



Fluorescent SAM analogues for methyltransferase based DNA labeling†

Vince Goyvaerts,^{id}^a Sven Van Snick,^a Laurens D'Huys,^{id}^a Raffaele Vitale,^{id}^{ab} Milena Helmer Lauer,^{id}^a Su Wang,^{ac} Volker Leen,^a Wim Dehaen^a and Johan Hofkens^{id}^{*ad}

Cite this: *Chem. Commun.*, 2020, 56, 3317

Received 15th November 2019,
Accepted 22nd January 2020

DOI: 10.1039/c9cc08938a

rsc.li/chemcomm

In this work, the preparation of new S-adenosyl-L-methionine (SAM) analogues for sequence specific DNA labeling is evaluated. These non-natural analogues, comprising cysteine rather than the natural homolog, were obtained in near quantitative conversions from readily available starting materials without relying on using an excess amount of labor intensive molecules. The synthetic strategy was used to generate fluorescent cofactors, with colours spanning the whole visible spectrum, and their applicability in methyltransferase based optical mapping is shown.

Site-specific DNA labeling holds the potential of unlocking promising advances in DNA-based applications.¹ Through modifications of known positions on DNA, researchers are able to extract vital information about the genetic material, making site-specific DNA labeling an indispensable tool for species identification or medical diagnostics.^{2–4} In the last two decades, several approaches to such site-specific labeling have been developed, mainly relying on hairpin polyamides,⁵ triplex-helix-forming oligodeoxynucleotides⁶ or enzyme-directed DNA modifications.^{7–9}

Perhaps one of the most exciting examples of site-specific labeling is DNA methyltransferases (MTases) based labeling. In nature, these enzymes play a key role in the DNA methylation process by catalyzing the transfer of a methyl group from an S-adenosyl-L-methionine (SAM) cofactor onto adenine (N⁶-position) or cytosine (N⁴- or C⁵-position).¹⁰ Generally, the catalytic transfer will only occur after the enzyme successfully binds its recognition sequence, ensuring site-specific labeling. While a methyl group is

an inert moiety, pioneering work by Weinhold and coworkers shows that MTase enzymes tolerate non-natural cofactors to effectively transfer functionalized labels onto DNA, and such strategies have been developed to introduce (fluorescent) labels which allow for localization,¹¹ capture,¹² photolabile caging,^{13,14} photocrosslinking,¹⁵ selective scission,¹⁶ ... and is a fundamental technology in the field of optical mapping.

In variants of this technology, replacing the methionine side chain of SAM with an aziridine ring resulted in functional cofactors with a pending aziridine moiety.¹⁷ MTase enzymes are tricked into catalyzing the nucleophilic ring opening, in line with nucleophilic attack by the nucleobase, thus covalently coupling the whole cofactor onto DNA. By introducing a functional group on the nucleobase, these molecules are able to label DNA. This research was later extended by Rajska and co-workers,¹⁸ who used reactive nitrogen mustards for *in situ* aziridine generation. However, during the enzymatic labeling, both research groups noticed a strong affinity of the labeled DNA for the enzymatic binding pocket, which inhibited further turnovers and thus forcing the use of stoichiometric amounts of enzyme to fully label DNA.

This downside prompted Weinhold *et al.* to develop more efficient cofactors, and a second class called the doubly activated cofactors was evaluated.⁹ Here, a double or triple bond is introduced at the β -position to the sulfonium center as part of the functionalized label which is transferred by the MTase enzymes. As only the label is transferred, this class is less affected by enzyme inhibition. The unsaturated bond stabilizes the transition state in the transfer reaction, and hence, no enzymatic transfer was observed for saturated labels.

Commonly, doubly activated cofactors are used to transfer a chemically reactive group, *e.g.* amine,¹⁹ azide²⁰ or alkyne,^{21,22} followed by conjugation to a fluorescent dye.^{23–26} This class was rapidly adopted by the research community and extended with synthetic modifications introduced on the sulfonium center,^{27–29} the amino acid^{30,31} or the transferrable linker.³² While this method can be used to quickly introduce different labels, the conjugation of the dye often results in a lower yield. Hence, in pursuit of high degrees of fluorescent labeling, the research was continued to

^a Laboratory of Molecular Imaging and Photonics, Department of Chemistry, KU Leuven, Celestijnenlaan 200F, 3001 Leuven, Belgium.
E-mail: johan.hofkens@kuleuven.be

