

Cite this: *J. Mater. Chem. A*, 2019, 7, 2686Received 21st October 2018
Accepted 5th January 2019

DOI: 10.1039/c8ta10130b

rsc.li/materials-a

Lone-pair effect on carrier capture in $\text{Cu}_2\text{ZnSnS}_4$ solar cells

Sunghyun Kim,^{ID}*^a Ji-Sang Park,^{ID}^a Samantha N. Hood^{ID}^a and Aron Walsh^{ID}*^{ab}

The performance of kesterite thin-film solar cells is limited by a low open-circuit voltage due to defect-mediated electron–hole recombination. We calculate the non-radiative carrier-capture cross sections and Shockley–Read–Hall recombination coefficients of deep-level point defects in $\text{Cu}_2\text{ZnSnS}_4$ (CZTS) from first-principles. While the oxidation state of Sn is +4 in stoichiometric CZTS, inert lone pair ($5s^2$) formation lowers the oxidation state to +2. The stability of the lone pair suppresses the ionization of certain point defects, inducing charge transition levels deep in the band gap. We find large lattice distortions associated with the lone-pair defect centers due to the difference in ionic radii between Sn(II) and Sn(IV). The combination of a deep trap level and large lattice distortion facilitates efficient non-radiative carrier capture, with capture cross-sections exceeding 10^{-12} cm². The results highlight a connection between redox active cations and ‘killer’ defect centres that form giant carrier traps. This lone pair effect will be relevant to other emerging photovoltaic materials containing ns^2 cations.

Introduction

In a semiconductor subject to above-band-gap illumination, the lifetime of charge carriers is determined by the kinetics of electron–hole recombination: radiative, Auger, and trap-assisted processes.^{1,2} Radiative and Auger recombination usually only become significant at high carrier concentrations such as in light-emitting diodes or solar cells using concentrated sunlight. In most photovoltaic technologies, defects limit carrier lifetimes and device efficiencies by acting as non-radiative electron–hole recombination centers.^{3,4}

Thin-film solar cells offer advantages over traditional silicon-based solar cells as they require less raw materials and energy to produce, and open up new application areas such as building-integrated photovoltaics. As an alternative to the current thin-film light absorbers such as CdTe and $\text{Cu}(\text{In,Ga})\text{Se}_2$ (CIGS) whose constituting elements are vulnerable to decreases in supply, kesterite minerals such as $\text{Cu}_2\text{ZnSnS}_4$ (CZTS) and $\text{Cu}_2\text{-ZnSnS}_4$ (CZTSe) (see Fig. 1(a)), have attracted much attention due to the earth-abundance of Cu, Zn, and Sn.^{5–7} In 2014, an alloyed CZTSSe kesterite solar cell reached a record light-to-electricity conversion efficiency of 12.6%.⁸ Recently, 11% efficiency is achieved in a pure sulfide CZTS solar cell.⁹ However, this technology suffers from a large open-circuit voltage (V_{OC}) deficit.^{10–12} The performance of current kesterite-based solar cells falls far below the Shockley–Queisser limit of $\sim 30\%$.^{13,14}

One likely origin of the V_{OC} deficit is a short minority carrier (electron) lifetime of below few ns due to fast non-radiative recombination pathways.^{12,15}

Thus, it is important to identify dominant recombination centers and to control their concentrations. According to Shockley–Read–Hall (SRH) statistics,^{16,17} a deep level in the band gap of a semiconductor acts as an efficient recombination channel that facilitates the sequential capture of minority and majority carriers. In addition to deep thermodynamic charge transition levels, large lattice distortions are required to achieve fast recombination rates.^{3,18} However, due to the strong interactions between an impurity and a host material, it is hard to

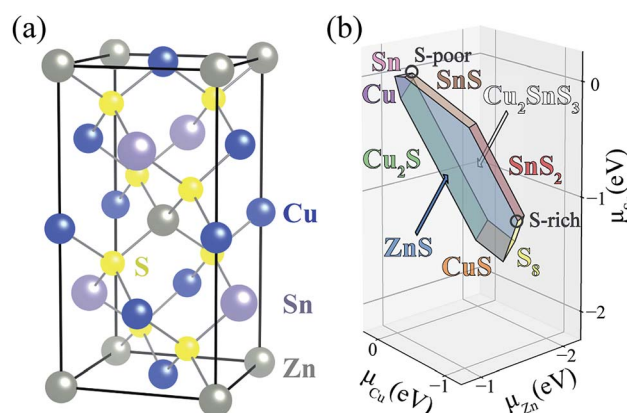


Fig. 1 (a) Atomic structure and (b) phase diagram of CuZnSnS_4 in chemical potential space, where $\mu_i = 0$ represent element i in its standard state. Blue, gray, purple and yellow balls represent Cu, Zn, Sn, and S atoms, respectively.

^aDepartment of Materials, Imperial College London, London SW7 2AZ, UK. E-mail: sunghyun.kim@imperial.ac.uk

^bDepartment of Materials Science and Engineering, Yonsei University, Seoul 03722, Korea. E-mail: a.walsh@imperial.ac.uk



predict the properties (charge transition level and lattice distortion) of impurities *a priori* or to find general trends in various host materials. Identification of recombination centers has relied on individual experimental or theoretical studies. If a simple criterion existed, then we could identify detrimental defects limiting the efficiency more easily and screen candidate optoelectronic materials more efficiently.

In this work, we argue that defects in semiconductors involving heavy post-transition metals are likely to act as fast non-radiative recombination centers not only due to their deep-level nature but also the large lattice distortions that accompany a change in oxidation state. As a representative case, we study the native point defects in CZTS containing multivalent ions of Sn and Cu. Through analysis of carrier capture rates from first-principles, we find that the dominant non-radiative recombination centers (V_S , V_S -Cu_{Zn} and Sn_{Zn}) are associated with Sn 5s² lone-pair configurations. These defects produce deep donor levels due to the Sn double reduction, and the recombination processes involve large structure distortions because of the change in the ionic radius of Sn during the carrier capture. We expect to find similar behaviour for other lone pair cations including Bi and Sb.

Here, we use the notion X_Y^q for a defect where X is the species (e.g. Cu, Zn, Sn, S or V for a vacancy), Y is the lattice site, and q is the charge state if specified. For example, V_S represents the sulfur vacancy and Sn_{Zn} is Sn-on-Zn antisite. Defect complexes are represented by symbols connected by a dash as in V_S -V_{Cu}.

