The intimate relationship between the dielectric response and the decay of intermolecular correlations and surface forces in electrolytes†

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A general, exact theory for the decay of interactions between any particles immersed in electrolytes, including surface forces between macroscopic bodies, is derived in a self-contained, physically transparent manner. It is valid for electrolytes at any density, including ionic gases, molten salts, ionic liquids, and electrolyte solutions with molecular solvent at any concentration. The ions, the solvent and any other particles in the system can have any sizes, any shapes and arbitrary internal charge distributions. The spatial propagation of the interactions in electrolytes has several decay modes with different decay lengths that are given by the solutions, \( k_\nu \) \( \nu = 1, 2, \ldots \), to a general equation for the screening parameter \( k_\nu \): an equation that describes the dielectric response. There can exist simultaneous decay modes with plain exponential decay and modes with damped oscillatory exponential decay, as observed experimentally and theoretically. In the limit of zero ionic density, the decay length \( 1/k_\nu \) of the mode with the longest range approaches the Debye length \( 1/k_D \). The coupling between fluctuations in number density and charge density, described by the density–charge correlation function \( H_{\text{inv}}(r) \), makes all decay modes of pair correlations and interaction free energies identical to those of the screened electrostatic potential, and hence they have the same values for the screening parameters. The density–density and charge–charge correlation functions, \( H_{\text{NN}}(r) \) and \( H_{\text{QQ}}(r) \), also have these decay modes. For the exceptional case of charge-inversion invariant systems, \( H_{\text{QQ}}(r) \) is identically zero for symmetry reasons and \( H_{\text{NN}}(r) \) and \( H_{\text{QQ}}(r) \) have, instead, decay modes with different decay lengths.

1 Introduction

Electrolytes are prevalent in various systems of great importance in physics, chemistry, biosciences, surface and colloid sciences and many applied fields, including industrial applications. Such systems have been studied theoretically and experimentally for a very long time, but recently some unexpected experimental observations have been made, in particular the existence of long-range interactions in systems with high ion densities like ionic liquids and concentrated electrolyte solutions. These observations have renewed the interest in the basics of interactions in electrolytes.

Oscillatory pair distribution functions and, consequently, oscillatory free energy of intermolecular interaction (potential of mean force) are well-established features of dense electrolytes like molten salts. The recent discovery of monotonic, exponentially decaying long-range forces with decay lengths of 4–11 nm in ionic liquids was therefore quite surprising, not least because traditional theories of electrolytes give screening lengths that are orders of magnitude shorter under the conditions in question. Such forces with long decay lengths have been observed for various systems with high ionic densities, like ionic liquids and concentrated electrolyte solutions. Simultaneously, there are often oscillatory contributions with shorter decay lengths simultaneously present in the force curves for such systems.

These observations are conceptually important because exponentially decaying forces between surfaces in electrolyte solutions have often been taken as an experimental verification of the correctness of the Poisson–Boltzmann (PB) approximation for surface interactions, which predicts such forces for large distances. The PB expressions that are commonly used contain adjustable parameters like surface charge density, surface potential etc. that can be fitted to the experimental data. However, any reasonable theory gives exponentially decaying forces in electrolytes at least for low ionic densities and exact analysis of statistical mechanics of fluids says that the forces must be exponential under such conditions – apart from contributions due to dispersion interactions that have power law decay and therefore ultimately must dominate the forces for sufficiently large distances. Therefore the mere existence of an exponential decay says nothing about the validity of the PB approximation.
In this connection we should note that an exponential decay of the interactions in planar geometry, i.e., proportional to \( e^{-\kappa r} \), where \( r \) is the distance and \( \kappa \) is the screening parameter, translates for spherical geometry into a Yukawa function decay, \( e^{-\kappa r}/r \), where \( r \) is the radial distance. Furthermore, it is a statistical mechanical fact that the exponential decay of surface forces for large separations has the same parameter \( \kappa \) as the pair correlations in the bulk fluid in equilibrium with the fluid in the slit between the surfaces. The same applies for exponentially damped oscillatory forces.

Another test of whether the PB approximation is applicable or not is the magnitude of the decay length, which in this approximation is given by the Debye length \( 1/\kappa_0 \) where the Debye screening parameter \( \kappa_0 = \kappa \) is defined for electrolyte solutions from

\[
\kappa_0^2 = \frac{\beta}{\varepsilon_0 \varepsilon_0} \sum_j q_j^2 n_j^b \quad \text{(ions in dielectric continuum),}
\]

where \( \beta = (k_B T)^{-1} \), \( k_B \) is Boltzmann’s constant, \( T \) is the absolute temperature, \( \varepsilon_0 \) is the dielectric constant of the pure solvent modeled as a dielectric continuum (this is assumed in the primitive model of electrolyte solutions), \( \varepsilon_0 \) is the permittivity of vacuum, \( q_j \) is the charge of an ion of species \( j \), and \( n_j^b \) is the number density in the bulk phase (superscript \( b \) stands for “bulk”). In the PB approximation one entirely neglects ion–ion correlations in the electrolyte adjacent to the object one considers. These ions are treated as point charges that do not correlate with each other. This applies also when the object one considers is an ion of the electrolyte itself (this is done when one deals with pair correlations in electrolytes), which means that this ion is treated in a different manner than the other ions of the same species. In a correct theory all ions should be treated on the same basis.

The PB approximation is quite accurate for low ion densities. The actual decay length for the electrolyte approaches \( 1/\kappa_0 \) when the ion density goes to zero and the importance of the ion–ion correlations decreases. A pertinent question is how low the ion density must be for the correlations to be unimportant. In fact, long-range electrostatic ion–ion correlations can give substantial deviation from the Debye length also for low ion densities provided the electrostatic coupling is sufficiently strong, for instance for multivalent ions in dilute aqueous solution as we will see later.

Furthermore, in contrast to the PB prediction of monotonic exponential decay with a single decay length, there appear in general more than one decay length and/or exponentially damped oscillatory decay. This can be illustrated by a simple approximation\(^{11,12} \) that is valid for simple ionic fluids with rather low electrostatic coupling, for example ionic fluids at very high temperatures or electrolyte solutions in the primitive model at high \( \varepsilon_0 \) for the dielectric continuum solvent; typical examples are monovalent ions in aqueous solution at room temperature and such ions in vacuum at \( T = 23 400 \text{ K} \) (these systems have the same value of \( \varepsilon_0 T \)). For a symmetric electrolyte with ionic diameter \( d \) this approximation gives the following expression for the screening parameter \( \kappa \)

\[
\frac{\kappa^2}{\kappa_0^2} = \frac{\phi^{cd}}{1 + \kappa d}
\]

which can be derived in a simple manner by making a very small extension of the linearized PB (LPB) approximation, whereby one requires that all ions in the systems should be treated on the same basis.\(^\dagger\) This expression can be written as

\[
\left( \frac{\kappa}{\kappa_0} \right)^2 = \frac{q^2 / \kappa_0^2}{1 + (\kappa/\kappa_0)^2 \kappa d}
\]

so it is an equation for \( \kappa/\kappa_0 \) as a function of \( \kappa d \). In the limit \( \kappa d \to 0 \) this equation has a solution with the property \( \kappa/\kappa_0 \to 1 \), but there is also another solution that we will denote as \( \kappa^* \). These solutions are shown by the curves marked by \( \kappa \) and \( \kappa^* \) in Fig. 1 and we see that when \( \kappa d \) is increased, the two solutions approach each other and at \( \kappa d = 1.35 \) they merge (one can show

\(^\dagger\) As shown in Section 3.2.4, for a binary symmetric electrolyte with ions of equal diameters we have the exact relationship \( \kappa^2/\kappa_0^2 = \phi^{cd} q [\text{eqn (101)}] \), where \( \phi^{cd} \) is an effective ionic charge and \( q = q_+ = -q_- \). Using the LPB approximation as a guide for an approximative value of \( \phi^{cd} \) for all ions (i.e., not only for the ion at the origin), we set \( \phi^{cd} = q^2 (1 + \kappa d) \) and obtain eqn (2); for details see Section 3.2.4.
that \( \kappa d = 1 + \sqrt{3} = 2.73 \) at this point), which for monovalent ions corresponds to a molar concentration of \( (4.10/d)^2 \) when \( d \) is measured in Å. For \( \kappa D > 1.35 \), the solutions are complex-valued; the two solutions are then complex conjugates to each other: \( \kappa = \kappa_R + i\kappa_I \) and \( \kappa' = \kappa_R - i\kappa_I \), where \( i \) is the imaginary unit. The appearance of the complex solutions, marked by a filled symbol in the figure, means that there is a cross-over from a plain exponential decay to an oscillatory exponential one, that is, proportional to \( e^{-\kappa R} \cos(\kappa I r + \gamma) \) in planar geometry and \( e^{-\kappa R} \cos(\kappa I r + \gamma)/r \) in spherical geometry, where \( \gamma \) is a phase shift. This behavior is well-known and is called a Kirkwood cross-over, named after John G. Kirkwood who already in the 1930s showed the existence of such a change to oscillatory behavior for electrolytes. Since then, several expressions for \( \kappa \) with the same behavior have been obtained, for example in the Mean Spherical Approximation (MSA), the Linearized Modified PB (LMPB) approximation by Outhwaite and the generalized MSA. This behavior has also been verified many times in the past in numerical calculations, including simulations.

The simple expression in eqn (2) is surprisingly accurate, considering its humble origin; in the figure its predictions are compared with the results from Monte Carlo (MC) simulations and Hypernetted Chain (HNC) calculations, and it is seen that the agreement is nearly quantitative. For the oscillatory decay, the decay length \( 1/\kappa_R \) increases quite rapidly with electrolyte concentration after the cross-over point. The wavelength \( 2\kappa R/\kappa_I \) immediately after the cross-over is very long (\( \kappa_R = 0 \) at the cross-over point), much longer than the decay length, so in practice the oscillations would not be observable. However, the wavelength decreases very rapidly with increased concentration and is quite soon comparable to the decay length. Therefore, for the forces between particles or between surfaces at increasing separations, some oscillations should be seen before their magnitude is too small.

The important points here are that there exist more than one decay mode, one with decay length \( 1/\kappa \) and one with \( 1/\kappa' \) and, furthermore, that the decay modes predicted by the equation for \( \kappa \) can be oscillatory. The objective of the simple expression (2) is to give a simple illustration of these facts. In the general case, other decay modes are also simultaneously present in electrolytes. These matters constitute the main theme of the current work. We will see how the decay modes can be treated in formally exact theory that is valid for much more complex systems. Such modes have a major role for the interactions in electrolytes for all conditions from low to high ionic densities and all temperatures (provided that the system remains fluid).

The approximation behind eqn (2) is not sufficient for higher electrostatic coupling than that of the example in the figure, i.e., lower temperatures, lower \( \varepsilon_i \) and higher ionic valencies. For divalent ions in aqueous solution at room temperature, results from HNC calculations and MC simulations show that the decay length at low ion densities is appreciably longer than the Debye length. In a plot as that in Fig. 1, the curve for \( \kappa/\kappa_D \) then lies below the value 1 for small \( \kappa_D d \); for example, for divalent ions with diameter \( d = 4.6 \) Å, \( \kappa/\kappa_D \) reaches a minimal value \( \approx 0.84 \) before it increases to values above 1, whereafter the qualitative behavior is similar to that for the monovalent case. In fact, the results from the HNC approximation show that \( \kappa/\kappa_D \) lies below the value 1 for very small \( \kappa_D d \) also for the monovalent ions, but this is not visible for monovalent electrolytes on the scale in Fig. 1.

It is quite significant that deviations in decay length from the Debye length occur not only for dense electrolytes. This is a fundamental property of electrolytes because in the limit of low ionic density, the ratio \( \kappa/\kappa_D \) for a \( z:z \) electrolyte (\( z \) is the valency) satisfies the exact limiting law:

\[
\left( \frac{\kappa}{\kappa_D} \right)^2 = 1 + \frac{A^2 \ln A}{6} \quad \text{when } A \to 0, \tag{4}
\]

where \( A = \kappa_D/\kappa \) is the coupling parameter, \( \kappa_D = \beta q_z^2/(4\pi \varepsilon_0 e_0) \) is the Bjerrum length, and \( q_z \) is the elementary (protonic) charge. The expression for \( \kappa_D \) used here is appropriate for the primitive model, otherwise \( \kappa_D = 1 \) both in \( \kappa_D \) and in eqn (1). The deviation of \( \kappa/\kappa_D \) from the value 1, as described by this law, is caused entirely by the long-range electrostatic correlations of the ions and is independent of other characteristics of the ions than their charges.\(^\S\) Note that in \( A \) is negative for small \( A \), so formula (4) shows that the decay length is larger than the Debye length for low electrolyte concentrations in agreement with the numerical theoretical results mentioned above.

This law was recently verified experimentally by Smith et al. for dilute 2:2 and 3:3 electrolytes in aqueous solution. Large negative deviations of \( \kappa/\kappa_D \) from the value 1 were observed for electrolyte concentrations in the interval 0.1–10 mM and the agreement with eqn (4) was nearly quantitative for a large part of this interval. For aqueous 1:1 electrolytes, however, the deviation from the value 1 was small.

The effect described by eqn (4) is not included in the approximation given in eqn (2) or in any other linear theory like MSA and LMPB, so \( \kappa/\kappa_D > 1 \) for small \( \kappa_D d \) in these approximations. However, if one includes nonlinear terms, one obtains agreement with eqn (4) in the limit of small \( A \). Since the HNC approximation is nonlinear, it is in agreement with eqn (4). The difference between \( \kappa/\kappa_D \) from eqn (2) and from the HNC approximation for small \( \kappa_D d \) is very small for the monovalent electrolyte in Fig. 1. Due to the factor of \( A^2 \) in eqn (4) the deviation from the Debye length increases rapidly with ionic valency for systems at low ion densities. For systems with much lower \( \varepsilon_i \) and/or lower temperatures, the deviation will be substantial also for monovalent ions.

\(^\S\) The general limiting law at high dilution for the decay parameter \( \kappa \) of the leading decay mode is from ref. 21:

\[
\left( \frac{\kappa}{\kappa_D} \right)^2 = 1 + \frac{A^2 \ln A}{4} \left[ \sum_{i} \frac{z_i^2}{\eta_i} \right]^2 + \frac{A^2}{6} \ln A \left[ \sum_{i} \frac{z_i^2}{\eta_i} \right] + \mathcal{O}(A^4),
\]

where \( z_i = q_i/q_z \) is the valency of species \( j \) and \( \eta_i \) is the stoichiometric coefficient.

The term proportional to \( A \), which was originally derived in ref. 22, vanishes for symmetric binary electrolytes. Both terms originate solely from the long-range, purely electrostatic correlations, while contributions that depend on other properties of the ions (like their sizes) affect the higher order terms, that here are included in \( \mathcal{O}(A^4) \) where \( A^2 \) is proportional to the total ion density.
Other very relevant illustrations of the subject that is studied formally in this work can be taken from the computer simulation work by Keblinski et al.,\textsuperscript{24} where they investigate NaCl for a large variety of conditions from thin gases to molten salt for various temperatures. They used an empirical, realistic model for the pair interaction potential for NaCl that has been shown to reproduce quite well the thermodynamic, structural and other properties of molten NaCl.

Keblinski and coworkers calculated the charge–charge and density–density correlation functions, $H_{QQ}(r)$ and $H_{NN}(r)$ respectively,\textsuperscript{8} and determined the decay lengths, wavelengths and other characteristics of the dominant contributions to these functions for various conditions. They found that the leading term of the asymptotic decay for large $r$ of the correlation functions can dominate also for shorter distances – in some cases down to a distance of one or two particle diameters. Their Fig. 1 in ref. 12 shows that in molten NaCl at 1000 K, the leading asymptotic term in $H_{QQ}(r)$, which decays like $e^{-kr_{00}/C_0}$, is practically indistinguishable from $H_{QQ}(r)$ down to $r \approx 2d$, where $d$ is the average ion diameter. Furthermore, it gives a very good description of $H_{QQ}(r)$ down to $r \approx d$. This means that an asymptotic analysis of the correlation function gives important information not only for very large $r$, but also for a quite wide range of shorter distances. It can in many cases be sufficient to include a couple of leading decay modes in the asymptotic analysis. Such a dominance of the leading asymptotic terms has also been found in earlier studies of other electrolyte systems.\textsuperscript{18,20,25}

The simulations by Keblinski and coworkers also included a study of the Kirkwood cross-over and they showed explicitly the presence of two exponentially decaying modes in in $H_{QQ}(r)$ with decay lengths $1/\kappa$ and $1/\kappa'$ (in our notation) for densities lower than the cross-over point and the leading oscillatory mode after the cross-over (see Fig. 5 in ref. 24). Furthermore, they showed the existence of two simultaneous decay modes of different types in the pair distribution function $g_{yy}(r)$, an oscillatory mode decaying as $e^{-kr_{00}/C_0}$, where $r_{00}$ is the average ion diameter. The decay lengths $1/\kappa$ and $1/\kappa'$ vary with ion density so that for some densities $1/\kappa < 1/\kappa'$ and for other densities the reverse is true (see Fig. 7 in ref. 24, where $1/\kappa_3$ is denoted by $\lambda_Q$ and $1/\kappa''$ is denoted by $\lambda_0$). The density where the decay lengths are equal is a so-called Fisher–Widom cross-over point,\textsuperscript{13} where changes in the decay lengths of $g_{yy}(r)$ make an oscillatory term to become leading for large distances instead of a monotonic term or vice versa. For NaCl at $T = 3000$ K, which is close to the critical temperature, this cross-over occurs at the density $n_{b0} = n_{c0}^{b0} = 0.1d^{-3}$. The monotonic term has the largest decay length for $n_{b0} < 0.1d^{-3}$, which is due to the proximity of criticality where the density–density correlations have a long range, and the oscillatory term has the largest decay length above this density.

In their analysis of the simulation data, Keblinski and coworkers extracted the decay length for the oscillatory decay mode from $H_{QQ}(r)$ and that for the monotonic mode from $H_{NN}(r)$. Therefore they denoted the former as the “screening length” $\lambda_Q$ and the latter as the “density–density correlation length” $\lambda_0$. A very important point to be made here is that there are terms in both correlation functions with these decay lengths, i.e., there is an oscillatory term in both $H_{QQ}(r)$ and $H_{NN}(r)$ with decay length $\lambda_Q$ and likewise a monotonic term in both functions with decay length $\lambda_0$. The magnitude of the prefactor for the term with decay length $\lambda_Q$ in $H_{NN}(r)$ is, however, small for the NaCl system and likewise the term with decay length $\lambda_0$ is small in $H_{QQ}(r)$, so these contributions are rather insignificant numerically. As we will see, this is due to the fact that $g_{yy}(r) \approx g_{xx}(r)$ in this system. For other systems the magnitudes of these kinds of terms in $H_{QQ}(r)$ and $H_{NN}(r)$ do not differ that much in general. In some cases they can have similar magnitudes.

In general, the leading terms of $H_{QQ}(r)$ and $H_{NN}(r)$ for large $r$ have the same decay length. This has been shown for binary electrolytes with spherical ions of different sizes or different valencies in ref. 13, 25 and 26. We will show in the current work that this applies to all decay modes in electrolytes of almost any kind, so the decay lengths are therefore both screening lengths and density–density correlation lengths – in this work they are simply called “screening lengths” and the $\kappa$ parameters are called “screening parameters.”

As we will see, there exist a few exceptions, for example model systems where the anions and cations are identical apart from the sign of their charges. The restricted primitive model (RPM) is such a model because all ions are charged hard spheres of equal diameter and have the same absolute valency. Then $g_{yy}(r) = g_{xx}(r)$ and, as a consequence of this symmetry, the density–charge correlation function $H_{QQ}(r)$ is identically equal to zero and $H_{QQ}(r)$ and $H_{NN}(r)$ have different decay lengths. It is quite astounding that the most common model for electrolytes is an exception as regards its screening behavior! However, as we saw from the case of NaCl where $g_{yy}(r) \approx g_{xx}(r)$ terms in $H_{QQ}(r)$ and $H_{NN}(r)$ with the same decay length have very different magnitudes and one dominates over the other, which means that results from models with $g_{yy}(r) = g_{xx}(r)$ may be quite reasonable as an approximation. These matters will be investigated for the general case in the current work and we will see that the class of systems that constitute this kind of exception consists of systems that are invariant when one reverses the sign of all charges in the system (including those in polar molecules). They will be called “charge-inversion invariant systems.”

The normal, realistic cases are, of course, systems that do not have such an invariance. Anions and cations virtually always differ by much more than the sign of their charges and, in addition, the positive and negative parts of the solvent molecules (if present) normally differ a lot apart from the sign of the charge. Thus $H_{QQ}(r)$ and $H_{NN}(r)$ normally have contributions with exactly the same set of decay lengths as the electrostatic potential and so do $g_{yy}$ and the free energy of interaction. It may appear puzzling that $H_{QQ}(r)$ and

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\textsuperscript{8} In the work by Keblinski et al. they use the notation $Q(r)$ and $G(r)$, where $Q(r)$ is proportional to $H_{QQ}(r)$ and the function $G(r) - 4n_{b0}$ is proportional to $H_{NN}(r)$. $n_{b0}$ is the total ion density. The definitions of $H_{QQ}(r)$ and $H_{NN}(r)$ can be found in Appendix B.
$H_{\text{SD}}(r)$ have the same decay length also in cases where long-range density fluctuations arise upon approach to a critical point. Since both functions have equal range, $H_{\text{QQ}}(r)$ is also an equally long-range function but we will see that the ratio $H_{\text{QQ}}(r)/H_{\text{SN}}(r)$ for large $r$ decreases quickly with increasing decay length, so the influence of $H_{\text{QQ}}(r)$ vanishes when the decay length diverges. We must, however, emphasize that the present analysis does not cover critical phenomena, so critical points are not included.

As mentioned earlier, simultaneous decay modes with long-range monotonic and shorter-range oscillatory exponential decays have been observed in ionic liquids and concentrated electrolyte solutions. The presence of oscillatory, exponentially damped contributions to the correlation functions reflect in many cases the granularity of the system at the molecular level. The wavelength can have about the same magnitude as the average molecular size or, for ionic systems, a combination of the anion and cation diameters. This is, however, not always the case, for example when the particle sizes are very different or when the electrostatic correlations dominate.

It is important to note that oscillatory surface forces that extend several oscillations, like those observed in electrolyte solutions and ionic liquids, have decay lengths and wavelengths that are the same as those in the bulk fluid in equilibrium with the fluid between the surfaces. It is common to think of such oscillations as being specifically caused by a layered structure close to the surfaces, but the corresponding structuring is present also for the pair distributions in the bulk. A specific structure may be induced by the surfaces that cause a few oscillations in the surface force for very short surface separations, but in order for the oscillations to continue when the separation is increased, they must be supported by a decay mode of the bulk phase. As we have seen, the leading modes of the decay of the correlation functions often dominate not only for very large distances, but also for surprisingly short ones. This applies also for surface forces. Therefore, the bulk properties of electrolytes that are analyzed in the current work have great significance also for surface forces.

Oscillatory contributions are expected for most dense liquid systems, including dilute electrolyte solutions with molecular solvent. The oscillations are then dominated by the structure of the solvent. This has been observed experimentally by Smith et al.\textsuperscript{9} in surface force experiments for mixtures of an ionic liquid and a polar solvent. Simultaneously, there was a monotonic long-range decay mode. Their findings regarding the oscillatory contributions (but not the long-range monotonic one) have been illustrated theoretically in calculations by Coupette et al.\textsuperscript{27,28} for a semi-primitive model of electrolyte solutions, where the monovalent ions and the solvent are hard spheres. They studied systems with equally-sized ions,\textsuperscript{27} where the model is a charge-inversion invariant system, and systems with unequally-sized ions.\textsuperscript{28} Several decay modes were found and analyzed. For the case of equally-sized ions, the charge-charge and the density-density correlations have different decay lengths and different wavelengths, while for unequally-sized ions all correlation functions have common lengths. The wavelengths and the decay lengths were determined by the bulk phase, as always when the oscillations are maintained for increased surface separation. These observations are in agreement with the findings in the current work and in ref. 29 and 31. The presence of multiple decay modes and the behavior of the decay lengths for this system are also in line with previous studies of the primitive model (without discrete solvent) in ref. 13, 18–20 and 25.

The oscillatory contributions due to solvent molecules must be present also for dilute aqueous solutions of simple salts and, as argued in ref. 29, such oscillatory forces determined by the bulk phase have been observed in both surface force experiments and in theoretical investigations of bulk systems with discrete water molecules; the oscillatory surface forces are therefore not caused by “hydration forces” specifically associated with the surfaces. Thus, for dilute electrolyte solutions in general, there are simultaneous decay modes with long-range monotonic decay (i.e., the traditional decay mode with $\kappa \approx \kappa_0$) and a shorter-range oscillatory exponential decay mode with a complex-valued screening parameter – the latter mode due primarily to the structure of the solvent. Most importantly, both of these decay modes are present in the correlation functions, the mean electrostatic potential and the interaction free energy.

