


 Cite this: *RSC Adv.*, 2020, 10, 24855

# A novel fabrication of electrospun polyacrylonitrile/NaYF<sub>4</sub>:Eu<sup>+3</sup> light emitting nanofibers

 Sanjeev Kumar, <sup>a</sup> Garima Jain, <sup>b</sup> Kuldeep Kumar, <sup>c</sup> Ashish Gupta, <sup>d</sup> B. P. Singh <sup>d</sup> and S. R. Dhakate <sup>d</sup>

Polyacrylonitrile/NaYF<sub>4</sub>:Eu<sup>+3</sup> nanophosphor composite nanofibers have been successfully prepared using the electrospinning technique. The electrospun nanofibers exhibited intense emission of gradient blue ( $X_2 = 0.254$ ,  $Y_2 = 0.152$  and  $X_3 = 0.233$ ,  $Y_3 = 0.139$ ) with different concentrations of nanophosphor under the excitation wavelength of 239 nm. The morphological and structural characterization of the nanofibers confirms the uniform dispersion of nanophosphor, while photoluminescence spectroscopy confirms tunability in luminescence properties.

 Received 2nd May 2020  
 Accepted 15th June 2020

 DOI: 10.1039/d0ra03984e  
[rsc.li/rsc-advances](http://rsc.li/rsc-advances)

## 1 Introduction

The one-dimensional (1D) nanostructures are receiving great attention from the scientific community especially in nanoscience, nanotechnology and biomaterials for tissue engineering of skin.<sup>1</sup> Nanotubes,<sup>2</sup> nanorod-superconductors<sup>3</sup> and molecular wires<sup>4</sup> exhibit a very high surface area to volume ratio that helped to enhance mechanical performance and luminescence characteristics.

Polyacrylonitrile (PAN), poly(methyl methacrylate) (PMMA), polyvinyl alcohol (PVA), polystyrene (PS), poly(ethylene oxide) (PEO), polyvinylpyrrolidone (PVP), polyvinylidene difluoride (PVdF), polyvinylcarbazole (PVK), poly[2-methoxy-5-(2-ethylhexyloxy)-*p*-phenylene vinylene] (MEH-PPV) and polydiallyldimethylammonium chloride PDAC are being used to prepare the luminescent composite nanofibers using electrospinning, while polypyrrole (PPy) used to synthesize material for neural tissue engineering to enhance cell interfacing.<sup>5</sup> The aspect ratio (length to diameter) of nanofibers leads them to show extraordinary properties. Large number of organic and inorganic functional additives such as nanoparticles, nanocrystals, nanophosphors and quantum dots have been incorporated into the above mentioned polymer matrices and their functionalization of properties to achieve the luminescent nanofibers for the solid state lighting.<sup>6–24</sup> The controlling of the

diameter of the fibers can be achieved by viscosity, evaporation rate, and charge density, tip of syringe to collector distance and collector speed.<sup>25–27</sup>

There are various nanomaterial processing techniques but electrospinning is one of the most fascinating technique for the preparation of polymer nanofibers with diameters ranging from micron to nanometer. The preparation of continuous nanofibers *via* electrospinning technique is simple, cost-effective and versatile.<sup>28–32</sup> Fabrication of composite nanofibers using electrospinning limited by % concentration of additives in polymer and potential applied.

In the past decade, numerous efforts have been made for the synthesis of rare earth (RE) doped nanocrystals. The RE ions have fantastic property to redistribute absorbed energy from shorter wavelength to longer wavelength region, which enables RE (Eu, Tb, and Sm) ions to be used in the optoelectronic device applications (luminosity in the visible region). The Eu ion has unique luminescent properties because of the 4f–4f transition available in Eu<sup>+3</sup> ions.<sup>33–38</sup> NaYF<sub>4</sub> is one of the marvelous down/up conversion host lattices among them. The doping of lanthanide ions with a size larger than Y<sup>3+</sup> ( $r = 1.159 \text{ \AA}$ ) in NaYF<sub>4</sub> host lattices should dominate the formation of pure hexagonal-phase NaYF<sub>4</sub> nanocrystals, luminescent properties of RE ions are being enhanced by changing the lattice parameters of the host matrix.<sup>39–42</sup> In recent years, Yu and Safaei *et al.* have been prepared electrospun PAN photoluminescent nanofibers using terbium complexes and carbon quantum dots which show characteristic peaks around at 545 nm and 660 nm respectively.<sup>35,36</sup> Although, the improvement in photoluminescence properties of PAN electrospun nanofibers have been observed by using organic europium complexes. The characteristics emission peaks were observed for these studies from 500 to 700 nm color spectrum region.<sup>46,47</sup>

<sup>a</sup>Department of Physics, R. K. (PG) College Shamli, C. C. S. University Meerut, UP-247776, India. E-mail: sanjeev.raonpl@gmail.com; Tel: +91 7906899520

<sup>b</sup>Department of Physics, D. A. V. (PG) College Muzaffarnagar, C. C. S. University Meerut, UP-251001, India

<sup>c</sup>Department of Physics, SGTB Khalsa College, University of Delhi, Delhi 110007, India. E-mail: kuldeep@sgtbkhalsa.du.ac.in; Tel: +91 8826076455

<sup>d</sup>Advanced Carbon Products and Metrology Section, Advanced Materials and Devices Metrology Division, CSIR-National Physical Laboratory, New Delhi-110012, India


Polyacrylonitrile (PAN) is considered to be an important engineering polymer and widely used for the fabrication of versatile synthetic functional ultrafine nanofibers.<sup>43–45</sup> Since the PAN provides the excellent processability and flexibility to nanophosphor during the electrospinning process. Barely, study has been done on preparation of PAN electrospun nanofibers by embedding of europium doped sodium yttrium fluoride ( $\text{NaYF}_4:\text{Eu}^{+3}$ ). Inspired by this, herein  $\text{NaYF}_4:\text{Eu}^{+3}$  nanophosphors were incorporated into the PAN matrix by using electrospinning technique.<sup>48–50</sup> The PL properties of PAN/ $\text{NaYF}_4:\text{Eu}^{+3}$  electrospun PLNs have been explored and showed the significant enhancement in the properties with the loading of  $\text{NaYF}_4:\text{Eu}^{+3}$ . To the best of our knowledge, no study has been reported on electrospun photoluminescent PAN composite nanofibers with inorganic additives having such an exciting improvement in photoluminescence properties till date.

## 2 Experimental

PAN used in the proposed study was purchased from RKB Industries, UK and *N,N*-dimethyl formamide (DMF), yttrium oxide ( $\text{Y}_2\text{O}_3$ ) and europium oxide ( $\text{Eu}_2\text{O}_3$ ) of 99.99% sodium fluoride (NaF) (99.9%), sodium hydroxide (NaOH), hydrochloric acid (HCl) and ethanol were purchased from Thomas Baker (Chemicals) Pvt. Ltd. (TBCPL), Mumbai, India. All reagents were of analytical grade and used without further purification in the experiment. Synthesis of  $\text{NaYF}_4:\text{Eu}^{+3}$  nanophosphor were similar to that reported by us previously.<sup>51</sup>

### 2.1 Preparation of polyacrylonitrile photoluminescent nanofibers

PAN and  $\text{NaYF}_4:\text{Eu}^{+3}$  nanophosphor have been used to prepare the polymeric PLNs by using electrospinning technique. In electrospinning process, high voltage DC is applied between the droplet of polymer solution and the grounded collector. Whenever the surface tension of polymer solution droplet gets overcome by the applied electric field, ultrafine nanofibers are produced. The electrospinning machine (ESPIN-NANO PICO,

Chennai) has been used to fabricate the PLNs. The systematic steps of process are shown in Fig. 1.

As produced 20 mg of  $\text{NaYF}_4:\text{Eu}^{+3}$  nanophosphor was dispersed in 4.6 g DMF. This solution was kept for the ultra-sonication upto 30 minutes at 40 °C temperature to avoid the agglomeration. After that 400 mg PAN was put into the above dispersed solution and kept it again for ultra-sonication for 90 minutes at same temperature. Further, to obtain the homogeneous PAN solution, this solution was kept under vigorous magnetic stirring for 10 h at room temperature. Two concentrations of nanophosphor, 5 and 8% were used to obtain the PLNs. Therefore, the PAN homogeneous solution was poured into the 5 ml plastic syringe having 0.7 mm diameter of needle. 13 cm × 14 cm aluminum foil sheet was used as a substrate for the PLNs which was wrapped on collector of electrospinning machine. The plastic disposal syringe was hanged at the stand of machine and the distance between the tip of syringe needle and collector was maintained at 20 cm. The applied voltages between the tip and collector were maintained at 15 kV with the flow rate of 0.15 ml h<sup>-1</sup> and collector speed of 2000 rpm. The reagents and solvents were determined by optimizing various solutions and instrumental parameters to get the nanofibers. The concentration of phosphor material (5 and 8%) was also optimized so that solution does not get accumulate on syringe. Above 8% of  $\text{NaYF}_4:\text{Eu}^{+3}$ , it is difficult to spun nanofibers from polymer solution. Hence, random nanoscale polymeric PLNs network were obtained on aluminum foil at room temperature.

## 3 Result and discussion

### 3.1 X-ray diffraction (XRD) of PAN/ $\text{NaYF}_4:\text{Eu}^{+3}$ nanofibers

The cubic crystal structure of  $\text{NaYF}_4$  [blue → sodium (Na), red → yttrium (Y) and green → fluorine (F)] is shown in the Fig. 2. The image was created by using Visualization for Electronic and SStructure Analysis (VESTA) package.<sup>52</sup> The ionic radius of  $\text{Eu}^{+3}$  (~1.07 Å) is comparable to  $\text{Y}^+$  (~1.159 Å) that indicates the  $\text{Eu}^{+3}$  can easily replace  $\text{Y}^+$  without any major change in structure.

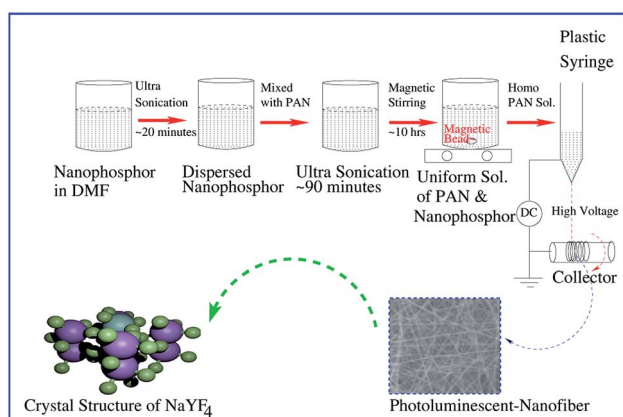


Fig. 1 Schematic illustration of the process and fabrication of PAN/ $\text{NaYF}_4:\text{Eu}^{+3}$  photoluminescent nanofibers (PLNs).

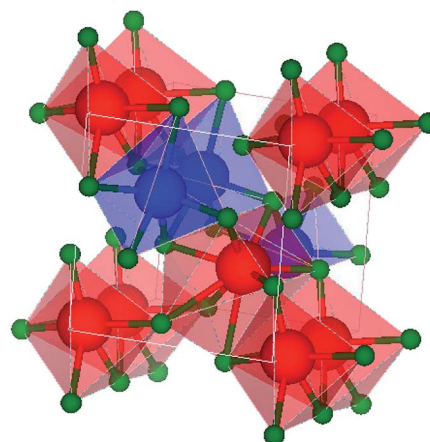


Fig. 2 The crystal structure of  $\text{NaYF}_4$  in cubic phase (created by [blue → sodium (Na), red → yttrium (Y) and green → fluorine (F)]).



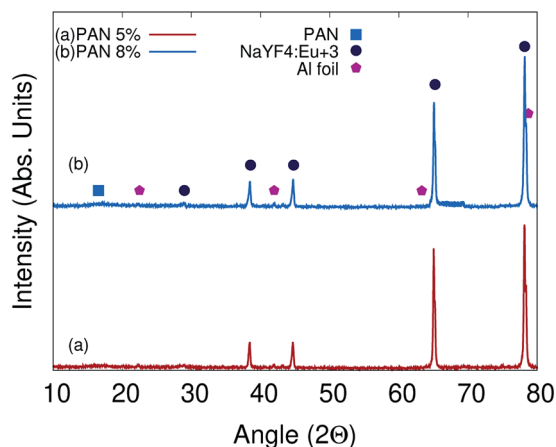


Fig. 3 XRD pattern of PAN/NaYF<sub>4</sub>:Eu<sup>+3</sup> electrospun PLNs @ 5% concentration of nanophosphor (a) and XRD pattern of PLNs @ 8% (b) shown.

The microscopic structure and crystalline nature of PAN/NaYF<sub>4</sub>:Eu<sup>+3</sup> electrospun photoluminescent nanofibers were confirmed by XRD measurement using X-ray diffractometer Rigaku Japan with Cu-K $\alpha$  radiation having a wavelength of  $\lambda = 1.5418$  Å. Fig. 3(a) and (b) exhibit the XRD pattern of PAN/NaYF<sub>4</sub>:Eu<sup>+3</sup> electrospun PLNs @ 5% and 8% concentrations of nanophosphor in PAN matrices respectively. The peak intensities are slightly higher at 8% which is the indicative of higher density of NaYF<sub>4</sub>:Eu<sup>+3</sup> in PAN nanofibers. The diffraction peak observed at around  $2\theta = 16.86^\circ$  which is attributed to PAN.<sup>53,54</sup>

The JCPDS card no 77-2042 has been used to investigate the cubic structure of NaYF<sub>4</sub>:Eu<sup>+3</sup> nanophosphor. The existence of peaks at angles  $2\theta = 28.8^\circ, 38.42^\circ, 43.17^\circ, 44.72^\circ, 65.50^\circ$  and  $78.72^\circ$  corresponding to crystal planes (111), (200), (220), (222), (400), and (420) respectively are attributed to the nanophosphors that conforms cubic crystal structure with lattice parameters of  $a = b = c = 5.369$  Å,  $\alpha = \beta = \gamma = 90^\circ$ .<sup>55</sup> It can be seen from XRD that the peaks of composite nanophosphor are slightly right shifted which indicates a tensile stress present on NaYF<sub>4</sub> due to the PAN molecule shell. The stress present on NaYF<sub>4</sub> is not so efficient to deform crystal structure.<sup>56-58</sup> The crystalline peaks centered at about angles  $26.62^\circ, 42.02^\circ, 65.12^\circ$

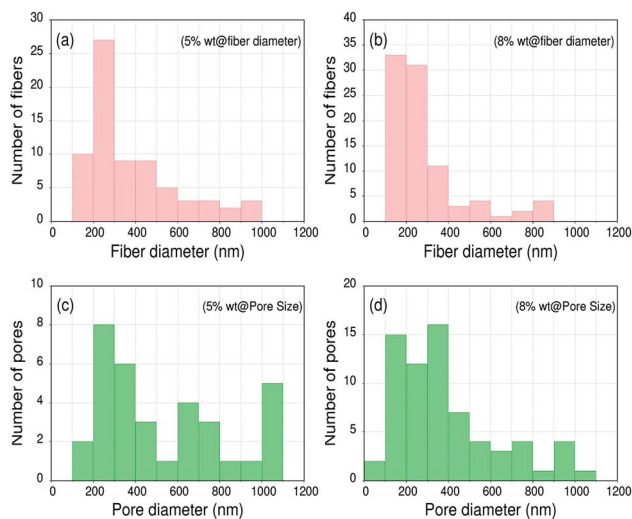


Fig. 5 SEM image texture analysis of PAN/NaYF<sub>4</sub>:Eu<sup>+3</sup> PLNs. (a and b) shows the fiber diameter distribution (FDD) @ 5 and 8% and (c and d) shows the pore size distribution (PSD) @ 5 and 8%.

and  $78.34^\circ$  are attributed to Al foil which was wrapped on collector to obtain the composite PLNs.<sup>59</sup>

### 3.2 Morphology of polyacrylonitrile (PAN)/NaYF<sub>4</sub>:Eu<sup>+3</sup> PLNS

The surface morphology of ultrafine electrospun composite nanofibers was recognized by the Scanning Electron Microscope (SEM) images using model ZEISS EVO 18. Fig. 4(a and b) show the SEM images of PAN/NaYF<sub>4</sub>:Eu<sup>+3</sup> nanophosphor composite nanofibers @ 5% and 8% respectively. The histograms for PAN/NaYF<sub>4</sub>:Eu<sup>+3</sup> composite nanofibers are shown in Fig. 5. The diameter and pore analysis shown in the graph illustrates the size distribution of PAN/NaYF<sub>4</sub>:Eu<sup>+3</sup> nanofibers @ 5 and 8%. In both cases, the nanofibers are semi aligned and form a network like structure. Fig. 5(a and b) shows the nanofibers embedded with 5% nanophosphor with smooth morphology and the average diameter measure as 383 nm using ImageJ analysis software. However on adding more phosphor *i.e.* 8% the diameter reduces due to the increase in metal content (5 to 8%) to an average size 286 nm. However, due to increased additive content more bead formation also take place

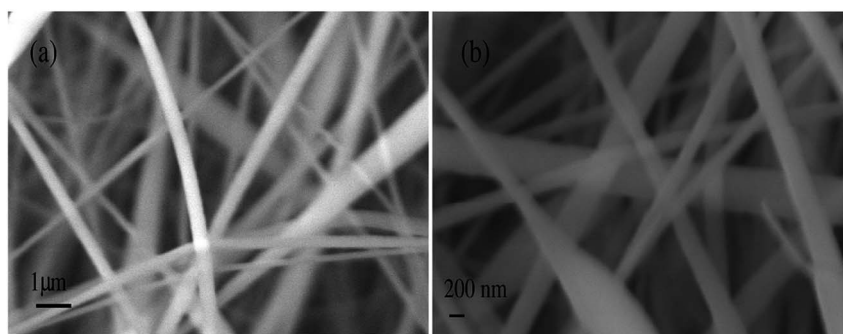


Fig. 4 SEM images of PAN/NaYF<sub>4</sub>:Eu<sup>+3</sup> electrospun PLNs with (a) 5% and (b) 8%.



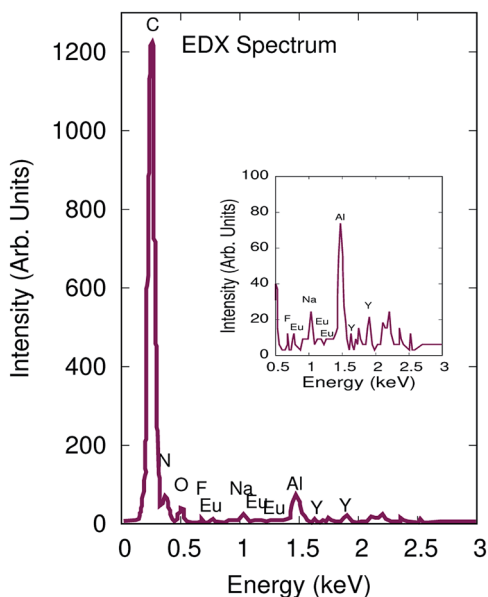


Fig. 6 EDX spectrum of PAN/NaYF<sub>4</sub>:Eu<sup>+3</sup> PLNs @ 8%.

in nanofibers. The average pore size of the composite nanofibers was observed  $\sim 1351$  nm and  $\sim 453$  nm @ 5 and 8% shown in Fig. 5(c and d).

Elemental analysis was performed using the energy-dispersive spectroscopy accessory connected to the SEM system (VPSEM ZEISS EVO M-10 with OXFORD INCA EDX DETECTOR). The energy dispersive X-ray spectroscopy (EDX) spectrum is shown in Fig. 6 which revealed homogeneous distribution of all elements. The EDX spectra confirm the presence of C, N, O, Na, Y, F, Eu elements in the PAN/NaYF<sub>4</sub>:Eu<sup>+3</sup> nanofibers. Aluminum (Al) peak is indicative of Al-foil substrate. This finding confirms that the nanofibers were fabricated efficiently with nanophosphor material. The EDX spectra show that rare earth elements (RE) are attached with the nanofibers. During the electrospinning, the polymers get stretched in the form of nanofibers along with NaYF<sub>4</sub>:Eu<sup>+3</sup> on applying field. Hence, electrostatic interaction plays a significant role during fiber fabrication but does not play any role to activate ageing of nanofibers.

Fig. 7(a) shows the High Resolution Transmission Microscopy (HRTEM-JEOL 2100F) image of nanofibers @ 5% which exhibits the NaYF<sub>4</sub>:Eu<sup>+3</sup> nanoparticles were homogeneously dispersed into the PAN fibers matrix. The presence of NaYF<sub>4</sub>:Eu<sup>+3</sup> nanophosphor inside the nanofibers can be understood by the artwork image Fig. 7(b) also.

The nano particles are well distributed into the PAN matrices due to a strong interaction between the PAN molecule chains and the Eu<sup>+3</sup> ions. The flow rate, distance between the tip of syringe needle and collector and applied voltages are the key factors which can influence the surface morphology of composite nanofibers during the electrospinning. Flow rates of 0.2 ml h<sup>-1</sup> and 0.3 ml h<sup>-1</sup> have been optimized and found that many beads were present in electrospun composite nanofibers. As a result, the flow rate played an important role to get the better morphology and diameter of the resulting nanofibers. So

