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***In vitro* kinetic release study, antimicrobial activity and *in vivo* toxicity profile of a kojic acid ester-based nanoemulsion for topical application**

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Nanoemulsions have emerged as novel vehicles for drug delivery that allow sustained or controlled release for topical application. In this study, kojic acid ester-based nanoemulsion (KAE-NA) was analyzed for *in vitro* permeation evaluation, kinetic release study, *in vitro* antimicrobial activity and *in vivo* toxicity profile on embryonic zebrafish (*Danio rerio*). Based on KAE-NA *in vitro* permeation evaluation, the percentage of permeation was significantly improved from 4.94% at 1 h to 59.64% at 8 h of application. The permeation rate of KAE-NA at 8 h was $4659.50 \mu\text{g cm}^{-2} \text{h}^{-1}$ (initial concentration, $C_0 = 2000 \mu\text{g mL}^{-1}$) with a permeability coefficient (K_p) value of 0.48 cm h^{-1} . The kinetic release analysis showed the Korsmeyer–Peppas model was the best fitted kinetic model with high linearity [$R^2 = 0.9964$]. Antimicrobial activity of KAE-NA was studied against the skin pathogen bacteria *Staphylococcus aureus* ATCC 43300. The results indicated that the inhibition zone size of the KAE-NA ($8.00 \pm 0.0 \text{ mm}$) was slightly bigger than that of its active ingredient, kojic acid ester ($6.5 \pm 0.0 \text{ mm}$). The toxicity profile of KAE-NA on embryonic zebrafish revealed less toxicity with LC_{50} (50% lethal concentration) more than $500 \mu\text{g mL}^{-1}$. The survival rate of the embryonic zebrafish was more than 80% when treated at doses ranging from 7.81 – $250 \mu\text{g mL}^{-1}$ and showed normal development throughout the experiment without any observed deformation. Hence, KAE-NA proved to be less toxic on the embryonic zebrafish.

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1. Introduction

Applying the new generation of nanotechnology for topical delivery systems offers numerous advantages in targeting active therapeutic ingredients to the desired site. This includes greater skin retention, improvement in the stability of cosmetic ingredients and long-lasting effect of sustained release of active ingredients that can be achieved. Nanotechnology-based novel carriers of cosmetics include nanoemulsions, nanocapsules, liposomes, niosomes, nanocrystals, solid lipid nanoparticles, carbon nanotubes, fullerenes and dendrimers.¹ Basically, nanoemulsions are a kinetically stable system (liquids in the nanoemulsion mixture will be separated very slowly) with droplet size ranging from 20 to 200 nm. Besides, these nanoemulsions performed as a delivery system in transporting various functional lipophilic compounds for drugs, antioxidants, flavors, antimicrobial agents and nutraceuticals products. To date, they are considered to be the most advanced

nanoparticulate system for cosmetic applications.² Research done by Ahmad N. and co-workers, 2019 were successfully prepared a novel curcumin nanoemulsion by ultrasonication with $93.64 \pm 6.48 \text{ nm}$ droplet size for wound healing treatment. The *in vitro* drug release using a pre-treated dialysis membrane was $79.64 \pm 5.34\%$ after 24 h application.³ Besides, copper peptide in a nanoemulsion system (120.7 nm) was developed to prevent aging. The permeability test results using Franz diffusion cell on cellulose acetate membrane was $21.89 \pm 0.53\%$ after 8 h application. Results from this permeation study show that nanoemulsion has the potential to be used as a carrier vehicle for the delivery of copper peptide.⁴

The advantages of nanoemulsions over micro- and macro-counterparts are their high surface area allowing effective transport properties (the smaller the size of the emulsion, the higher the stability and better capability to carry active ingredients) and they do not have inherent creaming like macroemulsions making the longer shelf life of the products. In terms of delivering lipophilic compounds, nanoemulsions are superior to liposomes due to their lipophilic interior.⁵ In addition, a topical delivery system is considered to be the best route for transferring active ingredients compare to an oral intravenous pathway. This type of route enhances the bioavailability of drugs, a pain-free method, avoidance of the first-pass effect, patient compliance, steady-state plasma drug pass effect and ease of application.⁶

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Considering these advantages of nanoemulsions as a carrier, kojic acid ester-based nanoemulsion (KAE-NA) can be used for the topical delivery system. Previous research reported kojic acid ester has therapeutic uses for melasma, antioxidant and preservative in foods, antibiotic, chemical intermediate, metal chelate, pesticide, and antimicrobial agents. In addition, kojic acid ester has a lower toxicity value and more stable in terms of storage stability, compatibility and oil-solubility compare to kojic acid itself.^{7,8} Marta and Jorge (2005) claimed that the composition for topical use of the present invention of kojic acid ester may be prepared in various forms such as emulsion, cream, and powder which can exhibit the melanin synthesis-inhibiting activity, whitening effect, and anti-suntan effect.⁹ Since emulsion, cream and powder are in micro and macro size, thus kojic acid ester was further study using a nanotechnology system (nanoemulsion) to give optimum stability and compatibility. In spite of that, the safety and efficacy of products are another part of the assessment that should be considered before providing to consumers. The latest research reported that the use of the zebrafish model (*Danio rerio*) has been accepted and validated for the *in vivo* toxicity and teratogenicity testing of colloidal nano-drug delivery systems.¹⁰

Its advantages as an intermediate model prior to *in vivo* studies in mammals are the high genetic homology with humans, the molecular and physiological similarities with mammals, the *ex utero* growth, the ability to perform high-throughput screenings due to the high number of fertilized eggs after spawning and its fast development that allows studying the establishment of the principal organ systems during the first week.¹¹ Unpredicted human safety events in clinical trials for new drugs are costly in terms of human health and money. The drug discovery industry attempts to minimize those events with diligent preclinical safety testing. Current standard practices are good at preventing toxic compounds from being tested in the clinic; however, false-negative preclinical toxicity results are still a reality. Therefore, the continual improvement must be pursued in the preclinical realm. Higher-quality therapies can be brought forward with more information about potential toxicities and associated mechanisms. The embryonic zebrafish model is a bridge between *in vitro* assays and mammalian *in vivo* studies. This model is powerful in its breadth of application and tractability for research. In the past two decades, our understanding of disease biology and drug toxicity has grown significantly owing to thousands of studies on this tiny vertebrate.¹²

