Conformationally rigid pyrazoloquinazoline \( \alpha \)-amino acids: one- and two-photon induced fluorescence†

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The synthesis and photophysical properties of a new class of \( \alpha \)-amino acid bearing a rigid pyrazoloquinazoline chromophore are described. Conformational constraint of the amino acid side-chains resulted in high emission quantum yields, while the demonstration of two-photon-induced fluorescence via near-IR excitation signifies their potential for sensitive bioimaging applications.

Fluorescence spectroscopy is a powerful technique for studying biological structure and function and for visualizing molecular interactions and intracellular processes. This is due to its high sensitivity and information content, its multiplexing capability, the ability to probe processes over a wide time range, and the availability of an array of versatile fluorescent probes. The investigation of cellular functions such as enzyme mechanisms and protein–protein interactions have often utilized extrinsic fluorescent labels such as green fluorescent protein (GFP) and its derivatives or intrinsic labels such as the proteinogenic fluorescent \( \alpha \)-amino acids, tyrosine or tryptophan. However, the attachment of a fluorescent protein can alter the stability and functionality of the fusion partner, while tyrosine and tryptophan have poor optical properties and the occurrence of multiple residues in different environments can complicate their analysis. In recent years these limitations have been overcome by the development of unnatural fluorescent \( \alpha \)-amino acids. The photophysical properties of such amino acids can be tuned for a particular application and they can be incorporated into proteins and peptides by solid phase peptide synthesis (SPPS) or by unnatural amino acid mutagenesis. Of particular interest are amino acids with side-chain chromophores that can be easily incorporated into peptides and proteins with minimal disruption to structure and function. While some of these fluorescent \( \alpha \)-amino acids have found application in biological and medicinal imaging, an inherent issue of small chromophores is that ultraviolet (UV) light is required for excitation. Prolonged exposure to UV light causes cell and tissue damage through photo-bleaching. In recent years, this limitation has been overcome by the use of two-photon fluorescence microscopy, which permits excitation of the chromophore at longer wavelength, allowing long-term imaging of biological specimens, deeper tissue penetration, with minimization of photo-bleaching and photo-toxicity. While this technique has been utilized with a wide range of fluorophores and applied to the imaging of various biological processes and diseases, there are relatively few reports with unnatural \( \alpha \)-amino acids. Examples include the seminal work of Mély and co-workers who used two-photon excitation of peptides containing 3-hydroxyflavone derived \( \alpha \)-amino acids (e.g. 1, Fig. 1) for monitoring interactions with cell membranes and intracellular RNA, while the Vendrell group used two-photon excitation of a cyclic peptide bearing a tryptophan-BODIPY conjugate (2) for the visualization of fungal infections.

We previously reported the synthesis of 5-arylpyrazole-derived \( \alpha \)-amino acids such as 3 (Fig. 1). These compounds possessed

\[ \text{Fig. 1} \quad \text{Selected unnatural fluorescent } \alpha \text{-amino acids.} \]
some interesting photophysical properties however, the rotational flexibility between the aryl and pyrazole rings resulted in low quantum yields and brightness. As conformationally rigid, molecular structures permit more efficient π-overlap and subsequently brighter chromophores, we proposed that conformational restriction of these 5-arylpyrazole motifs would lead to α-amino acids with enhanced photophysical properties. Here we describe a new class of α-amino acid (4), bearing planar pyrazoloquinazoline chromophores. We also report the photophysical properties of these compounds, demonstrating how conformational restraint leads to fluorophores with high quantum yields and spectroscopic properties that are sensitive to environment polarity. We also show the application of these α-amino acids to two-photon excitation, for potential use in imaging UV sensitive, biological systems.

