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Surface heterogeneity affects percolation and gelation of colloids: dynamic simulations with random patchy spheres†

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Surface heterogeneity of colloidal particles has a significant impact on their structure in solution and their rheological properties. During particle synthesis, heterogeneous chemical functionalization, processes of self-assembly, or phase separation, can all lead to heterogeneous colloidal surfaces which impart anisotropic interactions to suspended particles. Additionally, an important class of colloids, biological macromolecules, exhibit similar localized, short-ranged, anisotropic interactions, which have a significant impact on their solution properties. Therefore, understanding the assembly and rheology of such colloids can provide insight into a wide variety of relevant physical systems. In this computational study, we investigate dispersions of particles having surface patches with randomized functionality as a model for heterogeneous colloids. We use Brownian dynamics simulations with hydrodynamic interactions to explore the differences between these random patchy particles and homogeneous (or isotropic) particles. The common basis used for comparing dispersions of particles with different surface functionality is equality of the second virial coefficient, so that dispersions of particles with different patterns of surface heterogeneity are similar thermodynamically at low particle concentrations. We show that at modest particle concentrations, significant deviations from the isotropic model are evident in the dispersion micro-structure, giving drastically different percolation transition points depending on the degree of surface heterogeneity. However, these deviations can be rationalized and a universal percolation criteria derived in terms of the osmotic pressure of the dispersion. Heterogeneous interactions also impose extra constraints on the relative translation and rotation between neighboring particles, which increase the viscosity and elastic modulus of aggregated dispersions and gels built from heterogeneous colloids and shifts the gel point measurably.

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1 Introduction

Colloidal particles with short-range interactions are ubiquitous in functional materials and pharmaceuticals. Versatile physical properties can be achieved by different synthetic routes, which make them useful for a variety of applications.^{1–9} The ability to understand and predict the behavior of dispersions of these particles from the microscopic properties of individual particles will enable rational design of colloidal materials for specific applications. Inter-colloid interactions are commonly treated as isotropic,^{10–12} which for many materials can be a useful approximation for interpreting experiments.^{13–16} However, heterogeneities in the inter-particle interaction are common at the colloidal scale, and can have significant impact on micro-structure and rheology of colloidal dispersions. Anisotropic inter-particle

interactions are usually caused by heterogeneous chemical functionalization,^{17,18} self-assembly,^{19,20} or phase separation²¹ during particle synthesis. Anisotropic interactions are also common between proteins and other biological macromolecules.^{22–25} A fundamental understanding of the impact of surface heterogeneity on the micro-structure and rheology of colloidal particles will aid in the rational design of their functionalities.

One important effect of surface heterogeneity is a constraint on the relative translation and rotation of neighboring particles. This extra constraint can potentially change the dynamics and rheology of attractive colloidal dispersions. This is illustrated schematically in Fig. 1, where dispersions of spherical particles having homogeneous or heterogeneous surfaces are assumed to have identical percolated micro-structure. The homogeneous surface allows neighboring particles to freely rotate relative to each other. The rotational constraint between heterogeneous colloids generates additional bending moments that resist deformation. This changes the complex modulus and the gel point of the heterogeneous suspension relative to the homogeneous suspension.

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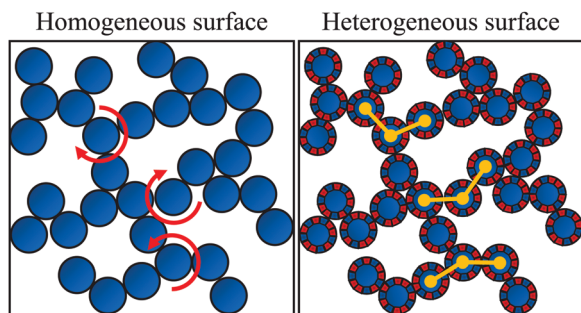


Fig. 1 Schematic comparing particles with homogeneous surfaces (left) and heterogeneous surfaces (right). An identical percolated structure is chosen for both particle types. Even for identical percolated micro-structures, the anisotropic interactions due to heterogeneous surfaces constrain the heterogeneous particles from translating or rotating relative to their neighbors and generate bending moments that resist deformation.

