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Diastereoselective synthesis of CF_3 -dihydrobenzofurans by [4+1] annulation of in situgenerated CF_3 -o-quinone methides and sulfur ylides†

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An efficient and highly diastereoselective synthesis of CF_3 -dihydrobenzofurans by the reaction of *in situ*-generated CF_3 -oQMs in the presence of a base with sulphur ylides is put forward. The generality of the present developed method was well studied with diverse substrates to access the corresponding products in excellent yields. The highly reactive CF_3 -oQM has been utilized first time for the annulation reaction.

Fluorine or fluoroalkyl group-containing organic molecules occupy a vital position in drug discovery due to the unique impact of fluorine atom in terms of lipophilicity, permeability, and protein-binding.¹ In the last two years, 45% of FDA-approved small molecule pharmaceuticals are fluorinated, which denotes the importance of synthesizing fluorinated molecules, with special emphasis on medicinal chemistry, for the identification of new scaffolds. Among the fluorinated functional groups, the trifluoromethyl group has emerged as one of the imperative fluoroalkyl groups to enhance the bioefficacy and metabolic stability of the corresponding motifs, which is needed for the identification of lead compounds.² Thus, finding new methods for the inclusion of the CF₃ group into novel biological entities is always desirable and challenging.

In this context, *ortho*-quinone methides (*o*QMs) are powerful reactive intermediates in synthetic organic chemistry to construct complex medium sized rings.^{3,4} Since *o*QM was first observed in 1907, it created a large impact in the synthesis of oxygen-containing benzannulated rings, which are of interest as photochromic materials and biologically active compounds.³ However, the reactions of *o*QMs were restricted to electron-rich substrates due to their high electrophilic nature. However, annulation reactions of diversely substituted *o*QMs (substitution on the exocyclic double bond) were explored extensively for the construction of oxygen-containing complex heterocyclic

Kato *et al.* were the first to report the nucleophilic addition of Grignard reagents and amines to the *in situ*-generated trifluoromethyl-substituted *ortho*- and *para*-quinone methides

Previous work:

a) Nucleophilic additions of in-situ generated CF₃-oQM

b) Nucleophilic additions of *in-situ* generated CF₃-pQM (Ref. 6)

Present work:

c) First annulation reaction of in-situ generated CF₃-oQM

Fig. 1 Previous work vs. the present work.

structures;³ however, it is quite surprising that annulation reactions involving trifluoromethyl-substituted *o*QM (CF₃-*o*QM) have not been reported yet, especially because CF₃-*o*QM is like a gold mine and could open the realm to construct versatile fluorinated oxygen architectures.

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(Fig. 1a).5 After that, there were no reports of this in situgenerated CF₃-oOM for any nucleophilic additions and annulations. In 2020, several papers were published on in situgenerated CF3-para-quinone methides (CF3-pQM) by quickly trapping them with different carbon and hetero atom-centered nucleophiles (Fig. 1b).6 Notably, Waser et al. demonstrated that CF₃-pQM has higher electrophilicity parameter (E) when compared to other substituted para-quinone methides. 6d Similarly, we hypothesized that CF₃-oQMs may also have higher "E" in comparison with the corresponding oQMs. Thus, the utilization of this highly reactive CF₃-oQM for annulation reactions is extremely challenging and equally desirable towards the synthesis of novel organofluorine molecules. In continuation of our research interest on oQM-based annulations and the development of new fluorinating methodologies,7 herein, we report for the first [4 + 1] annulation of in situ-generated CF₃oQM with sulphur ylide to access trifluoromethyl-substituted dihydrobenzofurans with high diastereoselectivity.

The study was initiated by exposing 2-(1-chloro-2,2,2trifluoroethyl) phenol 1a and sulphur vlide 2a to 1.2 equiv. of Cs₂CO₃ at room temperature in THF. Delightedly, the in situgenerated CF₃-oQM was successfully trapped with sulphur ylide to afford the corresponding trifluoromethyl-substituted dihydrobenzofuran 3a in 80% yield (Table 1, entry 1) with high diastereoselectivity. The investigation of a variety of solvents revealed that THF was the optimal solvent for the [4 + 1] annulation reaction (Table 1, entries 2-5). A quick survey was then conducted with different bases, and organic base DABCO was found to be the best to deliver 3a in 93% yield (Table 1, entries 6-9). The control experiment showed that the reaction in the absence of a base failed to produce the [4 + 1] annulation product (Table 1, entry 10). The reaction conditions in entry 6 (Table 1) were optimal and gave the product in 93% yield with >20:1 dr. The configuration of the obtained product was confirmed as trans from the X-ray crystallographic structure of compound 3a (CCDC 2023269).

Table 1 Optimization of reaction conditions^a

Entry	Base	Solvent	Yield ^b	$\mathrm{d}\mathrm{r}^c$
1	Cs ₂ CO ₃ (1.2 equiv.)	THF	80%	>20:1
2	Cs_2CO_3 (1.2 equiv.)	DCM	62%	>20:1
3	Cs_2CO_3 (1.2 equiv.)	CH_3CN	51%	>20:1
4	Cs_2CO_3 (1.2 equiv.)	Toluene	Trace	_
5	Cs_2CO_3 (1.2 equiv.)	MeOH	Trace	_
6	DABCO (1.2 equiv.)	THF	93%	>20:1
7	DBU (1.2 equiv.)	THF	65%	>20:1
8	Et ₃ N (1.2 equiv.)	THF	Trace	_
9	K ₂ CO ₃ (1.2 equiv.)	THF	45%	>20:1
10	_ ` ` ` `	THF	NR	_

 $[^]a$ Reaction conditions: 1a (0.6 mmol), 2a (0.5 mmol), base (0.6 mmol) in solvent (2 mL) at rt. b Isolated yields. c dr was determined by $^{19}{\rm F}$ NMR.

