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Stereocontrolled addition of Grignard reagents to oxa-bridged benzazepines: highly efficient synthesis of functionalized benzazepine scaffolds†

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An efficient and highly diastereoselective synthesis of 2-substituted benzo[b]azepin-5-ol via stereocontrolled addition of Grignard reagents to oxa-bridged benzazepines has been developed. The reaction proceeds efficiently starting from versatile skeletons with mild reaction conditions as well as simple operation. Furthermore, 2-substituted benzazepinones could been obtained by simple Dess–Martin oxidation in excellent yields.

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Benzofused azepines, a unique family of seven-member azaheterocycles, are widely found in numerous bioactive molecules, natural products and pharmaceuticals.¹⁻⁴ This is due to their chemotherapeutic properties, and exhibiting interesting biological activities,⁵⁻⁹ for instance, competitive vasopressin receptor antagonist (tolvaptan),¹⁰⁻¹² antidepressants (mianserin),¹³ zilpaterol (beef improvement agent),¹⁴ ACE inhibitor (benazepril)¹⁵ (Fig. 1).

Consequently, tremendous efforts have recently been dedicated to developing new methodologies to construct the benzazepine derivatives. Typically, the benzazepine skeletons could be assembled by expansion of smaller rings, rearrangements, ^{16,17} Dieckmann cyclization, ^{18,19} transition-metal-catalyzed coupling, ring closure metathesis, ^{15,20,21} and others. ^{22–25} Nevertheless, most of these protocols are limited to highly engineered starting materials, expensive catalysts and hazardous handling, obviously expeditious strategies for the diverse construction of benzazepine backbones from readily available starting materials, remains highly attractive and challenging.

Diversity-oriented synthesis (DOS), defined as a powerful synthetic strategy to the libraries of diverse highly valuable molecules from one parent compound, ^{26,27} is therefore well-suited for the timely design and execution of parallel (library) synthesis. ²⁸ In recent years, our group focused on the development of a more facile and efficient diversity-oriented synthesis strategy for the generation of this class of 7-membered heterocyclic compounds. ²⁹⁻³¹ This newly introduced ene-type cyclization reaction was used to prepare a series of bridged aromatic fused azepines, ²⁹ as a versatile building block, which could be

transformed into structurally different ring systems through selective ring opening of the cyclic acetals (Scheme 1A).^{30,31}

As an extension of our ongoing work toward the synthesis of the azepine skeleton, we suggested a new reaction model could be achieved if the suitable nucleophile could be carefully

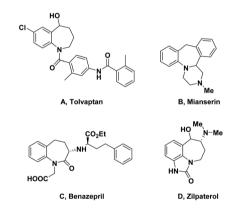
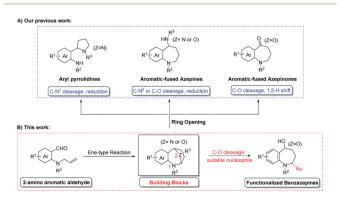


Fig. 1 Selected examples containing a benzazepine skeleton.



Scheme 1 Our previous work (A) and this work (B).

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designed. Recently, a couple of efficient approaches to access nitrogen-containing heterocycles has been developed through the nucleophilic addition of cyclic N,O-acetal with Grignard reagents. 32-35 Inspired by their excellent studies and to showcase the utility of cyclic N,O-acetal building blocks for the preparation of functionalized azepines, we present a facile approach to a stereoselective synthesis of 2,5-substituted benzazepine derivatives from oxa-bridged benzazepines by Grignard addition. This strategy is complementary to our recently published cascade reaction to prepare the benzazepinone scaffold. Herein, the details of this study is disclosed.

Our investigations commenced by exploring nucleophilic addition of 1a, which was readily prepared in two steps via substitution reaction and subsequent ene-type reaction (see the ESI†). We started our screening with 1-allyl-2,3,4,5-tetrahydro-1H-2,5-epoxybenzo[b]azepine (1a) as a model substrate for the optimization of the reaction conditions (Table 1). First, we chose the commonly used solvent tetrahydrofuran and dioxane for Grignard addition, no 2a was observed (Table 1, entries 1 and 2). The desired product 2a was obtained in 92% yield, along with low diastereoselectivity (dr = 24:76), when the reaction was carried out in diethyl ether (Table 1, entry 3). Subsequently, we conducted this nucleophilic addition of 1a in halogen-containing solvent instead of the more commonly used ether solvent to increase the coordination of organomagnesium to the substrates. 36,37 We were gratified to find that Grignard reagents could indeed be added with high selectivity (Table 1, entries 4-6).

However, there was no significant improvement in the diastereoselectivity was observed after the addition of magnesium bromide³⁶ (Table 1, entry 7). The experimental results show that the ring-opening reaction of N,O acetals were sensitive to size of substituent R¹, and the syn-selectivity became better as the size of Grignard reagents increased (Table 1, entries 8, 9).

