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All-in-one NMR spectroscopy of small organic molecules: complete chemical shift assignment from a single NMR experiment†

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A new class of NOAH NMR experiments (NOAH-AST and NOAH-AST_{PS}), with the abbreviations, A: 1,1-ADEQUATE, S: sensitivity improved version of multiplicity-edited (ME)-HSQC, T: TOCSY, and T_{PS}: pure shift TOCSY, are reported to obtain complete chemical shift assignments of small organic molecules from a single NMR experiment. While NOAH-AST provides ¹³C-¹³C, ¹H-¹³C, and ¹H-¹H connectivities for molecules with well resolved chemical shifts, NOAH-AST_{PS} experiments discern ¹H-¹H connectivities even in complex organic molecules such as steroids at ultra-high resolution. These methods are very flexible and allow to record data through non-uniform-sampling, which reduces the experimental time to a great extent. In order to make these methods friendly to non-NMR experts (especially organic chemists and natural product scientists), python scripts have been developed and they help researchers in using these methods.

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Introduction

In nuclear magnetic resonance (NMR) spectroscopy, chemical shift assignment is the primary and foremost task. Typically, in most cases, a set of 2D experiments are recorded to obtain this chemical shift information. Recently, it has been shown that, a whole set NMR data can be acquired in a single experiment by considering (i) NOAH (NMR by ordered acquisition using ¹H detection,¹ including multiple-receivers)² or (ii) MFA (multiple-FID acquisition)³ schemes. Both these methods use only one recycle delay block for all the experimental modules of the different pulse schemes combined. Therefore, experimental times are significantly reduced compared to the time taken to record the combined individual experiments, separately. In the case of NOAH pulse schemes, the unutilized magnetization of the *i*th experimental module is utilized in the (*i*+1)th experimental module. Whereas, in the MFA experiments, both the *i*th and (*i*+1)th experiments share common indirect evolution times. They both have their own inherent flexibilities in concatenating a set of experimental modules in a single experiment and several of such combinations have been proposed. ASAP versions have also been reported to minimize the relaxation artefacts in the NOAH schemes.⁴ Very recently our group has also reported a NOAH-based PROSMASH-HSQC² scheme⁵ for monitoring the small molecule–protein interactions in a single

experiment, wherein very high chemical shift resolution for the small molecule signals is achieved with the aid of real-time homonuclear broadband BIRD decoupling.⁶

Often, rigid ring systems are a part of many organic natural products such as steroids, terpenoids, macrocyclic antibiotics, and alkaloids. For such spin-systems, recording a 1,1-ADEQUATE^{7–9} experiment alone provides the complete backbone ¹³C-¹³C connectivity. The 1,1-ADEQUATE experiment filters the double-quantum ¹³C-¹³C magnetization, *i.e.*, at the natural abundance levels of carbon, it utilizes only one active spin-pair (¹³C-¹³C) out of 10 000 spin-pairs of carbons. Hence, the similar proportion of ¹H magnetization is only used for that experiment; therefore, the remaining magnetization which is not a part of the double-quantum ¹³C-¹³C spin-pair is discarded. Now, in the present work, in order to use that remaining magnetization, we have proposed a novel method, and named it as NOAH-AST. In general, along with the 1,1-ADEQUATE experiment, there is an immense need to record the multiplicity-edited (ME)-HSQC¹⁰ experiment, as it discriminates the multiplicity of CH, CH₂, and CH₃ carbons and the respective attached protons. Further, to verify ¹H-¹H spin network, we need to record either a COSY or a TOCSY¹¹ experiment. Therefore, by combining 1,1-ADEQUATE (A), sensitivity improved version of ME-HSQC (S), and TOCSY (T) in a single experiment (AST) in a NOAH fashion, NOAH-AST delivers the complete chemical shift details of various classes of organic molecules; needless to say that sufficient concentrations are required to be able to work at natural abundance of ¹³C.