The fact that the coupling between fluctuations in number density and fluctuations in charge density leads to equal decay lengths for the screened electrostatic potential, and all correlation functions, including $H_{\text{SD}}(r)$, are analyzed thoroughly in the present work. We will prove that all decay modes in electrolytes, both the monotonic and the oscillatory exponential ones, originate in the general case from the same fundamental equation, which describes the relevant dielectric response of the system and involves the static dielectric function $\varepsilon(k)$, where $k$ is the wave number. This equation, which constitutes the general exact equation for the screening parameter $\kappa$, can be written in several equivalent manners. Perhaps the most appealing manner is the following expression\textsuperscript{30,31} that is similar to eqn (1) for the Debye parameter

$$
\kappa^2 = \frac{\beta \sum n_i^b q_i^* q_i^*}{\varepsilon_r^*(k) \varepsilon_0},
$$

where $q_i^*$ is a renormalized charge (called the dressed particle charge) for particles of species $i$ and $\varepsilon_r^*(k)$ is called the dielectric factor, which is obtained from $\varepsilon(k)$ in a manner described later. Since the unknown variable $\kappa$ appears on both sides in the expression, it is an equation for $\kappa$. The different solutions $\kappa_n$, $n = 1, 2, \ldots$, to this equation comprise the set of screening parameters for the various decay modes. There exist in general both real and complex-valued solutions to the equation, which give rise to the monotonic and the oscillatory modes, respectively. An equivalent manner to write eqn (5) is, as we will see, $\varepsilon(k \kappa) = 0$, which explicitly involves the static dielectric function. The physical meaning of the appearance of the imaginary unit $i$ in $k = i \kappa$ is that the dielectric response is evaluated for an exponentially decaying field rather than a sinusoidal one (the latter corresponds to real $k$).
The exact theory presented in the present paper is very general and is valid for electrolytes from low to high ion densities: from thin gases to dense liquids and from dilute to concentrated electrolyte solutions with molecular solvent. The ions, the solvent and any other particles can have any sizes, arbitrary shapes, any internal charge distributions and can interact with any reasonable non-electrostatic pair interactions. For simplicity, the particles are, however, assumed to be rigid and un polarizable. The theory is a generalization of the dressed ion theory (DIT)\textsuperscript{21,25,32} and the dressed molecule theory (DMT)\textsuperscript{33-35} and it extends previous related work on the interactions in electrolytes.\textsuperscript{29-31} The basis of the theory is derived in the present paper in a new self-contained manner, which allows it to be done in a physically intuitive, yet correct way without use of advanced statistical mechanics. This derivation should allow a much broader readership than otherwise would be possible. It also allows a gradual introduction of the principally new results of the present work in an equally transparent manner. Although this theory treats quite complex phenomena and therefore is intrinsically complicated, the introduction of the concept of dressed particles, which is a key element, allows the general exact features of electrolytes to be cast in a form that is much simpler than otherwise is possible.

The contents of the paper are as follows. Section 2 contains a treatment of the linear polarization response of electrolytes due to the effect of weak external electrostatic fields. This is done in a special manner which focuses on the free energy of interaction for each constituent particle in the fluid and is formulated such that it is suitable as a basis for the general treatment of interparticle interactions in electrolytes. The nonlocal nature of electrostatic interactions in electrolytes is thereby demonstrated in a straightforward manner and is contrasted to the treatment of electrolytes in the PB approximation. The relationship to the static dielectric function $\varepsilon(k)$ is given. In Section 3, the concept of a dressed particle is defined from an elementary analysis of the polarization response of electrolytes. The charge density of a dressed particle of species $i$ is denoted by $\rho_i^*$ and it is an entity that is involved in all major aspects of screened interactions in electrolytes. The screened Coulomb potential $\phi_{\text{Coul}}^*(r)$, which describes the spatial propagation of electrostatic interactions in electrolytes, is defined for the general case and can be expressed in terms of $\tilde{\varepsilon}(k)$ in a simple manner. It is shown that one can use $\phi_{\text{Coul}}^*(r)$ and $\rho_i^*$ in Coulomb's law to calculate the mean electrostatic potential $\psi_i$ due to the particle. The general exact eqn (5) for the screening parameters $\kappa_i$ of the decay modes of $\phi_{\text{Coul}}^*(r)$ is derived. The potential $\psi_i$ has the same set of decay modes as $\phi_{\text{Coul}}^*(r)$ and the different magnitudes of the modes of $\psi_i$ are determined by projections of $\rho_i^*$ on each mode. The projections can be interpreted in terms of “effective” charges, as we will see. Section 4 deals with the pair distribution function $g_{ij}$ and the free energy of interaction $w_{ij}$ between particles in electrolytes, i.e., the pair potential of mean force. The distribution functions $g_{ij}^*$ of the dressers of the particles are also obtained. It is shown that all decay modes of $w_{ij}$ and $g_{ij}$ are, in general, the same as those of $\phi_{\text{Coul}}^*(r)$ and are hence determined by the dielectric response of the electrolyte. This is caused by the coupling between charge density and number density fluctuations and implies that $H_{\text{QQ}}(r)$, $H_{\text{SO}}(r)$ and $H_{\text{SOQ}}(r)$ also have the same set of decay modes. The exception is charge-inversion invariant systems where the modes of $H_{\text{SSQ}}(r)$ differ from those of $H_{\text{QQ}}(r)$ and of $\phi_{\text{Coul}}^*(r)$ since charge and density fluctuations in this case are independent of each other for symmetry reasons. Such invariant systems are investigated in some detail. Finally, the decays of $H_{\text{QQ}}(r)$, $H_{\text{SSQ}}(r)$ and $H_{\text{SOQ}}(r)$ for large $r$ are investigated for the general case. Section 5 contains a summary of the conclusions of this work and also some perspectives on the use of the results in experimental and theoretical investigations.

2 Polarization response and nonlocal electrostatics in electrolytes

2.1 Polarization and external electrostatic potential

Consider an inhomogeneous electrolyte that is exposed to an external electrostatic potential $\Psi_{\text{ext}}(r)$ from a source outside the system. The source may be a macroscopic body in contact with the electrolyte or a particle of any size and shape immersed in the electrolyte. The body or particle that is external to the system contains an arbitrary internal charge distribution that gives rise to $\Psi_{\text{ext}}$. Let us change the external potential somewhat by changing this charge distribution, so the system is instead exposed to the external potential $\Psi_{\text{ext}}(r) + \delta \Psi_{\text{ext}}(r)$ where $\delta \Psi_{\text{ext}}(r)$ is assumed to be small everywhere in the electrolyte. Initially the charge distribution of the inhomogeneous fluid phase is $\rho(r)$, which can be obtained from the density distribution of constituent particles of the electrolyte. For the case of spherically symmetric ions with a point charge at the center, we have $\rho(r) = \sum_i q_i n_i(r)$, where $n_i(r)$ is the number density distribution of ion species $i$ (the corresponding formula for nonspherical ions and other particles will be given later). The change $\delta \Psi_{\text{ext}}$ in external potential makes the charge distribution become $\rho(r') + \delta \rho(r')$, where $\delta \rho(r') = \sum_i \delta n_i(r')$ and $\delta n_i$ is the variation in number density induced by the change in $\Psi_{\text{ext}}$. For reasons that soon will be apparent, this variation of $\rho$ will be called a polarization charge density induced by $\delta \Psi_{\text{ext}}$, so we have $\delta \rho_{\text{pol}} \equiv \delta \rho$. The total electrostatic potential in the system is equal to $\Psi(r)$ given by

$$\Psi(r) = \Psi_{\text{ext}}(r) + \int \text{d}r' \rho(r') \phi_{\text{Coul}}(|r-r'|).$$

where $\phi_{\text{Coul}}(r) = 1/(4\pi\varepsilon_0 r)$ is the (unit) Coulomb potential and the integral is taken over the whole space (this is the convention throughout the paper for integrals without limits). The variation in $\Psi$ due to the influence of $\delta \Psi_{\text{ext}}$ is hence given by

$$\delta \Psi(r) = \delta \Psi_{\text{ext}}(r) + \delta \Psi_{\text{pol}}(r). \quad (6)$$

where

$$\delta \Psi_{\text{pol}}(r) = \int \text{d}r' \delta \rho_{\text{pol}}(r') \phi_{\text{Coul}}(|r-r'|). \quad (7)$$
is the potential from the polarization charge density $\delta \rho_{\text{pol}}$ induced by $\delta \Psi_{\text{ext}}$.

The density distribution $n_i(r')$ can be expressed in terms of the potential of mean force $W_i(r')$ as the Boltzmann relationship

$$n_i(r') = n_i^0 e^{-\beta W_i(r')},$$  \hspace{1cm} (8)

where $n_i^0$ is the density of ions of species $i$ in the bulk phase that is in equilibrium with the inhomogeneous phase. $W_i$ is set to zero in the bulk phase. When $n_i$ is changed by $\delta n_i$ and $W_i$ is changed by $\delta W_i$ it follows from eqn (8) that

$$\delta n_i(r') = n_i^0 e^{-\beta W_i(r')} [-\beta \delta W_i(r')],$$

so we have

$$\delta n_i(r') = -n_i^0 \delta W_i(r'),$$

(9)

which will be of use later. In the Poisson–Boltzmann (PB) approximation $W_i(r') = q_i \Psi(r')$, while in reality there are also contributions to $W_i$ that depend on ion–ion correlations, which are entirely neglected in the PB approximation.

For the special case of a bulk electrolyte, which is the main subject in this work, the density of species $i$ is equal to $n_i^0$ and we have $\rho(r) = 0$. Then, the electrostatic potential $\Psi$ is constant and we set it equal to zero by convention. Likewise we initially have $\Psi_{\text{ext}} = 0$. Let us expose the system to an external electrostatic potential $\delta \Psi_{\text{ext}}(r)$ that is small everywhere in the electrolyte. This potential polarizes the bulk electrolyte and gives rise to a polarization charge density $\delta \rho_{\text{pol}}$ and hence to nonzero $\delta \rho_{\text{pol}}$ and $\delta \Psi$ given by eqn (6) and (7). Henceforth, when we consider the polarization of a bulk electrolyte, all functions with a $\delta$, like $\delta \rho_{\text{pol}}$, $\delta \Psi_{\text{pol}}$, $\delta \Psi$, and $\delta \Psi_{\text{ext}}$, denote entities that are weak everywhere in the electrolyte; in principle they are infinitesimally small.

For a bulk fluid eqn (9) becomes

$$\delta n_i(r') = -n_i^0 \delta W_i(r'),$$

(10)

and the polarization charge density $\delta \rho_{\text{pol}}(r') = \sum_j q_j^0 \delta n_j(r')$ can be determined when we know $\delta W_i$. Thereby, a central question is: how can $\delta W_i$ be determined from $\delta \Psi_{\text{ext}}$ or, in other words, what is the free energy of interaction $\delta W_i(r')$ of an $i$-ion at $r'$ when it interacts with the weak electrostatic potential $\delta \Psi_{\text{ext}}$?

As we will show below

$$\delta W_i(r_1) = q_i \delta \Psi_{\text{ext}}(r_1) + \int \mathrm{d} r_2 \rho_i(\mathbf{r}_2) \delta \Psi_{\text{ext}}(\mathbf{r}_2),$$

(11)

where $\rho_i(\mathbf{r}_2)$ is the charge density that surrounds an $i$-ion in the bulk phase, i.e., the charge density of the surrounding ion cloud which is given by

$$\rho_i(\mathbf{r}_2) = \sum_j q_j n_j^0 g_j(\mathbf{r}_2) = \sum_j q_j n_j^0 h_j(\mathbf{r}_2),$$

(12)

where the sum is taken over all species, $h_j = g_j - 1$ is the total pair-correlation function and we have used the fact that $\sum_j q_j n_j^0 = 0$, which follows from charge electroneutrality of the bulk phase. Eqn (11) says that the potential of mean force $\delta W_i$ is given by the interactions of $\delta \Psi_{\text{ext}}$ with the $i$-ion itself, that is, with the charge $q_i$ at the center of the ion [the first term on the right-hand side (rhs)], and with the ion cloud with density $\rho_i$ [the integral], whereby the latter interaction affects the $i$-ion via the ion–ion correlations. Eqn (11) is an exact result, so it is precisely in this manner that ion–ion correlations enter into $\delta W_i$. For reasons that will be apparent in the following sections, we will, however, need to express the effects of such correlations in a more useful manner later.

To show eqn (11) we will make use of an exact relationship in statistical mechanics for fluids called Yvon’s first equation. It says that for a bulk fluid mixture that we expose to a weak external potential $\delta \Psi_{\text{ext}}(\mathbf{r}_1)$ acting on the various species $j$, we have

$$\delta n_i(r_1) = -n_i^0 \left[ \delta \Psi_i(r_1) + \sum_j \left[ \int \mathrm{d} r_2 n_j^0 h_j(\mathbf{r}_2) \delta \Psi_j(\mathbf{r}_2) \right] \right],$$

(13)

for spherical particles. For nonspherical particles a similar equation applies that we will consider later (see Appendix A).

In our case of a weak external electrostatic potential $\delta \Psi_{\text{ext}}$ we have $\delta \Psi_j(r_2) = q_j \delta \Psi_{\text{ext}}(r_2)$, which means that

$$\delta n_i(r_1) = -n_i^0 \left[ q_i \delta \Psi_{\text{ext}}(r_1) + \sum_j \left[ \int \mathrm{d} r_2 n_j^0 h_j(\mathbf{r}_2) q_j \delta \Psi_{\text{ext}}(\mathbf{r}_2) \right] \right]$$

$$= -n_i^0 \left[ q_i \delta \Psi_{\text{ext}}(r_1) + \rho_i(\mathbf{r}_2) \delta \Psi_{\text{ext}}(\mathbf{r}_2) \right]$$

(14)

where we have used the definition (12) of $\rho_i$. Since $\delta n_i(r_1) = -n_i^0 \delta W_i(r_1)$ [eqn (10)] we can identify $\delta W_i$ as the square bracket, which is equal to the rhs of eqn (11). Thereby the latter equation is demonstrated.

We can write the central charge $q_i$ of the ion as a charge density $q_i \delta \rho_{\text{pol}}(r)$, where $\delta \rho_{\text{pol}}(r)$ is the three-dimensional Dirac delta function. By introducing $\rho_i^{\text{tot}}(r) = q_i \delta \rho_{\text{pol}}(r) + \rho_i(r)$, which is the total charge density of the ion itself and the surrounding ion cloud, we can write eqn (14) as

$$\delta n_i(r_1) = -n_i^0 \int \mathrm{d} r_2 \rho_i^{\text{tot}}(\mathbf{r}_2) \delta \Psi_{\text{ext}}(\mathbf{r}_2)$$

(15)

and eqn (11) can be expressed as

$$\delta W_i(r_1) = \int \mathrm{d} r_2 \rho_i^{\text{tot}}(\mathbf{r}_2) \delta \Psi_{\text{ext}}(\mathbf{r}_2).$$

(16)

This simple, exact relationship accordingly gives the free energy of interaction $\delta W_i$ of the ion with the weak electrostatic potential $\delta \Psi_{\text{ext}}$.

From eqn (15) it follows that the polarization charge density for a bulk electrolyte exposed to the weak electrostatic potential $\delta \Psi_{\text{ext}}$ is given by

$$\delta \rho_{\text{pol}}(r_1) = \sum_j q_j \delta n_j(r_1) = -\beta \sum_j q_j n_j^0 \int \mathrm{d} r_2 \rho_j^{\text{tot}}(\mathbf{r}_2) \delta \Psi_{\text{ext}}(\mathbf{r}_2).$$

We can write this as

$$\delta \rho_{\text{pol}}(r_1) = \int \mathrm{d} r_2 \chi'(\mathbf{r}_2) \delta \Psi_{\text{ext}}(\mathbf{r}_2).$$

(17)

where

$$\chi'(\mathbf{r}_2) = -\beta \sum_j q_j n_j^0 \rho_j^{\text{tot}}(\mathbf{r}_2).$$

(18)
The function $\chi''(r_{12})$ links the polarization charge density at one point, $r_1$, to the external electrostatic potential at another point, $r_2$. It is the polarization response function (a static, linear response function) that gives the contribution to $\delta\rho^{\text{pol}}$ at $r_1$ from the influence of the external electrostatic potential at $r_2$. The response function is a property of the unperturbed bulk fluid. By inserting the definitions of $\rho_{\text{tot}}(r)$ and $\rho_i$ into eqn (18) it follows that

$$\chi''(r_{12}) = -\beta \left[ \sum_i q_i^2 n_i^b \delta^{(3)}(r_{12}) + \sum_{ij} q_i q_j n_i^b n_j^b \delta_0(r_{12}) \right].$$

We can express this as $\chi''(r_{12}) = -\beta q_e^2 H_{QQ}(r_{12})$, where $q_e$ is the elementary (protonic) charge and $H_{QQ}$ is the charge–charge correlation function. The latter gives the correlation between fluctuations in charge densities at $r_1$ and $r_2$ irrespectively of which species the charges belong to and is for spherical ions equal to the square bracket divided by $q_e^2$. Its Fourier transform is equal to the charge–charge structure factor. These results for $\chi''$ and $H_{QQ}$ are well-known, but we have derived them here in a manner that is suitable for further development of the exact theory for electrolytes done later in this work.

Let us now turn to electrolytes consisting of nonspherical ions and other particles, for example solvent molecules. We will for simplicity solely treat the case of rigid, unpolarizable particles, but for each species $i$, the particle size, shape and internal charge density $\sigma_i$ are arbitrary. For a particle with center of mass at $r_1$ and orientation $\omega_3$, the internal charge density at point $r_1$ is given by $\sigma_i(r_1, \omega_3)$. We use a normalized orientation variable $\omega$ so that $d\omega = 1$ where the integral is taken over all orientations.\textsuperscript{\dag} The charge of the particle is $q_i = \int d\omega \sigma_i(r_1, \omega_3)$, which is independent of $r_3$ and $\omega_3$. Note that $\sigma_i(r_1, \omega_3)$ for a given $\omega_3$ is a function of only $r_{31} = r_1 - r_3$, where the vector $r_{31}$ starts at the center of the particle. To simplify the notation we will henceforth write $(r_1, \omega_3) \equiv r_1$, whereby we have $\sigma_i(r_1, \omega_3) \equiv \sigma_i(r_1)$, which is the charge density at $r_1$ for a particle with coordinates $r_1$.

The pair interaction potential is $u_{ij} = u_{ij}^{\text{el}} + u_{ij}^{\text{ne}}$, which is the sum of the electrostatic (el) and nonelectrostatic (ne) interaction. The former is given by Coulomb's law as

$$u_{ij}^{\text{el}}(r_1, r_2) = \left[ dR_i \sigma_i(r_1) \phi_{\text{Coul}}(r_3 \delta) \right] \sigma_j(r_2).$$

and the latter is assumed to have a short range\textsuperscript{\ddag} unless something else is stated explicitly and to be strongly repulsive for small separations, but it is otherwise completely arbitrary.

The number density of particles with center of mass at $r_1$ and orientation $\omega_3$ is equal to $n_i(r_1, \omega_3) = n_i(r_1)$, and we have $n_i(r_1) = n_i^b e^{-\beta W_i(r_1)}$. The charge density of the system is

$$\rho(r) = \sum_i dR_i n_i(r) \sigma_i(r),$$

where $dR_i \equiv dR_i d\omega_3$ and the integral is taken over the whole space and over all orientations. When a variation in external potential gives rise to a change $\delta n_i$ in density, the resulting change in charge density is

$$\delta \rho^{\text{pol}}(r_1) = \sum_i dR_i \delta n_i(r) \sigma_i(r).$$

which is also applicable for the polarization charge density induced by the weak external potential $\delta \Psi^{\text{ext}}$ in a bulk phase with densities $n_j^b$ for the various species $j$.

As shown in Appendix A, the equation that corresponds to eqn (15) for a bulk fluid of nonspherical particles exposed to a weak external electrostatic potential $\delta \Psi^{\text{ext}}$ is

$$\delta n_i(r_1) = -\beta n_i^b \left[ \int dR_j \rho_{\text{tot}}^{\text{el}}(r_2 | R_1) \delta \Psi^{\text{ext}}(r_2) \right].$$

where

$$\rho_{\text{tot}}^{\text{el}}(r_2 | R_1) = \sigma_i(r_2 | R_1) + \rho_i(r_2 | R_1)$$

is the total charge density at $r_2$ around a particle with coordinates $R_1$ (i.e., it is located at $r_1$ and has orientation $\omega_3$). This density consists of the internal charge density $\sigma_i$ of the particle itself and the charge density $\rho_i$ of the surrounding cloud of ions, solvent molecules and other particles present in the electrolyte. Electroneutral particles contribute to the distribution since they have orientational order due to the interactions with the $i$-particle. For simplicity, $\rho_i$ will be denoted as the charge density of the ion/solvent cloud that surrounds the particle. In the same manner as before it follows that

$$\delta \Psi^{\text{el}}(r_1) = \left[ \int dR_j \rho_{\text{tot}}^{\text{el}}(r_2 | R_1) \delta \Psi^{\text{ext}}(r_2) \right].$$

The physical interpretation of this formula is identical to that for spherical particles.

For nonspherical particles the expression for the polarization response function $\chi''$ is somewhat more complicated than eqn (18). By inserting eqn (21) into eqn (20) we obtain

$$\delta \rho^{\text{pol}}(r_1) = -\beta \left[ \int dR_j n_j^b \left[ \int dR_k \sigma_j(r_1) \delta \Psi^{\text{ext}}(r_k) \right] \sigma_i(r_1) \right]$$

$$= -\beta \left[ \int dR_j \left[ \int n_j^b \left[ \int dR_k \sigma_j(r_1) \delta \Psi^{\text{ext}}(r_k) \right] \right] \right]$$

which can be written as eqn (17) when we make the identification

$$\chi''(r_{12}) = -\beta \left[ \int dR_j \sigma_j(r_1) \delta \Psi^{\text{ext}}(r_2) \right]$$

$$\equiv -\beta \left[ \int dR_j \int n_j^b \sigma_j(r_1) \delta \Psi^{\text{ext}}(r_2, \omega_3) \right],$$

where $r_{12} = |r_{12}|$ and $r_{12} = r_2 - r_1$. As before, the response function can be expressed in terms of the charge–charge correlation

\textsuperscript{\dag} We can, for example, take $\omega = \frac{\phi}{2\pi} \cos \theta \sin \eta$, where $\phi, \theta, \eta$ are the Euler angles of the particle or, for a linear molecule, $\omega = \frac{\phi}{2\pi} \cos \theta$ since $\eta$ is redundant in the latter case.

\textsuperscript{\ddag} The formalism has a general validity, but in the presence of nonelectrostatic interactions that decay like a power law [like dispersion interactions] there are terms that are not explicitly included in the asymptotic expressions for large distances obtained in this work; see the comments at the end of Section 4.1.
function as \( \chi'(r_{12}) = -\beta q_e^2 H_{\text{eq}}(r_{12}) \), where \( H_{\text{eq}} \) for the case of nonspherical particles is defined in Appendix B [eqn (134)].

We can write eqn (24) as

\[
\chi'(r_{12}) = -\beta \sum_i n_i^3 \int \left\langle \sigma_i(r_1|\mathbf{R}_1) p_i^{\text{pol}}(r_2|\mathbf{R}_2) \right\rangle_{\alpha_0}
\]

(25)

where we have written the integral over \( \omega_3 \) as an average since \( \int f(\omega)d\omega = \int f(\omega)d\omega' / \int f(\omega) d\omega \). Note that \( \chi'(r_{12}) \) depends only on the distance \( r_{12} \) because the average over orientations is taken.

2.2 Polarization and total electrostatic potential

To continue we note that \( \Psi_{\text{ext}} \) is the electrostatic potential due to the external source in the absence of the electrolyte, while \( \Psi \) is the total potential in the presence of the electrolyte. This means that \( \Psi \) is the mean potential that actually exists in the electrolyte. Since this is the system that we are interested in, it is a relevant task to express \( \delta \mu_{\text{pol}} \) in terms of \( \delta \Psi \) rather than in terms of \( \delta \Psi_{\text{ext}} \) as in eqn (17). Thereby we will obtain a relationship that can be used as an alternative to the latter equation.

The fact that the functions \( \delta \Psi_{\text{ext}} \), \( \delta \mu_{\text{pol}} \), and \( \delta \Psi \) are linearly related to each other implies that there exists a function \( \chi(r) \) that expresses the linear relationship between \( \delta \mu_{\text{pol}} \) and \( \delta \Psi \) as

\[
\delta \mu_{\text{pol}}(r_1) = \int \chi'(r_{12}) \delta \Psi(r_2),
\]

(26)

which is similar to eqn (17). The function \( \chi(r) \) will also be denoted as a “polarization response function,” although a true response function normally gives the response of a system due to an influence that we can completely control by external means. The total potential \( \delta \Psi \) contains \( \delta \Psi_{\text{pol}} \) from the polarization charge density that we cannot control directly, so \( \chi(r) \) is a slightly different kind of function than \( \chi(r) \).

In fact, for a given \( \chi'(r) \), the function \( \chi(r) \) is the solution to the equation

\[
\chi'(r_{14}) = \chi'(r_{14}) + \int \chi'(r_{12}) \delta \mu_{\text{pol}}(r_2|\mathbf{R}_2) \phi_{\text{Coul}}(r_3|\mathbf{R}_3),
\]

(27)

which we will derive below. The solution is readily obtained in Fourier space as

\[
\tilde{\chi}(k) = \frac{\tilde{\chi}^*(k)}{1 + \tilde{\phi}_{\text{Coul}}(k) \tilde{\chi}^*(k)},
\]

(28)

where the transform of the unit Coulomb potential is \( \tilde{\phi}_{\text{Coul}}(k) = 1/(\omega_0 k^3) \). We define the Fourier transform \( \tilde{f}(k) \) of a function \( f(r) \) as \( \tilde{f}(k) = \int f(r) e^{-i k \cdot r} \). For a function \( f(r) \) that only depends on \( r \) we have \( \tilde{f}(k) = \int f(r) \sin(kr)/(kr) \) where \( k = |k| \).

