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## Synthesis of anti-allergic drugs

Histamine is formed by the decarboxylation of histidine catalyzed by enzymes. It is an endogenous biologically active substance involved in multiple complex physiological processes as an important chemical transmitter. Histamine receptors have four subtypes,  $H_1$ ,  $H_2$ ,  $H_3$  and  $H_4$ , all of which are G protein coupling receptors (GPCRs) with different physiological functions. Histamine plays an important role in the pathophysiological mechanism of allergic diseases, and the antagonistic effect of histamine has become an important way to study anti-allergic drugs, wherein the anti-allergic drugs used in clinical practice are mainly  $H_1$  receptor antagonists. Currently, there are many varieties of  $H_1$  receptor antagonists in clinical applications, which can be divided into ethylenediamine antagonists, amino ether antagonists, propylamine antagonists, tricyclic antagonists, piperazine antagonists and piperidine antagonists depending on their chemical structures. This article mainly reviews the research progress of allergic reactions with histamine  $H_1$  receptor antagonists and expounds the important aspects of the design and synthesis of various new compounds.

### 1. Introduction

Histamine, chemically known as 4(5)-(2-aminoethyl)imidazole, is an endogenous biologically active substance involved in multiple complex physiological processes as an important chemical transmitter. Histamine is formed by the decarboxylation of histidine catalyzed by enzymes.  $^{1-6}$  It is widely found in a variety of plants, animals and microorganisms in nature.  $^{7-9}$  In animals, histamines are found in peripheral tissues and in peripheral and central nerves. Histamine is released under endogenous and exogenous stimulations and interacts with receptors. Histamine receptors have four subtypes,  $H_1$ ,  $H_2$ ,  $H_3$  and  $H_4$ , all of which are G protein coupling receptors (GPCRs) with different physiological functions.  $^{10-13}$ 

H<sub>1</sub> receptors are distributed in endothelial cells, smooth muscle, adrenal medulla, heart and central nervous system. Exogenous antigens can cause anaphylaxis, combine with antibody immunoglobulin on target cell mast cell nucleus and granulocyte, change the function of target cell membrane, release histamine and other allergic media, and make histamine distribute in H<sub>1</sub> receptor function of tissues and organs.<sup>14</sup> After the activation of the H<sub>1</sub> receptor, phosphatidase C is activated through G proteins, which promotes the increase in Ca<sup>2+</sup> concentration, thereby increasing the vasodilation and capillary permeability, leading to plasma exudation, local tissue redness and swelling, and allergic symptoms of bronchial and

gastrointestinal smooth muscle contraction.15-17 H2 receptors are distributed in gastric wall cells, vascular smooth muscles, heart and central nervous system. H2 receptor stimulation could promote gastric acid secretion, stimulate the heart and inhibit uterine contraction.18 Histamine, acetylcholine and gastrin receptors could be found on the surface of gastric mucosa cells. When the corresponding ligands interact with these receptors, they could activate the secretion of gastric acid. The H<sub>3</sub> receptor is located in the central and peripheral nerve endings of the presynaptic membrane, and it could function as their receptor, inhibiting histamine release and synthesis. 19-23 H3, as a different receptor, controls the release of other neurotransmitters. H<sub>3</sub> receptor has important physiological functions, such as cardiac function, gastric acid secretion, anaphylaxis, sleep wake cycle control, cognition and memory, convulsion, diarrhea or reducing gastric acid secretion.24-30 The H4 receptor is only found in the small intestine, spleen, thyroid and immunocompetent cells. The H<sub>4</sub> receptor is related to the regulation of immune functions, such as allergic reactions, asthma, cancer and other immune diseases.31-35

Anaphylaxis is an allergic disease that occurs when the body is exposed to specific allergens.<sup>36-39</sup> It could be caused by a variety of exogenous substances like allergens that include heterologous serum (such as tetanus antitoxin), certain animal proteins (such as that of fish, shrimp, and crabs), bacteria, viruses, parasites, animal fur, plant pollen, dust mites in the air, and chemicals and drugs. Allergens could stimulate human B cells to produce immunoglobulin E, which combines with antibodies on human mast cells and sensitized cells, damages the cell membrane and leads to degranulation, and releases histamine, 5-hydroxytryptamine, leukotrienes, bradykinin and other active substances in the granules.<sup>40-46</sup> The allergic reaction