Previously, NOAH version of HMBC-HSQC-COSY/TOCSY¹ and HMBC-H2OBC¹² experiments have been reported to achieve similar purposes; however, the 1,1-ADEQUATE module in

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the NOAH-AST experiment provides several significant benefits when compared to the HMBC/H2OBC pulse modules: (i) in the case of HMBC there is no specificity/directionality in showing the chemical shift correlations, sometimes it shows correlations up to 4–5 bonds, whereas 1,1-ADEQUATE shows correlations only for the adjacent carbons; therefore, attributing chemical shifts for different carbon atoms is easier than in the HMBC. (ii) Interestingly, H2OBC experiment is useful to obtain specific correlations between two-bond separated C–H pairs, but it is not useful in monitoring the correlations of carbonyls with the other spins; however, such information can easily be obtained from the 1,1-ADEQUATE. (iii) Notwithstanding the benefits of HMBC, in terms of its capability in showing the correlations with quaternary carbons, it suffers from significant resolution issues along the proton dimension as each carbon correlates with several protons. In such cases, it is rather difficult to monitor the precise chemical shift correlations, especially for molecules which exhibit complex proton NMR spectra. In this regard, 1,1-ADEQUATE outperforms both the H2OBC and HMBC, since, the H2OBC does not show correlations with quaternary carbons (iv).

Steroidal molecules often exhibit reasonable ^{13}C chemical shift resolution; thus, 1,1-ADEQUATE is a good choice over the H2OBC experiments, as the carbonyl chemical shift can be used as a starting point while analyzing the data. (v) Further, as 1,1-ADEQUATE shows relatively fewer chemical shift correlations

along the ^1H dimension when compared with the HMBC, it enables ^1H chemical shift deconvolution which would be at least at one of the ^1H sites of diastereotopic protons.

One particular concern with regard to 1,1-ADEQUATE is its low sensitivity, which may demand higher sample concentrations, in general. However, in the light of the above mentioned benefits, the low sensitivity of 1,1 ADEQUATE can be tolerated. Moreover, at the current times, the availability of high sensitivity cryogenic probes, reduces the severity of the sensitivity issue of the 1,1-ADEQUATE experiment, in practical terms. In fact, in 2015, applications of 1,1-ADEQUATE experiment have been demonstrated only with 0.7 mg of cryptospirolepine (molecular weight $\sim 504 \text{ g mol}^{-1}$).¹³ This reveals that by having access to such probes, low intrinsic sensitivity of 1,1-ADEQUATE would no longer be an issue of concern, and structural studies of any synthetic organic/natural product molecules available at very small amounts can be carried out. Hence, the present NOAH-AST experiments will also take the sensitivity benefits of 1.7 mm probes in the similar way.

Often, the chemical shift resolution in the TOCSY is impaired due to the severe overlap of ^1H - ^1H scalar multiplets; this can be circumvented by invoking ideas of homodecoupling in the TOCSY block of the NOAH-AST. Therefore, NOAH-AST_{PS} experiments have also been proposed; wherein T_{PS} represents the PSYCHE pure shift TOCSY.¹⁴