After establishing the optimized reaction conditions, we investigated the addition of Grignard reagents to a series of oxabridged azepine 1, and the results are listed in Table 2. We were pleased to observe that this reaction exhibited broad substrate scope, and the aliphatic and aromatic Grignard reagents reacted smoothly with 1a. These reactions generated aminoalcohols

Table 2 Nucleophilic addition with Grignard reagents on cyclic N,Oacetals^a

N,O-Acetals	R^1	\mathbb{R}^2	\mathbb{R}^3	Time	2	Yield (%)	syn/anti ^b
10	Mo	ш	A llv.l	10 min	20	06	69/31
			,				
1a	Et	Н	Allyl	10 min	2b	97	91/9
1a	<i>i</i> -Pr	Η	Allyl	7 h	2c	88	100/0
1a	Cy	Н	Allyl	18 h	2d	83	100/0
1a	Allyl	Н	Allyl	10 min	2e	98	0/100
1a	Ph	Н	Allyl	10 min	2f	98	100/0
1b	Me	Me	Allyl	10 min	2g	91	88/12
1b	Allyl	Me	Allyl	10 min	2h	90	0/100
1c	Me	Cl	Allyl	10 min	2i	92	67/33
1c	Allyl	Cl	Allyl	10 min	2j	91	0/100
1d	Me	Н	Me	10 min	2k	92	68/32
1d	i-Pr	Н	Me	5 min	21	91	100/0
1e	Me	Н	H	1 h	2m	80	91/9
1e	Allyl	Н	H	10 min	2n	81	75/25
	1a 1a 1a 1a 1a 1a 1b 1b 1c 1c 1d 1d 1d	1a Me 1a Et 1a i-Pr 1a Cy 1a Allyl 1b Me 1b Allyl 1c Me 1c Allyl 1d Me 1d i-Pr 1e Me	1a Me H 1a Et H 1a i-Pr H 1a Cy H 1a Allyl H 1b Me Me 1b Allyl Me 1c Me Cl 1c Allyl Cl 1d Me H 1d i-Pr H 1e Me H	1a Me H Allyl 1a Et H Allyl 1a i-Pr H Allyl 1a Cy H Allyl 1a Allyl H Allyl 1b Me Me Allyl 1b Allyl Me Allyl 1c Me Cl Allyl 1c Allyl Cl Allyl 1d Me H Me 1d i-Pr H Me 1e Me H H	1a Me H Allyl 10 min 1a Et H Allyl 10 min 1a i-Pr H Allyl 7 h 1a Cy H Allyl 18 h 1a Allyl H Allyl 10 min 1b Me Me Allyl 10 min 1b Allyl Me Allyl 10 min 1c Me Cl Allyl 10 min 1c Allyl Cl Allyl 10 min 1d Me H Me 5 min 1e Me H H 1 h	1a Me H Allyl 10 min 2a 1a Et H Allyl 10 min 2b 1a i-Pr H Allyl 7 h 2c 1a Cy H Allyl 18 h 2d 1a Allyl H Allyl 10 min 2e 1a Ph H Allyl 10 min 2f 1b Me Me Allyl 10 min 2h 1b Allyl Me Allyl 10 min 2h 1c Me Cl Allyl 10 min 2i 1c Allyl Cl Allyl 10 min 2j 1d Me H Me 10 min 2k 1d i-Pr H Me 5 min 2l 1e Me H H 1 h 1 h 2m	1a Me H Allyl 10 min 2a 96 1a Et H Allyl 10 min 2b 97 1a i-Pr H Allyl 7 h 2c 88 1a Cy H Allyl 18 h 2d 83 1a Allyl H Allyl 10 min 2e 98 1a Ph H Allyl 10 min 2f 98 1b Me Me Allyl 10 min 2g 91 1b Allyl Me Allyl 10 min 2h 90 1c Me Cl Allyl 10 min 2i 92 1c Allyl Cl Allyl 10 min 2k 92 1d Me H Me 10 min 2k 92 1d i-Pr H Me 5 min 2l 91 1e Me H H 1 h 2m 80

^a Unless indicated otherwise, the reaction was carried out on 1.0 mmol scale in DCM (10 mL). b Diastereoisomeric ratios were determined by H NMR analysis of the mixture, see the ESI for details.

Table 1 Optimization of the Grignard addition conditions^a

Entry	Solvent	Additive	Time	$Yield^{b}$ (%)	dr (syn/anti) ^c
1 ^d	THF	_	10 h	0	
2^d	Dioxane	_	10 h	0	_
3	Et ₂ O	_	30 min	92	2a, 24/76
4	1,2-Dichloroethane	_	10 min	90	2a , 68/32
5	CHCl ₃	_	10 min	95	2a, 66/34
6	$\mathrm{CH_2Cl_2}$	_	10 min	96	2a, 69/31
7	CH_2Cl_2	$MgBr_2$ (1.2 equiv.)	10 min	90	2a, 70/30
8	CH_2Cl_2		10 min	97	2b , 91/9
9	$\mathrm{CH_2Cl_2}$	_	10 min	88	2c, 100/0

^a Reaction conditions: 1a (1 mmol), MeMgBr in 10 mL of solvent at 0 °C under air. ^b Isolated yield after column chromatography. ^c Determined by 1 H NMR and X-ray crystallographic analysis. d The reaction was conducted at 0 $^\circ$ C for 1 h, then at 25 $^\circ$ C for 9 h.

