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A fluororous biphasic drug delivery system triggered by low frequency ultrasound: controlled release from perfluororous discoidal porous silicon particles†

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Conventional drug delivery systems face unsatisfactory loading efficiency, poor biological bypass, and uncontrollable release, which are great barriers for improving the treatment of many diseases. Herein, a proof-of-concept of a fluororous biphasic drug delivery system (FB-DDS) triggered by low frequency ultrasound (LFUS) is proposed for the first time, where promoted incorporation and stabilization of therapeutic agents in nanocarriers was achieved through fluorine–fluorine interactions, and the encapsulated drugs were controllably released by external sources, resulting in minimized nonspecific toxicity and enhanced therapeutic efficacy. The FB-DDS was constructed from monodisperse, discoidal porous silicon particles (PSP) and was functionalized with 1*H*,1*H*,2*H*,2*H*-perfluorodecyltrimethoxysilane (FAS17) for loading perfluoropentane (PFP) and fluorinated drugs through fluorine–fluorine interactions. This delivery system was demonstrated by utilizing model compounds including a fluororous-tagged fluorescein and a fluorine containing antibiotic ciprofloxacin. Loading of the model molecules into fluorocarbon-coated carriers was facilitated by fluororous interactions, whereas ejection of the model molecules was promoted by applying LFUS to rapidly evaporate PFP. In the *in vitro* test, these carriers loaded with fluorine containing ciprofloxacin exhibited excellent antimicrobial activity against *Pseudomonas aeruginosa* biofilm formation. Overall, this innovative stimulus-responsive fluororous biphasic drug delivery system will be a promising candidate for practical applications as well as encouraging further investigation of drug delivery and controlled release strategies.

Received 24th April 2020
Accepted 25th June 2020

DOI: 10.1039/d0na00324g

rsc.li/nanoscale-advances

Introduction

Clinical use of many drugs suffers from several important drawbacks, such as poor solubility, nonspecific distribution, rapid metabolism, low bioavailability, and toxic side effects during delivery.^{1,2} Owing to the impressive progress in materials science and pharmaceuticals, nanoparticle-based drug delivery platforms have emerged as intriguing candidates for overcoming these limitations associated with conventional drug formulations, thus increasing drug efficacy and intracellular penetration.^{3–8} Unfortunately, great efforts and vast investments over the past thirty years have yielded only a few nanoformulations that were approved by the US Food and Drug Administration (FDA) for clinical applications, *e.g.*, Doxil® and Abraxane®.¹⁰ Moreover, most of the conventional drug delivery

systems (DDS) depend on either the conjugation and functionalization methodologies for incorporation of therapeutic drugs or the unique porous architectures of nanocarriers for drug encapsulation, which are not yet satisfactory in clinical usage and call for the development of potential alternative strategies.

“Fluororous biphasic system (FBS)” was a concept initially proposed by Horváth *et al.* in the 1990s for the selective separation or delivery of materials based on their difference in solubility.^{11–13} A FBS consists of a fluororous phase that carries dissolved reagents, and a second phase that could be any solvent with limited or no solubility in the fluororous phase. The fluororous phase is mostly perfluorocarbons (PFC). The FBS compatible reagents comprise fluororous moieties and will be soluble only or preferentially in the fluororous phase. By incorporation of fluororous tags, therapeutic agents can be stabilized in fluororous carriers filled with PFC through noncovalent fluorine–fluorine interactions during transportation and storage.^{14–17} Inspired by the FBS, we hypothesize that a fluororous biphasic drug delivery system (FB-DDS) can dramatically enhance the drug loading efficiency and biological stability, particularly for the vast majority of fluorine containing pharmaceuticals (Table S1, ESI†),^{18,19} which have ever been reported.

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† Electronic supplementary information (ESI) available: Synthesis of fluororous-tagged compounds, bacterial cell culture, plate counting assay, particle characterization, ¹H, ¹³C, and ¹⁹F NMR spectra of perfluorooctan-1-yl fluorescein isothiocyanate (FITC-C₈F₁₅), the plot of fluorescence intensity vs. concentration of FITC-C₈F₁₅, the surface plot images at three different angles for the fluororous particles, fluorescence microscopy images of the fluorinated FITC molecules incorporated in fluororous particles. See DOI: 10.1039/d0na00324g

Delivery of drugs from the carrier to the target is mainly dependent on diffusion that can be blocked by physical barriers in the body system. To achieve a rapid and controlled release in targeted tissues or cells where the particles accumulate, as well as to reduce systemic toxicity, stimulus-responsive nanoparticles are often employed.^{20–22} These nanoparticles present an intriguing combination of properties that release drugs on demand through external energy sources or triggers at a desired time, location and/or dose, *i.e.*, temperature,^{23,24} pH,²⁵ magnetic field,^{26,27} illumination,²⁸ or ultrasound.^{29,30} Low frequency ultrasound (LFUS) triggered release has features such as being non-invasive, permeable, precisely controllable using frequency, power density, duty cycles, and time, and can be focused on the targeted regions with a resolution of 1–2 mm for release.³¹ A variety of LFUS responsive carriers such as liposomes,^{32,33} micelles,³⁴ hydrogels,^{35,36} and emulsion droplets^{37–39} have been used to encapsulate poorly soluble drugs, protect them from premature degradation in the body, and decrease their interaction with healthy tissues to reduce side effects.^{40–44} Among them, biocompatible porous silicon particles (PSP) have drawn particular attention,^{45,46} as an excellent candidate^{47–50} for controlled drug delivery.^{51,52}

In the present work, a proof-of-concept of a FB-DDS triggered by LFUS is demonstrated. We filled fluororous PSP with volatile PFC, *i.e.*, perfluoropentane (PFP),^{53–55} an FDA approved biomedical agent.^{56–58} Although PFP has a low boiling point at 28 °C, its micro and nanodroplet emulsions remain in a metastable state even above the boiling point, attributed to a higher internal pressure in these droplets caused by surface tension, or namely Laplace pressure.^{59,60} Thus, fluorine containing drugs can be stabilized in fluororous nanocarriers, which can form a fluororous biphasic system during delivery. Furthermore, upon interaction with LFUS, spontaneous vaporization of these droplets can occur, producing gas bubbles of equilibrium volume hundred times larger than that of the original droplets.⁶¹ The acoustic cavitation produces local extremes of temperature and pressure of these droplets confined in the nanochannels, which facilitates drug delivery across the endothelial barrier. To highlight the promising practical applications of the FB-DDS, we examined two model compounds including a short fluororous-tagged fluorescein and a fluorine containing antibiotic ciprofloxacin. These model compounds can be easily loaded into the fluororous discoidal carriers and the LFUS delivery system exhibits a desired controlled release profile.

