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Carbon dots for effective photodynamic inactivation of virus

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The antiviral function of carbon dots (CDots) with visible light exposure was evaluated, for which the model bacteriophages MS2 as a surrogate of small RNA viruses were used. The results show clearly that the visible light-activated CDots are highly effective in diminishing the infectivity of MS2 in both low and high titer samples to the host *E. coli* cells, and the antiviral effects are dot concentration- and treatment time-dependent. The action of CDots apparently causes no significant damage to the structural integrity and morphology of the MS2 phage or the breakdown of the capsid proteins, but does result in the protein carbonylation (a commonly used indicator for protein oxidation) and the degradation of viral genomic RNA. Mechanistically the results may be understood in the framework of photodynamic effects that are associated with the unique excited state properties and processes of CDots. Opportunities for potentially broad applications of CDots coupled with visible/natural light in the prevention and control of viral transmission and spread are highlighted and discussed.

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1. Introduction

Infectious diseases continue to affect hundreds of millions of people each year with serious health outcomes, accounting for ~20% global mortality, among which about one-third are caused by viral infections.¹ Influenza or “the flu”, HIV (human immunodeficiency virus), and herpes are some of the many viruses that kill millions of people every year. In addition, new viruses, such as Ebola, SARS, and SARS-CoV-2, are continuously emerging and cause serious public health problems. For prevention of viral infections, vaccination is known as the best practice, but only a limited number of vaccines are available. For many viruses, including HIV, noroviruses, and other newly emerged viruses, there is presently no vaccine. Also, constant antigenic shift in some viruses creates difficulties for vaccine development. In reality, the control and prevention of viral transmission and spread largely rely on effective hygiene and sanitation practices, though other tools are also being explored to help the control of viral infections.^{2,3} Many traditional disinfection agents have been used for such purposes, but

recently various nanomaterials of antiviral functions have been explored with very promising outcomes.

In the early exploration for inactivation of viruses or virus-like particles (VLPs), various metal and metal oxide nanoparticles (NPs), including NPs of silver,^{4,5} copper iodide,⁵ and titanium dioxide⁶ have been used. Their small sizes, high surface-to-volume ratio,⁷ ability of direct interaction with viral capsid proteins,⁸ and/or other unique properties of the materials at nanoscale have contributed to their observed antiviral activities, making their use a more popular alternative to other physical and chemical antiviral strategies. Moreover, the coupling of nanomaterials with photon energies for photodynamic inactivation (PDI) of microbials has also attracted much attention. In the PDI, a popular approach has been the use of UV light to illuminate on selected NPs such as colloidal TiO₂ to generate reactive oxygen species (ROS), including singlet molecular oxygen, superoxide and hydroxyl radicals, to kill pathogenic bacteria or viruses.⁹ Since UV light is hazardous for broader applications in living environment, various modifications to TiO₂ NPs have been made to extend absorption into the visible.^{10–12} Similarly, the uses of gold NPs with organic dyes,^{13–15} cationic fullerenes,¹⁶ and semiconductor quantum dots (QDs)^{17–19} have been explored in the development of antimicrobial agents that can be excited by visible photons. More recently, carbon dots (CDots), which are small carbon NPs with particle surface functionalized by organic molecules,^{20–31} have been found and demonstrated for their strong visible/natural light-activated antimicrobial functions.^{32,33}

CDots may be considered as a special kind of “core-shell” nanostructures, each with a small carbon nanoparticle core

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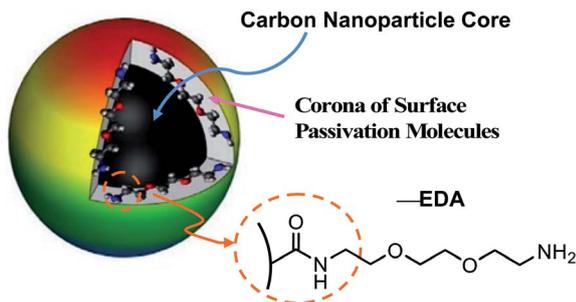


Fig. 1 A cartoon illustration on the structure of CDots in general, with the carbon nanoparticle surface functionalization by EDA molecules highlighted for EDA-CDots.

(mostly less than 10 nm in diameter) and a thin shell of soft materials (the organic or biological surface molecules) (Fig. 1). Their broad optical absorptions over the entire visible spectral range enable their efficient visible/natural light activation.^{21,23} Upon photoexcitation of CDots, there are efficient charge separations for the generation of separated electrons and holes, which are trapped at various surface defect sites in the dots and likely participate directly in the observed antimicrobial activities.^{34,35} As widely agreed,^{20,21,23–31,35} the radiative recombination of the electrons and holes results in the formation of emissive excited states, which are responsible for the famous bright and colorful fluorescence emissions of CDots and also contribute to the photodynamic antimicrobial functions. The antimicrobial activities of CDots are known to correlate with their surface functionality, surface charge, fluorescence emission quantum yields, and other structural/material characteristics.^{32,33,36}

Here, we report a study of visible light-activated CDots for effective inactivation of virus and an exploration on mechanistic implications of the experimental results using bacteriophage MS2 as a model (a surrogate of a number of small RNA viruses

such as norovirus, Ebola virus, Marburg virus). In the study, the CDots with 2,2'-(ethylenedioxy)bis(ethylamine) (EDA) for the carbon nanoparticle surface functionalization (thus EDA-CDots, Fig. 1) were selected for their being chemically and structurally well characterized³⁷ and effective in antibacterial activities against multiple laboratory model bacteria,^{32,33,36,38} as well as exhibiting significant antiviral potential in the initial evaluation of their inhibitory effect on the binding of norovirus VLPs to host cell receptor.³⁹ The use of bacteriophage MS2 in this study was to further reveal the effect of CDots on viral infectivity and viral genome, and to explore mechanistic insights into the action of CDots on the virus, both of which would not be possible with the VLPs.

2. Results and discussion

2.1 Bacteriophage infection and antiviral strategies

The process of viral infection to a host cell consists of multiple steps, generally including viral attachment, penetration (viral RNA/DNA entry), replication, assembly, release, and host cell lysis, but the details in these steps differ greatly between infections by different virus species.^{40,41} The viral attachment to a host cell is achieved when specific proteins on the viral capsid or viral envelope bind to specific host receptor proteins on the host cell surface. After attachment, animal virions enter into the host cell through receptor-mediated endocytosis or membrane fusion,^{42,43} while viruses enter into plant cells and fungi cells only after trauma to the cell walls.^{41,44} The general process of bacteriophage infection to a host bacterial cell is shown in Fig. 2. Generally bacteriophages use specialized surface receptor-binding proteins to interact with and adhere to their specific cognate host receptors, and then the phages use various mechanisms to breach the bacterial cell wall and inject their genome into the host bacterial cell, leaving the empty phage capsid outside the cell.⁴⁵ Once the viral genome is uncoated in

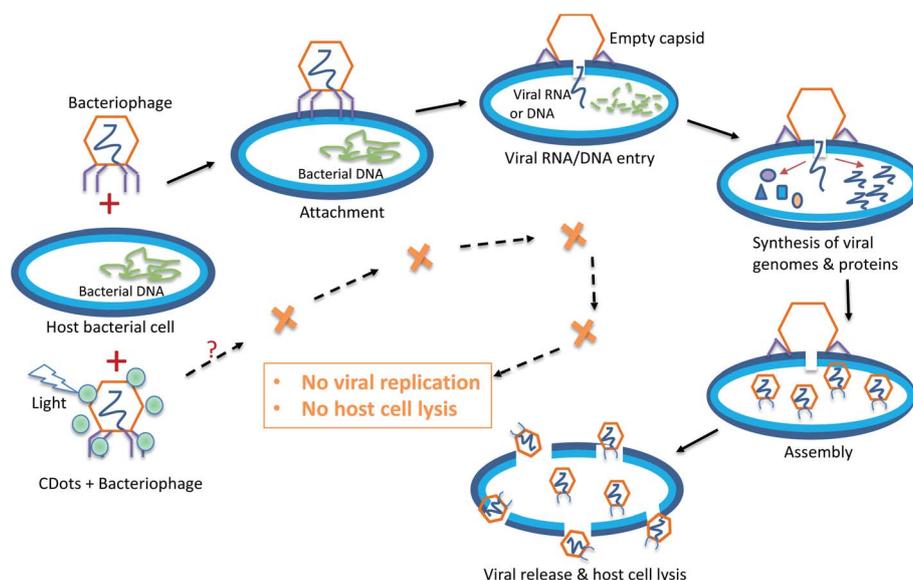


Fig. 2 The general process of bacteriophage infection to its host cells, and the expected outcome of PDI antiviral function of CDots.