A synthetic route for the preparation of pyrazoloquinazoline-containing α-amino acids 4 was developed from L-aspartic acid. Initially, L-aspartic acid was converted to phosphononorvaline derivative 5 in five steps and 84% overall yield using previously described methods. A range of 2-nitrobenzaldehydes was reacted with phosphononorvaline 5 via a Horner–Wadsworth–Emmons reaction, which gave the E-isomer of enones 6a–e in 42–82% yields (Scheme 1). The enone moiety was used to construct the pyrazole ring by reaction of 6a–e with hydrazine via a one-pot condensation/aza-Michael process. Low temperature oxidation of the resulting 2-pyrazolines with DDQ gave 5-(2-nitroaryl)pyrazoles 7a–e. Carbyonlation was then achieved by reaction with triphosgene under basic conditions. This gave pyrazoloquinazolines 8a–e in 30–75% yields over the two steps. Deprotection to the parent amino acids was either conducted in one step under acidic conditions (4b, 37%; 4c, 40%) or more efficiently by a two-step strategy involving ester hydrolysis, followed by removal of the Boc-protecting group under milder acidic conditions (4a, 80%; 4d, 82%; 4e, 78%).

The UV/Vis absorption and emission spectra of the amino acids were recorded in methanol at a concentration of 0.5 × 10⁻⁵ M (Fig. 2 and Table 1). As expected for chromophores that are planar in the ground state, the amino acids, particularly 4a–d, showed structured absorption spectra with well-resolved vibrational bands (Fig. 2a). The emission spectra of amino acids 4a–e, following one-photon excitation at the longest wavelength absorption band, showed strong fluorescence ranging from 342–414 nm, with a vibrational progression which mirrored that of the absorption spectra (Fig. 2b). Amino acid 4e bearing the electron rich dimethylamino substituent gave a red-shifted, structureless emission maximum at 414 nm, which is indicative of a combination of locally excited and internal charge transfer states. More importantly, the quantum yields of 4a–e were characteristic of conformational restriction and showed a ten-fold increase in comparison to the non-rigid pyrazole-derived amino acid 3. As a result, these α-amino acids are substantially brighter than the corresponding non-rigid systems.

As amino acid 4e showed the most interesting photophysical properties, this compound was further investigated. The fluorescence decay was measured at an emission wavelength of 430 nm in methanol using time-correlated single-photon counting, following excitation at 390 nm (see ESI†). Three decay components were fitted with lifetimes of 0.79, 2.17 and 7.90 ns, and A-factors of 0.60, 0.36 and 0.04, respectively. This suggests the molecule exists

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**Scheme 1** Synthesis of pyrazoloquinazoline α-amino acid 4a–e.

**Fig. 2** (a) Absorption spectra of amino acids 4a–4e recorded at 0.5 × 10⁻⁵ M in MeOH. (b) Emission spectra of 4a–4e at 0.5 × 10⁻⁵ M in MeOH.
predominantly in two states, as has been observed previously for dimethylamino-substituted fluorophores.\textsuperscript{21} The amplitude-weighted average lifetime is 1.57 ns, which is comparable to the weighted average lifetime of 1.57 ns, which is comparable to

Table 1 Photophysical data for 5′-(naphthyl-2′-yl)-1′-phenylpyrazole amino acid 3 and pyrazoloquinazoline-derived α-amino acids 4a–e\textsuperscript{a}

<table>
<thead>
<tr>
<th>Amino acid</th>
<th>λ_{abs} (nm)</th>
<th>ε (cm\textsuperscript{-1} M\textsuperscript{-1})</th>
<th>λ_{em} (nm)</th>
<th>Φ\textsubscript{p}\textsuperscript{b}</th>
<th>Brightness\textsuperscript{c} (cm\textsuperscript{-1} M\textsuperscript{-1})</th>
</tr>
</thead>
<tbody>
<tr>
<td>3</td>
<td>342</td>
<td>23700</td>
<td>356</td>
<td>0.04</td>
<td>950</td>
</tr>
<tr>
<td>4a</td>
<td>321</td>
<td>8300</td>
<td>342</td>
<td>0.42</td>
<td>3500</td>
</tr>
<tr>
<td>4b</td>
<td>350 (279)</td>
<td>7700 (24 800)</td>
<td>372 (373)</td>
<td>0.47 (0.27)</td>
<td>3600 (6700)</td>
</tr>
<tr>
<td>4c</td>
<td>328</td>
<td>10 000</td>
<td>353</td>
<td>0.47</td>
<td>4700</td>
</tr>
<tr>
<td>4d</td>
<td>337</td>
<td>11 200</td>
<td>363</td>
<td>0.56</td>
<td>6300</td>
</tr>
<tr>
<td>4e</td>
<td>349</td>
<td>12 200</td>
<td>414</td>
<td>0.48</td>
<td>5900</td>
</tr>
</tbody>
</table>

\textsuperscript{a} All spectra were recorded at 0.5 × 10\textsuperscript{-5} M in MeOH. \textsuperscript{b} Quantum yields (Φ\textsubscript{p}) were determined in MeOH using anthracene and 1-tryptophan as standards. \textsuperscript{c} Brightness was calculated as the product of the QY and the molar absorptivity at the wavelength maximum specified.

