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Access to N-unprotected 2-amide-substituted indoles from Ugi adducts via palladium-catalyzed intramolecular cyclization of o-iodoanilines bearing furan rings†

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A variety of N-unprotected 2-amide-substituted indoles were synthesized from readily available furfuralbased Ugi adducts in moderate to good yields via palladium-catalyzed intramolecular cyclization of oiodoanilines bearing furan rings. These reactions involved a cascade sequence consisting of dearomatizing arylation, opening of the furan ring, and deprotection of the N atom.

Polyfunctionalized indoles, including 2-amide-substituted indoles, are privileged motifs in medicinal chemistry and synthetic organic chemistry.1 The indole ring is probably the most common heterocycle found in natural products and pharmaceuticals,2 and functionalized indoles are versatile building blocks for the preparation of structurally complex and novel indolines, many of which show potent bioactivities (Fig. 1).3 Thus, much effort has been devoted to the development of strategies for the synthesis and functionalization of indoles and their derivatives.4 Among them, the most attractive routes are those involving transition-metal-catalyzed intermolecular or intramolecular cyclization of o-haloanilines with alkenes,⁵ alkynes,⁶ or allenes.⁷ Despite the attractiveness of these routes, it would be desirable to develop efficient catalytic methods for the preparation of functionalized indoles from o-haloanilines and furans, which are readily available, alternatives to alkenes for diversity-oriented synthesis strategies.8,9

We speculated that Ugi adducts might be useful for this purpose. Ugi reactions involve four components—an aldehyde or ketone, an isocyanide, an amine, and a carboxylic acidand afford a diverse array of functionalized α-acylamino amides, 10 which can be subjected to a wide variety of postcondensation transformations to achieve further structural diversity.11 Recently, we and other groups developed a route to functionalized indoles via palladium-catalyzed intramolecular arylative dearomatization of 2-bromo-N-(furan-2-ylmethyl) anilines. 5f,12 In this paper, we report a convenient protocol for the synthesis of α -amide-substituted indoles via palladium-

catalyzed intramolecular arylative cyclization of furans that were generated by Ugi reactions of furfurals and o-haloanilines (Scheme 1).

The success of this protocol relies on suppression of the following side reactions: β-arylation of the furan ring, protonation of the ArI, and intramolecular C-N coupling. With this in mind, we chose *N*-(*tert*-butyl)-2-(furan-2-yl)-2-(*N*-(2-iodophenyl) acetamido)acetamide (1a)-which was prepared by means of a Ugi reaction of furfural, 2-iodoaniline, acetic acid, and tertbutyl isocyanide-as the substrate for optimization of the reaction conditions. We were pleased to find that upon treatment of 1a with Pd(PPh₃)₄ (0.05 equiv.), PPh₃ (0.1 equiv.), and K_2CO_3 (2 equiv.) in 1,4-dioxane at 70 °C for 12 h, polysubstituted N-unprotected indole 2a was obtained in 30% yield along with unidentified by-products (Table 1, entry 1). This transformation clearly involved a cascade sequence consisting of arylation, ringopening, and N-deacetylation. The in situ N-deacetylation is particularly interesting and useful and may have resulted from the weaker nucleophilicity of the N atom of the indole ring

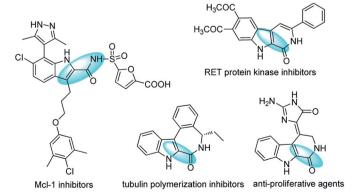


Fig. 1 Bioactive 2-amide-substituted indoles

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relative to that of the amide N of 1a. Other bases (Cs₂CO₃, NaHCO₃, Na₂CO₃, and 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU)) were also tested, but 2a was not detected in any of these reactions (entries 2-5). Stronger base of Cs₂CO₃ resulted in side-reaction of C-N coupling. NaHCO₃ and Na₂CO₃ as the base mostly led to the protonated product. DBU led to no reaction. Next, we attempted to improve the yield of 2a by increasing the reaction temperature (entries 6–9), and an 89% yield was obtained at 110 °C. Screening of various ligands other than PPh₃ failed to produce better results (entries 10-13), and Pd(PPh₃)₄ was the optimal catalyst (compare entry 2 with entries 14-17). Evaluation of other solvents (THF, toluene, and DMSO) did not improve the yield (entries 18-20). Therefore, we concluded that the optimal conditions involved the use of $Pd(PPh_3)_4$ (0.05 equiv.) as the catalyst, K_2CO_3 (2.0 equiv.) as the base, 1,4-dioxane as the solvent, and 110 °C as the reaction temperature.