Experimental

Materials

1H,1H,2H,2H-Perfluorodecyltrimethoxysilane (FAS17, 97%) was purchased from Apollo Scientific. Decafluorobutane (98%), dodecafluoro-*n*-pentane (99%), and perfluoro-*n*-hexane (99%) were purchased from SynQuest Labs. 1H,1H-Perfluorooctylamine (97%) was purchased from Alfa Aesar. Fluorescein isothiocyanate (FITC, 97%) and Dulbecco's Modified Eagle Medium (DMEM) were purchased from Thermo Scientific. Ciprofloxacin (98%), perfluoropentanoyl chloride (98%),

triethylamine (99%), and phosphate buffered saline (10× PBS) were purchased from Sigma-Aldrich. LB broth base (powder) and LB agar (powder) were purchased from Becton-Dickinson. Cell counting kit-8 (CCK-8) was purchased from Beyotime Biotechnology. All reactants were used as received without further purification.

Preparation of discoidal porous silicon particles

The fabrication of discoidal porous silicon particles (PSP) was modified from a reported procedure.⁵⁰ Briefly, a heavily doped P-type (100) silicon wafer with a resistivity of 0.005 ohm × cm (Silicon Quest, Inc.) was employed as a substrate. The substrate was exposed to 1 : 3 HF/ethanol solution and electrochemically etched with an applied current density of 10 mA cm^{−2} to form a porous silicon film with a mean pore size of 40 nm. Afterwards, a high electrical current (76 mA cm^{−2}) was applied to form a high porosity release layer. An 80 nm low temperature oxide (LTO) layer was deposited on the porous silicon film in a low pressure chemical vapor deposition (LPCVD) furnace. A standard photolithography process was used to pattern the arrays of 600 nm and 1000 nm circles with a contact aligner (K. Suss MA6 mask aligner) and NR9-500P photoresist (Futurrex Franklin). The patterns were transferred to LTO and porous silicon films by reactive ion etching in tetrafluoromethane plasma (Plasmatherm Batchtop VII). After stripping the photoresist and the LTO, the as-prepared particles were retained on the substrate as an array allowing for further surface functionalization. Noticeably, this fabrication strategy provides sufficient flexibility to fabricate multistage vectors⁶² that can be directly translated to large scale production within the framework of established good manufacturing practice semiconductor processes.

Functionalization of porous silicon particles with perfluorosilane

The as-made porous silicon particles on the wafer were rinsed with ethanol, dried with N₂, and subsequently immersed in a piranha solution (H₂SO₄/H₂O₂, v/v = 2 : 1, **caution**: hot and corrosive) for 10 min to oxidize the silicon surface to silanol. After cooling down to room temperature, the array was rinsed with copious Milli-Q water and dried under reduced pressure. The oxidized array was then immersed in 1 mL of 1 mM 1H,1H,2H,2H-perfluorododecyltrimethoxysilane (FAS17) in dichloromethane and was incubated statically for 2 hours at room temperature. Afterwards, the array was rinsed with ethanol and Milli-Q water successively, and dried with N₂. Finally, the array was placed in a 25 mL Schlenk flask and annealed in a N₂ atmosphere at 100 °C for 1 hour. The resultant fluororous particles were denoted as F-PSP.

Loading of perfluorocarbon (PFC) and fluoro-compounds

Perfluorocarbon. A 5 × 5 mm² array with fluororous porous silicon particles was transferred into a 5 mL vial with 1 mL of decafluorobutane (PFB, C₄F₁₀, b.p. = −1.7 °C) and kept in a dry CO₂/acetone bath, 1 mL of dodecafluoro-*n*-pentane (PFP, C₅F₁₂, b.p. = 28 °C), or 1 mL of perfluoro-*n*-hexane (PFH, C₆F₁₄, b.p. =



59 °C) and kept in an ice bath for 5 min. Subsequently, the dry CO₂/acetone bath or the ice bath was removed. The vial was decapped to evaporate excess C₄F₁₀ and C₅F₁₂ at room temperature or decant the excess C₆F₁₄ to give the PFC loaded fluororous particles, referred to as PFC-F-PSP.

Perfluorooctanil fluorescein isothiocyanate (FITC-C₈F₁₅). A 5 × 5 mm² array with fluororous porous silicon particles was introduced into a 5 mL vial with C₅F₁₂ (1 mL) and kept in an ice bath for 5 min, and then transferred into another 5 mL vial with FITC-C₈F₁₅ (0.79 mg, 1 μmol) in ethanol (1 mL) and kept at room temperature for 15 min. Afterwards, the array was dried with N₂, and rinsed with ethanol to remove the excess FITC-C₈F₁₅.

Ciprofloxacin (Cip). A 5 × 5 mm² array with fluororous porous silicon particles was introduced into a 5 mL vial with C₅F₁₂ (1 mL) and kept in an ice bath for 5 min and then transferred into another 5 mL vial with Cip (10 μg, 1.73 nmol) in 0.01 M PBS (1 mL) and kept for 30 min. Afterwards, the wafer was dried with N₂ and rinsed with 0.01 M PBS to remove the excess Cip.

Controlled release of FITC-C₈F₁₅

The FITC-C₈F₁₅ loaded fluororous porous silicon particles on the 5 × 5 mm² array were immersed in a 5 mL vial with 0.01 M PBS solution (1 mL). Low frequency ultrasound (LFUS) was applied to the sample for 5 seconds by using an ultrasound bath (Branson model 1510) operating at a frequency of 40 kHz. After that, the array was withdrawn from the solution and dried with N₂. The concentration of the released FITC-C₈F₁₅ in the solution was quantitated against a calibration curve of FITC-C₈F₁₅ in the range of 0–50 μM (Fig. S1, ESI†), through the measurement of FITC fluorescence intensity at λ_{ex}/λ_{em} 495/535 nm using a microplate reader (Tecan Spark®). The wafer was also examined by reflected bright field and fluorescence microscopy imaging with a Nikon eclipse 80i microscope at the excitation and emission wavelengths of 494/517 nm for FITC-C₈F₁₅. All

images were processed using NIS-Elements AR3.0 software (Nikon).

Biofilm inhibition assay

Preparation of bacterial suspension. A *Pseudomonas aeruginosa* strain expressing the green fluorescent protein and a carbenicillin resistance gene was referred to as PAO1-GFP. A single isolated PAO1-GFP colony was inoculated in a fresh LB broth medium (5 mL) containing carbenicillin (300 μg mL⁻¹) and orbitally shaken overnight at 37 °C. The optical density at 600 nm ([OD]₆₀₀) was adjusted to 0.25 for each bacterial culture after overnight incubation, corresponding to a bacterial concentration of ~10⁸ CFU mL⁻¹.

Controlled release of Cip. 1 mL of the fresh bacterial suspension ([OD]₆₀₀ = 0.25) was inoculated into a well containing the Cip loaded fluororous porous silicon particles (1 mg). Afterwards, the 48-well plate was sealed with a secondary container, sonicated in a LFUS bath for 5 seconds to release the Cip, and statically incubated at 37 °C for 24 hours. The control groups were treated without F-PSP, Cip, or LFUS. The negative control group was without any treatment. After completion, the culture solutions in the 48-well plate were gently removed and the biofilm formation on the bottom of the 48-well plate was examined by microscopy imaging under fluorescence with a Nikon eclipse 80i microscope. The images were processed using NIS-Elements AR3.0 software. The amount of Cip on the fluororous particles was quantified against the calibration curve of the Cip standard solution in the range of 0–100 μM (Fig. S2, ESI†) via LC-MS analysis. The calculation of load rate is summarized in Table S2 (ESI†).