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2 in good chemical yields and with excellent selectivity and the structure of 2m was further confirmed by X-ray crystallography analysis. The compound 2m was recrystallized in an ethyl acetate/petroleum ether solution to obtain a single configuration compound syn-2m, which was tested by single crystal X-ray diffraction (Fig. 2). From the diffraction pattern, it could be clearly seen that the hydroxyl group and the methyl group are on the same side. The stereochemistry in the rest of the series could be unambiguously assigned by comparison of their NMR spectra with those of syn-2m.

In general, the improved diastereoselectivities were observed with increasing steric bulk of the Grignard reagents, the synadduct was the major product in all cases (Table 2, entries1-4). The anomalous diastereoselection shown by allyl Grignard reagent is to be underlined. Possibly this is due to the peculiar nature of allyl metals (Table 2, entries 5, 8, 10).38-43 Even phenylmagnesium bromide could be added to 1a and provided the adduct in 98% yield and a single syn-adduct (Table 2, entry 6), despite the considerably higher basicity of this Grignard reagent.³⁷ 7-Substituented substrates (1b, 1c) were also tolerated by this process, as well as Me at the nitrogen atom of the substrate (1d) (Table 1, entries 7-12). To further expand the scope of this reaction, N-unsubstituted cyclic N,O-acetal 1e was employed under the established condition, the expected product was obtained smoothly and in good yield as well. Surprisingly, even lower proportions of syn-diastereoisomers were observed relative to methyl Grignard reagent correlated with increasing steric bulk of the Grignard reagent (Table 2, entries 13-14).

The observed diastereoselectivity for the Grignard reaction of N-substituted cyclic N,O-acetals 1 leading to 2 can be rationalized by assuming that the Grignard reagent coordinates with the oxygen atom of the cyclic N,O-acetal ring 1 and that the subsequent intramolecular delivery of the alkyl group occurs on the same face of the C-O bond of the incipient iminium salt A as shown in Fig. 3 (transition state A).36,44,45 However, this diastereoselectivity for Grignard addition to the N-unsubstituted cyclic N,O-acetal 1a may be attributed to a highly ordered transition state resulting from significant chelation of the alkoxy substituent and imino nitrogen to at least one magnesium cation as shown Fig. 3 (transition state B). 46,47

To give the intrinsic versatility of 2-substituted benzo[b]azepin-5-ol and as a complement to our recently published cascade reaction to prepare the benzazepinone scaffold, treatment of the above compounds 2 with Dess-Martin in CH2Cl2 gave the 2-substituted benzazepinones 3 in good to excellent yields (Table 3, entries 1-10), except the 3k (Table 3, entry 11).



Fig. 2 X-ray crystallographic structure of syn-2m.



3 Proposed transition the states accounting for diastereoselectivity

Transition State A

The synthetic versatility of 2-substituted benzazepinones has also been explored. The fused tricyclic compound 4 (ref. 48) could also be readily synthesized from 3e. Treatment of 3e with Grubbs II catalyst led to 4 in 95% yield [egn (1)].

Transition State B

In conclusion, we have demonstrated that the cyclic N,Oacetals were successfully applied to the diastereoselective addition of various Grignard reagents with encouraging levels of stereoselection. In the formation of 2,5-substituted 1-benzazepine derivatives, the reaction proceeds through a ring-opening/nucleophilic addition pathway. These benzo [b]azepin-5-ols then undergo simple Dess-Martin oxidation to afford the 2-substituted benzazepinones in excellent yields. In respect to the easy availability of the starting materials, simple manipulation, mild conditions and high diastereoselectivity, this reaction will be synthetically useful in organic chemistry.

Table 3 The synthesis of 2-substituted benzazepinones^a

Entry	R^1	R^2	R^3	Time (min)	3	Yield (%)
1	Н	Allyl	Me	30	3a	85
2	Н	Allyl	Et	30	3b	83
3	Н	Allyl	<i>i</i> -Pr	15	3 c	86
4	Н	Allyl	Cy	15	3d	80
5	H	Allyl	Allyl	15	3e	75
6	H	Allyl	Ph	15	3f	88
7	Me	Allyl	Me	20	3g	88
8	Me	Allyl	Allyl	45	3h	76
9	Cl	Allyl	Me	60	3i	84
10	Cl	Allyl	Allyl	60	3j	78
11	H	Н	Me	20	3k	38

^a Unless indicated otherwise, the reaction was carried out on 0.5 mmol scale in DCM (5 mL).

Conflicts of interest

There are no conflicts to declare.

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