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Anomalous boron isotope effects on electronic structure and lattice dynamics of CuB₂O₄†‡

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Copper metaborate had a unique crystal structure and exhibited noteworthy magnetic phase transitions at 21 and 10 K. The electronic structure and lattice dynamics of copper metaborate Cu¹¹B₂O₄ single crystals were investigated and compared with the optical properties of CuB₂O₄, to assess the boron isotope effect. The optical absorption spectrum at room temperature revealed two charge-transfer bands at approximately 4.30 and 5.21 eV with an extrapolated direct optical band gap of 3.16 ± 0.07 eV. Compared with the data on CuB₂O₄, the electronic transitions were shifted to lower energies upon the replacement of a heavier boron isotope. The band gap was also determined to be lower in Cu¹¹B₂O₄. Anomalies in the temperature dependence of the optical band gap were observed below 21 K. Furthermore, 38 Raman-active phonon modes were identified in the room-temperature Raman scattering spectrum of Cu¹¹B₂O₄, which were also observed in CuB₂O₄ with a shift to lower frequencies. No broadening caused by isotopic changes was observed. As the temperature decreased, phonon frequencies shifted to higher wavenumbers and the linewidth decreased. Anomalous softening in the Raman peaks below 21 K was also revealed.

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I. Introduction

Isotopes are atoms with the same number of protons and electrons but a different number of neutrons. Isotopes differ in mass but are chemically identical. Early theoretical works have established that only properties dependent on nuclear mass are altered in isotopes. The most obvious instance of such dependence is the mass-dependent effect on the harmonic lattice vibrational frequency, which is represented as $1/\sqrt{m}$. Isotope-dependent properties also include properties that are affected by changes in unit cell volume, atomic hopping mobility, and anharmonicities—for example, thermal conductivity, thermal expansion, melting temperature, nuclear magnetic resonance, and superconducting phase transition temperature. The electronic band structures were once believed to remain identical in isotopes under such changes. However, this belief has been justified only by atomic-spectra data. Evidence has revealed that in the case of molecular spectra, the effect of mass involves differences between isotopes through the mechanism of electron–phonon coupling.^{1,2} The presence of an isotope effect is an effective indicator of phonon-mediated superconductivity in materials and constitutes support for the Bardeen–Cooper–Schrieffer (BSC) theory of superconductivity.^{3–5}

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Studies on the isotope effects of different materials are crucial in clarifying these materials' properties, especially when investigating phonons and their interactions. For instance, isotope labeling in the study of graphene provided direct evidence that the growth mechanism is substrate dependent.⁶ Furthermore, isotopic substitution in diamond (¹²C and ¹³C) and in diamond-type semiconductors (Si and Ge) was noted to increase the indirect electronic energy gap because of isotopically induced changes in lattice volume and electron–phonon interaction.^{7,8} Isotopic shifts in the superconducting gap and binding energy of high T_c superconductors were used to analyze the nature of the electron–lattice coupling and elucidate the electron pairing mechanism in high T_c superconductors.^{3,9,10} Isotope shifts were demonstrated to be an essential test of superconductivity.

BCS superconductors with electron–phonon coupling are expected to have a total isotope effect coefficient of 0.5. The isotope effect coefficient α , is defined as $\alpha = -d \ln T_c / d \ln M$ where M is the atomic mass. The coefficient α was measured in the superconductor MgB₂. An analysis of the magnetization data revealed a large boron isotope effect, which indicated superconductivity in this material. The extrapolated T_c exhibited at least 1 K difference between the ¹¹B and ¹⁰B samples. The isotope effects from Mg were small compared with the boron isotope effects, which clearly indicated that phonons associated with boron vibrations are crucial in superconductivity. The findings further revealed superconductivity in transition metal diborides (*i.e.* NbB₂) and borocarbides (*e.g.*, YNi₂B₂C and LiNi₂B₂C).^{4,11} The natural abundance of stable boron isotopes



facilitated the assessment of isotopic effects, despite a lack of suitable isotopes for the other elements.

Although isotopic effects are well studied in high T_c superconductors, research on isotopic shifts in phase transition of magnetic materials is scarce. Copper metaborate, CuB_2O_4 exhibited complex magnetic properties that have been extensively investigated. Canted antiferromagnetic ordering occurred below 21 K followed by a commensurate-to-incommensurate magnetic structure at 10 K.^{12,13} CuB_2O_4 crystallizes in a unique noncentrosymmetric $I\bar{4}2d$ structure which contains BO_4 tetrahedrons with Cu^{2+} ions between them. These Cu^{2+} ions are located in two inequivalent crystallographic positions (see ESI Fig. 1†). Various studies reported that CuB_2O_4 exhibited the unusual magnetoelectric properties and complex magnetic, electric and optical coupling phenomenon.^{14–17} Notably, CuB_2O_4 containing ^{11}B and ^{10}B isotopes exhibited no significant shifts in the phase transition temperatures, as demonstrated by the nearly identical magnetic susceptibility illustrated in Fig. 1. This finding contrasted with the large isotope effects identified in high T_c superconductors. Studies on antiferromagnetic cuprates have also revealed negligible oxygen isotope effects in the magnetic phase transition temperatures.^{18,19}

The optical features of antiferromagnetic copper metaborate under isotopic substitution remain unexplored. In this paper, the boron isotope effects in copper metaborate were optically and vibrationally investigated. We analyzed the dependence of the electronic transition energy on the boron isotope mass as obtained from spectroscopic ellipsometry. Our measurements revealed notable isotope effects in the charge-transfer excitation energies and the direct band gap. The Raman isotopic shifts accorded with the calculated shifts because of changes in isotopic mass, which enabled the identification of the boron-related vibrations. Anomalies in the band gap, the peak energy of charge-transfer bands, and Raman-active phonon modes of $\text{Cu}^{11}\text{B}_2\text{O}_4$ were observed at 21 K, confirming the spin–

