



Cite this: *RSC Adv.*, 2020, **10**, 12060

Effects of substrates on proton irradiation damage of graphene

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In this paper, the irradiation damage of graphite sheets and monolayer graphene on Cu and Ni substrates after the proton irradiation with High Intensity D-T Fusion Neutron Generator (HINEG) were studied. The microstructure evolution of graphite sheets and monolayer graphene on different substrates was analyzed using Raman spectroscopy, X-ray photoelectron spectroscopy (XPS), and transmission electron microscopy (TEM). Results showed that the peak area ratio of carbon-oxides (C=O, C \equiv O) in graphene was reduced after irradiation. 2–50 nm nanopores were produced in monolayer graphene on both Cu and Ni substrates. The results of Raman spectroscopy showed that the relationship between D and G peak intensity ratios (I_D/I_G) of irradiated graphene on diverse substrates was $I_D/I_{G(Cu)} < I_D/I_{G(Ni)} < I_D/I_{G(graphite)}$, which indicated that the proton irradiation damage of graphene on the Cu substrate was the lightest. The reason for this could be speculated to be that different substrates had different damage self-repairing capabilities.

Received 29th October 2019

Accepted 11th February 2020

DOI: 10.1039/c9ra08905e

rsc.li/rsc-advances

1. Introduction

Graphene, two-dimensional (2D) carbon sheets arranged in a honeycomb lattice, is an outstanding material with various applications, due to its big theoretical specific surface area (2630 m 2 g $^{-1}$), high intrinsic mobility (200 000 cm 2 V $^{-1}$ s $^{-1}$), Young's modulus (\sim 1.0 TPa), thermal conductivity (\sim 5000 W m $^{-1}$ K $^{-1}$) and optical transmittance (\sim 97.7%).^{1–3} Nanoporous graphene (NPG), a kind of defective graphene with nano-scale pores on a two-dimensional base surface, is formed by removing the carbon atoms from the crystal lattice or transferring ones to the surface leaving a vacancy.⁴ The size of NPG can be grouped into micropores (<2 nm), mesopores (2–50 nm) and macropores (>50 nm).⁵ Therefore, compared to the inert perfect graphene surface, NPG not only retains the excellent properties, but its material transport efficiency also gets improved, especially for the separation of isotopes with little physicochemical difference. Lozada-Hidalgo *et al.*⁶ used the two-dimensional membranes of monolayer graphene to separate H $^+$ from D $^+$ successfully, with a separation factor up to 10 at room temperature. In addition, NPG has a pronounced application prospect in single-molecule sequencing, gas separation, seawater desalination, energy storage, and so on.

The research of NPG production mainly focuses on how to obtain high quality and high throughput array holes. Lots of techniques, like the mask-based lithography, chemical vapor deposition and particle beam irradiation, have been proposed in the recent years to produce NPG and control its structure. However, particle beam irradiation has higher energy transfer efficiency than mask-based lithography and chemical vapor deposition.⁷ The irradiation damages of graphene with particle beams had been widely carried out. Drndić *et al.*⁸ first fabricated nanometer-scale pores (\sim 3.5 nm) in multilayer graphene suspended on SiN $_x$ film under irradiation by transmission electron microscope (TEM) with the focused electron beam (200 kV). They got NPG with stable structures but have had no improvement so far. Similarly, scientists also studied the irradiation damage of graphene by other particle beams. For example, irradiation with Ar $^+$, Ga $^+$ and He $^{2+}$ ion beams produced nanopores with diameters of 80 nm, 10 nm and 1.3 nm in graphene, respectively.^{9–11}

In addition, some research studies on the irradiation damage of graphene by proton beams have also been reported. Ko *et al.*¹² irradiated few-layer graphene on a SiO $_2$ /Si substrate by 5 MeV proton beams and found that high-energy proton irradiation could cause point defect damage in graphene. Mathew *et al.*¹³ studied the impact of the substrate and layer number on the stability of graphene by using 2 MeV proton beams to irradiate the graphene set in SiO $_2$ /Si substrate. The damage threshold of graphene samples was discovered to present a layer number-dependent effect and behave the same when using a substrate to support the graphene layer. Yang *et al.*¹⁴ researched the effect of doses on proton-irradiation damage of