Methods

Non-radiative carrier capture

The phenomenon of carrier capture in semiconductors *via* multiphonon emission has been extensively studied following pioneering work by Huang and Rhys,¹⁹ and Henry and Lang.²⁰ The initial excited state of system, for example, a positively charged donor (D^+) with an electron in the conduction band (e^-), vibrates around the equilibrium geometry. Owing to the electron-phonon coupling, the deformation of the structure causes the electronic energy level of a state localized around the defect to oscillate. As the defect level approaches the conduction band, the probability for the defect to capture an electron increases significantly. When an electron is captured, the donor becomes neutral (D^0) and relaxes to a new equilibrium geometry by emitting multiple phonons as shown in Fig. 2. To describe and predict this process, quantitative accounts of the electronic and atomic structures as well as the vibrational properties of the defect are essential.

Recently, approaches have been developed for first-principles calculations of capture rates within a certain set of approximations.^{21,22} We have adopted a one-dimensional configuration coordinate for the effective vibrational wave function and the static coupling theory for electron-phonon coupling matrix elements as proposed by Alkauskas *et al.*^{22,23}

We described the degree of deformation using a one-dimensional configuration coordinate Q defined by

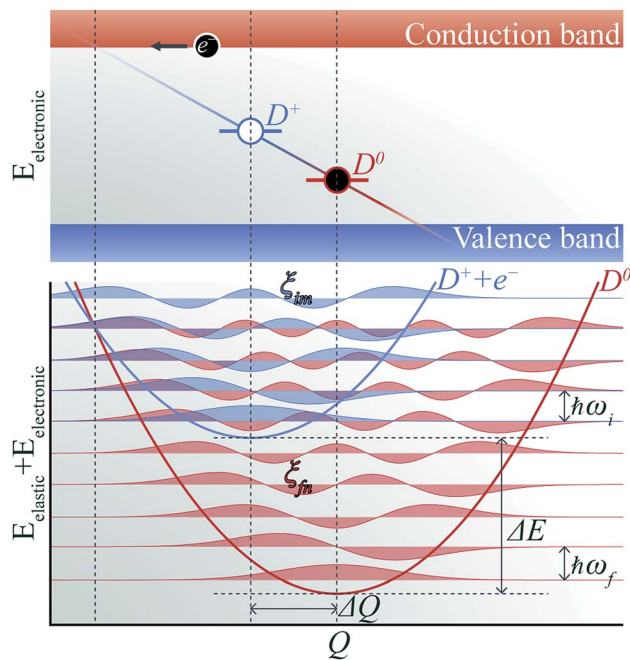


Fig. 2 Schematic of non-radiative electron (e^-) capture by a positively charged defect (D^+) turning into the neutral defect (D^0) in band diagram (upper panel) following Henry and Lang (ref. 20). The electron (black circle) is captured by the empty level (blue line) crossing the conduction band due to the thermal vibration of local geometry. The diagonal line represents evolution of the electronic energy level as the local geometry (Q) vibrates. The occupied level (red line) has a lower electronic energy due to the change in the equilibrium geometry (ΔQ). Configuration coordinate diagram (lower panel) for the same process shows potential energy surfaces of $D^+ + e^-$ (blue curve) and D^0 (red curve). The vibrational wave functions (ξ_{im} and ξ_{fn} , see the text) are shown in lighter shades.

$$Q^2 = \sum_{\alpha,i} m_{\alpha} \Delta R_{\alpha,i}^2, \quad (1)$$

where m_{α} and $\Delta R_{\alpha,i}$ are the atomic mass and the displacement along the direction i from the equilibrium position of atom α , respectively. The vibrational wave function of excited (ξ_{im}) and ground (ξ_{fn}) states, and associated frequencies ω_i and ω_f were obtained by solving the one-dimensional Schrödinger equation for potential energy surfaces around the equilibrium geometries. The capture coefficient is given by

$$C = V \frac{2\pi}{\hbar} g W_{if}^2 \sum_m w_m \sum_n |\langle \xi_{im} | Q | \xi_{fn} \rangle|^2 \delta(\Delta E + m\hbar\omega_i - n\hbar\omega_f), \quad (2)$$

where V , g and W_{if}^2 are the volume of supercell, the degeneracy factor and the electron-phonon coupling matrix element of initial and final states, respectively. w_m is the occupation number of the excited vibrational state ξ_{im} , and ΔE corresponds to the difference in energy of excited and ground states.

The Coulomb interaction at temperature T between a carrier with charge q and a defect in a charge state Q is accounted by the Sommerfeld factor $\langle s \rangle$;^{24,25}



$$\langle s \rangle = \begin{cases} 4|Z|(\pi E_R/k_B T)^{\frac{1}{2}}, & \text{for } Z < 0 \\ 8/\sqrt{3}(\pi^2 Z^2 E_R/k_B T)^{\frac{2}{3}} \times \exp\left(-3(Z^2 \pi^2 E_R/k_B T)^{\frac{1}{3}}\right), & \text{for } Z > 0 \end{cases}, \quad (3)$$

where k_B is the Boltzmann constant. $E_R = m^*q^4/(2\hbar^2\epsilon^2)$ is an effective Rydberg energy where m^* and ϵ are an effective mass of the carrier and a low-frequency dielectric constant, respectively. For an attractive center, $Z = Q/q$ is negative, while Z is positive for a repulsive center.