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The structural modification of loratadine analogues 4a-h.

can manifest as skin redness and swelling, rash, itching, patches as well as allergic rhinitis, bronchial asthma, laryngeal edema and other symptoms. The clinical manifestations of anaphylaxis include decreased blood pressure, increased heart rate, pale skin, edema, and shock and even death in severe cases.47-51

Histamine plays an important role in the pathophysiological mechanism of allergic diseases, and the antagonistic effect of histamine has become an important way to study anti-allergic drugs.52-55 Currently, the anti-allergic drugs used in clinical practice are mainly H<sub>1</sub> receptor antagonists.<sup>56-60</sup> H<sub>1</sub> receptor antagonists were first reported in 1933 to relieve bronchospasm in animals induced by the inhalation of excessive histamine. Since the 1930s, scientists around the world have conducted a large number of structural modifications by using structure activity relationship (SAR) studies, and dozens of classic antiallergy drugs have been listed for clinical use. In the history of drug development, they have been listed as the first generation

H<sub>1</sub> receptor antagonists. The second generation H<sub>1</sub> receptor antagonists were marketed after the 1980s. These compounds have characteristics of high selectivity towards the H<sub>1</sub> receptor, no sedation effect, separation of the antihistamine effect from the central nervous system and fewer side effects. They were known as non-sedation H<sub>1</sub> receptor antagonists.<sup>61</sup> The third generation H<sub>1</sub> receptor antagonists are active optical isomers of the second generation H<sub>1</sub> receptor antagonists or their metabolites. This generation has greater safety and lower toxic and side effects; thus, their application scope has been expanding in recent years. 62-65 Currently, there are many varieties of H<sub>1</sub> receptor antagonists in clinical application, which can be divided into ethylenediamine antagonists, amino ether antagonists, propylamine antagonists, tricyclic antagonists, piperazine antagonists and piperidine antagonists depending on their chemical structures. 66-70 This article mainly reviews the research progress of allergic reactions with histamine H1 receptor antagonists (2011–2017), including the research on the design and synthesis of various new compounds.

#### 2. Synthesis

Loratadine is a second-generation antihistamine because it has fewer central nervous system and anticholinergic side effects. Loratadine has a strong therapeutic effect because of its semirigid conformation. Lewis et al. have reported a series of compounds based on loratadine that have both H1 receptor antagonist activity and 5-lipoxygenase inhibitory activity. Liu et al. successfully synthesized a series of loratadine derivatives with C-alkyl butenamide, and found that several of them were able to inhibit histamine induction. Yue W. et al.71 further

Scheme 1 General synthesis of loratadine analogues 4a-h. Reagents and conditions: (a) methyl acrylate, MeOH, 40 °C, 12 h; (b) MeONa/MeOH 30% aq., toluene, 110 °C, 5 h; (c) HCl, 100 °C, 5 h; (d) TiCl<sub>4</sub>/Zn, THF dry, 70 °C, 17 h.

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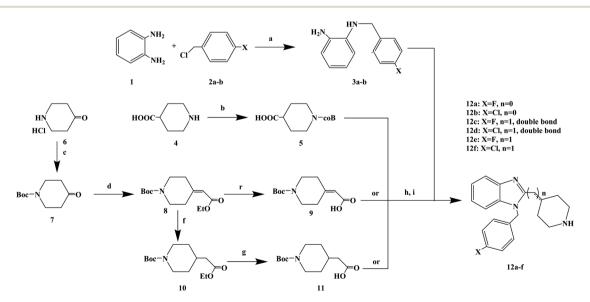
Fig. 2 The structural modification of benzimidazole derivatives 12a-f.

optimized the structure of loratadine based on previous studies, and designed and synthesized a series of loratadine derivatives (Fig. 1). In structural optimization, hydroxyl groups and chiral centers were introduced into the molecules to enhance their affinity for  $\rm H_1$  receptors. In the synthetic process (Scheme 1), a series of chiral hydroxyl groups containing N-substituted-4-piperidone were obtained by Michael addition. Subsequently, a Dieckmann cyclization and decarboxylation were performed with chiral or non-chiral aminoalcohols and methyl acrylate as raw materials, affording a yield of 35%. A variety of carbonyl groups were obtained by coupling piperidone with a tricyclic ketone using a low-priced titanium catalyst to obtain a series of target compounds at a yield of approximately 25%. The synthetic route is characterized by mild reaction conditions, good versatility and simple operation.