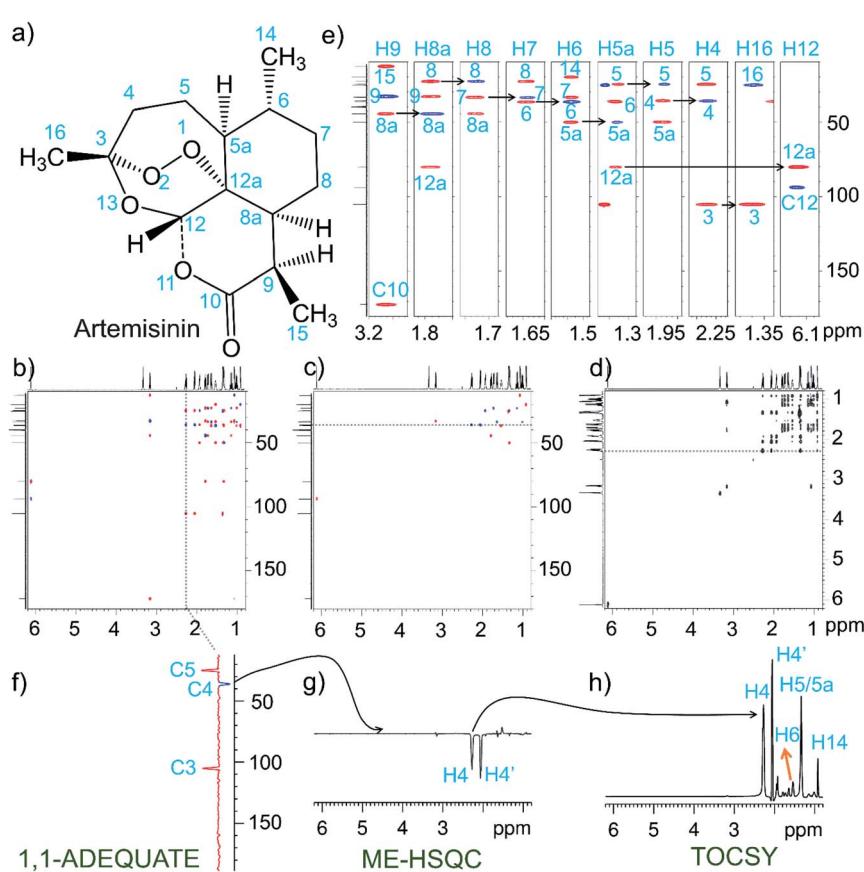


Fig. 1 1,1-ADEQUATE (b, e and f), ME-HSQC (c and g), and TOCSY (d and h) spectra obtained from a single NOAH-AST recorded on artemisinin (a) molecule ($\sim 140 \text{ mM}$ in DMSO-D_6), using an 800 MHz magnet equipped with cryoprobe.



Results and discussion

NOAH-AST NMR of artemisinin

Fig. 1 shows the NOAH-AST spectra recorded on the anti-malarial natural product drug, artemisinin (Fig. 1a). From the 1D projections which are added on the two-dimensional spectra, it is clear that ^1H and ^{13}C chemical shift resolution is sufficient to record the simple version of NOAH-AST experiment. The resultant 1,1-ADEQUATE, ME-HSQC, and TOCSY spectra are respectively shown in Fig. 1b, c, and d. Complete ^{13}C assignments can be obtained from the 1,1-ADEQUATE experiment while using the carbonyl chemical shift as a starting point. For better understanding of the analysis, strip plots are shown (Fig. 1e) with appropriate labelling. Initially, the analysis has been started by considering the strip which has carbonyl carbon and it has been assigned to H9, wherein the self-correlation, *i.e.*, H9–C9 is the negative peak and the remaining correlations are positive peaks. At this ^1H chemical shift (~ 3.2 ppm), two other correlations to carbons are seen; the upfield carbon chemical shift is for C15, and the other peak is for C8a. For the subsequent analysis, C8a is used as a starting point and we performed the analysis in a similar fashion. However, in order to have an unambiguous walk in the analysis process, we have to identify the multiplicity of carbons (multiplicity, $-\text{CH}$, CH_2 , and CH_3), which can be obtained from the ME-HSQC module of NOAH-AST module. For example, the chemical shift of H4 proton is used to identify the C4, C5, and C3 carbon chemical shifts

(Fig. 1f), and subsequently the ME-HSQC (Fig. 1g) has enabled attributing the multiplicity pattern to $-\text{CH}_2$. This can clearly be seen from the 1D internal projections provided. Finally, the TOCSY spectrum that has been obtained from the NOAH-AST is used to establish the ^1H spin system, *e.g.*, breaks in the TOCSY correlations provide insights about the presence of quaternary carbons and hetero-atoms. A similar analysis has been performed on the alkaloid, strychnine (see ESI, Fig. S1†).

NOAH-AST_{PS} NMR of estradiol

Next, NOAH-AST_{PS} experiments are demonstrated using the estradiol molecule (Fig. 2). As has been seen in the previous example, here too, whole ^{13}C chemical shift assignments are achieved from the 1,1-ADEQUATE data set (Fig. 2b, g and f). This molecule has only one methyl group (C18) on one of the quaternary carbons (C13), which is used as a starting point for the analysis. Next, the strip at H17 is considered, as it has a hydroxyl group and C12 is bonded with C13. This facilitated identification of C16, and then proceeding in a similar fashion the ^{13}C and ^1H analysis (Fig. 2f) could be completed. However, identification of both the diastereotopic protons from the 1,1-ADEQUATE experiment is not often possible as is the case for estradiol too. At least in this kind of steroid molecule, the presence of functional groups and aromatic rings facilitates discrimination of two diastereotopic protons with different chemical shifts. Thus, the well-separated proton chemical shifts