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graphene by using various proton doses extending from $1 \times 10^{16} \text{ cm}^{-2}$ to $3 \times 10^{16} \text{ cm}^{-2}$ to irradiate monolayer graphene set in SiO_2/Si substrate. The result showed that as the irradiation doses of proton beams increased, the monolayer graphene damages increased gradually. Lee¹⁵ and Zhang *et al.*¹⁶ both studied the effect of proton energy on damage formation in graphene by using protons with a different energy to irradiate monolayer graphene set in SiO_2/Si substrate. They detected that the irradiation damage decreased as the proton energy increased. Combining experiments and simulations, they thought it was due to the low energy transfer of higher energy protons to graphene. Most of the current reports demonstrated the effects of graphene layer number, proton beam dose and energy on the irradiation damage of graphene on SiO_2/Si substrates, while few have focused on the effect of substrates on proton irradiation damage of graphene. However, the researches of Compagnini *et al.*¹⁷ showed that when the graphene defect density was increased to exceed a certain threshold, the graphene ripples were suppressed intensely and the defective crystal adapted its shape to the substrate so as to maintain the thermodynamic stability. They also indicated that the reason of why monolayer graphene has a higher disorder than bilayer and multi-layer graphene is that they interact with the substrate differently. Therefore, it is quite necessary to explore the effects of substrates on proton irradiation damage of graphene.

In this paper, the effects of substrates on proton irradiation damage of graphene were investigated by proton beam irradiation. This work would play a crucial part in guiding the production of NPG through particle beam irradiation to graphene.

2. Experimental details

The Cu foil ($100 \times 100 \text{ mm}$), Cu and Ni substrates ($\Phi = 25 \text{ mm}$, $d = 1 \text{ mm}$, purity: 99.99%) were all commercially obtained from Hefei Vigon Material Technology Co., Ltd (Hefei, China). The graphite sheets ($\Phi = 25 \text{ mm}$, $d = 1 \text{ mm}$) were purchased from Dongteng Company. Poly (methyl methacrylate) (PMMA), chlorobenzene and acetone for the transfer of graphene were all purchased from Alfa Asha.

The graphene was transferred from Cu foil to Cu and Ni substrates with the wet-transfer method as shown in Fig. 1. Firstly, graphene was grown on Cu foil through the technique of chemical vapor deposition (CVD). The graphene film grown on

Cu foil has been analyzed with Raman spectra in Fig. 2. The results show that the intensity ratio of I_G/I_{2D} is lower than 1, and the 2D peak has a perfect single Lorentz peak type. It indicates that the prepared graphene is monolayer.^{13,16} Secondly, the graphene on Cu foil was spin-coated with a 6 wt% PMMA solution at 3600 rad s^{-1} and formed with a PMMA/graphene/Cu foil structure. Subsequently, we floated the PMMA/graphene/Cu foil structure on a Cu etchant (volume ratio of copper sulfate : hydrochloric acid : water = 1 : 5 : 5) for 30 min to form a PMMA/graphene structure. Then the PMMA/graphene structure was shifted onto a Cu or Ni substrate and the PMMA layer was removed by using acetone. Finally, the monolayer graphene/Cu and graphene/Ni substrate was prepared.

As-prepared samples were irradiated with proton beam generated by a High Intensity D-T Fusion Neutron Generator (HINEG)^{18,19} under vacuum conditions. The parameters are shown in Table 1.

Raman spectroscopy is an important method for quickly and non-destructively characterizing the crystal structure of graphene.^{20,21} The samples were measured by Raman spectra to investigate the disorder of the graphene. The samples were measured at room temperature with a Micro Raman spectrometer (Renishaw *in Via* Reflex, UK) of which the wavelength is 532 nm, the incident power is 5 mW (to avoid laser-induced heating and sample damage), and the spot size is about $2 \mu\text{m}$. The C element's chemical states in samples were analyzed by Escalab 250Xi X-ray photoelectron spectroscopy (XPS) using the excitation source of Al-K α . The binding energy of C element was determined at the pass energy of 100 eV, with an energy step of 1 eV. The Tecnai G2 F20 200 keV transmission electron microscope (TEM) has also been used as an indispensable instrument in the study of the structures of samples.