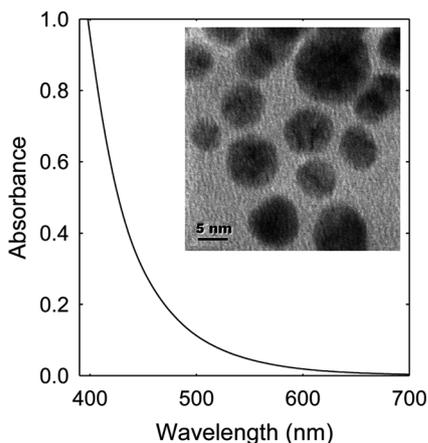


Fig. 3 The absorption spectrum of EDA-CDots in aqueous solution. Inset: A TEM image of EDA-CDots.

the host cell, the replication starts and the transcription or translation of the viral genome initiates. After the synthesis of viral genome and proteins, the proteins are packaged with the newly replicated viral genomes to assemble new virions ready to be released from the host cells, resulting in the lysis of infected host cells.⁴⁵ However, in some other cases, viruses undergo a lysogenic cycle where the viral genome is incorporated by genetic recombination into a specific site in the host's chromosome.⁴¹

Many antiviral agents/drugs are designed to target either viral proteins or host factors. Host-targeting antivirals aim at the host proteins that are involved in the viral life cycle, so as to regulate the function of the immune system or other cellular processes in host cells.⁴⁶ More relevant to the study reported here in terms of antiviral strategies, the virus-targeting

antivirals can function directly or indirectly to inhibit the functions of viral proteins, mostly enzymatic activities, or to block viral replication machinery. Among direct virus-targeting antivirals may be inhibitors of virus attachment, entry, protease, polymerase, integrase, methyltransferase, helicase, *etc.*⁴⁶ Other direct antiviral methods may physically destroy/break viral capsid, such as high pressure, radiation, and the action of some nanoparticles.^{47–50} The photodynamic inactivation (PDI) as an antiviral approach exploits the function of photosensitizers under light illumination to generate ROS to attack viral envelope lipids, core proteins, capsid, and nucleic acid, leading to the loss of virus infectivity.⁵¹ CDots due to their unique photoexcited state properties and processes represent a new class of particularly effective PDI agents, as already demonstrated in their visible/natural light-activated actions against bacterial cells, are expected to be similarly effective antiviral PDI agents (Fig. 2).

2.2 PDI with CDots to diminish viral infectivity

EDA-CDots (Fig. 1) were prepared and characterized by following the experimental protocols reported previously.^{37,52} These dots are 5–10 nm in diameter according to microscopy characterization results (Fig. 3, inset). They are readily soluble in aqueous media, and the resulting aqueous solution is strongly absorptive in the visible spectrum (Fig. 3) for photoexcitation with broad-band visible light.

The antiviral activity of EDA-CDots was evaluated under different conditions. The initial experiment was on MS2 samples containing low virus titers (~ 50 – 60 PFU per mL). The samples were treated with EDA-CDots at different concentrations ranging from 1 to $20 \mu\text{g mL}^{-1}$ with visible light illumination (from 36 W light box) or in the dark for 1 h. The infectivity of treated and untreated MS2 was evaluated by the

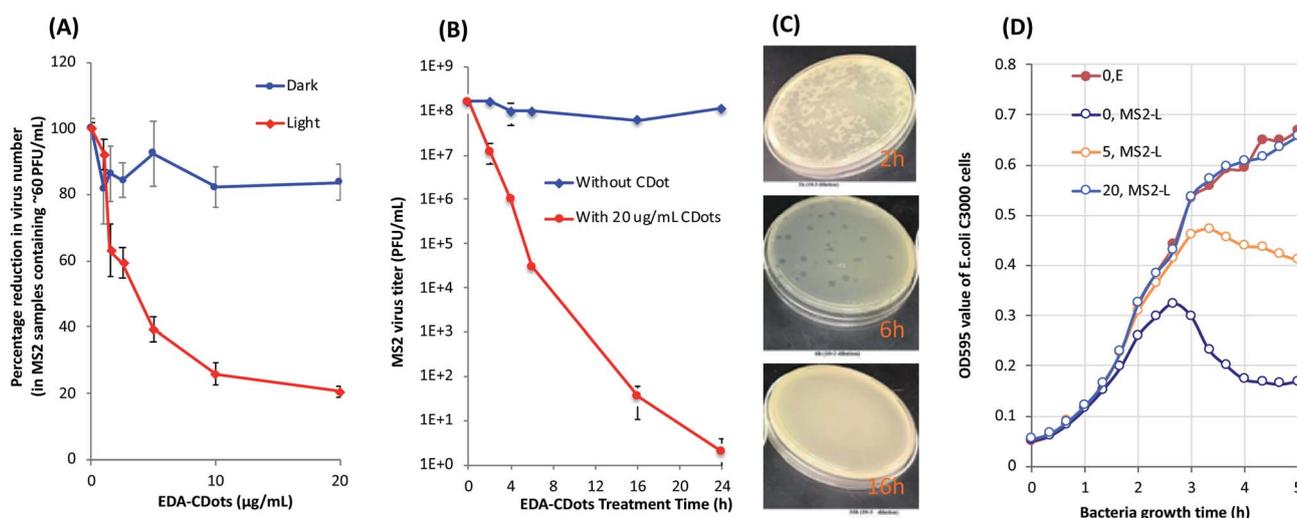


Fig. 4 (A) Percentage reduction in virus titers when MS2 samples containing low virus titers (<100 PFU per mL) were treated with different concentrations of EDA-CDots under visible light for 1 h. (B) Reduction in virus titers when MS2 samples containing high virus titers ($\sim 10^8$ PFU per mL) were treated with $20 \mu\text{g mL}^{-1}$ EDA-CDots under visible light for different times. (C) Images of the representative plates of MS2 plaque forming units on the lawn of *E. coli* C3000 after the MS2 samples were treated with $20 \mu\text{g mL}^{-1}$ EDA-CDots under visible light for 2, 6, and 16 h. (D) The growth curves of *E. coli* C3000 when they were infected with untreated and CDots-treated MS2 samples.



plaque assay. Fig. 4A shows the percentage reduction in MS2 plaque numbers upon 1 h treatment with EDA-CDots under visible light and in dark. The viral titer was significantly reduced with the increasing concentration of EDA-CDots in the treatment under visible light. Specifically, the viral titer reduced to 92.3, 63.2, 59.4, 39.4, 25.8 and 20.6% of the starting viral titer with 1, 1.5, 2.5, 5, 10, and 20 $\mu\text{g mL}^{-1}$ EDA-CDots, respectively, while the corresponding treatments in dark did not cause significant reduction in viral titer. The results confirmed that MS2 phages lost their infectivity to host cells upon the treatment with EDA-CDots under visible light in a dose concentration-dependent manner, but not the treatment in dark.

Next, MS2 samples containing higher virus titers were tested to see how well CDots were able to inactivate the viruses at high viral titers for more practical opportunities. In the experiments, MS2 phage samples of $\sim 10^8$ PFU per mL were treated with 20 $\mu\text{g mL}^{-1}$ EDA-CDots for different time periods ranging from 2 to 24 h under the visible light. Fig. 4B shows the viral titers of the samples at different treatment time points, as determined by the plaque assay, along with controls without CDots treatment. The results show that EDA-CDots were effective in inactivating MS2 phages in the high virus titer samples. With 20 $\mu\text{g mL}^{-1}$ EDA-CDots, the MS2 phages in the samples were completely inactivated (>8 log reduction in viral titer) with 24 h treatment, or >6 log reduction and approximately 1 log reduction (90% reduction) of virus with 16 h and 2 h treatments, respectively. The results indicated that CDots' photo-inactivation of MS2

phages was in a time-dependent manner, as prolonged treatment time could significantly increase the inactivation effectiveness. Fig. 4C shows the representative plates of the plaque forming units (PFU) of MS2 phage on the lawn of *E. coli* strain C3000, after the MS2 samples were treated with 20 $\mu\text{g mL}^{-1}$ EDA-CDots under visible light for 2, 6, and 16 h (all plated upon 10^{-3} dilutions). There was an obvious decrease of PFU in the samples with increasing treatment time from 2 h to 6 h, and the samples with 16 h treatment did not form PFU. The results show that CDots were capable of inactivating MS2 phages in high virus titer samples, suggesting excellent potential for practical applications.