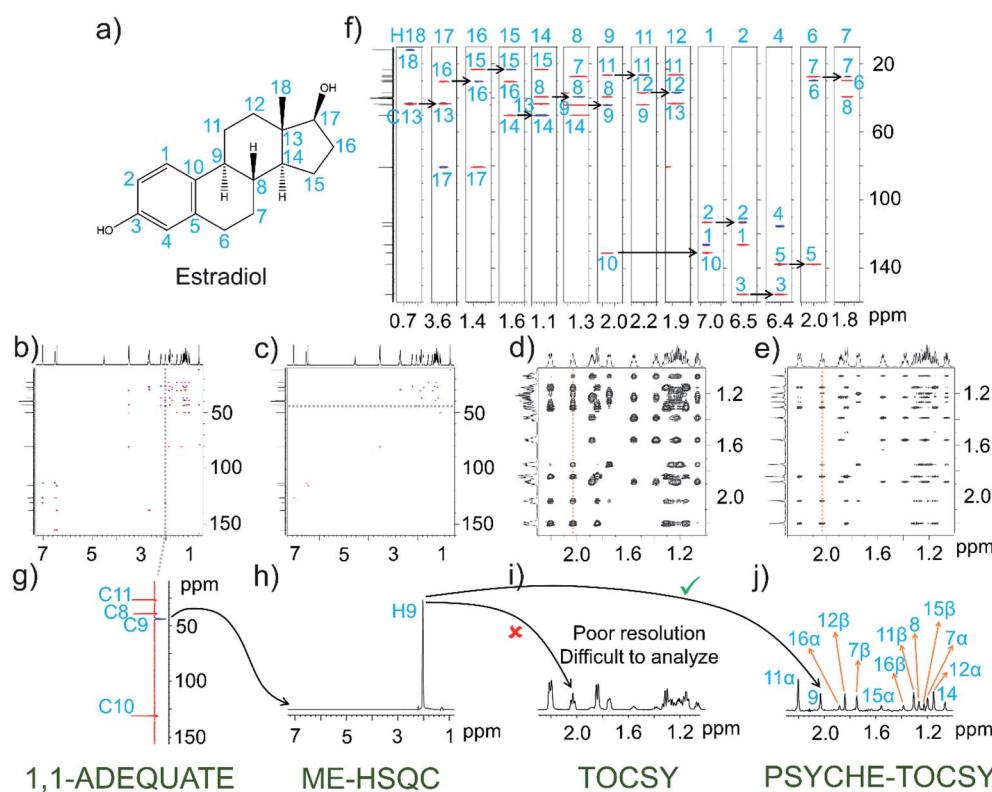


Fig. 2 1,1-ADEQUATE (b, f and g), ME-HSQC (c and h), and PSYCHE-TOCSY (e and j) spectra obtained from a single NOAH-AST_{PS} recorded on estradiol (a) molecule (~ 370 mM in DMSO-D_6), using an 800 MHz magnetic field strength. On the other hand, TOCSY (d and i) spectra is recorded from the NOAH-AST experiment.



Table 1 Typical time requirements to obtain the 1,1-ADEQUATE, ME-HSQC, and TOCSY spectra in independent and NOAH-AST (mixture of D-glucose and D-xylose) fashion while using different number of scans (which are given in the parenthesis)

Experiment type	1,1-ADEQUATE	ME-HSQC	TOCSY	Total experimental time
Independent	1 h 30 min (16 scans)	25 min (4 scans)	25 min (4 scans)	2 h 20 min
Independent	1 h 30 min (16 scans)	1 h 20 min (16 scans)	1 h 20 min (16 scans)	4 h 10 min
NOAH-AST	1 h 40 min (16 scans)			1 h 40 min