3. Results and discussions

3.1 Raman spectroscopy analysis

Due to the characteristic response to any variation in carbon hybridization state and defects or foreign species introduction, the Raman spectroscopy had been highly favored as the most important spectroscopic technique in studying the carbon-based materials all the time. Fig. 3 shows the development of Raman spectra of different samples non-irradiated and irradiated with the irradiation parameters of Group-A in Table 1.

Two typical characters of the monolayer graphene before irradiation are the G peak at 1580 cm^{-1} (with E_{2g} symmetry, in-

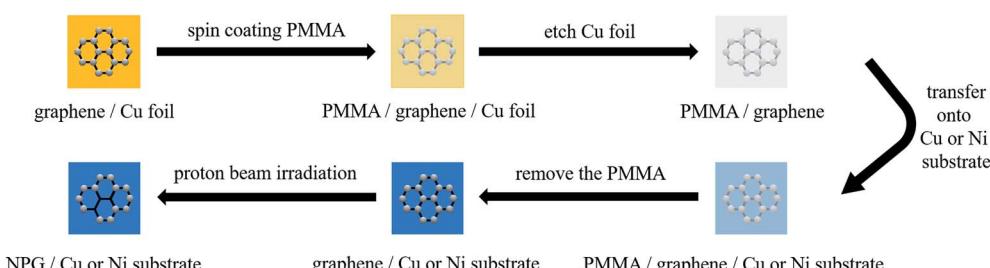


Fig. 1 The specific process of transferring the graphene on Cu foil to the Cu and Ni substrates by wet-transfer method.



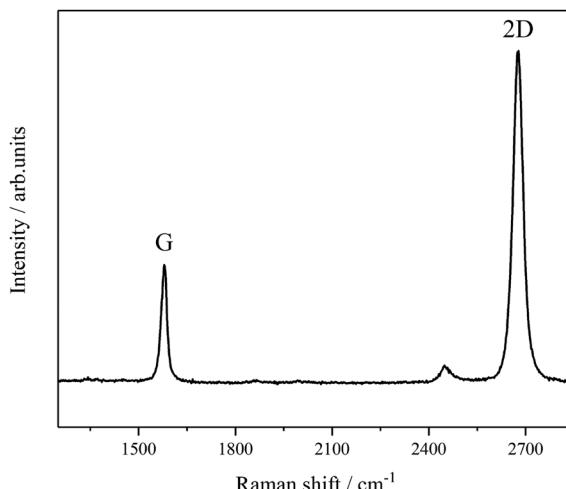


Fig. 2 Raman spectra of the monolayer graphene film on Cu foil by CVD.

plane vibration of sp^2 carbon atoms) and the 2D peak at 2690 cm^{-1} (two-phonon double-resonance Raman scattering process).^{20,21} The characteristics of pristine monolayer graphene used in the experiment are confirmed by three features: (1) the full-width-at-half-maximum (FWHM) of the 2D peak of graphene on Cu and Ni substrates are 34.8 cm^{-1} and 36.2 cm^{-1} , respectively. (2) The intensity ratio of G peak and 2D peak of the graphene on Cu and Ni substrates are lower than 1. (3) The 2D peaks of the graphene on Cu and Ni substrates both have a perfect single Lorentz peak type.^{13,16,22,23} Also, the D peaks at $\sim 1350\text{ cm}^{-1}$ reflect that the presence of Raman active defects is weak enough to be ignored. There are also three characteristic peaks of D, G and 2D around 1350 cm^{-1} , 1580 cm^{-1} , and 2710 cm^{-1} in the graphite sheets. The graphite sheets could be

considered as the multilayer graphene on graphite substrate due to a FWHM of $\sim 80.1\text{ cm}^{-1}$, $I_G/I_{2D} > 1$, and a 2D peak of a plurality of single Lorentz peak type. Meanwhile, the D peak of pristine graphite sheets is also very weak.

