


 Cite this: *Nanoscale*, 2020, **12**, 7995

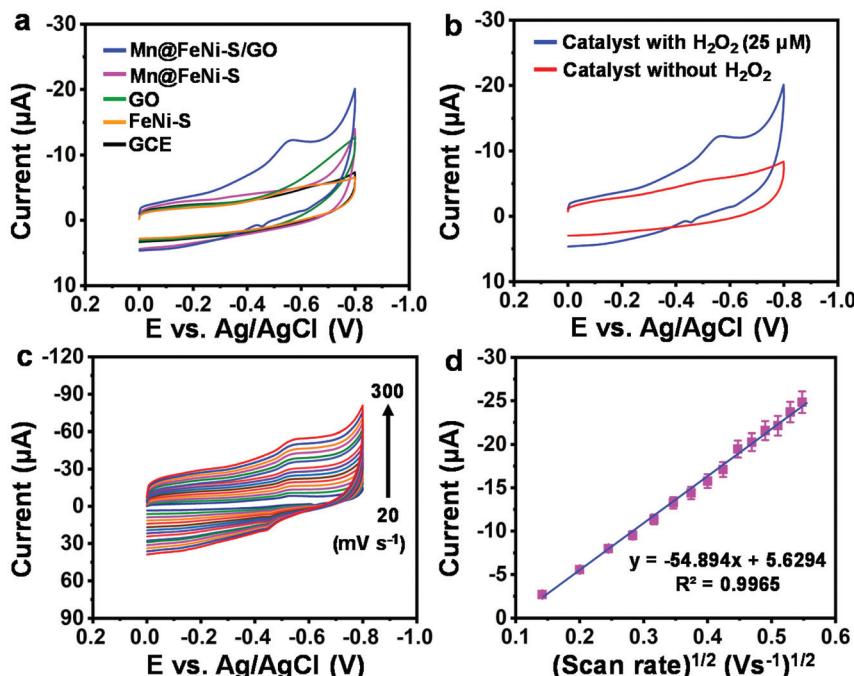
## Correction: A robust Mn@FeNi-S/graphene oxide nanocomposite as a high-efficiency catalyst for the non-enzymatic electrochemical detection of hydrogen peroxide

 Shaktivel Manavalan, <sup>a</sup> Jaysiva Ganesamurthi, <sup>a</sup> Shen-Ming Chen, <sup>a\*</sup> Pitchaimani Veerakumar <sup>b,c</sup> and Keerthi Murugan<sup>a</sup>

 DOI: 10.1039/d0nr90068k  
[rsc.li/nanoscale](http://rsc.li/nanoscale)

Correction for 'A robust Mn@FeNi-S/graphene oxide nanocomposite as a high-efficiency catalyst for the non-enzymatic electrochemical detection of hydrogen peroxide' by Shaktivel Manavalan *et al.*, *Nanoscale*, 2020, **12**, 5961–5972.

The authors have noticed that the original article contains an incorrect version of Fig. 6(b). Therefore, a corrected version of Fig. 6 is provided below:



**Fig. 6** (a) CV curves of bare GCE, GO, FeNi-S, Mn@FeNi-S, Mn@FeNi-S/GO-modified GCEs containing 25  $\mu$ M of  $\text{H}_2\text{O}_2$  at a scan rate of 50  $\text{mV s}^{-1}$ . (b) CV curves of Mn@FeNi-S/GO/GCE with and without addition of 25  $\mu$ M  $\text{H}_2\text{O}_2$ . (c) CV curves of Mn@FeNi-S/GO-modified GCE at different scan rates ranging from 20–300  $\text{mV s}^{-1}$  in 25  $\mu$ M of  $\text{H}_2\text{O}_2$ , and (d) the corresponding plot of peak current versus square root of the scan rate. All experiments were conducted in  $\text{N}_2$ -saturated 0.1 M PB (pH 7.0).

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

<sup>a</sup>Department of Chemical Engineering and Biotechnology, National Taipei University of Technology, Taipei 10608, Taiwan, Republic of China.  
 E-mail: smchen78@ms15.hinet.net; Fax: +886-2-27025238

<sup>b</sup>Department of Chemistry, National Taiwan University, Taipei 10617, Taiwan, Republic of China

<sup>c</sup>Institute of Atomic and Molecular Sciences, Academia Sinica, Taipei 10617, Taiwan, Republic of China