belonging to one of the diastereotopic chemical sites can be used for the spectral interpretation and it eventually provides the carbon chemical shift assignments. The remaining proton chemical shifts which have not been analyzed from the 1,1-ADEQUATE experiment can be obtained subsequently from the ME-HSQC module with multiplicity editing (Fig. 2c and h). Understanding the complete proton network connectivity is not possible from the conventional TOCSY experiment alone in steroid molecules (Fig. 2d). For example, the 1D-internal trace obtained for the H9 proton shows severe overlap in scalar-coupled multiplets that hampers the analysis (Fig. 2i). This necessitates recording the homodecoupled PSYCHE-TOCSY which results in ^1H singlets that simplify proton connectivity analyses (Fig. 2e and j). However, in order to obtain such very high resolution along the indirect dimensions of pure shift TOCSY, the NOAH-AST_{PS} experiment had to be recorded for almost 7 hours on 50 mg of estradiol (with more indirect increments). In such cases, invoking the ideas of non-uniform-

sampling (NUS)¹⁵ certainly helps in decreasing the experimental time to a significant extent. The NUS based sampling has been implemented for NOAH-AST_{PS} experiments, and 25% NUS sampling has taken only \sim 1 hour 45 min of instrument time, yet the spectral quality of NUS-NOAH-AST_{PS} is comparable to that of the NOAH-AST_{PS} spectra recorded in the conventional manner for about \sim 7 hours (see ESI, Fig. S2†). In the present case, while recording NOAH-AST_{PS} experiments, due to the low sensitivity of T_{PS} module, a large number of scans are required; this is also true for the 1,1-ADEQUATE module. Since both these modules require practically a similar number of scans to get desirable signal to ratio (SNR), combining them in a single NOAH module is worthwhile.

NOAH-AST NMR of mixture of D-glucose and D-xylose

Often, chemical constituents are present in a wide concentration range in the sample (leading to dynamic range problems in NMR). In such situations, NOAH-AST family of experiments

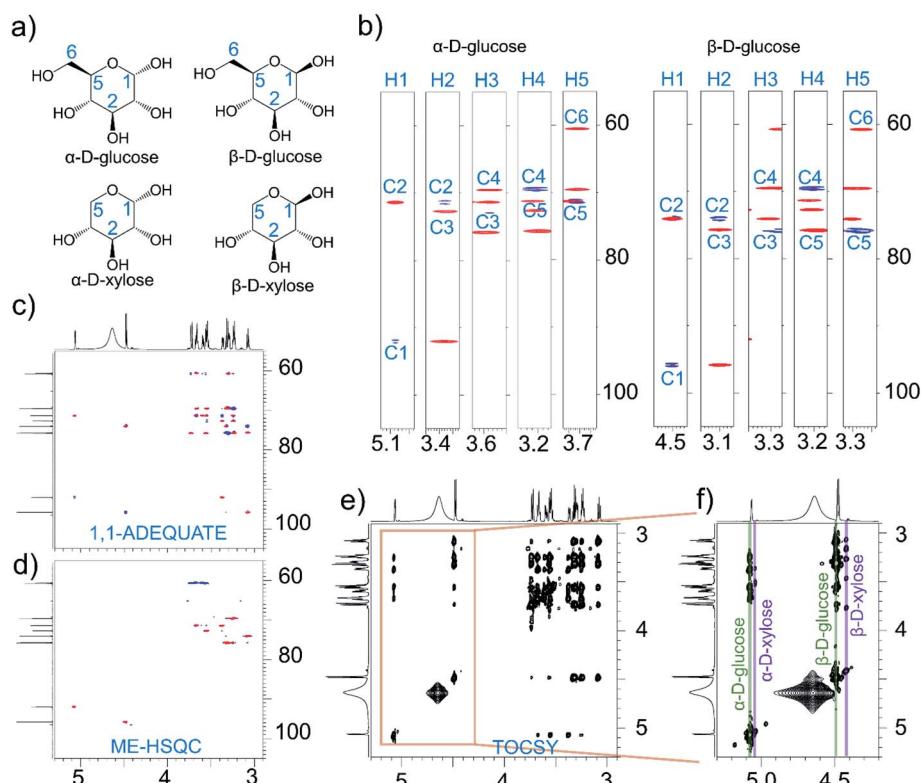


Fig. 3 1,1-ADEQUATE (b and c), ME-HSQC (d), and TOCSY (e and f) spectra obtained from a single NOAH-AST recorded on a mixture of D-glucose (\sim 110 mM) and D-xylose (\sim 6 mM) molecules (a) dissolved in D_2O , using an 800 MHz magnetic field strength.