After the proton beam irradiation, the Raman spectra of different samples are all greatly changed. The 2D peaks are significantly broadened and their intensities are reduced so that they almost disappeared. The G peaks also appear to be significantly broadened, and their intensities are reduced to varying degrees. More importantly, for the disordered D peaks, their intensities are enhanced and their FWHM are also significantly increased. Therefore, a Raman spectrum similar to graphene oxide is formed because the D and the G peaks significantly overlap.²⁴

It has been demonstrated that measuring the D peak intensity helps quantitatively analyze the defect density of graphene.²⁵ The ratio of the intensity of the D-G peaks (I_D/I_G) is often used as an important parameter to characterize the defect density of graphene.²¹ Studies showed that the following relationship exists between I_D/I_G , the average distance between defects (L_D), laser wavelength (λ_L , 532 nm) and defect density (n_D).²⁶

$$L_D^2(\text{nm}^2) = (1.8 \pm 0.5) \times 10^{-9} \times \lambda_L^4 \left(\frac{I_D}{I_G} \right)^{-1} \quad (1)$$

$$n_D(\text{cm}^{-2}) = \frac{(1.8 \pm 0.5) \times 10^{22}}{\lambda_L^4} \left(\frac{I_D}{I_G} \right) \quad (2)$$

The calculation results are shown in Table 2. It could be seen that under the same irradiation condition, from the monolayer graphene on Cu substrate, the monolayer graphene on Ni substrate to the graphite sheets, the intensity ratio of D-G peaks

Table 1 The experimental parameters of proton beam irradiation

Group	Ions energy (keV)	Beam intensity (mA)	Irradiation time (min)	Irradiation fluence (p cm^{-2})
A	100	1	10	5.58×10^{16}
B	250	1	10	5.58×10^{16}
C	250	0.35	30	5.86×10^{16}

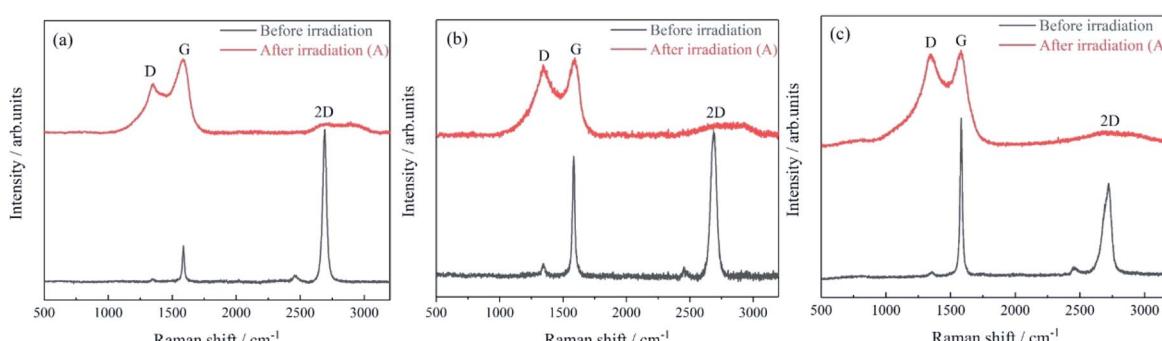


Fig. 3 Raman spectra of different samples before and after irradiation. (a) Monolayer graphene on Cu substrate; (b) monolayer graphene on Ni substrate; (c) graphite sheets.



Table 2 The average distance between defects and the defect density of samples after irradiation of Group-A

Substrates	I_D/I_G	Average distance between defects L_D (nm)	Defect density n_D (cm^{-2})
Cu	0.641	14.85 ± 2.10	$(1.44 \pm 0.40) \times 10^{11}$
Ni	0.914	12.44 ± 1.76	$(2.05 \pm 0.57) \times 10^{11}$
Graphite	0.967	12.09 ± 1.71	$(2.17 \pm 0.60) \times 10^{11}$

(I_D/I_G) is sequentially increased, the average distance between defects (L_D) is sequentially decreased, and the defect density (n_D) is sequentially increased.

After irradiating graphene with proton beams of the same irradiation condition, what could be known from Table 2 is: $I_D/I_G(\text{Cu}) < I_D/I_G(\text{Ni}) < I_D/I_G(\text{graphite})$, $L_D(\text{Cu}) > L_D(\text{Ni}) > L_D(\text{graphite})$, $n_D(\text{Cu}) < n_D(\text{Ni}) < n_D(\text{graphite})$. It could be found that after the proton beam irradiation of graphene on different substrates under the same irradiation condition, graphene on different substrates have different irradiation damages: the monolayer graphene on the Cu substrate produces the least damage, the monolayer graphene on the Ni substrate produces the second damage, and the graphene on the graphite substrate (graphite sheets) produces the highest damage.