^b LASIR CNRS, Université de Lille, 59655 Villeneuve d'Ascq, France

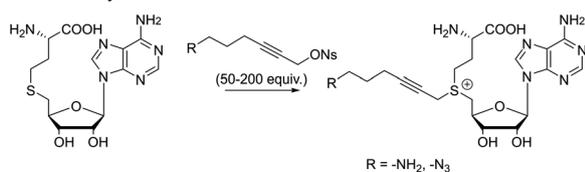
^c Laboratory for Quantitative Biology of the Nucleus, Department of Medical Biochemistry and Biophysics, Karolinska Institutet, SE-171 77 Stockholm, Sweden

^d Department of Chemistry, Max Planck Institute for Polymer Research, 55128 Mainz, Germany

† Electronic supplementary information (ESI) available: Experimental procedures, synthetic protocols and characterization details. See DOI: 10.1039/c9cc08938a



Previous Work: Homocysteine based cofactors



This Work: Cysteine based cofactors

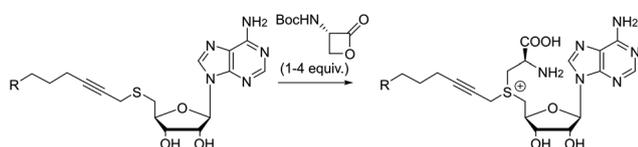


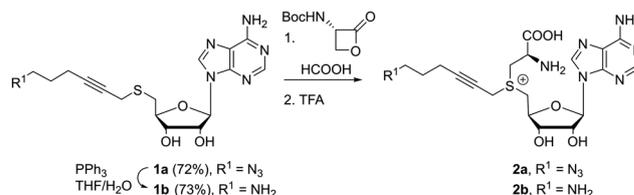
Fig. 1 Summary of previous work and present work.

fluorescent SAM analogues which contain the fluorescent molecule and directly transfer this functionality to its substrate.^{11,33}

Unfortunately, the described pathways towards these non-natural cofactors are time consuming and often result in low yields. This is especially true for the final step, where *S*-adenosyl-L-homocysteine (SAH) is coupled with a highly electrophilic alkyl triflate or allyl bromide and a high excess of this linker (50–200 equivalents) is needed to drive the reaction towards completion (Fig. 1). As the reagent used in large excess actually contains the functionality introduced to the substrate later-on, this imposes significant restraints on the flexibility of the approach. Furthermore, even with this large excess, reported yields do not exceed 25% for a single epimer after HPLC purification.⁹ Efforts to increase the formation of the correct epimer by a chemoenzymatic synthesis pathway also failed to give reasonable yields for larger systems.^{31,34}

In our pursuit of advanced applications of enzymatic DNA labeling, we were interested in the development of doubly activated cofactors, unhindered by the synthetic issues discussed above. To this end, we developed and evaluated new cysteine based SAM analogues. Our findings show that they are easily accessible and possess comparable transfer behavior as their homocysteine counterparts. The late stage introduction of the amino acid moiety allows to reach functional MTase cofactors without sacrificing labor intensive molecules.

Our first efforts were aimed at the synthesis of cofactors which can be used in a two-step labeling protocol. In such schemes, a reactive functionality, *e.g.* azide or amine, is appended to the transferable group, *e.g.* azide or amine, is appended to the transferable group, for modification at either the cofactor stage or after transfer to the substrate. Thus, we started with the coupling of 5'-thioadenosine³⁵ and 6-azido-1-bromohex-2-yne. These starting materials are well described in literature and their synthesis can be carried out at gram scale. The substitution reaction furnished thioether **1a** in good yield, which is readily reduced to the amino substituted product **1b** under Staudinger conditions. Much to our delight, these molecules are converted into the desired cofactors **2** in acidic media in the presence of one to four equivalents of the β -lactone (Scheme 1). In line with the soft nucleophilic character of the sulfur, the reaction with the strained β -lactone favors α -carbon attack over nucleophilic acylation.³⁶ For simple systems, conversion was quantitative without the formation of byproducts, and final removal of the protecting

Scheme 1 Formation of cofactors **2a** and **2b**.

group proceeded smoothly after addition of a stronger acid. Though lactone salts (tosylate, tetrafluoroborate, trifluoroacetate) can provide the desired compounds directly upon reaction with the thioethers **1**, reactions are sluggish, and the deprotection route is preferred. Due to the good conversion and low amount of byproducts formed, these cofactors can be used directly or with minimal purification (trituration or reverse phase silica gel filtration). While this is an effective and scalable method for cofactor synthesis, it should be noted that the formed cofactors are diastereomeric in nature, and the sulfonium centers are approximately a 50/50 mixture of *R* and *S* isomers. Diastereomerically pure cofactors can be obtained from HPLC purification, though the inactive isomer does not hinder further applications. The cofactors can be stored for several months at low temperatures and low pH.