The zebrafish embryo model is already accepted as a validated alternative assay to assess fish acute toxicity (OECD, No. 236)¹³ and currently, the zebrafish embryo is also being explored as a potential replacement for one of the regulatory *in vivo* mammalian embryo fetal developmental toxicity studies in view of the upcoming third revision of the ICH S5 guideline on detection of the toxicity to reproduction for human pharmaceuticals.¹⁴ Moreover, the zebrafish model allows evaluating parameters such as lethal dose, acute toxicity, teratogenicity, and specific-organ toxicity, in particular of the heart and the nervous system.^{15,16} To add, the transparent body of zebrafish allows us to study the biodistribution of actives and drug

delivery systems by diverse imaging techniques.¹⁷ This report focused on the safety and efficacy of kojic acid ester-based nanoemulsion significant for topical application.

2. Material and methods

2.1 Materials

Kojic acid ester-based nanoemulsion (110.01 ± 0.14 nm) was prepared using the method described by Syed *et al.*, 2018.¹⁸ Roughly, the nanoemulsion was prepared under high and low energy emulsification technique consisting of kojic acid ester (10% w/w), Tween 80 (3.19% w/w), castor oil and lemon oil (3.74% w/w), xanthan gum (0.70% w/w) and deionized water (81.68% w/w). Synthetic membrane (cellulose acetate) was purchased at Advantec, Japan. Phosphate-buffer saline (PBS) was purchased at Sigma Aldrich. Embryonic zebrafish (*Danio rerio*) kit was obtained from Danio Assay Laboratories Sdn. Bhd, Universiti Putra Malaysia, Malaysia.

2.2 *In vitro* permeation study

The permeation and percentage release studies of KAE-NA were performed across cellulose acetate (Advantec, Japan) using Franz diffusion cells (Perme Gear, Hellertown, PA, USA) and analyzed using UV-visible spectrophotometer (Shimadzu 1650PC, Japan) at $\lambda_{\text{max}} = 306$ nm. The Franz diffusion cells consist of two main compartments which were the donor and receptor medium where a cellulose acetate membrane (13 mm) was placed in between them. Both compartments were clipped tightly to avoid leaking of the sample. The area for the diffusion of nanoemulsion between the media was 0.64 cm^2 . The kojic acid ester based-nanoemulsion ($2000 \mu\text{g mL}^{-1}$) was placed on top of the donor's compartment whereas the receptor compartment contained phosphate-buffered saline (PBS) (5 mL) with pH 7.4 was maintained at 37°C and stirred at speed 600 rpm. Every hour, 3 mL was taken out from the receptor compartment and another 3 mL of PBS was replaced into the same compartment. The total volume of 5 mL PBS was maintained inside the receptor compartment throughout the process. A total of nine samples were collected at time $t = 0, 1, 2, 3, 4, 5, 6, 7$ and 8 h. The 8 h determination indicated nanoemulsion needs to be applied 2 to 4 times daily (or about 4 to 8 h) for topical application.¹⁹ The process was carried out in triplicate and samples were analyzed using a UV-visible spectrophotometer with respect to the percentage of cumulative kojic acid ester detected in the receptor compartment.

2.3 Kinetic release study

The kinetic measurement of KAE-NA was determined by the mathematical models; zero-order (cumulative amount of drug release against time, eqn (1)), first-order (log cumulative amount of drug remaining against time, eqn (2)), Higuchi (cumulative percentage of drug release against the square root of time, eqn (3)), Hixson–Crowell (cube root cumulative amount of drug remaining against time, eqn (4)) and Korsmeyer–Peppas (log cumulative percentage of drug release against log time, eqn (5)). The best model fitted the release data was evaluated based



on the coefficient determination (R^2) obtained from the plotted graph. The models were constructed based on the model's theoretical equation.

$$M_t = M_0 + K_0 t \quad (1)$$

$$\log M_t = \log M_0 + K_1 t / 2.303 \quad (2)$$

$$Q = K_H \times t^{1/2} \quad (3)$$

$$M_0^{1/3} - M_t^{1/3} = K_{HC} t \quad (4)$$

$$M_t / M_\infty = K_{kp} t^n \quad (5)$$

where M_0 is the initial amount of kojic acid ester in dissolution media, M_t is the amount of kojic acid ester released in time t , M_∞ is the amount of drug released after time ∞ , K_0 , K_1 , K_H , K_{HC} and K_{kp} are the release rate constants, Q is the cumulative amount of kojic acid ester released in time t per unit area, fraction of kojic acid ester release over time, n is the release exponent and t is the time.

2.4 In vitro antimicrobial activity

In vitro antimicrobial activity of kojic acid ester-based nano-emulsion and kojic acid ester against *Staphylococcus aureus* ATCC 43300 were studied by means of agar diffusion method. The test was carried out using 6 mm diameter of a paper disc containing antibiotics placed onto a plate for microbe's growth. The microbe culture was standardized to 0.5 McFarland standards (Mueller Hinton agar pH 7.3) with approximately 10^8 cells. *Streptomycin* standard was used for each bacterium. Plates were next inverted and incubated at 37 °C for 24 h. After incubation, each plate was examined. The diameters of the zones of complete inhibition were measured to the nearest whole millimeter (mm) using sliding calipers. The clear zone formation (inhibition zone) around the disc indicated antimicrobial activity. Positive (*Streptomycin*) and negative (deionized water) controls were performed to verify the testing conditions. All tested samples were carried out triplicate.

2.5 In vivo toxicology profile on embryonic zebrafish (*Danio rerio*)

The *in vivo* toxicity profile on embryonic zebrafish (*Danio rerio*) was performed in accordance with the guidelines for care and use of Animal Biochemistry & Biotechnology Laboratory, Faculty of Biotechnology & Biomolecular Sciences, Universiti Putra Malaysia and approved by the Institutional Animal Care and Use Committee, Universiti Putra Malaysia (UPM/IACUC/AUP No. R059/2018).

The collected synchronized embryonic zebrafish were arrayed by pipette into a 96-well plate, one embryo per well with a 200 mL embryo medium. The prepared inhibitor solutions (in 1% DMSO) were added to the embryo medium from 0 to 120 hpf (hours post-fertilization, total 120 h exposure). Stereomicroscope was employed for observing the effects on the pigmentation of zebrafish. Phenotype based evaluation of the body pigmentation was performed at 120 hpf. Embryos were deteriorated by forceps, anesthetized in tricaine methanesulfonate solution (Sigma-

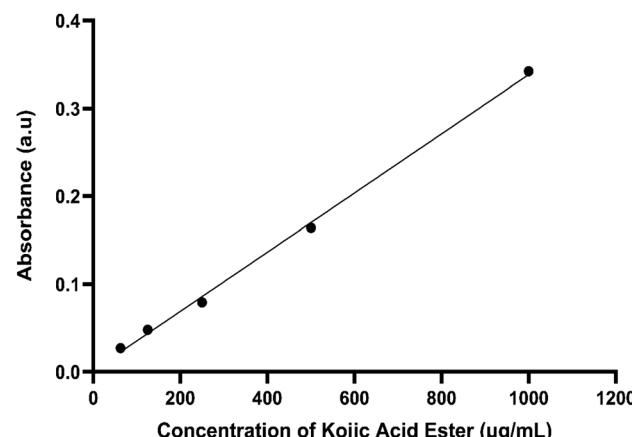


Fig. 1 Calibration curve of kojic acid ester ($\mu\text{g mL}^{-1}$).

Aldrich), mounted in 1% methylcellulose on a depression slide (Aquatic Eco-Systems, Apopka, FL, USA) and photographed under the stereomicroscope Z16 (Leica Microsystems, Ernst-Leitz-Strasse, Germany) for observation. Images capturing and pixel measurement analyses were carried out.

2.6 Statistical analysis

Statistical analysis was carried out using GraphPad Prism 8.0.2. Each experiment was performed in triplicate and the data were expressed as mean \pm standard deviation.

3. Results and discussion

3.1 In vitro permeation evaluation

Permeation studies aimed to achieve good skin penetration by assessing the relationship between the skin, active ingredient and formulation. In this study, cellulose acetate membrane was used in Franz cell diffusion to stimulate the skin instead of

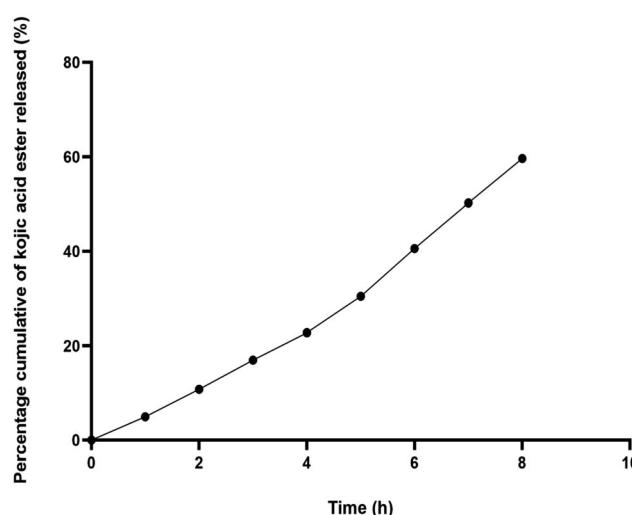


Fig. 2 *In vitro* permeation study for kojic acid ester based-nano-emulsion (KAE-NA).



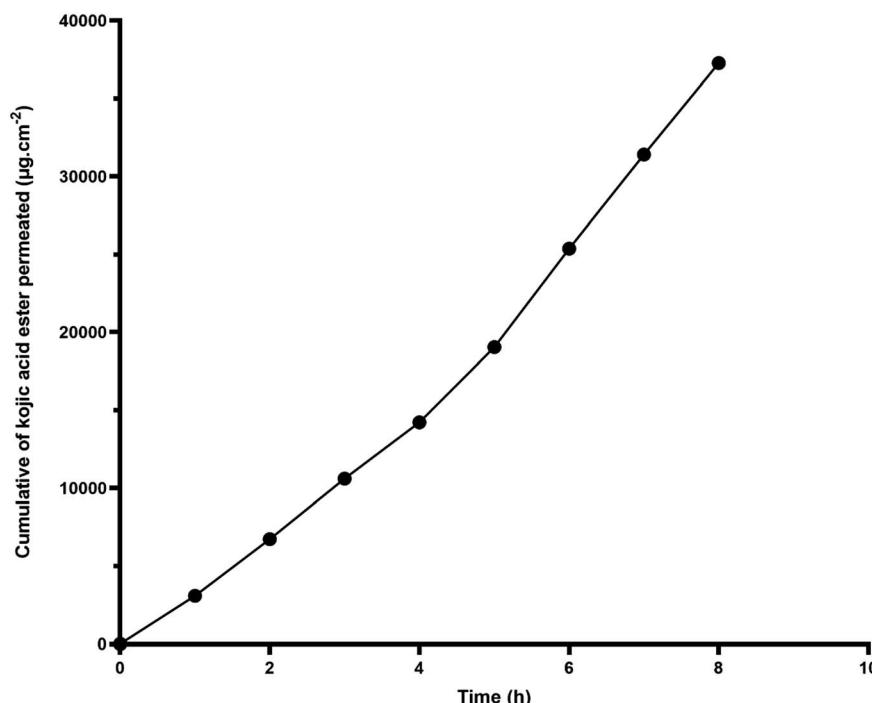


Fig. 3 Cumulative kojic acid ester permeation from nanoemulsion (KAE-NA) through cellulose acetate membrane.

animal skin or human cadaver skin to perform the *in vitro* permeation study.²⁰ As reported by Ng. *et al.*, (2010), the porous synthetic membrane-like cellulose acetate membrane was suggested by the Food and Drug Administration (FDA) for topical assessment due to the non-rate limiting barrier.²¹ Usually, this assessment is used excised human or animal skin but when biological skin is not readily available, synthetic membranes are employed. The ultimate function of synthetic membrane employed in Franz cell drug diffusion studies are for simulation of the skin^{22,23} and quality control.²⁴ The absorption of the active matter was indicated by transporting the active matter through the cellulose acetate membrane.²⁵ Shah *et al.*, 1989 from FDA used different microporous membranes, namely cellulose acetate, cellulose and polysulfone of similar pore sizes and thicknesses to examine the permeation of hydrocortisone (HC) from two commercial creams. They found that the HC flux was consistent irrespective of the types of synthetic membrane.²⁶

Fig. 1 shows the calibration curve of kojic acid ester. The calibration showed excellent linearity over the concentration range from 62.5 to 1000 $\mu\text{g mL}^{-1}$. The mean linear regression equation of the calibration curve is $y = 0.0003x + 0.0015$, with a correlation coefficient of 0.998. These results were used to find

Table 1 The permeation parameters of kojic acid ester based-nanoemulsion

Sample	Flux at 8 h ($\mu\text{g cm}^{-2} \text{h}^{-1}$)	Permeated amount at 8 h (%)	Permeation coefficient, K_p (cm h^{-1})
Kojic acid ester based-nanoemulsion	4659.50	59.64	0.48

the concentration of kojic acid ester in the receptor cells of the Franz diffusion cell and calculate the amount of kojic acid ester transferred through the cellulose acetate membrane.

Fig. 2 shows the percentage of cumulative permeation of KAE-NA. The permeability of the kojic acid ester was significantly improved and the release of the active matter increased from 4.94% at 1 h to 59.64% at 8 h. Based on the release study, it seems that kojic acid ester released faster particularly at the first 4 h and this may due to the smaller droplet size of the nanoemulsion which has a higher surface area, therefore made the initial release faster.

Fig. 3 demonstrates a permeation profile of KAE-NA. The permeation rate and permeation coefficient, K_p are shown in Table 1. Kojic acid ester based-nanoemulsion showed the highest permeation rate of kojic acid ester at $4659.50 \mu\text{g cm}^{-2} \text{h}^{-1}$ with K_p value of 0.48 cm h^{-1} .

Apart from the small droplet size of nanoemulsion, the surfactant used in the system was able to compromise with the barrier function of the cellulose acetate membrane thus, facilitate the passage of kojic acid ester. In addition, according to Tsai *et al.*, (2014) the nanoemulsions could change the surface electrical charge of the drug and enhance its' permeability.

Table 2 The regression coefficient (R^2) of five different kinetic models for kojic acid ester-based nanoemulsion

Kinetic model	Regression coefficient (R^2)
Zero-order	0.9865
First-order	0.8210
Higuchi	0.9390
Hixson–Crowell	0.9009
Korsmeyer–Peppas	0.9964



Based on the results, nanoemulsion can be used as a successful carrier vehicle for the delivery of kojic acid ester.²⁷

3.2 Kinetic release analysis

In order to determine the kinetic release pattern of kojic acid ester-based nanoemulsion, the *in vivo* permeation data were substituted

with five different kinetic models. The release rate was calculated from the slope of appropriate plots and the regression coefficient (R^2) was determined as shown in Table 2. Based on Fig. 4, the kinetic release of kojic acid ester-based nanoemulsion was best fitted with the Korsmeyer–Peppas model with high linearity ($R^2 = 0.9964$). Also, the model Korsmeyer–Peppas equation states the type of diffusion, which was evaluated by release exponent, $n =$

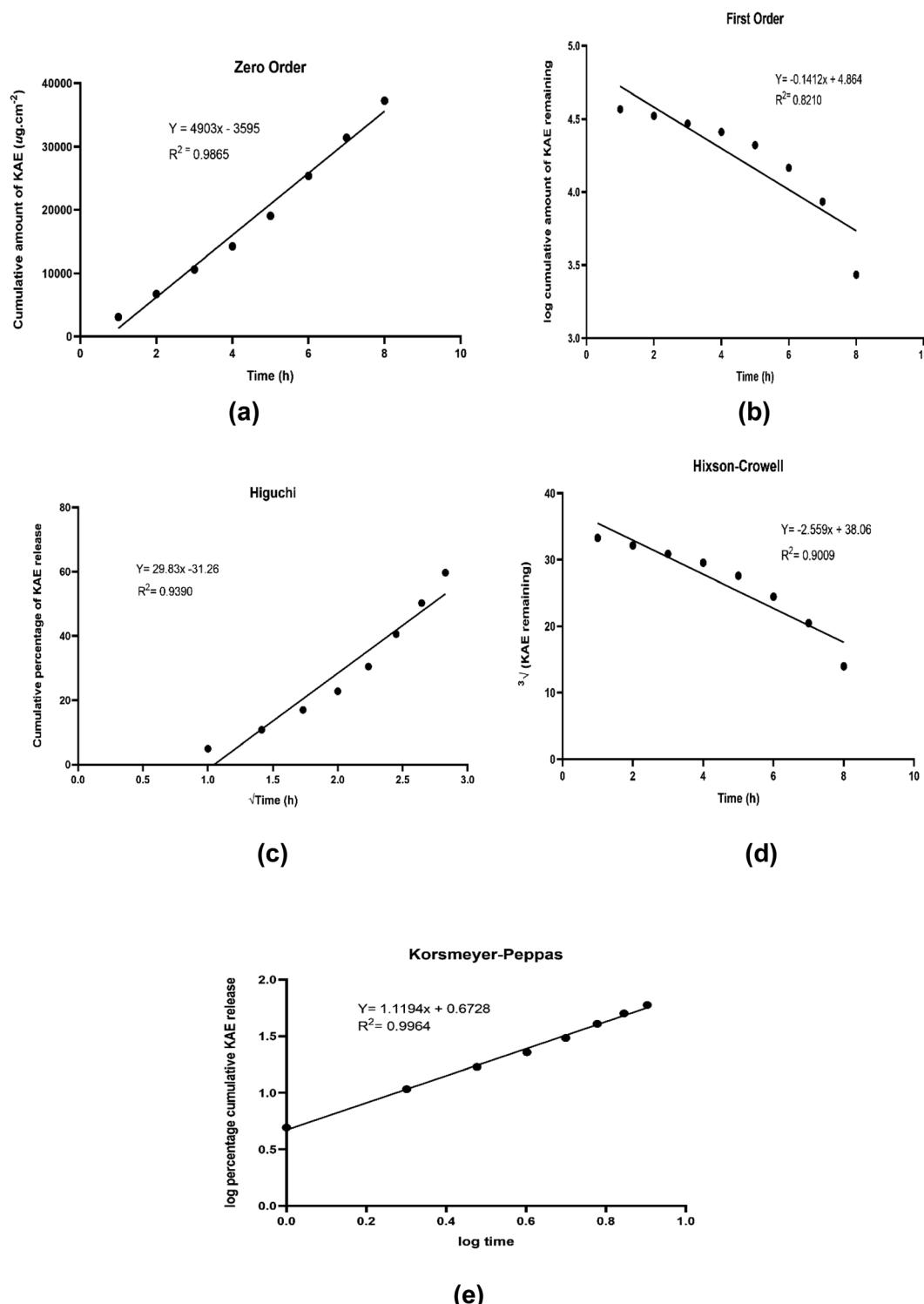


Fig. 4 Graph of kinetic models (a) zero order (b) first-order (c) Higuchi (d) Hixson–Crowell (e) Korsmeyer–Peppas.



Table 3 The inhibition zone size effect of kojic acid ester-based nanoemulsion, kojic acid ester and controls towards *Staphylococcus aureus* ATCC 43300

Samples	Diameter zone inhibition on bacteria <i>Staphylococcus aureus</i> ATCC 43300 (mm)
Kojic acid ester	6.5 ± 0.0
Kojic acid ester-based nanoemulsion	8.0 ± 0.0
Positive control (Streptomycin)	28.0 ± 0.1
Negative control (deionized water)	No inhibition

1.1194 which is higher than 0.89. This implies that the KAE-NA release from the system follows Super case II transport.

3.3 In vitro antimicrobial activity

The *in vitro* antimicrobial activity against the pathogen bacteria *Staphylococcus aureus* ATCC 43300 (*S. aureus*) on kojic acid ester-based nanoemulsion (KAE-NA) and kojic acid ester (KAE) were investigated. Based on the literature, *Staphylococcus aureus* (*S. aureus*) is a Gram-positive bacterium that capable of causing numerous diseases of the human skin. The incidence of *S. aureus* skin infections reflects the conflict between the host skin's immune defenses and the *S. aureus*' virulence elements. Besides, *S. aureus* is a common component of human skin microbiota and it colonizes the nasal mucosa of the world's population.²⁸ Furthermore, *S. aureus* colonization has a negative impact on some chronic inflammatory dermatose such as atopic dermatitis (AD), a multifactorial complex disease that causes skin barrier dysfunction.^{29,30} It is also the primary cause of delayed healing and infection in both acute and chronic wounds and often produce invasive infections that can even cause sepsis.^{31,32} *S. aureus* represent the most common source of skin and soft tissue infections

(SSTIs) for Gram-positive bacteria.³³ Therefore, the antimicrobial assessment of *Staphylococcus aureus* on kojic acid ester-based nanoemulsion for topical application was studied.

The results of the inhibition zone size activity of *Staphylococcus aureus* ATCC 43300 on both samples were shown in Table 3 and Fig. 5 respectively. Based on the results, kojic acid ester-based nanoemulsion gave better inhibition size (8.0 mm) as compared to the active ingredient, kojic acid ester (6.5 mm). These findings may due to the composition of nanoemulsion which possesses high stability towards particle aggregation and gravitational separation showing improved and efficient activity such as antimicrobial activity.³⁴ Even though the active ingredient in the nanoemulsion, kojic acid ester was reported to have antibacterial properties against bacteria and fungi,^{35,36} the encapsulation of kojic acid ester in nanoemulsion gave better inhibition results due to its improvement in stability, controlled particle size and utilization for topical delivery. Since kojic acid ester has low solubility in water, it needs to be encapsulated into a suitable delivery system, in order to be active in topical application, where microorganism are more likely to grow and proliferate.¹⁸

Presumably, the mechanism action of the antimicrobial nanoemulsion against *S. aureus* involved by bringing the kojic acid ester into proximity with the cell membrane. The trapped kojic acid ester molecules in the inner phase of nanoemulsion may result in an extended drug release and a prolonged activity. Delivery systems at the nanoscale can potentially increase the passive cellular absorption mechanisms, thus reducing mass transfer resistances and increasing antimicrobial activity. Besides, the addition of lemon essential oil and potassium sorbate in the kojic acid ester-based nanoemulsion may also help in the inhibition growth of bacteria. The antimicrobial action of essential oil has been attributed to their phenolic compounds and their interaction with microbial cell membranes. They are known to penetrate through the microbial membrane and cause the leakage of ions and cytoplasmatic content thus leading to a cellular breakdown.³⁷ Previous

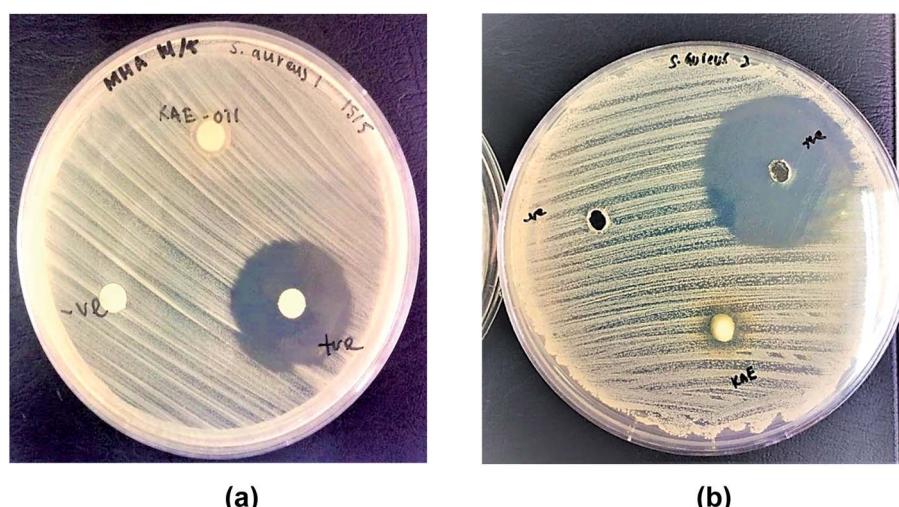


Fig. 5 Inhibition zone of (a) kojic acid ester and (b) kojic acid ester-based nanoemulsion towards *Staphylococcus aureus* ATCC 43300.



Table 4 Toxicity effect of kojic acid ester-based nanoemulsion on embryonic zebrafish after 96 hpf

Sample	Concentration ($\mu\text{g mL}^{-1}$)	Total embryos tested	Percentage (%)		Percentage (%)		$\text{LC}_{50} (\mu\text{g mL}^{-1})$
			Survival	Death	Normal	Abnormal	
Kojic acid ester-based nanoemulsion	500.00	36.00	75.00	25.00	66.67	33.33	>500
	250.00	36.00	83.33	16.67	83.33	16.67	
	125.00	36.00	83.33	16.67	83.33	16.67	
	62.50	36.00	83.33	16.67	83.33	16.67	
	31.25	36.00	91.67	8.33	83.33	16.67	
	15.63	36.00	91.67	8.33	83.33	16.67	
	7.81	36.00	91.67	8.33	83.33	16.67	
Embryo medium + 0.1% DMSO	—	36.00	100.00	0.00	100.00	0.00	

research by Pannu *et al.* 2011, developed novel NB-002 oil-in-water nanoemulsions with broad antifungal activity against dermatophytes and activity against propionibacterium acne for topical treatment of skin, hair and nail infections.³⁸ Besides, Hamouda *et al.* 2001, have presented a novel nanoemulsion formulation with a unique topical antimicrobial activity against bacteria, enveloped viruses and fungi including *H. influenzae*, *N. gonorrhoeae* and *V. cholera*, *P. aeruginosa*.³⁹

3.4 Toxicology evaluation using embryonic zebrafish (*Danio rerio*)

The exposure concentrations of kojic acid ester-based nanoemulsion (110.01 nm) for *in vivo* toxicity evaluation using embryonic zebrafish were ranging from 7.81–500 $\mu\text{g mL}^{-1}$. Besides the survival rate, hatching rate and teratogenic defects were also studied as suggested by Rizzo *et al.*, 2013.¹⁷ Table 4 shows the toxicity effect of kojic acid ester-based nanoemulsion expresses as survival and mortality of embryos after 96 hours, post-exposure is the concentration-dependent whereby the

embryo mortality increased with increased concentration of sample. Based on the results, there were only a few embryos were coagulated after 96 hpf even at high concentration. The calculated LC_{50} values of kojic acid ester-based nanoemulsion showed less toxic effect with LC_{50} more than 500 $\mu\text{g mL}^{-1}$. Fig. 6 demonstrates the survival rate of embryonic zebrafish treated with kojic acid ester-based nanoemulsion at 0 to 120 h. The survival rate of the embryonic zebrafish showed more than 80% when treated at doses ranging from 7.81–250 $\mu\text{g mL}^{-1}$. Embryos that survived showed normal development normally throughout the experiment without any observed deformation. This indicates that the concentrations tested are less toxic to the embryos and may offer for further cosmeceutical application.

Hatching rate of the embryonic zebrafish treated with kojic acid ester-based nanoemulsion at 24 to 120 h as shown in Table 5. Fig. 7 shows the healthy images of zebrafish embryogenesis showing stages of development at different hours of post-fertilization (hpf) captured using an inverted microscope at 10 \times digital magnification. According to previous studies, embryonic zebrafish is considered healthy if the embryos