3.2 XPS analysis

As described in Section 3.1, the D peak and G peak of the graphene form a Raman spectrum similar to graphene oxidation.

The C element's chemical states in samples unirradiated and irradiated are both dissected by XPS so as to check the oxidative degree of graphene. Fig. 4 shows the development of XPS spectra of the C 1s of different samples non-irradiated and irradiated by the parameters of Group-A in Table 1.

The C 1s core peaks are divided into three symmetric Gaussian curves: a predominant sp^2 C-C (binding energy $E_B = 284.6$ eV), C-O ($E_B = 286.4$ eV), C=O ($E_B = 288.5$ eV).²⁷ As shown in Fig. 4(a)–(c), the peak area ratios of carbon-oxide-related (C-O and C=O) are reduced from 25.16%, 33.78% and 29.81% to 19.67%, 32.34% and 28.42%, respectively. After proton beam irradiation, the peak area ratios of C-O and C=O in graphene on Cu, Ni, and graphite substrates are reduced to different degrees, which are reduced by 5.49%, 1.44%, and 1.39%, respectively. During the irradiation process, the samples are all in a vacuum state and the peak area ratio of C-O and C=O is low. There are two possible reasons for the reduction: (1) when graphene is bombarded by the proton beam, protons are combined with oxygen atoms to reduce carbon oxides. (2) The irradiation process raises the temperature of the graphene samples, boosting the thermal reduction of the carbon oxide. Therefore, the Raman spectrums are not formed by graphene oxide but by the damages of graphene after proton beam irradiation.

3.3 TEM analysis

According to the analysis in Section 3.1 and 3.2, after proton beam irradiation, damages are generated in graphene. The

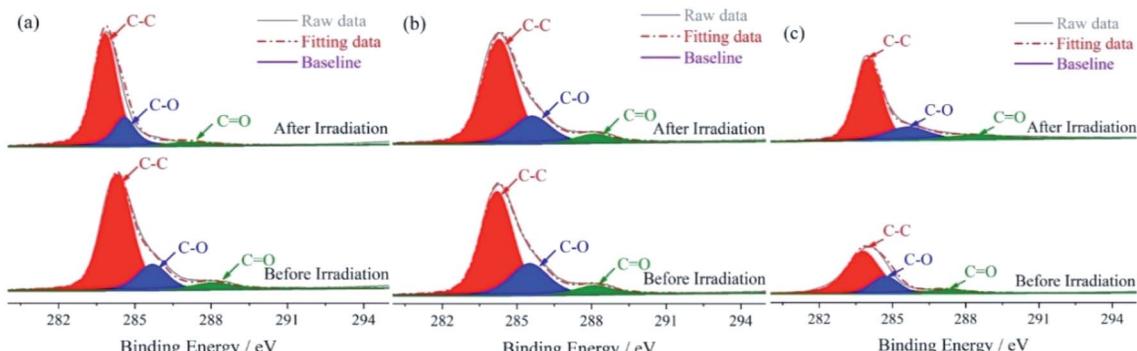


Fig. 4 XPS spectra of different samples' C 1s before and after irradiation (a) monolayer graphene on Cu substrate; (b) monolayer graphene on Ni substrate; (c) graphite sheets.

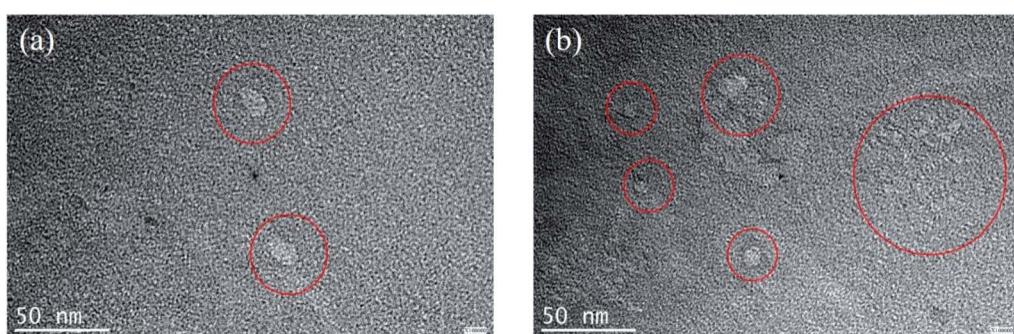


Fig. 5 TEM images of different samples after irradiation (a) monolayer graphene on Cu substrate; (b) monolayer graphene on Ni substrate.



microscopic morphology of damages in graphene could be observed by TEM. After proton beam irradiation with the parameters of Group-A in Table 1, the monolayer graphene on Cu and Ni substrates were transferred to 200 mesh TEM copper grids by the wet-transfer method. Fig. 5 displays the microstructure of graphene after irradiation.