Eqn (27) is easily derived as follows. By inserting (6) into eqn (26) and then using eqn (7) we obtain

\[
\delta \mu_{\text{pol}}(r_1) = \int \chi'(r_{12}) \left[ \delta \Psi_{\text{ext}}(r_2) + \delta \Psi_{\text{pol}}(r_2) \right]
\]

\[
= \int \chi'(r_{12}) \delta \Psi_{\text{ext}}(r_2)
\]

\[
+ \int \chi'(r_{12}) \int \delta \mu_{\text{pol}}(r_3) \phi_{\text{Coul}}(r_3),
\]

where we have changed the integration variable in the first integral on the rhs from \( r_2 \) to \( r_3 \). We now insert eqn (17) and obtain

\[
\delta \mu_{\text{pol}}(r_1) = \int \chi'(r_{14})
\]

\[
+ \int \chi'(r_{12}) \phi_{\text{Coul}}(r_3) \delta \Psi_{\text{ext}}(r_3).
\]

Since the left-hand side (lhs) is equal to \( \int \chi'(r_{14}) \delta \Psi_{\text{ext}}(r_4) \) and \( \delta \Psi_{\text{ext}}(r_4) \) is arbitrary, eqn (27) follows.

It is illustrative to see what the PB approximation says about the polarization response. In this approximation we have for spherical ions \( \delta \Psi_{\text{pol}}(r_1) = q \delta \Psi(r_1) \) and hence from eqn (10) we obtain \( \delta n_i(r_1) = -\beta n_i q \delta \Psi(r_1) \). Therefore

\[
\delta \mu_{\text{pol}}(r_1) = \int q_i \delta n_i(r_1) = -\beta \int q_i n_i^3 \delta \Psi(r_1) \quad \text{(PB)}.
\]

(29)

where (PB) means that the equation is valid only in the PB approximation. This equation can be written as

\[
\delta \mu_{\text{pol}}(r_1) = -\beta \int q_i n_i^3 \delta \mu_{\text{pol}}(r_1) \quad \text{(PB)}
\]

and by comparing with the expression (26) for \( \chi^* \), we can identify

\[
\chi'(r_{12}) = -\beta \int q_i n_i^3 \delta \mu_{\text{pol}}(r_{12}) \quad \text{(PB)}.
\]

(30)

In eqn (29) we see that the polarization charge density \( \delta \mu_{\text{pol}}(r_1) \) is proportional to the total mean electrostatic potential \( \delta \Psi(r_1) \) at the same point \( r_1 \). This fact is likewise expressed by the appearance of the Dirac delta function \( \delta^{(3)}(r_{12}) \) in \( \chi'(r_{12}) \) for the PB case. Hence, electrostatics is local in the PB approximation: \( \delta \mu_{\text{pol}}(r_1) \) is not influenced by the values of \( \delta \Psi(r_3) \) at other points \( r_3 \neq r_1 \).

In reality, we have nonlocal electrostatics in the sense that the polarization at one point \( r_1 \) is influenced by the electrostatic potential at other points in the surroundings. This can be seen in eqn (26). The polarization response function \( \chi'(r_{12}) \) is non-zero for \( r_{12} \neq 0 \) so \( \delta \mu_{\text{pol}}(r_1) \) is influenced by \( \delta \Psi(r_3) \) for all points \( r_3 \) in the neighborhood of \( r_1 \). This feature is caused by the correlations between the particles in the electrolyte. The nonlocal nature of the response can be understood in the following manner. Any ion located at \( r_2 \) interacts with the electrostatic potential and since this ion correlates with other ions it will affect the probability for ions to be at \( r_1 \). The density of ions at
r1 accordingly depends on the potential elsewhere. In other words, the potential of mean force 〈δω(r1)⟩ hence the charge density 〈δρpol(r1)〉 depend on the values of 〈δΨ(r2)〉 for all points r2 in the vicinity of r1. This fact is expressed by the nonlocality in the general exact relationships we have obtained.

The response functions 〈χ′(r)〉 and 〈χ''(r)〉 are closely related to the dielectric response of the electrolyte in terms of the static dielectric function 〈δΨ(k)〉, where k is a wave number. To see this we introduce the electrostatic fields corresponding to 〈Ψ(r)〉 and 〈Ψext〉, which are 〈E(r)〉 = −∇〈Ψ(r)〉 and 〈Eext(r)〉 = −∇〈Ψext(r)〉, respectively. E is sometimes called the Maxwell field and 〈Eext〉 can be expressed in terms of the displacement field 〈D〉 given by 〈D(r)〉 = 〈ε0Eext(r)〉 but we will use 〈Eext〉 in the present work. As we have seen 〈Eext〉 is the field from the external source in the absence of the fluid. The reasoning is performed most easily in Fourier space, so we will consider 〈E(k)〉 and 〈Eext(k)〉.

The static (longitudinal) dielectric function 〈δΨ(k)〉 relates these electrostatic fields when they are weak and therefore linearly related to each other. In general we have for a homogeneous and isotropic fluid

\[ \mathbf{k} \cdot \mathbf{E}(k) = \frac{\mathbf{k} \cdot \mathbf{E}^{\text{ext}}(k)}{\tilde{\epsilon}(k)} \quad \text{(weak fields)}, \]

where \( \mathbf{k} = k / k \) is a unit vector and where the field components along the wave vector \( k \) (the longitudinal components) are projected out. In electrostatics \( \mathbf{E} \) and \( \mathbf{E}^{\text{ext}} \) are parallel to \( k \), which can be seen from the fact that in Fourier space we have \( \mathbf{E}(k) = -i k \mathbf{P}(k) \) (the factor \( i k \) corresponds to \( \nabla \) in ordinary space) and likewise for \( \mathbf{E}^{\text{ext}} \). Hence we have \( \mathbf{k} \cdot \mathbf{E}(k) = -i k \mathbf{P}(k) = -i k \mathbf{P}^{\text{ext}}(k) \) and \( \mathbf{k} \cdot \mathbf{E}^{\text{ext}}(k) = -i k \mathbf{P}^{\text{ext}}(k) \). Therefore only the longitudinal components of the fields matter. In electrodynamics the transversal components, which are perpendicular to \( k \), also matter, but in the present work we limit ourselves to the static equilibrium case. The frequency dependence of the dielectric function, that is commonly considered in the study of the polarization of fluids, is therefore not included in the treatment.

In line with the notation above, when we expose the electrolyte to a weak field we put \( \mathbf{A} \) on the potentials \( \Psi \) and \( \Psi^{\text{ext}} \). Therefore we write \( \delta \Psi \) and \( \delta \Psi^{\text{ext}} \) instead of \( \Psi \) and \( \Psi^{\text{ext}} \), whereby eqn (31) implies

\[ \delta \Psi(k) = \frac{\delta \Psi^{\text{ext}}(k)}{\tilde{\epsilon}(k)} \quad \text{(32)} \]

In Fourier space eqn (7), (17) and (26) are

\[ \delta \Psi^{\text{pol}}(k) = \tilde{\phi}_{\text{Coul}}(k) \delta \rho^{\text{pol}}(k) \]
\[ \delta \Psi^{\text{pol}}(k) = \tilde{\chi}'(k) \delta \Psi^{\text{ext}}(k) \]
\[ \delta \Psi^{\text{pol}}(k) = \tilde{\chi}''(k) \delta \Psi^{\text{pol}}(k) \]

and since \( \delta \Psi = \delta \Psi^{\text{ext}} + \delta \Psi^{\text{pol}} \) we obtain

\[ \delta \Psi^{\text{ext}}(k) = \delta \Psi^{\text{ext}}(k) + \tilde{\phi}_{\text{Coul}}(k) \tilde{\chi}'(k) \delta \Psi^{\text{ext}}(k) \]
\[ \delta \Psi^{\text{ext}}(k) = \delta \Psi^{\text{ext}}(k) + \tilde{\phi}_{\text{Coul}}(k) \tilde{\chi}''(k) \delta \Psi^{\text{pol}}(k). \]

By comparing the last two equations with eqn (32) we can identify

\[ \tilde{\epsilon}(k) = \frac{1}{1 + \tilde{\phi}_{\text{Coul}}(k) \tilde{\chi}'(k) \tilde{\chi}''(k)} = 1 - \tilde{\phi}_{\text{Coul}}(k) \tilde{\chi}''(k). \quad \text{(34)} \]

Note that the last equality is equivalent to eqn (28). These expressions for \( \tilde{\epsilon}(k) \) in terms of \( \tilde{\chi}'(k) \) and \( \tilde{\chi}''(k) \) are equivalent to standard expressions for \( \tilde{\epsilon}(k) \) given in ref. 36–38.

The nonlocal nature of electrostatics in the microscopic domain, which is described by the nonlocal response functions \( \tilde{\chi}'(r) \) and \( \tilde{\chi}''(r) \), is hence included in the \( k \) dependence of the dielectric function 〈δΨ(k)〉. In the PB approximation, where electrostatics is local, we have from eqn (30) \( \tilde{\chi}'(k) = -\beta \sum_j q_j^2 n_j^3 = -e_0 K D^2 \), where \( K \) is the Debye screening parameter for ions in vacuum given by

\[ K D^2 = \frac{\beta}{e_0} \sum_j q_j^2 n_j^3 \quad \text{(ions in vacuum)} \quad \text{(35)} \]

and \( 1 / K D \) is the Debye length. We see that \( \tilde{\chi}''(k) \) is independent of \( k \) in the PB case and we obtain

\[ \tilde{\epsilon}(k) = 1 + \beta \sum_j q_j^2 n_j^3 \tilde{\phi}_{\text{Coul}}(k) = 1 + \frac{K D^2}{k^2} \quad \text{(PB)}. \quad \text{(36)} \]

In this case, the entire \( k \) dependence of 〈δΨ(k)〉 originates from the Coulomb potential 〈δΨCoul(k)〉.

For completeness, we note that in electrostatics it is common to use the electric susceptibility 〈χ〉, which gives the polarization density 〈P〉 in terms of the total field as 〈P〉 = 〈ε0ε0E〉 for weak fields.†† In our notation we have 〈P〉 = 〈ε0 ε0 Epol〉, where 〈Epol(r)〉 = −∇〈Ψpol(r)〉. For weak fields we thus have −e0ε0 Epol = e0ε0 E0 and in terms of potentials 〈δΨpol〉 = −ε0ε0 E0 E by comparing with eqn (33) we see that

\[ \tilde{\chi}_c(k) = -\tilde{\phi}_{\text{Coul}}(k) \tilde{\chi}'(k) = \frac{\tilde{\chi}'(k)}{e_0 k^2} \]

so the electric susceptibility is related to 〈δΨ〉 as 〈δΨ〉 = 1 + 〈χ〉 as usual. In traditional treatments of polar media, 〈P〉 expresses dipolar polarization, while in the general case, including electrolytes, 〈Epol〉 and hence 〈P〉 originate from all kinds of polarization, for example from changes in dipolar, quadrupolar, octupolar orientations and ion distributions.31 For a pure polar fluid (no ions), the dielectric constant (relative dielectric permittivity) is defined microscopically as \( \varepsilon_r = \lim_{k \to 0} \tilde{\chi}(k) \) and in this limit solely the dipolar polarization contributes. In the case of electrolytes this limit does not exist since 〈δΨ〉 diverges at \( k = 0 \) and, as we will see, other \( k \) values matter a lot.

To minimize the number of symbols we use, we will refrain ourselves from using 〈χc〉 in what follows. Instead we use 〈χ〉 in ordinary space 〈χ(r)〉, which plays a central role in the present work. Obviously, 〈χ〉 has a prominent role also in ordinary electrostatics.

†† In macroscopic electrostatics we have 〈P〉 = 〈ε0E〉 in ordinary space, while in the present microscopic case for bulk fluids we have the corresponding relationship in Fourier space.
3 Dressed particles and screened electrostatic interactions

3.1 Definition of dressed particles

The potential of mean force \( \delta W_i \) due to a weak external electrostatic potential \( \delta \Psi \text{ext} \) was expressed in eqn (23) as the interaction energy between \( \delta \Psi \text{ext} \) and the total charge density \( \rho_i ^{\text{tot}} \) associated with an i-particle, where \( \rho_i ^{\text{tot}} \) is the sum of the internal charge densities \( \sigma_i \) of the particle and \( \rho_i \) of the surrounding ion/solvent cloud, i.e., the surrounding cloud of particles of any kind present in the system. We will now determine the corresponding relationship between \( \delta W_i \) and the total electrostatic potential \( \delta \Psi \). Thereby we are led to the definition of the concept of dressed particles and the corresponding charge density \( \rho_i ^* \), which plays a fundamental role in the understanding of interactions in electrolytes.

By inserting \( \delta \Psi \text{ext} \) in eqn (7) into eqn (23) we obtain

\[
\delta W_i (R_1) = \int d_r \rho_i ^{\text{tot}} (r_2 | R_1) \delta \Psi (r_2) - \int d_r \delta \rho_i ^{\text{tot}} (r_2 | R_1) \delta \Psi (r_2)
\]

(37)

and it remains to express the last term in terms of \( \delta \Psi \). By inserting \( \delta \rho_i ^{\text{tot}} \) from eqn (7) into the last term, it can be written

\[
\int d_r \rho_i ^{\text{tot}} (r_2 | R_1) \delta \rho_i ^{\text{pol}} (r_2) = \int d_r \rho_i ^{\text{tot}} (r_2 | R_1) \int d_r \delta \rho_i ^{\text{pol}} (r_3) \phi_{\text{Coul}} (r_2, r_3) = \int d_r \int d_r \rho_i ^{\text{tot}} (r_2 | R_1) \delta \rho_i ^{\text{pol}} (r_3) = \int d_r \psi_i (r_3 | R_1) \delta \rho_i ^{\text{pol}} (r_3),
\]

(38)

where

\[
\psi_i (r_3 | R_1) = \int d_r \rho_i ^{\text{tot}} (r_2 | R_1) \phi_{\text{Coul}} (r_3, r_2)
\]

is the electrostatic potential from the charge density \( \rho_i ^{\text{tot}} \), that is, the mean electrostatic potential due to the i-particle located at \( r_1 \) and with orientation \( \omega_i \). We can now express \( \delta \rho_i ^{\text{pol}} \) in terms of \( \delta \Psi \) by using eqn (26) and by inserting the result into eqn (37) we obtain

\[
\delta W_i (R_1) = \int d_r \rho_i ^{\text{tot}} (r_2 | R_1) - \int d_r \psi_i (r_3 | R_1) \chi^* (r_2) \delta \Psi (r_2).
\]

(39)

If we define

\[
\rho_i ^* (r_2 | R_1) = \rho_i ^{\text{tot}} (r_2 | R_1) - \int d_r \psi_i (r_3 | R_1) \chi^* (r_2)
\]

(40)

we can write eqn (39) as

\[
\delta W_i (R_1) = \int d_r \rho_i ^* (r_2 | R_1) \delta \Psi (r_2).
\]

(41)

By comparing this expression with eqn (23) we see that when \( \delta W_i \) is expressed in terms of \( \delta \Psi \) instead of \( \delta \Psi \text{ext} \), the charge density \( \rho_i ^* \) takes the role corresponding to that of \( \rho_i ^{\text{tot}} \) in eqn (23). Since \( \delta n_i (R_1) = - \beta n_i ^0 \delta W_i (R_1) \) [cf. eqn (10)] we obtain from eqn (41)

\[
\delta n_i (R_1) = - \beta n_i ^0 \int d_r \rho_i ^* (r_2 | R_1) \delta \Psi (r_2),
\]

(42)

which can be compared with eqn (21).

We will now interpret the physical meaning of \( \rho_i ^* \). Let us imagine a particle of species \( i \) with orientation \( \omega_i \) into a bulk electrolyte at the point \( r_1 \). Initially the charge density is zero and after the immersion the density at \( r_2 \) is \( \rho_i ^{\text{tot}} (r_2 | R_1) \). We have \( \rho_i ^{\text{pol}} = \rho_i ^* + \rho_i ^{\text{lin}} \) [eqn (22)] and \( \rho_i ^{\text{pol}} \) can be regarded as the total polarization charge density that the particle induces in the surrounding electrolyte due to all kinds of interactions between this particle and the other particles (not only the polarization due to electrostatic interactions). Since these interactions are strong close to the particle, one cannot treat the polarization by linear response. If, however, the total electrostatic potential \( \psi_i \) due to the particle were weak, we would according to eqn (26) have the polarization charge density due to this potential

\[
\rho_i ^{\text{pol}} (r_2 | R_1) = \int d_r \psi_i (r_3 | R_1) \chi^* (r_2) \quad \text{(weak)}
\]

(43)

The rest of the polarization of the bulk electrolyte caused by the interactions between the immersed i-particle and the other particles is contained in \( \rho_i ^{\text{tot}} - \rho_i ^{\text{lin}} \). We now define the dressed particle charge-density \( \rho_i ^* \) of the particle as

\[
\rho_i ^* (r_2 | R_1) = \rho_i ^{\text{tot}} (r_2 | R_1) - \rho_i ^{\text{lin}} (r_2 | R_1),
\]

(44)

which is the same as eqn (40). Since \( \rho_i ^{\text{tot}} = \rho_i ^* + \rho_i ^{\text{dres}} \) we can write

\[
\rho_i ^* (r_2 | R_1) = \sigma_i (r_2 | R_1) + \rho_i ^{\text{dress}} (r_2 | R_1),
\]

(45)

where \( \rho_i ^{\text{dress}} \equiv \rho_i ^* - \rho_i ^{\text{lin}} \) is the charge density of the “dress” of the i-particle. The dress expresses the effects of many-body correlations between the particles in the electrolyte in addition to those included in the linear response. Like the charge density \( \rho_i \) of the ion/solvent cloud, which is given by the pair distribution function \( g_{io} \), the dress is a distribution of particles of all kinds surrounding the i-particle. The pair distribution function \( g_i ^* \) that gives the charge distribution \( \rho_i ^{\text{dress}} \) will be investigated in Section 4.

The density \( \rho_i ^* \) plays a key role in what follows. For instance, as we have seen from eqn (23) and (41), the dressed charge density \( \rho_i ^* \) has the same role vis-à-vis the total potential \( \delta \Psi \) as...
the total charge density \( \rho_{i}^{\text{tot}} \) has \( \text{vis-à-vis} \) the external potential \( \delta \Psi_{\text{ext}} \). Eqn (41) says that in the linear domain, the free energy of interaction \( \delta W_{i} \) of the \( i \)-ion with the total electrostatic field is equal to the electrostatic interaction energy between \( \delta \Psi \) and the dressed particle charge–density \( \rho_{i}^{\ast} \) associated with the ion. As we will see \( \rho_{i}^{\ast} \) has very important roles in the interparticle interactions in electrolytes.

Note that \( \rho_{i}^{\ast} \) for each \( i \) can be expressed in terms of \( \rho_{j}^{\text{tot}} \) for all \( j \) via eqn (40) where \( \psi_{\text{t}} \) is given in terms of \( \rho_{i}^{\text{tot}} \) by eqn (38) and where \( \chi^{\ast} \) can be expressed in terms of \( \chi^{\ast} \) via eqn (28) and finally in terms of \( \rho_{i}^{\text{tot}} \) via eqn (25). Thus, given \( \rho_{i}^{\text{tot}} \) for all \( j \), one can calculate \( \rho_{i}^{\ast} \) at least in principle.

Let us return now to the case when a bulk electrolyte is polarized by a weak electrostatic potential. Then, \( \delta n_{i} \) is given by eqn (42) and we can readily derive

\[
\chi^{\ast}(r_{12}) = - \beta \sum_{i} n_{i}^{b} \left[ \int d\mathbf{r}_3 \langle \sigma_{i}(\mathbf{r}_3) \rho_{i}^{\ast}(\mathbf{r}_2|\mathbf{r}_3) \rangle_{\mathbf{r}_3} \right]_{\omega_{i}}^{(46)}
\]

where the first term on the rhs is the sole contribution to \( \delta \Psi \) included in the PB approximation and the last term describes the effects of ion–ion correlations. In the PB approximation we thus neglect that the ions have dresses when evaluating \( \delta W_{i} \).

For cases where all particles in the electrolyte are spherical eqn (46) reduces to

\[
\chi^{\ast}(r_{12}) = - \beta \sum_{i} n_{i}^{b} q_{i} \rho_{i}^{\ast}(r_{12}),
\]

which should be compared with eqn (18). Note that \( \chi^{\ast}(r_{12}) \) for the the PB case, eqn (30), is obtained from eqn (54) when we set \( \rho_{i}^{\ast}(r_{12}) = q_{i} \delta^{(3)}(r_{12}) \), that is, the charge density of a bare ion without its dress. This is in agreement with eqn (53) in the absence of its last term. In the PB approximation, all particles in the electrolyte are treated as being bare, except for the particle that causes the potential \( \delta \Psi \). That particle does have a dress in this approximation, as we will now see for the special case when the particle is an ion of species \( j \). This is very illustrative because it gives some insights into the concept of a dressed particle in this simple approximation.

Let us consider a single nonspherical \( j \)-particle immersed in an electrolyte consisting of spherical ions of equal sizes. In this case the PB approximation says that

\[
\rho_{j}^{\text{tot}}(\mathbf{r}|\mathbf{R}_1) = \begin{cases} \sum_{i} q_{i} n_{i}^{b} e^{-\delta \Psi_{\text{d}}(\mathbf{r})}, & \text{inside (PB)} \\ \sum_{i} q_{i} n_{i}^{b} e^{-\delta \Psi_{\text{d}}(\mathbf{r})}, & \text{outside (PB)} \end{cases}
\]

where \( \delta \Psi_{\text{d}} \) is the difference between the PB and the exact potential. Eqn (49) implies that the dielectric function \( \tilde{\varepsilon}(k) = 1 - \frac{\chi^{\ast}(k)}{\delta \Psi_{\text{d}}(k)} \) can be expressed as

\[
\tilde{\varepsilon}(k) = 1 + \frac{\beta}{\omega_{k}^{2}} \sum_{i} n_{i}^{b} \langle \tilde{\sigma}_{i}(\mathbf{k}) \rho_{i}^{\ast}(\mathbf{k}|\mathbf{R}_1) \rangle_{\mathbf{R}_1}.
\]

The reciprocal function \( \frac{1}{\tilde{\varepsilon}(k)} = 1 + \frac{\delta \Psi_{\text{d}}(k)}{\chi^{\ast}(k)} \) can likewise be expressed in terms of \( \rho_{i}^{\text{tot}} \) and thereby in terms of \( \delta \Psi_{\text{d}} \) via the Fourier transform of eqn (24). The latter is the usual path to \( \tilde{\varepsilon}(k) \), while we will see that eqn (50) is much more useful.

Let us yet again connect with the PB approximation. For this purpose we apply the general, exact equations above to the special case of spherical ions with a charge \( q_{i} \) at the center, whereby eqn (43) becomes

\[
\rho_{i}^{\text{lin}}(r_{12}) = \int d\mathbf{r}_3 \tilde{\psi}_{i}(r_{13}) \chi^{\ast}(r_{32})
\]

and we have \( \rho_{i}^{\ast}(r_{12}) \equiv \rho_{i}^{\text{tot}}(r_{12}) - \rho_{i}^{\text{lin}}(r_{12}) \). In such cases eqn (45) becomes

\[
\rho_{i}^{\ast}(r_{12}) = \sigma_{i}(r_{12}) + \rho_{i}^{\text{dress}}(r_{12}) = q_{i} \delta^{(3)}(r_{12}) + \rho_{i}^{\text{dress}}(r_{12}),
\]

where \( \rho_{i}^{\text{dress}}(r_{12}) \equiv \rho_{i}(r_{12}) - \rho_{i}^{\text{lin}}(r_{12}) \). Eqn (41) can be written as

\[
\delta W_{j}(r_{1}) = \int d\mathbf{r}_2 \rho_{i}^{\ast}(r_{12}) \delta \Psi(r_{2})
\]

and we have

\[
\rho_{i}^{\ast}(r_{12}) \equiv \rho_{i}(r_{12}) - \rho_{i}^{\text{lin}}(r_{12}) \].
so from eqn (44) it follows that
\[ \rho^\text{tot}_i(r|\mathbf{R}_1) = \left\{ \begin{array}{ll} \sigma_i(r|\mathbf{R}_1) + \epsilon_0 k_D^2 \tilde{\psi}_i(r|\mathbf{R}_1), & \text{inside} \\
\frac{1}{2} \sum_j \beta_j^2 q^*_j \tilde{\psi}_j(r|\mathbf{R}_1) + \text{other nonlinear terms}, & \text{outside} \end{array} \right. \]
\[ \text{(PB)} \]

since the linear term in \( \rho^\text{tot}_i \) is canceled by \( -\rho^\text{lin}_i \) outside the particle. We see that \( \rho^*_i \) is equal to the nonlinear terms of \( \rho^\text{lin}_i \) on the outside. Since \( \psi_i(r|\mathbf{R}_1) \) and \( \rho^\text{tot}_i(r|\mathbf{R}_1) \) in the PB approximation decay as \( e^{-\kappa_D r} \) when \( r \to \infty \), we see that \( \rho^*_i(r|\mathbf{R}_1) \) decays as \( (e^{-\kappa_D r})^3 \). Thus \( \rho^*_i \) has a considerably shorter range than \( \rho^\text{lin}_i \); the former has half the decay length of the latter.