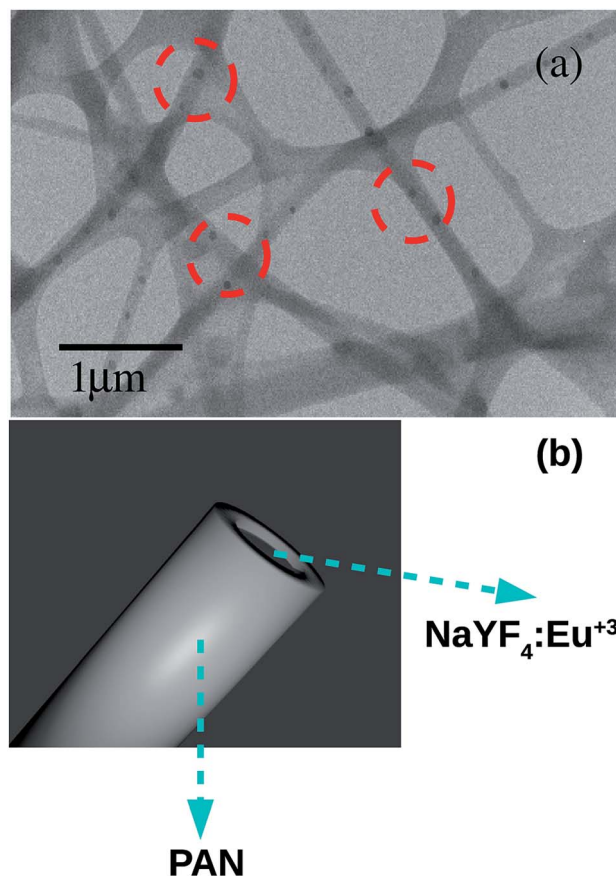


Fig. 7 (a) HRTEM image of PLNs showing the nanoparticles of NaYF<sub>4</sub>:Eu<sup>+3</sup> homogeneously embedded inside the nanofibers and (b) an artwork to represent NaYF<sub>4</sub>:Eu<sup>+3</sup> crystal inside PAN nanofiber.

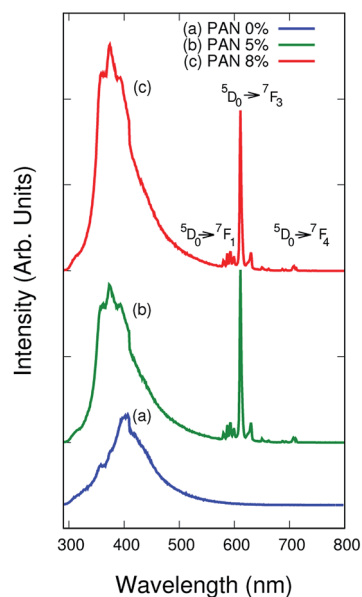


Fig. 8 Photoluminescence spectra of (a) pure PAN, (b) PAN/NaYF<sub>4</sub>:Eu<sup>+3</sup> PLNs @ 5% and (c) @ 8% upon excitation wavelength of 239 nm.



that 0.15 ml h<sup>-1</sup> flow rate for PAN homogeneous solution was best suited in the proposed study. One cannot ignore the importance of concentrations of nanophosphor in PAN precursor which has the considerable affect on surface morphology of the nanofibers. When the amounts of NaYF<sub>4</sub>:Eu<sup>+3</sup> nanophosphor was increased from 0 to 8%, the ultrafine uniform and continuous PAN/NaYF<sub>4</sub>:Eu<sup>+3</sup> electrospun composite nanofibers were achieved on collector. The needle of plastic syringe nozzle did not allow the concentrations of nanophosphor in PAN matrices beyond the 8% for smooth fabrication of nanofibers. NaYF<sub>4</sub>:Eu<sup>+3</sup> nanophosphor has been prepared separately for the fabrication of electrospun PLNs. These nano particles were incorporated into the PAN matrix *via* electrospinning.

### 3.3 Photoluminescence properties of PAN/NaYF<sub>4</sub>:Eu<sup>+3</sup> nanofibers

The photoluminescence spectra<sup>60–63</sup> of bare PAN and the electrospun composite nanofibers were investigated using spectrometer model PerkinElmer LS 55 as shown in Fig. 8. NaYF<sub>4</sub>:Eu<sup>+3</sup> nanophosphor were doped in PAN to understand the photoluminescence properties of the composite nanofibers.

Fig. 8(a–c) reveal the photoluminescence spectrum of pure PAN and composite nanofibers under 239 nm excitation wavelength at room temperature. The bare PAN fibers have narrow emission peak at around 405 nm whereas PAN/NaYF<sub>4</sub>:Eu<sup>+3</sup> electrospun composite nanofibers exhibit very sharp peaks at around 593, 625, 628 and 710 nm. The characteristic sharp peaks were observed with the <sup>5</sup>D<sub>0</sub> → <sup>7</sup>F<sub>J</sub>, where J ranges from 0 to 4, transition of Eu<sup>+3</sup> ions in the range of 300 to 710 nm region. The transition <sup>5</sup>D<sub>0</sub> → <sup>7</sup>F<sub>1</sub> at about 593 nm was observed in the orange spectral region due to the magnetic dipole transitions. While the intense sharp transitions <sup>5</sup>D<sub>0</sub> → <sup>7</sup>F<sub>3</sub> and <sup>5</sup>D<sub>0</sub> → <sup>7</sup>F<sub>4</sub> at about 625 nm and 710 nm were observed in the red

spectral region of the photoluminescence spectrum due to the electric dipole transitions respectively.

The quantum efficiency (QE) is a measure of luminescence efficiency. The QE of nanofibers is calculated by eqn (1) using the integrated sphere method.<sup>64</sup>

$$QE = \frac{\int L_S}{\int E_R - \int E_S} \quad (1)$$

where L<sub>S</sub> is the luminescence emission spectra of the material, E<sub>S</sub> is the spectrum of the excitation light to excite the material, and E<sub>R</sub> is the spectrum of the light without sample. The maximum QEs of composite nanofibers were calculated as 83.1% @ 5% and 87.5% @ 8% under the excitation wavelength of 239 nm.

In such a case the orange peak intensity was almost negligible and hypersensitive red peaks appeared.<sup>65–67</sup> The observations suggested that the Eu<sup>+3</sup> ions were homogeneously impinged in the PAN matrices and placed on a molecule chain due to the effect of electric field during the electrospinning process.<sup>68</sup> When the concentrations of NaYF<sub>4</sub>:Eu<sup>+3</sup> nanophosphor increased up to 8%, the emission intensity increased sharply is because of Eu<sup>+3</sup> ions get distributed homogeneously in the PAN chains. Fig. 9 shows the International Commission on Illumination (CIE) diagram from the photoluminescence spectra of electrospun nanofibers. The chromaticity diagram demonstrates the color model which is a numerical model.<sup>69</sup> This color model represents the color spectrum visible to the ordinary man. The color space chromaticity diagram was drawn using the CIE 1931 color matching functions with the CIE coordinates (X<sub>1</sub> = 0.169, Y<sub>1</sub> = 0.10) @ PAN, (X<sub>2</sub> = 0.254, Y<sub>2</sub> = 0.152) @ 5% and (X<sub>3</sub> = 0.233, Y<sub>3</sub> = 0.139) @ 8% at the excitation wavelength of 239 nm. These CIE coordinates shows a variation in photoluminescence with different % concentrations.

The work has limitation that not higher content of nanophosphor material (>8% in this work) can be added due to limitation of solution flow and agglomeration on syringe's needle. In our study, smooth nanofibers with 8% of NaYF<sub>4</sub>:Eu<sup>+3</sup> were fabricated and these are showing good photoluminescence intensity, which is sufficient for its flexible lightening application, although, more intensity will be a plus point. Novelty of the work lies in the optimization of process parameters in fabricating inorganic composite nanofibers having 8% of nanophosphor showing efficient performance of composite nanofibers. Also, no study shows the incorporation of NaYF<sub>4</sub>:Eu<sup>+3</sup> nanophosphor in PAN polymer as nanofibers. The high effective surface of nanofibers and embedment of nanophosphor in the PAN both contributed equally in the marvelous performance of composite nanofibers.

## 4 Conclusion

NaYF<sub>4</sub>:Eu<sup>+3</sup> PLNs were prepared successfully *via* electrospinning technique. The average diameters of PLNs were observed ~ 310 nm. The SEM and HRTEM images clearly distinguished the existence of Eu<sup>+3</sup> ions which are embedded

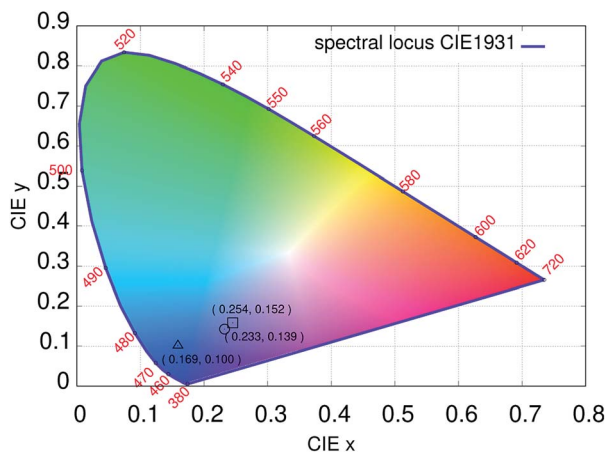


Fig. 9 CIE (1931) chromaticity diagram for the electrospun nanofibers corresponding to excitation wavelength of 239 nm with CIE coordinates (X<sub>1</sub> = 0.169, Y<sub>1</sub> = 0.10) @ PAN indicated by Δ, (X<sub>2</sub> = 0.254, Y<sub>2</sub> = 0.152) @ 5% indicated by □ and (X<sub>3</sub> = 0.233, Y<sub>3</sub> = 0.139) @ 8% indicated by ○.



homogeneously in the PAN molecule chain. The  $\text{Eu}^{+3}$  ions have strong coordination interaction with polymers therefore the prepared composite nanofibers exhibit the significant effect on photoluminescent properties. The composite nanofibers demonstrated the marvelous emission peaks from 593 to 710 nm region which lead the nanofibers for the stable solid state lighting applications whereas organic PAN composite nanofibers for flexible lighting applications. The results are pretty illuminating as compared to the existing studies on inorganic additives based electrospun polymeric light emitting nanofibers. After this fundamental study, there is a need to check the real time applications of the nanofibers. Future research will focus on the possible use of these nanofibers in different applications such as combating counterfeiting of expensive products, solar cell, energy storage and biomedical fields.

## Conflicts of interest

There are no conflicts to declare.

## Acknowledgements

Authors are grateful to Professor P. D. Sahare, Department of Physics and Astrophysics University of Delhi, Delhi for providing the experimental support for the preparation of nanophosphor. We would like to thank Dr Govind, Senior Principal Scientist and Mr Jai Tawale from CSIR-NPL, New Delhi for Photoluminescence (PL) and Energy Dispersive X-ray Spectroscopy (EDX) characterization respectively. Also, we thank to Professor Beer Pal Singh (Head), and Dr Yogendra Kumar Gautam, Department of Physics, C. C. S. University, Meerut, UP for providing facilities of SEM. One of the authors, Dr Ashish Gupta thanks CSIR for RA.

## Notes and references

- Q. Zhang, Q. Du, Y. Zhao, F. Chen, Z. Wang, Y. Zhang, H. Ni, H. Deng, Y. Li and Y. Chen, *RSC Adv.*, 2017, 7, 28826–28836.
- H. Dai, E. W. Wong, Y. Z. Lu and C. M. Lieber, *Nature*, 1995, 375, 769–772.
- P. D. Yang and C. M. Lieber, *Science*, 1996, 273, 1836.
- C. Nacci, F. Ample, D. Bleger, S. Hecht, C. Joachim and L. Grill, *Nat. Commun.*, 2015, 6, 7397.
- W. R. Stauffer and X. T. Cui, *Biomaterials*, 2004, 25, 2477–2488.
- Y. J. Chen, J. B. Li, Y. S. Han, X. Z. Yang and J. H. Dai, *J. Cryst. Growth*, 2002, 245, 163–170.
- J. Liu, Z. Yang, B. Ye, Z. Zhao, Y. Ruan, T. Guo, X. Yu, G. Chen and S. Xu, *J. Mater. Chem. C*, 2019, 7, 4934–4955.
- M. A. Sobhan, A. Lebedev, L. L. Chng and F. Anariba, *J. Nanomater.*, 2018, 1980357.
- M. Yang, J. Yu, S. Jiang, C. Zhang, Q. Sun, M. Wang, H. Zhou, C. Li, B. Man and F. Lei, *Opt. Express*, 2018, 26, 20649–20660.
- S. G. Itankar, M. P. Dandekar, S. B. Kondawar and B. M. Bahirwar, *Luminescence*, 2017, 32, 1535–1540.
- M. M. A. Abualrejal, H. Zou, J. Chen, Y. Song and Y. Sheng, *Adv. Nanopart.*, 2017, 6, 33–47.
- J. Y. Lei, W. Wang, M. X. Song, B. Dong, Z. Y. Li and C. Wang, *React. Funct. Polym.*, 2011, 71, 1071–1076.
- R. Tatavarty, E. T. Hwang, J. W. Park and J. H. Kwak, *React. Funct. Polym.*, 2011, 71, 104–108.
- D. R. Son, A. V. Raghu, K. R. Reddy and H. M. Jeong, *J. Macromol. Sci., Part B: Phys.*, 2016, 55, 1099–1110.
- K. Reddy, K. Karthik, S. B. Prasad, S. Soni, H. Jeong and A. Raghu, *Polyhedron*, 2016, 120, 169–174.
- K. R. Reddy, K. P. Lee and A. I. Gopalan, *J. Nanosci. Nanotechnol.*, 2007, 7, 3117–3125.
- K. R. Reddy, B. C. Sin, K. S. Ryu, J. C. Kim, H. Chung and Y. Lee, *Synth. Met.*, 2009, 159, 595–603.
- S. J. Han, H. I. Lee, H. M. Jeong, B. K. Kim, A. V. Raghu and K. R. Reddy, *J. Macromol. Sci., Part B: Phys.*, 2014, 53, 1193–1204.
- K. R. Reddy, K. P. Lee and A. I. Gopalan, *J. Appl. Polym. Sci.*, 2007, 106, 1181–1191.
- M. Hassan, K. R. Reddy, E. Haque, S. N. Faisal, S. Ghasemi, A. I. Minett and V. G. Gomes, *Compos. Sci. Technol.*, 2014, 98, 1–8.
- K. R. Reddy, K. P. Lee, Y. Lee and A. I. Gopalan, *Mater. Lett.*, 2008, 62, 1815–1818.
- Y. R. Lee, S. C. Kim, H. Lee, H. M. Jeong, A. V. Raghu, K. R. Reddy and B. K. Kim, *Macromol. Res.*, 2011, 19, 66–71.
- M. U. Khan, K. R. Reddy, T. Snguanwongchai, E. Haque and V. G. Gomes, *Colloid Polym. Sci.*, 2016, 294, 1599–1610.
- S. H. Choi, D. H. Kim, A. V. Raghu, K. R. Reddy, H. I. Lee, K. S. Yoon, H. M. Jeong and B. Kyu Kim, *J. Macromol. Sci., Part B: Phys.*, 2012, 51, 197–207.
- M. L. Shofner, F. J. Rodriguez-Macias and R. Vaidyanathan, *Composites*, 2003, 34, A1207.
- C. J. Otten, O. R. Louire, M. F. Yu, J. M. Cowley, M. J. Dyer, R. S. Ruoff and W. E. Buhro, *J. Am. Chem. Soc.*, 2002, 124, 4564–4565.
- R. Stepanyan, A. Subbotin, L. Cuperus, P. Boonen, M. Dorsch, F. Oosterlinck and M. Bulters, *Appl. Phys. Lett.*, 2014, 105, 173105–173109.
- D. H. Reneker, A. L. Yarin, H. Fong and S. Koombhongse, *J. Appl. Phys.*, 2000, 87, 4531–4547.
- A. Formhals, *US Pat.*, 1975504, 1934.
- L. M. Larrondo and R. S. J. Manley, *J. Polym. Sci., Polym. Phys. Ed.*, 1981, 12, 921–932.
- P. Khude, *J. Mater. Sci. Eng.*, 2017, 6, 1000399.
- Y. Liu, R. Liu, X. Wang, J. Jiang, W. Li, J. Liu, S. Guo and G. Zheng, *Micromachines*, 2018, 9, 427.
- G. Y. Adachi and N. Imanaka, *Chem. Rev.*, 1998, 98, 1479–1514.
- B. C. Hong and K. Kawano, *Sol. Energy Mater. Sol. Cells*, 2003, 80, 417–432.
- Z. Yu, L. Shen, D. Li, E. Y. B. Pun, X. Zhao and H. Lin, *Sci. Rep.*, 2020, 10, 926.
- B. Safaei, M. Youssefi, B. Rezaei and N. Irannejad, *Smart Science*, 2017, 6, 117–124.
- H. G. Liu, F. Xiao, W. S. Zhang and Y. Chung, *J. Lumin.*, 2005, 114, 187–196.