Astemizole (Fig. 2) is an effective histamine  $H_1$  receptor antagonist ( $K_i = 2 \text{ nmol } L^{-1}$ ). However, it could lead to arrhythmias with acquired long QT syndrome, which includes torsades DE pointes, and sudden death due to the blockage of rapidly activated delayed rectification  $K^+$  current  $I_{Kr}$  and its underlying human ether-a-gogo related gene (hERG)  $K^+$ 

channel ( $IC_{50} = 0.9 \text{ nmol L}^{-1}$ ). In order to find a new type of  $H_1$  receptor antagonist with low hERG blocking activity, Xiao J. *et al.*<sup>72</sup> carried out systematic structural modification based on astemizole (Fig. 2). The synthetic route of the benzimidazole derivative **12a–12f** was summarized in Scheme 2. In addition, compounds **9** and **11** were synthesized from compound **6** according to previous literature. The compounds **12a–12f** were produced by following acylation, cyclation, and deprotection procedures. The synthetic route was characterized by mild reaction conditions, high yield and good versatility.

Numerous uses of substituted triazines have been reported, such as anti-inflammatory, analgesic activity, purine confrontation, antitumoral and trypanocidal activity. Very little studies have been conducted on heterocyclic compounds such as 1, 2, 3-triazine and other triazines, this kind of compounds and similarity based on the structure of these compounds and other triazines like 1,2,4-triazine and 1,3,5-triazin. It was found that 1,2,3-triazine derivatives possess prostaglandin inhibition property, analgesic, anti-inflammatory and antihistaminic properties. Gollapalle L. *et al.*<sup>73</sup> further optimized 1,2,3-triazine



Scheme 2 General synthesis of benzimidazole derivatives 12a-f. Reagents and conditions: (a)  $K_2CO_3$ , DMF, 50 °C, 6 h; (b)  $Boc_2O$ , 1 mol  $L^{-1}$  NaOH, THF, 25 °C, 2 h; (c)  $Boc_2O$ , 2 mol  $L^{-1}$  NaHCO<sub>3</sub>, THF, 25 °C, 6 h; (d) ethyl diethoxyphosphoryl acetate,  $K_2CO_3$ , DMF, 78 °C, 12 h; (e) 1 mol  $L^{-1}$  NaOH, CH<sub>3</sub>OH, 25 °C, 30 min; (f) 10% Pd/C, H<sub>2</sub>,  $C_2H_5OH$ , 25 °C, 7 h; (g) 1 mol  $L^{-1}$  NaOH, CH<sub>3</sub>OH, 25 °C, 30 min; (h) ethyl chloroformate, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, 30 min; (i) AcOH, reflux, 2 h.

Scheme 3 General synthesis of 3H-benzo[4,5]thieno[2,3-d][1,2,3]triazin-4-ones 4a-c. Reagents and conditions: (a)  $CH_3COOH/CH_3COONH_4$ , glacial acetic/cyclohexane, reflux 10 h; (b)  $NaNO_2/HCl$ , glacial acetic, 0-5 °C.

on the basis of previous studies, and designed and synthesized three new compounds with antihistamine effects. In the process of synthesis (Scheme 3), the starting compounds 2-amino-3-(N-substituted carboxamido)-4 and 5-tetramethylene thiophenes (compounds 3a–c) were synthesized in three steps by adapting the Gewald reaction. Afterwards, compounds 3a–c were diazotized to yield a series of 3-substituted amino-5,6-tetramethylene thieno[2,3-d][1,2,3]-triazin-4(3H)-ones (compounds 4a–c). In this reaction, the starting compounds 2-amino-3-(N-substitutes carboxamido)-4 and 5-tetramethylene thiophenes (compounds 3a–c) react with NaNO<sub>2</sub> in the presence of HCl yielding the respective triazine-4-ones. The synthetic route was characterized by mild reaction conditions, simple operation and cheap and easy to obtain reagents.