As noted earlier, these results in the PB approximation apply only to a single particle in an electrolyte. The surrounding ions are treated as bare point ions that do not correlate with each other, so they do not have any ion clouds of their own and no dresses. For these ions the potential of mean force is \( \tilde{\psi}_i(r|\mathbf{R}_1) \) as assumed in the exponent of eqn (55). When the distribution of ions around an ion is calculated in the PB approximation, \( i.e. \), the pair distributions, one accordingly treats the latter ion differently than the rest. This unequal treatment leads to an infamous feature that \( \tilde{\psi}_i \neq \tilde{\psi}_j \) in the (nonlinear) PB approximation.

As we saw in the simple example in Fig. 1 based on the approximation in eqn (2), if we instead treat all ions on an equal basis so all have dresses and, as we will see, therefore have effective charges different from the bare charges, the consequences are quite dramatic with the appearance of multiple screening parameters and oscillatory decay instead of a simple screening parameter \( \kappa_D \).

3.2 The screened Coulomb potential in the general case

3.2.1 The general Green’s function for screened Coulomb interactions. So far we have seen that the dressed particle charge–density \( \rho^*_i \) plays an important role in the free energy of interaction of a particle with the total electrostatic potential in the linear domain [eqn (41)]. Furthermore \( \rho^*_i \) determines the polarization response function \( \chi^* \) [eqn (46) and (54)] and the dielectric function \( \varepsilon(k) \) [eqn (50)]. We will now see that the dressed particle charge–density has yet another important role, namely as the source of the screened electrostatic potential from a particle.

The mean electrostatic potential \( \psi_i(r|\mathbf{R}_1) \) from a particle with coordinates \( \mathbf{R}_1 \) can be obtained from the total charge density \( \rho^\text{tot}_i \) associated with the particle via Coulomb’s law as expressed in eqn (38). This potential satisfies Poisson’s equation
\[ -\epsilon_0 \nabla^2 \psi_i(r|\mathbf{R}_1) = \rho^\text{tot}_i(r|\mathbf{R}_1) \]
\[ \text{(57)} \]
with the boundary condition \( \psi_i(r|\mathbf{R}_1) \to 0 \) when \( r \to \infty \). By subtracting \( \rho^\text{lin}_i(r|\mathbf{R}_1) \) from both sides of Poisson’s equation we obtain
\[ -\epsilon_0 \nabla^2 \psi_i(r|\mathbf{R}_1) - \rho^\text{lin}_i(r|\mathbf{R}_1) = \rho^\text{tot}_i(r|\mathbf{R}_1) - \rho^\text{lin}_i(r|\mathbf{R}_1) \equiv \rho^*_i(r|\mathbf{R}_1) \]
so we have, using eqn (43),
\[ -\epsilon_0 \nabla^2 \psi_i(r|\mathbf{R}_1) - \int \nabla' \psi_i(r'|\mathbf{R}_1) \chi^*(|r-r'|) = \rho^*_i(r|\mathbf{R}_1). \]
\[ \text{(58)} \]

The solution \( \psi_i \) of this equation, which is linear in \( \psi_i \), can be written in terms of the Green’s function of the equation. The Green’s function, which we denote by \( \phi^\text{Coul}_i(r) \), is by definition the solution of
\[ -\epsilon_0 \nabla^2 \phi^\text{Coul}_i(r) - \int \nabla' \phi^\text{Coul}_i(r') \chi^*(|r-r'|) = \delta^3(r). \]
\[ \text{(59)} \]
and the solution of eqn (58) is thereby given by
\[ \psi_i(r|\mathbf{R}_1) = \int \nabla' \phi^\text{Coul}_i(r'|\mathbf{R}_1) \phi^\text{Coul}_i(|r-r'|) \]
\[ \text{(60)} \]
as can be verified by inserting this expression into eqn (58) and using eqn (59). The Green’s function \( \phi^\text{Coul}_i(r) \), which accordingly is defined by eqn (59), is called the (unit) screened Coulomb potential for the general case. It has a key role in the spatial propagation of electrostatic interactions in the electrolyte.

Incidentally we note that in the PB approximation, where \( \chi^*(r) \) given by eqn (30), eqn (59) becomes
\[ -\epsilon_0 \nabla^2 \phi^\text{Coul}_i(r) - \kappa_D^2 \phi^\text{Coul}_i(r) = \delta^3(r) \]
\[ \text{(PB)} \]
which has the solution
\[ \phi^\text{Coul}_i(r) = \frac{e^{-\kappa_D r}}{4\pi\epsilon_0 r} \]
\[ \text{(62)} \]
a monotonic Yukawa function. Therefore
\[ \psi_i(r|\mathbf{R}_1) = \frac{1}{4\pi\epsilon_0} \int \nabla' \phi^\text{Coul}_i(r'|\mathbf{R}_1) \frac{e^{-\kappa_D |r-r'|}}{|r-r'|} \]
\[ \text{(PB)} \]
in this approximation.

In the general case, by taking the Fourier transform of eqn (59) we obtain
\[ \tilde{\phi}^\text{Coul}_i(k) = \frac{1}{\epsilon_0 k^2 - \chi^*(k)} = \frac{\tilde{\phi}^\text{Coul}_i(k)}{1 - \phi^\text{Coul}_i(k) \tilde{\chi}(k)} = \frac{\tilde{\phi}^\text{Coul}_i(k)}{\tilde{\chi}(k)}, \]
\[ \text{(64)} \]
where we have used eqn (34). If we expose the electrolyte to an external electrostatic potential \( \delta \Phi^\text{ext}(r) = \delta q \phi^\text{Coul}(r) \equiv \Phi^\text{ext}_q(r) \), which is the potential at distance \( r \) from a small point charge \( \delta q \), we see from eqn (32) with \( \delta \Phi^\text{ext}(k) = \delta q \phi^\text{Coul}(k) \) that the resulting total potential is \( \delta \Phi(r) = \delta q \phi^\text{Coul}(r) \equiv \Phi^\text{ext}_q(r) \) at least if \( r \) is sufficiently large and \( \delta q \) is sufficiently small, so the linear response is adequate. In this sense we may say that
\[ \phi^\text{Coul}_i(r) = \lim_{\delta q \to 0} \frac{\Phi^\text{ext}_q(r)}{\delta q} \]
for \( r > 0 \).

The right hand side (rhs) of eqn (60) has the same form as Coulomb’s law in eqn (38), which is no surprise since the usual (unscreened) Coulomb potential \( \phi^\text{Coul}(r) \) is a Green’s function of Poisson’s equation (57). Note that \( \psi_i(r|\mathbf{R}_1) \) given by eqn (38) and (60) is exactly the same for all \( r \); what differs in the two equations is the following: when \( \psi_i \) is expressed in terms of \( \phi^\text{Coul}_i(r) \) the source is \( \rho^\text{tot}_i \) while when it is expressed in terms of \( \phi^\text{Coul}_i(r) \) the source is \( \rho^*_i \).
We can write eqn (38) and (60) in Fourier space by using the notation in eqn (47), whereby we obtain
\[
\tilde{\rho}_i (k, \omega_1) = \frac{1}{\tilde{\rho}_0 (k, \omega_1)} \tilde{\phi}_{\text{Coul}} (k) \tilde{\rho}_i (k, \omega_1) \tilde{\phi}_{\text{Coul}}^* (k).
\] (65)

By using \( \tilde{\phi}_{\text{Coul}}^* (k) = \tilde{\phi}_{\text{Coul}} (k) \tilde{\omega} (k) = 1 / |\omega_k|^2 \tilde{\omega} (k) \) we see from the last equality that
\[
\tilde{\rho}_i (k, \omega_1) = \frac{1}{\tilde{\rho}_0 (k, \omega_1)} \tilde{\omega} (k)
\] (66)
when the factor \(1/|\omega_k|^2\) has been removed. This simple relationship can also be derived from eqn (40).

In the PB approximation where \( \tilde{\omega} (k) \) is given by eqn (36) we have
\[
\tilde{\phi}_{\text{Coul}}^* (k) = \tilde{\phi}_{\text{Coul}} (k) \frac{1 + \frac{\kappa D}{k^2}}{1 + \frac{\kappa D}{k^2} \epsilon_0 (k^2 + \kappa D^2)} \quad (\text{PB}),
\] (67)
which is the Fourier transform of eqn (62). The feature that the decay behavior of \( \tilde{\phi}_{\text{Coul}}^* (r) \) in the PB approximation is proportional to \( e^{-\kappa_D r} \) is associated with the fact that the denominator in eqn (67) is zero when \( k = \pm i \kappa_D \), that is, \( \tilde{\phi}_{\text{Coul}}^* (k) \) has simple poles at \( k = \pm i \kappa_D \). This holds in general,\(^\dagger\dagger\) so a pair of simple zeros \( k = \pm i \kappa_D \) of the denominator of eqn (64) corresponds to a term in \( \tilde{\phi}_{\text{Coul}}^* (r) \) that decays as \( e^{-\kappa(r)} \). At such zeros, i.e., poles of \( \tilde{\phi}_{\text{Coul}}^* (k) \), we have
\[
[\epsilon_0 (k^2 + \kappa D^2) \tilde{\omega} (k)]_{k = \pm i \kappa_D} = 0, \quad \tilde{\omega} (k) = 0,
\] (68)
where the latter holds for \( \kappa = 0 \). Since \( \tilde{\phi}_{\text{Coul}}^* (k) \) is the Fourier transform of a real function of \( r \), \( \tilde{\phi}_{\text{Coul}}^* (k) \) is an even function of \( k \) and therefore it is sufficient to consider a pole at \( i \kappa_D \), whereby the pole at \( -i \kappa_D \) follows.

In order to see what these facts lead to, let us first consider an electrolyte consisting of spherical simple ions in vacuum. From the Fourier transform of eqn (54) we see that \( \tilde{\phi}_{\text{Coul}}^* (k) \) has a pole \( k = i \kappa_D \) when
\[
[\epsilon_0 (k^2 + \kappa D^2) \tilde{\omega} (k)]_{k = i \kappa_D} = -\epsilon_0 \kappa_D^2 + \beta \sum_i n_i^b q_i \tilde{\rho}_i^* (i \kappa_D) = 0,
\]
which implies that
\[
\kappa_D^2 = \frac{\beta}{\epsilon_0} \sum_i n_i^b q_i \tilde{\rho}_i^* (i \kappa_D)
\] (69)
where we have used the Fourier transform of eqn (52) and the definition (35) of \( \kappa_D \). It follows from eqn (69) that the dresses of the ions, which describe the effects of ion-ion correlations, make \( \kappa \neq \kappa_D \).

When the density of the ions goes to zero, the charge density \( \rho_i (r) \) of the ion cloud goes to zero for each \( r \) and likewise the linear part \( \tilde{\rho}_i^\text{lin} (r) \), so \( \tilde{\rho}_i^\text{dress} \equiv \tilde{\rho}_i - \tilde{\rho}_i^\text{lin} \) goes to zero. Thus \( \kappa / \kappa_D \rightarrow 1 \) in this limit and the PB result is approached. Likewise, in the same limit we have \( \delta W (r_i) \approx q_i \delta q (r_i) \) from eqn (53), so the PB approximation is reasonable at least where the electrostatic potential is sufficiently small.

At finite densities the pole at \( k = i \kappa_D \) gives as, we will see, the leading term in the decay of \( \phi_{\text{Coul}}^* \)
\[
\phi_{\text{Coul}}^* (r) \sim A^* \frac{e^{-\kappa_D r}}{4\pi r} \quad \text{when} \quad r \rightarrow \infty,
\] (70)
where \( A^* \) is a constant that will be determined later [eqn (70) holds at least when the ion density is not too high]. In contrast to the PB result (62), this decay formula is only valid asymptotically for large \( r \) because, as we will see, there are other terms in \( \phi_{\text{Coul}}^* (r) \) that decay faster to zero than the contribution on the rhs. They give non-negligible contributions for small \( r \).

3.2.2 The general equation for the screening parameter \( \kappa \) vs. the dielectric function. Let us now focus on the dielectric function \( \tilde{\omega} (k) \) and use the condition \( \tilde{\omega} (0) = 0 \) in eqn (68) for the pole of \( \tilde{\phi}_{\text{Coul}}^* (k) \). Since \( \tilde{\omega} (k) = 1 - \chi^* (k) \tilde{\phi}_{\text{Coul}} (k) \) we can write
\[
\tilde{\omega} (k) = 1 - \frac{\chi^* (k)}{\epsilon_0 k^2} \frac{(\kappa_D^2 + \kappa D^2)}{\epsilon_0 k^2 - \kappa D^2}
\]
(71)
which is singular when \( k \rightarrow 0 \) (provided that \( \chi^* (0) \neq 0 \)), and
\[
\tilde{\omega}_{\text{reg}} (k) = 1 - \frac{\chi^* (k) - \chi^* (0)}{\epsilon_0 k^2}
\]
(73)
which is regular (non-singular) at \( k = 0 \) since its value there is
\[
\tilde{\omega}_{\text{reg}} (0) = 1 + \frac{1}{6 \epsilon_0} \int \text{d} \mathbf{r}^2 \chi^* (\mathbf{r}).
\]
These integrals converge provided \( \chi^* (r) \) decays to zero sufficiently rapidly with increasing \( r \).

For the case of spherical simple ions we have from the Fourier transform of eqn (54)
\[
\tilde{\chi}^* (0) = -\beta \sum_i n_i^b q_i \tilde{\rho}_i^* (0) = -\beta \sum_i n_i^b q_i q_i^* \quad (\text{spherical ions}),
\] (74)
where
\[
q_i^* = \int \text{d} \mathbf{r}^* \tilde{\rho}_i^* (r) = \tilde{\rho}_i^* (0) \quad (\text{spherical ions})
\] (75)
is the dressed particle charge of a particle of species \( i \), that is, the total charge of the dressed particle charge–density \( \rho_i^* \). This charge consists of the charge \( q_i \) of the ion and the total charge of its dress. Note that
\[
q_i^* = \int \text{d} \mathbf{r}^* \tilde{\rho}_i^* (r) = \int \text{d} \mathbf{r}^* \tilde{\rho}_i^\text{tot} (r) - \int \text{d} \mathbf{r}^* \tilde{\rho}_i^\text{lin} (r) = -\int \text{d} \mathbf{r}^* \tilde{\rho}_i^\text{lin} (r)
\]
because the total charge of $\rho_0^\text{tot}(r)$ is zero due to the local electroneutrality condition. Since $\rho_0^\text{lin}$ is the linear part of the polarization response due to $\psi$, from the ion, $q_i^*$ is normally nonzero and has the same sign as $q_i$. It is, however, possible for $q_i^*$ for a species to change sign and thereby attain the opposite sign to $q_i$, so in rare cases $q_i^*$ for a species can fortuitously be zero for certain parameter values of the system. One can show that $\sum_i n_i^b q_i^* > 0$ for electrolytes, so it is not possible to have $q_i^* < 0$ for all ionic species simultaneously. In Appendix A of ref. 31 it is described how $\rho_0^\text{tot}(r)$ and $q_i^*$ can be calculated from $\rho_0^\text{tot}(r)$ for all $j$ or from the pair distribution functions $g_{ij}(r)$.

For nonspHERical particles eqn (49) yields for $k = 0$

$$\chi^0(0) = - \beta \sum_i n_i^b \langle \hat{\sigma}_i(r, \omega_3) \hat{\sigma}_i^*(0, \omega_3) \rangle_{\omega_3}$$

$$= - \beta \sum_i n_i^b \langle q_i q_i^* \rangle_{\omega_3} = - \beta \sum_i n_i^b q_i q_i^*,$$

(76)

where we have used

$$q_i^* = \int \, \mathrm{d}r \rho_i(r, \omega_3) = \rho_i^*(0, \omega_3),$$

which is independent of the orientation $\omega_3$, and the corresponding relationship for $\sigma_i$. Again, $q_i^*$ is the total charge of the dressed particle charge–density $\rho_i^*$. Thus we always have $\chi^0(0) = - \beta \sum_i n_i^b q_i q_i^*$, where $q_i^*$ is the dressed particle charge.

By inserting this result into eqn (72) we obtain

$$\tilde{e}_{\text{sing}}(k) = \frac{\beta \sum_i n_i^b q_i q_i^*}{\varepsilon_0 k^2}. $$

(77)

Due to the presence of $\tilde{e}_{\text{sing}}(k)$, the dielectric function $\tilde{e}(k)$ for electrolytes diverges to infinity when $k \to 0$. This divergence is commonly called perfect screening.

Using eqn (77) we see that the condition for a pole, $\tilde{e}(\text{i}k) = \tilde{e}_{\text{reg}}(\text{i}k) + \tilde{e}_{\text{sing}}(\text{i}k) = 0$, is

$$\tilde{e}_{\text{reg}}(\text{i}k) = - \frac{\beta \sum_i n_i^b q_i q_i^*}{\varepsilon_0 k^2} = 0,$$

which can be written

$$\kappa^2 = \frac{\beta \sum_i n_i^b q_i q_i^*}{\tilde{e}_{\text{reg}}'(\text{i}k) \varepsilon_0}, $$

(78)

where

$$\tilde{e}_{\text{reg}}'(\text{i}k) = \tilde{e}_{\text{reg}}(\text{i}k),$$

(79)

is the dielectric factor. Eqn (78) is the general equation for the screening parameter $\kappa$, eqn (5). It is obviously equivalent to $\tilde{e}(\text{i}k) = 0$. Note that eqn (78), which is an exact equation for $\kappa$ with general applicability, has an appearance very similar to the definition of the Debye parameter $\kappa_D$; only the factor $q_i^*$ and the presence of $\tilde{e}_{\text{reg}}'(\text{i}k)$ in the denominator differ. From eqn (73) it follows that

$$\tilde{e}_{\text{reg}}'(\text{i}k) = 1 + \frac{1}{\varepsilon_0} \int \mathrm{d}r^2 \left[ \frac{\sinh(kr)}{(kr)^3} \right] \tilde{\chi}'(r).$$

(80)

As we will see, the fact that $\tilde{e}_{\text{reg}}'(\text{i}k)$ is a function of $\kappa$ is crucial for the understanding of the properties of electrolytes. When the particle density goes to zero, we have $q_i^* \to q_i$ and $\tilde{e}_{\text{reg}}'(\text{i}k) \to 1$ with screening parameter $\kappa$ near zero, so eqn (78) approaches the expression for $\kappa_D$ in eqn (35) which implies that $\kappa_D / \kappa_D \to 1$ in this limit. For electrolyte models with a dielectric continuum solvent, the initial 1 on the rhs of eqn (80) should be replaced by $\varepsilon_r$ for the solvent.

For a binary electrolyte, the deviation in $\kappa$ from $\kappa_D$ can be obtained from

$$\left[ \frac{\kappa}{\kappa_D} \right]^2 = \frac{q_{+}^* - q_{-}^*}{(q_{+} + |q_{-}|) \tilde{e}_{\text{reg}}'(\text{i}k)}$$

(81)

which follows from eqn (35) and (78) and the fact that $n_i^b q_i = -n_i^b q_i^*$. Note that $\kappa_D$ used here is calculated for particles in vacuum.

#### 3.2.3 Decay modes, multiple screening parameters and effective dielectric permittivities

Let us now determine the coefficient $A^*$ in eqn (70). From eqn (64) and the results above we see that

$$\tilde{e}_{\text{reg}}'(\text{i}k) \approx \frac{1}{\varepsilon_0 k^2 \tilde{e}_{\text{reg}}(\text{i}k) + \varepsilon_{\text{sing}}(\text{i}k)} \approx \frac{1}{\varepsilon_0 k^2 \tilde{e}_{\text{reg}}(\text{i}k) + \beta \sum_i n_i^b q_i q_i^*}.$$ 

Since the denominator is zero for $k = \pm \text{i} \xi$ we can write

$$\tilde{e}_{\text{reg}}'(\text{i}k) = \frac{1}{k^2 + \kappa^2} \varepsilon_0 k^2 \tilde{e}_{\text{reg}}(\text{i}k) + \beta \sum_i n_i^b q_i q_i^*$$

$$\sim \frac{A^*}{k^2 + \kappa^2} \text{ when } k \to \text{i} \xi$$

where $A^*$ is given by

$$A^* = \lim_{k \to \text{i} \xi} \frac{k^2 + \kappa^2}{\varepsilon_0 k^2 \tilde{e}_{\text{reg}}(\text{i}k) + \beta \sum_i n_i^b q_i q_i^*} = \frac{2k}{\varepsilon_0 [2\varepsilon_{\text{reg}}(\text{i}k) + k^2 \tilde{e}_{\text{reg}}'(\text{i}k)]} \bigg|_{k = \text{i} \xi}. $$

Here $\tilde{e}_{\text{reg}}'(\text{i}k) = \tilde{e}_{\text{reg}}'(\text{i}k) / \mathrm{d}k$ and we have used l'Hospital’s rule to obtain the last equality. If we define

$$\tilde{e}_{\text{reg}}'(\text{i}k) = \tilde{e}_{\text{reg}}(\text{i}k) + \frac{\mathrm{i} k}{2} \tilde{e}_{\text{reg}}'(\text{i}k),$$

(82)

it follows that $A^* = 1 / [\varepsilon_0 \tilde{e}_{\text{reg}}'(\text{i}k)]$ and hence we have

$$\phi_{\text{Coul}}(r) \sim \frac{q_{-}^* - q_{+}^*}{4\pi \varepsilon_0 \tilde{e}_{\text{reg}}'(\text{i}k)} \text{ when } r \to \infty.$$ 

(83)

By comparing with the PB result in eqn (62) we see that a correct treatment of the correlations between all particles in the system has given rise to a change in magnitude of $\phi_{\text{Coul}}(r)$ for large $r$ by a factor $1 / \tilde{e}_{\text{reg}}'(\text{i}k)$. In addition there is a change in the
value of the screening parameter from $\kappa_D$ to $\kappa$. Note that $\varepsilon_{\text{eff}}^*(k)$ differs from $\varepsilon_{\text{eff}}(k)$ by the last term in eqn (82).

Alternative expressions for $\varepsilon_{\text{eff}}^*(k)$ are

$$\varepsilon_{\text{eff}}^*(k) = \frac{\alpha}{\beta} \left[ \frac{k^2 \varepsilon_{\text{reg}}(k)}{2k} \right]_{k=\text{i}c} - \frac{k^2 \varepsilon(k)}{2k} \right]_{k=\text{i}c},$$

where the first expression yields eqn (82) after the differentiation and where we have used $\varepsilon(\text{i}c) = 0$ to obtain the last equality. From eqn (73) and the first equality in eqn (84) we readily obtain

$$\varepsilon_{\text{eff}}^*(k) = 1 + \frac{1}{2\alpha_0} \left( \int \mathrm{d}r' \left[ k^2 \varepsilon(r) \cosh(kr) - \sinh(kr) \right] \right) ^\prime(r),$$

which can be compared with eqn (80). When the density of the particles in the electrolyte goes to zero, we have $\varepsilon_{\text{eff}}^*(k) \to 1$. Since $\kappa/\kappa_D \to 1$ in this limit, eqn (83) approaches the PB result for $\phi_{\text{coul}}(r)$ in eqn (62), at least for sufficiently large $r$. (For electrolyte models with a dielectric continuum solvent, the initial 1 on the rhs should be replaced by $\varepsilon_I$ for the solvent.)

These results can be applied to the case of an electrolyte solution with a molecular solvent consisting of polar, uncharged molecules. Since such molecules have $q_I = 0$ they do not contribute to $\varepsilon_{\text{sing}}(k)$, so the sum in eqn (77) runs in practice over the ionic species only. Furthermore, since these molecules have no net charge they do not contribute to $q_I^*$ for any species $i$. In the expression for $\kappa$, eqn (78), the solvent molecules solely contribute to the dielectric factor $\varepsilon_{\text{eff}}^*(k)$ [part from their indirect influence on the distribution functions and other entities, of course].

For a pure polar solvent without ions, $\varepsilon_{\text{sing}}(k)$ vanishes so $\varepsilon(k) = \varepsilon_{\text{reg}}(k)$. As mentioned earlier the dielectric constant of a pure polar medium is defined microscopically as $\varepsilon_I = \lim \varepsilon(k)$, so we have $\varepsilon_I = \varepsilon(0) = \varepsilon_{\text{reg}}(0)$. In a very dilute electrolyte solution $\kappa \approx 0$ and hence $\varepsilon_{\text{eff}}^*(k) \approx \varepsilon_{\text{eff}}(k) \approx \varepsilon_I$. Eqn (78) then becomes

$$\kappa^2 \approx \frac{\beta \sum_i n_i^b q_i q^*}{\varepsilon_{\text{reg}}(0)} = \frac{\varepsilon_I}{\varepsilon_{\text{reg}}(0)} \approx \kappa_D^2 \quad \text{(dilute solutions)},$$

where we have used the fact that $q_I^* \approx q_I$ for a dilute electrolyte. In the last equality we have also identified $\kappa_D^2$ for the electrolyte solution when the solvent is a dielectric continuum with dielectric constant $\varepsilon_I$, eqn (1).

In dilute solutions we have $\varepsilon_{\text{eff}}^*(k) \approx \varepsilon_{\text{eff}}^*(0) \approx \varepsilon_I$ since the last term in eqn (82) vanishes when $\kappa \to 0$ because of the prefactor $i \kappa$. This implies that we approximately have when $r \to \infty$

$$\phi_{\text{coul}}(r) \approx \frac{e^{-\kappa r}}{4\pi\varepsilon_0 e r} \quad \text{(dilute solutions)}$$

with $\kappa \approx \kappa_D$, that is, the same as in the PB approximation when the pure solvent has dielectric constant $\varepsilon_I$.