In a separated experiment, the infectivity of MS2 phages with and without the CDots treatment to the host *E. coli* C3000 cells in suspensions was examined. The tests were performed by monitoring the growth curves of *E. coli* C3000 cells that were infected with CDots treated and untreated MS2 phages. As shown in Fig. 4D, *E. coli* C3000 cells without MS2 exhibited a normal growth curve consisting of lag phase, log phase and stationary phase, with the optical density (OD) value reaching ~ 0.66 at 5 h. Untreated MS2 were able to infect *E. coli* cells and cause cell lysis over time. Although an increasing rate of growth of *E. coli* was observed until the time point of ~ 2.7 h, after that, the OD started to decline and continued progressively until reaching a plateau (OD at ~ 0.17), indicating the onset of cell lysis due to the infection of MS2 to the *E. coli* cells. The MS2 treated with 5 $\mu\text{g mL}^{-1}$ EDA-CDots for 16 h under visible light induced *E. coli* cell lysis at a later time point of ~ 3.3 h, and the

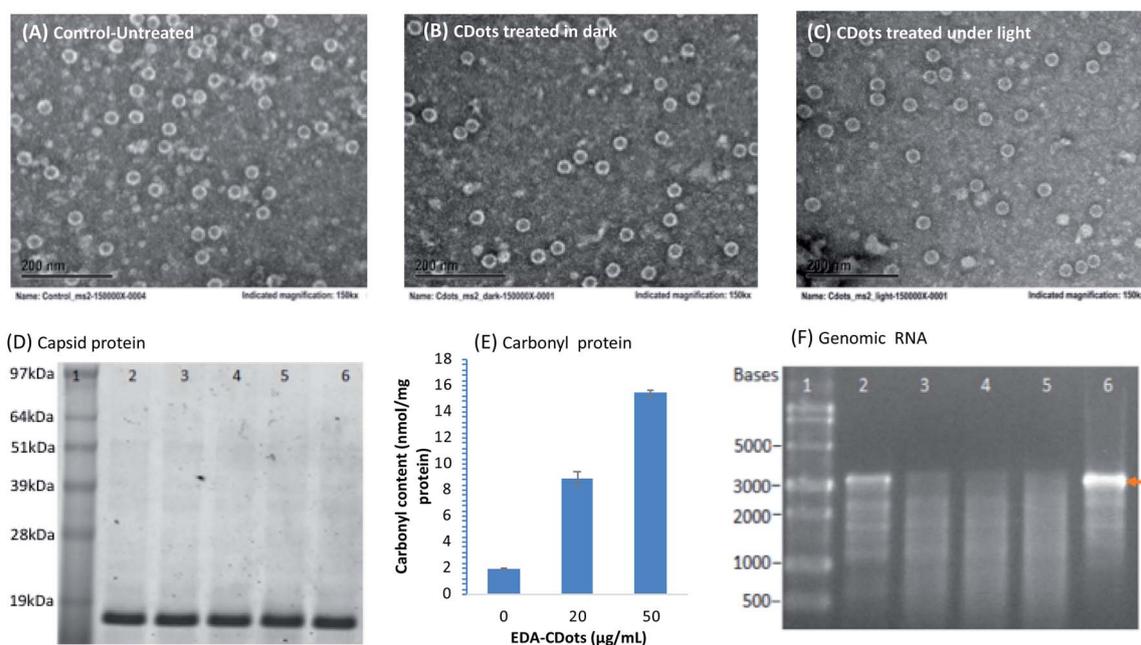


Fig. 5 TEM images of MS2 phages (A) untreated, (B) treated with CDots in dark, and (C) treated with CDots under visible light. (D) Image of SDS-PAGE gel of 13.7 kDa MS2 coat proteins in CDots treated MS2 phages. Lane 1: SeeBlue Plus2 Pre-stained Protein Standard (Invitrogen, Carlsbad, CA); lane 2–5: protein from MS2 phages treated with 0, 10, 20, and 50 $\mu\text{g mL}^{-1}$ CDots under light for 16 h; and lane 6: proteins from freshly thawed MS2 phages without CDots treatment. (E) The carbonyl contents in proteins from MS2 phages that were treated with 20 and 50 $\mu\text{g mL}^{-1}$ EDA-CDots under visible light illumination for 2 h, along with those of the untreated controls. (F) Image of agarose gel electrophoresis of RNA from CDots treated MS2 phages. Lane 1: ssRNA ladder; lane 2–5: RNA from phages treated with 0, 5, 10, and 20 $\mu\text{g mL}^{-1}$ CDots under light for 16 h; and lane 6: RNA isolated from freshly thawed MS2 phages without the CDots treatment.



decline in OD continued at a slower rate than that with untreated MS2, until reaching the OD of ~ 0.41 at 5 h, indicating reduced infectivity of the treated MS2 to *E. coli* cells. However, the MS2 treated with $20 \mu\text{g mL}^{-1}$ EDA-CDots for 16 h did not show a decline in OD within the experimental time, and the growth curve of *E. coli* was the same as the control with uninfected *E. coli* cells, suggesting that the treated MS2 phages had lost their infectivity to *E. coli* cells. In the same experiments, the corresponding MS2 samples treated with CDots in dark also did not cause *E. coli* cell lysis, namely that light exposure in the CDots treatment was necessary to reducing the infectivity of MS2. On the effect of CDots concentrations in the treatments, the growths of *E. coli* C3000 cells infected with MS2 phages that were treated with 0, 5, 20, 30, 40, and $50 \mu\text{g mL}^{-1}$ CDots under visible light for 1 h were examined. The cell lysis apparently decreased with the increasing CDots concentration used in the treatment of MS2 phages, consistent with more MS2 phages losing their infectivity. However, for the effect of treatment time at a constant CDots concentration, 1 h vs. 16 h, the longer time was obviously more effective in inactivating MS2, as expected. Even at a high CDots concentration like $50 \mu\text{g mL}^{-1}$, the 1 h treatment could not completely inactivate the MS2 phages. Nevertheless, these results did consistently confirm that CDots coupled with visible light were effective for inactivating MS2 phages, making them lose their infectivity to host cells, and that the antiviral activity of CDots was dots concentration- and treatment time-dependent.

2.3 PDI damages to the integrity of MS2 phage, capsid proteins, and viral genomic RNA

MS2 phages treated with CDots under visible light and in dark, as well as the untreated phages were examined using TEM (Fig. 5A–C). Surprisingly, the CDots treated MS2 phages remained intact without obvious morphologic changes according to the TEM imaging, comparable to the untreated ones or those treated with CDots in dark. On the viral structural integrity, this observation is consistent with what was previously observed in the study of human norovirus virus-like particles (VLPs) treated with CDots, in which the VLPs maintained their morphology after CDots treatment despite of the loss of their binding capability to host receptors.³⁹ This is also similar to what was found in the use of CDots for PDI against bacteria, in which the bacterial cells photodynamically inactivated by CDots also remained intact.^{53,54} However, the outcomes were significantly different by the action of many metal or metal oxide NPs and carbon nanotubes, in which physical damages to cell membrane or viral capsids were observed.^{47,48}

Next, the integrity of phage capsid protein was examined using SDS PAGE gel (Fig. 5D). The MS2 capsid consists of 2 structural proteins: 178 copies of coat protein (~ 13.7 kDa) and 1 copy of A-protein (~ 44 kDa).⁵⁵ As shown in Fig. 5D, the bands corresponding to 13.7 kDa were the coat proteins from MS2 phages that were treated with 0, 10, 20, and $50 \mu\text{g mL}^{-1}$ CDots under visible light for 16 h, showing no obvious difference in the band intensity among the MS2 samples treated with different concentrations of CDots, and the controls. Due to the

low copy number of A-protein, it was undetectable on the SDS PAGE gel. The results suggest that the CDots treatment at tested concentrations did not break down the coat proteins in the capsid of MS2 phages, consistent with the intact morphology of MS2 observed in TEM imaging.

To further assess the consequence at the protein level in MS2 phages upon the CDots treatment with visible light, the oxidation of protein was examined. Proteins have often been described as major targets for oxidative modifications, with the formation of protein carbonyls being among the most prevalent oxidative lesions of proteins.⁵⁶ In fact, protein carbonyls are commonly used as a biomarker to measure oxidative stress *in vivo* for biological samples.^{57–59} In this study, the carbonyl formation in CDots treated MS2 phages was probed and analyzed. Fig. 5E shows the carbonyl contents in MS2 phages that were treated with 20 and $50 \mu\text{g mL}^{-1}$ EDA-CDots under visible light illumination for 2 h, along with those of the controls without CDots treatment. The carbonyl contents were 1.9, 8.9, and $15.4 \text{ nmol mg}^{-1}$ protein in the MS2 phages treated with 0, 20, and $50 \mu\text{g mL}^{-1}$ CDots, respectively, corresponding to 3.6 and 7.1 fold increases in the treated MS2 phages. Carbonylation often causes major changes in protein structure and function, including unfolding and/or aggregation.^{56,60} It has been reported that changes to host receptor recognition sites on the virus capsid could inhibit host receptor attachment, thereby resulting in its inability to infect the host cell.⁵⁵ Although MS2 phages and their coat proteins apparently remained intact upon the CDots treatment with visible light, there was oxidation of the capsid proteins by the photo-generated ROS during treatment, resulting in the observed high level of protein carbonylation, which could lead to structural changes/dysfunction in host receptor recognition sites on the capsid proteins, thus responsible for the loss of their infectivity.

Further examined was the degradation of the genomic RNA in MS2 phages due to the CDots treatment with visible light. Experimentally, the total RNA from the CDots-treated MS2 phages was isolated and analyzed on agarose gel after electrophoresis (Fig. 5F). The MS2 genome is comprised of 3569 nucleotides (nt) of single-stranded RNA, consisting of 1487 nt *mat* gene, 510 nt *cp* gene, 295 nt *lys* gene, and 2055 nt *rep* gene, while *lys* gene overlaps with both the 3'-end of the upstream *cp* gene and the 5'-end of the *rep* downstream *rep* gene.^{61,62} As shown in Fig. 5F, the fresh MS2 samples and the controls (without CDots but processed the same as all other samples) showed a clear band at size of ~ 3569 nt, corresponding to the total genomic RNA of MS2. However, all the CDots treated MS2 samples did not show the clear band of their total genomic RNA, indicating the degradation of genomic RNA due to the treatment. In fact, the extent of degradation increased with the increasing CDots concentrations (from 5, 10 to $20 \mu\text{g mL}^{-1}$) used in the treatment with visible light.

The results above show that CDots could readily be activated by visible light to effectively diminish the infectivity of MS2 phages, but not at all with the same CDots treatment in dark. The former apparently oxidized the capsid proteins of MS2, thus most likely disabled the viral recognition/binding ability to the host cells, and also degraded the viral genomic RNA, though



without breaking apart the viral particles. To the best of our knowledge, this is apparently the first demonstration on the potent PDI actions of CDots under visible light for effectively diminishing viral infectivity. The observed high performances of CDots justify these simple carbon nanomaterials as an extremely promising platform in the development of countermeasures for the control and prevention of viral transmission/spread. Beyond the high performances, the CDots platform also offers other major advantages, CDots have higher absorptivity in visible light wavelength than traditional organic dye photosensitizers, and more efficient PDI due to the generation of highly reactive redox pairs beyond the traditional ROS in the light activation processes.⁶³ Especially with their known benign and nontoxic nature,^{64–66} CDots offer excellent opportunities for broad applications in many healthcare settings and human living environments.

2.4 Mechanistic issues/implications of the PDI diminishing viral infectivity

The results presented above suggest the necessity of visible light activation in CDots' antiviral function, pointing to a similar photodynamic inactivation (PDI) scheme to what has been adopted for the observed antibacterial activities of CDots with visible/natural light illumination.^{32–34} The potent PDI function of CDots could be attributed to their unique photoexcited state properties and processes.^{34,35}

As illustrated in Fig. 6, the photoexcitation of CDots drives the initial ultrafast charge separation to form charge separated

species, which are electrons and holes trapped at various defect sites. These are highly reactive species, similar to or even more reactive than the classical ROS by organic dyes *via* photodynamic effects. Their recombination results in the formation of emissive excited states (Fig. 6), which are not only responsible for the observed bright and colorful fluorescence emissions but also capable of the same organic dye-like ROS generation. The collective actions of the initially formed charged separated species and the subsequently generated ROS make the PDI of CDots particularly effective and efficient, as made evident by the already reported excellent antibacterial outcomes.^{34,35} The same PDI framework may be applicable to the observed effective antiviral action of CDots under visible-light illumination to result in the loss of infectivity of MS2 phages.

On the traditional photodynamic effects with organic dyes as photosensitizers, ROS scavengers such as L-histidine and *tert*-butanol are often used for the quenching of singlet molecular oxygen and hydroxyl radicals, respectively.^{67,68} These scavengers were added to the CDots treatment of MS2 phages for an evaluation of their quenching effects on the antiviral activities of EDA-CDots with visible light. In the experiments, MS2 phage samples containing $\sim 10^8$ PFU per mL were treated with $20 \mu\text{g mL}^{-1}$ EDA-CDots under visible light illumination for 2 h, which resulted in ~ 1.16 log virus reduction (Fig. 7). The same MS2 samples treated with each of the two scavengers only without any CDots under the same light illumination caused no virus reduction. In the CDots treatment of MS2 samples with visible light, the addition and presence of up to 90 mM *tert*-butanol resulted in no quenching effect, still the same virus reduction without *tert*-butanol (Fig. 7). On the other hand, the presence of

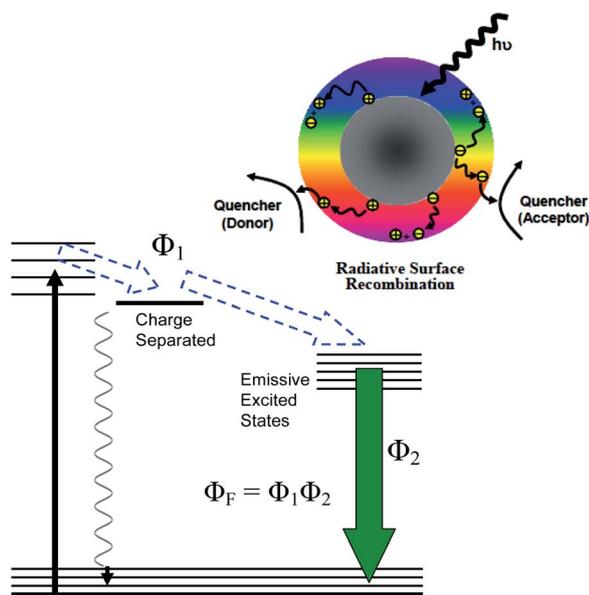


Fig. 6 Upper: Cartoon illustration on the photoexcited state processes in CDots, including the rapid charge separation, and the trapping of electrons and holes thus formed (which can be quenched by external electron acceptor and donor, respectively) and their radiative recombinations. Lower: The energy diagram for the same processes, with the observed fluorescence quantum yield Φ_F as a product of Φ_1 (denoting the overall yield for the charge separation and radiative recombinations) and Φ_2 (the yield of radiative process).

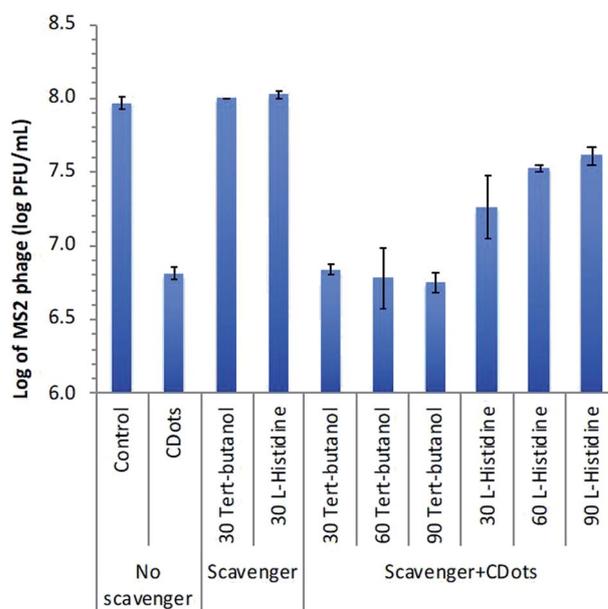


Fig. 7 The effects of scavengers L-histidine and *tert*-butanol on the antiviral function of CDots toward MS2 phages. The concentrations of the respective scavengers used in the experiments were 30, 60, and 90 mM. MS2 phage samples: $\sim 10^8$ PFU mL⁻¹; treatment: $20 \mu\text{g mL}^{-1}$ EDA-CDots under visible light illumination for 2 h.