1-(4-Fluorobenzyl)-2-(4-(4-methyl-1*H*-pyrazol-1-yl)piperidin-1-yl)-1*H*-benzo[*d*]imidazole (lead compound **1**, Fig. 3) is a high

affinity and selective H<sub>1</sub>-antihistamine. The lead compound 1 exhibits limitations of poor metabolic stability and solubility. Further optimization of compound 1 led to the development of benzyl)-2-(1-methylpiperidine-3-acyl)-1*H*-benzimidazole (lead compound 2, Fig. 3) with high affinity for H<sub>1</sub>antihistamine. The lead compound 2 showed improved selectivity and stability for hERG channel when compared with lead compound 1. Satheesh B. et al.74 performed relevant structure activity relationship (SAR) studies, determining that a change in the direction of the basic center in lead compound 2 was a key step in establishing the required H<sub>1</sub> affinity and selectivity. Based on this discovery, Satheesh B. et al. designed and synthesized a series of H<sub>1</sub>-antihistamines (Fig. 3) after structural optimization of the lead compounds 1 and 2. In this process, different substituents R1 were introduced into the imidazole ring. The polar group R2 was added to the piperidine

Fig. 3 The structural modification of 2-(piperidin-3-yl)-1H-benzimidazole analogues 5a-n. 6a-d and 7a-h.

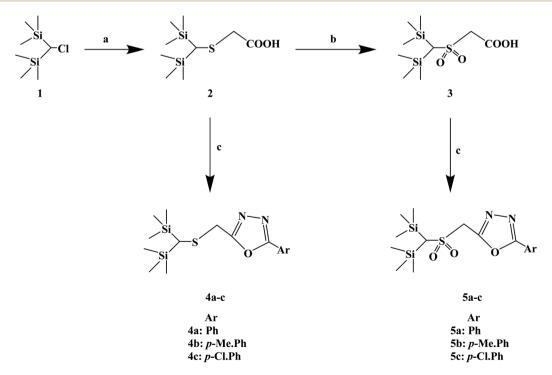
Scheme 4 General synthesis of 2-(piperidin-3-yl)-1H-benzimidazole analogues 5a-n, 6a-d and 7a-h. Reagents and conditions: (a) EtOH, 120 °C; (b) R<sup>1</sup>CH<sub>2</sub>Br, K<sub>2</sub>CO<sub>3</sub>, DMF, 80 °C; (c) TFA, CH<sub>2</sub>Cl<sub>2</sub>; (d) R<sup>2</sup>-( $\rightleftharpoons$ O)H, Na(OAc)<sub>3</sub>BH or R<sub>2</sub>X, Et<sub>3</sub>N, THF; (e) separation of enantiomers by chiral SFC when chiral acids were not used.

5a-n, 6a-d, 7a-h

ring nitrogen, hence,  $pK_a$  and/or  $\log P$  were reduced by the electron attraction effect. The synthesis of analogues described in Scheme 4 relies on the formation of the benzimidazole core (compound 3), after the coupling of a suitably protected amino acid (compound 1) with *O*-phenylene diamine (compound 2). Introduction of the appropriate arene-containing moiety  $R^1$  was accomplished by alkylation of compound 3. The final step included the alkylation or reductive amination and the removal

of Boc to obtain the final product (compounds **5a-n**, **6a-d** and **7a-h**). Further separation of enantiomers was achieved by chiral SFC when chiral acids were not used for the synthesis of benzimidazole. The synthetic route was characterized by mild reaction conditions, cheap and easy to obtain reagents, relatively high yield and good versatility.

Silicon substitution (C/Si exchange) has attracted extensive attention in drug-like scaffold materials due to its unique



Scheme 5 General synthesis of 2-((bis(trimethylsilyI) methylthio/methylsulfonyI) methyl)-5-aryl-1,3,4-oxadiazoles. Reagents and conditions: (a)  $HSCH_2COOH$ , MeOH, NaOH, reflux 4 h; (b)  $H_2O_2$ , AcOH, 24 h; (c)  $ArCONHNH_2$ ,  $POCl_3$ , reflux 6 h.

