Highly effective organic light-emitting diodes containing thermally activated delayed fluorescence emitters with horizontal molecular orientation†

Chan Hee Lee, a Shin Hyung Choi, a Sung Joon Oh, a Jun Hyeon Lee, a Jae Won Shim, b,c Chihaya Adachi c,d and Sae Youn Lee e,a

In this study, we report new thermally activated delayed fluorescence (TADF) emitters, AcPYM (10,10’-(pyrimidine-2,5-diylbis(4,1-phenylene))bis(9,9-dimethyl-9,10-dihydroacridine)) and PxPYM (10,10’-(pyrimidine-2,5-diylbis(4,1-phenylene))bis(10H-phenoxazine)), by employing donor units at the 2,5-positions of the pyrimidine acceptor unit. The donor–acceptor–donor (D–A–D) units combined in the linear molecular structure of AcPYM or PxPYM enhanced the horizontally oriented alignment, and the horizontal transition dipole moments were realized by up to 87% in the host matrix. Organic light-emitting diodes (OLEDs) containing AcPYM and PxPYM emitters realized external quantum efficiencies ($\eta_{\text{ext}}$) of 16.8% for blue and green emissions.

Introduction

After the first research into the organic light-emitting diodes (OLEDs) was reported in 1987, significant improvements in OLEDs regarding efficiency, color purity, and lifetime have been achieved, leading to them becoming the next-generation light source for lighting and flexible display applications. Over the past several decades, various studies into organic emitters based on fluorescence and phosphorescence in the visible-light region have been conducted to improve the internal quantum efficiency ($\eta_{\text{int}}$). Even though fluorescence emitters have outstanding reliability and stability, low exciton production efficiency ($\eta_{\text{pro}}$) of fluorescence in electrical excitation limits the theoretical $\eta_{\text{out}}$ to 25%. Phosphorescence emitters can achieve a theoretical $\eta_{\text{out}}$ of nearly 100% from singlet ($S_1$) and triplet ($T_1$) exciton harvesting through intersystem crossing (ISC) between the $S_1$ and $T_1$ states using heavy transition metals. However, phosphorescence emitters suffer from a strong exciton annihilation process and a drastic efficiency decrease under the high current density derived from the high density of $T_1$ excitons originating from their long radiative decay time ($\tau_1$; ms–ns).

Recently, a thermally activated delayed fluorescence (TADF) emitter that can substitute for fluorescence and phosphorescence emitters in OLEDs due to its effective $S_1$ and $T_1$ excitons harvesting by reverse ISC (RISC) from the $T_1$ to the $S_1$ state has been reported. The up-conversion through RISC can only arise with a small $S_1$–$T_1$ state energy gap ($\Delta E_{\text{ST}} < 0.3$ eV) in TADF emitters, which is attained by providing less overlapping between the highest occupied molecular orbital (HOMO) and the lowest unoccupied molecular orbital (LUMO) levels. Furthermore, numerous studies have focused on molecular designs utilizing twisted molecular structures among the donor and acceptor units for reducing $\Delta E_{\text{ST}}$ and achieving efficient intramolecular charge transfer (ICT) based on pure organic derivatives. Meanwhile, despite the development of various TADF emitters with $\eta_{\text{out}}$ of nearly 100%, conventional OLEDs have a 20–25% external quantum efficiency ($\eta_{\text{ext}}$) limitation because of low light out-coupling efficiency ($\eta_{\text{out}}$). External and internal optical modification of OLEDs such as attaching a microlens array, optimizing the layer thickness, and refractive index control of the internal layers have been widely studied to increase $\eta_{\text{out}}$. Another promising approach for enhancing $\eta_{\text{out}}$ is to introduce horizontally oriented emitters. Previous research by Yokoyama and co-workers demonstrated an increase of $\eta_{\text{out}}$ in OLEDs by applying horizontally aligned emitters having horizontal transition dipole moment orientation on the substrate. Furthermore, the orientation degree of the emitters can be controlled by the shape of the molecules: rod- or disk-like structures prefer high degrees of horizontal transition dipole moments for perpendicular light emission.
In this study, we developed two pyrimidine-based TADF emitters: 10,10′-(pyrimidine-2,5-diylbis(4,1-phenylene))bis(9,9-dimethyl-9,10-dihydroacridine) (AcPYM) and 10,10′-(pyrimidine-2,5-diylbis(4,1-phenylene))bis[10H-phenoxazine] (PxPYM) and adopted donor-acceptor-donor (D–A–D) combined linear molecular structures based on a combination of donor units of 9,9-dimethylacridin (Ac) or phenoxazine (Px) and one of the pyrimidine acceptor units. The donor units were introduced at the 2,5-position of the pyrimidine unit to compose a linear molecular structures based on a combination of donor units of 1,3,2-dioxaborolan-2-yl)-10-(4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl)-9,10-dihydroacridine (1) and 10-(4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl)-10H-phenoxazine (2), were synthesized as reported previously.42