For electrolyte solutions in general, $\varepsilon_{\text{eff}}^*(k)$ and $\varepsilon_{\text{eff}}^*(k)$ contain, as we have seen, contributions from the ions. Since $1/\varepsilon_{\text{eff}}^*(k)$ determines the magnitude of the screened Coulomb potential for large $r$, $\varepsilon_{\text{eff}}^*(k)$ can be designated as the effective relative dielectric permittivity of the entire electrolyte solution. Likewise, the dielectric factor $\varepsilon_{\text{eff}}^*(k)$ acts as a kind of relative dielectric permittivity of the solution since it takes the role that the dielectric constant of the pure solvent has in the expression for $\kappa_D$. Remember that $\varepsilon_{\text{eff}}^*(k)$ and $\varepsilon_{\text{eff}}^*(k)$ are different from each other in general.

The only difference between the general equation for the screening parameter $\kappa$, eqn (78), and the definition (35) of the Debye parameter $\kappa_D$ is, apart from the factor $\varepsilon_{\text{eff}}^*(k)$, that the former contains the factor $q_I q_I^*$ instead of $q_I^*$. While $\kappa_D$ is given solely in terms of the system parameters $n_i$, $q_i$, and $T$, the actual screening parameter $\kappa$ depends on the state-dependent entities $q_I^*$ and $\varepsilon_{\text{eff}}^*(k)$. The latter are, as we have seen, defined in terms of dressed charge densities or, equivalently, in terms of $\rho_i^\text{tot}$ for all $i$ via its relationship to $\rho_i^*$ as explained earlier.

In the expression (78) for $\kappa$, the dressed ion charge $q_I^*$ is a constant for each system with given system parameters, but the dielectric factor $\varepsilon_{\text{eff}}^*(k)$ is a function of $\kappa$. The latter fact makes a huge difference compared to the predictions of the PB approximation. While this approximation provides a unique screening parameter $\kappa_D$ from eqn (35), the exact equation, eqn (78), is an equation for $\kappa$. Apart from the solution $\kappa_\text{sol}$, which gives the longest decay length, eqn (78) has in general several other solutions $\kappa_\text{sol}, \kappa_\text{sol}^\prime$ etc. (the three solutions can alternatively be denoted $\kappa_\text{sol}$, for $\nu = 1, 2$ and 3). Each solution gives rise to a term in $\phi_{\text{coul}}(r)$ like the rhs of eqn (83), so we have

$$\phi_{\text{coul}}(r) = \frac{1}{4\pi\varepsilon_0} \left[ \frac{e^{-\kappa r}}{\varepsilon_{\text{eff}}^*(k)r} + \frac{e^{-\kappa r}}{\varepsilon_{\text{eff}}^*(k_\text{sol})r} + \frac{e^{-\kappa r}}{\varepsilon_{\text{eff}}^*(k_\text{sol}^\prime)r} \right] + \text{other terms},$$

where the “other terms” are analogous Yukawa function terms with shorter decay lengths [i.e., with other $\kappa$ values that are solutions to eqn (78)] and functions with different functional dependences of $r$ (more about this later). Note that each Yukawa term has its own value of $\varepsilon_{\text{eff}}^*(k)$.

Furthermore, solutions to eqn (78) can be complex-valued, in the case in which there are always two solutions that are complex conjugates to each other, say, $\kappa = \kappa_\text{sol} + \text{i} \kappa_\text{sol}^\prime$ and $\kappa = \kappa_\text{sol} + \text{i} \kappa_\text{sol}^\prime$, where $\kappa_\text{sol}$ and $\kappa_\text{sol}^\prime$ are real. Since the sum of a complex number $Z$ and its complex conjugate $Z^\prime$ is given by $Z + Z^\prime = 2\Re(Z)$, where $\Re(Z)$ stands for the real part of $Z$, we then have

$$\frac{1}{4\pi\varepsilon_0} \left[ \frac{e^{-\kappa r}}{\varepsilon_{\text{eff}}^*(k)r} + \frac{e^{-\kappa r}}{\varepsilon_{\text{eff}}^*(k_\text{sol})r} \right] = \frac{1}{2\pi\varepsilon_0} \Re \left[ \frac{e^{-\kappa r + \text{i} \kappa_\text{sol} r}}{\varepsilon_{\text{eff}}^*(k)r} \right] = \frac{1}{2\pi\varepsilon_0} \left[ \frac{e^{-\kappa_\text{sol}(\kappa_\text{sol}^\prime + \text{i}) r}}{\varepsilon_{\text{eff}}^*(k)r} \right] = \frac{\cos(\kappa_\text{sol} r - \text{i} \kappa_\text{sol}^\prime r)}{r},$$

where we have written $\varepsilon_{\text{eff}}^*(k_\text{sol} + \text{i} \kappa_\text{sol}^\prime) = |\varepsilon_{\text{eff}}^*(k)| e^{-\text{i} \kappa_\text{sol}^\prime r}$ with a real $\theta_\text{sol}$. The two poles hence give rise to an exponentially decaying, oscillatory term with decay length $1/\kappa_\text{sol}$, wavelength $2\pi\kappa_\text{sol}$ and phase shift $-\theta_\text{sol}$. This can be shown by contour integration and residue calculus for $\phi_{\text{coul}}(k)$ in complex $k$-space.
We will denote such a function as an “oscillatory Yukawa function.” The function $e^{-\kappa r}$ with real $\kappa$ will be designated as a “monotonic Yukawa function.”

Thus there exist several decay modes for the screened electric potential, some that give monotonic and others that give oscillatory decay. In the limit of low ionic densities, one of the modes approaches the single mode that is included in the PB approximations. Henceforth, we will use the term “screening parameters” to denote a set like $\kappa, \kappa', \kappa''$ etc. that are solutions to eqn (78), so this term includes the inverse decay lengths $k$ or $k_R$ and the inverse wavelengths $\kappa\sqrt{2\pi}$ of all monotonic and oscillatory Yukawa function contributions to $\phi_{\text{Coul}}^*$. More generally, the concept of “decay parameters” denotes the inverse decay lengths and inverse wavelengths of all kinds of contributions to the various functions.

A well-known example of the occurrence of an oscillatory term as in eqn (87) is the Kirkwood cross-over\(^\text{13}\) mentioned in the Introduction, where two real solutions turn into two complex conjugate solutions when a system parameter like the ion density is changed. We take the example of two real solutions $\kappa$ and $\kappa'$, where $\kappa$ is the smallest and $\kappa'$ is the second smallest solution, i.e., for large $r$ they give the two leading terms in $\phi_{\text{Coul}}^*(r)$ as given by eqn (86) and we have

$$\phi_{\text{Coul}}^*(r) \sim \frac{1}{4\pi\alpha_0} \left[ e^{-\kappa r} r^\epsilon_{\text{eff}}(\kappa') + e^{-\kappa' r} r^\epsilon_{\text{eff}}(\kappa) \right] \text{ when } r \to \infty \quad (88)$$

with wavelengths $1/k > 1/k'$. For low ionic densities we have seen that $\kappa \approx \kappa_0$ and when the ionic density is increased, the two solutions $\kappa$ and $\kappa'$ of eqn (78) approach each other as in the example of Fig. 1 and then merge when the density reaches the cross-over point. For even higher densities, $\kappa$ and $\kappa'$ become two complex conjugate solutions, so the leading term for large $r$ is oscillatory as shown in eqn (87).

Recall that eqn (78) is equivalent to $\delta(\xi) = 0$, so before the cross-over $\kappa$ and $\kappa'$ are two consecutive zeros of $\delta(\xi)$ as a function of the real variable $\xi$. As we have seen, $\delta_{\text{eff}}^{(2)}(\kappa) > 0$ for low ion densities, so from the rhs of eqn (84) we see that $\delta_{\text{eff}}^{(2)}(\xi) > 0$ for $\xi = \kappa$. The next zero of $\delta_{\text{eff}}^{(2)}(\xi)$ must have an opposite derivative, so $\delta_{\text{eff}}^{(2)}(\xi) < 0$ for $\xi = \kappa'$, which implies that $\delta_{\text{eff}}^{(2)}(\kappa') < 0$. Thus the two terms in eqn (88) have opposite signs. At the $\kappa$-cross-over point, where the two zeros of $\delta_{\text{eff}}^{(2)}(\xi)$ merge, we must have $\delta_{\text{eff}}^{(2)}(\kappa) = \delta_{\text{eff}}^{(2)}(\kappa') = 0$ but the sum of the two terms remains finite there.

There may also appear another kind of cross-over between a monotonic and an oscillatory Yukawa function decay, namely the Fisher–Widom cross-over\(^\text{13}\) mentioned in the Introduction. Say that the term with screening parameter $\kappa''$ in eqn (86) initially has a shorter decay length $1/\kappa''$ than the oscillatory term we have just discussed. When the ionic density is increased it is possible that the former term becomes the leading term because $1/\kappa''$ becomes larger than the decay length $1/k_R$ of the latter. This means that the decay behavior of $\phi_{\text{Coul}}^*(r)$ for large $r$ changes from oscillatory to monotonic at the density value where $1/\kappa''$ and $1/k_R$ are equal. This kind of cross-over may, of course, also occur in the reverse direction, i.e., the decay changes from monotonic to oscillatory.

### 3.2.4 Multipolar effective charges.

The mean electrostatic potential $\psi_i$ due to an $i$-particle is given by eqn (60) and for each Yukawa function term in $\phi_{\text{Coul}}^*(r)$ there is a term in $\psi_i$ with the same screening parameter, say $\kappa_i$,

$$\psi_i(r_i | R_i) \text{ contribution } :$$

$$\frac{1}{4\pi\alpha_0} \left[ \delta_{\text{eff}}^{(2)}(\kappa_i) r_i^2 | \frac{e^{-\kappa_i r}}{r} r - r_i | \right]$$

(90)

Each of these contributions are like the single term in the PB result, eqn (63). The screening parameter $\kappa_i$ of some contributions may, as we have seen, be complex-valued and give rise to an oscillatory contribution to $\psi_i$.

Let us investigate some consequences of eqn (89) and we start with real $\kappa_i$. We will use the fact that

$$e^{-\kappa r} \sim \frac{1}{r} e^{-\kappa r} \quad \text{when } r' \ll r \to \infty,$$

where $r = r_i$ and $r' = r_{i3}$ (note that $r - r' = r_i - r_i$). We will use the fact that $\kappa_i$ is the smallest and $\kappa'_{i3}$ is the second smallest

$$\psi_i(r_{i2}, \omega_1) \text{ contribution } :$$

$$\frac{1}{4\pi\alpha_0} \left[ e^{-\kappa_{i2} r} \delta_{\text{eff}}^{(2)}(\kappa_{i2}) r_{i2}^2 | \frac{e^{-\kappa_{i2} r}}{r} r - r_{i2} | \right]$$

(90)

provided that $\rho_i$ decays sufficiently rapidly with distance. Thus each contribution decays like a Yukawa function with distance $r$ but has a magnitude that is different depending on the direction of the vector $r_{i2}$, whereby the integral contains the direction dependence as expressed via $r_{i3}$ in the exponent.

For a spherically symmetric particle, the integral becomes

$$\int r_i^2 r_{i2} e^{-\kappa_{i2} r} = \int r_i^2 r_{i2} \sinh(\kappa_{i2} r_i) = q_i^{\text{eff}}(\kappa_{i2})$$

(91)

and we obtain from the terms in eqn (86) in the limit $r_{i2} \to \infty$

$$\psi_i(r_{i2}, \omega_1) = \frac{1}{4\pi\alpha_0} \left[ \frac{q_i^{\text{eff}}(\kappa_{i2}) e^{-\kappa_{i2} r_{i2}}}{\delta_{\text{eff}}^{(2)}(\kappa_{i2}) r_{i2}} + \frac{q_i^{\text{eff}}(\kappa') e^{-\kappa' r_{i2}}}{\delta_{\text{eff}}^{(2)}(\kappa') r_{i2}} \right]$$

+ other terms,

where $q_i^{\text{eff}}$ is an “effective charge” of the $i$-particle. Note that each decay mode has its own value of the effective charge $q_i^{\text{eff}}(\kappa_i)$ depends on $\kappa_i$. Incidentally we also note that $q_i^{\text{eff}}$ is different from the dressed particle charge $q_i^*$, which is a constant that is independent of $\kappa_i$. 


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In the general case of nonspherical particles eqn (90) implies that

$$
\psi_i(\mathbf{r}_{12}, \omega_1) \sim \frac{1}{4\pi \varepsilon_0} \left[ \frac{Q_i^{\text{eff}}(\mathbf{r}_{12}, \omega_1, \mathbf{k})}{\varepsilon_r^{\text{eff}}(\mathbf{k}) r_{12}} + \frac{Q_i^{\text{eff}}(\mathbf{r}_{12}, \omega_1, \mathbf{k}^\prime)}{\varepsilon_r^{\text{eff}}(\mathbf{k}^\prime) r_{12}} + \frac{Q_i^{\text{eff}}(\mathbf{r}_{12}, \omega_1, \mathbf{k}^\prime\prime)}{\varepsilon_r^{\text{eff}}(\mathbf{k}^\prime\prime) r_{12}} + \text{other terms}, \right]
$$

where

$$
Q_i^{\text{eff}}(\mathbf{r}_{12}, \omega_1, \mathbf{k}_i) \equiv \int d\mathbf{r} \rho_i^\star(\mathbf{r}, \omega_1) e^{i\mathbf{k}_i \cdot \mathbf{r}}
$$

is a direction dependent entity that has the unit of charge. We may call it a "multipolar effective charge" of the particle, where "multipolar" indicates that it is direction dependent. Each decay mode has its own value since $Q_i^{\text{eff}}$ depends on $k_i$. The orientation independent part of the potential, the monopolar part, is given by the average

$$
\psi_i(\mathbf{r}_{12}) = \langle \psi_i(\mathbf{r}_{12}, \omega_1) \rangle_{\omega_1}
$$

$$
\sim \frac{1}{4\pi \varepsilon_0} \left[ \frac{Q_i^{\text{eff}}(\mathbf{k})}{\varepsilon^{\text{eff}}(\mathbf{k}) r_{12}} + \text{other terms} \right]
$$

where

$$
Q_i^{\text{eff}}(\mathbf{k}) = \langle Q_i^{\text{eff}}(\mathbf{k}, \omega, \mathbf{k}) \rangle_{\omega}
$$

is the orientation average of the effective multipolar charge. $Q_i^{\text{eff}}$ can be described as a kind of monopolar effective charge of the particles. The rest of $Q_i^{\text{eff}}(\mathbf{r}_{12}, \omega_1, k_i)$ has an orientational angle dependence with a combination of dipolar, quadrupolar and higher multipolar characteristics.\textsuperscript{34} Note that electroneutral particles, like solvent molecules, acquire nonzero $\tilde{Q}_i^{\text{eff}}$ due to the presence of ions in their neighborhood, so they have nonzero effective charges in general.

For a pair of complex-valued screening parameters, there is an oscillatory Yukawa term in $\psi_i$ with a direction dependent coefficient that can be determined from eqn (93) in a similar manner as in eqn (87).

Since the theory is valid for particles of any size and shape, the result in eqn (93) is also applicable to macroscopic particles. Therefore the decay of the potential from, for example, a planar surface is also given by the same decay modes. In such cases the effective particle charges can be translated into effective surface charge densities by dividing by the surface area.\textsuperscript{30}

The results in eqn (93) and (94) can alternatively be obtained in Fourier space where $\psi_i$ is given by [eqn (65)]

$$
\tilde{\psi}_i(\mathbf{k}, \omega_1) = \tilde{\psi}_i(\mathbf{k}, \omega_1) \tilde{\phi}_{\text{Coal}}(\mathbf{k}) = \frac{\tilde{\psi}_i(\mathbf{k}, \omega_1)}{\tilde{\phi}(\mathbf{k})_{\omega_1}}.
$$

For each pole $k = i k_i$, of $\tilde{\phi}_{\text{Coal}}(\mathbf{k})$, that is, zero of $\tilde{\phi}(\mathbf{k})$, we obtain a contribution to $\psi_i(\mathbf{r}_{12}, \omega_1)$ that decays as $e^{-k_i \cdot r}$ with the same prefactor $1/(4\pi \varepsilon_0 \varepsilon_r^{\text{eff}}(\mathbf{k}))$ as in eqn (86) times the factor

$$
\tilde{\rho}_i^\star(\mathbf{k}, \omega) \equiv \int d\mathbf{r} \tilde{\rho}_i^\star(\mathbf{r}, \omega) e^{i \mathbf{k} \cdot \mathbf{r}} = Q_i^{\text{eff}}(\mathbf{k}, \omega, \mathbf{k}_i),
$$

where we in ordinary space apply this result with $\mathbf{k} = \mathbf{k}_i$ since the origin is selected at the particle center. We see that $Q_i^{\text{eff}}$ for the mode with screening parameter $k_i$ is the projection of $\rho_i^\star$ on this mode.

There exists an intimate relationship between the screening parameters $k_i$ and the effective charges, i.e., $q_i^{\text{eff}}$ for spherical and $Q_i^{\text{eff}}$ for nonspherical particles. The parameter $k$ is a solution to $\tilde{\phi}(\mathbf{k}) = 0$ and, equivalently, to the general equation for $k$, eqn (78). By inserting $k = i k$ into the expression (50) for $\tilde{\phi}(\mathbf{k})$ we obtain

$$
\tilde{\phi}(\mathbf{k}) = 1 - \frac{\beta}{\varepsilon_0} \sum_i n_i^b \tilde{\phi}(i \mathbf{k}, \omega) \tilde{\phi}(i \mathbf{k}, \omega) = 0 .
$$

Inserting eqn (96) and using $\tilde{\phi}(0) = 0$ we find that

$$
k^2 = \frac{\beta}{\varepsilon_0} \sum_i n_i^b q_i q_i^{\text{eff}}(k) .
$$

(Eqn (97) is an equation for $k$ that is equivalent to eqn (78).)

For a spherical ion with charge $q_i$ at the center we have $\tilde{\phi}(\mathbf{k}) = q_i$ so we have $Q_i^{\text{eff}}(\mathbf{k}, \omega, \mathbf{k}) = q_i$ and $Q_i^{\text{eff}}(\mathbf{k}, \omega, \mathbf{k}) = q_i^{\text{eff}}(\mathbf{k})$. When we deal with a system consisting solely of such ions, eqn (97) becomes [cf. eqn (69)]

$$
k^2 = \frac{\beta}{\varepsilon_0} \sum_i n_i^b q_i q_i^{\text{eff}}(k) .
$$

which is similar to the general eqn (78). The differences are that $q_i^{\text{eff}}(k)$ depends on $k$ while $q_i$ is constant and that $\tilde{\phi}(\mathbf{k})$ in the denominator. Incidentally, we may note that for spherical ions we have $\sum_i n_i^b q_i q_i^{\text{eff}}(k) = \sum_i n_i^b q_i q_i^{\text{eff}}(k)$, but in general $q_i^{\text{eff}}(k) \neq q_i q_i^{\text{eff}}(k)$. For a binary electrolyte of spherical ions, the deviation in $k$ from $k_D$ can be obtained from

$$
\frac{k^2}{k_D^2} = \frac{q_i^{\text{eff}}(k) - q_i^{\text{eff}}(k)}{q_i + |q_i|} \quad \text{(spherical ions),}
$$

which should be compared to eqn (81).

For the special case of a binary symmetric electrolyte $n_i^h = n_i^b$ and $q_i = -q_i^\ast = q_i$. If the spherical anions and cations differ only by the sign of their charges, as in the restricted primitive model, we have $q_i^{\text{eff}}(k) = -q_i^{\text{eff}}(k)$ and $q_i = -q_i^\ast = q_i^*$. In this case $q_i^{\text{eff}}(k) = q_i^{\ast} \tilde{\phi}(\mathbf{k})$ and from eqn (100) it follows that

$$
\frac{k^2}{k_D^2} = \frac{q_i^{\text{eff}}(k) - q_i^{\ast} \tilde{\phi}(\mathbf{k})}{q_i^{\ast} \tilde{\phi}(\mathbf{k})} \quad \text{(RPM).}
$$

For an RPM electrolyte in the linearized PB approximation, the mean electrostatic potential due to an ion of species $i$, the "central" ion located at the origin, is

$$
\psi_i(r) = \frac{q_i \epsilon_0 \epsilon_r}{1 + kd} \frac{e^{-\kappa r}}{4\pi \mu \varepsilon_0 r} \quad \text{for } r \geq d \quad \text{(LPB)},
$$

where $k = k_D$ and prefactor for $\psi_i(r)$ is dictated by local electroneutrality, $4\pi \int_0^\infty drr^2 \rho_i(r) = -q_i$. We can identify the effective
charge as \( q_i^{\text{eff}} = q_i e^{\kappa d_i/(1 + \kappa d_i)} \) whereby we have \( \Psi_i(r) = q_i e^{-\kappa r}/[4\pi\varepsilon_0 r^2] \). In the PB approximation, only the central ion has \( q_i^{\text{eff}} \neq q_i \) while all ions in its ion cloud are assumed to be bare, so for them \( q_i^{\text{eff}} = q_i \) and eqn (101) yields \( \kappa/\kappa_0 = 1 \). However, since all ions should be treated on the same basis, they should all have \( q_i^{\text{eff}} \neq q_i \). If we in eqn (101) set \( q_i^{\text{eff}} \) equal to the value from the LPB case but with \( \kappa \neq \kappa_0 \), we obtain the approximation in eqn (2).

We finally note that each Yukawa function term for non-spherical particles is always accompanied by terms that decay according to eqn (41)

\[ \frac{1}{r^2} \] for \( \kappa \rightarrow 0 \), the term with \( l = 2 \) goes over to a purely dipolar term that decays with distance as \( 1/r^3 \) and the term with \( l = 3 \) goes to a purely quadrupolar term that decays as \( 1/r^4 \). Thereby the usual multipole expansion is obtained, which applies to the electrostatic potential from a fixed charge distribution immersed in a pure polar liquid.

### 4 Screened interactions in electrolytes in the general case

#### 4.1 The main case: all decay modes of the correlations are determined by the dielectric function

As we saw earlier, the electrostatic potential from a particle can be regarded as an external potential for the system even when the particle belongs to the same species as one of the constituent ones. The total mean potential \( \varphi_i(R_i) \) from a \( j \)-particle with coordinates \( R_j \) can then be treated in the same manner as \( \varphi_i(r) \) and when \( \psi_i \) is weak, the pair potential of mean force \( w_{ij}(R_i, R_j) \) for a particle of species \( i \) with coordinates \( R_i \) satisfies according to eqn (41)

\[
w_{ij}(R_i, R_j) \approx \int dr \rho_i^{\text{eff}}(r|R_i) \psi_j(r|R_j) \quad \text{(weak \( \psi_j \) and \( w_{ij} \))}
\]

provided that the screened electrostatic interactions between the two particles dominate in \( w_{ij} \). Here \( w_{ij} \) has taken the role of \( \delta W_i \) and \( \psi_i \) of \( \delta \varphi \) in eqn (41). With this in mind, we define in the general case the screened electrostatic part of the potential of mean force as

\[
w_{ij}^{\text{el}}(R_i, R_j) = \int dr \rho_i^{\text{eff}}(r|R_i) \varphi_j(r|R_j)
\]

\[
= \int dr dr' \rho_i^{\text{eff}}(r|R_i) \varphi_{\text{Coul}}^{\text{eff}}(r-r') \rho_j^{\text{eff}}(r'|R_j)
\]

where we have made use of eqn (60) for the potential \( \psi_j \) due to the \( j \)-particle. Note that the \( i \)-particle and the \( j \)-particle are treated in a symmetric manner and that we can also write

\[
w_{ij}^{\text{el}}(R_1, R_2) = \int dr \rho_i^{\text{eff}}(r|R_1) \varphi_j(r|R_2)
\]

\[
= \int dr dr' \rho_i^{\text{eff}}(r|R_1) \varphi_{\text{Coul}}^{\text{eff}}(r-r') \rho_j^{\text{eff}}(r'|R_2)
\]

(103)

The integral on the rhs of eqn (103) has the same form as a Coulomb interaction energy between the charge densities \( \rho_i^{\text{eff}} \) and \( \rho_j^{\text{eff}} \), but with the screened Coulomb potential \( \varphi_{\text{Coul}}^{\text{eff}}(r) \) instead of the unscreened one. As we will see \( w_{ij}^{\text{el}} \) has a central role for the interparticle interactions in electrolytes. For spherically symmetric particles we have

\[
w_{ij}^{\text{el}}(R_1, R_2) = \int dr dr' \rho_i^{\text{eff}}(r|R_1) \varphi_{\text{Coul}}^{\text{eff}}(r|R_2)\rho_j^{\text{eff}}(r|R_2)
\]

\[
= \int dr \rho_i^{\text{eff}}(r|R_1) \varphi_j(r|R_2)
\]

(104)

Since eqn (103) and (104) constitute definitions of \( w_{ij}^{\text{el}} \), they can be used irrespectively of the magnitude of the potentials.

A very important task is to relate the decay behavior of \( \varphi_{\text{Coul}}^{\text{eff}}(r) \) to that of \( w_{ij} \) and the pair distribution functions \( g_{ij} = 1 + h_{ij} = e^{-\beta w_{ij}} \). We have the expansions

\[
h_{ij} = -\beta w_{ij} + \beta w_{ij}^2/2! - \ldots
\]

\[
\beta w_{ij} = -\ln(1 + h_{ij}) \sim -h_{ij} + h_{ij}^2/2 - \ldots,
\]

so for large separations, where \( h_{ij} \) and \( w_{ij} \) are small, these two functions decay in the same manner, that is, \( h_{ij} \sim -\beta w_{ij} \).