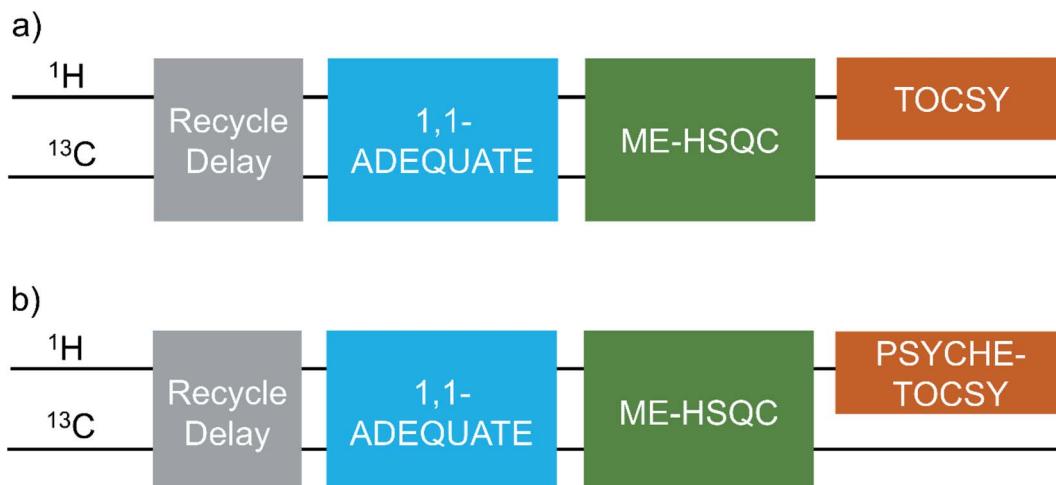


Fig. 4 Schematic of NOAH-AST (a) and NOAH-AST_{PS} (b) pulse schemes developed in the present study. Full pulse schemes and experimental details are given in the ESI.[†]

have particular advantages. This arises from the fact that all the three experiments are recorded using only one relaxation delay time; if the three experiments were to be recorded independently, each one would require the delay time at the beginning of every scan. In the NOAH experiments, the number of scans required will be dictated by the least sensitive of the experiments; in the present case, 1,1 ADEQUATE. Table 1 shows typical time requirements for the three experiments being done in NOAH style *versus* being done independently with different scans, as dictated by the sensitivity of the experiments. The first row assumes that 16 scans are required for 1,1 ADEQUATE and 4 scans each are enough for ME-HSQC and TOCSY. The second row assumes 16 scans for all of them, as in the NOAH-AST experiment. It is clearly seen from the last column that NOAH-AST has better performance in terms of time with respect to independent measurements as in row 1, and additionally it would have better signal-to-noise ratio for the ME-HSQC and TOCSY experiments. Thus, even constituents present in smaller concentration would get detected at least in the more sensitive of the three experiments.

This application is demonstrated using a mixture of *D*-glucose (10 mg, ~110 mM) and *D*-xylose (0.5 mg, ~6 mM) in 0.5 ml of D₂O (Fig. 3). In the resultant 1,1-ADEQUATE (Fig. 3b and c) spectrum of NOAH-AST, only the resonances belonging to *D*-glucose are observed, since the concentration of *D*-xylose anomers is significantly less to detect. However, signals of *D*-xylose are seen in the ME-HSQC (Fig. 3d) and TOCSY spectra (Fig. 3e and f). Then, if required, one can record the 1,1-ADEQUATE spectrum only of *D*-xylose separately using higher concentration if possible.

NOAH-AST and NOAH-AST_{PS} NMR pulse sequences

The pulse schemes developed in the present study, NOAH-AST and NOAH-AST_{PS} are shown respectively in the Fig. 4a and b, as block diagrams. The full pulse schemes (Fig. S3 and S4[†]) and experimental details are given in the ESI.[†] These NOAH-AST and NOAH-AST_{PS} experiments require significant user intervention

in optimization, which is not simple for non-NMR specialists. Hence, we wrote two python scripts (see ESI[†]), which take care of the entire experimental setup with minimal user intervention.