Based on the principle of detailed balance, the steady-state recombination rate R via a defect with electron-capture cross section σ_n and hole-capture cross section σ_p is given by^{16,17}

$$R = \frac{np - n_i^2}{\tau_p(n + n_1) + \tau_n(p + p_1)}, \quad (4)$$

where

$$\begin{aligned} \tau_n^{-1} &= N_T \sigma_n v_{th,n} = N_T C_n, \\ \tau_p^{-1} &= N_T \sigma_p v_{th,p} = N_T C_p. \end{aligned} \quad (5)$$

Here, n , p and N_T denote concentrations of electrons, holes and defects, respectively. n_1 and p_1 represent the densities of electrons and holes, respectively, when the Fermi level is located at the trap level. The thermal velocities of electrons $v_{th,n} = \sqrt{3k_B T/m_e^*}$ and holes $v_{th,p} = \sqrt{3k_B T/m_h^*}$ are calculated from the effective masses of electron (m_e^*) and hole (m_h^*) in the electronic band structure. C_n and C_p are the capture coefficients for electron and hole, respectively. Finally, the Shockley–Read–Hall recombination coefficient A is given by

$$A = R/\Delta n. \quad (6)$$

We assume an excess carrier concentration $\Delta n = 1 \times 10^{14} \text{ cm}^{-3}$, which results in an open-circuit voltage V_{OC} of 0.9 V. The Shockley–Read–Hall coefficient is not sensitive to the excess carrier concentration as long as Δn is much lower than the majority carrier concentration.

Electronic structure theory

The atomic and electronic structure of defects were calculated from first-principles within the framework of density functional theory (DFT).^{26,27} We employed the projector-augmented wave (PAW) method²⁸ and the hybrid exchange–correlation functional of Heyd–Scuseria–Ernzerhof (HSE06),²⁹ as implemented in VASP.³⁰ The wave functions were expanded in plane waves up to an energy cutoff of 380 eV. A Monkhorst–Pack k -mesh³¹ with a grid spacing less than $2\pi \times 0.03 \text{ \AA}^{-1}$ was used for Brillouin zone integration. The atomic coordinates were optimized until the residual forces were less than 0.02 eV \AA^{-1} . The lattice vectors were relaxed until stress was below 0.5 kbar. For defect formation, a $2 \times 2 \times 1$ supercell expansion (64 atoms) of the conventional cell was employed.

We calculated the formation energy $\Delta E_{\text{form}}(D^q)$ of a defect D in the charge state q which is given by³²

$$\Delta E_{\text{form}}(D^q) = E_{\text{tot}}(D^q) - E_{\text{tot}}(\text{bulk}) - \sum_i n_i \mu_i + qE_F + E_{\text{corr}}, \quad (7)$$

where $E_{\text{tot}}(\text{bulk})$ and $E_{\text{tot}}(D^q)$ are the total energies of a bulk supercell and a supercell containing the defect D^q , respectively. In the third term on the right-hand side, μ_i and n_i are the chemical potential and number of atoms i added to the supercell, respectively. E_F is the Fermi level, and E_{corr} is a correction term to account for the artificial electrostatic interaction due to periodic boundary conditions.^{33,34} The formation energy is a function of the Fermi level, while the Fermi level is determined by the concentration of charged defects. Thus, we calculated the equilibrium concentration of defects and the Fermi level self-consistently, under the constraint of charge neutrality condition for overall system of defects and charge carriers, using the SC-FERMI code.³⁵

Results and discussion

Equilibrium phase diagram

A challenge to achieving high efficiency from kesterite thin-film solar cell is to synthesize homogeneous CZTS without unintentional formation of secondary phases.^{15,36–39} The thermodynamic chemical potential μ of each element depends on the growth environment including partial pressures and temperature. We compare the DFT total energies of CZTS and its competing phases in the chemical potential space (Fig. 1(b)), showing the range of chemical potentials that favors the formation of CZTS, using CPLAP.⁴⁰ The narrow range and complex shape of the phase diagram implies that it is hard to get a single-phase and homogeneous CZTS sample without the secondary phases. Even ‘pure’ CZTS is expected to contain an equilibrium population of point defects whose concentrations are controlled by the chemical potentials. We calculate the formation energies of the native defects under S-poor and S-rich conditions depicted in the phase diagram (Fig. 1(b)).

S-poor growth environment

Under S-poor conditions, which could be realized by annealing in a low sulfur partial pressure, the most dominant native defects are Cu_{Zn} and Zn_{Cu} antisites which are shallow and responsible for the p-type behaviour with a Fermi level close to the valence band (see Fig. 3(a)). At the Fermi level of 0.22 eV determined self-consistently, we predict high concentrations of V_S ($1.3 \times 10^{16} \text{ cm}^{-3}$), $V_S\text{-Cu}_{\text{Zn}}$ ($3.0 \times 10^{17} \text{ cm}^{-3}$) and Sn_{Zn} ($1.1 \times 10^{18} \text{ cm}^{-3}$). Here, we assume the growth and annealing temperature of 853 K resulting in defect populations. The



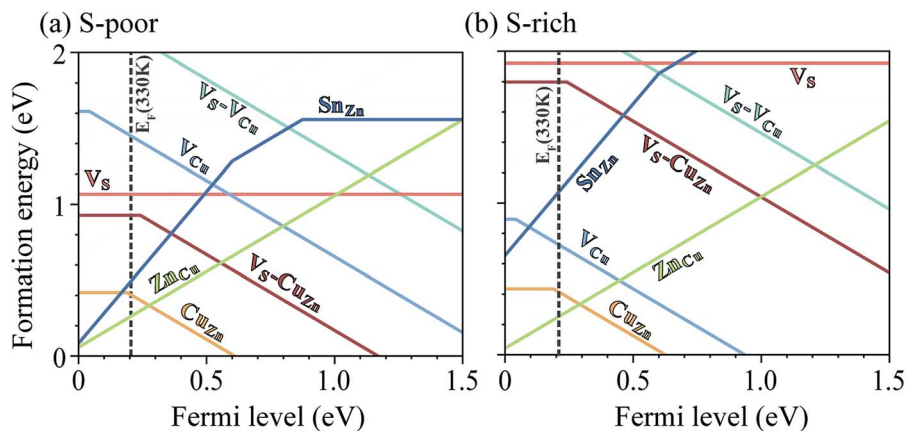


Fig. 3 Calculated formation energies of native defects in CZTS (a) under S-poor conditions and (b) under S-rich conditions. The self-consistent Fermi level resulting from the equilibrium defect population is shown as a vertical black dashed line. The top of valence band is set to 0 eV, while the bottom of conduction band is 1.46 eV.

operating temperature of a solar cell is a complex function of a level of irradiation, wind speed, humidity, and mounting type.⁴¹ We assumed a typical operating temperature of 330 K to equilibrate the Fermi level.