L-histidine in the treatment of the same MS2 samples with CDots and visible light did quench the antiviral effect, with the level of quenching dependent on the L-histidine concentration (Fig. 7). At the highest L-histidine concentration of 90 mM, the antiviral effect decreased from the originally ~ 1.16 log to 0.36 log, namely that the presence of L-histidine significantly protected MS2 phages from the antiviral action of the photoexcited CDots.

In the study of photodynamic effects by organic dyes, *tert*-butanol and L-histidine as ROS scavengers are often used to probe the sub-species in the ROS, specifically hydroxyl radicals and singlet molecular oxygen, respectively.^{67,68} However, CDots are hardly organic dyes-equivalent, despite the fact that some of their photoexcited state properties and processes may share similar characteristics with organic dye photosensitizers (Fig. 6). The observed no quenching by *tert*-butanol may not be sufficient for a conclusion on the absence of hydroxyl radicals, because such radicals, if present, could be too short lived and/or protected by the dot surface EDA molecules from the scavenging action of *tert*-butanol. In a similar consideration, L-histidine could be scavenging singlet molecular oxygen, but alternatively might be quenching the action of CDots by interfering their interactions with MS2 phages. These mechanistic issues are complicated, requiring separate systematic investigations. However, one significant outcome for sure from the scavenging experiments is that the antiviral function of photoexcited CDots could not be quenched completely, which might be correlated with expected activities of the initially formed highly reactive yet short-lived charge separated species (Fig. 6). Again, more dedicated investigations for mechanistic insights are needed.

3. Conclusions

The results from this study show unambiguously that CDots activated by visible light are highly effective antiviral agents, diminishing the infectivity of bacteriophage MS2 to the host cells. The action of photoexcited CDots apparently did not significantly change the integrity and morphology of the MS2 phage or break down the capsid proteins, but did result in the protein carbonylation (a commonly used indicator for protein oxidation) and the degradation of genomic RNA. Mechanistically the antiviral action and outcome are attributed to similar photodynamic inactivation (PDI) to what is more established for the visible/natural light-activated antibacterial function of CDots. With more investigations for an improved understanding of mechanistic details, CDots may be established as a new class of antiviral agents for broad applications in the prevention and control of viral infection/spread.

4. Materials and methods

4.1 Synthesis and characterization of EDA-CDots

The CDots were synthesized by chemical functionalization of pre-processed and selected small carbon nanoparticles. The processing and selection to harvest the nanoparticles from a commercially acquired sample of carbon nano-powders (US

Research Nanomaterials, Inc.) were similar to those reported previously.^{37,69} In a typical experiment, a portion of the as-supplied carbon nano-powders (2 g) was refluxed in aqueous nitric acid (8 M, 200 mL) for 48 h. The reaction mixture was cooled back to room temperature, and centrifuged at 1000g to discard the supernatant. The residue was re-dispersed in deionized water, dialyzed in a membrane tubing (molecular weight cut-off ~ 500) against fresh water for 48 h, and then centrifuged at 1000g to retain the supernatant. Upon the removal of water, small carbon nanoparticles were recovered and used in the subsequent functionalization reaction.

For the synthesis of EDA-CDots,^{37,70} the small carbon nanoparticles were refluxed in neat thionyl chloride for 12 h. Upon the removal of excess thionyl chloride, the treated sample (50 mg) was mixed well with carefully dried 2,2'-(ethylenedioxy) bis(ethylamine) (EDA, Sigma-Aldrich, ~ 1 g) liquid in a round-bottom flask, heated to 120 °C, and vigorously stirred under nitrogen protection for 3 days. The reaction mixture back at room temperature was dispersed in water and then centrifuged at 20 000g to retain the supernatant. It was dialyzed in a membrane tubing (cutoff molecular weight ~ 500) against fresh water to remove unreacted EDA and other small molecular species to obtain EDA-CDots as an aqueous solution. The characterization by using NMR, microscopy, and optical spectroscopy techniques confirmed the structure and properties of EDA-CDots as previously reported.^{37,52}

4.2 Propagation of bacteriophage MS2 and the plaque assay

E. coli C 3000 (ATCC 15597) and *E. coli* bacteriophage MS2 (ATCC 15597-B1) were purchased from American Type Culture Collection (ATCC, Manassas, VA). *E. coli* C-3000 cells were cultured at 37 °C in tryptic soy broth (TSB) with supplements of 0.1% glucose, 0.001% thiamine, and 5.52 mM CaCl₂. The supplements were prepared and sterilized by filtration and then added into the sterile TSB media which were cooled to ~ 55 °C or below. The overnight grown *E. coli* cells in TSB were diluted 100 times with fresh TSB, and then grown to the optical density (OD) value of ~ 0.5 at the wavelength of 600 nm for further uses to propagate MS2 phages.

The propagation of bacteriophage MS2 were performed using the double agar layer method (DLA).^{71,72} The double agar layer plates were prepared using tryptic soy agar (TSA) plates. Briefly, for the top layer, the stock of MS2 phages were serially diluted with TSB. Then 100 μ L diluted MS2 phages (approximately 10^8 PFU per mL) and 20 μ L freshly grown *E. coli* cells (OD value of ~ 0.5 at the wavelength of 595 nm) and were mixed with 3 mL melted TSA which contained 0.7% low-melting agarose pre-warmed at 46 °C. The mixtures were poured onto the plates containing the bottom layer of TSA which were pre-warmed at 37 °C. Once the top TSA solidified, the plates were incubated at 37 °C overnight. To harvest the MS2 phases, the plates with well lysis were flooded with 3 mL TSB and the top TSA was scraped off into a 50 mL sterile tube. TSB was added to the volume of 40 mL, followed by adding 0.2 g EDTA and 0.03 g lysozyme. The tube was incubated in a shaker at 37 °C for 2 h, and then centrifuged at 8000g for 10 min. The supernatant containing the



bacteriophage was filtered (pore size 0.22 μm) and stored at $-80\text{ }^\circ\text{C}$ as a stock of MS2 for further experimental uses.

The titer of MS2 was determined by the plaque assay using the double layer TSA plates and *E. coli* C3000 as the host. Briefly, an aliquot of 1 mL serially diluted MS2 sample was added to 7 mL of molten agarose overlay along with 200 μL of freshly grown log phase *E. coli* cells. The agarose containing the *E. coli* and the MS2 phages were poured onto prewarmed TSA plates. Allowed to solidify for 20–30 min, the plates were then flipped and incubated overnight, and the plaques formed on the plates were counted the next day, and calculated into plaque forming unit per mL (PFU per mL).

4.3 CDots treatment to MS2 phages

To test the antiviral activity of CDots on MS2 phages, the phages from stock were first thawed on ice. Aliquots of 20 μL of MS2 phages at the titer of $\sim 10^8$ PFU per mL were placed in the wells of 96-well plates, various volumes of 0.75 mg mL^{-1} CDots stock were added to reach the final concentration ranging from 0 to 50 $\mu\text{g mL}^{-1}$, then $2/3 \times \text{TSB}$ and $\text{DI-H}_2\text{O}$ were added to make the final volume of 150 μL in each well. The plates were then placed on a shaker with constantly shaking under the visible light (~ 10 cm above the plate) or in the dark (the plate wrapped with aluminum foil) with desired treatment time. Samples without CDots with light or in dark were used as controls. After the treatments, the MS2 phage samples were then serially diluted and the plaque assay was performed. The reduction in viral titers due to CDots treatment compared to the controls was used to evaluate the effect of CDots on the infectivity MS2 phages.

4.4 Growth curves of *E. coli* C3000 cells infected with untreated and CDots-treated MS2 phage

The growth curves of *E. coli* C3000 cells infected with CDots-treated and untreated MS2 phages were also examined for examining the amplification of MS2 phages in host *E. coli* cells in broth. MS2 phages were treated with EDA-CDots at the concentration of 0, 5, 20, 30, 40, and 50 $\mu\text{g mL}^{-1}$ in the 96-well plate under the light for 1 h or 16 h. Considering the possible influences of CDots on the growth curves of host bacteria, the TSB with CDots but without phages were used as control samples. The treated MS2 phages and the control samples were added into the wells containing *E. coli* C3000 cells in TSB at the OD_{595} of ~ 0.05 . The bacterial growths were monitored by OD at 595 nm every 20 min for 6 h. A decline in OD on the growth curve of *E. coli* cells indicated *E. coli* cell lysis by MS2 phage infection.

4.5 Effect of ROS scavengers on CDots' antiviral activity

Tert-butanol and *L*-histidine were used in the experiments as ROS scavengers to scavenge hydroxyl radicals ($\cdot\text{OH}$) and singlet oxygen ($^1\text{O}_2$), respectively. Stock solutions of *tert*-butanol and *L*-histidine were prepared in DI H_2O . In the treatments of MS2 phages with EDA-CDots (20 $\mu\text{g mL}^{-1}$), various concentrations of *tert*-butanol or *L*-histidine at 30 mM, 60 mM, and 90 mM were added to the MS2 samples, appropriate controls including

samples treated with scavenger only, CDots only, and untreated samples were included. The samples were mixed and kept in the dark for 15 min and then were placed on the shaker under visible light for 2 h. The titer of MS2 in all samples were determined using the plaque assay as described above.

4.6 Transmission electron microscopy (TEM)

The integrity of MS2 phages after EDA-CDots treatment was examined by TEM imaging. MS2 phage samples were treated with 50 $\mu\text{g mL}^{-1}$ CDots under the visible light or in the dark for 2 h. The samples were loaded on copper grids and air dried, followed by a negative staining step by placing uranyl acetate on the grids for 30 s. The excess stain was wicked with filter paper and the grids were air dried. TEM imaging of MS2 samples was performed in the Microscopy Services Laboratory Core Facility at the University of North Carolina (UNC) at Chapel Hill School of Medicine, using a JEOL JEM-1230 transmission electron microscope operating at 80 kV (JEOL USA, Peabody, MA), equipped with the Gatan Orius SC1000 CCD camera and Gatan Microscopy Suite 3.0 software (Gatan, Inc., Pleasanton, CA).

4.7 Assay of protein oxidation in MS2 phages

Protein oxidation in MS2 phages after CDots treatment was assessed by the content of carbonyl protein per gram of total protein. MS2 phages were treated with 20 and 50 $\mu\text{g mL}^{-1}$ EDA-CDots under visible light for 2 h. The total protein content was determined by the use of Pierce Rapid Gold BCA Protein Assay Kit (Thermo Scientific, Rockford, IL). The carbonyl content was determined using Protein Carbonyl Content Assay Kit (Bio-Vision, Inc., Milpitas, CA). The carbonyl content was calculated according to the instruction from the manufacturer as follows:

$$C = [(\text{OD}_{375 \text{ nm}})/6.364] \times 100 \text{ nmol per well}$$

$$\text{Carbonyl content: nmol carbonyl per mg protein} = (C/P) \times 1000 \times D$$

where: C is the nmol carbonyl in the sample; 6.364 is the mM extinction coefficient using the provided 96 well plate; P is the protein from standard curve $\times 20$ μg per well; D is the dilution or concentration factor if applied to a sample; 1000 is the factor to convert μg to mg.

4.8 SDS-PAGE for analysis of MS2 capsid protein

SDS-PAGE gel was used to examine the effect of CDots on the MS2 capsid protein integrity. MS2 phages were treated with CDots at the concentrations of 0, 10, 20, and 50 $\mu\text{g mL}^{-1}$ under visible light for 16 h. The freshly thawed phages without CDots treatments were also used for comparison. Aliquots of 5 μL of the treated and untreated MS2 samples were mixed with 1 μL NuPAGE Sample Reducing Agent (Invitrogen, Carlsbad, CA), 2.5 μL LDS-Sample Buffer (Invitrogen, Carlsbad, CA), and 1.5 μL DI- H_2O , and sat at $72\text{ }^\circ\text{C}$ in a hot block for 8 min. The samples were then analyzed using the SDS-PAGE gels, which were run in NuPAGE MOPS SDS Running Buffer (Life Technologies, Carlsbad, CA) at 200 V for 50 min. The gels were fixed with a 50%



methanol and 7% acetic acid solution, rinsed with DI-H₂O, and then stained with GelCode Blue Stain Reagent (Thermo Scientific, Rockford, IL) for 1 h with gently shaking. After de-stain with DI-H₂O for 2 h, the gels were imaged by using the Li-Cor Odyssey imager (Li-Cor, Inc., Lincoln, NE).

4.9 Agarose gel for viral genomic RNA analysis

For examine the effect of CDots treatment on the genomic RNA integrity, MS2 phages were treated with EDA-CDots at concentrations of 0, 5, 10, and 20 $\mu\text{g mL}^{-1}$ under visible light for 16 h. Genomic RNA was extracted from 250 μL samples by using Invitrogen PureLink™ Viral RNA/DNA Mini Kit (Thermo Fisher Scientific, Carlsbad, CA) according to the instructions from the manufacturer. The extracted RNA was eluted with 25 μL of RNase-/DNase-free water and was loaded at the volume of 5 μL in a 1.2% agarose gel for electrophoresis. The gel was stained for 20 min with ethidium bromide solution in the dark, and the image was captured under UV light using Kodak Image Station 4000R Pro (Carestream Health, Inc., Rochester, NY).

Authors' contributions

XD: methodology, investigation, data curation, writing-original draft. RE: methodology, investigation. FY: methodology, investigation. YT: methodology, data curation, resources, funding acquisition. PW: methodology, investigation, data curation. SYP: conceptualization, resource, supervision, writing – reviewing and editing, finding acquisition. LY: conceptualization, resources, supervision, writing – reviewing and editing, funding acquisition.

Conflicts of interest

The authors declare no conflict of interest.

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References

- R. Lozano, M. Naghavi, K. Foreman, S. Lim, K. Shibuya, V. Abovans, J. Abraham, T. Adair, R. Aggarwal, S. Y. Ahn, M. Alvarado, H. R. Anderson, L. M. Anderson, K. G. Andrews, C. Atkinson, L. M. Baddour, S. Barker-Collo, D. H. B. Bartels, M. L. Bell, E. J. Benjamin, D. Bennett, K. Bhalla, B. Bikbov, A. Bin Abdulhak, G. Birbeck, F. Blyth, I. Bolliger, S. Boufous, C. Bucello, M. Burch, P. Burney, J. Carapetis, H. L. Chen, D. Chou, S. S. Chugh, L. E. Coffeng, S. D. Colan, S. Colquhoun, K. E. Colson, J. Condon, M. D. Connor, L. T. Cooper, M. Corriere, M. Cortinovis, K. C. de Vaccaro, W. Couser, B. C. Cowie, M. H. Criqui, M. Cross, K. C. Dabhadkar, N. Dahodwala, D. De Leo, L. Degenhardt, A. Delossantos, J. Denenberg, D. C. Des Jarlais, S. D. Dharmaratne, E. R. Dorsey, T. Driscoll, H. Duber, B. Ebel, P. J. Erwin, P. Espindola, M. Ezzati, V. Feigin, A. D. Flaxman, M. H. Forouzanfar, F. G. R. Fowkes, R. Franklin, M. Fransen, M. K. Freeman, S. E. Gabriel, E. Gakidou, F. Gaspari, R. F. Gillum, D. Gonzalez-Medina, Y. A. Halasa, D. Haring, J. E. Harrison, R. Havmoeller, R. J. Hay, B. Hoen, P. J. Hotez, D. Hoy, K. H. Jacobsen, S. L. James, R. Jasrasaria, S. Jayaraman, N. Johns, G. Karthikeyan, N. Kassebaum, A. Keren, J. P. Khoo, L. M. Knowlton, O. Kobusingye, A. Koranteng, R. Krishnamurthi, M. Lipnick, S. E. Lipshultz, S. L. Ohno, J. Mabweijano, M. F. MacIntyre, L. Mallinger, L. March, G. B. Marks, R. Marks, A. Matsumori, R. Matzopoulos, B. M. Mayosi, J. H. McAnulty, M. M. McDermott, J. McGrath, G. A. Mensah, T. R. Merriman, C. Michaud, M. Miller, T. R. Miller, C. Mock, A. O. Mocumbi, A. A. Mokdad, A. Moran, K. Mulholland, M. N. Nair, L. Naldi, K. M. V. Narayan, K. Nasserli, P. Norman, M. O'Donnell, S. B. Omer, K. Ortblad, R. Osborne, D. Ozgediz, B. Pahari, J. D. Pandian, A. P. Rivero, R. P. Padilla, F. Perez-Ruiz, N. Perico, D. Phillips, K. Pierce, C. A. Pope, E. Porrini, F. Pourmalek, M. Raju, D. Ranganathan, J. T. Rehm, D. B. Rein, G. Remuzzi, F. P. Rivara, T. Roberts, F. R. De Leon, L. C. Rosenfeld, L. Rushton, R. L. Sacco, J. A. Salomon, U. Sampson, E. Sanman, D. C. Schwebel, M. Segui-Gomez, D. S. Shepard, D. Singh, J. Singleton, K. Sliwa, E. Smith, A. Steer, J. A. Taylor, B. Thomas, I. M. Tleyjeh, J. A. Towbin, T. Truelsen, E. A. Undurraga, N. Venketasubramanian, L. Vijayakumar, T. Vos, G. R. Wagner, M. R. Wang, W. Z. Wang, K. Watt, M. A. Weinstock, R. Weintraub, J. D. Wilkinson, A. D. Woolf, S. Wulf, P. H. Yeh, P. Yip, A. Zabetian, Z. J. Zheng, A. D. Lopez and C. J. L. Murray, *Lancet*, 2012, **380**, 2095–2128.
- P. Moitra, *Bioorg. Chem.*, 2019, **86**, 15–27.
- P. Moitra, M. Alafeef, K. Dighe, M. B. Frieman and D. Pan, *ACS Nano*, 2020, **14**, 7617–7627.
- S. Park, H. H. Park, S. Y. Kim, S. J. Kim, K. Woo and G. Ko, *Appl. Environ. Microbiol.*, 2014, **80**, 2343–2350.
- N. Shionoiri, T. Sato, Y. Fujimori, T. Nakayama, M. Nemoto, T. Matsunaga and T. Tanaka, *J. Biosci. Bioeng.*, 2012, **113**, 580–586.
- G. W. Park, M. Cho, E. L. Cates, D. Lee, B. T. Oh, J. Vinje and J. H. Kim, *J. Photochem. Photobiol., B*, 2014, **140**, 315–320.
- A. Ingle, A. Gade, S. Pierrat, C. Sonnichsen and M. Rai, *Curr. Nanosci.*, 2008, **4**, 141–144.
- T. Cromeans, G. W. Park, V. Costantini, D. Lee, Q. Wang, T. Farkas, A. Lee and J. Vinje, *Appl. Environ. Microbiol.*, 2014, **80**, 5743–5751.
- M. R. Hamblin and S. B. Brown, in *Advances in Photodynamic Therapy: Basic, Translational and Clinical*, ed. M. R. Hamblin and P. Mroz, Artech House, Inc., Norwood, MA, 2008, ch. 19.
- G. V. Anehosur, R. D. Kulkarni, M. G. Naik and R. K. Nadiger, *Journal of Gerontology and Geriatric Research*, 2012, **1**, 1–8.
- Y. L. Chen, Y. S. Chen, H. Chan, Y. H. Tseng, S. R. Yang, H. Y. Tsai, H. Y. Liu, D. S. Sun and H. H. Chang, *PLoS One*, 2012, **7**, e31212.