Shifting from homocysteine appended cofactors to cysteine containing systems will have an effect on the position of the cofactor within the enzymatic pocket. Fortunately, gel based DNA restriction analysis indicated that MTase enzymes tolerate cysteine based cofactors, though at increased concentration (ESI,† Fig. S1–S3), proving that this pathway provides efficient access to azide and amine substituted functional cofactors.

Next, these cofactors were converted to fluorescent versions through coupling of their reactive handles with dyes. However, these reactions were always accompanied by severe side product formation, attributed to high pH (amide coupling) or incompatible reagents (Cu-salts in click-chemistry). Thus, fluorescent dye attachment was effected before the actual introduction of the amino acid.

Here, it was postulated that enzyme binding might be influenced by the fluorescent molecules, which tend to be quite bulky and often carry charges. To counter these effects, we attempted to increase the distance between dye and cofactor core. This distance can be acquired by adding aliphatic or oligo-ethylene glycol spacers. To evaluate the effect of the length and structure of the spacer, three different examples **3–5a** were prepared: (1) no linker (2) an aliphatic extension of six carbons (3) extension with three ethylene glycol units, respectively. Chain extension proceeded smoothly when using *N*-Boc protected carboxylic acids and **1b**. After removal of the protecting group, the free amine can be used in the next step without intermediate purification. Dye attachment was carried out through amide formation using similar reaction conditions as for the extension. In the presence of four equivalents of the β -lactone, the desired cofactors **3–5a** (Fig. 2) were obtained in good conversions after reverse phase silica gel filtration.

To further extend this product portfolio of chain extended direct cofactors, the dye repertoire was expanded and a small



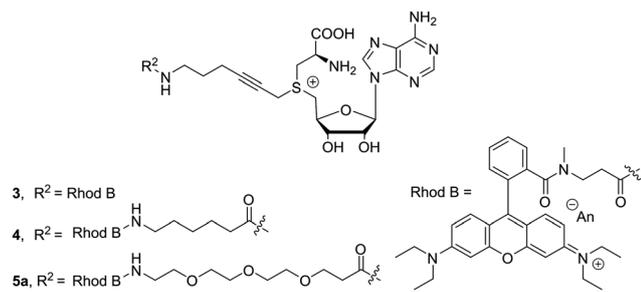


Fig. 2 Fluorescent cysteine-based SAM analogues with various spacers.

library, spanning the visible spectrum, was prepared (Fig. 3, Full structures in ESI†). To ensure solubility for cofactors conjugated to hydrophobic dyes, cofactor synthesis was started from the ethylene glycol extended molecule **S2**.

After successful synthesis of the fluorescent cofactors **3–5j**, the compatibility in MTase based labeling was validated through gel based DNA restriction analysis (Fig. 4 and ESI,† Fig. S4–S9). Interestingly, all extended cofactors **4–5** were accepted by the MTase enzyme M.TaqI. In a next step, the labeling performance was assessed through a counting assay²³ (ESI,† Fig. S10). Given their superior labeling performance, Rhodamine B containing cofactors **4** and **5a** were selected to use in further optical mapping experiments. In short, DNA fragments of the bacteriophage lambda were labeled using the M.TaqI methyltransferase enzyme (recognition sequence 5'-TCGA-3') and the fluorescent cofactor **4–5a**. Labeled DNA is linearized on a zeonex coated coverslip using the “rolling droplet” technique³⁷ and the fluorescence intensity trace (map) is extracted using Structured Illumination Microscopy (SIM). The obtained intensity profiles are cross-correlated to the theoretical profile and a matching score is assigned to each DNA strand (Fig. 5).³⁸ DNA labeled by cofactors **4** and **5a** shows excellent

