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Semi-transparent quaternary oxynitride photoanodes on GaN underlayers†

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Conformal atomic layer deposition (ALD) technique is employed to make semi-transparent TaO_xN_y , providing the possibility to build semi-transparent oxy(nitride) heterojunction photoanodes on conductive substrates. A generalized approach was developed to manufacture semi-transparent quaternary metal oxynitrides on conductive substrates beyond semi-transparent binary Ta_3N_5 photoanodes aiming for wireless tandem photoelectrochemical (PEC) cells.

A sustainable solution for storing the solar energy in principle can be achieved by PEC technology, which is based on splitting of earth-abundant water into the renewable energy carrier hydrogen and oxygen.¹ Connecting the photoanode and the photocathode in a stacked configuration enables the devices to maximize sunlight harvesting by using narrow band gap semiconductors in tandem.² Thus, the photovoltage generated in each photoabsorber can be summed to drive water splitting at high efficiency. In the tandem geometry, the photoanode as first photoelectrode absorbs photons of higher energy while the photocathode absorbs the transmitted photons to drive water reduction.³ Fig. S1 (ESI†) depicts a wireless tandem PEC cell configuration, in which the solar photons not absorbed by the semi-transparent photoelectrode at the front side will pass through the intermediate transparent conductive substrate (TCS).

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Subsequently, the remaining photons can be harvested by the second photoelectrode with a smaller band gap energy (E_g) .

Additionally, the other advantages of the wireless PEC tandem cell are the (i) flexibility in materials selection, (ii) low resistance losses and (iii) high proton conductivity,^{5,6} which target for high solar-to-hydrogen (STH) conversion efficiency.⁷ Furthermore, the wireless configuration is more suitable for industrial manufacturing in comparison to the wired tandem configuration in which the photoabsorbers are connected via an external circuit.⁸ With respect to the TCS materials required for a PEC tandem cell, the most commonly used substrates are transparent conductive oxides (TCOs) such as fluorine/indium doped tin oxide (FTO/ITO) and aluminum doped zinc oxide (AZO) .⁹ These TCS materials, however, cannot be applied to materials that require synthetic conditions including ammonia at elevated temperatures (>873) K). As such, common TCS materials are limited in their implantation for nitrogen-containing photoelectrode materials.^{10,11} COMMUNICATION
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For commercialization of PEC technology, it is widely accepted that the STH efficiency needs to be greater than 10%.¹² Among the often studied metal oxide-based photoanodes such as $TiO₂, WO₃,$ $ZnFe₂O₄$, BiVO₄, the most representative one is the latter,¹³ exhibiting the highest STH conversion efficiency of 3.7% in a bias-free tandem cell for PEC water splitting.⁷ To achieve higher STH efficiency, the front photoanode in wireless tandem PEC cells has to be semi-transparent. Thus, it must absorb photons with energy larger than its band gap and transmit the remaining photons to the photocathode behind it, avoiding scattering or parasitic absorption of the transmitted light.

Metal nitrides, oxynitrides and N-doped oxides exhibit narrower band gaps than the best-performing oxidic photoanode candidates. An example is Ta_3N_5 with a bandgap of 2.1 eV and that can be obtained by ammonolysis of Ta₂O₅. This synthetic approach makes it challenging to produce semi-transparent metal (oxy) nitrides on FTO, because the transparent SnO_x becomes metallic Sn under the nitridation conditions.14 It should be noted that fabrication of semi-transparent photoanodes by means of electrophoretic deposition of powder samples displays significant worse mechanical stability under operating conditions. Recently, the

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groups of Hamann^{10,14} and Domen^{15–17} have demonstrated alternative methods to fabricate semi-transparent Ta_3N_5 photoanodes. Although semitransparent films of Ta₃N₅ can reach 6.3 mA cm⁻² at 1.23 V vs. reversible hydrogen electrode $(RHE),¹⁷$ the material suffers from a high onset potential of 0.3–0.8 V vs. RHE, 14,18,19 which then must be generated by the tandem photocathode for overall water splitting to occur.²⁰

The quaternary metal oxynitrides $ATa(0,N)$ ₂ display a photocurrent onset at smaller applied potentials and, for several compounds, can potentially exhibit a higher theoretical STH efficiency than the binary Ta_3N_5 .^{21,22} This is due to the narrower band gaps of quaternary metal oxynitrides, depending on the chemical composition and being tunable within the range of 1.7 and 2.4 eV .²²⁻²⁴ This renders semi-transparent quaternary metal oxynitrides generally as very promising photoanode candidates for PEC tandem cells, which could outperform Ta_3N_5 -based devices. So far, semi-transparent quaternary metal oxynitride films have been mainly fabricated by means of reactive radiofrequency (RF) magnetron sputtering²⁵⁻²⁸ or pulsed laser deposition (PLD)^{29,30} on MgO or Nb:SrTiO₃ substrate. In case of the RF approach, the oxynitride sputtering targets have to be made first with multi-step processes, and harsh conditions (usually up to 1073 K for film deposition) are required, too. 27 The development of photoelectrode materials with more narrow band gaps than Ta_3N_5 and hematite is mainly motivated by the higher theoretical light absorption capacity.¹¹ Recently, Domen et al. have demonstrated a core–shell heterojunction photoanode of Ta₃N₅-nanorods/BaTaO₂N, generating a stable photocurrent owing to efficient generation and extraction of charge carriers.³¹ For the reproducible fabrication of thin films exhibiting controllable thickness on the atomic scale and conformal features, Atomic Layer Deposition (ALD) is a promising technique for the deposition of oxidic precursor layers for subsequent nitridation.³² Communication