Experimental

Materials

All reagents and solvents were prepared from Alfa Aesar, Sigma-Aldrich, and Tokyo Chemical Industry. Two intermediates, 9,9-dimethyl-10-(4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl)-9,10-dihydroacridine (1) and 10-(4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)phenyl)-10H-phenoxazine (2), were synthesized as reported previously.42

Synthesis of AcPYM. 5-Bromo-2-chloropyrimidine (0.50 g, 2.58 mmol) and 1 (2.34 g, 5.69 mmol) were dissolved in 1,4-dioxane (30 mL) under an N₂ atmosphere. Next, sodium carbonate (2.4 M, 15 mL) and tetrakis(triphenylphosphine)-palladium(0) (0.24 g, 0.21 mmol) were added and the mixture refluxed for 12 h. After cooling down to room temperature, the solution was poured into chloroform and distilled water for extraction. The chloroform layer was washed with distilled water several times and dried over magnesium sulfate (MgSO₄). The crude product was filtered by using Celite 545 and purified via column chromatography on silica gel (eluent: dichloromethane/hexane, 4:1) to give a yellow powder (yield ¼ 89.5%). The product was dried under vacuum and analytically determined by ¹H NMR (500 MHz, CDCl₃) data as follows: 7.90 (d, J = 8.0 Hz, 2H), 7.56 (d, J = 8.5 Hz, 2H), 7.53 (d, J = 8.5 Hz, 2H), 6.74–6.60 (m, 12H), 6.05 (dd, J = 6.5, 1.5 Hz, 2H), 6.02 (dd, J = 6.0, 1.5 Hz, 2H). ¹³C NMR (500 MHz, CDCl₃) data are as follows: 206.91, 155.40, 144.03, 134.10, 132.10, 131.14, 130.97, 129.47, 123.30, 121.69, 121.53, 115.67, 115.56, 113.38, 113.28, 110.73, 110.41, 31.24, 30.92; Anal. calcd (%) for C₆₀H₃₅N₄: C 80.79; H 9.42; N 4.41; found: C 80.91; H 9.47; N 4.26.

Results and discussion

Synthesis and thermal properties

AcPYM and PxPYM were synthesized through Suzuki coupling reactions with either Ac or Px, respectively, with the bromine-substituted pyrimidine (Scheme 1).42 The chemical structures of the newly synthesized emitters were analyzed using ¹H and ¹³C NMR spectroscopy, MALDI-TOF mass spectrometry, and elemental analysis. In addition, Differential Scanning Calorimetry (DSC) and thermogravimetric analysis (TGA) was performed to analyze the thermal properties of the emitters (Fig. S4 and S5†); it was observed that the decomposition temperatures (T_d) of AcPYM and PxPYM were 435 and 463 °C, respectively.

Density functional theory (DFT) calculations

The optimized molecular structures, excited states energy levels, and frontier molecular orbital contributions of AcPYM and PxPYM were calculated via a time-dependent DFT (TD-DFT) at the B3LYP/6-31G(d,p) level (Fig. 1). These emitters have dihedral angles between the phenyl linkers and the Ac or Px units (θ_d,p) of 89.5–89.8° and 87.8–89.3° for AcPYM and PxPYM, respectively. Moreover, the LUMO and HOMO levels of AcPYM and PxPYM are localized on the pyrimidine acceptor and either the Ac or Px donor units, respectively. This good separation of the frontier molecular orbital distributions of AcPYM and PxPYM originating from the highly sterically hindered structure contributed to the relatively small calculated ΔE_ST of 0.01 eV.

