## Materials Horizons



## CORRECTION

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## Correction: Epoxy resin with exchangeable disulfide crosslinks to obtain reprocessable, repairable and recyclable fiber-reinforced thermoset composites

Alaitz Ruiz de Luzuriaga, PROBERTO Martin, Nerea Markaide, Alaitz Rekondo, FROMENTA REMONDO PROBERTO PROBERTO PROBERTO PROBER PROBERTO PRO

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Correction for 'Epoxy resin with exchangeable disulfide crosslinks to obtain reprocessable, repairable and recyclable fiber-reinforced thermoset composites' by Alaitz Ruiz de Luzuriaga *et al.*, *Mater. Horiz.*, 2016, **3**, 241–247, DOI: 10.1039/C6MH00029K.

The authors have detected an error in the calculation of the activation energy and vitrimer temperature. In Fig. 1b,  $\log \tau$  should be plotted on the y axis instead of  $\ln \tau$ . Due to this error, the calculated activation energy is incorrect and the energy should be multiplied by 2.3 ( $1/\log e$ ). The correct activation energy is 126.5 KJ mol<sup>-1</sup> and the correct calculated vitrimer temperature is 75 °C. The authors apologise for this error and state that this error does not affect any of the scientific findings and interpretations.

A detailed list of changes is provided as follows:

- The correct version of Fig. 1b is shown below
- On page 243, "From this equation, an activation energy  $(E_a)$  of 55 kJ mol<sup>-1</sup> was calculated for the dynamic epoxy network 5. Such activation energy is lower than other reported systems..." should be changed to "From this equation, an activation energy  $(E_a)$  of 126 kJ mol<sup>-1</sup> was calculated for the dynamic epoxy network 5. Such activation energy is similar to other reported systems..."
- On page 243, "For our dynamic epoxy system 5, the hypothetical  $T_{\rm v}$  value obtained was -13 °C, which is well below its  $T_{\rm g}$  (127 °C from DSC)." should be changed to "For our dynamic epoxy system 5, the hypothetical  $T_{\rm v}$  value obtained was 75 °C, which is well below its  $T_{\rm g}$  (127 °C from DSC)."

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

Correction

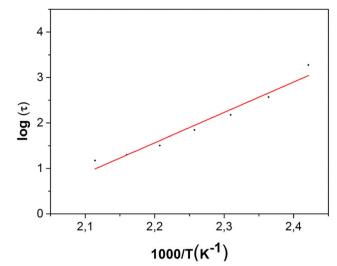


Fig. 1 (b) Fitting of the relaxation times to the Arrhenius' equation (R-square = 0.946).