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In this communication, we demonstrate the synthesis of semi-transparent quaternary photoanodes on the examples of $SrTaO₂N$ and $LaTiO₂N$. First, ALD was explored to obtain semitransparent TaO_xN_y thin films on conductive n-type GaN substrate. GaN was selected as the substrate, because it can maintain chemical stability at high temperature in $NH₃$ without degradation¹⁵ and under ambient conditions.³³ Low pressure metalorganic vapor phase epitaxy reactor (MOVPE) was employed to grow unintentionallydoped GaN epitaxial film on double-sided polished Al_2O_3 (sapphire) substrates (see ESI†).

The ALD cycle procedure for Ta_2O_5 deposition is illustrated in Fig. 1 (see ESI†). Nitridation at 1148 K for 2 h converted the Ta_2O_5 to TaO_xN_y , *i.e.* N-doped Ta_2O_5 , which was confirmed by X-ray photoelectron spectroscopy (XPS) (Fig. S2, see ESI† for experimental details). It has been reported that ammonolysis of Ta $_{2}\mathrm{O}_{5}$ at higher temperatures may yield the binary Ta $_{3}\mathrm{N}_{5}$. 34 The transmittance of the prepared thin film was evaluated by means of Ultraviolet-visible spectroscopy (UV-vis) as a function of the wavelength (Fig. 2a). After depositing Ta_2O_5 onto Al_2O_3/GaN substrate, the transmittance only decreased slightly because of thin Ta₂O₅ layer and its wide band gap. After nitridation, the transmittance of the resulting $Al_2O_3/GaN/TaO_xN_y$ thin film decreased to around 70% at wavelengths above 600 nm.

Fig. 1 Scheme of ALD procedure for Ta_2O_5 on substrate.

Fig. 2 (a) UV-vis transmittance spectra. (b) LSV of corresponding $Al_2O_3/$ $\text{GaN/TaO}_{x}\text{N}_{v}$ photoanode measured in 0.1 M NaOH under chopped AM 1.5G illumination (1 sun, 100 mW cm $^{-2}$).

The light-absorbing thin film TaO_xN_v on the Al_2O_3/GaN substrate was composed of grains with approximately size of 100 nm due to sintering at high temperature substrate as observed by scanning electron microscopy (SEM) (Fig. S3 in the ESI†). The grains are distributed uniformly throughout the substrate as observed at the SEM micrograph (Fig. S4, ESI†). The $Al_2O_3/GaN/TaO_xN_y$ photoanode, which was prepared by MOVPE and ALD, generated a photocurrent of 0.21 mA cm-2 at 1.23 V vs. RHE in 0.1 M NaOH under AM 1.5G illumination (1 sun) as shown in the linear sweep voltammetry curve (LSV, Fig. 2b). The photocurrent is lower than other reported semitransparent Ta_3N_5 film because of smaller thickness and smaller amount of N incorporation. $15-17$ The photocurrent maintained stable during chronoamperometry (CA) (Fig. S5, ESI†).

Given the suitability of the deposited GaN layer for electron transport of nitridic light absorbers, we elucidated the possibility to fabricate quaternary metal oxynitrides on the transparent substrate Al_2O_3/GaN . Fig. 3 shows the preparative procedure of semitransparent quaternary oxynitride thin films (see ESI† for details).

Fig. 3 Schematic illustration of the fabrication procedure for semitransparent quaternary metal oxynitride thin film photoanode on transparent conductive GaN/Al₂O₃ substrate.

A spin-coating step followed to deposit a sol–gel film on the substrate from a precursor solution containing La and Ti with molar ratio of 1 : 1. Afterwards, the substrate was annealed at 773 K for 1 h in air to remove organics from the sol–gel to form the La and Ti oxide precursor. The oxide was converted to $LaTiO₂N$ by nitridation under flowing NH₃ and H₂ at 1148 K for 2 h at a ramping rate of 10 K min^{-1} .