With the optimized conditions in hand, we prepared a series of Ugi adducts 1 with various R^1-R^4 groups and a furan moiety in moderate yields, and we subjected the resulting compounds to the arylative cyclization conditions to investigate the substrate scope (Table 2). In all cases, the reaction proceeded smoothly to afford corresponding indoles 2 in moderate to good isolated yields (40–77%). Specifically, with $R^1 = H$, $R^2 = Me$, and $R^4 = t$ -Bu, several R^3 groups (H, Me,

Scheme 1 Pd-catalyzed approaches to polyfunctionalized indoles from o-haloanilines.

2-amide-indole

F, and Cl) were screened and found to provide corresponding indolyl aldehydes 2a-2d in 45-66% yields (entries 1-4). Reaction of 1c, which bears an electron-withdrawing 4-F group, gave a substantial amount of a by-product generated by protonation without opening of the furan ring, which resulted in a relatively low yield of 2c (45%). Similarly, with R¹ = Me, R^2 = Me, and R^4 = t-Bu, compounds with H, Me, MeO, and CF₃ at R^3 afforded 2e-2h in 60-77% yields (entries 5-8). Substrate 1h, which has an electron-withdrawing 4-CF₃ at R³, gave a lower yield (60%) than the other three substrates. In addition to H or Me, R¹ could be Ph or 4-Me-Ph: 2i and 2j were obtained in 67% and 72% yields, respectively (entries 9 and 10). Notably, when R² was an aryl group (4-MeO-Ph), 2e was produced in 77% yield (entry 11). In contrast, when R³ was n-Pr, the yield of 2e was only 40% (entry 12). Finally, when R⁴ was cyclohexyl, 2m-2o were obtained in good yields (entries 13-15).

Products 2 bear amide, carbonyl and alkenyl functional groups, all of which are amenable to numerous further

Table 1 Optimization of reaction conditions

Entry	[Pd]	Ligand	Base	T (°C)	Yield ^b (%)
1	Pd(PPh ₃) ₄	PPh_3	K ₂ CO ₃	70	30
2	$Pd(PPh_3)_4$	PPh_3	Cs_2CO_3	70	ND
3	$Pd(PPh_3)_4$	PPh_3	NaHCO ₃	70	ND
4	$Pd(PPh_3)_4$	PPh_3	Na_2CO_3	70	ND
5	$Pd(PPh_3)_4$	PPh_3	DBU	70	ND
6	$Pd(PPh_3)_4$	PPh_3	K_2CO_3	80	31
7	$Pd(PPh_3)_4$	PPh_3	K_2CO_3	100	44
8	$Pd(PPh_3)_4$	PPh_3	K_2CO_3	110	89
9	$Pd(PPh_3)_4$	PPh_3	K_2CO_3	120	25
10	$Pd(PPh_3)_4$	DPPP	K_2CO_3	110	18
11	$Pd(PPh_3)_4$	DPPB	K_2CO_3	110	19
12	$Pd(PPh_3)_4$	DPPF	K_2CO_3	110	12
13	$Pd(PPh_3)_4$	Xantphos	K_2CO_3	110	48
14	$Pd_2(dba)_3$	PPh_3	K_2CO_3	110	50
15	$Pd(OAc)_2$	PPh_3	K_2CO_3	110	18
16	$Pd(PPh_3)_2Cl_2$	PPh_3	K_2CO_3	110	54
17	$Pd(CH_3CN)_2Cl_2$	PPh_3	K_2CO_3	110	31
18 ^c	$Pd(PPh_3)_4$	PPh_3	K_2CO_3	110	45
19^d	$Pd(PPh_3)_4$	PPh_3	K_2CO_3	110	21
20^e	$Pd(PPh_3)_4$	PPh_3	K_2CO_3	110	66