With the determination of optimized conditions, the substrate scope for the [4 + 1] annulation reaction was scrutinized by the reaction of compound 1a with a broad array of sulphur ylides (Table 2). First, the reaction of electron-rich substrates 2b (CH₃) and 2c (OCH₃) afforded CF₃-dihydrobenzofurans 3b (91%) and 3c (90%) in very good yields, respectively. The halogen-containing sulphur ylides 2d, 2e, and 2f (F, Cl, and Br) also proceeded smoothly to furnish the required products 3d-f in excellent yield (up to 85%) and we observed a slight improvement in the yield from fluoro to bromo substrates. Further, the reaction of naphthalene-derived sulphur ylide 2h also participated well in the reaction to deliver the required CF₃-dihydrobenzofuran 3h in 74% yield. The electron-deficient substrate 2i (CN) also underwent the [4 + 1] annulation reaction very well to give CF₃-dihydrobenzofuran 3i in 71% yield. CF₃-oQM generated in situ from compound 1a, was trapped with a wide range of sulphur ylides without any effect on the substituents to yield the required trifluoromethylsubstituted dihydrobenzofurans in good yields with high diastereoselectivity (dr > 20:1).

Next, we investigated the substrate scope with respect to *ortho*-hydroxy- CF_3 -benzyl chlorides **1b-d** to delineate the generality of the present [4 + 1] annulation under the standard reaction conditions (Table 3). The reactions with CF_3 -benzyl chlorides having electronically dissimilar groups as substituents, proceeded well to furnish the desired products in good yields. The reaction of methyl (**1b**)- and methoxy (**1c**)-substituted CF_3 -benzyl chlorides with a variety of sulphur ylides delivered the corresponding CF_3 -dihydrobenzofurans **3j-o** in

Table 2 The substrate scope of sulphur ylides $(2a-i)^a$

 $[^]a$ Reaction conditions: 1 (0.6 mmol), 2 (0.5 mmol), DABCO (0.6 mmol), THF (2 mL). Isolated yield. dr was determined by $^{19}{\rm F}$ NMR.

Table 3 The substrate scope of *ortho*-hydroxy CF_3 -benzyl chloride (1b-e)

 a Reaction conditions: 1 (0.6 mmol), 2 (0.5 mmol), DABCO (0.6 mmol), THF (2 mL) at rt. Isolated yield. dr was determined by 19 F NMR.

excellent yields with good diastereoselectivity. Further, CF_3 -oQMs generated from the bromo (1d)- and chloro (1e)-substituted CF_3 -benzyl chlorides were also successfully trapped to deliver the desired products 3p-r in good yields (up to 79% with >20 : 1 dr).

To determine the synthetic utility of the present transformation, we executed a gram scale [4+1] annulation reaction of compound **1a** with **2a** under applied reaction conditions, which gave CF₃-dihydrobenzofuran **3a** in 85% yield (Scheme 1). Later, we exposed compound **3a** to vinyl magnesium bromide at 0 °C in THF to furnish the corresponding alcohol **4** in good yield with excellent diastereoselectivity (dr > 19:1).

The plausible reaction mechanism for the base-catalyzed [4+1] annulation of *ortho*-hydroxy-CF₃-benzyl chloride **1a** with compound **2a** is depicted in Fig. 2. Initially, CF₃-oQM was generated in the presence of a stoichiometric amount of base.^{5,8} This highly electrophilic CF₃-oQM undergoes nucleophilic addition with compound **2a** to form a new C–C bond in **TS I**; the diastereoselectivity in **TS I** arises due to the favourable steric repulsions between the trifluoromethyl group and sulphur ylide, resulting in the final compound with *trans* configuration. Finally, the intramolecular nucleophilic substitution in **TS I** by

Scheme 1 The synthetic transformation of compound 3a.

Fig. 2 The plausible reaction mechanism.

oxygen with a sulphonium moiety furnishes the desired CF₃-dihdrobenzofuran 3a in good yield.

Conclusions

In conclusion, we demonstrated a novel method for the synthesis of CF_3 -dihdrobenzofurans 3 via [4 + 1] annulation of *ortho*-hydroxy- CF_3 -benzyl chlorides 1 with sulphur ylides 2 under basic conditions in good yields (up to 93%) and diastereoselectivities (>20:1). The highly reactive CF_3 -oQM, due to the electron-withdrawing nature of the CF_3 group, was trapped successfully in the present [4+1] annulation. This annulation is the first example for the trapping of trifluoromethyl-substituted oQM. The core skeleton of dihydrobenzofuran obtained in the present protocol has received huge attention in literature, 9 and CF_3 present at a strategic position may improve the biological activities of molecules tremendously. Further, the expansion of annulation reactions via in situ-generated CF_3 -oQMs is in progress in our laboratory to construct versatile trifluoromethyl-substituted oxygen-containing heterocycles.

Conflicts of interest

There are no conflicts of interest to declare.

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