Evaluation of cytotoxicity

The *in vitro* cytotoxicity was evaluated using a cell counting kit-8 (CCK-8). Briefly, HeLa cells were seeded at a concentration of 1 × 10⁴ cells per well in a 96-well plate containing DMEM (100 μL) that was supplied with 10% FBS, streptomycin (100 μg mL⁻¹),

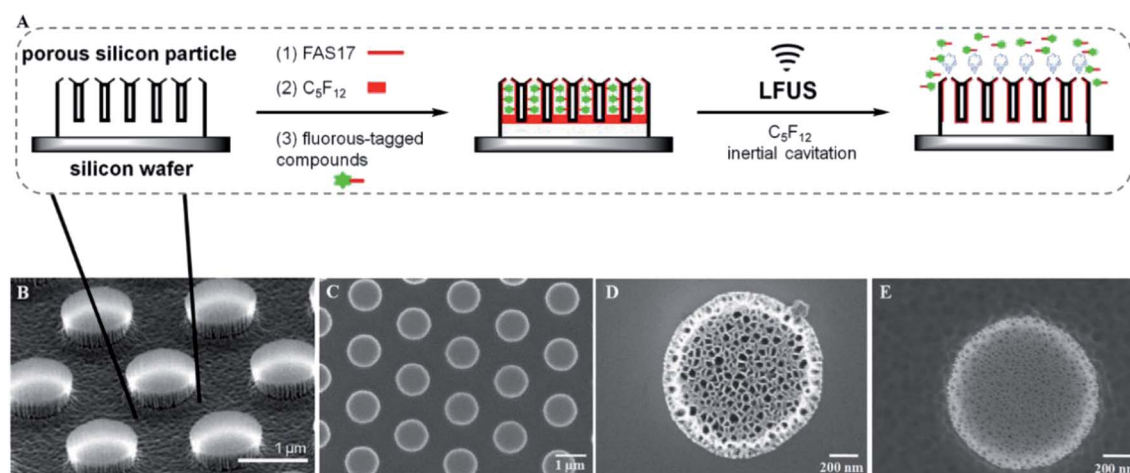


Fig. 1 (A) Schematic illustration of the general procedure for the low frequency ultrasound triggered release of fluororous-tagged compounds. The tilted view (B) and top view (C) of the SEM images of the discoidal porous silicon particles deposited on a silicon wafer. SEM images of the porous morphology of the release layer (D) and the device layer (E).



and penicillin ($100 \mu\text{g mL}^{-1}$). The culture medium was then replaced with fresh medium supplemented with C_5F_{12} loaded fluoros particles (C_5F_{12} -F-PSP, 1 mg mL^{-1}), fluoros particles (F-PSP, 1 mg mL^{-1}), and unmodified particles (1 mg mL^{-1}) after culturing for 24 hours (37°C , $5\% \text{ CO}_2$) and was continued to incubate for additional 24–48 h. The untreated cells served as the negative control. At the selected time point, the medium was replaced by CCK-8 solution and statically incubated at 37°C for 1 hour. The absorbance at 450 nm was read using a microplate reader. The percentage of cell viability was calculated by normalizing the absorbance of the treated samples with that of the untreated sample.

Results and discussion

In this study, we investigated ultrasound triggered release by utilizing discoidal porous silicon particles (PSP) as the biocompatible carriers. The particle surface was modified with perfluorosilane molecules to allow the incorporation of perfluorocarbon (PFC) and fluoros-tagged compounds into the pores through fluorine–fluorine interactions.^{63,64} As illustrated in Fig. 1A, the discoidal particles deposited on a silicon wafer were functionalized with 1H,1H,2H,2H-perfluorodecyltrimethoxysilane (FAS17) to generate a fluorinated surface.⁶⁵ The amplified area represents a pore containing PFC and fluoros-tagged compounds. Remarkably, loading of PFC and fluoros-tagged compounds is facilitated by fluorine–fluorine interactions, and after evaporation of PFC under LFUS, the fluoros-tagged compounds released rapidly from the pores.

Characterization of discoidal porous silicon particles

The discoidal particles were fabricated by the combination of photolithography and controlled electrochemical etching.⁵⁰ Fig. 1B shows the formation of the ordered discoidal particle array on a silicon wafer. The average particle size was determined as $1.11 \pm 0.01 \mu\text{m}$ with a thickness of $320 \pm 17 \text{ nm}$ and a constant distance between particles of $0.75 \pm 0.02 \mu\text{m}$ (Fig. 1C). The particles consisted of two layers of different pore sizes. The pore sizes of the top layer – release layer (Fig. 1D) and the bottom layer – device layer (Fig. 1E) were determined as $75.4 \pm 9.3 \text{ nm}$ and $13.5 \pm 3.5 \text{ nm}$, respectively. Noticeably, this bilayer structure is anticipated to form internal cavities capable of efficient encapsulation and reduce the diffusion of the payload.

Prior to the fluorination, a systematic optimization of perfluorosilane concentration in the range of 0.1 mM to 10 mM indicated that 1 mM was optimal to generate a high density fluorine film on the particles (Fig. S3, ESI†). Successively, the fluoros particles were characterized by SEM, TEM, EDS, FTIR, and XPS. Both TEM (Fig. 2A–C) and SEM (Fig. 2D) analyses confirmed the preservation of the bilayer porous morphology in the particles after fluorination, with the pore sizes of the release layer (Fig. 2A and S4A, ESI†) and the device layer (Fig. 2B and S4B, ESI†) being $57.5 \pm 15.2 \text{ nm}$ and $14.4 \pm 5.6 \text{ nm}$. The tilt view (Fig. 2C) of the fluoros particles shows dense cavities inside the particles. The EDS analysis revealed the presence of fluorine

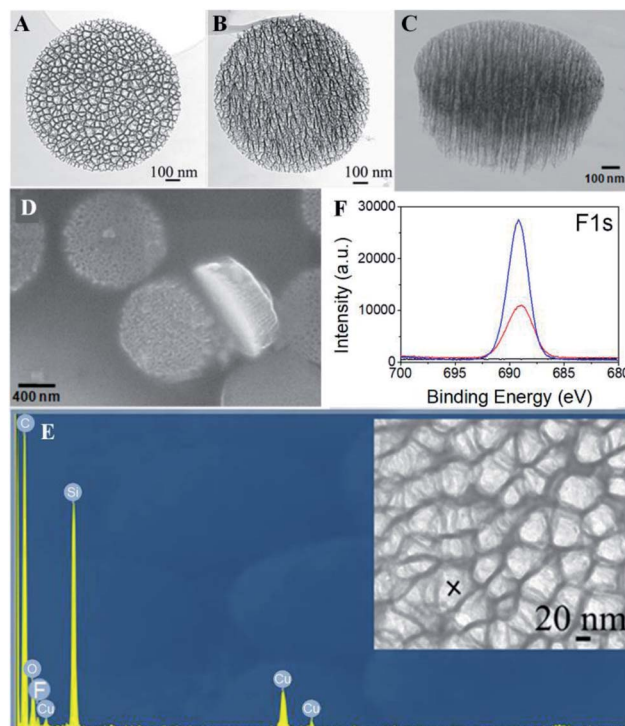


Fig. 2 TEM images of the release layer (A) and device layer (B), and tilt view (C) and SEM image (D) of the fluoros discoidal porous silicon particles. (E) EDS spectrum of the area marked with an "X" on the fluoros porous silicon particles. (F) High-resolution F1s scans of the unmodified porous silicon particles (black line), fluoros porous silicon particles (red line), and FITC- C_8F_{15} incorporated fluoros porous silicon particles (blue line).

groups within the porous structure, together with silicon, oxygen, and carbon as the particle compositional elements (Fig. 2E). The detected copper was assumed to be from the copper grid.

Further, the fluorine film density on the particles was determined by XPS analysis by using a self-assembled monolayer (SAM) of $\text{CF}_3(\text{CH}_2)_{13}\text{SH}$ on Au(111) as a reference.⁶⁶ The F1s peak at 688 eV in the high resolution XPS indicated the presence of fluorine groups in the fluoros particles (Fig. 2F, red line) and the absence of the F1s signal in the unmodified particles (Fig. 2F, black line). The F1s peak areas of the fluoros particles and the CF_3 -terminated SAM surface were $A_{\text{F-PSP}} = 16\,717 \pm 2459$ and $A_{\text{F-SAM}} = 7577 \pm 286$. The numbers of fluorine atoms on the FAS17 and the $\text{CF}_3(\text{CH}_2)_{13}\text{SH}$ are $N_{\text{F-PSP}} = 17$ and $N_{\text{F-SAM}} = 3$, respectively. Considering the known density of the SAM on Au(111) as a standard ($D_{\text{F-SAM}} = 4.6 \times 10^{14}$ molecules per cm^2),⁶⁶ the fluorine density on the particles can be calculated as $(1.8 \pm 0.1) \times 10^{13}$ molecules per cm^2 , and the perfluorosilane coverage is approximately 39%.

The FTIR spectra (Fig. 3A) of the pure FAS17 molecules (red line) and the fluoros particles (black line) were collected, and the FTIR spectra (Fig. 3B) of the fluoros particles (red line) and the unmodified particles (black line) were also recorded. In the fluoros particle spectrum (Fig. 3A), the peak at $\sim 1154 \text{ cm}^{-1}$ is assigned to the asymmetric stretching of $-\text{CF}_2-$, and the peaks



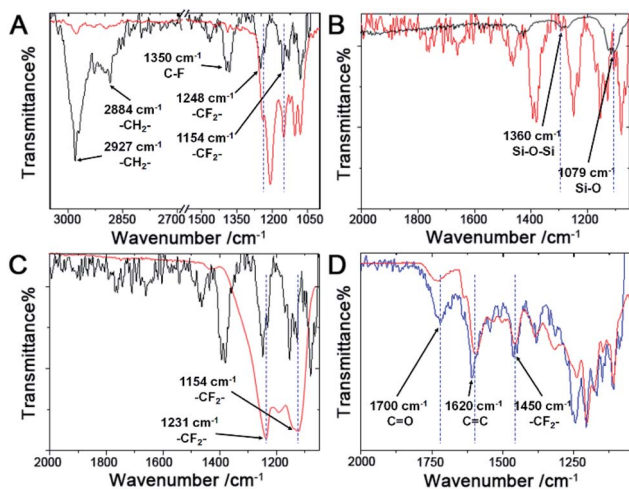


Fig. 3 (A) FTIR spectra of the fluororous porous silicon particles (black line) and pure FAS17 molecules (red line). (B) FTIR spectra of the unmodified porous silicon particles (black line) and the fluororous porous silicon particles (red line). (C) FTIR spectra of the fluororous porous silicon particles (black line) and the C_5F_{12} incorporated fluororous porous silicon particles (red line). (D) FTIR spectra of the fluororous porous silicon particles after incorporation of FITC- C_8F_{15} (blue line) and the pure FITC- C_8F_{15} molecules (red line).

at 1248 and 1231 cm^{-1} are attributed to the symmetric stretching of $-CF_2-$ and $-CF_3$. The peaks at ~ 2884 and 2927 cm^{-1} are attributed to the symmetric and asymmetric stretching of $-CH_2-$, respectively. The peak at $\sim 1350\text{ cm}^{-1}$ is attributed to immobilized C-F groups.^{48,67} The peaks at $\sim 1360\text{ cm}^{-1}$ and $\sim 1079\text{ cm}^{-1}$ are the stretching of Si-O-Si and Si-O groups.⁶⁸

Loading of perfluorocarbon (PFC)

Fluorous chemistry is dependent on unique fluorine-fluorine interactions with which fluorocarbons are attracted to each other, much more than to other media.⁴⁵ On this basis, fluororous interactions can be applied for loading of PFC and cargoes with a fluororous tag.

The boiling points of C_4F_{10} (PFB), C_5F_{12} (PFP), and C_6F_{14} (PFH) are $-1.7\text{ }^{\circ}C$, $28\text{ }^{\circ}C$, and $59\text{ }^{\circ}C$, but they remain as superheated liquids in the porous structure due to the Laplace pressure. To investigate the loading efficiency of PFC in the fluororous particles, FTIR was employed. In Fig. 3C, two signature peaks at ~ 1154 and 1231 cm^{-1} , attributed to the asymmetric and symmetric stretching of $-CF_2-$, appeared after the loading of PFC. The higher intensity peak at 1231 cm^{-1} was utilized to monitor the PFC loading in the fluororous particles. The release profiles at room temperature (Fig. 4) were plotted using absorbance at 1231 cm^{-1} vs. time and were fitted using the first-order release equation $\ln\left(1 - \frac{A_t}{A_{\infty}}\right) = -kt$. As a result, the release rate constant (k) followed the order: $k(C_4F_{10}) \gg k(C_5F_{12}) \approx k(C_6F_{14})$, which was consistent with the corresponding boiling point of each PFC. In order to achieve both a steady release profile and a rapid explosion effect, C_5F_{12} was therefore chosen as a model PFC in the following experiments.

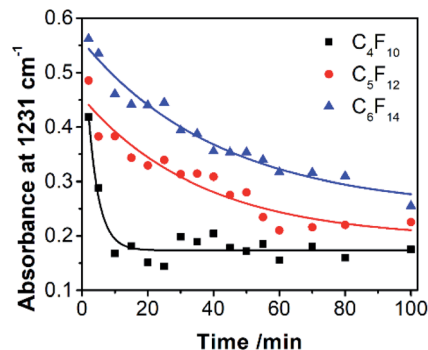


Fig. 4 The release profiles of PFC at room temperature were plotted using absorbance at 1231 cm^{-1} vs. time and were fitted using the first-order release equation $\ln\left(1 - \frac{A_t}{A_{\infty}}\right) = -kt$, where A is the absorbance of PFC at 1231 cm^{-1} , and t is the release time. The release rate constants (k) are 0.3092 (C_4F_{10}), 0.02753 (C_5F_{12}), and 0.02372 (C_6F_{14}). The correlation coefficient (r^2) is 0.9575 (C_4F_{10}), 0.8897 (C_5F_{12}), and 0.9148 (C_6F_{14}), respectively.