Previously, we have shown that V_S can act as an efficient non-radiative recombination center in CZTS.⁴² However, for electron capture, V_S needed to be activated. Firstly, as the ground state of V_S^0 involving Sn(II) is neutral and produces a state resonant within the valence band, thermal excitation is required to access

V_S^+ . As shown in Fig. 4a, the hole capture barrier for V_S^+ is so high that the thermal motion can not overcome it. Instead, the optical absorption can trigger the vertical transition from V_S^+ to V_S^{2+} , which corresponds to Sn(III) to Sn(IV) oxidation.

Here, we find that V_S can also be activated by forming a defect complex with Cu_{Zn} . In $(V_S-Cu_{Zn})^0$, the electronic wave function is localized around the Sn 5s lone-pair orbital similar to that of V_S^{1+} (Fig. 5(a) and (b)), suggesting that the ionized acceptor Cu_{Zn}^{1-} ionizes the neutral donor V_S^0 . Thus, Sn(III)

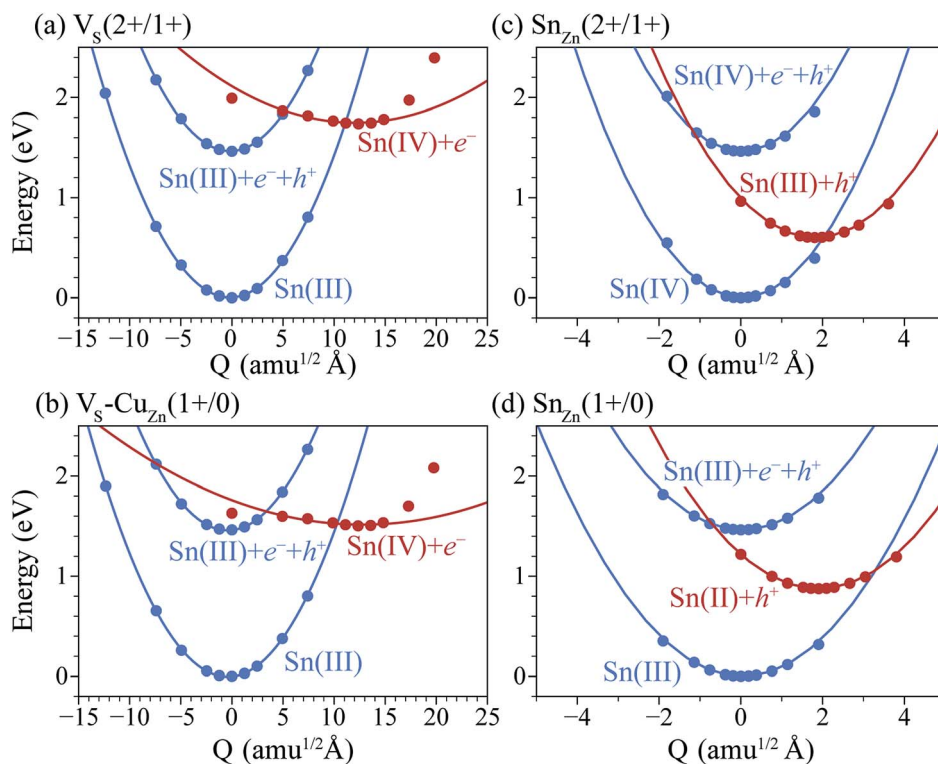


Fig. 4 Configuration coordinate diagrams for (a) V_S (2+/1+), (b) V_S-Cu_{Zn} (0/1+), (c) Sn_{Zn} (2+/1+) and (d) Sn_{Zn} (1+/0). The dot represents the formation energy calculated by DFT, and the line is a quadratic fit to the change in energy as the structure is distorted along the configuration coordinate. Q defines a pathway between the minimum-energy structure for each charge state.



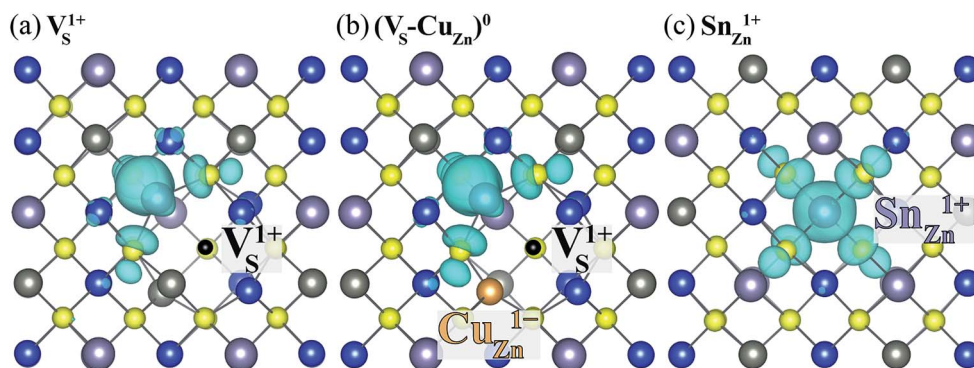


Fig. 5 Charge density of the lowest unoccupied Kohn–Sham orbitals associated with (a) V_S^{1+} , (b) neutral $(V_S-Cu_{Zn})^0$ and (c) Sn_{Zn}^{1+} . The black and orange balls represent V_S and Cu_{Zn} , respectively.

becomes the ground-state electronic configuration in the neutral V_S-Cu_{Zn} complex, indicating, unlike the isolated V_S , thermal excitation is not necessary.

We further find that optical excitation is not required for hole capture by the V_S-Cu_{Zn} complex. As a stronger Coulomb force binds the negatively charged acceptor Cu_{Zn}^{1-} , the formation energy difference between $Sn_{(III)}$ and $Sn_{(IV)}$ is reduced in the V_S-V_{Cu} complex (Fig. 4(b)). Accordingly, the reduced barrier for hole capture facilitates carrier recombination *without* optical excitation. The subsequent electron capture process will be fast due to the negligible energy barrier (see Fig. 4(b)). The V_S-V_{Cu} complex shows similar behavior, but its concentration is low under standard growth conditions.