It could be seen from the TEM images that the nanopores (circled in red) are generated in the monolayer graphene on the Cu and the Ni substrates. In Fig. 5(a), a small number of nanopores (pore density: $1\text{--}3$ nanopores/ $2.5 \times 10^3 \text{ nm}^2$) with similar pore sizes are observed in the monolayer graphene on Cu substrate. In Fig. 5(b), a large number of nanopores with higher pore density (pore density: $3\text{--}6$ nanopores/ $2.5 \times 10^3 \text{ nm}^2$) with different pore sizes are observed in the monolayer graphene on Ni substrate. Pore density of the graphene on the Ni substrate is higher than that of the graphene on the Cu substrate, which is consistent with the difference of defect densities ($n_{D(\text{Ni})} > n_{D(\text{Cu})}$) in Table 2. Although the pore sizes of the nanopores produced in graphene on different substrates are different, their pore sizes are in the same range of 2–50 nm, which belong to mesopores.⁵

3.4 Reliability of experimental results

In order to verify the experimental data and the consistency of results, experiments under two other different irradiation conditions of Groups B and C of Table 1 were carried out. The samples before and after irradiation were also analyzed by Raman spectroscopy in Fig. 6.

After the proton beam irradiation with parameters of Group-B and Group-C in Table 1, the obvious Raman spectra changes

of the graphene on different substrates are the same as that of Fig. 3. The 2D peaks of graphene on different substrates have been significantly broadened and their intensities are reduced. The G peaks also appear to be significantly broadened and their intensities are reduced to varying degrees. The intensities of the disordered D peaks are enhanced and their FWHM are significantly increased. The L_D and n_D were both calculated by formulas (1) and (2) and the calculation results are shown in Table 3.

Comparing results of Group-A in Table 2 with that of Group-B in Table 3, it could be found that for graphene on the same substrate (Cu or Ni or graphite), at the same beam intensity (1 mA), irradiation time (10 min) and irradiation fluence ($5.58 \times 10^{16} \text{ p cm}^{-2}$), the greater the ions energy of the proton beam, the smaller the damages produced in graphene. This conclusion is consistent with the study of Zhang *et al.*¹⁶ Comparing results of Group-B with Group-C in Table 3, it could be found that for graphene on the same substrate (Cu or Ni or graphite), at the same ionic energy (250 keV), the damages of graphene are increased as the irradiation fluence increased and this conclusion is consistent with the study of Zeng *et al.*²⁸ The above comparison of Group-A with Group-B and Group-B with Group-C shows that the data measured by Raman spectroscopy in three irradiation conditions is reliable. In Table 3, under different irradiation conditions of Group-B and Group-C, changes are the same for graphene on different substrates: $I_D/I_{G(\text{Cu})} < I_D/I_{G(\text{Ni})} < I_D/I_{G(\text{graphite})}$, $L_{D(\text{Cu})} > L_{D(\text{Ni})} > L_{D(\text{graphite})}$, $n_{D(\text{Cu})} < n_{D(\text{Ni})} < n_{D(\text{graphite})}$. The results show that different irradiation damages are produced in graphene on different substrates after proton beam irradiation under the same irradiation condition. The