- 12 C. L. Cheng, D. S. Sun, W. C. Chu, Y. H. Tseng, H. C. Ho, J. B. Wang, P. H. Chung, J. H. Chen, P. J. Tsai, N. T. Lin, M. S. Yu and H. H. Chang, *J. Biomed. Sci.*, 2009, **16**, 7.
- 13 A. J. T. Naik, S. Ismail, C. Kay, M. Wilson and I. P. Parkin, *Mater. Chem. Phys.*, 2011, **129**, 446–450.
- 14 S. Noimark, E. Allan and I. P. Parkin, *Chem. Sci.*, 2014, **5**, 2216–2223.
- 15 S. Perni, C. Piccirillo, J. Pratten, P. Prokopovich, W. Chrzanowski, I. P. Parkin and M. Wilson, *Biomaterials*, 2009, **30**, 89–93.
- 16 L. Y. Huang, M. Terakawa, T. Zhiyentayev, Y. Y. Huang, Y. Sawayama, A. Jahnke, G. P. Tegos, T. Wharton and M. R. Hamblin, *Nanomedicine*, 2010, **6**, 442–452.
- 17 C. M. Courtney, S. M. Goodman, J. A. McDaniel, N. E. Madinger, A. Chatterjee and P. Nagpal, *Nat. Mater.*, 2016, **15**, 529–534.
- 18 C. M. Courtney, S. M. Goodman, T. A. Nagy, M. Levy, P. Bhusal, N. E. Madinger, C. S. Detweiler, P. Nagpal and A. Chatterjee, *Sci. Adv.*, 2017, **3**, e1701776.
- 19 Z. Lu, C. M. Li, H. Bao, Y. Qiao, Y. Toh and X. Yang, *Langmuir*, 2008, **24**, 5445–5452.
- 20 L. Cao, M. J. Meziani, S. Sahu and Y. P. Sun, *Acc. Chem. Res.*, 2013, **46**, 171–180.
- 21 L. Cao, X. Wang, M. J. Meziani, F. S. Lu, H. F. Wang, P. J. G. Luo, Y. Lin, B. A. Harruff, L. M. Veca, D. Murray, S. Y. Xie and Y. P. Sun, *J. Am. Chem. Soc.*, 2007, **129**, 11318–11319.
- 22 K. A. S. Fernando, S. Sahu, Y. M. Liu, W. K. Lewis, E. A. Guliants, A. Jafariyan, P. Wang, C. E. Bunker and Y. P. Sun, *ACS Appl. Mater. Interfaces*, 2015, **7**, 8363–8376.
- 23 G. E. LeCroy, S. T. Yang, F. Yang, Y. M. Liu, K. A. S. Fernando, C. E. Bunker, Y. Hu, P. G. Luo and Y. P. Sun, *Coord. Chem. Rev.*, 2016, **320**, 66–81.
- 24 S. Y. Lim, W. Shen and Z. Q. Gao, *Chem. Soc. Rev.*, 2015, **44**, 362–381.
- 25 P. J. G. Luo, S. Sahu, S. T. Yang, S. K. Sonkar, J. P. Wang, H. F. Wang, G. E. LeCroy, L. Cao and Y. P. Sun, *J. Mater. Chem. B*, 2013, **1**, 2116–2127.
- 26 P. J. G. Luo, F. Yang, S. T. Yang, S. K. Sonkar, L. J. Yang, J. J. Broglie, Y. Liu and Y. P. Sun, *RSC Adv.*, 2014, **4**, 10791–10807.
- 27 P. Miao, K. Han, Y. G. Tang, B. D. Wang, T. Lin and W. B. Cheng, *Nanoscale*, 2015, **7**, 1586–1595.
- 28 Y. P. Sun, *US Pat.*, #7829772, 2010 and #8932877, 2015.
- 29 Y. P. Sun, B. Zhou, Y. Lin, W. Wang, K. A. S. Fernando, P. Pathak, M. J. Meziani, B. A. Harruff, X. Wang, H. F. Wang, P. J. G. Luo, H. Yang, M. E. Kose, B. L. Chen, L. M. Veca and S. Y. Xie, *J. Am. Chem. Soc.*, 2006, **128**, 7756–7757.
- 30 Y. F. Wang and A. G. Hu, *J. Mater. Chem. C*, 2014, **2**, 6921–6939.
- 31 A. D. Zhao, Z. W. Chen, C. Q. Zhao, N. Gao, J. S. Ren and X. G. Qu, *Carbon*, 2015, **85**, 309–327.
- 32 X. L. Dong, W. X. Liang, M. J. Meziani, Y. P. Sun and L. J. Yang, *Theranostics*, 2020, **10**, 671–686.
- 33 M. J. Meziani, X. Dong, L. Zhu, L. P. Jones, G. E. LeCroy and F. Yang, *ACS Appl. Mater. Interfaces*, 2016, **8**, 10761–10766.
- 34 D. Abu Rabe, O. Mohammed, X. Dong, A. Patel, C. Overton, Y. Tang, S. Kathariou, Y. Sun and L. Yang, *Materials Advances*, 2020, **1**, 321–325.
- 35 Y. P. Sun, *Carbon Dots – Exploring Carbon at Zero-Dimension*, Springer International Publishing, 2020.
- 36 D. I. Abu Rabe, M. M. Al Awak, F. Yang, P. A. Okonjo, X. Dong, L. R. Teisl, P. Wang, Y. Tang, N. Pan, Y. P. Sun and L. Yang, *Int. J. Nanomed.*, 2019, **14**, 2655–2665.
- 37 G. E. LeCroy, S. K. Sonkar, F. Yang, L. M. Veca, P. Wang, K. N. Tackett 2nd, J. J. Yu, E. Vasile, H. Qian, Y. Liu, P. G. Luo and Y. P. Sun, *ACS Nano*, 2014, **8**, 4522–4529.
- 38 M. M. Al Awak, P. Wang, S. Y. Wang, Y. A. Tang, Y. P. Sun and L. J. Yang, *RSC Adv.*, 2017, **7**, 30177–30184.
- 39 X. L. Dong, M. M. Moyer, F. Yang, Y. P. Sun and L. J. Yang, *Sci. Rep.*, 2017, **7**, 519.
- 40 G. F. Brooks, K. C. Carroll, J. S. Butel, S. A. Morse and T. A. Mietzner, in *Jawetz, Melnick, & Adelberg's Medical Microbiology*, The McGraw-Hill Companies, Inc., McGraw Hill, 26th edn, 2013.
- 41 C. Rye, R. Wise, V. Jurukovsk, J. DeSaix, J. Choi and Y. Avissar, *General Biology*, [https://bio.libretexts.org/Bookshelves/Introductory_and_General_Biology/Book%3A_General_Biology_\(Boundless\)/](https://bio.libretexts.org/Bookshelves/Introductory_and_General_Biology/Book%3A_General_Biology_(Boundless)/).
- 42 J. A. Thorley, J. A. McKeating and J. Z. Rappoport, *Protoplasma*, 2010, **244**, 15–24.
- 43 H. Sobhy, *Arch. Virol.*, 2017, **162**, 3567–3585.
- 44 J. P. Buchmann and E. C. Holmes, *Microbiol. Mol. Biol. Rev.*, 2015, **79**, 403–418.
- 45 J. Callanan, S. R. Stockdale, A. Shkoporov, L. A. Draper, R. P. Ross and C. Hill, *Viruses*, 2018, **10**, 388.
- 46 Z. Y. Lou, Y. N. Sun and Z. H. Rao, *Trends Pharmacol. Sci.*, 2014, **35**, 86–102.
- 47 E. Addae, X. Dong, E. McCoy, C. Yang, W. Chen and L. Yang, *J. Biol. Eng.*, 2014, **8**, 11.
- 48 J. J. Broglie, B. Alston, C. Yang, L. Ma, A. F. Adcock, W. Chen and L. Yang, *PLoS One*, 2015, **10**, e0141050.
- 49 K. Feng, E. Divers, Y. Ma and J. Li, *Appl. Environ. Microbiol.*, 2011, **77**, 3507–3517.
- 50 E. Araud, E. DiCaprio, Z. Yang, X. Li, F. Lou, J. H. Hughes, H. Chen and J. Li, *Appl. Environ. Microbiol.*, 2015, **81**, 6669–6678.
- 51 L. Costa, M. A. Faustino, M. G. Neves, A. Cunha and A. Almeida, *Viruses*, 2012, **4**, 1034–1074.
- 52 G. E. LeCroy, F. Messina, A. Sciortino, C. E. Bunker, P. Wang, K. A. S. Fernando and Y.-P. Sun, *J. Phys. Chem. C*, 2017, **121**, 28180–28186.
- 53 X. Dong, A. E. Bond, N. Pan, M. Coleman, Y. Tang, Y. P. Sun and L. Yang, *Int. J. Nanomed.*, 2018, **13**, 8025–8035.
- 54 X. L. Dong, M. Al Awak, N. Tomlinson, Y. G. Tang, Y. P. Sun and L. J. Yang, *PLoS One*, 2017, **12**, e0185324.
- 55 H. Majiya, O. O. Adeyemi, N. J. Stonehouse and P. Millner, *J. Photochem. Photobiol., B*, 2018, **178**, 404–411.
- 56 M. D. Driessen, S. Mues, A. Vennemann, B. Hellack, A. Bannuscher, V. Vimalakanthan, C. Riebeling, R. Ossig, M. Wiemann and J. Schneckeburger, *Part. Fibre Toxicol.*, 2015, **12**, 36.



- 57 I. Dalle-Donne, R. Rossi, D. Giustarini, A. Milzani and R. Colombo, *Clin. Chim. Acta*, 2003, **329**, 23–38.
- 58 E. Shacter, J. A. Williams, M. Lim and R. L. Levine, *Free Radicals Biol. Med.*, 1994, **17**, 429–437.
- 59 E. Shacter, *Drug Metab. Rev.*, 2000, **32**, 307–326.
- 60 E. R. Stadtman, *Free Radical Res.*, 2006, **40**, 1250–1258.
- 61 W. Fiers, R. Contreras, F. Duerinck, G. Haegeman, D. Iserentant, J. Merregaert, W. Min Jou, F. Molemans, A. Raeymaekers, A. Van den Berghe, G. Volckaert and M. Ysebaert, *Nature*, 1976, **260**, 500–507.
- 62 J. van Duin and N. Tsareva, in *The Bacteriophages*, ed. R. L. Calendar, Oxford University Press, Oxford, 2nd edn, 2006, vol. 1, ch. 4, pp. 117–167.
- 63 X. Dong, L. Ge, D. I. Abu Rabe, O. O. Mohammed, P. Wang, Y. Tang, S. Kathariou, L. Yang and Y.-P. Sun, *Carbon*, 2020, **170**, 137–145.
- 64 J.-H. Liu, Y. Wang, G.-H. Yan, F. Yang, H. Gao, Y. Huang, H. Wang, P. Wang, L. Yang, Y. Tang, L. R. Teisl and Y.-P. Sun, *J. Nanosci. Nanotechnol.*, 2019, **19**, 2130–2137.
- 65 Y. Wang, P. Anilkumar, L. Cao, J. H. Liu, P. G. Luo, K. N. Tackett 2nd, S. Sahu, P. Wang, X. Wang and Y. P. Sun, *Exp. Biol. Med.*, 2011, **236**, 1231–1238.
- 66 S. T. Yang, X. Wang, H. Wang, F. Lu, P. G. Luo, L. Cao, M. J. Meziani, J. H. Liu, Y. Liu, M. Chen, Y. Huang and Y. P. Sun, *J. Phys. Chem. C*, 2009, **113**, 18110–18114.
- 67 K. Hara, S. Holland and J. Woo, *J. Exp. Microbiol. Immunol.*, 2004, 62–66.
- 68 K. Ishiyama, K. Nakamura, H. Ikai, T. Kanno, M. Kohno, K. Sasaki and Y. Niwano, *PLoS One*, 2012, e37871.
- 69 F. Yang, G. E. LeCroy, P. Wang, W. Liang, J. Chen, K. A. S. Fernando, C. E. Bunker, H. Qian and Y.-P. Sun, *J. Phys. Chem. C*, 2016, **120**, 25604–25611.
- 70 Y. M. Liu, P. Wang, K. A. S. Fernando, G. E. LeCroy, H. Maimaiti, B. A. Harruff-Miller, W. K. Lewis, C. E. Bunker, Z. L. Hou and Y. P. Sun, *J. Mater. Chem. C*, 2016, **4**, 6967–6974.
- 71 J. You, Y. Zhang and Z. Hu, *Colloids Surf., B*, 2011, **85**, 161–167.
- 72 A. R. Badireddy, E. M. Hotze, S. Chellam, P. Alvarez and M. R. Wiesner, *Environ. Sci. Technol.*, 2007, **41**, 6627–6632.