Photophysical properties

Fig. 2 displays spectra of the ultraviolet-visible (UV-Vis) absorbance and the PL of the emitters in toluene. The broad 375 and 394 nm-centered absorption peaks and the 477 and 532 nm-centered blue and green PL emission peaks (λPL) of AcPYM

![Scheme 1 Synthesis of AcPYM and PxPYM.](Image)
and PxPYM were observed in the UV-Vis and PL spectra under photoexcitation of oxygen-free toluene, respectively. Furthermore, 6 wt% AcPYM and PxPYM doped in 2,8-bis(diphenylphosphoryl)-dibenzo[b,d]thiophene (PPT) host thin films were prepared to confirm the photophysical properties of the emitters in the solid-state. Because of the higher T1 energy level of PPT (\(E_T = 3.1\) eV) compared with AcPYM and PxPYM (\(E_T = 2.65\) and 2.58 eV, Fig. S7†), the PPT host was chosen to contain the reverse energy transfer between the emitter and the host.\textsuperscript{17,19,43}

As shown in Fig. 2, the 6 wt% AcPYM:PPT-doped film exhibited blue emission with 478 nm \(\lambda_{PL}\), while the 6 wt% PxPYM:PPT-doped film emitted green light at around 537 nm.

The PL quantum yields (PLQY, \(\Phi_{PL}\)) of the 6 wt% AcPYM- and PxPYM-doped films in PPT hosts were measured using an integrating sphere, with which \(\Phi_{PL}\) levels of 53.3% and 78.0% were observed. We also examined temperature-dependent transient PL decays to identify the TADF characteristics of the emitters’ temperature range from 100 to 300 K using a streak camera (Fig. 3). The delayed component PL intensity increased while the temperature raised from 100 to 300 K, thereby demonstrating that the thermal energy at high temperature accelerated the RISC process.\textsuperscript{31} Furthermore, 19 and 17 ns of the prompt fluorescence lifetimes (\(\tau_p\)) and 486 and 287 \(\mu\)s of the delayed fluorescence lifetimes (\(\tau_d\)) were observed in the double-exponential function fitted transient PL decay curves of AcPYM and PxPYM, respectively. \(\Phi_{PL}\) and the ISC and RISC rate constants are summarized in Table S3.\textsuperscript{44}
The horizontal alignment of the emitters was confirmed via angle-dependent PL measurements (Fig. 4). To verify the molecular orientation of the AcPYM and PxPYM, the angle dependence of the PL intensity of emitters in the host matrix was estimated to fit the experimental data. Moreover, the horizontal molecular orientation degree was estimated using anisotropy factor $\Theta$ calculated from the perpendicularly oriented fraction to the total amount of transition dipole moments:16,39,45

$$\Theta = [p_x]/([p_y] + [p_x] + [p_z])$$

where $[p_x]$ is vertical, and $[p_y]$ and $[p_z]$ are horizontal transition dipole moments. A $\Theta$ value of 1/3 means complete isotropy $([p_x] = [p_y] = [p_z] = 1)$, and the value of zero means perfectly horizontal $([p_x] = [p_z] = 1$, $[p_y] = 0)$. From the experimental results, $\Theta$ was equal to 0.13 and 0.18 for AcPYM and PxPYM:PPT-doped films, respectively, demonstrating that the AcPYM:PPT-doped film has 87% horizontal transition dipole moments and the PxPYM:PPT-doped film has 83%. Moreover, the values are higher than those of previously reported emitters having random molecular orientation, indicating that the devices developed using AcPYM and PxPYM emitters showed higher $\eta_{ext}$ values.

Electroluminescence (EL) performance

The EL characteristics of TADF-OLEDs using AcPYM and PxPYM emitters in devices of indium tin oxide (ITO)/1,4,5,7,8,11-hexaazatriphenylene-hexacarbonitrile (HATCN, 10 nm)/4,4′-bis[N-(1-naphthyl)-N-phenylamino]-1,1′-biphenyl (α-NPD, 30 nm)/1,3-bis(9-carbazolyl)benzene (mCP, 5 nm)/6 wt% emitter: PPT (30 nm)/PPT (5 nm)/1,3,5-tris(N-phenylbenzimidazol-2-yl) benzene (TPBi, 40 nm)/lithium fluoride (LiF, 0.8 nm)/aluminum (Al, 80 nm) (Fig. 5(a)). To prevent triplet exciton quenching from emitting layer (EML) to hole- and electron-transporting layer in these devices, 5 nm thin layers of mCP and PPT having high T1 energy levels (2.9 and 3.1 eV, respectively) were added as the neighboring layers of the EML.