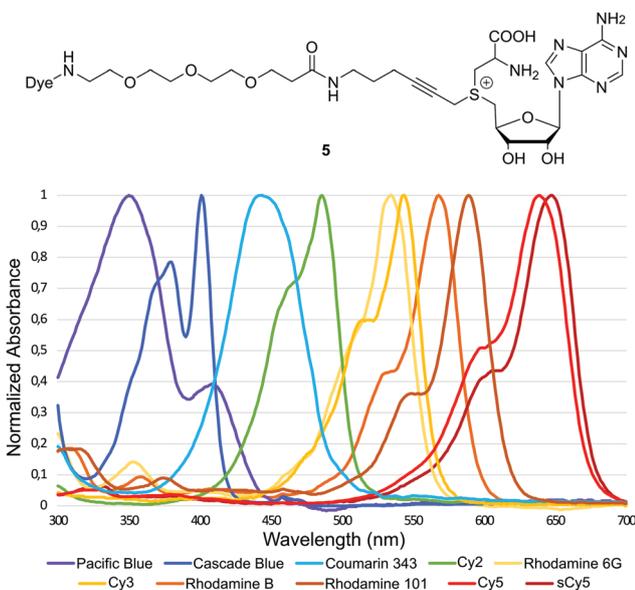


Fig. 3 General structure and normalized absorbance spectra of the synthesized fluorescent cofactors **5a–5j**.

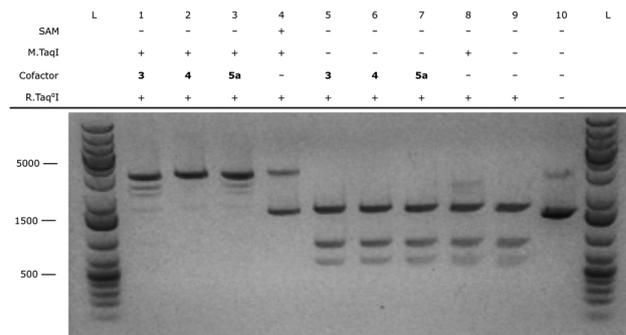


Fig. 4 TaqI restriction assay on pUC19 DNA using oligonucleotide-treated M.TaqI ($0.03 \mu\text{g} \mu\text{l}^{-1}$) with **3**, **4** and **5a**. All samples were reacted with TaqI restriction enzyme unless stated otherwise. From left to right: GeneRuler 1 kb plus (ladder); (1) M.TaqI with $50 \mu\text{M}$ **3**, (2) M.TaqI with $50 \mu\text{M}$ **4**, (3) M.TaqI with $50 \mu\text{M}$ **5a**, (4) control sample with $50 \mu\text{M}$ natural SAM cofactor, (5) control sample without M.TaqI enzyme and with $50 \mu\text{M}$ **3**, (6) control sample without M.TaqI enzyme and with $50 \mu\text{M}$ **4**, (7) control sample without M.TaqI enzyme and with $50 \mu\text{M}$ **5a**, (8) control sample without cofactor, (9) control sample without M.TaqI enzyme and without cofactor, (10) pUC19 DNA, GeneRuler 1 kb plus (ladder).

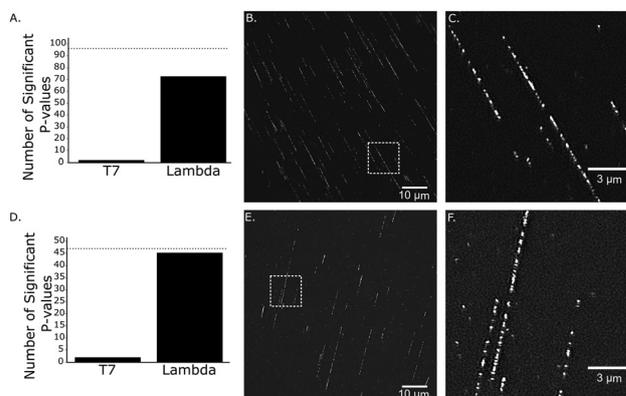


Fig. 5 Matching results for **5a** and **4**. (A) Result obtained after matching 96 experimental maps (dashed line), obtained through **5a**-mediated labeling of bacteriophage lambda. Out of 96 maps, 73 matched significantly ($\alpha = 0.05$) to the ground truth species. (B) Field of view of **5a** labeled bacteriophage lambda DNA obtained through SIM. (C) Cropped image from B as indicated by white dashed rectangle. (D) Result obtained after matching 47 experimental maps (dashed line), obtained through **4**-mediated labeling of bacteriophage lambda. Out of 47 maps, 45 matched significantly ($\alpha = 0.05$) to the ground truth species. (E) Field of view of **4** labeled bacteriophage lambda DNA obtained through SIM. (F) Cropped image from E as indicated by white dashed rectangle. 20 kilobasepairs was used as a size threshold, smaller molecules were not analyzed.

matching to the ground-truth species, indicating the applicability of these novel cofactor systems for optical mapping approaches.