To assess the environmental sensitivity of amino acid 4e, its photophysical properties were examined in a range of solvents. The absorption maxima (342–348 nm) were found to be independent of polarity (see ESI†), indicating negligible intramolecular interaction between the electron-rich dimethylamino and electron-deficient N-acylpyrazole moieties in the ground state. In contrast, the emission spectra were found to be highly sensitive to the solvent used, with increasing polarity leading to concomitantly broadened, structureless emission spectra at longer wavelengths (Fig. 3a).\textsuperscript{23,24}

Fig. 3 (a) Solvatochromic study of 4e. Spectra were recorded using a concentration of 0.5 × 10\textsuperscript{-5} M. (b) Emission of 4e after two-photon absorption (red) and one-photon (blue) absorption in MeOH. For the two-photon measurement, a solution of 200 μM in MeOH and an excitation centred around 700 nm with FWHM of 10 nm was used.

stabilization of both the locally excited and intramolecular charge transfer states.

Application of amino acid 4e for two-photon spectroscopy was next investigated. Two- and three-photon excitation has recently been shown to be a promising approach for the ultra-sensitive analysis of fluorescent nucleobases, suggesting a similar approach may be applicable to α-amino acids.\textsuperscript{25} Two-photon absorption was confirmed with a quadratic dependence of the emitted intensity on the excitation intensity (see ESI†). Furthermore, the molecule has no one-photon absorption in the near IR, precluding linear excitation. The emission spectra of amino acid 4e in methanol after one- and two-photon excitation have a similar profile (Fig. 3b). The two-photon cross section was measured to be 0.38 ± 0.06 GM using rhodamine B as a reference compound (see ESI†). This value is comparable to other molecules of similar size.\textsuperscript{25} Thus, 4e can undergo two-photon absorption using near-IR excitation, thereby avoiding the use of UV excitation and the issues of photobleaching.

The potential of these amino acids for incorporation into peptides via SPPS methodology was also investigated. A model pentapeptide (Val-Pro-Thr-Leu-Lys), based on the Bax-binding domain of Ku70, a multifunctional protein involved in DNA repair and cell-death regulation was chosen.\textsuperscript{26} This pentapeptide and similar analogues have been shown to have low cytotoxicity and are effective at penetrating living cells.\textsuperscript{26b} The pentapeptide was prepared in an automated, peptide synthesizer using TentaGel\textsuperscript{TM} S Rink Amide (RAM) resin as the polymer support\textsuperscript{27} and routine SPPS methodology (Scheme 2). Coupling of each Fmoc-protected amino acid was performed by HCTU\textsuperscript{28} activation, followed by piperidine-mediated N-deprotection, which gave the N-terminal unprotected pentapeptide. Boc-Protected pyrazoloquinazoline-derived amino acid 9 (see ESI† for preparation) was then coupled manually onto the polymer-supported pentapeptide. A TFA cleavage cocktail was used to remove the protecting groups and release the hexapeptide from the polymer support. Purification by reverse phase-HPLC gave hexapeptide 10 in 72% yield with >95% purity. Analysis of the photophysical properties of hexapeptide 10 showed that apart from a small hypsochromic shift in the emission spectra

Scheme 2 SPPS and photophysical data of hexapeptide 10.
Notes and references
17 The absorption and emission spectra for each amino acid are presented in the ESI.
19 It should be noted that while excitation of the absorption band at longest wavelength gave the most intense emission spectrum for most of the amino acids, this was not the case for 4b. Excitation of the peak at 279 nm gave the most intense fluorescence and highest brightness value (Table 1).
28 HCTU: O-(1-H-6-Chlorobenzotriazole-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate.