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Regio- and stereoselective thiocyanatothiolation of alkynes and alkenes by using NH_4SCN and N-thiosuccinimides†

Liang Qi,^a Shiwen Liu^b and Linxia Xiao (1)**

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A highly regioselective thiocyanatothiolation of alkynes and alkenes assisted by hydrogen bonding under simple and mild conditions is developed. Our thiocyanatothiolation reagents are readily available ammonium thiocyanate and *N*-thiosuccinimides. This metal-free system offers good chemical yields for a wide range of alkyne and alkene substrates with good functional group tolerance.

Previous works

Sulfur-containing molecules are ubiquitous structural motifs and widely exist in natural products, 1,2 pharmaceuticals 3,4 and agrochemicals.5-7 Examples include the nonsteroidal antiinflammatory drug Sulindac,8 the basal-cell carcinoma treatment drug Vismodegib,9 and drugs for the treatment of Parkinson's disease.10 Therefore, efficient introduction of sulfur into organic molecules has drawn much attention.11-15 And numerous approaches for the formation of C-S bonds have been developed. 16-20 The most used organosulfur sources for the formation of C-S bonds are thiols and thiophenols, which have an unpleasant smell. Recently, inorganic metal sulfides have been extensively used to construct C-S bonds, such as sodium metabisulfite,21 K2S,22 Na2S23 and Na2S2O3.24 Compared to thiols and thiophenols, inorganic metal sulfides are cheaper and generally stable. Thus, introduction of sulfur-containing groups into molecules by using inorganic metal sulfides is one of the desired approaches. Among them, thiocyanates commonly serve as important precursors for the preparation of thioethers,25 trifluoromethyl sulfides,26 heteroaromatic compounds.27 In general, the sources of SCN used to introduce a sulfur-containing group into molecules are thiocyanate salts²⁸⁻³⁵ such as KSCN, NaSCN, AgSCN and NH₄SCN. For example, thiocyanate salts were employed in thiocyanation of bromoalkenes via photocatalysis (Scheme 1a).36 Besides, the vinyl thiocyanates could be also obtained by thiocyanation of haloalkynes (Scheme 1b),37 iodothiocyanation of alkynes (Scheme 1c).38 Obviously, difunctionalization of alkynes is the most straightforward protocol to prepare vinyl thiocyanates.

Recently, our group has focused on hydrogen-bonding network or cluster³⁹ assisted transformations such as hydrofluorination of ynamides⁴⁰ and alkenes,⁴¹ the addition of sulfonic acids to haloalkynes,⁴² fluorothiolation of alkenes,²⁰ dihalogenation of alkynes⁴³ and hydrochlorination of alkynes,⁴⁴⁻⁴⁶halothiolation of alkynes.⁴⁷ Along this line, herein, we are glad to report a hydrogen bond network-enabled regio- and stereoselective thiocyanatothiolation of alkynes using NH₄SCN and *N*-thiosuccinimides.

Initially, according to the previous report, 20 we started the investigation of thiocyanatothiolation protocol using NH₄SCN and N-(phenylthio)succinimide as thiolation reagents in DCM under air and carried out the reaction at 60 $^{\circ}$ C (Table 1). To our delight, the desired product 3a was obtained in 42% yield

Scheme 1 Methods for thiocyanatothiolation of alkynes and alkenes.

J. Ora. Chem., 1990, 55, 1801 AgOAc 60 °C Org. Chem. Front., 2020, 7, 119 This work NH₄SCN HFIP, 60 °C 1) catalyst-free R1: aryl, alkyl; 2) highly regioselective 3) good functional group tolerace R2: H, aryl, alkyl, halogens; 4) mild reaction conditions R3: aryl, alkyl 5) introduction of two sulfurcontaining groups

[&]quot;Jiangsu Vocational College of Medicine, Jie Fang South Road 283 th, Yancheng, 224000, China. E-mail: xiaolinxiaxlx@126.com

^bCollege of Textiles and Clothing, Yancheng Institute of Technology, Yancheng, 224051, China

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Table 1 Optimization for the reaction conditions