In conclusion, cysteine based SAM analogues bring forward an easily accessible class of non-natural methyltransferase cofactors which are obtained through a high yielding pathway with limited byproduct formation. These compounds can be used to introduce various dyes onto DNA and provide excellent matching in optical mapping experiments. Despite a transfer efficiency that is somewhat reduced when compared to similar



homocysteine cofactor analogues, their synthetic accessibility warrants widespread use.

This work was supported by the European Union's Horizon 2020 research and innovation program under grant agreement No. 686271, "AD-Gut" and European Union's Horizon 2020 research and innovation program under grant agreement No. 634890, "BeyondSeq". J. H. gratefully acknowledges financial support of the Flemish government through long-term structural funding Methusalem (CASAS2, Meth/15/04). Mass spectrometry was made possible by the support of the Hercules Foundation of the Flemish Government (grant 20100225-7). L. D. gratefully acknowledges the Fonds voor Wetenschappelijk Onderzoek (FWO) Aspirant for a PhD fellowship (11D3718N). Open Access funding provided by the Max Planck Society.

Conflicts of interest

J. H. and V. L. are inventors on a patent application around the cofactor structures described in this manuscript.

Notes and references

- 1 A. Gottfried and E. Weinhold, *Biochem. Soc. Trans.*, 2011, **39**, 623–628.
- 2 R. K. Neely, P. Dedecker, J. Hotta, G. Urbanavičiūtė, S. Klimašauskas and J. Hofkens, *Chem. Sci.*, 2010, **1**, 453–460.
- 3 R. K. Neely, J. Deen and J. Hofkens, *Biopolymers*, 2011, **95**, 298–311.
- 4 M. Levy-Sakin and Y. Ebenstein, *Curr. Opin. Biotechnol.*, 2013, **24**, 690–698.
- 5 M. Mrksich, M. E. Parks and P. B. Dervan, *J. Am. Chem. Soc.*, 1994, **116**, 7983–7988.
- 6 N. T. Thuong and C. Hélène, *Angew. Chem., Int. Ed. Engl.*, 1993, **32**, 666–690.
- 7 J. Deen, C. Vranken, V. Leen, R. K. Neely, K. P. F. Janssen and J. Hofkens, *Angew. Chem., Int. Ed.*, 2017, **56**, 5182–5200.
- 8 G. Pljevaljčić, F. Schmidt and E. Weinhold, *ChemBioChem*, 2004, **5**, 265–269.
- 9 C. Dalhoff, G. Lukinavičius, S. Klimašauskas and E. Weinhold, *Nat. Chem. Biol.*, 2006, **2**, 31–32.
- 10 X. Cheng, *Annu. Rev. Biophys. Biomol. Struct.*, 1995, **24**, 293–318.
- 11 A. Grunwald, M. Dahan, A. Giesbertz, A. Nilsson, L. K. Nyberg, E. Weinhold, T. Ambjörnsson, F. Westerlund and Y. Ebenstein, *Nucleic Acids Res.*, 2015, **43**, e117.
- 12 F. Kunkel, R. Lurz and E. Weinhold, *Molecules*, 2015, **20**, 20805–20822.
- 13 L. Anhäuser, F. Muttach and A. Rentmeister, *Chem. Commun.*, 2018, **54**, 449–451.
- 14 M. Heimes, L. Kolmar and C. Brieke, *Chem. Commun.*, 2018, **54**, 12718–12721.
- 15 F. Muttach, F. Mäsing, A. Studer and A. Rentmeister, *Chem. – Eur. J.*, 2017, **23**, 5988–5993.
- 16 L. R. Comstock and S. R. Rajski, *J. Am. Chem. Soc.*, 2005, **127**, 14136–14137.
- 17 M. Pignot, C. Siethoff, M. Linscheid and E. Weinhold, *Angew. Chem., Int. Ed.*, 1998, **37**, 2888–2891.
