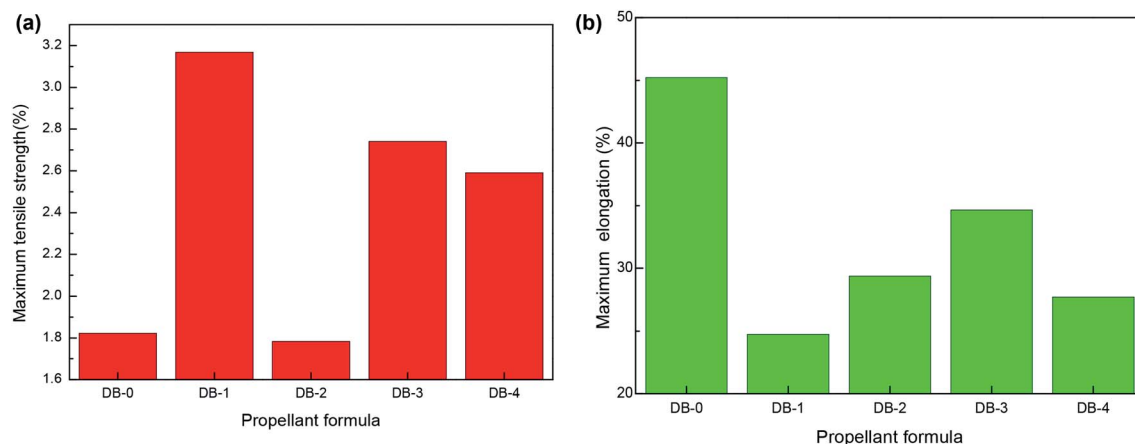


Fig. 4 The static mechanical properties of double-base propellants at high temperature ($50\text{ }^{\circ}\text{C}$). (a) Maximum tensile strength. (b) Maximum elongation.



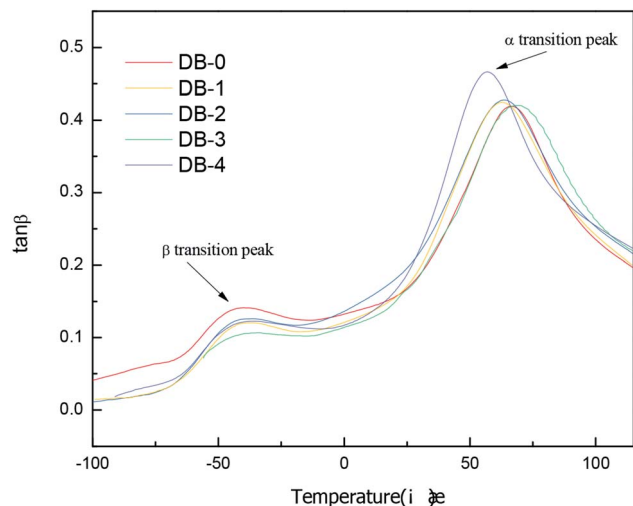


Fig. 5 Loss tangent curve of double-base propellants.

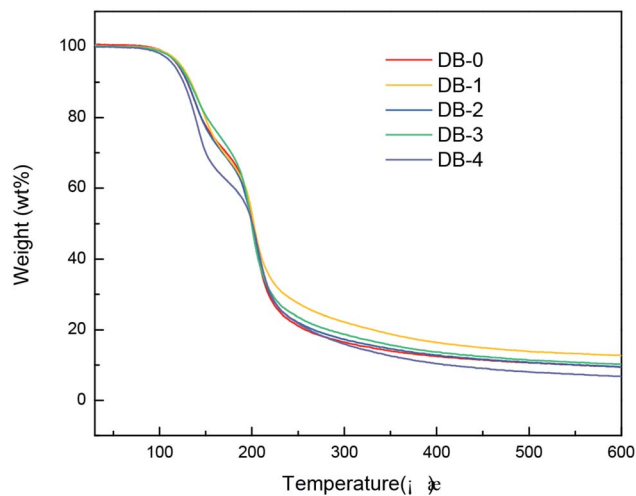


Fig. 6 TG curves of double-base propellants.

temperature of double-base propellants in the past.²¹ However, during the β transition of the double-base propellant, in addition to the modulus, there were some complicated relaxation behaviors such as sudden changes of specific heat capacity and thermal expansion coefficient. The β transition temperature was in a wide range, so it could not be accurately used as the glass transition temperature of the double-base propellant. In the DMA test, at low frequencies, the loss modulus E'' of the α relaxation of the double-base propellant was a shoulder peak corresponding to β relaxation, which was difficult to measure accurately, but the peak of loss tangent $\tan \delta$ of α relaxation was the largest and clearest, and it had good repeatability and reproducibility by many experiments. Therefore, in the DMA test, the α -transition temperature of the loss tangent was usually used as T_g of the double-base propellant by some researchers,^{22,23} which was also a common reference in the present study.

In this study, DB-0, DB-1, DB-2, DB-3, and DB-4 were tested by DMA. The change in the loss tangent of the elastomer was studied, and the corresponding α transition temperature was obtained. The results are shown in Fig. 5.

Table 3 shows the loss tangent α transition temperature and peak value results of double-base propellants by the DMA test.