X-ray diffraction (XRD) patterns presented in Fig. 4a reveal that the grown LaTiO₂N thin film on Al_2O_3/GaN substrate is of single-phase purity. The LaTiO₂N patterns fully match the simulated powder patterns (ICSD 239555). 35 The UV-vis spectrum (Fig. 4b) also shows an absorption edge at approximately 600 nm, consistent with the LaTiO₂N band gap of 2.0–2.1 eV.³⁶ The transmittance decreased from 80% on the Al_2O_3/GaN substrate to roughly 60% on the $Al_2O_3/GaN/LaTiO_2N$ substrate for wavelengths above 600 nm. The high transparency (see inset of Fig. 4b) of LaTiO₂N is above the threshold for constructing a tandem PEC cell by integration with a photocathode of smaller band gap. The damping-like feature of the spectrum is due to the interference fringes of multiple layers, i.e. Al_2O_3 , GaN and LaTiO₂N.¹⁵ The SEM micrograph in Fig. 4c shows the morphology of the prepared $LATiO₂N$ film in the form of a microstructure comprised of nanoparticulate agglomerations that were several hundred nanometers in diameter. A similar morphology is usually obtained for other ceramic thin films prepared by chemical solution deposition (CSD). 37 The SEM at low magnification (see Fig. S6, ESI†) demonstrates that the film is uniform and crack-free, the thickness is around 230 nm from the cross-sectional SEM (Fig. S7, ESI†). We also synthesized SrTaO₂N in order to verify the generality of the CSD-based synthesis for semi-transparent quaternary oxynitride photoanodes. Its XRD pattern (Fig. 4a) ChernComm

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Fig. 4 (a) XRD patterns of LaTiO₂N and SrTaO₂N film on Al₂O₃/GaN substrate and simulated patterns (ICSD 239555 and 95373). (b) UV-vis transmittance spectra of LaTiO₂N and SrTaO₂N film on Al_2O_3/GaN substrate and the substrate before oxynitride formation. The inset depicts a photographic image of the films. (c) SEM micrograph of LaTiO₂N film on Al_2O_3/GaN substrate. (d) LSV curves of LaTiO₂N and SrTaO₂N films measured in 0.1 M NaOH under chopped AM 1.5G simulated solar light $(100 \, \text{mW cm}^{-2})$.

confirms the formation of single phase $SrraO₂N$ crystallizing with the tetragonally-distorted perovskite structure in space group $I4/mcm$ (ICSD 95373).³⁸ The prepared SrTaO₂N thin film also exhibited high transparency (see inset of Fig. 4b). UV-vis measurements showed transmittance of approximately 70% for wavelengths above 550 nm, which corresponds to the absorption edge of SrTaO₂N.³⁹ The surface of SrTaO₂N (SEM, Fig. S8, ESI[†]) is smoother than for LaTiO₂N. However, the thin film thickness is difficult to determine due to strong charging during SEM analysis. These results prove that the CSD-based process can indeed be used for other semi-transparent quaternary oxynitride thin films on transparent Al_2O_3/GaN substrate.

The critical feature of such prepared electrodes is the transport of electrons through the GaN pathway over the external circuit to the counter-electrode. The PEC water oxidation performances of the prepared LaTiO₂N and SrTaO₂N films were assessed in 0.1 M NaOH electrolyte (pH 13) under simulated sunlight (1 sun, AM 1.5G). Fig. 4d displays the LSV curves measured at a scan rate of 10 mV s^{-1} under chopped illumination. The LaTiO₂N thin film developed a net photocurrent of approximately 20 μ A cm⁻² at 1.23 V vs. RHE. The photocurrent is comparable with the LaTiO₂N film deposited on MgO/TiN and $Nb:STIO₃$ substrates via PLD and RF sputtering reported by Lippert²⁹ and Domen,²⁶ respectively. The dark-current contribution stems from photocorrosion of the oxynitride surface, which is a common phenomenon for unprotected metal oxynitrides and can be suppressed with protective layers.^{40,41} The pristine semitransparent SrTaO₂N photoanode, having a slightly larger band gap than LaTiO₂N, generated a photocurrent of app. 7 μ A cm $^{-2}$ at 1.23 V vs. RHE. Both LaTiO₂N and SrTaO₂N films show photoresponse at app. 0 V vs. RHE, indicating relative negative onset potentials (see enlarged view in Fig. S9, ESI†).

In conclusion, we have successfully fabricated semi-transparent quaternary oxynitride LaTiO₂N and SrTaO₂N thin films on transparent conductive $\text{Al}_2\text{O}_3/\text{GaN}$ substrate. PEC measurements confirmed their potential for use as the top photoanode in a PEC water-splitting cell, and the band gaps suggest that they may have higher theoretical STH than Ta₃N₅-based photoanodes. The CSD-based process can be adopted as a general method towards the manufacture of semi-transparent quaternary oxynitride thin films on large scale. Additionally, the conductive Al_2O_3/GaN substrate have been coated with TaO_xN_y by means of ALD to yield semi-transparent thin films. Nonetheless, the photocurrent of the quaternary oxynitride thin film is far lower in comparison to their maximum theoretical values. This requires future synthetic strategies for nanostructuring below 100 nm of the light absorbing metal oxynitride layer and its subsequent semi-transparent catalysts.

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Conflicts of interest

There are no conflicts to declare.

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