biological activity, chemical reactivity and physical and chemical properties compared with the parent carbon compounds. 1,3,4-Oxadiazole derivatives have been reported as promising antiallergic drugs. Dinneswara R. et al.75 designed and synthesized a variety of substituted 1,3,4-oxadiazole compounds (Scheme 5) in a study related to silicon substitution drugs, and applied them to the study of antiallergic drugs. In the structural design process, -Ph, 4-Me-Ph and 4-Cl-Ph substituents were selected. In the process of synthesis, 2-(bis(trimethylsilyl) methylthio) acetic acid (compound 2) was synthesized by the reaction of bis(trimethylsilyl)chloromethane (compound 1) and mercaptoacetic acid in the presence of NaOH/MeOH. Compound 3 was subsequently oxidized in the presence of H<sub>2</sub>O<sub>2</sub>/CH<sub>3</sub>COOH. The required heterocyclic compounds, 2-((bis(trimethylsilyl)methylthio)methyl)-5-aryl-1,3,4-oxadiazoles (compound 4) and 2-((bis(trimethylsilyl)-methylsulfonyl) methyl)-5-aryl-1,3,4-oxadiazoles (compound 5) were synthesized by the reaction of compound 2/compound 3 with benzohydrazides in the presence of phosphorus oxychloride. The synthetic route has high reaction yields (79% to 88%), mild reaction conditions and conventional reagents.

Over the years, Kujin has been widely used to treat allergic diseases and many others. Phytochemical studies have shown that it contains quinolizidine, alkaloids, flavonoids, and

triterpenoids. Kujin and its active components have many biological properties, such as anti-inflammatory, antiasthmatic, antitumor and antibacterial. Hiroyuki M. et al.76 found that the extract of Kujin has inhibited the up-regulation of H1R and IL-4 gene expression in rats sensitized by TDI, and confirmed (R)maackiain as the main anti-allergy component of Kujin. (R)-Maackiain is found in low concentration in plants, which made it difficult to extract; so, it was obtained by chemical synthesis (Scheme 6). In the process of synthesis, maackiain bis(benzonitrile) palladium(II) dichloride [PdCl<sub>2</sub>(PhCN)<sub>2</sub>] was prepared as reported previously. The synthesis of (R)-3-benzylmaackiain was performed as follows. To a solution of 7-benzyloxy-2H-1-benzopyran and 2-bromo-4,5-methylene dioxyphenyl in anhydrous DMF suspended in potassium acetate, PdCl<sub>2</sub>(PhCN)<sub>2</sub> was added in one portion. The resulting suspension was stirred at 40  $^{\circ}$ C. After 24 h, the addition of PdCl<sub>2</sub>(PhCN)<sub>2</sub> to the resulting reaction mixture was carried out. The latest procedure was repeated after 48 h and 72 h. Finally, the resulting mixture was stirred for additional 24 h. This synthetic route achieved the first chemical synthesis of (R)-maackiain, but it had disadvantages, such as long synthetic route, low yield, the use of precious metal Pd in the synthesis process, and long overall reaction time, which need to be further optimized.

Scheme 6 General synthesis of ( $\pm$ )-maackiain. Reagents and conditions: (a) acrylonitrile, triton B, NaOH, 100 °C, 10 h; (b) H<sub>2</sub>SO<sub>4</sub>, CH<sub>3</sub>COOH, H<sub>2</sub>O, 100 °C, 8 h; (c) benzyl chloride, NaI, K<sub>2</sub>CO<sub>3</sub>, acetone, 60 °C, 18 h; (d) NaBH<sub>4</sub>, ethanol, rt, 3 h; (e) *p*-toluenesulfonic acid, toluene, 110 °C, 2 h; (f) PCl<sub>2</sub>(PhCN)<sub>2</sub>, KoAc, DMF, 40 °C, 4 days; (f) H<sub>2</sub>-Pd/C, 3 h.

compound 3 (3-(3-chlorophenyl)-2-thioxo-quinazolin-4(3*H*)-one) was belonged to the key intermediate. Dithiocarbamic

acid methyl ester was prepared with 3-chloro aniline, carbon

disulfide and sodium hydroxide in dimethyl sulfoxide, then

methylated with dimethyl sulfate, which was reacted to

produce the required compound 4 in ethanol has a good yield (86%). Compound 3 was dissolved in a solution of 2%

