Chem Soc Rev

CORRECTION

View Article Online
View Journal | View Issue

Cite this: *Chem. Soc. Rev.*, 2020, **49**, 643

DOI: 10.1039/c9cs90105a

rsc.li/chem-soc-rev

Correction: Voltage issue of aqueous rechargeable metal-ion batteries

Zhuoxin Liu,[ab] Yan Huang,[c] Yang Huang,[a] Qi Yang,[b] Xinliang Li,[b] Zhaodong Huang[b] and Chunyi Zhi*[bd]

Correction for 'Voltage issue of aqueous rechargeable metal-ion batteries' by Zhuoxin Liu *et al.*, *Chem. Soc. Rev.*, 2020, DOI: 10.1039/c9cs00131j.

The authors regret that there were errors in Fig. 6, 10, 15 and 19 in the original article. The labels for H_2 evolution and O_2 evolution were misplaced in the original figures. The corrected versions of these figures are as shown below.

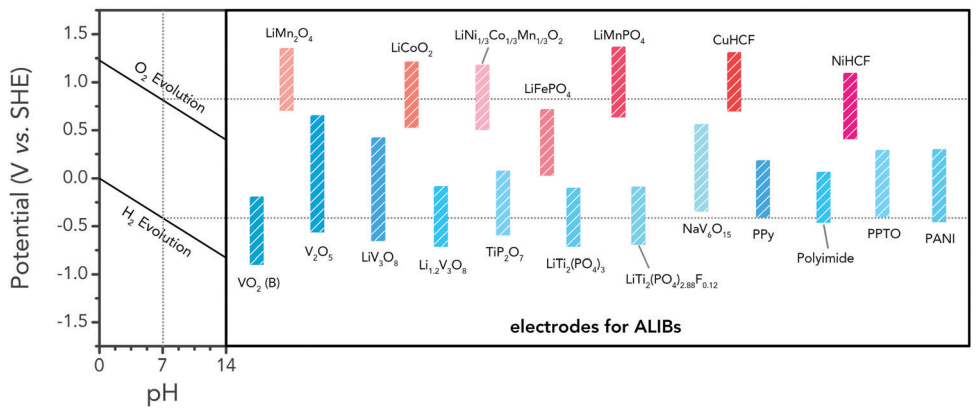

Fig. 6 Comparison of redox potentials of representative electrode materials for ALIBs. Red-colour columns and blue-colour columns represent cathodes and anodes, respectively.

[a] *College of Materials Science and Engineering, Shenzhen University, Shenzhen 518060, China*
[b] *Department of Materials Science and Engineering, City University of Hong Kong, 83 Tat Chee Avenue, Hong Kong 999077, China. E-mail: cy.zhi@cityu.edu.hk*
[c] *State Key Laboratory of Advanced Welding and Joining, Harbin Institute of Technology (Shenzhen), Shenzhen 518055, China*
[d] *Shenzhen Research Institute, City University of Hong Kong, Shenzhen 518057, China*

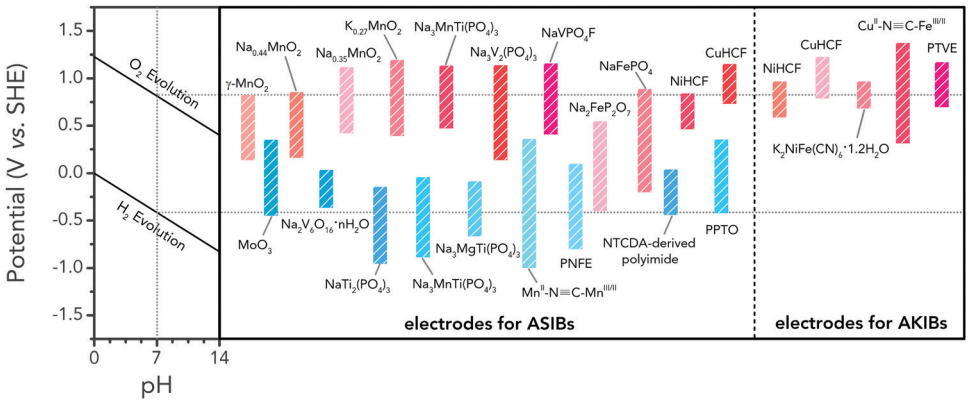

Fig. 10 Comparison of redox potentials of representative electrode materials for ASIBs and AKIBs. Red-colour columns and blue-colour columns represent cathodes and anodes, respectively.

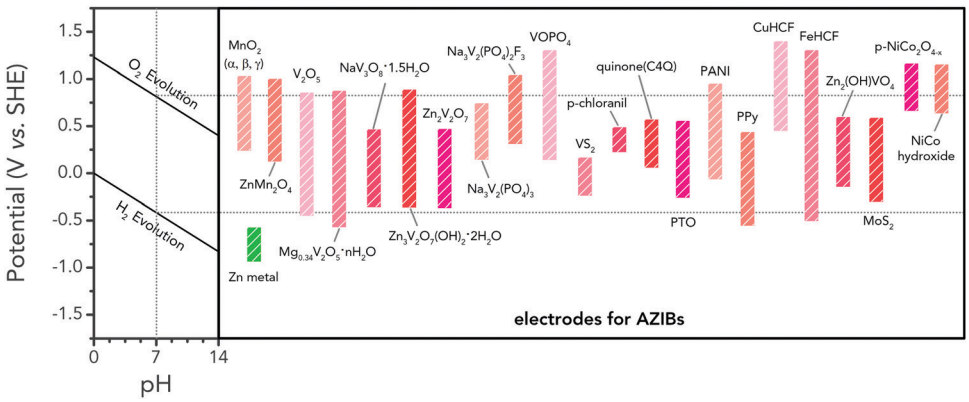

Fig. 15 Comparison of redox potentials of representative electrode materials for AZIBs. Red-colour columns and green-colour column represent cathodes and zinc metal anode, respectively. (Some conversion-type NiCo compounds are also included.)

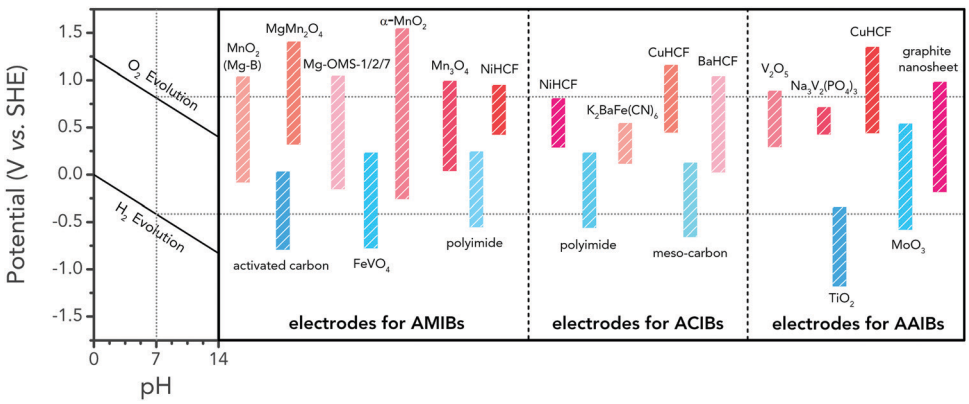

Fig. 19 Comparison of redox potentials of representative electrode materials for AMIBs, ACIBs and AAIBs. Red-colour columns and blue-colour columns represent cathodes and anodes, respectively.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