To find the connection between the decays of \( \varphi_{\text{Coul}}^{\text{eff}} \) and \( h_{ij} \), let us introduce another pair correlation function \( h_{ij}^{\text{corr}} \), which for spherical simple ions is defined by

\[
h_{ij}^{\text{corr}}(r) = h_{ij}(r) + \beta \int dr \psi_j(r|R_4) \rho_i^{\text{eff}}(r|R_4).
\]

(106)

Recall that the total charge density \( \rho_i^{\text{tot}} \) associated with a spherical \( i \)-ion is given by

\[
\rho_i^{\text{tot}}(r) = \sigma_i(r) + \rho_i^{\text{eff}}(r) = \sigma_i(r) + \sum_j q_j n_{ij}^3 g_{ij}(r).
\]

\[
= \sigma_i(r) + \sum_j q_j n_{ij}^3 h_{ij}(r).
\]

We will now show that the dressed ion charge density is given by

\[
\rho_i^{\text{dressed}}(r) = \sigma_i(r) + \rho_i^{\text{dressed}}(r) = \sigma_i(r) + \sum_j q_j n_{ij}^3 g_{ij}(r)
\]

\[
= \sigma_i(r) + \sum_j q_j n_{ij}^3 h_{ij}^{\text{corr}}(r).
\]

(107)

which means that the functions \( h_{ij}^{\text{corr}} \) and \( g_{ij} \equiv 1 + h_{ij}^{\text{corr}} \) have the same roles \( \text{vis-à-vis} \) \( \rho_i^{\text{eff}} \) as the usual pair functions \( h_{ij} \) and \( g_{ij} \) have \( \text{vis-à-vis} \) \( \rho_i^{\text{tot}} \). In other words, the function \( g_{ij} \) is the distribution function of the dress. Eqn (107) can easily be
realized from the fact that eqn (106) implies that
\[
\sigma_i(r_{12}) + \sum_j q_i n_j h_{ij}^*(r_{12}) = \sigma_i(r_{12}) + \sum_j q_i n_j h_{ij}(r_{12}) \\
+ \beta \sum_j q_i n_j \int d r_4 \psi_j(r_4) \rho_j^*(r_{24}) \\
= \rho_i^\text{tot}(r_{12}) - \int d r_4 \psi_j(r_4) \chi_j(r_{24}).
\]
where we have used eqn (54) to obtain the last equality. The rhs of this equation is equal to \( \rho_i^\ast(r_{12}) \) according to its definition (40), so eqn (107) follows.

In the general case we define in an analogous manner
\[
h_{ij}^*(\mathbf{R}_1, \mathbf{R}_2) \equiv h_{ij}(\mathbf{R}_1, \mathbf{R}_2) + \beta \int d r_4 \psi_j(r_4) \rho_j^*(r_4) \\
+ \beta \int (d r_3 d r_4) \rho_j^*(r_3) \phi_{\text{Coul}}^\ast(r_4) \rho_j^*(r_4) \\
\]
where we have inserted eqn (60). The charge densities \( \rho_i^\text{tot} \) and \( \rho_i^\ast \) are in this case given by
\[
\rho_i^\text{tot}(\mathbf{r}_2|\mathbf{R}_1) = \sigma_i(\mathbf{r}_2|\mathbf{R}_1) + \sum_j \int d \mathbf{r}_3 n_j h_{ij}(\mathbf{R}_1, \mathbf{R}_3) \sigma_j(\mathbf{r}_2|\mathbf{R}_3) \\
\rho_i^\ast(\mathbf{r}_2|\mathbf{R}_1) = \sigma_i(\mathbf{r}_2|\mathbf{R}_1) \\
+ \sum_j \int d \mathbf{r}_3 n_j h_{ij}^*(\mathbf{R}_1, \mathbf{R}_3) \sigma_j(\mathbf{r}_2|\mathbf{R}_3),
\]
where one can prove the expression for \( \rho_i^\ast \) by using eqn (40) and (46) and analogous arguments as in the derivation of eqn (107). Also in this case, \( h_{ij}^\ast \) gives \( \rho_i^\ast \) in the same way as \( h_{ij} \) gives \( \rho_i^\text{tot} \).

By using the definition (103) of \( w_{ij}^\ast \) we see that eqn (109) can be written as
\[
h_{ij}(\mathbf{R}_1, \mathbf{R}_2) = h_{ij}^\ast(\mathbf{R}_1, \mathbf{R}_2) - \beta w_{ij}^\ast(\mathbf{R}_1, \mathbf{R}_2). \\
\]

The pair correlation function \( h_{ij} \) has accordingly been split into two parts: the function \( h_{ij}^\ast \) of the dresses and the part \( -\beta w_{ij}^\ast \) that contains the screened Coulomb interaction. As we will see, the electrostatic term \( w_{ij}^\ast \) determines the decay behavior of pair correlation function \( h_{ij} \) in terms of Yukawa functions. More precisely, in the overwhelming number of cases

(i) the term \( -\beta w_{ij}^\ast \) in eqn (112) gives rise to all contributions to \( h_{ij} \) (and thereby to \( -\beta w_{ij} \)) that decays exponentially like a monotonic Yukawa function \( e^{-\alpha r} \) or an oscillatory one \( e^{-\alpha r} \cos(\alpha_2 r + \delta) / r \) with \( \alpha_R = \Re(a) \) and \( \alpha_3 = \Im(a) \), the imaginary part

(ii) the decay parameter \( a \) of each such contribution satisfies \( \tilde{a}(\alpha) = 0 \), which means that \( a \) is a solution to the general eqn (78) for \( \kappa \), so \( a \) is a screening parameter. This implies that \( h_{ij} \) has the same set of screening parameters as \( \phi_{\text{Coul}}^\ast(r) \) and that the various are equal to \( \kappa, \kappa^\prime \) etc. in eqn (86).

Thus, for each Yukawa term in eqn (86) there is a corresponding contribution in \( h_{ij} \) with the same screening parameters (but with different prefactor and phase shift, if any). The first term in eqn (112), the function \( h_{ij}^\ast \), also has contributions that decay like Yukawa functions (with other values of the decay parameters), but as shown below they do not give any contribution to \( h_{ij} \) (apart from some rare exceptions to be described later). Thus, the screened Coulomb potential and the pair correlation function normally have the same decay modes, each mode having its own values of the screening parameter, dielectric factor, effective relative dielectric permittivity and effective charges.

As we have seen, a contribution that decays like a monotonic or oscillatory Yukawa function corresponds to a simple pole in complex \( k \)-space. Let us therefore investigate the functions in Fourier space. Since the pair correlation function \( h_{ij}(\mathbf{R}_1, \mathbf{R}_2) \) in the bulk phase depends on the separation vector \( r_{12} = r_2 - r_1 \) we can write \( h_{ij}(\mathbf{R}_1, \mathbf{R}_2) = h_{ij}(r_{12}, \omega_1, \omega_2) \), so in Fourier space we obtain from eqn (103) and (112)
\[
\tilde{h}_{ij}(k, \omega_1, \omega_2) = \tilde{h}_{ij}^\ast(k, \omega_1, \omega_2) \\
= \frac{\beta \tilde{\rho}^\ast_i(k, \omega_1) \tilde{\phi}_{\text{Coul}}^\ast(k) \tilde{\rho}^\ast_j(-k, \omega_2)}{\tilde{c}(k)},
\]
where we have used eqn (64). For spherical particles this equation reduces to
\[
\tilde{h}_{ij}(k) = \tilde{h}_{ij}^\ast(k) - \frac{\beta \tilde{\rho}^\ast_i(k) \tilde{\phi}_{\text{Coul}}^\ast(k) \tilde{\rho}^\ast_j(k)}{\tilde{c}(k)).
\]
These two results, which we have obtained here by physical reasoning, are key equations in DMT and DIT, respectively, that have been derived earlier\(^{21,25,33}\) from the Ornstein–Zernike (OZ) equation. The entire theory of the present work, including the concept of dressed particles, can be formulated in terms of direct correlation functions, but in this work this has been avoided because the theory is then more accessible for a wider readership.

An important result of the current work is that the coupling between fluctuations in charge density and in number density, which normally takes place, makes all poles of \( \tilde{h}_{ij} \) to be given by the zeros of \( \tilde{c}(k) \), whereby they coincide with the poles of \( \tilde{\phi}_{\text{Coul}}^\ast(k) \). This follows from the fact, shown in Appendix C, that \( \tilde{h}_{ij} \) and \( \tilde{h}_{ij}^\ast \) cannot have poles for the same \( k \) values (apart from a few exceptional cases that will be treated in Section 4.2). This result is a consequence of the fact that each pole of \( \tilde{h}_{ij}^\ast \) is cancelled by a corresponding pole in the last term in eqn (113), as shown explicitly in Appendix C. For binary simple electrolytes this cancellation in eqn (114) has previously been found\(^{25}\). Since \( \tilde{\rho}^\ast_i \) and \( \tilde{\rho}^\ast_j \) are given by linear combinations of \( \tilde{h}_{ij}^\ast \), their poles are also poles of \( \tilde{h}_{ij}^\ast \) and cannot coincide with any pole of \( \tilde{h}_{ij} \). Therefore, all poles of the rhs of eqn (113) [for spherical ions eqn (114)] originate from the zeros of \( \tilde{c}(k) \) in the denominator and the assertions in (i) and (ii) above follow. The dielectric function \( \tilde{c}(k) \) is also a linear combination of \( \tilde{h}_{ij}^\ast \) since \( \tilde{\rho}^\ast_i \) in eqn (50) is such a linear combination. The poles and zeros of \( \tilde{c}(k) \) occur, of course, for different \( k \) values.
For spherical particles, these results and eqn (114) imply that [cf. eqn (92)]

\[
h_{ij}(r_{12}) \sim -\frac{\beta}{4\pi e_0} \left[ \frac{q_{ij}^\text{eff}(k)q_{ij}^\text{eff}(k')e^{-\kappa r_{12}}}{\delta_T(k')r_{12}} + \frac{q_{ij}^\text{eff}(k')q_{ij}^\text{eff}(k)e^{-\kappa r_{12}}}{\delta_T(k)r_{12}} \right] + \frac{q_{ij}^\text{eff}(k')q_{ij}^\text{eff}(k)e^{-\kappa r_{12}}}{\delta_T(k')r_{12}} + \text{other terms}
\]

in the limit \( r_{12} \to \infty \). The "other terms" in this equation are additional Yukawa function terms with different \( \kappa \) values (shorter decay lengths) that are solutions to eqn (78) and functions with different functional dependences of \( r_{12} \) that normally decay faster than the leading term. The latter kind of functions always arises for reasons explained at the end of the current section.

For nonspherical particles, an expression for \( h_{ij}(r_{12},o_1,o_2) \) analogous to that in eqn (115) applies when \( r_{12} \to \infty \)

\[
h_{ij}(r_{12},o_1,o_2) \sim -\frac{\beta}{4\pi e_0} \left[ \frac{Q_{ij}^\text{eff}(r_{12},o_1,k)Q_{ij}^\text{eff}(-r_{12},o_2,k)}{\delta_T(k)r_{12}} \right] + \text{other terms},
\]

where we have only shown the first term for \( \kappa_s = \kappa, \kappa' \) and \( \kappa'' \) [cf. eqn (93)]. Note that the vectors \( r_{12} \) in \( Q_{ij}^\text{eff} \) and \(-r_{12}\) in \( Q_{ij}^\text{eff} \) point towards the center of the other particle along the connecting line. As noted in Section 3.2.4, for these kinds of systems, each Yukawa function term is always accompanied by terms that decay as \( e^{-\kappa r} \) with \( l = 2, 3, \ldots \). They are included in "other terms" together with additional terms that will be discussed at the end of this section.

When \( \kappa \) is complex, the \( \kappa \) and \( \kappa' \) terms combine and form an oscillatory term [cf. eqn (87)]. By writing \( Q_{ij}^\text{eff}(r_{12},o,k) = [Q_{ij}^\text{eff}(r_{12},o,k)]e^{i\gamma l(r_{12})} \) with a real-valued \( \gamma_l \) for \( l = i, j \), we then have for \( r_{12} \to \infty \)

\[
h_{ij}(r_{12},o_1,o_2) \sim -\frac{\beta}{2\pi e_0} \left[ Q_{ij}^\text{eff}(r_{12}) \left| e^{-\kappa r_{12}} \right| \delta_T(k)r_{12} \right] \times \cos \left[ \kappa_s r_{12} - \gamma_l + \gamma_l \right] + \text{other terms},
\]

where \( \gamma_l = \gamma_l(r_{12},o_1,k) \), \( \gamma_l = \gamma_l(-r_{12},o_2,k) \) and where we likewise have suppressed the arguments of \( Q_{ij}^\text{eff} \) and \( Q_{ij}^\text{eff} \). Both \( \gamma_l \) and \( [Q_{ij}^\text{eff}] \) depend on orientation. This kind of oscillatory term is, of course, formed from any pair of complex conjugate screening parameters \( \kappa_s \) that are solutions to eqn (78), say, \( \kappa_s \) and \( \kappa_{s+1} = \kappa_s \).

As noted in Section 3.2.4, the theory is valid for particles of any size and shape. Therefore results in eqn (116) and (117) are applicable for the correlations between, for example, a macro-particle and an ion. Thereby the potential of mean force acting on the ion as a function of distance from the macro-particle decays as \( w_{ij} \sim -\beta h_{ij} \). Likewise, the results can be applied for the interaction between two macro-particles, for example, the surface forces between two macroscopic surfaces, which hence have the decay modes that are determined by the bulk electrolyte that the fluid phase between the surfaces is in equilibrium with.

The density–density, charge–charge, density–charge correlation functions, \( H_{\text{S}}(r), H_{\text{QQ}}(r) \) and \( H_{\text{SQ}}(r) \) defined in Appendix B, have the same poles in Fourier space as \( h_{ij} \), which are in general determined by the zeros of \( \tilde{z}(k) \). This can be realized as follows. \( \tilde{H}_{\text{NN}}(k) \), given in Appendix B, eqn (138), is a linear combination of \( h_{ij} \)

\[
\tilde{H}_{\text{QQ}}(k) = n_b^b \sum_\gamma n_b^b \left\{ d_{\gamma} \tilde{h}_{ij}(k_\gamma, \omega_\gamma, \omega_{\gamma 2}) \right\}
\]

so it cannot have any other poles than those of \( h_{ij} \). By inserting \( \text{eqn (66) into eqn (116) and } \tilde{H}_{\text{QQ}}(k) \) given in eqn (136) and \( \tilde{H}_{\text{SQ}}(k) \) given in eqn (140) we obtain

\[
\tilde{H}_{\text{QQ}}(k) = \frac{q_{e}^{-2} \sum_\gamma n_b^b \tilde{\sigma}(k_{\gamma}, \omega_{\gamma}, \omega_{\gamma 2}) \tilde{h}_{ij}(k_{\gamma}, \omega_{\gamma}, \omega_{\gamma 2})}{\tilde{z}(k)}
\]

and we see that the zeros of \( \tilde{z}(k) \) are poles of these functions. Thus \( H_{\text{S}}(r), H_{\text{QQ}}(r), H_{\text{SQ}}(r), h_{ij}, w_{ij} \) and \( \phi_{\text{Coul}}(r) \) have the same screening parameters, apart from in the exceptional cases mentioned earlier. They will be treated in the next section.

As mentioned earlier there always exist terms in eqn (115) that have a different \( r \) dependence than Yukawa functions. They appear because when \( h_{ij}(r) \) and therefore \( w_{ij}(r) \) contain a term that decays as \( e^{-\kappa r} \) with real \( \kappa \), it follows from eqn (105) that \( h_{ij}(r) \) contains terms that decay as \( e^{-\kappa r} \), \( e^{-\kappa r} \), etc. and that this also applies to \( w_{ij}(r) \). These higher order terms that appear because the system is intrinsically nonlinear have decay lengths \( (2\kappa)^{1}, (3\kappa)^{1} \) etc. so they decay faster than the "original" Yukawa function term, with the same \( \kappa \), that has generated them. Therefore they give important contributions mainly for small \( r \). For large \( r \) some of them can, however, dominate over Yukawa function terms with larger \( \kappa \) values in eqn (115), for example the term with \( e^{-\kappa r} \) provided \( \kappa > 2k \). There also exist terms that decay as \( \zeta(r) e^{-\kappa r} \), where \( \zeta(r) \) is a slowly varying function and \( l \geq 2 \). Furthermore, there are cross-terms from two or more Yukawa functions with different decay lengths. All these higher order terms are included in "other terms" in eqn (115) and it is not worthwhile to consider more than the leading ones individually and explicitly. For complex-valued \( \kappa \), similar conclusions are valid for exponentially decaying oscillatory terms and the same applies in eqn (116) for nonspherical particles.

The higher order terms give singularities in complex Fourier space that are different from simple poles, for instance \( e^{-2\kappa r} \) gives a logarithmic branch point at \( k = 2\kappa \).
All of these contributions are generated by the set of fundamental decay modes with screening parameters $\kappa_i$, that are solutions to the general eqn (78) for $\kappa$ and, equivalently, solutions to $\tilde{e}(\kappa_i) = 0$ and to eqn (97). Thus, the decay behaviors of both the screened electrostatic potential and the correlation functions originate from fundamental decay modes.

In cases where the non-electrostatic pair interactions $u^{ne}_{ij}(r)$ do not have a short range (contrary to our assumptions earlier), but instead have a power law decay like the $r^{-6}$ dispersion interactions, the functions $h_{ij}(r)$, $w_{ij}(r)$ and $\phi_{\text{Coul}}(r)$ ultimately decay like a power law when $r \to \infty$. These functions still have Yukawa function terms*** that are given by the present formalism and the power law terms are included in “other terms” in the various equations. The Yukawa terms can give dominant contributions for short to intermediate $r$ values, but they can never dominate for very large $r$ because they decay faster than any power law. In many cases the Yukawa function terms have a dominant influence in $h_{ij}(r)$ for most $r$. This is, for example, seen in the simulation results by Keblinski et al.24 mentioned in the Introduction for the realistic model for NaCl that includes dispersion interactions. Their calculations show that the monotonic and oscillatory Yukawa function terms give the dominant contributions.

4.2 Exceptions; charge-inversion invariant systems

Let us now turn to the exceptional cases, that is, cases where $\tilde{h}_{ij}$ and $\tilde{h}_{ij}^*$ can have poles for the same $\kappa$ values and where Yukawa function contributions to $h_{ij}$ therefore originate both from the last term in eqn (113) and from $\tilde{h}_{ij}^*$. As shown in Appendix D of ref. 29 this can occur for most systems at exceptional points in the system’s parameter space (for instance a critical point).††† This will not be dealt with any further because the systems behave in the way we have just described for all other parameter values.

The other exceptional case is a category of systems where $\tilde{h}_{ij}^*$, in contrast to the main case, always gives Yukawa function contributions to $h_{ij}$, namely model systems that are invariant when we invert the sign of all charges of the particles. Such systems, which we will call charge-inversion invariant systems, remain exactly the same when one does such a charge inversion, whereby the internal charge distribution $\sigma_i(r|R_i)$ for each particle changes to $-\sigma_i(r|R_i)$ (positive regions become negative and vice versa) without change in the absolute value of $\sigma_i$ for each $r$. This means, for example, that the anions become cations and vice versa during charge inversion. For an invariant system, for each cation species there must exist an anion species that is identical in all respects apart from the sign of $\sigma_i(r|R_i)$, like the same size, same shape and same non-electrostatic interactions with other particles, and have the same number density. Examples of such systems include the restricted primitive model, where the anions and cations of the same absolute valency are charged hard spheres of equal size. As soon as there is any difference, however small, between anions and cations apart from the sign of their charges, the main result applies, so all Yukawa function terms in $h_{ij}$ originate from the last term in eqn (113) and all poles of $\tilde{h}_{ij}$ arises from the zeros of $\tilde{z}(k)$. For charge-inversion invariant electrolyte systems, all electroneutral species present must also satisfy invariance conditions. Examples include electrolyte models with explicit solvent where the solvent molecules turn into themselves during a charge inversion, for instance spherical particles with a dipole at the center.

The reason why charge-inversion invariant systems are exceptions is that the extreme symmetry forces the density–charge correlation function $H_{\text{QQ}}(r)$ to be identically zero, so fluctuations in charge density and in number density are uncoupled from each other. It is simple to realize that $H_{\text{QQ}}(r)$ must be identically equal to zero in such systems. Suppose that $H_{\text{QQ}}(r)$ is, say, positive for a certain $r$ value and we invert all charges in the system, $H_{\text{QQ}}(r)$ would become negative but since the system is charge-inversion invariant $H_{\text{QQ}}(r)$ must remain the same, which implies that $H_{\text{QQ}}(r)$ must be zero. Alternatively, this can be realized when we consider correlations between fluctuations in density and fluctuations in charge at two points separated by distance $r$, because for a given fluctuation in density, the probability for positive and negative fluctuations in charge must be equal, so the fluctuations will average to zero. Mathematically, the fact that $H_{\text{QQ}}(r) \equiv 0$ can be realized from the rhs of the definition of $H_{\text{QQ}}(r)$ in Appendix B [eqn (139)]. For each positive contribution on the rhs there must exist an equally large negative contribution due to the charge-inversion invariance. This can likewise be realized from eqn (140). Exactly the same argument applies to eqn (120) because during a charge inversion $\tilde{\rho}_i^*(k, \omega_1)$ for each particle changes to $-\tilde{\rho}_i^*(k, \omega_1)$, so we must have

$$\sum_j h_{ij}^* \delta(\omega_1 - \omega_2) = 0 \quad (121)$$

and hence $\tilde{R}_{\text{QQ}}(k) \equiv 0$.

Let us now consider the density–density correlation function for charge-inversion invariant systems. By inserting $\tilde{h}_{ij}$ from eqn (113) into eqn (118) we see that the contribution from the last terms in eqn (113) cancels identically in $\tilde{R}_{\text{NN}}(k)$ because of eqn (121). Therefore we have

$$\tilde{R}_{\text{NN}}(k) = h_{\text{NN}}^* + \sum_q h_{ij}^* \delta_{\omega_2} \left( \tilde{h}_{ij}^* \langle k, \omega_1, \omega_2 \rangle \right)_{\omega_1, \omega_2} \quad (122)$$

so the poles of $\tilde{R}_{\text{NN}}(k)$ must be poles of $\tilde{h}_{ij}^*$. Eqn (119) for $\tilde{R}_{\text{QQ}}(k)$ remains, however, valid and the poles of $\tilde{R}_{\text{QQ}}(k)$ are hence given by the zeros of $\tilde{z}(k)$ also in the present case. Thus there are two different sets of poles that are relevant, the poles of $\tilde{R}_{\text{QQ}}(k)$ [zeros of $\tilde{z}(k)$] and those of $\tilde{R}_{\text{NN}}(k)$. The poles of the pair correlation function $\tilde{h}_{ij}$ belong to both sets, but as we will see they enter in $\tilde{h}_{ij}$ in different manners.

Consider two species of ions that swap their identities as anions and cations during a charge inversion. Let us call them $A^+$ and $A^-$, so we have $\sigma_{A^+}(r|R_i) = -\sigma_{A^-}(r|R_i)$. The common notation A indicates that they are identical apart from the sign

*** At least for low ionic densities, the leading term Yukawa function term in the presence of dispersion interactions is oscillatory with a wavelength that is much larger than the decay length,35 so in practice it appears like monotonic Yukawa functions.

††† A full treatment of the theory that uses and generalizes the approach in Appendix D of ref. 29 will be published in a separate paper.
of their internal charge distributions. We likewise consider two other species of ions $B^-$ and $B^{+}$ with $\sigma_{g}^s(r|R_1) = -\sigma_{g}^s(r|R_2)$. They are also identical to each other apart from the sign of $\sigma$. The charge-inversion invariance implies that we have the symmetry
\[
\begin{align*}
    h_{A,-A} &= h_{A,-A} \\
    h_{B,-B} &= h_{B,-B} \\
    h_{A,+A} &= h_{A,+A} \\
    h_{A,-B} &= h_{A,-B} = h_{B,+A} = h_{B,+B}.
\end{align*}
\]
(we also have $h_{A,-A} = h_{A,-A}$ and $h_{B,-B} = h_{B,-B}$, as always). The first two lines are special cases of the last two (with $A = B$), so in the following we will only deal with the latter. The same symmetry relationships are valid for $h_{ij}^*$. In the presence of more than four species of ions the corresponding relationships are valid for $A$, $B$, $C$, $D$, etc.