The current density–voltage–luminance ($J$–$V$–$L$) properties are displayed in Fig. 5(b). The TADF-OLEDs employing AcPYM and PxPYM as emitters exhibited blue and green EL emission peaks ($\lambda_{EL}$) at 487 and 524 nm, respectively (Fig. 5(c)). In addition, below a turn-on voltage of 3.6 V for all devices exhibited maximum luminance ($L_{max}$) values of 20 370 cd m$^{-2}$ for AcPYM and 19 390 cd m$^{-2}$ for PxPYM. The AcPYM-based TADF-OLEDs achieved $\eta_{ext}$ of 11.9%, current efficiency ($\eta_{I}$) of 27.0 cd A$^{-1}$, and power efficiency ($\eta_{P}$) of 25.0 lm W$^{-1}$, while the PxPYM-based OLEDs achieved 16.8%, 52.2 cd A$^{-1}$, and 41.0 lm W$^{-1}$, respectively (Table 1).

Conclusions

We designed and synthesized linear D–A–D-structured TADF emitters AcPYM and PxPYM with Ac or Px donor units,

<table>
<thead>
<tr>
<th>Emitter</th>
<th>$\lambda_{EL}$ (nm)</th>
<th>$V_{turn-on}$ (V)</th>
<th>$L_{max}$ (cd m$^{-2}$)</th>
<th>$\eta_{I}$ (cd A$^{-1}$)</th>
<th>$\eta_{P}$ (lm W$^{-1}$)</th>
<th>$\eta_{ext}$ (%)</th>
<th>CIE (x,y)</th>
</tr>
</thead>
<tbody>
<tr>
<td>AcPYM</td>
<td>487</td>
<td>3.4</td>
<td>20 370</td>
<td>27.0</td>
<td>25.0</td>
<td>11.9 [3.4 V]</td>
<td>(0.17, 0.32)</td>
</tr>
<tr>
<td>PxPYM</td>
<td>524</td>
<td>3.6</td>
<td>19 390</td>
<td>52.2</td>
<td>41.0</td>
<td>16.8 [4.0 V]</td>
<td>(0.29, 0.51)</td>
</tr>
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* The peak wavelength of EL emission. $^b$ Turn-on voltage at 1 cd m$^{-2}$. $^c$ maximum luminance. $^d$ Maximum current efficiency. $^e$ Maximum power efficiency. $^f$ External quantum efficiency. $^g$ Commission Internationale de l’Eclairage (CIE) color coordination.
respectively, at the 2,5-positions of the pyrimidine acceptor unit. These newly designed TADF emitters realized a small $\Delta E_{ST}$, which accelerated the RISC and increased the exciton harvesting efficiency. Moreover, the molecular orientation of the emitters was estimated by angle-dependent PL measurements, indicating that both AcPyM and PyPyM had high rates of horizontal molecular alignment (87% and 83%, respectively). The TADF-OLEDs fabricated using either AcPyM or PyPyM as the emitter exhibited $\eta_{ext}$ values of 11.9% and 16.8%, respectively, both of which exceed the theoretical $\eta_{ext}$ evaluated for a random distribution of emitting dipoles. We also believe that our approach clearly presents a move toward horizontally oriented TADF emitters that will inspire the realization of highly effective TADF-OLEDs in the future.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This research was supported by the Dongguk University Research Fund of 2017 for S. Y. Lee. C. Adachi acknowledges support from the Japan Science and Technology Agency (JST) and ERATO, the Adachi Molecular Exciton Engineering Project, under JST ERATO (grant number: JPMJER1305), Japan, and the International Institute for Carbon Neutral Energy Research (WPI-I2CNER) funded by the Ministry of Education, Culture, Sports, Science & Technology (MEXT).

Notes and references