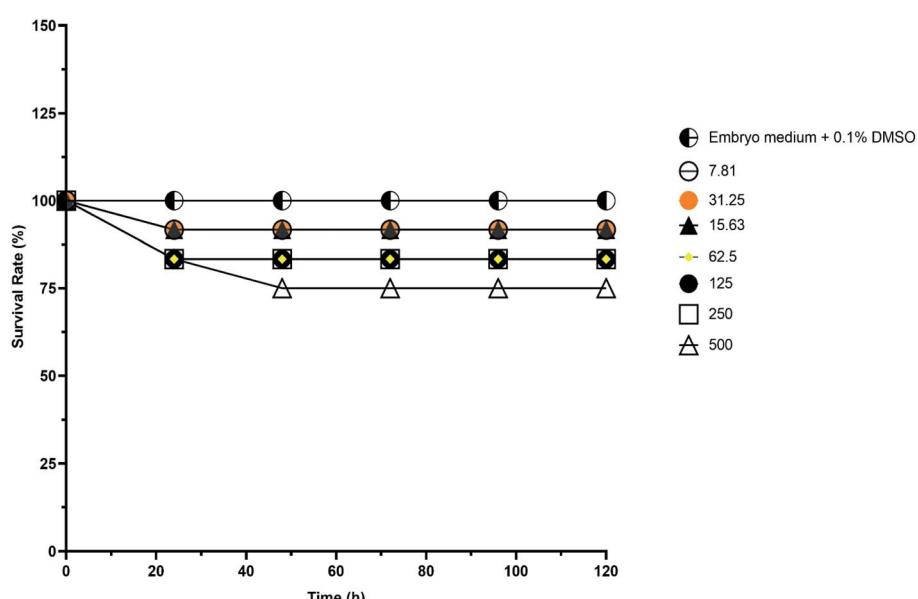


Fig. 6 The survival rate of embryonic zebrafish treated with kojic acid ester-based nanoemulsion at 0 to 120 h at different sample concentrations.



Table 5 Hatching rate of embryonic zebrafish treated with kojic acid ester-based nanoemulsion at 24 to 120 h

Sample Concentration ($\mu\text{g mL}^{-1}$)	Time of exposure (h)				
	24	48	72	96	120
500.00	0.00	16.67	66.67	66.67	66.67
250.00	0.00	0.00	83.33	83.33	83.33
125.00	0.00	0.00	83.33	83.33	83.33
62.50	0.00	0.00	83.33	83.33	83.33
31.25	0.00	0.00	91.67	91.67	91.67
15.63	0.00	0.00	91.67	91.67	91.67
7.81	0.00	0.00	91.67	91.67	91.67

exhibit all the epical endpoint characteristics such as the full formation of somite (body segment), presence of a heartbeat, tails completely detached from the yolk sac and no coagulation inside the embryos.⁴⁰⁻⁴²

Apart from that, the advantages of using zebrafish is that the eggs from fertilization to pharyngulation stage are transparent where the tissues become dense and pigmentation started to appear. Thus, this enables monitoring of all of the organs and systems for any changes during earlier developmental stages. In addition, teratogenic zebrafish embryos with lacking specific organs or having abnormal organs function can survive well

beyond the time where organs normally start to function in healthy individuals.⁴³

Fig. 8 shows images of malformation defect in zebrafish embryo and larvae after 120 h of exposure to a $500 \mu\text{g mL}^{-1}$ of KAE-NA captured using microscope $100\times$ digital magnification. Common teratogenic effects of the zebrafish embryo include the curved tail, short body length, curved body, pericardial edema. Wang, 2011 reported the fact that the induction of proteratogenic or procarcinogenic agents without the involvement of exogenous metabolic activation systems in the embryonic zebrafish can be promoted by activating phase I enzyme activities at very clear embryonic stages.⁴⁴ Therefore, the presence of teratogenicity in the developing larvae was slightly increased with the increasing concentration of the sample. The toxicity effect on morphological features of the zebrafish embryo was not dose dependent as evident by different teratogenic defects across all concentrations of KAE-NA.

This may be because the morphological defect was recorded at a low concentration of kojic acid ester-based-nanoemulsion. These teratogenic effects arise concurrently and non-concurrently in individual zebrafish larvae. The previous researcher documented the fact that the presence of organic fatty acids may affect the metamorphosis and formation of the larval skin which predisposes to high risk of morphological

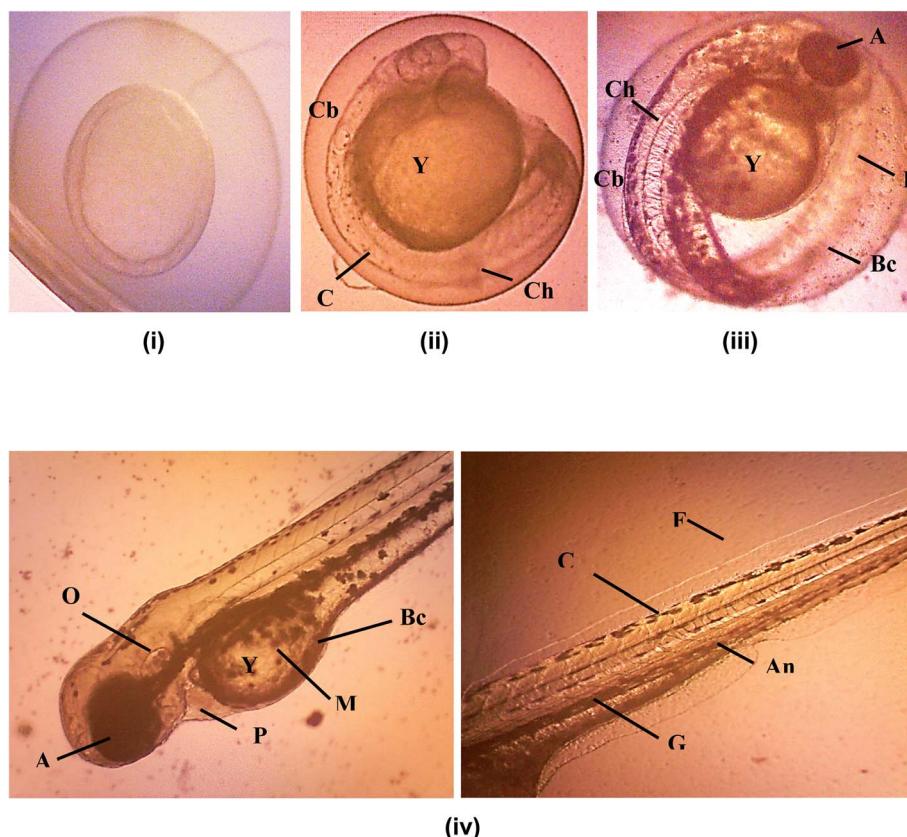


Fig. 7 Healthy images of zebrafish embryogenesis showing 4 stages of development at different hours of post fertilization (hpf) captured using inverted microscope at $10\times$ digital magnification. (i) Blastula period (4 hpf); (ii) segmentation period (24 hpf); (iii) pharyngula period (48 hpf); (iv) hatching period (72 hpf). A-eye anlage; An-anus; Bc-blood cells; C-chorda; Ch-chorion; Cb-curved body; F-fin; G-gut; M-melanophores; O-ear bud; P-pericard; Y-yolk sac. Scale bar = 0.5 mm.



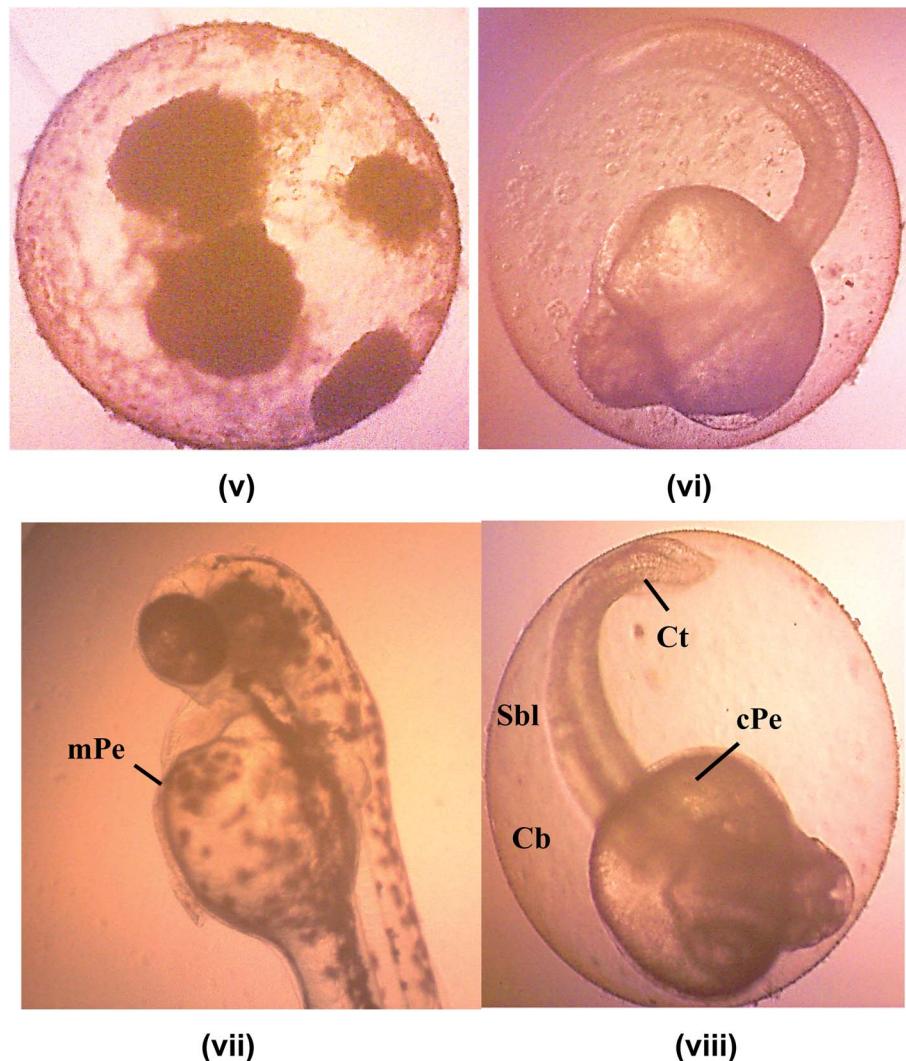


Fig. 8 Images of malformation defect in zebrafish embryo and larvae after 120 hours of exposure to a different concentration of KAE-NA captured using microscope 100 \times digital magnification. (v) Coagulated embryo; (vi) unhatched embryo; (vii) mPe-mild pericardial edema; (viii) Sbl-short body length, Cb-curve body, Ct-curve tail, cPe-chronic pericardial. Scale bar = 0.5 mm.

defects.⁴⁵ Besides, research had shown that injury, disease, parasite, stress-related spawning, unusual water quality conditions (high temperature, low dissolved oxygen, low pH), poor nutrition and toxic algal bloom are some of the factors that contribute to the teratogenic defects.⁴⁶

4. Conclusion

The permeation studies of kojic acid ester based-nanoemulsion revealed that the permeability of the kojic acid ester was increased from 4.94% at 1 h to 59.64% at 8 h. The permeation rate of kojic acid ester based-nanoemulsion at 8 h was 4659.50 $\mu\text{g cm}^{-2} \text{h}^{-1}$ (initial concentration, $C_0 = 2000 \mu\text{g mL}^{-1}$) with the permeability coefficient, K_p value of 0.48 cm h^{-1} shows nanoemulsion could be used as a vehicle carrier for the topical delivery system. The kinetic model suggested that Korsmeyer-Peppas model was the best fitted with high linearity of $R^2 = 0.9964$ indicate that the kinetic release of KAE-NA followed the

Super case II transport system. The LC₅₀ of zebrafish embryos (*Danio rerio*) showed no toxicity effect with more than 500 $\mu\text{g mL}^{-1}$. Antimicrobial activity of kojic acid ester based-nanoemulsion showed inhibition zone size of the kojic acid ester based-nanoemulsion (8.00 mm) was slightly bigger than kojic acid ester (6.5 mm). This study has revealed that KAE-NA has less toxicity and potentially used for further cosmeceutical applications.

Conflicts of interest

There are no conflicts to declare.

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