ethanol sodium hydroxide, stirred at room temperature, and

methylated with dimethyl sulfate to give 3-(3-chlorophenyl)-

2-(methylthio) quinazolin-4(3H)-one (compound 4). The

nucleophilic displacement of the methylthio group of

compound 4 with hydrazine hydrate was carried out using Quinazolinones have a wide range of biological and ethanol as the solvent, affording compound 5 (3-(3-chlorpharmacological effects, such as analgesic. inflammatory, antifungal, antiviral and antihistamine activophenyl)2-hydrazinyl-quinazolin-4(3H)-one). Finally, the ities. Veerachamy A. et al. proved that quinazolinone derivcompound 5 with a variety of one carbon donors such as atives have strong antihistamine activity with alow sedative acetic acid, propionic acid, formic acid, butyric acid and effect. Manavalan G. et al. 77 designed and synthesized a series chloroacetyl chloride under refluxing provided 1-substitutedof 1-substituted-4-(3-chlorophenyl)-[1,2,4]triazolo[4,3-a]qui-4-(3-chlorophenyl)-[1,2,4]triazolo[4,3-a]quinazolin-5(4H)nazolin-5(4H)-ones (Scheme 7) based on the structural ones (compounds 6a-e). The synthetic route has high reacmodifications of quinazolone. In that process, the triazolo tion yield (80% to 85%), mild reaction conditions and uses group has been substituted with different groups (-H, -CH<sub>3</sub>, conventional reagents. -CH<sub>2</sub>CH<sub>3</sub>, -(CH<sub>2</sub>)<sub>2</sub>CH<sub>3</sub>, -CH<sub>2</sub>Cl) to obtain compounds with The compounds containing the phthalazinone scaffold were improved biological activity. In the process of synthesis,

obtained through structural optimization and subsequently used as histamine H1 receptor antagonists for the treatment of allergic reactions. These compounds bind to both H<sub>1</sub> and H<sub>3</sub> receptors. Based on the structure optimization of phthalazinone, Panayiotis A. et al.78 introduced different substituents, and designed and synthesized a series of novel phthalazinone amide compounds (Scheme 8). In the molecular structure design, -Et, n-Pr, iso-Pr,  $c-C_5H_9$ , -CH<sub>2</sub>OCH<sub>3</sub>, -CH<sub>2</sub>CH<sub>2</sub>OCH<sub>3</sub>, -CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>OCH<sub>3</sub>, 4-tetrahydropyranyl were selected as the substituent. During the synthetic procedure, compound 1 was alkylated with N-(2-bromoethyl) phthalimide to give compound 2 (yield 87%), and then, deprotected using

CI

A

CI

NH

SCH3

B

CI

NH

SCH3

CI

O

O

NH

SCH3

A

CI

O

NH

SCH3

A

CI

O

NH

SCH3

A

CI

O

NH

SCH3

A

Scheme 7 General synthesis of 1-substituted-4-(3-chlorophenyl)-[1,2,4]triazolo[4,3-a] quinazolin-5(4H)-ones 6a-e. Reagents and conditions: (a) CS<sub>2</sub>, NaOH, stirring 30 min, (CH<sub>3</sub>)<sub>2</sub>SO<sub>4</sub>; (b) methyl anthranilate/EtOH, reflux 19 h; (c) NaOH/EtOH, string 2 h; (d) NH<sub>2</sub>NH<sub>2</sub>, H<sub>2</sub>O/EtOH, reflux 24 h; (e) reflux 29 h.

6а-е

Scheme 8 General synthesis of phthalazinone amide 3a-g, 7a-g and 10a-h. Reagents and conditions: (a) N-(2-bromoethyl)phthalimide,  $K_2CO_3$ , 2-butanone, 80 °C, 18 h; (b)  $NH_2NH_2 \cdot H_2O$ , EtOH, 80 °C, 1.25 h; (c)  $RCO_2H$ , TBTU,  $Et_3N$ , DMF, 2 h, or RCOCI,  $Et_3N$ , DCM; (d) 2,2,2-10,  $Et_3N$ ,  $Et_3N$ , trifluoro-N-(2-iodoethyl)acetamide, DIPEA, 2-butanone; (e) (i) MeI, K2CO3, DMF; (ii) K2CO3, H2O, MeOH; (f) BrCH2CH2CO2Et, K2CO3, DMF, 150 °C, microwave, 110 min; (g) NaOH, H<sub>2</sub>O, MeOH, 20 °C, 1 h; (h) TBTU, Et<sub>3</sub>N, DMF.