Loading of fluororous-tagged compounds

A fluorinated fluorescein isothiocyanate (FITC) derivative was exploited as a model compound in this study. Fluorinated FITC- C_8F_{15} molecules were incorporated into the fluororous particles as indicated in the Experimental section. The successful incorporation of fluorinated FITC molecules in the fluororous particles was demonstrated by XPS and FTIR. In the XPS spectra (Fig. 2F), an increase in the F1s peak intensity of the FITC- C_8F_{15} incorporated fluororous particles (blue line) in comparison with the fluororous particles (red line) and unmodified particles (black line) indicated the presence of FITC- C_8F_{15} in the fluororous particles. In the FTIR spectra (Fig. 3D), the feature peaks of the fluorinated FITC molecules (red line) at 1450 cm^{-1} , 1620 cm^{-1} ,

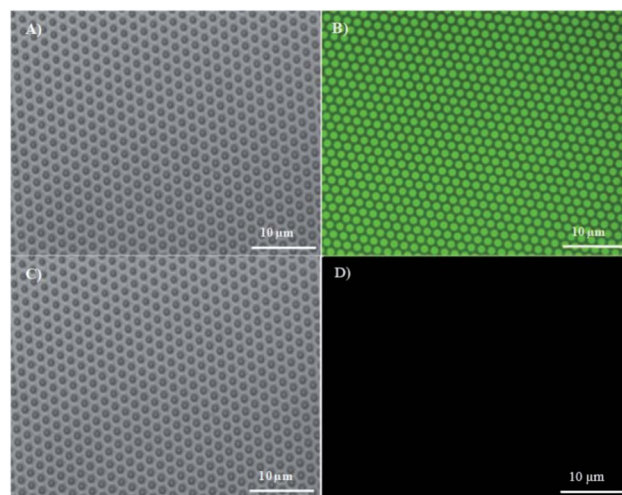


Fig. 5 Reflected bright-field (A) and fluorescence microscopy images (B) for the fluororous porous silicon particles deposited on a silicon wafer after treatment with FITC- C_8F_{15} . Reflected bright-field (C) and fluorescence microscopy images (D) for the unmodified porous silicon particles deposited on a silicon wafer after treatment with FITC- C_8F_{15} .



and 1700 cm^{-1} appeared for the FITC- C_8F_{15} incorporated fluororous particles (blue line).

The incorporation of cargoes was also visualized by fluorescence microscopy. A monodisperse array of particles on the silicon wafer was observed in the reflected bright-field image (Fig. 5A). After treatment with FITC- C_8F_{15} , each particle exhibited green fluorescence as shown in the fluorescence microscopy image (Fig. 5B), indicating the presence of FITC- C_8F_{15} in the fluororous particles. Moreover, the surface plot image at three different angles shown in Fig. S5 (ESI[†]) confirmed the presence of FITC- C_8F_{15} in each fluororous particle. A control experiment was carried out by using a monodisperse array of unmodified particles after the same treatment with FITC- C_8F_{15} (Fig. 5C). In contrast, an overwhelming absence of fluorescence (Fig. 5D) implied the failure of loading of fluorinated molecules into the particle pores. This result validated the relevance of perfluorosilane functionalization for a successful incorporation of fluorinated molecules into the fluororous particles.

Low frequency ultrasound triggered controlled release of fluororous-tagged compounds

The primary goal of the fabrication of this ultrasound triggered structure is to apply it as a delivery system for the controlled release of therapeutic molecules in biomedical applications. The low frequency ultrasound (10–60 kHz) is known to enhance the permeability of biological membranes and improve the release efficacy of therapeutic agents.⁴⁴ The ultrasonic power density was determined calorimetrically as described by Hill *et al.*⁶⁹ In this study, the application of fabricated fluororous particles with LFUS triggered controlled release was explored. The release experiments were performed with the fluororous particles on a silicon wafer aiming to establish a convenient observation by optical microscopy. In order to simulate physiological conditions, the array was immersed in a vial containing 0.01 M phosphate buffered saline (PBS), and sonicated for 5 seconds in an ultrasonic bath, operating at 40 kHz at room temperature.

Before applying LFUS, each particle incorporated with fluorescent molecules displayed a green dot on the surface (Fig. 6A). After applying LFUS, the fluorescent molecules were ejected out of the particles and two different regions were observed in the microscopy images. In region (I), the FITC molecules were released simultaneously from the porous particles and the silicon wafer surface (Fig. 6B), and transferred to the PBS solution. In region (II), the FITC molecules were released selectively from the porous particles but remained on the silicon wafer surface (Fig. 6C). Fig. 6D presents a mixed region of (I) and (II). The presence of FITC molecules on the silicon wafer accounted for the multilayer structure of the pores. The mean fluorescence intensity (MFI) was measured as 11 106 before sonication and was decreased to 1476 in region (I) and 5666 in region (II) after sonication. The MFI of the mixed regions of (I) and (II) was 2411. The inertial cavitation during ultrasound produced the release of FITC- C_8F_{15} molecules through the release layers. The result suggests that a multi-level controlled release of cargoes can be tuned by adjusting the sonication time.

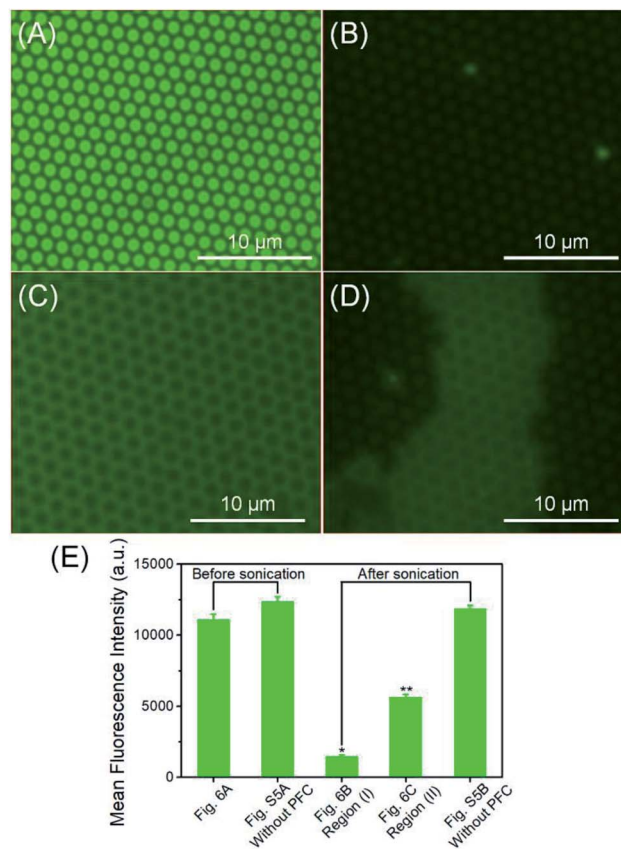


Fig. 6 Fluorescence microscopy images of the fluorinated FITC loaded fluororous particles deposited on a silicon wafer before (A) and after (B), (C), and (D) after 5 seconds of 40 kHz ultrasound application (note: the acquisition time was 200 ms in all the images). The mean fluorescence intensity (MFI) was extracted from these images ($A = 642\text{ }\mu\text{m}^2$). (E) Comparison of the MFI before and after applying LFUS. The control groups are the fluorinated FITC loaded fluororous particles deposited on a silicon wafer without C_5F_{12} incorporation. * $P < 0.001$ and ** $P < 0.001$ from the MFI extracted from (A).