We now define $h_{S,AB}(r,o_1,o_2)$, $h_{D,AB}(r,o_1,o_2)$ and the corresponding $h^*$ functions from
\[
\begin{align*}
    h_{S,AB} &= h_{A,+B} + h_{A,-B} = h_{A,+B} + h_{A,-B} \\
    h_{D,AB} &= h_{A,+B} - h_{A,-B} = h_{A,+B} - h_{A,-B},
\end{align*}
\]
($S$ stands for sum and $D$ for difference) so we have
\[
\begin{align*}
    h_{A,+B} = h_{A,+B} &= h_{A,-B} = h_{B,-A} = 1/2 [h_{S,AB} + h_{D,AB}] \\
    h_{A,+B} = h_{A,+B} &= h_{A,-B} = 1/2 [h_{S,AB} - h_{D,AB}],
\end{align*}
\]
and likewise for $h_{ij}^*$. Note that $+ sign on the rhs of eqn (123) applies to ions with equal sign of their charges (like $A^+$) and $-$ sign applies to ions with different signs (like $A^- B^+$). One can also define $S$ and $D$ functions for the electroneutral particles (like solvent molecules) in charge-invariant systems, but we will not enter into any details here.\footnote{The details will be published elsewhere.}

Due to the charge-inversion invariance we have $\tilde{\rho}_{A,+}^* = -\tilde{\rho}_{A,-}^* \equiv \tilde{\rho}_{A}^*$, which defines $\tilde{\rho}_{A}^*$, and likewise for $\tilde{\rho}_{B}^*$. Therefore we have from eqn (113)
\[
\begin{align*}
    \tilde{h}_{S,AB}(k,o_1,o_2) &= \tilde{h}_{S,AB}^*(k,o_1,o_2) \\
    \tilde{h}_{D,AB}(k,o_1,o_2) &= \tilde{h}_{D,AB}^*(k,o_1,o_2)
\end{align*}
\]
\[
\begin{align*}
    &= 2\beta \tilde{\rho}_{A}^*(k,o_1,o_2) \frac{\phi_{\text{Coul}}(k) \tilde{\rho}_{B}^*(-k,o_2)}{\epsilon(k)}
\end{align*}
\]
\[
\begin{align*}
\int (\sigma_{g}^s(r|R_1) - \sigma_{g}^s(r|R_2)) \, d^3r
\end{align*}
\]
Analogously to the arguments about the poles of $\tilde{h}_y^*$ and $\tilde{h}_i$, it follows that $\tilde{h}_{D,AB}$ cannot have poles common to $\tilde{h}_{S,AB}$ so the poles of $\tilde{h}_{D,AB}$ are due to the zeros of $\bar{\epsilon}(k)$ [possibly apart from exceptional points in the system’s parameter space]. The functions $\tilde{h}_{S,AB}$ and $\tilde{h}_{D,AB}$ have different decay parameters; the poles of $\tilde{h}_{S,AB}$ belong to one of the sets mentioned earlier [the poles of $\tilde{H}_{NN}(k)$] and those of $\tilde{h}_{D,AB}$ belong to the other set [the poles of $\tilde{H}_{QQ}(k)$]. Thus only $\tilde{h}_{D,AB}$ has the same screening parameters as $\phi_{\text{Coul}}^*(r)$ and $\tilde{h}_{D,AB}/2$ decays analogously to the rhs of eqn (115)-(117). The contribution from $\tilde{h}_{D,AB}$ in eqn (123) appears with a plus sign for ions of the same sign of charge and with a minus sign for ions of opposite sign. The function $\tilde{h}_{S,AB}$ (with different decay parameters than $\tilde{h}_{D,AB}$) appears with a plus sign in eqn (123) for all ions irrespectively of their signs of charge.

In fact, $H_{NN}(r)$ is a linear combination of the $S$-functions and $H_{QQ}(r)$ is a linear combination of the $D$-functions (including those for any electroneutral particles, if present). Furthermore, the total particle density $\sum g_i^{s}\rho_i^0$ around each of the particles in charge-inversion invariant systems is determined by $S$-functions and the total charge density $\rho_i^{0\text{ex}}$ is determined by $D$-functions, so these two entities have different decay parameters.

All these facts are well-known for the restricted primitive model where there are only two species $A^+$ and $A^-$ in a dielectric continuum solvent, but the new findings here are that this applies in general for charge-inversion invariant systems with any number of components including solvent molecules and other electroneutral particles. All particles can have any shape, size, internal charge density and nonelectrostatic interactions subject to the conditions of charge-inversion invariance.

In the RPM, where we have $h_{i \pm}(r) \equiv h_{A,+A}^*(r) = 1/2 [h_{S,AA}(r) \pm h_{D,AA}(r)]$, it can happen that $h_{S,AA}(r)$ has a longer decay length than $h_{D,AA}(r)$ when the ion density is high.\footnote{The details will be published elsewhere.}

Then, the tail of $h_{ij}(r)$ for large $r$ has the same sign irrespective of the signs of the $i$ and $j$-ions. This has been denoted as “core dominance,”\footnote{The details will be published elsewhere.} because $h_{S,AA}(r)$ is decoupled from electrostatics and is mainly determined by the hard core packing of the ions. For low ion densities, where $h_{D,AA}(r)$ has the longest decay length, the tails of $h_{+,i}(r)$ and $h_{-,i}(r)$ have different signs and there is “electrostatic dominance.” Such strict distinction between core dominance and electrostatic dominance exists only in charge-inversion invariant systems.

It is interesting to consider systems that are close to being charge-inversion invariant. This is the case, for example, in the primitive model of binary symmetric electrolyte solutions when anions and cations have equal absolute valency but differ slightly in size. Then, there is no longer a decoupling between electrostatic and nonelectrostatic correlations, so there is electrostatic coupling in the decay modes where hard core packing of the ions dominates and vice versa. Systems close to charge-inversion invariance also occur in more realistic models of symmetric electrolytes where the anions and cations have nearly the same nonelectrostatic pair interactions with their own species $u_{\text{nn}}(r) \approx u_{\text{nn}}^*(r)$, as in the model for NaCl mentioned in the Introduction. Another example is a model with explicit solvent where anions and cations are identical apart from their signs but the solvent molecules are modeled as spheres with a radially aligned dipole that lies somewhat off-center.

For these systems all poles of $\tilde{h}_y$ are given by the zeros of $\bar{\epsilon}(k)$, while for charge-inversion invariant systems the decay parameter values of $\tilde{h}_y$ belong, as we have seen, to two distinct sets: those that are due to poles of $\tilde{h}_y^*$ and those that are due to zeros of $\bar{\epsilon}(k)$. Since the two kinds of systems differ very little from each
other, their decay parameters must be closely related and vary continuously when one changes a system from being nearly charge-inversion invariant to being invariant or vice versa. This must apply not only for the decay parameters that are given by zeros of \( \bar{\sigma}(k) \) all the time, but also for those that initially are due to such zeros and then, for the invariant system, belong to the set that are due to poles of \( \bar{h}_j \) but not zeros of \( \bar{\sigma}(k) \). In Appendix C it is shown why and how this happens. One can visualize the findings by following the “trajectories” of the poles of \( \bar{h}_j \) and \( \bar{h}_j^* \) as functions of the system parameters like ion sizes etc. For a system that is nearly charge-inversion invariant, it is found in the appendix that there exist one set of poles of \( \bar{h}_j \) [zeros of \( \bar{\sigma}(k) \)] where each pole is close to a pole of \( \bar{h}_j^* \) and where the trajectories of the poles of \( \bar{h}_j \) and \( \bar{h}_j^* \) cross each other at the point where the system become invariant. At the crossing point the pole of \( \bar{h}_j \) ceases to be a zero of \( \bar{\sigma}(k) \) and becomes instead a pole of both \( \bar{h}_j^* \) and \( \bar{h}_j^* \). The second set of poles of \( \bar{h}_j \) do not have such crossings and these poles remain zeros of \( \bar{\sigma}(k) \) throughout. These two sets correspond to the sets for the invariant system introduced above.

More precisely, it is shown in the appendix that there exist zeros of \( \bar{\sigma}(k) \) for the nearly invariant system that lie close to the poles of \( \bar{h}_j^* \) for the same system. The latter are not poles of \( \bar{h}_j \) while the zeros of \( \bar{\sigma}(k) \) are such poles. Each of these zeros moves continuously with the system parameters so that when the system is turned into a charge-inversion invariant one, it merges with the corresponding pole of \( \bar{h}_j^* \) and ceases to be a zero of \( \bar{\sigma}(k) \) but remains as a pole of \( \bar{h}_j^* \).

4.3 The decay of \( H_{\text{QQ}}(r), H_{\text{SN}}(r) \) and \( H_{\text{NQ}}(r) \)

Let us now return to the general case of systems without charge-inversion symmetry, which is the normal, realistic case since anions and cations virtually always differ by much more than the sign of their charge and since the positive and negative parts of the solvent molecules normally differ a lot apart from the sign of the charge.

Let us consider cases where the leading term in \( \bar{h}_j \) is given by the monotonic Yukawa term shown explicitly in eqn (116). This term originates from the last term in eqn (113) evaluated at the leading zero of \( \bar{\sigma}(k) \), i.e., \( k = i \kappa \) and we assume that \( \kappa \) is real. We will investigate the leading term when \( r \to \infty \) for \( H_{\text{SN}}(r), H_{\text{QQ}}(r) \) and \( H_{\text{NQ}}(r) \), respectively. The decay length is \( 1/\kappa \) for all three functions and we will find that the magnitudes of the leading term of these functions are interdependent via common prefactors, which can be obtained from the effective multipole charges \( Q_{\text{eff}}^i \). We will also show that the intimate relationship between the screening parameter \( \kappa \) and \( Q_{\text{eff}}^i \) found in Section 3.2.4 has direct consequences for the relative importance of the different correlation functions.

These results can be obtained from eqn (118)-(120) together with eqn (116) and we obtain for \( r \to \infty \)

\[
H_{\text{NN}}(r) \sim -\beta \left[ \sum_i n_i^b (Q_{\text{eff}}^i (\bar{\mathbf{r}}, \omega, \kappa))_{\text{eff}} \right]^2 \frac{e^{-\kappa r}}{4\pi\varepsilon_0 \varepsilon_{\text{eff}}(\kappa) r^2} \]

where the coefficient is independent of \( \bar{\mathbf{r}} \) due to the average over the orientations,

\[
H_{\text{QQ}}(r) \sim -q_e^2 \sum_i n_i^b (Q_i \bar{Q}_i^\text{eff} (\bar{\mathbf{k}}, \omega, \kappa))_{\text{eff}} \frac{k^2 e^{-\kappa r}}{4\pi\varepsilon_0 \varepsilon_{\text{eff}}(\kappa) r^2}
\]

\[
H_{\text{NQ}}(r) \sim -q_e^{-1} \sum_i n_i^b (Q_i \bar{Q}_i^\text{eff} (\bar{\mathbf{k}}, \omega, \kappa))_{\text{eff}} \frac{k^2 e^{-\kappa r}}{4\pi\varepsilon_0 \varepsilon_{\text{eff}}(\kappa) r^2}
\]

where we can select \( \bar{\mathbf{k}} = \bar{\mathbf{r}} [\text{eqn (94) and (96)}] \) and where \( q_i \) is defined in eqn (98). The Yukawa function factor arises from \( 1/\bar{\sigma}(k) = k^2 e^{\phi_{\text{eff}}^\text{ion}(\kappa)} \) evaluated in \( r \) space. By defining the average entities

\[
\bar{Q}_{\text{eff}}^n(k) = \sum_i x_i^b (Q_{\text{eff}}^i (\bar{\mathbf{r}}, \omega, \kappa))_{\text{eff}} = \sum_i x_i^b \bar{Q}_{\text{eff}}^i (k)
\]

\[
\bar{Q}_{\text{eff}}^i(k) = q_e^{-1} \sum_i n_i^b (Q_i \bar{Q}_i^\text{eff} (\bar{\mathbf{k}}, \omega, \kappa))_{\text{eff}}
\]

\[
= k^2 \left[ \frac{\beta n_i^b q_i^b}{\varepsilon_0} \right]^{-1}
\]

where \( x_i^b = n_i^b/n_{\text{tot}}^b \) is the mole fraction of species \( i \) and where we have used eqn (97) to obtain the last equality, we can write these equations as

\[
\frac{H_{\text{NN}}(r)}{(n_{\text{tot}}^b)^2} \sim -\beta \left[ \bar{Q}_{\text{eff}}^n(k) \right]^2 \frac{e^{-\kappa r}}{4\pi\varepsilon_0 \varepsilon_{\text{eff}}(\kappa) r^2}
\]

\[
\frac{H_{\text{QQ}}(r)}{(n_{\text{tot}}^b)^2} \sim -\beta \left[ \bar{Q}_{\text{eff}}^n(k) \right]^2 \frac{e^{-\kappa r}}{4\pi\varepsilon_0 \varepsilon_{\text{eff}}(\kappa) r^2}
\]

\[
\frac{H_{\text{NQ}}(r)}{(n_{\text{tot}}^b)^2} \sim -\beta \bar{Q}_{\text{eff}}^n(k) \bar{Q}_{\text{eff}}^n(k) \cdot \frac{e^{-\kappa r}}{4\pi\varepsilon_0 \varepsilon_{\text{eff}}(\kappa) r^2}
\]

when \( r \to \infty \). While \( \bar{Q}_{\text{eff}}^n(k) \) is simply the average over all particles of the orientation independent part of \( Q_{\text{eff}}^i \) [eqn (95)], \( \bar{Q}_{\text{eff}}^n(k) \) contains a weighing with respect to \( Q_i \) that depends on the bare charge density \( \sigma_i \) of the particles [see eqn (98)].

Since \( \sigma_i^2 = \beta n_i^b q_i^b \bar{Q}_{\text{eff}}^i(k)_{\text{eff}} \), the decay length \( 1/\kappa \) is directly linked to the magnitude of \( H_{\text{QQ}}(r) \), while the magnitude of \( H_{\text{NN}}(r) \) does not have such a direct link. For a situation with long-range density-density correlations, the screening parameter \( \kappa \) is small and the charge-charge correlations must be small and become even smaller when the decay length increases. This is, for example, relevant when a critical point is approached, whereby the charge-charge correlations become less and less significant although they have the same decay length as the density-density correlations. In the present analysis we do, however, avoid the immediate neighborhood of critical points, where other considerations have to be made. We may, however, note that in the limit of \( 1/\kappa \to \infty \) the charge-charge correlations vanish as they must do.
For charge-inversion invariant systems $\tilde{Q}_{\text{eff}}^\text{NN} \equiv 0$ and $H_{\text{QQ}}(r) \equiv 0$, while the decay of $H_{\text{NN}}(r)$ can be obtained from

$$\frac{H_{\text{NN}}(r)}{n_{\text{bot}}^{b}} = \frac{\delta^{(3)}(r)}{n_{\text{bot}}^{b}} + \sum_{j} n_{j}^{b} \sum_{m} \left\langle \tilde{g}_{ij}(r, \omega_{1}, \omega_{2}) \right\rangle_{\omega_{1}, \omega_{2}}$$

as follows from eqn (122). $H_{\text{QQ}}(r)$, on the other hand, is given by the same expression as in the main case, eqn (128). In this case the behaviors of $H_{\text{NN}}(r)$ and $H_{\text{QQ}}(r)$ are independent of each other and the decay length of $H_{\text{QQ}}(r)$ remains finite even in situations where the decay length of $H_{\text{NN}}(r)$ becomes very large. This constitutes a considerable difference compared to the general case.

Returning to the general case, the conclusions above expressed in eqn (127)–(129) are valid for the leading term in $H_{\text{NN}}(r)$, $H_{\text{QQ}}(r)$ and $H_{\text{QQ}}(r)$ when $\kappa$ is real. Analogous conclusions are valid for the Yukawa terms with other $\kappa_0$ values in these correlation functions, so similar relationships to these equations are valid for all decay modes (including the oscillatory case where there is also a cosine factor). The relative magnitudes of $|\tilde{Q}_{\text{NN}}^\text{eff}|$ and $|\tilde{Q}_{\text{QQ}}^\text{eff}|$ vary for the different modes, so the contributions with some decay lengths may have $|\tilde{Q}_{\text{NN}}^\text{eff}| > |\tilde{Q}_{\text{QQ}}^\text{eff}|$ while the reverse can be true for those with other decay lengths. In this regard it is particularly illustrative to consider systems that are close to being charge-inversion invariant, but let us start with the corresponding invariant systems.

For charge-inversion invariant systems we have seen that there are two distinct sets of poles for $\tilde{h}_{ij}$: the poles of $H_{\text{NN}}(k)$ [poles of $\tilde{h}_{ij}^*$] and those of $H_{\text{QQ}}(k)$ [zeros of $\tilde{u}(k)$]. This implies that the Yukawa function terms in $H_{\text{NN}}(r)$ and $H_{\text{QQ}}(r)$ have distinct decay parameter values; a Yukawa term that is present in $H_{\text{NN}}(r)$ is absent in $H_{\text{QQ}}(r)$ and vice versa. A system that is close to being charge-inversion invariant has, however, all such terms present in both $H_{\text{NN}}(r)$ and $H_{\text{QQ}}(r)$. These two functions then have the same decay parameters and, as we saw at the end of Section 4.2, when one breaks the charge-inversion invariance all decay parameters vary continuously with the system parameters. We also saw that the two sets of poles for $\tilde{h}_{ij}$ remain in the sense that each pole in one set is close to a pole of $\tilde{h}_{ij}^*$ (and merge with it for the invariant system) and the other set consists of poles that do not behave in this manner and are poles of $\tilde{u}(k)$ throughout. For continuity reasons, the magnitudes of the Yukawa terms in $H_{\text{QQ}}(r)$ with decay parameters in the first set must be close to zero (they are exactly zero for the invariant system), while the magnitudes of the terms in $H_{\text{NN}}(r)$ belonging to the second set must be close to zero for the same reason. Thus $|\tilde{Q}_{\text{NN}}^\text{eff}| \gg |\tilde{Q}_{\text{QQ}}^\text{eff}|$ for the first set and $|\tilde{Q}_{\text{NN}}^\text{eff}| \ll |\tilde{Q}_{\text{QQ}}^\text{eff}|$ for the second. This is the situation for the NaCl system mentioned in the Introduction.

Finally, we will turn to dilute electrolyte solutions with molecular solvent. The leading term in $\tilde{h}_{ij}$ for large distances then has a decay parameter $\kappa$ that approaches the Debye parameter $\kappa_D$ at high dilution, as we saw in eqn (85). In a dilute solution where $\kappa$ is small we have $\tilde{Q}_{\text{eff}}^\text{NN}(k) \approx \frac{q_{e}}{4\pi\varepsilon_{0}k_{B}T}$ which implies that $n_{\text{bot}}^{b}\tilde{Q}_{\text{eff}}^\text{NN}(k)$ is very small because $\sum_{i}n_{i}^{b}\tilde{Q}_{ij}(k) \approx \sum_{i}n_{i}^{b}q_{i} = 0$ due to electroneutrality ($q_{\text{solvent}} = 0$ since the solvent molecules are uncharged). Thus, the leading term in $H_{\text{NN}}(r)$, which is proportional to $\tilde{Q}_{\text{eff}}^\text{NN}(k)$, is very small. $H_{\text{QQ}}(r)$ is much larger because its coefficient is proportional to the square of

$$n_{\text{bot}}^{b}\tilde{Q}_{\text{eff}}^\text{NN}(k) \approx q_{e}^{-1}\sum_{i, \text{ions}}n_{i}^{b}\left\langle Q_{i}(-\mathbf{k}, \omega, \mathbf{k})\right\rangle_{\text{ions}} \approx \sum_{i, \text{ions}}n_{i}^{b}q_{i}^{2}/q_{e}$$

where all terms in the sum are positive.

As in other systems, there exists other decay modes with terms of different decay lengths in the correlation functions, for example contributions where the density–density fluctuations of the solvent are prominent. In contrast to the very small terms in $H_{\text{NN}}(r)$ with decay length $1/\kappa$ that we investigated above, these contributions to $H_{\text{NN}}(r)$ are accordingly substantial. Such solvent-dominated contributions are given by Yukawa functions with screening parameters $\kappa_0$ that satisfy eqn (78) and each mode yields terms in $H_{\text{NN}}(r)$, $H_{\text{QQ}}(r)$ and $H_{\text{QQ}}(r)$ with the same decay length. As we saw from the examples in the Introduction, for dilute electrolyte solutions at least one pair of these screening parameters is complex-valued and gives rise to an oscillatory component of the correlation functions that reflects the structure of the molecular solvent.

In the limit of zero electrolyte concentration, the screened Coulomb potential decays as $\phi_{\text{Coul}}^*(r) \sim 1/(4\pi\varepsilon_{0}r)$ which originates from the leading term $e^{-1/4\pi\varepsilon_{0}q_{e}^{2}r^{2}}$ when $\kappa \rightarrow 0$. However, the solvent-dominated decay modes make $\phi_{\text{Coul}}^*(r)$ oscillatory for small to intermediate $r$ values as discussed in more detail for aqueous systems in ref. 29. These oscillations dominate in $\phi_{\text{Coul}}^*(r)$ for pure water up to quite large $r$ values where $1/r$ tail eventually takes over.

At higher concentrations of electrolyte, one cannot clearly distinguish between contributions to the correlation functions that are solvent-dominated and those that are electrolytedominated because there is an intricate coupling between the different constituents of the solution. In all cases there are several decay modes with different screening parameters that are simultaneously present.

5 Summary and concluding remarks

In this section we will give an overview of the key results and also give some perspectives on their usage in experimental and theoretical investigations.

Perhaps the most important finding in this work is the fact that all decay modes in electrolytes, with the exception of charge-inversion invariant electrolyte systems, are governed by the dielectric response and that the decay lengths hence are determined by the parameters $\kappa_0$ that are solutions to eqn (5), which are the screening parameters for the electrostatic potential. Accordingly, all decay modes of the pair potential of mean force $w_{ij}$ for the particles are the same as for the screened Coulomb potential $\phi_{\text{Coul}}^*(r)$; the latter modes are shown explicitly in eqn (86). This applies for both the monotonic and the
oscillatory exponential modes, including those that are dominated by “packing” of molecules in dense liquids. It also applies to the long-range monotonic modes at high ionic densities and the long-range density-density correlations on approach of criticality (but not at the critical point itself). All these facts are a consequence of the coupling between fluctuations in charge density and fluctuations in number density.

Another very significant result is that each decay mode has its own value of the effective relative dielectric permittivity \( \varepsilon_{\text{eff}}(\kappa) \), which determines the magnitude of the mode’s contribution to \( \phi_{\text{Coul}}^*(r) \) [eqn (86)] and, for oscillatory modes, determines also the phase. \( \varepsilon_{\text{eff}}(\kappa) \) is the physical entity for electrostatic interactions in electrolytes that corresponds to the dielectric constant \( \varepsilon_r \) for pure polar fluids and other non-electrolytes. In contrast to \( \varepsilon_r \) it is not given by the infinite wavelength dielectric response (\( k \to \infty \)) and does not solely contain the dipolar orientational dielectric polarization. As discussed in more detail in ref. 31, the values of the effective dielectric permittivities result from all kinds of polarizations: orientational polarizations (dipolar, quadrupolar, octupolar etc.) and from changes in ion distributions, including transient aggregate formations like ion pairing. The value of \( \varepsilon_{\text{eff}}(\kappa) \) reflects the polarization response of the electrolyte to an exponentially decaying disturbance with decay parameter \( \kappa_r \).

Furthermore, the dielectric factor \( \varepsilon_{\text{eff}}^*(\kappa) \) that appears in the expression for the screening parameter \( \kappa_{\text{eff}} \), eqn (5), has different values for the various modes. As we have seen, \( \varepsilon_{\text{eff}}^*(\kappa) \) replaces the dielectric constant \( \varepsilon_r \) that appears in the expression for the Debye parameter \( \kappa_{\text{D}} \) [eqn (1)]. It is not correct to use a dielectric permittivity obtained from the dielectric polarization response at infinite wavelength instead of \( \varepsilon_{\text{eff}}^*(\kappa) \) and \( \varepsilon_{\text{eff}}^*(\kappa) \), except possibly as an approximation for modes with long decay lengths. Thus, when \( \kappa_r \) is not close to zero, one cannot use experimental “dielectric constants” for electrolytes evaluated macroscopically at \( k = 0 \) from measurements at low frequencies. Furthermore, theoretical estimates of such dielectric constants based on dipolar polarizations including the effect of ions on dipolar orientations are clearly inadequate, since they leave out the polarization from changes in ion distributions and multipole orientations.

It is shown in Section 4.1 that the decay modes of \( h_{ij} \) and \( w_{ij} \) are determined by the screened electrostatic pair interaction \( w^0_{ij} \), which is a part of \( w_{ij} \). The entity \( w^0_{ij} \) is equal to the electrostatic interaction energy of the dressed particles of species \( i \) and \( j \) (with charge densities \( \rho_{*i} \) and \( \rho_{*j} \)) as mediated by the screened Coulomb potential \( \phi_{\text{Coul}}^*(r) \) [eqn (103) and (104)], that is, \( w^0_{ij} = \rho_{*i} \odot \phi_{\text{Coul}}^* \odot \rho_{*j} \), where \( \odot \) denotes a convolution integral. Alternatively this can be expressed as follows: \( w^0_{ij} \) is equal to the interaction energy between the mean electrostatic potential \( \psi_i \) due to the particle of species \( i \) and the charge distribution \( \rho_{*i} \) of a dressed particle of species \( j \) (or vice versa). Thereby we have used the fact that \( \psi_i = \rho_{*i} \odot \phi_{\text{Coul}}^* \), that is, \( \psi_i \) is obtained via Coulomb’s law for the screened potential, eqn (60), where \( \rho_{*i} \) is the source for \( \psi_i \).

The screened Coulomb potential \( \phi_{\text{Coul}}^*(r) \) is a Green’s function that describes the spatial propagation of electrostatic interactions in electrolytes. It is closely related to the dielectric function \( \varepsilon(k) \); in Fourier space we have \( \phi_{\text{Coul}}^*(k) = \phi^*_{\text{Coul}}(k) \varepsilon(k) \), where \( \phi^*_{\text{Coul}}(k) = 1/[(\varepsilon(k) k^2)] \) is the Fourier transform of the ordinary unscreened Coulomb potential \( \phi_{\text{Coul}}(r) \). The decay modes of \( \phi_{\text{Coul}}^*(r) \) are given by the poles of \( \phi_{\text{Coul}}^*(k) \) in complex \( k \)-space, i.e., the zeros of \( \varepsilon(k) \), whereby the screening parameter \( \kappa \) satisfies \( \varepsilon(i\kappa) = 0 \), where the imaginary unit \( i \) appears because the dielectric response is evaluated for an exponentially decaying field. The condition \( \varepsilon(i\kappa) = 0 \) is equivalent to the general exact equation for the screening parameter, eqn (5).