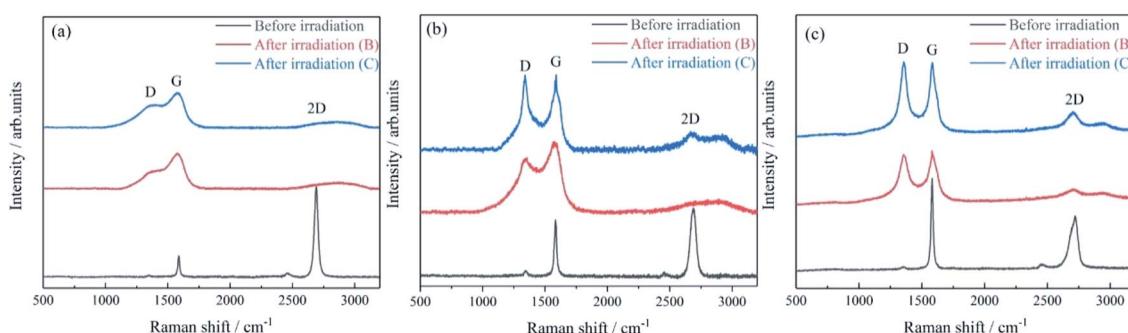


Fig. 6 Raman spectra of different samples before and after irradiation (a) monolayer graphene on Cu substrate; (b) monolayer graphene on Ni substrate; (c) graphite sheets.

Table 3 The average distance between defects and the defect density of samples after irradiation

Group	Substrate	I_D/I_G	Average distance between defects L_D (nm)	Defect density n_D (cm^{-2})
B	Cu	0.456	17.61 ± 2.49	$(1.02 \pm 0.29) \times 10^{11}$
	Ni	0.755	13.68 ± 1.94	$(1.70 \pm 0.47) \times 10^{11}$
	C	0.942	12.25 ± 1.74	$(2.12 \pm 0.59) \times 10^{11}$
C	Cu	0.679	14.43 ± 2.04	$(1.53 \pm 0.42) \times 10^{11}$
	Ni	0.977	12.03 ± 1.70	$(2.20 \pm 0.61) \times 10^{11}$
	C	1.008	11.84 ± 1.68	$(2.27 \pm 0.63) \times 10^{11}$



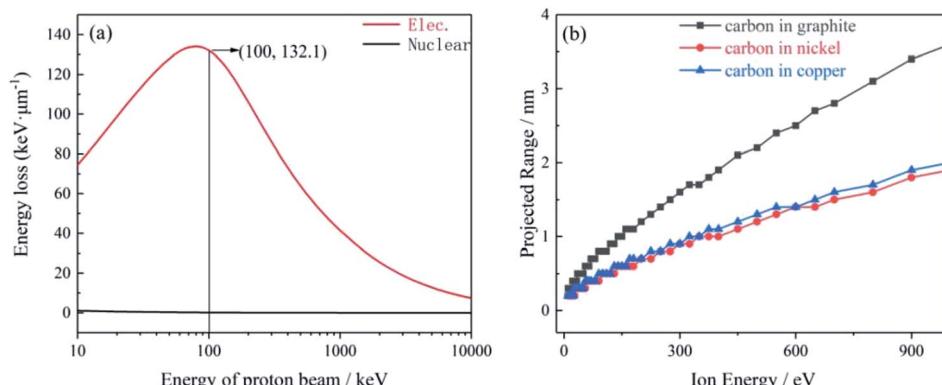


Fig. 7 Curve of SRIM simulation calculation. (a) Curve of energy loss proton in carbon material; (b) curve of projected range carbon ions in Cu, Ni and graphite.

order of irradiation damage is: Cu < Ni < graphite. This conclusion is completely consistent with the conclusions obtained in Section 3.1.

3.5 Discussion

The reason of why graphene on different substrates has different damages after irradiation under the same proton irradiation condition may be that different substrates have different effects on self-repairing property of damages in graphene.

The energy loss of protons in irradiated graphene was simulated by the particle transport software SRIM.²⁹ The range of incident proton energy was designed varying from 10 keV to 10 MeV, and the carbon density was 1.8 g cm⁻³. The simulation results are shown in Fig. 7(a). The theoretical thickness of monolayer graphene is only 0.335 nm. It could be seen from the calculation that the energy loss of proton with 100 keV is about 44.25 eV and this energy is mainly transferred to the kinetic energy of the detached carbon ions. At the same time, the penetration depth of carbon ions in Cu, Ni and graphite were imitated by SRIM and the consequences are shown in Fig. 7(b). With the same incident energy of carbon ions, the immersion depth of carbon ions in graphite is the largest, while the depth in Cu and Ni are almost the same but smaller than that in graphite.