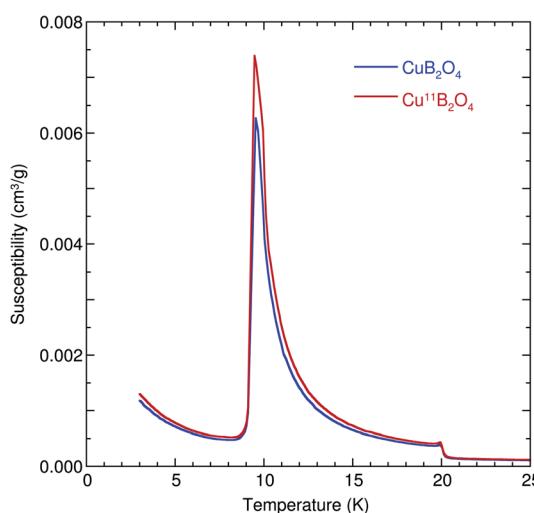


Fig. 1 The temperature dependence of magnetic susceptibility measured in an applied magnetic field of 50 Oe for $H \perp [001]$ of single-crystal $\text{Cu}^{11}\text{B}_2\text{O}_4$ and CuB_2O_4 .

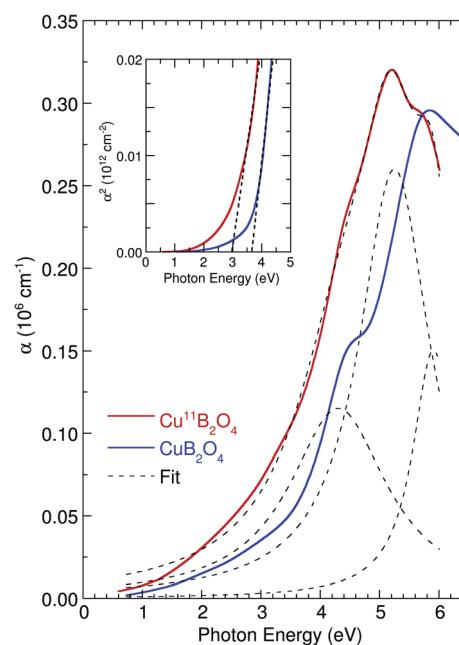


Fig. 2 Optical absorption spectra of $\text{Cu}^{11}\text{B}_2\text{O}_4$ and CuB_2O_4 at room temperature. The dashed lines illustrate the best fit with the Lorentz function. Inset illustrates the direct band gap analysis of $\text{Cu}^{11}\text{B}_2\text{O}_4$ and CuB_2O_4 at 300 K.

charge–lattice interaction in this material. These results elucidate the intricate lattice dynamics and the influence of the isotope effects on the electronic states, which are vital for the isotope engineering of optoelectronic and photonic devices.

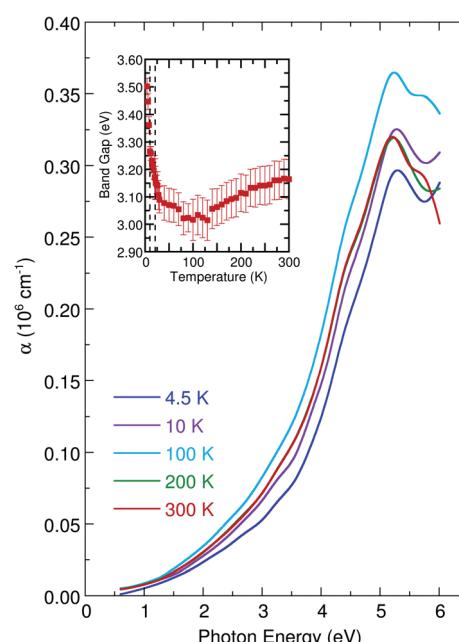


Fig. 3 Temperature-dependent optical absorption spectra of $\text{Cu}^{11}\text{B}_2\text{O}_4$. Inset illustrates the temperature-dependent band gap energy. The vertical dashed lines denote the magnetic phase transition temperatures at 10 and 21 K.



II. Experimental details

The flux method was used to grow $\text{Cu}^{11}\text{B}_2\text{O}_4$ single crystals.²⁰ The crystals with a (110) surface used in this study had approximate dimensions of $3 \times 3 \times 2 \text{ mm}^3$. For each batch, crystals were characterized using X-ray powder diffraction (see ESI Fig. 2 and 3[†]) and magnetization measurements (Fig. 1). The X-ray powder diffraction profile, the Rietveld refinement pattern, and the difference pattern for $\text{Cu}^{11}\text{B}_2\text{O}_4$ confirmed the formation of a single crystal. The lattice constants and the sample's crystallographic data, obtained from the Rietveld refinement, accorded with those reported in the literature on CuB_2O_4 .²⁰ The magnetization data revealed low-temperature magnetic phase transitions at 21 and 10 K. Above 21 K, $\text{Cu}^{11}\text{B}_2\text{O}_4$ exhibited paramagnetism. Between 21 and 10 K, the material revealed commensurate antiferromagnetism. Below 10 K, the material transitioned to incommensurate antiferromagnetism. The spectroscopic ellipsometry measurements were

performed under angles of incidence of 60° – 75° using a J. A. Woollam Co. M-2000U ellipsometer over a spectral range of 0.73–6.42 eV. For temperature-dependent measurements between 4.5 and 300 K, the ellipsometer was equipped with a Janis ST-400 ultrahigh-vacuum cryostat. Only a single angle of incidence was possible because of the 70° angle of the two cryostat windows.²¹ The micro-Raman scattering spectra were measured in a backscattering configuration using a laser with an excitation wavelength of 488 nm and a SENTERRA spectrometer with a 1024-pixel-wide charge-coupled detector. The spectral resolution using this spectrometer was typically lower than 0.5 cm^{-1} , and the laser power was lower than 4.0 mW to avoid heating effects. The polarized Raman scattering spectra were obtained in backscattering geometry with incident and scattered lights parallel to the [110] and [110] directions, denoted as Y' and \bar{Y}' , respectively. Three scattering configurations were thus obtained: $Y'(ZZ)\bar{Y}'$, $Y'(X'X)\bar{Y}'$, and $Y'(ZX')\bar{Y}'$. In this Porto notation, the direction parallel to the crystallographic

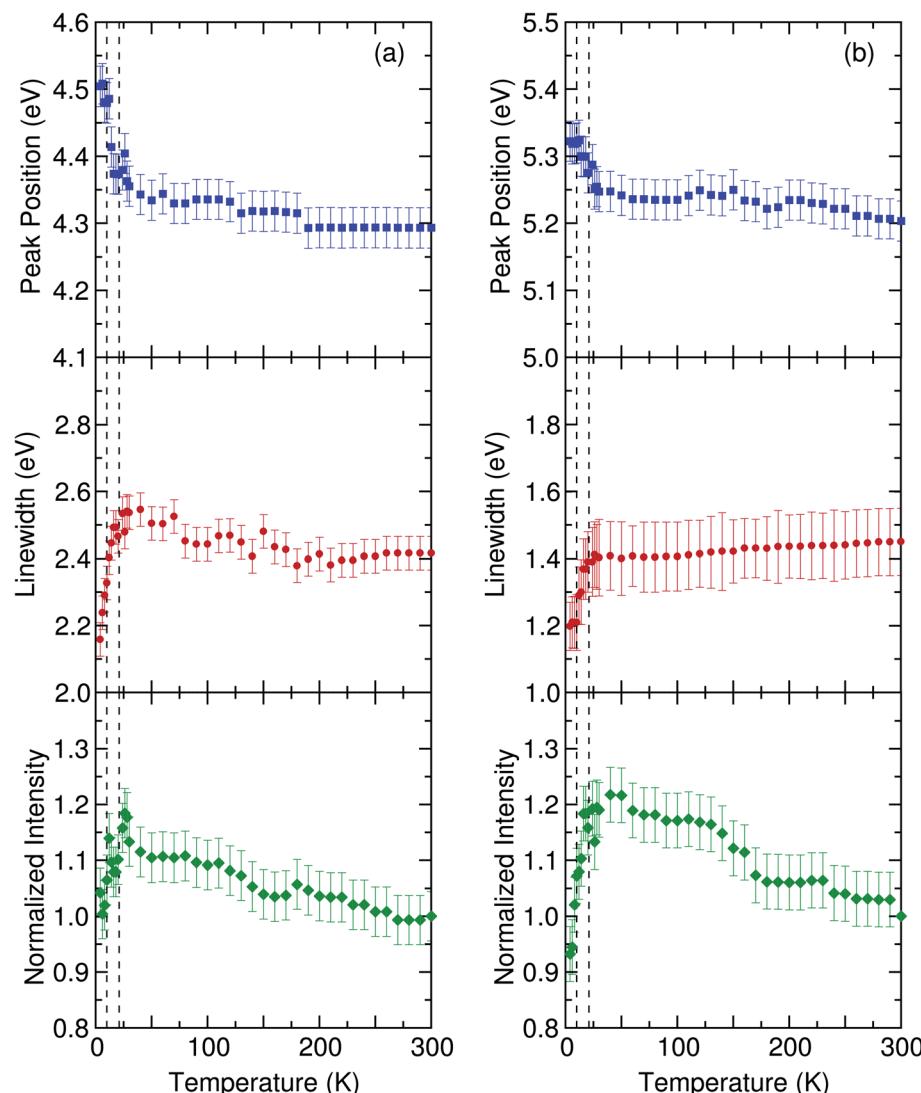


Fig. 4 Temperature dependence of the peak energy, linewidth, and normalized intensity of the (a) 4.30 and (b) 5.21 eV optical transitions. The vertical dashed lines denote the magnetic phase transition temperatures at 10 and 21 K.



(c) axis was labelled as *Z*, whereas the [110] crystal direction was labelled as *X'*. The sample was placed in a continuous-flow helium cryostat, which enabled measurements in the temperature range of 10–300 K.²¹

III. Results and discussion

A. Electronic excitations

The room-temperature optical absorption spectrum of Cu¹¹B₂O₄ is displayed in Fig. 2. The bands display different features as compared to the absorption spectrum of CuB₂O₄. We fitted the absorption spectrum using the Lorentz oscillators, as illustrated by the dashed lines. The spectrum displayed two peaks approximately at 4.30 and 5.21 eV. These peaks originated from the electronic transitions from the 2p states of the oxygen ions to the 3d states of the copper ions.^{22,23} The direct band gap E_g of Cu¹¹B₂O₄, estimated from the absorption coefficient $\alpha(E)$, was determined to be 3.16 ± 0.07 eV, as displayed in the inset of Fig. 2. The band gap of a normal solid, which holds contributions from direct and indirect transitions, can be estimated from the absorption coefficient $\alpha(E)$ as follows:²⁴

$$\alpha(E) = A(E - E_{g,\text{dir}})^{0.5} + B(E - E_{g,\text{indir}} \mp E_{\text{ph}})^2, \quad (1)$$

where $E_{g,\text{dir}}$ and $E_{g,\text{indir}}$ are the magnitudes of direct and indirect gaps, respectively; E_{ph} is the emitted (absorbed) phonon energy, and A and B are constants.

The temperature-dependent optical absorption spectra of Cu¹¹B₂O₄ are displayed in Fig. 3. As the temperature decreased, the absorption peaks displayed a gradual increase in their intensity and peak position, whereas the linewidth narrowed. The temperature-dependent band gap is displayed in the inset of Fig. 3. The band gap decreased as the temperature decreased consistent with the anomalous behavior of CuB₂O₄. We find $\frac{dE_g}{dt} = 6.41 \times 10^{-4}$ eV K⁻¹ which is a little lower compared to CuB₂O₄ but still in the same order of magnitude with those found for MAPbI₃.²⁵ Below 21 K, the band gap exhibited a gradual increase as the temperature decreased. This anomaly and discontinuity were correlated with the magnetic ordering at this temperature point. These findings demonstrate that the shifts were caused by spin-charge interactions, rather than changes induced by thermal contraction or alterations in the unit cell volume.

We examined the peak energy, linewidth, and normalized intensity of the fitted Lorentzian peaks displayed in Fig. 4. Above 21 K, the energy peak and linewidth exhibited minimal temperature variation, whereas the peak intensity increased as the temperature decreased. Anomalies near the magnetic phase transition temperature were evident in Cu¹¹B₂O₄. Below 21 K, the peak position shifted to higher energies and the linewidth decreased for two absorption peaks. These phenomena were similar to those observed in CuB₂O₄. The temperature dependences of Cu¹¹B₂O₄ and CuB₂O₄ were somewhat similar. The hardening as well as damping and intensity changes of the absorption bands below 21 K were approximately of the same magnitude in both samples. A notable isotope effect was

indicated by the shift of the absorption band toward lower energies in Cu¹¹B₂O₄ (4.30 and 5.21 eV) compared with CuB₂O₄ (4.49 and 5.83), as illustrated in Fig. 2. This isotopic shift has an opposite sign with the generally established behavior.^{9,26,27} Moreover, the extrapolated band gap was determined to be lower in Cu¹¹B₂O₄ with a difference of ~ 0.6 eV, as illustrated in the inset of Fig. 3. The replacement of a light isotope with a heavy isotope caused an increase in the energy gap in superconducting LiH and in most semiconductors – such as Si, Ge, and diamond (¹³C and ¹²C).^{7-9,26-29} Exception to this behavior has been observed in isotopic shifts in the energy gap of CuCl and other Cu salts. The dependence of the semiconductors band gap on the isotopic composition at constant temperature is governed mainly by two contributions expressed as:³⁰

$$\left(\frac{\partial E_g}{\partial M_k} \right)_T = \left(\frac{\partial E_g}{\partial M_k} \right)_{\text{EP}} + \left(\frac{\partial E_g}{\partial M_k} \right)_{\text{TE}}. \quad (2)$$

The first term is a contribution from the electron–phonon interaction (EP). The second term stems from lattice constant changes and is labeled as TE since it has similar behavior with thermal lattice expansion. Moreover, the second term can be extended in terms of the isotopic mass. The changes in isotopic mass translate to the changes in pressure within the crystal structure and a relative change in the lattice constants. We have observed the negligible changes in the X-ray powder diffraction profiles of CuB₂O₄ and Cu¹¹B₂O₄ (see ESI Fig. 2†), and thus ruled out the contributions from the second term. We speculate that the negative isotope effect in Cu¹¹B₂O₄ may arise from the changes in the electronic band structures induced by the d-electron–phonon interactions.^{9,27} The degeneracy of the d states

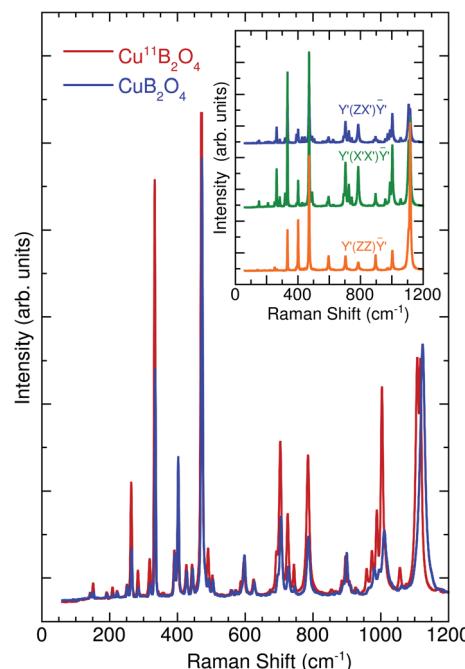


Fig. 5 Unpolarized room-temperature Raman scattering spectrum of Cu¹¹B₂O₄ and CuB₂O₄. Inset illustrates the polarized Raman scattering spectra of Cu¹¹B₂O₄.



of the Cu ions in the upper valence band was partially lifted and mixed with the p states of B ions under a crystal field. This p-d hybridization was verified in CuB_2O_4 and accounted for its magnetoelectric properties.¹⁷ Substitution of a heavier B isotope in $\text{Cu}^{11}\text{B}_2\text{O}_4$ may enhance the p-d hybridization through the electron-phonon interactions and raise the maximum of the valence band, resulting in a decrease of the band gap. Similar phenomenon has been confirmed in energy band transition studies on other Cu containing compounds such as CuCl, CuBr, and CuInS₂.^{30,31} Further theoretical calculations of the electronic band structures of $\text{Cu}^{11}\text{B}_2\text{O}_4$ and CuB_2O_4 are needed to clarify the origin of this negative isotope effect.

B. Vibrational properties

Fig. 5 displays the room-temperature unpolarized Raman scattering spectrum of $\text{Cu}^{11}\text{B}_2\text{O}_4$. The phonon peaks were fitted using a standard Lorentzian profile. The spectrum comprised

38 first-order Raman-active phonon modes, which is the same number of phonon modes observed in CuB_2O_4 . We previously identified 12A₁, 12B₂, and 14E modes in polarized Raman scattering measurements of CuB_2O_4 . The inset of Fig. 5 illustrates the polarized Raman scattering spectra of $\text{Cu}^{11}\text{B}_2\text{O}_4$ in different configurations. The peaks that appeared in the $Y'(ZZ)\overline{Y'}$, $Y'(XX')\overline{Y'}$, and $Y'(ZX')\overline{Y'}$ polarizations were identical to those observed in CuB_2O_4 in the same configurations. The phonon peaks and their assigned symmetries are summarized in Table 1. We assigned the phonon peaks of $\text{Cu}^{11}\text{B}_2\text{O}_4$ to the same atomic vibrations assigned to CuB_2O_4 . Low-frequency modes below 200 cm^{-1} were rotational vibrations of Cu ions. Peaks higher than 200 cm^{-1} but lower than 400 cm^{-1} were modes associated with changes in the bond angles that tilted and bent the BO_4 tetrahedra. These phonon peaks were coupled to the Cu–O vibrations, which were distributed over the entire frequency range. The highest vibrations from Cu–O stretching modes were observed in the range of 400–900 cm^{-1} . The Cu–O

Table 1 Raman peaks observed at room temperature for $\text{Cu}^{11}\text{B}_2\text{O}_4$

Phonon frequency (cm^{-1})	Symmetry	$\Delta\omega$ (cm^{-1}) ($\omega_{10_B} - \omega_{11_B}$)	$\omega_{11_B}/\omega_{10_B}$	Assignment
ω_1	142	E	1	
ω_2	151	B ₂	1	0.990
ω_3	192	E	2	0.993
ω_4	209	B ₂	2	0.992
ω_5	222	E	1	0.993
ω_6	250	A ₁	2	0.994
ω_7	264	B ₂	1	0.993
ω_8	284	B ₂	1	0.995
ω_9	318	B ₂	2	0.995
ω_{10}	333	A ₁	2	0.995
ω_{11}	391	E	1	0.994
ω_{12}	401	A ₁	1	0.997
ω_{13}	426	B ₂	1	0.997
ω_{14}	443	E	2	0.997
ω_{15}	471	A ₁	2	0.997
ω_{16}	491	B ₂	1	0.997
ω_{17}	505	E	1	0.998
ω_{18}	559	E	3	0.998
ω_{19}	572	E	2	0.995
ω_{20}	587	B ₂	2	0.997
ω_{21}	596	A ₁	2	0.997
ω_{22}	625	E	2	0.996
ω_{23}	677	E	3	0.997
ω_{24}	692	B ₂	4	0.995
ω_{25}	704	A ₁	2	0.997
ω_{26}	726	A ₁	2	0.997
ω_{27}	744	B ₂	3	0.996
ω_{28}	785	A ₁	2	0.998
ω_{29}	885	E	3	0.995
ω_{30}	896	A ₁	4	0.996
ω_{31}	907	E	4	0.995
ω_{32}	926	B ₂	39	0.997
ω_{33}	959	E	23	0.977
ω_{34}	974	E	21	0.979
ω_{35}	988	A ₁	23	0.977
ω_{36}	1004	B ₂	59	0.944
ω_{37}	1057	A ₁	58	0.948
ω_{38}	1107	A ₁	16	0.985



stretching modes in the range 700–900 cm^{−1} were mixed with the B–O vibrations, whereas modes above 900 cm^{−1} were considered to have occurred mainly from B–O stretching vibrations. Our experimental results of the phonon modes demonstrated good agreement with the theoretical *ab initio* calculations by Pisarev *et al.*³² The data on the isotope boron effects, summarized in Table 1, elucidates the coupling involved in the phonon vibrations occurring primarily between Cu and B.

The Raman scattering spectrum of Cu¹¹B₂O₄ was shifted to lower frequencies by at least one wavenumber compared with the Raman scattering spectrum of CuB₂O₄, as illustrated in Fig. 5. The softening of the frequencies of the heavier isotope mass was expected based on the simple harmonic model. In this case, Cu¹¹B₂O₄ contains the heavier ¹¹B isotope compared with the ¹⁰B isotope in CuB₂O₄. The simple harmonic model is expressed as follows:

$$\omega_{11_B} = \omega_{10_B} \sqrt{\frac{m_0}{m_0 + \Delta m}} \quad (3)$$

where ω_{11_B} represents the Raman frequency of the ¹¹B enriched sample, ω_{10_B} represents the frequency of the ¹⁰B sample, m_0 is the mass of the ¹⁰B, and Δm is the mass difference between the ¹¹B and ¹⁰B isotopes. Using this model, we calculated a ratio of $\frac{\omega_{11_B}}{\omega_{10_B}} \approx 0.95$ which accords with the observed redshift in the

high-frequency Raman modes (Table 1). For frequencies below 900 cm^{−1}, we obtained $\omega_{11_B}/\omega_{10_B}$ ratio of approximately 0.99. For frequencies above 900 cm^{−1}, the ratio was approximately 0.95 for ω_{36} and ω_{37} , with the largest possible shift of 59 cm^{−1}. These modes are thus confirmed to be pure B vibrations. Modes below 200 cm^{−1} exhibited the smallest shift in frequency because these peaks mostly originated from the Cu atom vibrations. The shifts in frequency, which were 200–900 cm^{−1}, exhibited differences of approximately 1–2 cm^{−1}. This finding indicates that the boron atoms influence these vibrational modes. High-frequency peaks above 900 cm^{−1} exhibited more obvious softening in the phonon frequencies shifting by at least 16 cm^{−1}. We observed a clear double peak constituted by ω_{37} (1057 cm^{−1}) and ω_{38} (1107 cm^{−1}) because of the nonuniform shift in the peaks, which contrasted with the overlapping peaks observed in CuB₂O₄. The phonon frequency ω_{38} most likely resulted from oxygen stretching vibrations. We surmise this because it is not heavily shifted with isotopic substitution, similar to other phonon modes in this range.

The temperature-dependent unpolarized Raman scattering spectra of Cu¹¹B₂O₄ are displayed in Fig. 6. For normal anharmonic solids, the phonon frequency should increase and the linewidth should decrease as the temperature decreased. The Raman scattering spectrum at 10 K was fitted with 38 Lorentz oscillators (inset of Fig. 6). The phonon modes of the ¹¹B enriched sample were redshifted. However, no significant changes in the linewidth were observed, and temperature-dependent behavior was similar to that of CuB₂O₄. An examination of the intense peak at 333 and 443 cm^{−1} is displayed in Fig. 7. Data for all other phonon modes were reported in ESI Fig. 4.‡ Above 21 K, the peak frequencies and linewidth behaved

according to the anharmonic model represented by the thin solid lines in Fig. 7. The anharmonic model is expressed as follows:³³

$$\omega(T) = \omega_0 + A \left(1 + \frac{2}{\exp\left(\frac{\Theta}{T}\right) - 1} \right) \quad (4)$$

$$\gamma(T) = \gamma_0 + B \left(1 + \frac{2}{\exp\left(\frac{\Theta}{T}\right) - 1} \right) \quad (5)$$

where ω_0 is the intrinsic frequency of the optical phonon mode, γ_0 is the linewidth broadening caused by defects, Θ is the Debye temperature, and A and B are the anharmonic coefficients. The values of the fitting parameters are summarized in Table 2. Negative values of A indicate that the phonon frequency increased as the temperature decreased, whereas positive values of B for the linewidth indicates the contrary. The Debye temperature obtained for different phonon frequencies was within 400–600 K in the same range obtained for CuB₂O₄. The anharmonic model revealed that the Raman phonon modes were mostly temperature independent at low temperature. Below the magnetic ordering temperature (21 K), deviations from the anharmonic model remained evident for all intense peaks. A gradual softening of up to 0.3 cm^{−1} was observed in Cu¹¹B₂O₄.

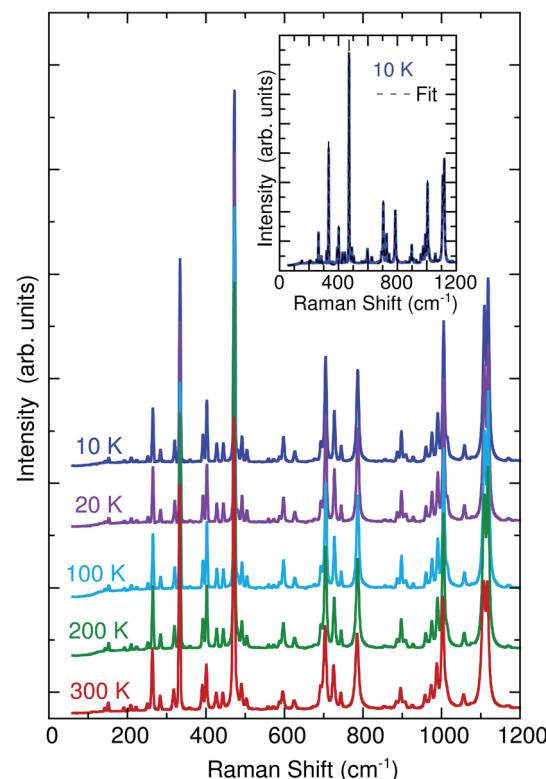


Fig. 6 Temperature-dependent unpolarized Raman scattering spectra of Cu¹¹B₂O₄. Inset illustrates the results of fitting the spectrum obtained at 10 K using the Lorentzian model.



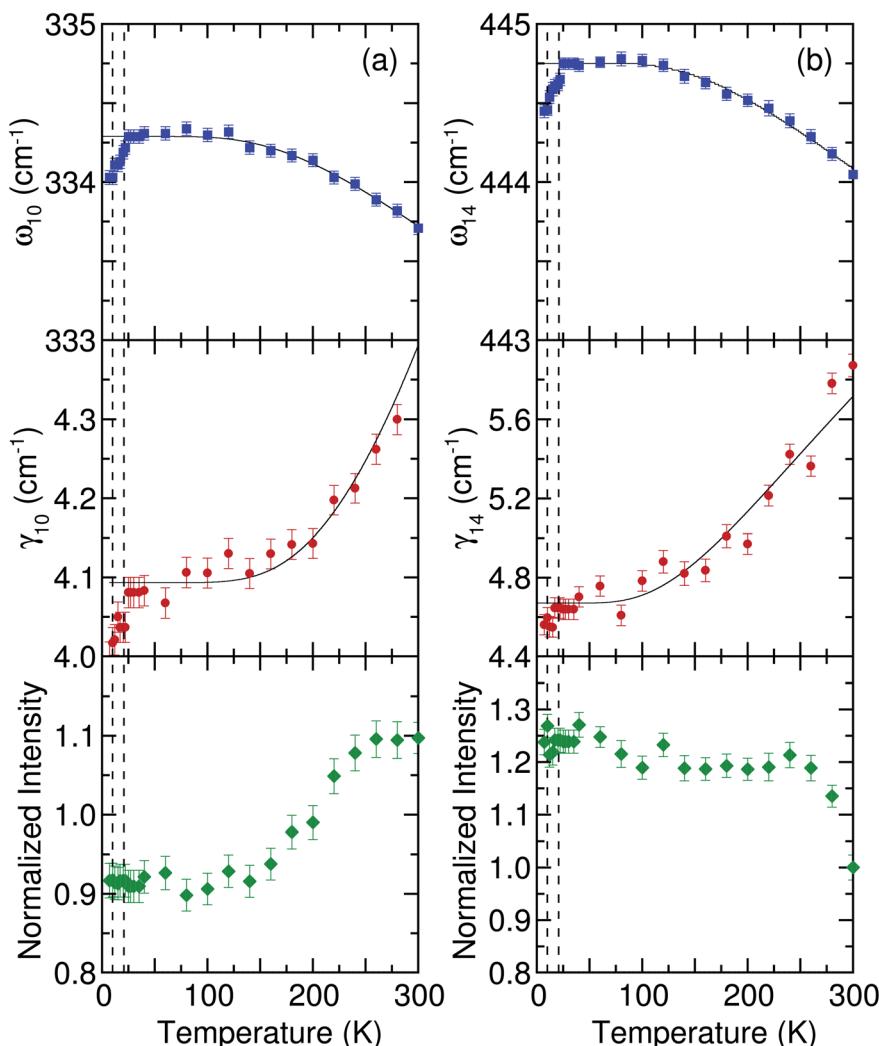


Fig. 7 Temperature dependence of the frequency, linewidth, and normalized intensity of (a) 333 cm^{-1} and (b) 443 cm^{-1} phonon modes. Thin solid lines are the fitting results obtained with the anharmonic model. The vertical dashed lines denote the magnetic phase transition temperatures at 10 and 21 K.

Softening in the phonon peaks at the magnetic phase transition temperature usually indicates the spin–phonon interactions. The sensitivity of the phonon frequency to correlations of spins of nearest-neighbor pairs can be expressed as follows:³⁴

$$\Delta\omega(T) \approx \lambda \langle S_i \times S_j \rangle \quad (6)$$

Table 2 Values of parameters obtained from fitting temperature dependence of the phonon frequencies and linewidth based on the anharmonic model expressed in eqn (4) and (5)

	$\omega_0\text{ (cm}^{-1}\text{)}$	$\gamma_0\text{ (cm}^{-1}\text{)}$	$A\text{ (cm}^{-1}\text{)}$	$B\text{ (cm}^{-1}\text{)}$	$\Theta\text{ (K)}$
ω_{10}	338	4.20	-3.06	0.89	498
ω_{11}	396	3.21	-2.49	2.20	554
ω_{12}	406	3.50	-3.68	1.61	611
ω_{14}	448	4.96	-3.41	1.02	560
ω_{15}	478	3.02	-4.02	1.87	528
ω_{21}	605	3.80	-5.98	3.79	551
ω_{25}	710	2.68	-4.24	3.84	551
ω_{28}	788	3.09	-1.96	0.56	543

where λ is the spin–phonon coupling constant and $\langle S_i \times S_j \rangle$ is the nearest neighbor spin correlation. The $\langle S_i \times S_j \rangle$ can be estimated from $4 \left(\frac{M_{\text{sub}}(T)}{M_s} \right)^2$ where $M_{\text{sub}}(T)$ is the sublattice magnetization per magnetic ion and M_s is the saturation magnetization. The factor of 4 was based on the number of nearest neighbors.³⁴ As displayed in Fig. 8, the sublattice magnetization data¹³ obtained from a prior study scaled well with the shift in phonon frequency. Using this correlation, we extracted the spin–phonon coupling constant λ for Cu–O stretching phonon modes tabulated in Table 3.

A slightly higher value of the spin–phonon coupling constant was obtained for $\text{Cu}^{11}\text{B}_2\text{O}_4$ compared with the values obtained for CuB_2O_4 . This slight difference coincides with the quantitative estimate of the spin–phonon coupling λ using a simplified

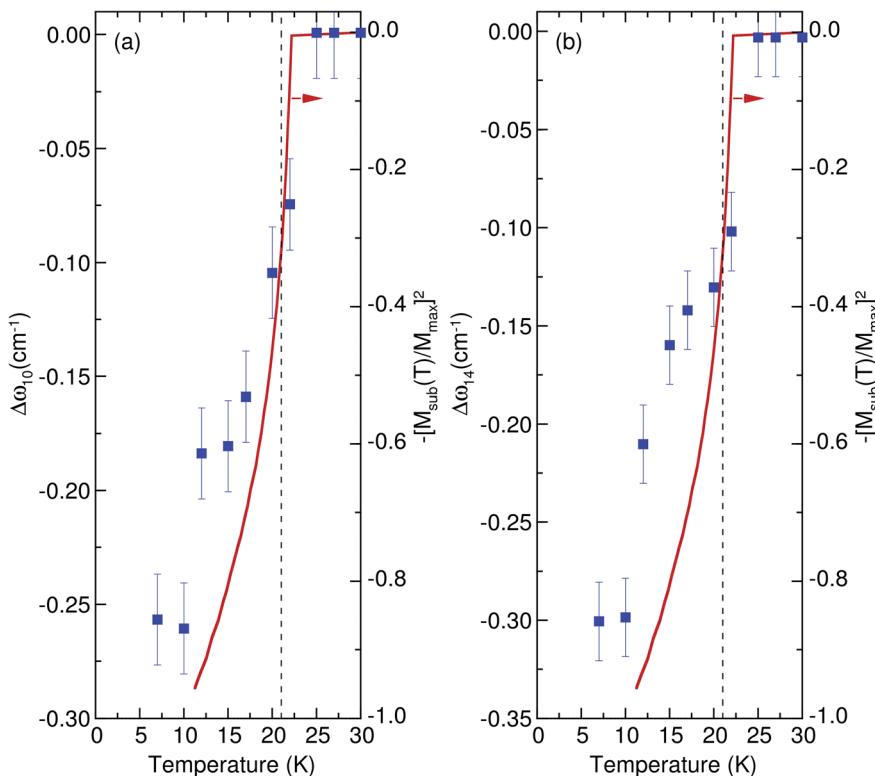


Fig. 8 Temperature dependence of the shift in the phonon frequency of (a) 333 and (b) 443 cm^{-1} modes plotted against the normalized square of the magnetic susceptibility.¹⁵ The vertical dashed line denotes the magnetic phase transition temperatures at 21 K.

lattice model proposed by Sushkov *et al.*,³⁵ which can be calculated as follows:

$$\lambda \approx \frac{2\alpha^2 J}{m\omega} \quad (7)$$

where m is the mass of the magnetic ion, ω is the mode frequency, J is the nearest neighbor exchange coupling constant, and α can be calculated using $\alpha = 2z/3a_B$ (a_B is the Bohr radius and z is the nearest neighbor coordination number). Eqn (7) is inversely related to the phonon frequencies. The phonon frequencies of $\text{Cu}^{11}\text{B}_2\text{O}_4$ were redshifted from those of CuB_2O_4 , which causes the spin-phonon coupling to increase. Considering only the nearest neighbors ($z = 4$), we

obtained $\alpha = 5.04 \text{ \AA}^{-1}$. Moreover, we estimated the exchange coupling constant using³⁵

$$J = 3k_B\Theta_{\text{cw}}/zS(S+1) \quad (8)$$

where k_B is the Boltzmann constant, Θ_{cw} is the Currie-Weiss temperature, and S is the spin angular moment. We used $z = 4$, $S = 1/2$, and $\Theta_{\text{cw}} = 21 \text{ K}$ to get $J = 0.95 \text{ meV}$.¹³ Therefore, we obtained a quantitative spin-phonon coupling constant $\lambda \approx 0.037 \text{ cm}^{-1}$ which is slightly higher than the values obtained for CuB_2O_4 ($\lambda \approx 0.035 \text{ cm}^{-1}$) and scales well with the empirical spin-phonon coupling λ tabulated in Table 3.

IV. Summary

We used spectroscopic ellipsometry and Raman scattering spectroscopy to study the electronic structure and lattice dynamics of $\text{Cu}^{11}\text{B}_2\text{O}_4$ single crystals. We assessed the boron isotope effects by comparing the results obtained with the results for CuB_2O_4 . The room-temperature optical absorption spectrum of $\text{Cu}^{11}\text{B}_2\text{O}_4$ revealed the occurrence of charge-transfer transitions from the 2p states of O to the 3d states of Cu at approximately 4.30 and 5.21 eV, which were lower than the values measured for CuB_2O_4 . The direct optical band gap was determined to lower in $\text{Cu}^{11}\text{B}_2\text{O}_4$ (extrapolated at $3.16 \pm 0.07 \text{ eV}$). The temperature-dependent band gap and the peak energy of charge-transfer bands exhibited anomalies through the canted antiferromagnetic ordering temperature at 21 K. The

Table 3 Spin-phonon coupling constant of $\text{Cu}^{11}\text{B}_2\text{O}_4$ obtained for Cu–O stretching related modes

Mode frequency (cm^{-1})	λ (cm^{-1})
ω_{10}	333
ω_{11}	391
ω_{12}	401
ω_{14}	443
ω_{15}	471
ω_{21}	596
ω_{25}	704
ω_{28}	785
ω_{30}	896



Raman frequencies of $\text{Cu}^{11}\text{B}_2\text{O}_4$ shifted lower, whereas the linewidth exhibited no significant change. The extent of the isotopic shift in frequencies accorded with the inverse square root dependence of the mean atomic mass. The temperature dependence of the phonon frequency, linewidth, and normalized intensity exhibited anomalies near 21 K. The temperature dependence of the Cu–O stretching modes also exhibited an anomalous softening at a temperature of less than 21 K. The spin–phonon coupling constant, λ , were slightly higher than CuB_2O_4 , estimated to be 0.030–0.038 cm^{-1} .

Author contributions

H. L. L. conceived the research and was responsible for the experimental design. R. D. M. conducted the experiments. C. H. L. and C. H. D. were responsible for the sample preparation. R. D. M. and H. L. L. drafted the paper. All contributing authors have discussed the results and provided the comments regarding the manuscript.

Data availability statement

The data that support the findings of this study are available from the corresponding author upon reasonable request.

Conflicts of interest

The authors declare no competing interests.

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