In addition, the fluorinated FITC released into the PBS solution was analyzed using a fluorescence spectrometer to estimate the fluorinated FITC concentration. As a precaution to avoid artifacts from the detached particles, the solution was centrifuged at 12 000 rpm for 40 min prior to the measurement. The fluorescence intensity of the supernatant was measured as 42 546, corresponding to 48 μM fluorinated FITC in 400 μL , indicating that all the FITC molecules originated from the sonication application.

Finally, the relevance of the treatment with PFC was confirmed by a control experiment developed by sonication of the fluororous particles incorporated only with fluorinated FITC molecules. The fluorescence microscopy images of the fluororous particles on a silicon wafer before (Fig. S6A, ESI[†]) and after (Fig. S6B, ESI[†]) applying LFUS without C_5F_{12} incorporation are presented. As expected, the MFI was 12 353 and 11 878, demonstrating that the controlled release of the fluorinated FITC molecules was produced as a consequence of rapid PFC evaporation during ultrasound application (Fig. 6E).



Application for biofilm inhibition

The same methodology was used to design an antibiotic controlled release system on the fluororous particles, aiming to prevent pathogenic biofilm formation. Ciprofloxacin (Cip) is a well-known fluorine containing antibiotic and was recently reported as an active antibiotic to modify contact lenses against the notorious “superbug” pathogenic *Pseudomonas aeruginosa* strain (PAO1) via fluorine chemistry.¹⁷ A genetically modified PAO1-GFP strain to express the green fluorescent protein was used in this study. The PAO1-GFP suspension was inoculated into a 48-well plate containing Cip and C₅F₁₂ incorporated fluororous particles. The load rate of Cip on the fluororous particles was determined to be 58%. The control groups contained the fluororous particles without Cip or LFUS application, or bare LFUS application. The bacterial culture without any treatment was the negative control. As shown in Fig. 7A, no biofilm was formed on the Cip loaded fluororous particles subjected to LFUS. In contrast, densely packed biofilms were observed on the control groups, including the bare LFUS application (Fig. 7B), the fluororous particles without Cip (Fig. 7C), the Cip loaded fluororous particles without LFUS application (Fig. 7D), and the negative control groups (Fig. 7E). Further, the viable bacterial cells in these biofilms were counted (Fig. 7F). The sharp reduction by >5 orders of

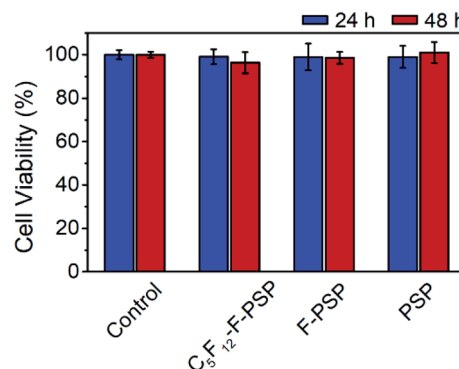


Fig. 8 Hela cell viability evaluated by CCK-8 assay after treatment with the indicated conditions for 24 to 48 h. Each value represents the relative viability compared to the control group. Each data point represents an average of five replicates and the error bars represent one standard deviation from the mean.

magnitude in the bacterial cell number in the experiment group compared to that of control groups demonstrated the effectiveness of this delivery system.

Cytotoxicity

The cytotoxicity of the C₅F₁₂ loaded fluororous particles (C₅F₁₂-F-PSP, 1 mg mL⁻¹), fluororous particles (F-PSP, 1 mg mL⁻¹) and unmodified particles (1 mg mL⁻¹) was evaluated using Hela cells as a model. The *in vitro* cell viability was determined using a commercial cell counting kit-8 (CCK-8). No apparent impact on cell viability was observed up to 1 mg mL⁻¹ concentration of all these particles within 48 h of incubation, which is similar to that of the control group (Fig. 8). The results suggested that the C₅F₁₂ loaded fluororous particles were biosafe to mammalian cells.

Conclusions

In conclusion, we have constructed a biphasic fluororous drug delivery system (FB-DDS) triggered by low frequency ultrasound (LFUS). The discoidal porous silicon particles were synthesized by a combined procedure of photolithography and controlled electrochemical HF etching, and then were functionalized with perfluorosilane (FAS17). This delivery system was demonstrated by utilizing model compounds including a fluororous-tagged fluorescein and a fluorine containing antibiotic ciprofloxacin. Loading of the model molecules into fluorocarbon-coated carriers was facilitated by fluororous interactions, whereas ejection of the model molecules was promoted by applying LFUS to rapidly evaporate perfluoropentane. In the *in vitro* test, these carriers loaded with fluorine containing ciprofloxacin exhibited excellent antimicrobial activity against *Pseudomonas aeruginosa* biofilm formation. Overall, this innovative stimulus-responsive fluororous biphasic drug delivery system will be a promising candidate for practical applications as well as encouraging further investigation of drug delivery and controlled release strategies.

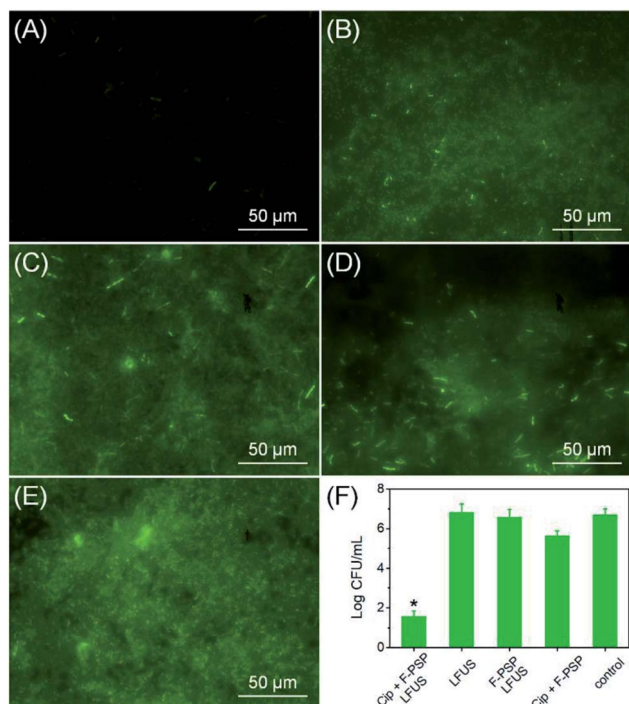


Fig. 7 (A) Representative fluorescence image of GFP-transformed *Pseudomonas aeruginosa* (PAO1-GFP) cultured in a 48-well plate with Cip and C₅F₁₂ incorporated fluororous particles after LFUS application for 24 h at 37 °C. The representative fluorescence images of the biofilms of PAO1-GFP cultured in a 48-well plate without the Cip loaded fluororous particles (B), Cip (C) and LFUS (D) treatment, and the negative control (E) after 24 h incubation at 37 °C. (F) Viable bacterial CFU counts of PAO1-GFP remaining on the bottom of the 48-well plate. **P* < 0.001 from data obtained for the control.



Conflicts of interest

The authors declare no competing financial interest.

Acknowledgements

This study was supported by the National Natural Science Foundation of China (Grant No. 31800800) and Natural Science Foundation of Shandong Province, China (Grant No. ZR2019BC101).

Notes and references

- 1 M. M. Rubiana and P. S. Luciano, *Curr. Drug Targets*, 2004, **5**, 449–455.
- 2 E. Blanco, H. Shen and M. Ferrari, *Nat. Biotechnol.*, 2015, **33**, 941–951.
- 3 T. Simon-Yarza, A. Mielcarek, P. Couvreur and C. Serre, *Adv. Mater.*, 2018, **30**, 1707365.
- 4 S. T. Gao, G. S. Tang, D. W. Hua, R. H. Xiong, J. Q. Han, S. H. Jiang, Q. L. Zhang and C. B. Huang, *J. Mater. Chem. B*, 2019, **7**, 709–729.
- 5 L. Wang, M. Zheng and Z. G. Xie, *J. Mater. Chem. B*, 2018, **6**, 707–717.
- 6 E. R. Ruskowitz and C. A. DeForest, *Nat. Rev. Mater.*, 2018, **3**, 17.
- 7 H. Wang, Q. W. Chen and S. Q. Zhou, *Chem. Soc. Rev.*, 2018, **47**, 4198–4232.
- 8 F. Q. Tang, L. L. Li and D. Chen, *Adv. Mater.*, 2012, **24**, 1504–1534.
- 9 Y. Barenholz, *J. Control. Release*, 2012, **160**, 117–134.
- 10 M. J. Hawkins, P. Soon-Shiong and N. Desai, *Adv. Drug Deliv. Rev.*, 2008, **60**, 876–885.
- 11 I. T. Horvath and J. Rabai, *Science*, 1994, **266**, 72–75.
- 12 I. T. Horváth, *Acc. Chem. Res.*, 1998, **31**, 641–650.
- 13 B. Barabás, O. Fülöp, R. Molontay and G. Pályi, *ACS Sustain. Chem. Eng.*, 2017, **5**, 8108–8118.
- 14 F. A. Jaipuri, B. Y. M. Collet and N. L. Pohl, *Angew. Chem., Int. Ed.*, 2008, **47**, 1707–1710.
- 15 R. Berger, G. Resnati, P. Metrangolo, E. Weber and J. Hulliger, *Chem. Soc. Rev.*, 2011, **40**, 3496–3508.
- 16 C. M. Santos, A. Kumar, S. S. Kolar, R. Contreras-Caceres, A. McDermott and C. Cai, *ACS Appl. Mater. Interfaces*, 2013, **5**, 12789–12793.
- 17 G. T. Qin, Z. L. Zhu, S. H. Li, A. M. McDermott and C. Z. Cai, *Biomaterials*, 2017, **124**, 55–64.
- 18 J. Wang, M. Sánchez-Roselló, J. L. Aceña, C. del Pozo, A. E. Sorochinsky, S. Fustero, V. A. Soloshonok and H. Liu, *Chem. Rev.*, 2014, **114**, 2432–2506.
- 19 Y. Zhou, J. Wang, Z. Gu, S. Wang, W. Zhu, J. L. Aceña, V. A. Soloshonok, K. Izawa and H. Liu, *Chem. Rev.*, 2016, **116**, 422–518.
- 20 S. Mura, J. Nicolas and P. Couvreur, *Nat. Mater.*, 2013, **12**, 991–1003.
- 21 M. Karimi, A. Ghasemi, P. S. Zangabad, R. Rahighi, S. M. M. Basri, H. Mirshekari, M. Amiri, Z. S. Pishabad, A. Aslani, M. Bozorgomid, D. Ghosh, A. Beyzavi, A. Vaseghi, A. R. Aref, L. Haghani, S. Bahrami and M. R. Hamblin, *Chem. Soc. Rev.*, 2016, **45**, 1457–1501.
- 22 O. C. Farokhzad and L. Robert, *ACS Nano*, 2009, **3**, 16–20.
- 23 R. R. Costa, C. A. Custódio, F. J. Arias, J. C. Rodríguez-Cabello and J. F. Mano, *Nanomedicine*, 2013, **9**, 895–902.
- 24 X. Wu, Z. Wang, D. Zhu, S. Zong, L. Yang, Y. Zhong and Y. Cui, *ACS Appl. Mater. Interfaces*, 2013, **5**, 10895–10903.
- 25 J. Cui, Y. Yan, Y. Wang and F. Caruso, *Adv. Funct. Mater.*, 2012, **22**, 4718–4723.
- 26 A. A. McBride, D. N. Price, L. R. Lamoureux, A. A. Elmaoued, J. M. Vargas, N. L. Adolphi and P. Muttli, *Mol. Pharm.*, 2013, **10**, 3574–3581.
- 27 S.-H. Hu, T.-Y. Liu, H.-Y. Huang, D.-M. Liu and S.-Y. Chen, *Langmuir*, 2008, **24**, 239–244.
- 28 A. Schroeder, M. S. Goldberg, C. Kastrup, Y. Wang, S. Jiang, B. J. Joseph, C. G. Levins, S. T. Kannan, R. Langer and D. G. Anderson, *Nano Lett.*, 2012, **12**, 2685–2689.
- 29 V. Frenkel, *Adv. Drug Deliv. Rev.*, 2008, **60**, 1193–1208.
- 30 R. Deckers and C. T. W. Moonen, *J. Contr. Release*, 2010, **148**, 25–33.
- 31 T. G. Leighton, *Prog. Biophys. Mol. Biol.*, 2007, **93**, 3–83.
- 32 S. L. Huang, *Adv. Drug Deliv. Rev.*, 2008, **60**, 1167–1176.
- 33 J. Tang, R. F. Liu and Z. F. Dai, *Prog. Chem.*, 2018, **30**, 1669–1680.
- 34 C. Oerlemans, W. Bult, M. Bos, G. Storm, J. F. W. Nijssen and W. E. Hennink, *Pharm. Res.*, 2010, **27**, 2569–2589.
- 35 N. Huebsch, C. J. Kearney, X. H. Zhao, J. Kim, C. A. Cezar, Z. G. Suo and D. J. Mooney, *Proc. Natl. Acad. Sci. U. S. A.*, 2014, **111**, 9762–9767.
- 36 M. L. Fabiilli, C. G. Wilson, F. Padilla, F. M. Martin-Saavedra, J. B. Fowlkes and R. T. Franceschi, *Acta Biomater.*, 2013, **9**, 7399–7409.
- 37 M. L. Fabiilli, J. A. Lee, O. D. Kripfgans, P. L. Carson and J. B. Fowlkes, *Pharm. Res.*, 2010, **27**, 2753–2765.
- 38 W. J. Duncanson, L. R. Arriaga, W. L. Ung, J. A. Kopechek, T. M. Porter and D. A. Weitz, *Langmuir*, 2014, **30**, 13765–13770.
- 39 K. Stark, J. P. Hitchcock, A. Fiaz, A. L. White, E. A. Baxter, S. Biggs, J. R. McLaughlan, S. Freear and O. J. Cayre, *ACS Appl. Mater. Interfaces*, 2019, **11**, 12272–12282.
- 40 D. Kagan, M. J. Benchimol, J. C. Claussen, E. Chuluun-Erdene, S. Esener and J. Wang, *Angew. Chem., Int. Ed.*, 2012, **51**, 7519–7522.
- 41 S.-T. Kang and C.-K. Yeh, *Langmuir*, 2011, **27**, 13183–13188.
- 42 N. Y. Rapoport, A. M. Kennedy, J. E. Shea, C. L. Scaife and K.-H. Nam, *J. Control. Release*, 2009, **138**, 268–276.
- 43 M. Javadi, W. G. Pitt, C. M. Tracy, J. R. Barrow, B. M. Willardson, J. M. Hartley and N. H. Tsosie, *J. Control. Release*, 2013, **167**, 92–100.
- 44 A. Schroeder, R. Honen, K. Turjeman, A. Gabizon, J. Kost and Y. Barenholz, *J. Control. Release*, 2009, **137**, 63–68.
- 45 J. L. Coffer, M. A. Whitehead, D. K. Nagesha, P. Mukherjee, G. Akkaraju, M. Totolici, R. S. Saffie and L. T. Canham, *Phys. Status Solidi A*, 2005, **202**, 1451–1455.
- 46 S. H. C. Anderson, H. Elliott, D. J. Wallis, L. T. Canham and J. J. Powell, *Phys. Status Solidi A*, 2003, **197**, 331–335.



- 47 A. P. Mann, T. Tanaka, A. Somasunderam, X. Liu, D. G. Gorenstein and M. Ferrari, *Adv. Mater.*, 2011, **23**, H278–H282.
- 48 G. Osei-Prempeh, H.-J. Lehmler, S. E. Rankin and B. L. Knutson, *Ind. Eng. Chem. Res.*, 2011, **50**, 5510–5522.
- 49 J. Salonen, A. M. Kaukonen, J. Hirvonen and V.-P. Lehto, *J. Pharm. Sci.*, 2008, **97**, 632–653.
- 50 B. Godin, C. Chiappini, S. Srinivasan, J. F. Alexander, K. Yokoi, M. Ferrari, P. Decuzzi and X. Liu, *Adv. Funct. Mater.*, 2012, **22**, 4225–4235.
- 51 C. Argyo, V. Weiss, C. Bräuchle and T. Bein, *Chem. Mater.*, 2014, **26**, 435–451.
- 52 J. Zhu, Y. Niu, Y. Li, Y. Gong, H. Shi, Q. Huo, Y. Liu and Q. Xu, *J. Mater. Chem. B*, 2017, **5**, 1339–1352.
- 53 N. Reznik, O. Shpak, E. C. Gelderblom, R. Williams, N. de Jong, M. Versluis and P. N. Burns, *Ultrasonics*, 2013, **53**, 1368–1376.
- 54 R. Singh, G. A. Hussein and W. G. Pitt, *Ultrason. Sonochem.*, 2012, **19**, 1120–1125.
- 55 O. D. Kripfgans, J. B. Fowlkes, D. L. Miller, O. P. Eldevik and P. L. Carson, *Ultrasound Med. Biol.*, 2000, **26**, 1177–1189.
- 56 I. Gorelikov, A. L. Martin, M. Seo and N. Matsuura, *Langmuir*, 2011, **27**, 15024–15033.
- 57 O. Couture, P. D. Bevan, E. Cherin, K. Cheung, P. N. Burns and F. S. Foster, *Ultrasound Med. Biol.*, 2006, **32**, 73–82.
- 58 M. M. Kaneda, S. Caruthers, G. M. Lanza and S. A. Wickline, *Ann. Biomed. Eng.*, 2009, **37**, 1922–1933.
- 59 J. W. Mo, J. J. Sha, D. K. Li, Z. G. Li and Y. F. Chen, *Nanoscale*, 2019, **11**, 8408–8415.
- 60 L. Wu, Z. C. Dong, F. Y. Li and Y. L. Song, *Langmuir*, 2018, **34**, 639–645.
- 61 N. Y. Rapoport, A. L. Efros, D. A. Christensen, A. M. Kennedy and K. H. Nam, *Bubble Sci. Eng. Technol.*, 2009, **1**, 31–39.
- 62 E. Tasciotti, X. Liu, R. Bhavane, K. Plant, A. D. Leonard, B. K. Price, M. M.-C. Cheng, P. Decuzzi, J. M. Tour, F. Robertson and M. Ferrari, *Nat. Nanotechnol.*, 2008, **3**, 151–157.
- 63 C. Hirayama, H. Ihara, S. Nagaoka and T. Wada, *Polym. J.*, 1994, **26**, 499–503.
- 64 R. J. Baker, P. E. Colavita, D. M. Murphy, J. A. Platts and J. D. Wallis, *J. Phys. Chem. A*, 2012, **116**, 1435–1444.
- 65 S. Suzuki, A. Nakajima, N. Yoshida, M. Sakai, A. Hashimoto, Y. Kameshima and K. Okada, *Langmuir*, 2007, **23**, 8674–8677.
- 66 S. Li, P. Cao, R. Colorado, X. Yan, I. Wenzl, O. E. Shmakova, M. Graupe, T. R. Lee and S. S. Perry, *Langmuir*, 2005, **21**, 933–936.
- 67 J.-D. Brassard, D. K. Sarkar and J. Perron, *ACS Appl. Mater. Interfaces*, 2011, **3**, 3583–3588.
- 68 D. B. Mawhinney, J. A. Glass and J. T. Yates, *J. Phys. Chem. B*, 1997, **101**, 1202–1206.
- 69 C. R. Hill, J. C. Bamber and G. R. ter Haar, *Detection and Measurement of Acoustic Fields, Physical Principles of Medical Ultrasonics*, John Wiley & Sons, 2004, pp. 69–91.