Reaction conditions: 1a (0.2 mmol), catalyst (0.05 equiv.), ligand (0.1 equiv.), and base (2 equiv.) in 2.0 mL of 1,4-dioxane were allowed to react under nitrogen for 12 h. DBU, 1,8-diazabicyclo[5.4.0]undec-7-ene; DPPP, 1,3-bis(diphenylphosphino)propane; DPPB, 1,4-bis(diphenylphosphino)butane; DPPF, 1,1'-bis(diphenylphosphino) ferrocene; xantphos, 4,5-bis(diphenylphosphino)-9,9-dimethylkanthene. b Yields were determined by H NMR spectroscopy.
ND = not detected. THF was the solvent. d Toluene was the solvent. DMSO was the solvent.

Table 2 Substrate scope⁶

Entry	R^1	R^2	\mathbb{R}^3	\mathbb{R}^4	1 (% yield ^b)	2 (% yield ^b)
1	Н	Me	Н	<i>t</i> -Bu	1a (50)	2a (66)
2	Н	Me	Me	t-Bu	1b (45)	2b (63)
3	Н	Me	F	t-Bu	1c (52)	2c (45)
4	Н	Me	Cl	t-Bu	1d (41)	2d (64)
5	Me	Me	H	t-Bu	1e (42)	2e (77)
6	Me	Me	Me	<i>t</i> -Bu	1f (42)	2f (63)
7	Me	Me	MeO	t-Bu	1g (40)	2g (70)
8	Me	Me	CF_3	<i>t</i> -Bu	1h (40)	2h (60)
9	Ph	Me	Н	t-Bu	1i (46)	2i (67)
10	<i>p</i> -Tolyl	Me	Н	<i>t</i> -Bu	1j (33)	2j (72)
11	Me	PMB	H	<i>t</i> -Bu	1k (55)	2e (77)
12	Me	<i>n</i> -Pr	Н	<i>t</i> -Bu	1l (32)	2e (40)
13	Me	Me	H	Cy	1m (57)	2m (50)
14	Me	Me	MeO	Су	1n (53)	2n (61)
15	Me	Me	CF_3	Cy	10 (42)	20 (66)

 $[^]a$ Reaction conditions: 1 (0.2 mmol), catalyst (0.05 equiv.), ligand (0.1 equiv.), and base in 2.0 mL solvent were allowed to react at 110 $^{\circ}$ C for 12 h. Cy, cyclohexyl. b Isolated yields are given.

transformations that can be used to prepare structurally diverse indoles. For example, hydrogenation of the double bonds of 2e–2g and 2i afforded the corresponding products (3e–3g and 3i) in good yields (Scheme 2).

In Scheme 3, we depict two possible pathways for this transformation (electrophilic palladation and carbopalladation) on the basis of the above-described experimental results and previously reported results regarding arylation of furans. Pecifically, an oxidative addition reaction between aryl iodide 1 and palladium(0) forms intermediate A. Intramolecular electrophilic palladation of the furan ring of A at the α -position results in the generation of intermediate B, which undergoes base-mediated furan ring-opening and β -elimination to afford intermediate C. A reductive elimination reaction of C provides F and palladium(0), completing the catalytic cycle. Deprotection of F yields 2. Alternatively, A undergoes carbopalladation to form intermediate D, which isomerizes to π -allylic palladium complex E. Ring-opening of E produces F.

Scheme 2 Hydrogenation of 2.

Scheme 3 Possible pathway for the formation of 2.

Conclusions

In summary, we have developed a protocol for the synthesis of *N*-unprotected 2-amide-substituted indoles by means of Pd-

catalyzed dearomatizing intramolecular arylation reactions of readily available furfural-based Ugi adducts. This protocol involves an intramolecular condensation of an *o*-haloaniline bearing a furan ring and a subsequent cascade involving dearomatizing arylation, opening of the furan ring, and *N*-deprotection. The bioactivities of the obtained polysubstituted indoles are being explored in our laboratory, and the results will

Conflicts of interest

be reported in due course.

Paper

There are no conflicts to declare.

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