In analyses of experimental measurements of interparticle interactions in electrolytes, including surface force measurements, it is important to realize the existence of the various decay modes with different values of the screening parameter \( \kappa_r \). It is therefore suitable to analyze the data using several decay modes with various decay lengths and, when appropriate, wavelengths. Thereby, the various modes may dominate in different distance intervals.

In both experimental and theoretical work, one can make curve fits to force curves and similar data plots on a log scale in order to find the decay lengths and the wavelengths. In cases where the decay modes have quite different decay lengths, it should be feasible to identify the modes, but if, say, two modes have nearly the same decay lengths, it is in general difficult to distinguish them. It can also be difficult to know if one has determined the functions for sufficiently large distances so the results cover the ultimate decay where the mode with the largest decay length dominates. Furthermore, if the magnitude of a mode is small it may not be possible to detect it.

In theoretical work, including simulations, one should, if practically feasible, solve eqn (5) for the decay parameter or the equivalent equation \( \varepsilon(i\kappa) = 0 \) in order to determine the various \( \kappa_r \). One can then also determine modes that are difficult to detect by fitting. For spherical ions, eqn (99) is an option. The solutions should then be compared with the decay parameters obtained by fitting. Thereby, one has two independent ways to determine the decay modes. Ref. 18 gives examples of how this can be done in practice in simulations of electrolytes with spherical ions.

For binary electrolytes, in particular, it can be useful to plot \( \kappa_i/\kappa_{\text{D}} \) as a function of, for example, temperature or concentration since this ratio has a simple relationship to \( q^* \) and \( \varepsilon^*(\kappa) \), see eqn (81). For spherical ions there is also a simple relationship to the effective ionic charges, see eqn (100).

A central theme of the present work is the demonstration of the fact that when one deals with the screened electrostatic interactions in electrolytes, it is appropriate and often advantageous to use the concept of a dressed particle instead of the entity composed of the particle together with the entire ion/solvent cloud that surrounds it. This cloud consists of the charge distribution \( \rho_i \) due to excess ions and solvent molecules close to the particle, including the effects of their orientational ordering due to interactions. A dressed particle is the particle together with its dress, which has a charge distribution \( \rho_{\text{dress}} \) that is different from \( \rho_i \). The total charge density of the particle and its surrounding cloud is \( \rho_{\text{total}} = \sigma_i + \rho_i \), where \( \sigma_i \) is the internal charge density of the particle. Likewise, we have \( \rho_{\text{dress}} = \sigma_i + \rho_{\text{dress}} \).

While the
ion/solvent cloud $\rho_i$ is described by the pair distribution functions $g_i = h_i + 1$, the dressed $\rho_{i\text{dressed}}$ is described by the distribution functions $g_i^* = h_i^* + 1$ that are related to the former by the simple relationship $g_i^* = g_i + \beta w_i^2 [\text{eqn (112)}]$. The dressed particle charge $q_i^*$ that occurs in eqn (5) for the screening parameter is the total charge of $\rho_i^*$. The reason why it is appropriate to use dressed particles is apparent already when considering the linear polarization response of the bulk electrolyte due to a weak external electrostatic potential $\delta\Phi_{\text{ext}}$, as discussed in Section 2. This response can be described microscopically from the point of view of the free energy of interaction for each constituent particle in the fluid (ion, solvent molecule or any other particle). The interactional free energy $\delta W_i$ for a particle of species $i$ is simply equal to the electrostatic interaction energy $\delta W_i = \rho_{i\text{tot}} \otimes \delta\Phi_{\text{ext}}$ between $\delta\Phi_{\text{ext}}$ and $\rho_{i\text{tot}} [\text{eqn (16)}$ and (23)]. However, $\delta\Phi_{\text{ext}}$ is the potential from the external charges in the absence of the electrolyte, while the total potential $\delta\Phi$, which also includes the potential $\delta\Phi_{\text{ind}}$ from the induced polarization charge density, is the actual potential in the presence of the electrolyte. The latter gives the real situation for the particles in the system and, as shown in Section 3, the free energy $\delta W_i$ is equal to the electrostatic interaction $\delta W_i = \rho_{i\text{tot}} \otimes \delta\Phi_{\text{ext}}$ between $\delta\Phi_{\text{ext}}$ and $\rho_{i\text{tot}} [\text{eqn (41)}]$. Thus, the use of dressed particles reflects the actual conditions for the particles in the weakly polarized electrolyte. The strong, nonlinear polarization in the immediate neighborhood of the $i$-particle, which is due to the interactions between the particle and its surroundings, is included in the dress. The linear part of the electrostatic polarization due to the particle, $\rho_{i\text{lin}}$, is included in $\rho_{i\text{tot}}$ but not in $\rho_{i\text{tot}}^*$, and we have $\rho_{i\text{tot}}^* = \rho_{i\text{tot}}^* - \rho_{i\text{lin}} [\text{eqn (44)}]$. The contribution from this linear part to the potential of mean force $\delta W_i = \rho_{i\text{tot}}^* \otimes \delta\Phi_{\text{ext}}$ is instead included in the factor $\delta\Phi_{\text{ind}}$ via the collective response to the external potential from all particles in the electrolyte.

A similar decomposition in linear and nonlinear parts of the electrostatic polarization due to a particle is done in the calculation of $\psi_i$ via Coulomb’s law. The mean potential $\psi_i$ can be obtained from $\rho_{i\text{tot}}^*$ by using the usual unscreened Coulomb potential, $\phi_{\text{Coul}}(r) = \frac{1}{4\pi\epsilon_0 \kappa r}$ in this law [eqn (38)]. As mentioned above, exactly the same potential $\psi_i$ can alternatively be obtained from $\rho_{i\text{tot}}^*$ by using $\phi_{\text{Coul}}(r)$ in Coulomb’s law [eqn (60)]. The difference is that the linear part of the electrostatic polarization in the latter case is included via $\phi_{\text{Coul}}(r)$, which contains an electrostatic linear response $\kappa(k)$, while the former case contains a factor $\kappa(k)$.

The decay modes of $\psi_i$ are always the same as those of $\phi_{\text{Coul}}(r)$. The magnitude of each of these decay modes of $\psi_i$ is proportional to a kind of effective charge $Q_i^{\text{eff}}$ of the dressed $i$-particle, which has different values for the different modes, and inversely proportional to $\delta\Phi_i$ [eqn (93)]. The values of $Q_i^\text{eff}$ are determined by the projections of $\rho_{i\text{tot}}^*$ on the various modes, which define $Q_i^{\text{eff}}$ for each mode [eqn (94)]. For the oscillatory modes there is also a phase shift determined by $\rho_{i\text{tot}}^*$. In this case, $Q_i^{\text{eff}}$ is complex-valued and its absolute value gives the magnitude of the mode and its phase gives the phase shift. It is clear that the concept of dressed particle has a fundamental role since $\rho_{i\text{tot}}^*$ directly determines the magnitudes and the phase shifts of the modes of $\psi_i$. For a nonspherical particle, the effective charge and the phase shift are direction dependent quantities which reflect the anisotropy of the mean potential $\psi_i$. The former is therefore denoted as the “multipolar effective charge” [Section 3.2.4].

The fundamental role of the dressed particles is further accentuated from their appearance in $w_i^0$ as mentioned earlier and the fact that $w_i^0$ in general determines all decay modes of $h_j^\text{ext}$ and $w_j^\text{ext}$. The magnitude of each decay mode of $h_j^\text{ext}$ and $w_j^\text{ext}$ is proportional to the product $Q_i^{\text{eff}}Q_j^{\text{eff}}$ as evaluated for each mode [eqn (116)]. For oscillatory modes the phase shift contains the sum of the phases of $Q_i^{\text{eff}}$ and $Q_j^{\text{eff}}$ [eqn (117)].

The correlation functions $H_{\text{NN}}(r)$, $H_{\text{NNQ}}(r)$ and $H_{\text{QQQ}}(r)$ have in general the same set of decay modes as $w_i^0$ and all three functions therefore have the same decay lengths. Their magnitudes are proportional to a set of prefactors $Q_i^{\text{eff}}$ and $Q_j^{\text{eff}}$ that constitute averages of the effective charges of the particles in the electrolyte [eqn (126)]. The modes of $H_{\text{NNN}}(r)$ are proportional to $(Q_i^{\text{eff}})^2$, those of $H_{\text{NNQ}}(r)$ proportional to $Q_i^{\text{eff}}Q_j^{\text{eff}}$ and those of $H_{\text{QQQ}}(r)$ proportional to $(Q_i^{\text{eff}})^3$ [eqn (127)–(129)].

For charge-inversion invariant systems, where $H_{\text{NNQ}}(r)$ is identically equal to zero, the decay modes of $H_{\text{QQQ}}(r)$ are determined by $w_i^0$, but those for $H_{\text{NNQ}}(r)$ are different. The latter are instead determined by the correlation functions $h_j^\text{ext}$ of the dresses [eqn (122)], so the concept of dressed particles has in this case yet another fundamental role. The decay modes of $H_{\text{NNN}}(r)$ are here given by the poles of $h_j^\text{ext}$, which are different from the poles of $\phi_{\text{Coul}}(k)$. Only in charge-inversion invariant systems there is a strict distinction between “electrostatic dominance” and so-called “core dominance” (in the latter, packing effects dominate), for example in the restricted primitive model.39 The former kind of dominance occurs when $H_{\text{QQQ}}(r)$ has the longest decay length and the latter when $H_{\text{NNQ}}(r)$ has the longest one.

There exist electrolytes that are nearly charge-inversion invariant, for example the NaCl system mentioned in the Introduction and any binary system in the primitive model that have anions and cations of the same valency but with slightly different sizes. For such systems, the conclusions for the main case apply and $h_{\text{NNQ}}$, $h_{\text{NNQ}}$, $h_{\text{NNQ}}$, $H_{\text{NNQ}}(r)$, $H_{\text{QQQ}}(r)$ and $H_{\text{QQQ}}(r)$ have the same set of decay modes. In this case the poles of $\phi_{\text{Coul}}(k)$ [i.e., zeros of $\kappa(k)$] can be divided into two categories depending on what takes place when the system is converted into one that is charge-inversion invariant; in the given primitive model example this happens when the ion sizes are made equal. Each pole in the first category lies close to a pole of $h_j^\text{ext}$ and the two poles move towards each other when the system is converted into an invariant one. When the system becomes invariant the pole of $\phi_{\text{Coul}}(k)$ vanishes. The decay mode remains but is instead determined by the pole of $h_j^\text{ext}$ for the invariant case. Each pole of $\phi_{\text{Coul}}(k)$ in the second set continues to be such a pole throughout. For the decay modes given by first set of poles $Q_i^{\text{eff}} > |Q_j^{\text{eff}}|$ and $Q_i^{\text{eff}}$ becomes zero when the system becomes charge-inversion invariant, while for the second set $|Q_i^{\text{eff}}| < |Q_j^{\text{eff}}|$ and $Q_i^{\text{eff}}$ becomes zero for the invariant case. Thereby, the various modes are smoothly changed when the electrolyte is turned into a charge-inversion invariant one.
Conflicts of interest

There are no conflicts to declare.

Appendices

A Response due to variation in external electrostatic potential for the nonspherical case

For an inhomogeneous fluid of nonspherical particles Yvon’s first equation is

\[ \delta n_i(R_1) = \delta n_1(R_1) + \sum_j \left[ dR_2 h(y(R_1, R_2)) \delta n_j(R_2) \right] \]

and for a bulk fluid perturbed by the weak external potential \( \delta \nu_i(R) \equiv \delta \nu_i(r, \omega) \) this becomes

\[ \delta n_i(R_1) = \delta n_1(R_1) + \sum_j \left[ \int dR_2 h(y(R_1, R_2)) \delta n_j(R_2) \right] \]

When the external potential is solely due to the electrostatic potential \( \delta \nu^ext(r) \) we have

\[ \delta \nu(R_2) = \int dr_R \sigma_i(r | R_2) \delta \nu^ext(r) \]

and hence we obtain

\[ \delta n_i(R_1) = \delta n_1(R_1) + \sum_j \left[ \int dR_2 h(y(R_1, R_2)) \sigma_i(r | R_2) \delta \nu^ext(r) \right] \]

We now introduce the charge density \( \rho_i(r_3 | R_1) \) at \( r_3 \) around a particle with coordinates \( R_1 \) (i.e., it is located at \( r_1 \) and has orientation \( \omega_1 \))

\[ \rho_i(r_3 | R_1) = \sum_j \left[ \int dR_2 g_0(R_1, R_2) n_j \sigma_i(r_3 | R_2) \right] \]

\[ = \sum_j \left[ \int dR_2 h(y(R_1, R_2)) n_j \sigma_i(r_3 | R_2) \right] \]

is hence given by

\[ \delta \rho_i(r_1) = \sum_j \left[ \int dR_1 \sigma_i(r_1 | R_1) \delta n_i(R_1) \right] \]

\[ = - \beta \epsilon_{i}^2 \int dR_1 H_{QQ}(r_1) \delta \psi^ext(r_1), \]

where \( H_{QQ} \) is the charge–charge correlation function given in eqn (134).

B The \( H_{QQ}, H_{NN} \) and \( H_{SN} \) correlation functions

The charge–charge correlation function \( H_{QQ} \) is defined as

\[ H_{QQ}(r_{12}) = H_{QQ}(r_1, r_2) \equiv \frac{1}{q^2} \int dR_3 h(y(R_1, R_2)) h(y(R_1, R_3)) h(y(R_2, R_4)) h(y(R_3, R_4)), \]

which can be written as

\[ H_{QQ}(r_{12}) = \frac{1}{q^2} \int dR_3 h(y(R_1, R_2)|R_3) h(y(R_1, R_2)|R_3) \]

\[ = \frac{1}{q^2} \int dR_3 \langle \delta n_i(r_3 | R_1) \delta n_j(r_3 | R_2) \rangle \]

and in Fourier space this is

\[ \tilde{H}_{QQ}(k) = \frac{1}{q^2} \int dR_3 \langle \delta n_i(-k, \omega_3) \delta n_j(k, \omega_3) \rangle \]

which is independent of \( \hat{k} = k/k. \)

The density–density correlation function \( H_{NN} \) is defined as

\[ H_{NN}(r_{12}) = H_{NN}(r_1, r_2) \]

\[ \equiv \int dR_3 \delta^3(r_{12}) + \int dR_3 \langle h(y(R_1, R_2), R_3) \rangle \]

\[ = \frac{1}{q^2} \int dR_3 \delta^3(r_{12}) + \int dR_3 \langle h(y(R_1, R_2), R_3) \rangle \]

where we have written \( h(y(R_1, R_2), R_3) = h(y(R_1, R_2), R_3) \). \( H_{NN}(r_{12}) \) describes the correlations in fluctuations in total number densities at \( r_1 \) and at \( r_2 \). Its Fourier transform is

\[ \tilde{H}_{NN}(k) = \frac{n_0^3}{q^2} + \int dR_3 \langle h(y(k, \omega_1, \omega_2)) \rangle \]

where \( n_0^3 = \frac{1}{q^2} \int dR_3 \delta^3(r_{12}) + \int dR_3 \langle h(y(R_1, R_2), R_3) \rangle \)
Finally, the density-charge correlation function $H_{\text{NQ}} \equiv H_{\text{QN}}$ is defined as

$$H_{\text{NQ}}(r_12) = H_{\text{NQ}}(r_1, r_2) \equiv \frac{1}{q_c} \sum_{i} \int d\omega_1 n_i^b \sigma_i(r_1, \omega_1)$$

$$+ \frac{1}{q_c} \sum_{q} \int d\omega_1 d\omega_2 n_i^b \times \hat{h}_{ij}(r_{13}, \omega_1, \omega_3) n_j^b \sigma_j(r_{32}, \omega_2)$$

$$= \frac{1}{q_c} \sum_{i} \int d\omega_1 n_i^b \rho^\text{qo}_i(r_1, \omega_1).$$

(139)

It describes the correlations in fluctuations in total number density at $r_1$ and the charge density at $r_2$ and vice versa. In Fourier space this is

$$\tilde{H}_{\text{NQ}}(k) = \frac{1}{q_c} \sum_{i} n_i^b \langle \rho^\text{qo}_i(\mathbf{k}, \omega_1) \rangle.$$  

(140)

C Relationships between the poles of $\tilde{h}_{ij}$ and $\tilde{h}_{ij}^*$

In this appendix we will investigate the relationships between the poles of $\tilde{h}_{ij}$ and $\tilde{h}_{ij}^*$. In particular we will show that each pole of $\tilde{h}_{ij}^*(k)$ in eqn (113) is normally cancelled by a corresponding pole in the last term of this equation. This implies that $\tilde{h}_{ij}$ and $\tilde{h}_{ij}^*$ cannot have poles for the same $k$ values (apart from some exceptional cases treated in detail in Section 4.2). We will also investigate some cases where poles of $\tilde{h}_{ij}^*$ can influence where the poles of $\tilde{h}_{ij}$ appear.

The existence of a simple pole for $\tilde{h}_{ij}^*$ at $k = \pm iz$ means that $\tilde{h}_{ij}^*$ diverges like

$$\tilde{h}_{ij}^*(k, \omega_1, \omega_2) \sim \frac{F_{ij}(iz \mathbf{k}, \omega_1, \omega_2)}{k^2 + z^2}$$

where $F_{ij}(k, \omega_1, \omega_2)$ is a function that is finite at $k = \pm iz$. One can show that $F_{ij}$ factorizes$^\text{§§}$ at $k = iz$ as $F_{ij} = \tau_{ij}/K$, where $\tau_i$ for $i = j$ is a function of orientation for a given $z$ and $\mathbf{k}$ and where $K = K(z)$ is a constant, and we have

$$\tilde{h}_{ij}^*(k, \omega_1, \omega_2) = \frac{\tau_{ij}(z \mathbf{k}, \omega_1, \omega_2)}{k^2 + z^2} + \tilde{h}_{ij}^\text{rest}(k, \omega_1, \omega_2)$$

(141)

where $\tilde{h}_{ij}^\text{rest}$ is a part of $\tilde{h}_{ij}^*$ that stays finite at $k = iz$. This means that this pole gives a term in $\tilde{h}_{ij}^*$ in ordinary space that decays as

$$h_{ij}^*(r, \omega_1, \omega_2) \text{ term: } \frac{\tau_i(r, \omega_1, \omega_2)}{4\pi K(z)r} e^{-ar}$$

when $r \rightarrow \infty$.

The factorization in eqn (141) is crucial for the cancellation of poles as we now will see. The Fourier transform of eqn (111) is given by the first line in the following equation and by inserting eqn (141) we obtain the second line

$$\tilde{\rho}^*_i(k, \omega_1) = \tilde{\sigma}_i(k, \omega_1) + \int d\omega_2 h_{ij}^*(k, \omega_1, \omega_2) \tilde{\sigma}_j(k, \omega_2)$$

$$= \tilde{\sigma}_i(k, \omega_1) + \tau_i(k, \omega_1, z)E(z, x) + \tilde{\rho}^\text{rest}_i(k, \omega_1),$$

(142)

where

$$E(z, x) = \int d\omega_1 \tilde{\sigma}(z, \omega_1);$$

and $\tilde{\rho}^\text{rest}_i$ is the part of $\tilde{\rho}^*_i$ originating from $\tilde{h}_{ij}^\text{rest}$. $E(z, x)$ does not depend on the direction of $\mathbf{k}$ due to the averaging over $\omega_1$. The dielectric function $\varepsilon(k)$ [given by eqn (50)] can be written as the first line in the following equation and by inserting eqn (142) we obtain the second line

$$\tilde{\varepsilon}(x) = 1 + \beta \phi_{\text{Coul}}(k) \sum_i \int d\omega_1 \rho^\text{qo}_i(k, \omega_1) \tilde{\sigma}_i(-k, \omega)$$

$$= 1 + \beta \phi_{\text{Coul}}(k) \frac{E^2(k, x)}{(k^2 + x^2)K(x)} + \tilde{\varepsilon}^\text{rest}(k),$$

where $\tilde{\varepsilon}^\text{rest}$ is the part of $\tilde{\varepsilon}(k)$ originating from $\tilde{\sigma}_i + \tilde{\rho}^\text{rest}_i$. Since $E(k, x)$ is normally nonzero, the functions $\tilde{h}_{ij}^*, \tilde{\rho}^*_i$ and $\tilde{\varepsilon}(k)$ diverge when $k \rightarrow iz$ due to $k^2 + x^2$ in the denominator. The remaining terms in the expressions stay finite for $k = iz$.

We now insert eqn (142) and (144) into the last term in eqn (113) and since $\tilde{\rho}^*_i$, $\tilde{\rho}^\text{rest}_i$ and $\tilde{\varepsilon}(k)$ are dominated by the diverging contributions for $k$ values close to $iz$, it follows that we have when $k \rightarrow iz$

$$\beta \phi_{\text{Coul}}(k) \frac{\tilde{\phi}_{\text{Coul}}(k) \tilde{\rho}^*_i(k, \omega_1) \tau_i(-k, \omega_2, x) E(k, x)}{(k^2 + x^2)K(x)}$$

$$= \frac{\tau_i(z \mathbf{k}, \omega_1, \omega_2)}{k^2 + z^2} E(k, x)$$

so on the rhs of eqn (113) this pole cancels the pole of $\tilde{h}_{ij}^*$ given by the first term on the rhs of eqn (141). This means that provided $E(k, x)$ is nonzero, the only poles of the entire rhs of
eqn (113) are those that originate from the zeros of $\tilde{a}(k)$ in the denominator.

Charge-inversion invariant systems, which are treated in Section 4.2, are, however, different. For such systems $E(k,\alpha)$ is identically equal to zero because for each positive contribution on the rhs of eqn (143) there is an equally large negative contribution. Therefore, the pole of $\tilde{h}_y^*$ at $k = i\alpha$ is not cancelled in $\tilde{h}_y$. For other systems it may happen at exceptional points in the system's parameter space that $E(k,\alpha)$ fortuitously happens to be equal to zero, which means that the cancellation may not take place at such points. Otherwise the cancellation always occurs and the poles of $\tilde{h}_y$ are given by the zeros of $\tilde{a}(k)$.

For systems that are close to being charge-inversion invariant, $E(k,\alpha)$ is small but nonzero. Such systems should have properties that are very similar to the corresponding charge-inversion invariant system. For the latter, it is shown in Section 4.2 that the decay parameter values of $\tilde{h}_y$ belong to two groups: those that are poles of $\tilde{h}_y^*$ and those that zeros of $\tilde{a}(k)$. When the system deviates only slightly from being charge-inversion invariant and all poles of $\tilde{h}_y$ are given by the zeros of $\tilde{a}(k)$, the former group must have turned into zeros of $\tilde{a}(k)$ because the poles of $\tilde{h}_y^*$ are not poles of $\tilde{h}_y$. This is indeed the case as we now are going to see.

When $E(k,\alpha)$ is small but nonzero and $\tilde{h}_y^*$ has a pole at $k = i\alpha$, the function $\tilde{h}_y$ does not have a pole there, but instead there is, in fact, a pole of $\tilde{h}_y$ close to this $k$ value, that is, there exist a zero of $\tilde{a}(k)$ close to $k = i\alpha$, say, at $k = i\kappa_0$. This can be realized as follows. Since $\tilde{a}(i\kappa_0) = 0$ we obtain from eqn (144)

$$1 - \frac{\beta E^2(k,\alpha)}{\tilde{v}_0\kappa_0^2(\alpha^2 - \kappa_0^2)K(\alpha)} + \tilde{v}_{\text{rest}}^2(k_0) = 0.$$  \hspace{1cm} (145)

Provided that $E(k,\alpha) \neq 0$ we can write this as

$$\left(1 + \tilde{v}_{\text{rest}}^2(k_0)\right)(\alpha^2 - \kappa_0^2) = \frac{\beta E^2(k,\alpha)}{\tilde{v}_0\kappa_0^2K(\alpha)}.$$  \hspace{1cm} (146)

When $E^2(k,\alpha) \approx 0$ this equation has a solution $\kappa_0 \approx \pm \alpha$ with the property $\kappa_0 \to \pm \alpha$ when $E(k,\alpha) \to 0$. This is the solution that we are looking for. Its existence can also be inferred directly from eqn (145) because $E^2(k,\alpha)/[\alpha^2 - \kappa_0^2]$ must remain finite in this limit since $\tilde{v}_{\text{rest}}^2(k_0)$ stays finite there. Likewise, there exist solutions of $\tilde{a}(k) = 0$ that are connected in the same manner to other poles of $\tilde{h}_y^*$. The remaining solutions $\kappa_{0'}$ of eqn (146) are not interesting here because in the same limit they go to the solutions of $1 + \tilde{v}_{\text{rest}}^2(k_{0'}) = 0$, which is the same as $\tilde{a}(k_{0'}) = 0$ as seen from eqn (144) with $E(k,\alpha) = 0$. Thus they belong to the second group mentioned above for charge-inversion invariant systems (these solutions are poles of $\tilde{h}_{D,\text{AB}}$ in eqn (125) and not poles of $\tilde{h}_y^*$).

References