- 18 R. L. Weller and S. R. Rajski, *ChemBioChem*, 2006, **7**, 243–245.
- 19 G. Lukinavičius, V. Lapienė, Z. Staševskij, C. Dalhoff, E. Weinhold and S. Klimašauskas, *J. Am. Chem. Soc.*, 2007, **129**, 2758–2759.
- 20 G. Lukinavičius, M. Tomkuvienė, V. Masevičius and S. Klimašauskas, *ACS Chem. Biol.*, 2013, **8**, 1134–1139.
- 21 W. Peters, S. Willnow, M. Duisken, H. Kleine, T. Macherey, K. Duncan, D. Litchfield, B. Lüscher and E. Weinhold, *Angew. Chem., Int. Ed.*, 2010, **49**, 5170–5173.
- 22 C. Vranken, J. Deen, L. Dirix, T. Stakenborg, W. Dehaen, V. Leen, J. Hofkens and R. K. Neely, *Nucleic Acids Res.*, 2014, **42**, e50.
- 23 M. H. Lauer, C. Vranken, J. Deen, W. Frederickx, W. Vanderlinden, N. Wand, V. Leen, M. H. Gehlen, J. Hofkens and R. K. Neely, *Chem. Sci.*, 2017, **8**, 3804–3811.
- 24 J. Jeffet, A. Kobo, T. Su, A. Grunwald, O. Green, A. N. Nilsson, E. Eisenberg, T. Ambjörnsson, F. Westerlund, E. Weinhold, D. Shabat, P. K. Purohit and Y. Ebenstein, *ACS Nano*, 2016, **10**, 9823–9830.
- 25 J. Deen, S. Wang, S. Van Snick, V. Leen, K. Janssen, J. Hofkens and R. K. Neely, *Nucleic Acids Res.*, 2018, **46**, e64.
- 26 N. O. Wand, D. A. Smith, A. A. Wilkinson, A. E. Rushton, S. J. W. Busby, I. B. Styles and R. K. Neely, *Nucleic Acids Res.*, 2019, **47**, e68.
- 27 I. R. Bothwell, K. Islam, Y. Chen, W. Zheng, G. Blum, H. Deng and M. Luo, *J. Am. Chem. Soc.*, 2012, **134**, 14905–14912.
- 28 I. R. Bothwell and M. Luo, *Org. Lett.*, 2014, **16**, 3056–3059.
- 29 S. Willnow, M. Martin, B. Lüscher and E. Weinhold, *ChemBioChem*, 2012, **13**, 1167–1173.
- 30 T. D. Huber, F. Wang, S. Singh, B. R. Johnson, J. Zhang, M. Sunkara, S. G. V. Lanen, A. J. Morris, J. George, N. Phillips and J. S. Thorson, *ACS Chem. Biol.*, 2016, **11**, 2484–2491.
- 31 T. D. Huber, B. R. Johnson, J. Zhang and J. S. Thorson, *Curr. Opin. Biotechnol.*, 2016, **42**, 189–197.
- 32 B. W. Lee, H. G. Sun, T. Zang, B. J. Kim, J. F. Alfaro and Z. S. Zhou, *J. Am. Chem. Soc.*, 2010, **132**, 3642–3643.
- 33 S. Flade, J. Jasper, M. Gieß, M. Juhasz, A. Dankers, G. Kubik, O. Koch, E. Weinhold and D. Summerer, *ACS Chem. Biol.*, 2017, **12**, 1719–1725.
- 34 F. Wang, S. Singh, J. Zhang, T. D. Huber, K. E. Helmich, M. Sunkara, K. A. Hurley, R. D. Goff, C. A. Bingman, A. J. Morris, J. S. Thorson and G. N. P. Jr, *FEBS J.*, 2014, **281**, 4224–4239.
- 35 M. Pignot, G. Pljevaljčić and E. Weinhold, *Eur. J. Org. Chem.*, 2000, 549–555.
- 36 L. D. Arnold, T. H. Kalantar and J. C. Vederas, *J. Am. Chem. Soc.*, 1985, **107**, 7105–7109.
- 37 J. Deen, W. Sempels, R. De Dier, J. Vermant, P. Dedecker, J. Hofkens and R. K. Neely, *ACS Nano*, 2015, **9**, 809–816.
- 38 A. Bouwens, J. Deen, R. Vitale, L. D'Huys, V. Goyvaerts, A. Descloux, D. Borrenberghs, K. Grussmayer, T. Lukes, R. Camacho, J. Su, C. Ruckebusch, T. Lasser, D. V. De Ville, J. Hofkens, A. Radenovic and K. P. F. Janssen, *NAR Genomics and Bioinformatics*, 2020, **2**, lqz007.