- 38 H. Liu, Y. Lee, W. P. Qin, K. Jang and X. S. Feng, *Mater. Lett.*, 2004, **58**, 1677–1682.
- 39 F. Wang, Y. Han, C. S. Lim, Y. Lu, J. Wang, J. Xu, H. Chen, C. Zhang, M. Hong and X. Liu, *Nature*, 2010, **463**, 1061–1065.
- 40 O. Graeve, S. Varma, G. George, D. Brown and E. Lopez, *J. Am. Chem. Soc.*, 2006, **89**, 926–931.
- 41 C. Li, C. Zhang, Z. Hou, L. Wang, Z. Quan, H. Lian and J. Lin, *J. Phys. Chem. C*, 2009, **113**, 2332–2339.
- 42 K. Kramer, D. Biner, G. Frei, H. Güdel, M. Hehlen and S. Lüthi, *Chem. Mater.*, 2004, **16**, 1244–1251.
- 43 E. Zussman, X. Chen, W. Ding, L. Calabri, D. A. Dikin, J. P. Quintana and R. S. Ruoff, *Carbon*, 2005, **43**, 2175–2185.
- 44 V. E. Kalayci, P. K. Patra, Y. K. Kim, S. C. Ugbohue and S. B. Warner, *Polymer*, 2005, **46**, 7191–7200.
- 45 K. Yoon, K. Kim, X. Wang, D. Fang, B. S. Hsiao and B. Chu, *Polymer*, 2006, **47**, 2434–2441.
- 46 Y. Wang, L. Huang, J. Tang, Y. Wang, X. Li and W. Ma, *Int. J. Electrochem. Sci.*, 2016, **11**, 2058–2065.
- 47 H. Wang, Q. Yang, L. Sun, C. Zhang, Y. Li, S. Wang and Y. Li, *J. Alloys Compd.*, 2009, **488**, 414–419.
- 48 Y. Zhou, L. Y. Hao, Y. Hu, R. Y. Zhu and Z. K. Chen, *Chem. Lett.*, 2000, 1308–1309.
- 49 A. Gole and C. J. Murphy, *Chem. Mater.*, 2005, **17**, 1325–1330.
- 50 C. Wang, E. Yan, Z. Y. Sun, Z. J. Jiang, Y. B. Tong, Y. Xin and Z. H. Huang, *Macromol. Mater. Eng.*, 2007, **292**, 949–1008.
- 51 S. Kumar, G. Jain, B. P. Singh and S. R. Dhakate, *J. Nanomater.*, 2020, 1023589.
- 52 K. Momma and F. Izumi, *J. Appl. Crystallogr.*, 2011, **44**, 1272–1276.
- 53 M. Panapoy, A. Dankeaw and B. Ksapabutr, *Thammasat Int. J. Sci. Tech.*, 2008, **13**, 11–17.
- 54 P. S. Saud, Z. K. Ghouri, B. Pant, T. An, J. H. Lee, M. Park and H. Y. Kim, *Carbon Lett.*, 2016, **18**, 30–36.
- 55 A. K. Gangwar, A. Gupta, G. Kedawat, P. Kumar, B. P. Singh, N. Singh, A. K. Srivastava, S. R. Dhakate and B. K. Gupta, *Chem.–Eur. J.*, 2018, **24**, 9477–9484.
- 56 K. Kumar, P. Arun, C. R. Kant and B. K. Juluri, *Appl. Phys. Lett.*, 2012, **100**, 243106.
- 57 K. Kumar, P. Arun, C. R. Kant, N. C. Mehra and V. Mathew, *Appl. Phys. A*, 2010, **99**, 305–310.
- 58 K. Kumar, P. Arun, C. R. Kant, N. C. Mehra, L. Makinistian and E. A. Albanesi, *J. Phys. Chem. Solids*, 2010, **71**, 163–169.
- 59 M. Manikandan, J. Gopal and S. Chun, *RSC Adv.*, 2016, **6**, 32405–32413.
- 60 M. Hassan, E. Haque, K. R. Reddy, A. I. Minett, J. Chenc and V. G. Gomes, *Nanoscale*, 2014, **6**, 11988–11994.
- 61 M. J. Kade, D. J. Burke and C. J. Hawker, *J. Polym. Sci., Part A: Polym. Chem.*, 2010, **48**, 1477–1484.
- 62 K. R. Reddy, K. P. Lee, J. Y. Kim and Y. Lee, *J. Nanosci. Nanotechnol.*, 2008, **8**, 5632–5639.
- 63 N. L. Reddy, V. N. Rao, M. V. kumar, R. Santhosh, S. Anandan, M. Karthik, M. V. Shankar, K. R. Reddy, N. P. Shetti, M. N. Nadagoudae and T. M. Aminabhavi, *Int. J. Hydrogen Energy*, 2019, **44**, 10453–10472.
- 64 S. Miao, Z. Xia, J. Zhang and Q. Liu, *Inorg. Chem.*, 2014, **53**, 10386–10393.
- 65 R. He, X. F. Qian, J. Yin, L. J. Bian, H. A. Xi and Z. K. Zhu, *Mater. Lett.*, 2003, **57**, 1351–1354.
- 66 Q. Li, T. Li and J. Wu, *J. Phys. Chem. B*, 2001, **105**, 12293–12296.
- 67 A. Tiwari, A. K. Mishra, H. Kobayanshi and A. P. F. Turner, *Contemp. Phys.*, 2013, **54**, 74.
- 68 J. Kalinowskia, W. Stampor, M. Cocchi, D. Virgili and V. Fattori, *J. Appl. Phys.*, 2006, **100**, 034318.
- 69 A. M. Ortega, M. T. Lopez, F. C. Garcia, J. A. Reglero, M. M. Alonso, G. Espino and J. M. Garcia, *Eur. Polym. J.*, 2017, **95**, 119–126.