Activation by passivation. It has been suggested that donor–acceptor complexes passivate deep donors in kesterite CZTS⁴³ and chalcopyrite CIGS,⁴⁴ which make them more tolerant to defects. However, we show that the neutral donor V_S^0 , which is deactivated by double Sn reduction, can be reactivated by forming complexes with the ionized acceptor Cu_{Zn}^{1-} and thus become an efficient recombination center. This is partially because the dominant defect–defect interaction is the classical Coulomb attraction instead of quantum mechanical level repulsion as is often considered.⁴⁵

We also examine recombination pathways *via* the donor levels of Sn_{Zn} . Fig. 5(c) shows the defect charge density of Sn_{Zn} is well localized around the Sn lone pair, suggesting the transitions involving Sn reduction and oxidation could trigger the carrier recombination similar to those in V_S and V_S-Cu_{Zn} . The

recombination path involving the double donor level $Sn_{Zn}(2+/1+)$ has a relatively high electron capture barrier of 0.23 eV (Fig. 4(c) and Table 1). On the other hand, $Sn_{Zn}(1+/0)$ – corresponding to the transition between $Sn_{(III)}$ and $Sn_{(II)}$ – has a smaller energy barrier of 0.05 eV, implying a faster recombination process.

In Fig. 6(a), we present the capture cross section calculated within the static coupling approximation.²² V_S-V_{Cu} and Sn_{Zn} can be classified as a giant electron trap whose electron capture cross section ($\sim 10^4 \text{ \AA}^2$) far exceeds the size of its atomic structure.⁴⁸ The calculated capture cross sections of the *native* defects in CZTS are orders of magnitudes larger than *extrinsic* transition metal impurities in silicon solar cells including Ti, V, Cr, Mo, Fe, Au and Zn whose cross sections range from 10^{-1} \AA^2 to 10^3 \AA^2 .^{46,47} This analysis suggests that V_S-Cu_{Zn} and Sn_{Zn} are the main sources of non-radiative recombination that limit the efficiency of CZTS solar cells (see Table 1). Note that due to the small energy barrier of V_S-Cu_{Zn} , the recombination is expected to be fast even at low temperature. At high temperature, the slight decrease in the capture cross section is attributed to the high Landau–Zener velocity;⁴⁸ the faster the defect level crosses the conduction bands, the less likely the defect captures electrons. The calculated capture cross section of Sn_{Zn}^{1+} is an order of magnitude higher than that of Sn_{Zn}^{2+} (Fig. 6 and Table 1).

In an operating solar cell, the recombination rate due to Sn_{Zn} may depend on the spatial position because the distance from the interface between CZTS and CdS determines the Fermi level (electronic band bending) and, hence, the charge state of Sn_{Zn} .

Table 1 Equilibrium point defect concentrations (N_T), carrier capture cross section at 330 K ($\sigma_{n/p}$) for electrons (n) and holes (p), thermal activation energy (E_t) and capture barrier (E_b) of V_S-Cu_{Zn} (0/+), Sn_{Zn}^{2+} (2+/1+), Sn_{Zn}^{1+} (1+/0), and Cu_{Sn} (2–/1–) charge transitions. The thermal velocities of electron and hole are 2.9×10^7 and $1.9 \times 10^7 \text{ cm s}^{-1}$, respectively. The Shockley–Read–Hall coefficient ($A = R/\Delta n$) is calculated for an excess carrier concentration $\Delta n = 1 \times 10^{14} \text{ cm}^{-3}$

Defect	$N_T (\text{cm}^{-3})$		$\sigma (\text{cm}^2)$		$E_t (\text{eV})$	$E_b (\text{meV})$		$A (\text{s}^{-1})$	
	S-poor	S-rich	n	p		n	p	S-poor	S-rich
$V_S-Cu_{Zn} (+/0)$	3.0×10^{17}	1.2×10^{12}	1.5×10^{-12}	8.4×10^{-14}	0.24	9	190	9.8×10^{12}	3.81×10^7
$Sn_{Zn} (2+/1+)$	1.1×10^{18}	3.3×10^{14}	1.5×10^{-14}	3.2×10^{-15}	0.60	230	5	4.8×10^{11}	1.4×10^8
$Sn_{Zn} (1+/0)$	1.8×10^{12}	6.7×10^8	5.5×10^{-13}	5.5×10^{-15}	0.87	54	184	2.7×10^7	1.0×10^4
$Cu_{Sn} (2–/1–)$	2.7×10^8	6.7×10^{11}	2.6×10^{-34}	1.3×10^{-16}	0.31	1693	427	2.0×10^{-18}	4.9×10^{-15}



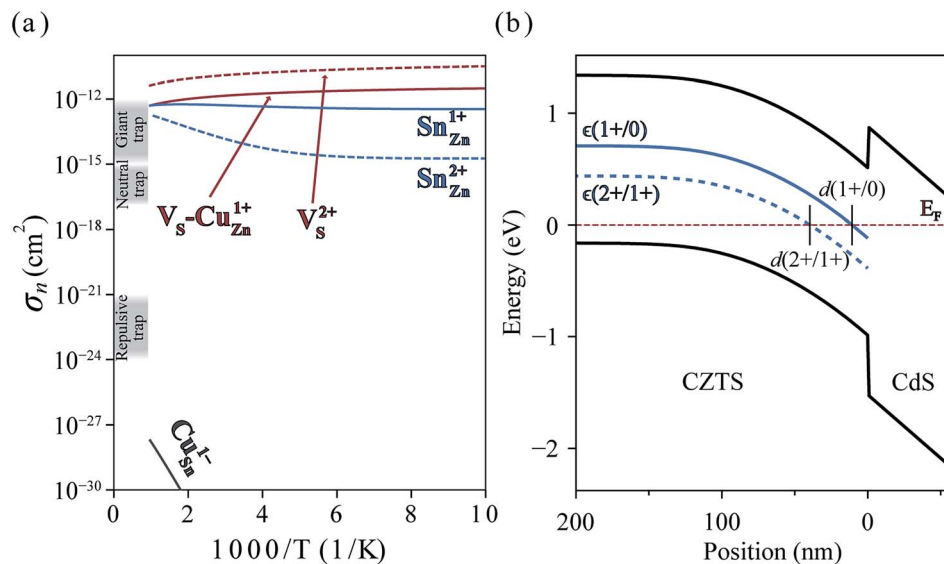


Fig. 6 (a) Electron capture cross-sections of V_S , V_S-Cu_{Zn} complex, Sn_{Zn} , and Cu_{Sn} . Gray shades represent the typical orders of magnitude of cross sections of giant, neutral and repulsive traps.¹⁸ (b) Band diagram of CZTS/CdS heterojunction. The solid and dashed blue curves represent the single $\epsilon(1+/0)$ and double donor levels $\epsilon(2+/1+)$ of Sn_{Zn} , respectively. $d(q_1/q_2)$ is defined by the distance from the interface where the charge transition levels of Sn_{Zn} $\epsilon(q_1/q_2)$ equals the Fermi level E_F (red dashed line). The band diagram and $d(2+/1+) = 40$ nm are obtained by solving the Poisson–Boltzmann equation using https://pythonhosted.org/eq_band_diagram.

In the undepleted region ($d > d(2+/1+)$) in Fig. 6(b), most of Sn_{Zn} is in the form of +2 charge state which is a slower recombination channel. However, in the depletion region ($d < d(2+/1+)$), Sn_{Zn} favors a +1 charge state which has much larger capture cross section. In this case, a recombination pathway is activated by band bending in a photovoltaic device.

S-rich growth environment

Under S-rich conditions, the formation of V_S and V_S-Cu_{Zn} is strongly suppressed (see Fig. 3(b)). We associate this with the experimentally observed increase in V_{OC} under a high S partial pressure during the annealing of a photovoltaic device.⁴⁹ However, even under S-rich conditions, a considerable concentration of Sn_{Zn} is still expected, which can limit the lifetime of carriers to below 7.1 ns (see Table 1). This shows good agreement with the reported photoluminescence (PL) decay times of kesterite materials which range from 1 ns to 10 ns.^{9,50} Moreover, the electron paramagnetic resonance (EPR) signal in CZTS⁵¹ supports the existence partially oxidized $Sn(III)$ with an unpaired electron ($5s^1$), which is the active state in the proposed recombination pathways.

Inert-pair effect

The heavy post-transition metals (elements in groups 13, 14, 15 and 16) often exhibit oxidation states two less than the group valency, referred to as the inert-pair effect. The inert-pair effect is explained by the insufficient screening by d^{10} electrons resulting in the s^2 lone-pair electrons tightly bound to the ion.^{52–55} However, the role of the inert-pair effect on the properties of defects in semiconductors has not been fully explored.

Deep defect nature. We find that the inert-pair effect of Sn produces deep defects, consistent with the previous theoretical studies.^{56,57} The ability of Sn to accommodate excess charges stabilizes the neutral state over the ionized state. In a multi-valent compound, such as CZTS, the variation of Madelung potential between cation sites with formal +1, +2 and +4 oxidation states favours particular defect charge states, so defect ionization is suppressed relative to elemental or binary semiconductors.

Large lattice distortion. Sn also produces defects with large distortions during carrier capture. Electron addition or removal from V_S , V_S-Cu_{Zn} and Sn_{Zn} are followed by the oxidation or reduction of Sn and are therefore accompanied by large structure change. We find, in the Sn-related defects in CZTS, large lattice distortion quantified by a Huang–Rhys factor $S \gg 1$.¹⁹ Especially, in V_S and V_S-Cu_{Zn} , a radiative transition pathway is impossible due to the very large lattice distortion where the minimum of the excited state ($Sn(IV)$) is located outside of the potential energy surface of the ground-state ($Sn(III)$).⁵⁸

Thus, the inert-pair effect in Sn is responsible for both the deep charge transition levels and the large lattice distortion of V_S , V_S-Cu_{Zn} and Sn_{Zn} , which make them efficient non-radiative recombination centers. Similarly, we find the deep acceptor level of Cu_{Sn} owing to the oxidation of Cu. However, the electron capture rates by Cu_{Sn} ($1-/2-$) are low (Table 1). The multi-valency of Cu is due to the change in the occupation of 3d orbital: from $3d^{10}$ in Cu(I) to $3d^9$ in Cu(II). Thus, the local relaxation after the oxidation is not significant. The small lattice distortion in Cu_{Sn} produces an electron capture barrier above 1.6 eV, making capture unlikely (Fig. 6(a)).

Emergence of deep defects induced by the formation of lone pairs has also been reported in CIGS. Extrinsic dopants of Bi



and Sb are deep in CuInSe₂ due to the lone-pair s² states.⁵⁹ Han *et al.*⁶⁰ have also found that the formation of lone pairs in amorphous oxide semiconductors, such as InGaZnO₄ and ZnSnO₃, is responsible for electron trapping.

While Sn reduction captures electrons, in lone-pair compounds whose cations, such as Tl(I), Pb(II) and Bi(III), have occupied s² orbitals in the stoichiometric compounds, the oxidation of cations could capture holes. Several EPR measurements show that Pb(III) is responsible for the hole traps in lead halides.^{61–64}

The efficiency and lifetime of optoelectronic devices can be severely damaged by a defect with fast non-radiative recombination. Stoneham¹⁸ suggests several characteristics of such *killer* centers, including: (1) defects producing many and closely spaced electronic levels; (2) defects with large lattice distortions. The first type can be directly related to the transition metal impurities with partially filled d orbitals (*e.g.* Ni in GaP). While a simple vacancy center was suggested as a candidate for the second-type,¹⁸ a wide variety of vacancies are not recombination centers in photovoltaic materials (*e.g.* V_{Cu} is a shallow acceptor in CZTS, and V_I is a shallow donor in CH₃NH₃PbI₃.) Because of the strong interaction between impurities and host materials, it is difficult to find a general trend of the properties of defects in the absence of detailed calculations. On the other hand, we find that the inert lone-pair of Sn is the origin of the large cross-section of a wide range of defects and not significantly altered by a specific configuration or an electronic state of the defect. Thus, we speculate that the inert-pair effect could likely cause *killer* centers with the ability to act as giant carrier traps in a broad range of semiconductors.

Many photovoltaic materials with band gaps close to the theoretical optimum of 1.3–1.5 eV (ref. 13) show poor performances; in particular a low V_{OC}. The record efficiency of Cu₂SnS₃, whose band gap is around 1 eV, is still low (4.63%) even with the high current density J_{SC} of 37.3 mA cm⁻² mainly because of low V_{OC} of 283 mV.⁶⁵ The carrier lifetime of Cu₂SnS₃ was reported to be very short (0.1–10 ps).⁶⁶ Sb₂Se₃ solar cells also exhibit low V_{OC} with short carrier lifetime of 1.3 ns.⁶⁷ The first light-to-electricity conversion efficiency of Cu₃BiS₃ solar cell (0.17%) has been achieved only recently with V_{OC} of 190 meV,⁶⁸ and, to our knowledge, a successful operation of CuBiS₂ solar cell has not been reported. The presence of lone-pair cations is a common feature in these technologies.

Conclusion

The lone-pair effect associated with Sn is responsible for both the deep defect levels and large lattice distortions that facilitate rapid electron–hole recombination in the semiconductor Cu₂-ZnSnS₄. By employing a first-principles approach to predict the defect levels, concentrations, and capture rates, we can distinguish between active and inactive defect centres. For a material grown under S-poor conditions, V_S, V_S-Cu_{Zn} and Sn_{Zn} act as dominant recombination centers, while Sn_{Zn} limits the minority carrier lifetime under S-rich conditions. We propose that a similar mechanism could underpin the low performance of other emerging photovoltaic compounds. We emphasise the

need for further experimental and theoretical investigation of defects in semiconductors composed of heavy post-transition metals to further evaluate the *general* role of the inert-pair effect on the non-radiative electron–hole recombination process.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

We acknowledge support from the Royal Society, the EPSRC (Grant No. EP/K016288/1), and the EU Horizon2020 Framework (STARCELL, Grant No. 720907). We are grateful to the UK Materials and Molecular Modeling Hub for computational resources, which is partially funded by EPSRC (EP/P020194/1). *Via* our membership of the UK's HEC Materials Chemistry Consortium, which is funded by EPSRC (EP/L000202), this work used the ARCHER UK National Supercomputing Service (<http://www.archer.ac.uk>). This work was also supported by a National Research Foundation of Korea (NRF) grant funded by the Korean government (MSIT) (No. 2018R1C1B6008728).

Notes and references

- 1 J. Nelson, *The Physics of Solar Cells*, Imperial College Press, London, UK, 2003.
- 2 J.-S. Park, S. Kim, Z. Xie and A. Walsh, *Nat. Rev. Mater.*, 2018, **50**, 797.
- 3 A. M. Stoneham, *Rep. Prog. Phys.*, 2000, **44**, 1251.
- 4 T. Kirchartz and U. Rau, *Adv. Energy Mater.*, 2018, **356**, 1703385.
- 5 M. Kumar and C. Persson, *Int. J. Theor. Appl. Sci.*, 2013, **5**, 1.
- 6 M. Kumar, A. Dubey, N. Adhikari, S. Venkatesan and Q. Qiao, *Energy Environ. Sci.*, 2015, **8**, 3134.
- 7 K. Kaur, N. Kumar and M. Kumar, *J. Mater. Chem. A*, 2017, **5**, 3069.
- 8 W. Wang, M. T. Winkler, O. Gunawan, T. Gokmen, T. K. Todorov, Y. Zhu and D. B. Mitzi, *Adv. Energy Mater.*, 2013, **4**, 1301465.
- 9 C. Yan, J. Huang, K. Sun, S. Johnston, Y. Zhang, H. Sun, A. Pu, M. He, F. Liu, K. Eder, L. Yang, J. M. Cairney, N. J. Ekins-Daukes, Z. Hameiri, J. A. Stride, S. Chen, M. A. Green and X. Hao, *Nat. Energy*, 2018, **1**, 15015.
- 10 D. A. R. Barkhouse, O. Gunawan, T. Gokmen, T. K. Todorov and D. B. Mitzi, *Prog. Photovolt: Res. Appl.*, 2011, **20**, 6.
- 11 T. K. Todorov, J. Tang, S. Bag, O. Gunawan, T. Gokmen, Y. Zhu and D. B. Mitzi, *Adv. Energy Mater.*, 2012, **3**, 34.
- 12 L. Grenet, M. A. A. Suzon, F. Emieux and F. Roux, *ACS Appl. Energy Mater.*, 2018, **1**, 2103.
- 13 W. Shockley and H. J. Queisser, *J. Appl. Phys.*, 1961, **32**, 510.
- 14 S. Rühle, *Sol. Energy*, 2016, **130**, 139.
- 15 S. K. Wallace, D. B. Mitzi and A. Walsh, *ACS Energy Lett.*, 2017, **2**, 776.
- 16 R. N. Hall, *Phys. Rev.*, 1952, **87**, 387.
- 17 W. Shockley and J. W. T Read, *Phys. Rev.*, 1952, **87**, 835.



- 18 A. M. Stoneham, *Theory of Defects in Solids*, Oxford University Press, Oxford, UK, 1975.
- 19 K. Huang and A. Rhys, *Proc. R. Soc. A*, 1950, **204**, 406.
- 20 C. H. Henry and D. V. Lang, *Phys. Rev. B*, 1977, **15**, 989.
- 21 L. Shi and L.-W. Wang, *Phys. Rev. Lett.*, 2012, **109**, 245501.
- 22 A. Alkauskas, Q. Yan and C. G. Van de Walle, *Phys. Rev. B*, 2014, **90**, 075202.
- 23 A. Alkauskas, M. D. McCluskey and C. G. Van de Walle, *J. Appl. Phys.*, 2016, **119**, 181101.
- 24 R. Pässler, *Phys. Status Solidi B*, 1976, **78**, 625.
- 25 P. T. Landsberg, *Recombination in Semiconductors*, Cambridge University Press, Cambridge, UK, 2009.
- 26 P. Hohenberg and W. Kohn, *Phys. Rev.*, 1964, **136**, B864.
- 27 W. Kohn and L. J. Sham, *Phys. Rev.*, 1965, **140**, A1133.
- 28 P. E. Blöchl, *Phys. Rev. B*, 1994, **50**, 17953.
- 29 J. Heyd, G. E. Scuseria and M. Ernzerhof, *J. Chem. Phys.*, 2003, **118**, 8207.
- 30 G. Kresse and D. Joubert, *Phys. Rev. B*, 1999, **59**, 1758.
- 31 H. J. Monkhorst and J. D. Pack, *Phys. Rev. B*, 1976, **13**, 5188.
- 32 C. Freysoldt, B. Grabowski, T. Hickel, J. Neugebauer, G. Kresse, A. Janotti and C. G. Van de Walle, *Rev. Mod. Phys.*, 2014, **86**, 253.
- 33 C. Freysoldt, J. Neugebauer and C. G. Van de Walle, *Phys. Rev. Lett.*, 2009, **102**, 016402.
- 34 Y. Kumagai and F. Oba, *Phys. Rev. B*, 2014, **89**, 195205.
- 35 SC-FERMI, <https://github.com/jbuckeridge/sc-fermi>, accessed: 2019-01-03.
- 36 J. J. S. Scragg, P. J. Dale, L. M. Peter, G. Zoppi and I. Forbes, *Phys. Status Solidi B*, 2008, **245**, 1772.
- 37 S. Chen, X. G. Gong, A. Walsh and S.-H. Wei, *Appl. Phys. Lett.*, 2010, **96**, 021902.
- 38 A. J. Jackson and A. Walsh, *J. Mater. Chem. A*, 2014, **2**, 7829.
- 39 M. Dimitrievska, A. Fairbrother, E. Saucedo, A. Perez-Rodriguez and V. Izquierdo-Roca, *Sol. Energy Mater. Sol. Cells*, 2016, **149**, 304.
- 40 J. Buckeridge, D. O. Scanlon, A. Walsh and C. R. A. Catlow, *Comput. Phys. Commun.*, 2014, **185**, 330.
- 41 E. Skoplaki, A. G. Boudouvis and J. A. Palyvos, *Sol. Energy Mater. Sol. Cells*, 2008, **92**, 1393.
- 42 S. Kim, J.-S. Park and A. Walsh, *ACS Energy Lett.*, 2018, **3**, 496.
- 43 S. Chen, J.-H. Yang, X. G. Gong, A. Walsh and S.-H. Wei, *Phys. Rev. B*, 2010, **81**, 245204.
- 44 S. B. Zhang, S.-H. Wei, A. Zunger and H. Katayama-Yoshida, *Phys. Rev. B*, 1998, **57**, 9642.
- 45 A. Walsh and A. Zunger, *Nat. Mater.*, 2017, **50**, 797.
- 46 D. Macdonald and L. J. Geerligs, *Appl. Phys. Lett.*, 2004, **85**, 4061.
- 47 A. R. Peaker, V. P. Markevich, B. Hamilton, G. Parada, A. Dudas, A. Pap, E. Don, B. Lim, J. Schmidt, L. Yu, Y. Yoon and G. Rozgonyi, *Phys. Status Solidi A*, 2012, **209**, 1884.
- 48 C. Zener, *Proc. R. Soc. A*, 1932, **137**, 696.
- 49 Y. Ren, N. Ross, J. K. Larsen, K. Rudisch, J. J. S. Scragg and C. Platzer-Björkman, *Chem. Mater.*, 2017, **29**, 3713.
- 50 C. J. Hages, A. Redinger, S. Levchenko, H. Hempel, M. J. Koeper, R. Agrawal, D. Greiner, C. A. Kaufmann and T. Unold, *Adv. Energy Mater.*, 2017, **7**, 1700167.
- 51 C. Chory, F. Zutz, F. Witt, H. Borchert and J. Parisi, *Phys. Status Solidi C*, 2010, **7**, 1486.
- 52 N. V. Sidgwick and H. M. Powell, *Proc. R. Soc. London, Ser. A*, 1940, **176**, 153.
- 53 R. J. Gillespie and R. S. Nyholm, *Q. Rev., Chem. Soc.*, 1957, **11**, 339.
- 54 L. Shimoni-Livny, J. P. Glusker and C. W. Bock, *Inorg. Chem.*, 1998, **37**, 1853.
- 55 A. Walsh, D. J. Payne, R. G. Egdell and G. W. Watson, *Chem. Soc. Rev.*, 2011, **40**, 4455.
- 56 K. Biswas, S. Lany and A. Zunger, *Appl. Phys. Lett.*, 2010, **96**, 201902.
- 57 D. Han, Y. Y. Sun, J. Bang, Y. Y. Zhang, H.-B. Sun, X.-B. Li and S. Zhang, *Phys. Rev. B*, 2013, **87**, 155206.
- 58 D. L. Dexter, C. C. Klick and G. A. Russell, *Phys. Rev.*, 1955, **100**, 603.
- 59 J.-S. Park, J.-H. Yang, K. Ramanathan and S.-H. Wei, *Appl. Phys. Lett.*, 2014, **105**, 243901.
- 60 W. H. Han and K. J. Chang, *Phys. Rev. Appl.*, 2016, **6**, 044011.
- 61 M. Iwanaga, M. Shirai, K. Tanaka and T. Hayashi, *Phys. Rev. B*, 2002, **66**, 064304.
- 62 I. A. Shkrob and T. W. Marin, *J. Phys. Chem. Lett.*, 2014, **5**, 1066.
- 63 D. Cortecchia, J. Yin, A. Bruno, S.-Z. A. Lo, G. G. Gurzadyan, S. Mhaisalkar, J.-L. Brédas and C. Soci, *J. Mater. Chem. C*, 2017, **5**, 2771–2780.
- 64 S. Colella, M. Todaro, S. Masi, A. Listorti, D. Altamura, R. Caliendo, C. Giannini, E. Carignani, M. Geppi, D. Meggiolaro, G. Buscarino, F. De Angelis and A. Rizzo, *ACS Energy Lett.*, 2018, **3**, 1840–1847.
- 65 M. Nakashima, J. Fujimoto, T. Yamaguchi and M. Izaki, *Appl. Phys. Express*, 2015, **8**, 042303.
- 66 L. L. Baranowski, K. McLaughlin, P. Zawadzki, S. Lany, A. Norman, H. Hempel, R. Eichberger, T. Unold, E. S. Toberer and A. Zakutayev, *Phys. Rev. Appl.*, 2015, **4**, 044017.
- 67 X. Wen, C. Chen, S. Lu, K. Li, R. Kondrotas, Y. Zhao, W. Chen, L. Gao, C. Wang, J. Zhang, G. Niu and J. Tang, *Nat. Commun.*, 2018, **9**, 2179.
- 68 J. Li, X. Han, Y. Zhao, J. Li, M. Wang and C. Dong, *Sol. Energy Mater. Sol. Cells*, 2018, **174**, 593.