$Entry^a$	[SCN]	Solvent	Temp. (°C)	$Yield^{b}$ (%)
1	NH₄SCN	DCM	60	42
2	NH ₄ SCN	DCE	60	47
3	NH ₄ SCN	THF	60	0
4	NH ₄ SCN	Acetone	60	0
5	NH ₄ SCN	DMF	60	0
6	NH ₄ SCN	iPrOH	60	0
7	NH ₄ SCN	AcOH	60	24
8	NH ₄ SCN	TFE	60	18
9	NH ₄ SCN	HFIP	60	87
10	LiSCN	HFIP	60	36
11	NaSCN	HFIP	60	42
12	KSCN	HFIP	60	49
13	NH_4SCN	HFIP	25	63
14	$\mathrm{NH_{4}SCN}$	HFIP	80	83

 $[^]a$ Reaction conditions: 1 (0.1 mmol), 2 (0.12 mmol), NH₄SCN (0.2 mmol), solvent (0.5 mL), under air for 12 h at 60 $^{\circ}$ C. b Determined by GC.

without any isomers found in the reaction mixture detected by GC-MS (Table 1, entry 1). Screening of solvents indicated that this transformation could not proceed in the polar solvents, such as acetone, THF, dioxane, i-PrOH, DMF (Table 1, entries 3-6) probably due to the solvation of electrophiles while moderate yield could be obtained in non-polar solvent (Table 1, entry 2). Strong hydrogen-bond donor solvents such as hexafluoro-2propanol (HFIP), could form an H-bond network activating the electrophiles through a strong hydrogen bonding interaction.48 In order to enhance the H-bond interaction between the hydroxyl and 2, so AcOH was chosen to compare with HFIP (Table 1, entry 7). Along this line, hydrogen-bond donor solvents were used and further optimization of hydrogen-bond donor solvents indicated that HFIP was superior to AcOH and trifluoroethanol (Table 1, entries 7-9). Moreover, a screening of thiocyanate salts showed that NH₄SCN was the best SCN source for this transformation compared with lithium thiocyanate, sodium thiocyanate and potassium thiocyanate (Table 1, entries 10–12). Additionally, decreasing the temperature from 60 °C to room temperature resulted in a lower yield (Table 1, entry 13) and the reaction yield was not improved significantly by raising the temperature from 60 °C to 80 °C (Table 1, entry 14).

With the optimized conditions in hand, we next turned our attention to explore the substrate scope (Table 2). Firstly, *N-(p-methoxyphenylthio)*succinimide was used as electrophile to explore the scope of alkynes. In general, the reaction proceeded well to provide the desired products 3 in moderate to excellent yields with satisfactory regio- and stereoselectivity. Diverse aryl alkynes containing electron-donating groups such as isopropyl, hydroxy, methoxy, hydroxyethyl, *tert*-butyl and trifluoromethoxy groups (Table 2, 3e-3g and 3m-3p) at the *ortho*, *meta*, or *para* positions of aryl rings all reacted with *N-*thiosuccinimides to

Table 2 Scope for thiocyanatothiolation of alkynes and N-arylsulfenylsuccinimides a,b

R ¹ 1 N-SAr +	NH ₄ SCN HFIP 60 °C, air 3	SAr SAn:			
Et SCN	SCN	SCN OH			
I I	JON	SCN OH			
SPh	SAn	SAn			
30 839/	R B	r V			
3a , 82%		3w, 72%			
R SCN	3k , R = F, 83%	ŞCN			
i joit	3I, R = CI, 85% c				
	3m ,R = <i>t</i> Bu, 80%	1441			
SAn	3n, R = Ph, 70%	SAn			
~	3o, R = OCF ₃ , 88%	3x, 61%			
3b , R = CI, 73%	3p, R = OMe, 77%	2011			
3c, R = <i>i</i> Pr, 77%	3q, R = Ac, 65%	SCN			
3d, R = OMe, 69%	34, K = AC, 05%	Ph			
3e, R = CH ₂ OH, 83%	《 】	SAn			
SCN	SCN N Ts	3y, 62%			
R. A.	∧ R I³	oy, 0270			
1 4 9					
SAn	SAn	OSCN			
05 D - 014- 700/	3r, R = Pr, 76%				
3f, R = OMe,79%	3s, R = Ph, 64%	Ans			
3g,R = OH, 62%	3t, R = CI, 52%				
3h,R = CN, 51%					
3i, R = CO ₂ Me 60% c	3u, R = Br, 60%	3z, 68%			
3j, R = CI, 71%	3v, R = I, 46% °	0			
0,					
)— NCS					
	An An				
	All s	O AnS			
322	, 74% 3a	b, 60%			
, 0 — 3aa	, 74% 3a	D, 60%			
SCN	SCN	SCN			
. I	~ I				
s s	Ś	S			
Br	R I ∥ ∃R	OMe			
	OMe	3al, 71%			
3ac, R = H, 81%		SCN			
3ad, R = 2-F, 88%	3ag, R = 3-Br, 68%	SCN			
3ae, R = 4-F, 90%	3ah, R = 4-Cl, 70%				
3af, R = 4-Me, 79%	3ai, R = 4-I, 57%	S.			
3aj , R = 4-Ac, 50%					
	$3ak$, R = $4-NO_2$, 42%	OMe 3am, 76%			
ŞCN	ŞCN	SCN			
\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	\^\\	SCN			
SAn	ÓH ŚPh	SPh			
3an, 66%, (1:1) ^d	3ao, 60%, (1:1) ^d	3ap, 71%			

 $[^]a$ Reaction conditions: 1 (0.1 mmol), 2 (0.12 mmol), NH₄SCN (0.2 mmol), HFIP (0.5 mL), under air for 12 h at 60 °C. b Isolated yield. c Ar = Ph. d Determined by NMR.

give the corresponding adducts in moderate to excellent yields. Besides, halide substitutes (F, Cl, Br) (Table 2, **3b**, **3j**–**3l** and **3w**) and electron-withdrawing groups such as cyano and ester (Table 2, **3h** and **3i**) on phenyl ring were well tolerated. Furthermore, asymmetric or symmetrical internal alkynes also could be transformed into vinyl thiocyanates (Table 2, **3r**, **3s** and **3w**) without any isomers. Remarkably, vinyl thiocyanates containing halogens could be obtained by using haloalkynes (Table 2, **3t**–**3v**). Additionally, slightly low yields were observed for fused aromatic such as naphthalene and heterocyclic aromatic (Table 2, **3x** and **3y**). Due to good functional-group tolerance, derivatives of diacetone-p-glucose (Table 2, **3z**), natural products L-menthol (Table 2, **3aa**) and pharmaceuticals such as zaltoprofen (Table 2, **3ab**) also worked well.

Next, we started to explore the scope of *N*-arylsulfenylsuccinimides. Various *N*-arylsulfenylsuccinimides can be obtained easily by the method in ESI.† To our delight, the introduction of electron-donating groups or halide substitutes to the phenyl ring of *N*-arylsulfenylsuccinimides had little influence on this

Table 3 Scope for thiocyanatothiolation of alkenes^{a,b}

 a Reaction conditions: 4 (0.1 mmol), 2 (0.12 mmol), NH₄SCN (0.2 mmol), DCE (1.0 mL), under air for 12 h at 60 $^{\circ}$ C. b Isolated yield.

reaction, providing the corresponding products in 57–90% yields (Table 2, 3ac-3ak) while electron-withdrawing groups on the phenyl ring such as acetyl or nitro resulted in lower yields (Table 2, 3aj and 3ak) probably due to the decrease of electrophilicity of N-arylsulfenylsuccinimides. Notably, the scope of N-sulfenylsuccinimides could be extended to N-alkylsulfenylsuccinimides (Table 2, 3al and 3am), affording the desired products with good yields and high selectivity. Unfortunately, the thiocyanatothiolated products (Table 2, 3an and 3ao) with poor stereoselectivity (Z/E = 1:1) were obtained when the unsymmetrically aliphatic alkynes were employed. We speculated that the steric hindrance of the aliphatic side chain maybe is small, resulting in a low Z/E ratios.

Encouraged by the success of thiocyanatothiolation of alkynes, we next turned our focus to the thiocyanatothiolation of alkenes. Under the optimized conditions, no product was observed. And the HFIP as nucleophile replaced the $\mathrm{NH_4SCN}$, giving hexafluoroisopropanol thiolated product. As result, nonpolar solvent DCE was used to avoid the hexafluoroisopropanol thiolat of alkenes. To our delight, the thiocyanatothiolation of alkenes could proceed smoothly though moderate or lower chemical yields were obtained. Among them, aromatic alkenes

Scheme 2 Gram-scale preparation of 3aq

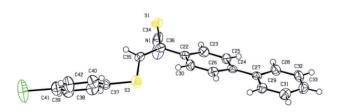


Fig. 1 Single crystal structure of 3ag.

Scheme 3 Plausible mechanism

gave moderate yields without any isomers (Table 3, 5a-5e) and aliphatic alkenes gave lower yields (Table 3, 5f-5h).

To demonstrate the scalability of this protocol, a gram-scale reaction of 1,1′-biphenyl-4-ethynyl (6 mmol) with *N*-(4-bromo thio)succinimide was carried out, and the corresponding product 3aq was obtained in 62% yield (Scheme 2).

To identify the configuration, the single crystal of product **3aq** was cultivated by solvent evaporation. And the regio- and stereoselectivity of products were further confirmed the X-ray crystallographic analysis of the obtained product **3aq** (Fig. 1).

Based on our previous work,⁴⁷ a plausible reaction pathway was proposed in Scheme 3. The interaction of HFIP hydrogen bonding linear aggregates⁴⁸ with sulfenylation reagent 2a may strongly activate the sulfenylation reagent, which generates the active intermediate **B** (Scheme 3). Sequentially, a sulfonium **C** is produced from intermediate **B** with an alkyne, followed by a nucleophilic attack of SCN anion to obtain the products 3.

Conclusions

In summary, we have developed a widely applicable regio- and stereoselective thiocyanatothiolation of alkynes and alkenes under simple and mild conditions. This metal-free system offers good chemical yields and functional group tolerance. At present, the fluorinated reagent HFIP, which is not a green solvent, is indeed a limitation of this method, but as scientific research continues, we believe that green fluorinated reagents can be discovered. Other similar thiolation systems are currently investigated in our laboratory.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

- 1 M. Fontecave, S. Ollagnier-de-Choudens and E. Mulliez, Biological Radical Sulfur Insertion Reactions, Chem. Rev., 2003, 103, 2149-2166.
- 2 S. Oida, Y. Tajima, T. Konosu, Y. Nakamura, A. Somada, T. Tanaka, S. Habuki, T. Harasaki, Y. Kamai, T. Fukuoka, S. Ohya and H. Yasuda, Synthesis and Antifungal Activities of R-102557 and Related Dioxane-Triazole Derivatives, Chem. Pharm. Bull., 2000, 48, 694-707.
- 3 E. A. Ilardi, E. Vitaku and J. T. Njardarson, Data-Mining for Sulfur and Fluorine: An Evaluation of Pharmaceuticals To Reveal Opportunities for Drug Design and Discovery, I. Med. Chem., 2014, 57, 2832-2842.
- 4 I. P. Beletskaya and V. P. Ananikov, Transition-Metal-Catalyzed C-S, C-Se, and C-Te Bond Formation via Cross-Coupling and Atom-Economic Addition Reactions, Chem. Rev., 2011, 111, 1596-1636.
- 5 C. Jacob, E. Battaglia, T. Burkholz, D. Peng, D. Bagrel and M. Montenarh, Control of Oxidative Posttranslational Cysteine Modifications: From Intricate Chemistry to Widespread Biological and Medical Applications, Chem. Res. Toxicol., 2012, 25, 588-604.
- 6 J. E. Casida, Unexpected Metabolic Reactions and Secondary Targets of Pesticide Action, J. Agric. Food Chem., 2016, 64, 4471-4477.
- 7 J. E. Casida and K. A. Durkin, Pesticide Chemical Research in Toxicology: Lessons from Nature, Chem. Res. Toxicol., 2017, 30, 94-104.
- 8 A. A. Constantinescu, K. Caliskan, O. C. Manintveld, R. van Domburg, L. Jewbali and A. H. M. M. Balk, Weaning from inotropic support and concomitant beta-blocker therapy in severely ill heart failure patients: take the time in order to improve prognosis, Eur. J. Heart Failure, 2014, 16, 435-443.
- 9 A. M. Giannetti, H. Wong, G. J. P. Dijkgraaf, E. C. Dueber, D. F. Ortwine, B. J. Bravo, S. E. Gould, E. G. Plise, B. L. Lum, V. Malhi and R. A. Graham, Identification, Characterization, and Implications of Species-Dependent Plasma Protein Binding for the Oral Hedgehog Pathway Inhibitor Vismodegib (GDC-0449), J. Med. Chem., 2011, 54, 2592-2601.
- 10 S. F. Nielsen, E. Ø. Nielsen, G. M. Olsen, T. Liljefors and D. Peters, Novel Potent Ligands for the Central Nicotinic Acetylcholine Receptor: Synthesis, Receptor Binding, and 3D-QSAR Analysis, J. Med. Chem., 2000, 43, 2217-2226.
- 11 D. Huang, J. Chen, W. Dan, J. Ding, M. Liu and H. Wu, A Metal-Free Sulfenylation and Bromosulfenylation of Indoles: Controllable Synthesis of 3-Arylthioindoles and 2-Bromo-3-arylthioindoles, Adv. Synth. Catal., 2012, 354, 2123-2128.
- 12 C. J. Nalbandian, E. M. Miller, S. T. Toenjes and J. L. Gustafson, A conjugate Lewis base-Brønsted acid catalyst for the sulfenylation of nitrogen containing heterocycles under mild conditions, Chem. Commun., 2017, 53, 1494-1497.

- Zhu, Q. 13 D. Y. Gu, L. Lu Shen, and N-Difluoromethylthiophthalimide: Α Shelf-Stable, Electrophilic Reagent for Difluoromethylthiolation, J. Am. Chem. Soc., 2015, 137, 10547-10553.
- 14 S. Song, Y. Zhang, A. Yeerlan, B. Zhu, J. Liu and N. Jiao, Cs₂CO₃-Catalyzed Aerobic Oxidative Cross-Dehydrogenative Coupling of Thiols with Phosphonates and Arenes, Angew. Chem., Int. Ed., 2017, 56, 2487-2491.
- 15 S. Vásquez-Céspedes, A. Ferry, L. Candish and F. Glorius, Heterogeneously Catalyzed Direct C·H Thiolation of Heteroarenes, Angew. Chem., Int. Ed., 2015, 54, 5772-5776.
- 16 S. Bhunia, G. G. Pawar, S. V. Kumar, Y. Jiang and D. Ma, Selected Copper-Based Reactions for C-N, C-O, C-S, and C-C Bond Formation, Angew. Chem., Int. Ed., 2017, 56,
- 17 X. Li, L. Li, X. Mo and D. Mo, Transition-metal-free synthesis of thiocyanato- or nitro-arenes through diaryliodonium salts, Synth. Commun., 2016, 46, 963-970.
- 18 S. Liu, X. Zeng and B. Xu, Hydrogen-Bonding-Network-Assisted Regioselective Trifluoromethylthiolation and Sulfenylation of Electron-Rich (Hetero)arenes, Asian J. Org. Chem., 2019, 8, 1372-1375.
- 19 S. Liu, X. Zeng and B. Xu, Regio- and stereoselective halothiolation of alkynes using lithium halides and Nthiosuccinimides, Org. Chem. Front., 2020, 7, 1690-1695.
- 20 S. Liu, X. Zeng and B. Xu, Practical fluorothiolation and difluorothiolation of alkenes using pyridine-HF and Nthiosuccinimides, Org. Chem. Front., 2020, 7, 119-125.
- 21 M. Wang, Q. Fan and X. Jiang, Metal-free construction of primary sulfonamides through three diverse salts, Green Chem., 2018, 20, 5469-5473.
- 22 W. Tan, C. Wang and X. Jiang, Green carbon disulfide surrogate via a combination of potassium sulfide and chloroform for benzothiazine-thione and benzothiazolethione construction, Org. Chem. Front., 2018, 5, 2390-2394.
- 23 J. Wei, Y. Li and X. Jiang, Aqueous Compatible Protocol to Both Alkyl and Aryl Thioamide Synthesis, Org. Lett., 2016, 18, 340-343.
- 24 J. T. Reeves, K. Camara, Z. S. Han, Y. Xu, H. Lee, C. A. Busacca and C. H. Senanayake, The Reaction of Grignard Reagents with Bunte Salts: A Thiol-Free Synthesis of Sulfides, Org. Lett., 2014, 16, 1196-1199.
- 25 F. Ke, Y. Qu, Z. Jiang, Z. Li, D. Wu and X. Zhou, An Efficient Copper-Catalyzed Carbon-Sulfur Bond Formation Protocol in Water, Org. Lett., 2011, 13, 454-457.
- 26 B. Exner, B. Bayarmagnai, F. Jia and L. J. Goossen, Iron-Catalyzed Decarboxylation of Trifluoroacetate and Its Application to the Synthesis of Trifluoromethyl Thioethers, Chem.-Eur. J., 2015, 21, 17220-17223.
- 27 V. Aureggi and G. Sedelmeier, 1,3-Dipolar Cycloaddition: Click Chemistry for the Synthesis of 5-Substituted Tetrazoles from Organoaluminum Azides and Nitriles, Angew. Chem., Int. Ed., 2007, 46, 8440-8444.
- 28 H. Yang, X.-H. Duan, J.-F. Zhao and L.-N. Guo, Transition-Metal-Free Tandem Radical Thiocyanooxygenation of Olefinic Amides: A New Route to SCN-Containing Heterocycles, Org. Lett., 2015, 17, 1998-2001.

- 29 L.-N. Guo, Y.-R. Gu, H. Yang and J. Hu, Transition-metal free thiocyanooxygenation of functionalized alkenes: facile routes to SCN-containing dihydrofurans and lactones, *Org. Biomol. Chem.*, 2016, 14, 3098–3104.
- 30 B. Chen, S. Guo, X. Guo, G. Zhang and Y. Yu, Selective Access to 4-Substituted 2-Aminothiazoles and 4-Substituted 5-Thiocyano-2-aminothiazoles from Vinyl Azides and Potassium Thiocyanate Switched by Palladium and Iron Catalysts, *Org. Lett.*, 2015, 17, 4698–4701.
- 31 Y. Chen, S. Wang, Q. Jiang, C. Cheng, X. Xiao and G. Zhu, Palladium-Catalyzed Site-Selective sp³ C–H Bond Thiocyanation of 2-Aminofurans, *J. Org. Chem.*, 2018, **83**, 716–722.
- 32 L. Zhen, K. Yuan, X.-y. Li, C. Zhang, J. Yang, H. Fan and L. Jiang, Cascade Reaction of Propargyl Amines with AgSCF₃, as Well as One-Pot Reaction of Propargyl Amines, AgSCF₃, and Di-tert-butyl Peroxide: Access to Allenyl Thiocyanates and Allenyl Trifluoromethylthioethers, *Org. Lett.*, 2018, **20**, 3109–3113.
- 33 Y.-F. Zeng, D.-H. Tan, Y. Chen, W.-X. Lv, X.-G. Liu, Q. Li and H. Wang, Direct radical trifluoromethylthiolation and thiocyanation of aryl alkynoate esters: mild and facile synthesis of 3-trifluoromethylthiolated and 3-thiocyanated coumarins, *Org. Chem. Front.*, 2015, 2, 1511–1515.
- 34 Q. Lin, W. Yang, Y. Yao, S. Chen, Y. Tan, D. Chen and D. Yang, Copper-Catalyzed Diastereoselective 1,2-Difunctionalization of Oxabenzonorbornadienes Leading to β-Thiocyanato Thioethers, *Org. Lett.*, 2019, **21**, 7244–7247.
- 35 X. Zeng, B. Chen, Z. Lu, G. B. Hammond and B. Xu, Homogeneous and Nanoparticle Gold-Catalyzed Hydrothiocyanation of Haloalkynes, *Org. Lett.*, 2019, 21, 2772–2776.
- 36 T. Kitamura, S. Kobayashi and H. Taniguchi, Photolysis of vinyl halides. Reaction of photogenerated vinyl cations with cyanate and thiocyanate ions, *J. Org. Chem.*, 1990, 55, 1801–1805.
- 37 G. Jiang, C. Zhu, J. Li, W. Wu and H. Jiang, Silver-Catalyzed Regio- and Stereoselective Thiocyanation of Haloalkynes: Access to (Z)-Vinyl Thiocyanates, *Adv. Synth. Catal.*, 2017, **359**, 1208–1212.

- 38 X. Zeng and L. Chen, Iodine-mediated regio- and stereoselective iodothiocyanation of alkynes in aqueous ethanol, *Org. Biomol. Chem.*, 2018, **16**, 7557–7560.
- 39 T. Steiner, The Hydrogen Bond in the Solid State, *Angew. Chem.*, *Int. Ed.*, 2002, **41**, 48-76.
- 40 X. Zeng, J. Li, C. K. Ng, G. B. Hammond and B. Xu, (Radio) fluoroclick Reaction Enabled by a Hydrogen-Bonding Cluster, *Angew. Chem., Int. Ed.*, 2018, 57, 2924–2928.
- 41 Z. Lu, X. Zeng, G. B. Hammond and B. Xu, Widely Applicable Hydrofluorination of Alkenes via Bifunctional Activation of Hydrogen Fluoride, *J. Am. Chem. Soc.*, 2017, **139**, 18202–18205.
- 42 X. Zeng, S. Liu, Z. Shi and B. Xu, Hydrogen Bonding Cluster-Enabled Addition of Sulfonic Acids to Haloalkynes: Access to Both (E)- and (Z)-Alkenyl Sulfonates, *Org. Lett.*, 2016, **18**, 4770–4773.
- 43 X. Zeng, S. Liu, Y. Yang, Y. Yang, G. B. Hammond and B. Xu, Regio- and Stereoselective Synthesis of 1,2-Dihaloalkenes Using In-Situ-Generated ICl, IBr, BrCl, I₂, and Br₂, *Chem*, 2020, **6**, 1018–1031.
- 44 R. Ebule, S. Liang, G. B. Hammond and B. Xu, Chloride-Tolerant Gold(I)-Catalyzed Regioselective Hydrochlorination of Alkynes, *ACS Catal.*, 2017, 7, 6798– 6801.
- 45 X. Zeng, S. Liu, G. B. Hammond and B. Xu, Hydrogen-Bonding-Assisted Brønsted Acid and Gold Catalysis: Access to Both (E)- and (Z)-1,2-Haloalkenes via Hydrochlorination of Haloalkynes, *ACS Catal.*, 2018, **8**, 904–909.
- 46 J. Oliver-Meseguer, A. Doménech-Carbó, M. Boronat, A. Leyva-Pérez and A. Corma, Partial Reduction and Selective Transfer of Hydrogen Chloride on Catalytic Gold Nanoparticles, Angew. Chem., Int. Ed., 2017, 56, 6435–6439.
- 47 S. Liu, X. Zeng and B. Xu, Regio- and stereoselective halothiolation of alkynes using lithium halides and N-thiosuccinimides, *Org. Chem. Front.*, 2020, 7, 1690–1695.
- 48 S. Henkel, M. C. Misuraca, P. Troselj, J. Davidson and C. A. Hunter, Polarisation effects on the solvation properties of alcohols, *Chem. Sci.*, 2018, 9, 88–99.