Conclusions

Overall, the present work describes development of a new set of NOAH pulse sequences, NOAH-AST and NOAH-AST_{PS}. The applications of these methods are demonstrated on different kinds of organic molecules, artemisinin (terpene), strychnine (alkaloid), estradiol (steroid), and a mixture of *D*-glucose and *D*-xylose (monosaccharides). As has been demonstrated herein, complete ¹H and ¹³C chemical shift assignments could be obtained from a single NOAH-AST experiment (¹³C-¹³C from 1,1-ADEQUATE, ¹H-¹³C from ME-HSQC, and ¹H-¹H from TOCSY), which works well for the well-resolved chemical shifts. On the other hand, for molecules which exhibit complex NMR spectra, NOAH-AST_{PS} experiments can provide pure shift TOCSY. This NOAH-AST_{PS} experiment demands a larger number of indirect increments (having PSYCHE homodecoupling), which can be reduced to a significant extent by invoking a non-uniform-sampling approach. The present set of NOAH-AST/AST_{PS} experiments may require rather high concentrations of samples due to intrinsic low sensitivity of the 1,1-ADEQUATE experiment. However, this limitation would be overcome by having access to special probes such as cryogenically cooled 1.7 mm probe. On the other hand, if samples are available at moderate concentrations, acquiring NOAH-AST with more scans helps in obtaining the desired results in an overnight experiment, even without such special probes. Thus, the proposed NOAH methods may provide an efficient alternative to acquisition of several 2D data sets on small and complex organic molecules.

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Conflicts of interest

The authors declare no competing financial interest.

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References

- 1 Ě. Kupče and T. D. W. Claridge, *Angew. Chem., Int. Ed.*, 2017, **56**, 11779–11783.
- 2 Ě. Kupče and T. D. W. Claridge, *J. Magn. Reson.*, 2019, **307**, 106568.
- 3 K. Motiram-Corral, M. Pérez-Trujillo, P. Nolis and T. Parella, *Chem. Commun.*, 2018, **54**, 13507–13510.
- 4 E. Kupče and T. D. W. Claridge, *Chem. Commun.*, 2018, **54**, 7139–7142.
- 5 V. M. R. Kakita, K. Rachineni, M. Bopardikar and R. V. Hosur, *J. Magn. Reson.*, 2018, **297**, 108–112.
- 6 L. Paudel, R. W. Adams, P. Kiraly, J. A. Aguilar, M. Foroozandeh, M. J. Cliff, M. Nilsson, P. Sandor, J. P. Walther and G. A. Morris, *Angew. Chem., Int. Ed.*, 2013, **52**, 11616–11619.
- 7 B. Reif, M. Koeck, R. Kerssebaum, H. Kang, W. Fenical and C. Griesinger, *J. Magn. Reson., Ser. A*, 1996, **118**, 282–285.
- 8 B. Reif, M. Koeck, R. Kerssebaum, J. Schleucher and C. Griesinger, *J. Magn. Reson., Ser. B*, 1996, **112**, 295–301.
- 9 M. Koeck, R. Kerssebaum and W. Bermel, *Magn. Reson. Chem.*, 2003, **41**, 65–69.
- 10 R. D. Boyer, R. Johnson and K. Krishnamurthy, *J. Magn. Reson.*, 2003, **165**, 253–259.
- 11 J. Cavanagh and M. Rance, *J. Magn. Reson.*, 1990, **88**, 72–85.
- 12 T. M. Nagy, T. Gyöngyösi, K. E. Kövér and O. W. Sørensen, *Chem. Commun.*, 2019, **55**, 12208–12211.
- 13 J. Saurí, W. Bermel, A. V. Buevich, E. C. Sherer, L. A. Joyce, M. H. Sharaf, P. L. Schiff Jr, T. Parella, R. T. Williamson and G. E. Martin, *Angew. Chem., Int. Ed.*, 2015, **54**, 10160–10164.
- 14 M. Foroozandeh, R. W. Adams, M. Nilsson and G. A. Morris, *J. Am. Chem. Soc.*, 2014, **136**, 11867–11869.
- 15 T. D. W. Claridge, M. Mayzel and Ě. Kupče, *Magn. Reson. Chem.*, 2019, **57**, 946–952.