It can be seen from Fig. 5 that the tangent curve of double-base propellants has two-loss peaks. The peak position of the

β transition was between -60°C and 10°C and was related to the joint movement of NG and NC molecular side groups. The position of the α transition peak was at about $50\text{--}70^\circ\text{C}$, which showed that TPE-modified double-base propellants were in the glassy state above room temperature. Compared with DB-0, the peak value of the α transition increased with the addition of TPEs. DB-4 had the highest peak value of α -transition. The α transition temperature of DB-0, DB-1, DB-2, DB-3, and DB-4 were 66.9°C , 62.9°C , 63.9°C , 69.3°C , and 56.5°C , respectively. Compared with DB-0, the addition of PB-TPE increased the α transition temperature of the double-base propellant, while the addition of PEBA-TPE, GAP-ETPE, and PET-TPE decreased T_g , particularly the formula containing PET-TPE, which decreased T_g to 10°C .

From the glassy state to the high elastic state, the motion of the small units occurred first, and then the chain segmental motion of the molecular main chains occurred. The addition of thermoplastic elastomers reduced the α transition temperature of the propellant system and increased the operating

Table 3 Loss tangent α transition temperature and peak values of double-base propellants

Propellant formula	α transition peak value	α transition temperature/ $^\circ\text{C}$
DB-0	0.419	66.9
DB-1	0.424	62.9
DB-2	0.428	63.9
DB-3	0.420	69.3
DB-4	0.466	56.5

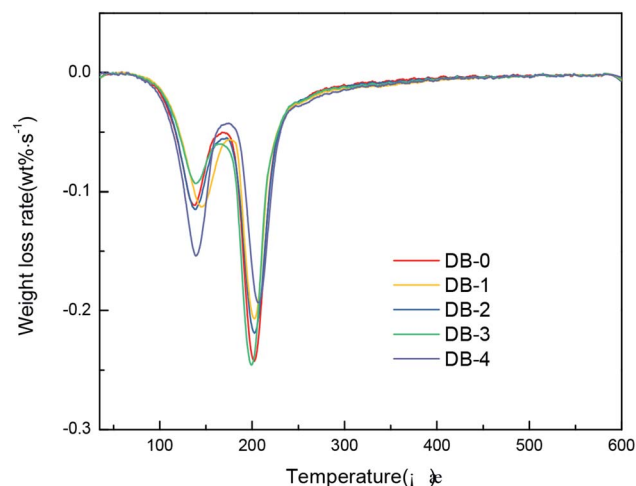


Fig. 7 DTG curves of double-base propellants.



Table 4 The thermogravimetric decomposition temperature of the double-base propellants^a

Propellant formula	Temperature of initial decomposition	Stage 1		Stage 2	
		Maximum decomposition rate/% s ⁻¹	Temperature of maximum decomposition rate/°C	Maximum decomposition rate/% s ⁻¹	Temperature of maximum decomposition rate/°C
DB-0	119.4	−0.11	138.2	−0.24	202.3
DB-1	120.6	−0.09	138.8	−0.25	199.3
DB-2	119.9	−0.11	138.6	−0.22	202.5
DB-3	122.4	−0.11	144.8	−0.21	202.6
DB-4	114.8	−0.15	139.2	−0.19	206.6

^a The initial decomposition temperature is referred to as the temperature when the weight loss rate of the sample was 5%.

temperature range of the propellant. It can be seen from Section 3.2 that the addition of thermoplastic elastomers not only improved the tensile strength of the propellant at a low temperature but also ensured that the tensile strain was almost unchanged. Therefore, the mechanical properties of the double-base propellant at the low temperature were improved, which was consistent with the increase of the peak value of α transition and the decrease in the α transition temperature in the DMA curve.

3.4 Thermogravimetric analysis of the thermoplastic elastomer-modified double-base propellant

The thermogravimetric tests of the five propellant formulas were carried out. The results are shown in Fig. 6 and 7.

Table 4 showed initial decomposition temperature, maximum decomposition rate, and temperature of the maximum decomposition rate of double-base propellants obtained by the TG test.

The thermal decomposition reaction of NC and NG in the double-base propellant was autocatalytic. The decomposition products played a catalytic role in the decomposition, and the decomposition released a lot of heat to accelerate the chemical reaction of the system. The thermal decomposition of the double-base propellant was divided into two stages, the first stage was the decomposition of nitroglycerin, and the second stage was the decomposition of nitrocellulose. It can be seen from TG and DTG curves that the addition of four thermoplastic elastomers did not change the thermal decomposition behavior of the double-base propellant. The change in the initial decomposition temperature of four elastomer-modified double-base propellants were within 5 °C. In terms of the decomposition rate and the temperature of maximum decomposition rate in the two stages, the four elastomers had a minimal effect.

4 Conclusion

In this study, the theoretical calculation properties of different formulas were compared, and the effects of thermoplastic elastomers on the dynamic and static mechanical and thermal properties of the double-base propellant were studied. The optimal formula was comprehensively determined. The main results are as follows:

(1) The theoretical calculation showed that the addition of four thermoplastic elastomers would result in a loss of the energy level of the double-based propellant; therefore, it was not suitable to add too much elastomer. The energy loss of the double-base propellant with GAP-ETPE was the least.

(2) The dynamic thermal-mechanical analysis was used to study the changes in the thermal transition behavior of four TPE-modified double-base propellants as compared to the original formula. It was found that the other three thermoplastic elastomers could reduce α transition temperature of double-base propellant except for PB-TPE. The thermogravimetric analysis showed that the addition of four thermoplastic elastomers did not change the thermal decomposition.

(3) By a comprehensive consideration of mechanical properties and energy level, GAP-ETPE had the best effect on improving the low-temperature mechanical properties of double-base propellant. Its maximum tensile strength and maximum elongation at low temperature could reach 43.30 MPa and 6.24%, respectively, which was 52.4% and 20.7% higher than that of the original formula.

Conflicts of interest

There are no conflicts to declare.

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