The microscopic process of irradiating the graphene on a substrate by proton beam is shown in Fig. 8. The carbon atoms in graphene are directly bombarded by proton beam to form detached carbon ions with a certain kinetic energy. Then the detached carbon ions are infiltrated into the substrate. The detached carbon ions in the substrate would be precipitated to the surface of the substrate due to their low solubility. Finally, the precipitated carbon ions are combined with the C atoms in the vicinity of the nanopores to form a C-C bond to repair the nanopores in graphene produced by the proton beam irradiation.

Under the same irradiation condition, the detached carbon ions produced by proton beam bombardment of graphene on different substrates have the same amount. Then, the detached

carbon ions with a certain kinetic energy are permeated into Cu, Ni and graphite substrates at the same amount. The solubility of the detached carbon ions in Cu is much smaller than that in Ni and graphite.^{30,31} And the detached carbon ions penetrate almost the same depth in Cu and Ni, which is smaller than the penetration depth in graphite. The capability of the detached carbon ions to precipitate from the Cu substrate to its surface is the strongest due to the minimal carbon solubility and small penetration depth of Cu substrate. The capability of the detached carbon ions to precipitate from the graphite substrate to its surface is the weakest due to the maximal carbon solubility and the largest penetration depth of graphite substrate. The greater the capability of detached carbon ions to precipitate from the substrate to the surface, the more carbon ions are precipitated; the better the self-repairing property of damage in graphene, the smaller the damages are produced in graphene after proton beam irradiation.

The reason of why irradiation damages of graphene on different substrates are different after proton beam irradiation under the same irradiation conditions is that different

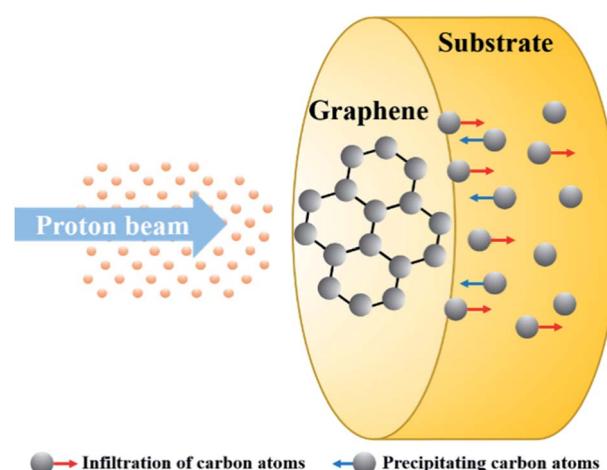


Fig. 8 Microscopic process figure of graphene on a substrate by proton beam irradiation.



substrates have different capabilities for the damage self-repairing in graphene. The greater the capabilities of different substrates, the smaller the damages of graphene are produced after irradiation. The capability from large to small is: Cu > Ni > graphite. Therefore, the damage of graphene on different substrates after irradiation from small to large is: Cu < Ni < graphite.

4. Conclusions

Proton beam irradiation has been used to introduce nanopores into graphene in the way of controllability. The joint use of Raman spectroscopy, XPS and TEM allow to verify the mechanisms of different damages formed in graphene on different substrates and to observe the microscopic morphology of damages. The defect densities (n_D) of graphene on different substrates are $n_{D(Cu)} = (1.44 \pm 0.40) \times 10^{11} \text{ cm}^{-2}$, $n_{D(Ni)} = (2.05 \pm 0.57) \times 10^{11} \text{ cm}^{-2}$, $n_{D(graphite)} = (2.17 \pm 0.60) \times 10^{11} \text{ cm}^{-2}$, respectively. It indicates that the graphene on the Cu substrate after proton beam irradiation suffered the least damages and the graphite sheets suffered the most. Different damages couldn't be completely ascribed to damages, stemming from the direct proton-C collisions. It's also because different substrates have different self-repairing capabilities of damages in graphene, which is demonstrated by the simulation results of SRIM and the different solubility of detached carbon ions in substrates. The self-repairing capability of damages in graphene of Cu substrate is the strongest among Cu, Ni and graphite. The graphite substrate has the weakest capability.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This research was supported by grants from the National Natural Science Foundation of China (Grant No. 21773250 and 51802308). Thanks to the members of the FDS team for their help.

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