7g 

4a-h

10b, 10f

HO OH OH

Fig. 4 The structural modification of fexofenadine derivatives 3a-e.

Fexofenadine

hydrazine to afford compound 3 with 77% yield. Afterwards, compound 3 was acylated with a variety of carboxylic acids in the presence of TBTU to give compounds 4d, 4f, 4g and 4h, or with the corresponding acid chloride to give compounds 4a, 4b, 4c and 4e. Alkylation of compound 1 with 2,2,2-trifluoro-*N*-(2-iodoethyl)acetamide resulted in trifluoroacetamide (compound 5), which was alkylated with iodomethane and deprotected, to obtain the secondary amine (compound 6). The latter was acylated in the presence of 4-methoxybutanoic acid and TBTU to obtain tertiary amide (compound 7g). Compound 1 reacted with ethyl 4-bromobutyrate to give compound 8, which was converted to a carboxylic acid 9, and finally, transformed into amides 10b and 10f. The synthetic route has good versatility and uses conventional reagents, but the partial yield was low (yield 29%), which should be further improved.

 $H_1$  antagonists are competitive drugs that inhibit histamine action on tissues containing  $H_1$ . Fexofenadine (2-[4-[1-hydroxy-4-[4-(hydroxydiphenylmethyl)piperidino]butyl]phenyl]-2-methylpropanoic acid) (Fig. 4) is a selective and peripherally acting  $H_1$  receptor antagonist and is a nonsedative active

metabolite of terfenadine. Fexofenadine was clinically used to treat seasonal allergic rhinitis, but there were associated adverse effects such as headache, throat irritation, viral infection, nausea, dysmenorrhea, drowsiness, dyspepsia and fatigue. Saeed A et al. 79 used fexofenadine as the lead compound (Fig. 4), and designed and synthesized fexofenadine derivatives with hydrazine hydrochloride, urea, p-phenelediamine, acetamide and semicarbazide (Scheme 9). Herein, carboxylbenadine 2methylpropionamide and benadrylamide derivatives were synthesized using carboxylbenadine as the nucleophilic substituent and tertiary C-19 was replaced by a benzene ring. The carboxyl group on the tertiary C-19 was esterified by a condensation reaction, and the nucleophilic substitution reaction on the acyl group formed a tetrahedral intermediate, which transformed back to the carbonyl group after the leaving group was eliminated. On the basis of the above reaction mechanism, various amines were reacted to produce amide compounds, and various reactions were identified. This synthetic route was characterized by a high yield, short reaction time and good versatility.

Target compounds

Scheme 9 General synthesis of fexofenadine derivatives 3a-e. Reagents and conditions: (a) MeOH,  $H_2SO_4/H_2O$ , reflux 7–8 h; (b) amine, reflux 2–3 h.

#### 3. Conclusions

Histamine plays an important role in the pathophysiological mechanism of allergic diseases, and the antagonistic effect of histamine has become an important way to study anti-allergic drugs. The anti-allergic drugs used in clinical practice were mainly H<sub>1</sub> receptor antagonists. This article mainly reviews the research progress of allergic reactions with histamine H<sub>1</sub> receptor antagonists and expounds the important aspects of design, synthesis and biological activity of various new compounds. These novel H<sub>1</sub> receptor antagonists, with regards to their chemical structure design, are mostly obtained by optimizing the structure of the reported compounds with antiallergic activity (including natural products, Schemes 6 and 9) or already sold drugs. These compounds have novel structures, and the design of H1 receptor antagonists in the future will tend to optimize the structure of natural products as lead compounds. In the process of synthesis, most synthetic routes use conventional solvent reaction to synthesize target compounds. These synthetic routes were characterized by simple operation, mild reaction conditions, cheap reagents and a good overall yield, which will provide certain foundation for future industrialization. However, some synthetic routes use multiple catalysts or precious metals (such as Pd, Scheme 6), or the reaction time was too long (Scheme 7). In the selection of synthesis methods, researchers should focus on the modern synthesis methods, such as microwave synthesis, and the organic synthesis process will have more room for development. As can be seen from the reagent and reaction condition perspective, the choice direction in the future will be to use cheap and readily available raw materials, conventional solvents, no catalyst or conventional catalyst, and medium or low reaction temperature.

#### Conflicts of interest

The authors declare that they have no conflict of interest.

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