Colloidal gels are important soft material for their unique mechanical properties. They are widely applied in paints,⁵ cosmetics,⁶ and food products,⁷ and are being developed for new applications such as tissue engineering⁸ and 3D printing.⁹ Despite the numerous applications of colloidal gels, the location of the gel point is not well understood. The gel transition, or critical gel point, is commonly defined as the condition at which the power law scaling with respect to frequency of both storage modulus G' and loss modulus G'' coincide for a broad range frequencies,^{26,27} and is typically determined using small amplitude oscillatory shear (SAOS) experiments. However, other definitions of the critical gel point have been used.^{28,29} For example, “percolation”, which is defined as the appearance of an infinitely spanning network in the suspension, is sometimes used interchangeably with term “gelation”.³⁰ Percolation, a structural criterion, does not necessarily ensure gelation, a rheological criterion. Eberle and coworkers^{31,32} found that the critical gel point overlapped perfectly with the percolation transition for polymer brush grafted nanoparticles. However, other works^{16,28} have shown that spinodal decomposition, which requires much stronger inter-particle attractions than exhibited at the percolation transition, is necessary for some colloidal dispersions to form gels. These diverse experimental results from previous works suggest there are different physical attributes of the particles critical for understanding their gelation. Surface heterogeneity is a candidate, which will be investigated in this work.

Understanding the influence of surface heterogeneity has an especial implication for protein solution rheology. Protein molecules are highly heterogeneous colloidal particles having a widely varied surface charge distribution and hydrophobic patches.^{15,24,33} Protein surface functionalities significantly impact the solution micro-structure and viscosity,^{23,34} which is a phenomenology that is not explained by isotropic liquid theories^{22,35} based on spherically averaged attractions and lumped charges. Our recent work³⁶ has shown that constraint on relative motion between nearly touching proteins explains the high viscosity observed in concentrated solutions of monoclonal antibodies.

The patchy particle model is one of the simplest and most flexible models that can capture heterogeneity. Many different

types of patchy particles have been synthesized with regular arrangements of surface patches, and their phase behavior^{22,37–39} and self-assembly^{40,41} has been extensively studied. However, these studies primarily focus on particles with geometrically ordered patch architectures.^{42,43} For colloidal dispersions and biomacromolecules in general, a disordered, or random distribution of patches is more representative. In the present work, we develop a flexible random patchy sphere model and use Brownian dynamics simulations with hydrodynamic interactions to investigate the micro-structure, thermodynamic properties, and rheology of concentrated dispersions. Patchy particles are compared with uniform particles to elucidate the effect of surface heterogeneity on percolation and gelation. Recent work by Blanco and Shen (2016),⁷⁰ has analyzed spherical particles with a small number of symmetrically positioned patches reflecting different dipolar and quadrupolar charge distributions that could be correlated with phase behavior and percolation.

2 Methods

2.1 Random patchy sphere model

We utilized a model colloidal particle similar to that proposed by Van Lehn and Alexander-Katz for nanoparticle–lipid bilayer simulations,⁴⁴ and apply this model to study dispersions. We allow that the surface of our model spherical particle can be occupied by two types of patches, as shown in (Fig. 2a). The patches are identified on the surface of the particle with beads whose positioning is determined by a geodesic decomposition of the particle surface into a Goldberg polyhedron. Beginning with an initial icosahedral tessellation, the edges of the polyhedron are iteratively bisected to generate a new set of vertices on which the beads centers are positioned. Each bead on the polyhedron is randomly assigned one of two states, +1 (red) or –1 (blue), as shown in (Fig. 2a). To control the spatial characteristics of this assignment process, a Monte Carlo simulation of the Ising model on the tessellated spherical surface is applied to randomly assign the states to the beads. The energy of the Ising model is defined as:

$$E_{\text{Ising}} = J \sum_{\langle i,j \rangle} \sigma_i \sigma_j, \quad (1)$$

where σ_i is either +1 or –1, representing two types of sites, J is the interaction energy factor or Ising coupling parameter, and $\langle i,j \rangle$ represents all neighboring beads i and j on the spherical surface. A schematic plot showing converged samples from the Monte Carlo simulation can be found in (Fig. 2b). By tuning the dimensionless Ising coupling parameter βJ , which is the energy factor J normalized by the thermal energy scale ($\beta = 1/k_B T$) in the Monte Carlo simulation, the patchiness can be controlled systematically. (Fig. 2d) shows that at lower βJ sites prefer to have the same type as their neighbors, while at higher βJ values neighboring sites of the same type are energetically penalized. In addition, a constraint forcing the total number of each type of site to be equal can be imposed (called “balanced”) or not (called “unconstrained”). As the βJ value decreases and becomes negative, “phase separation” can be



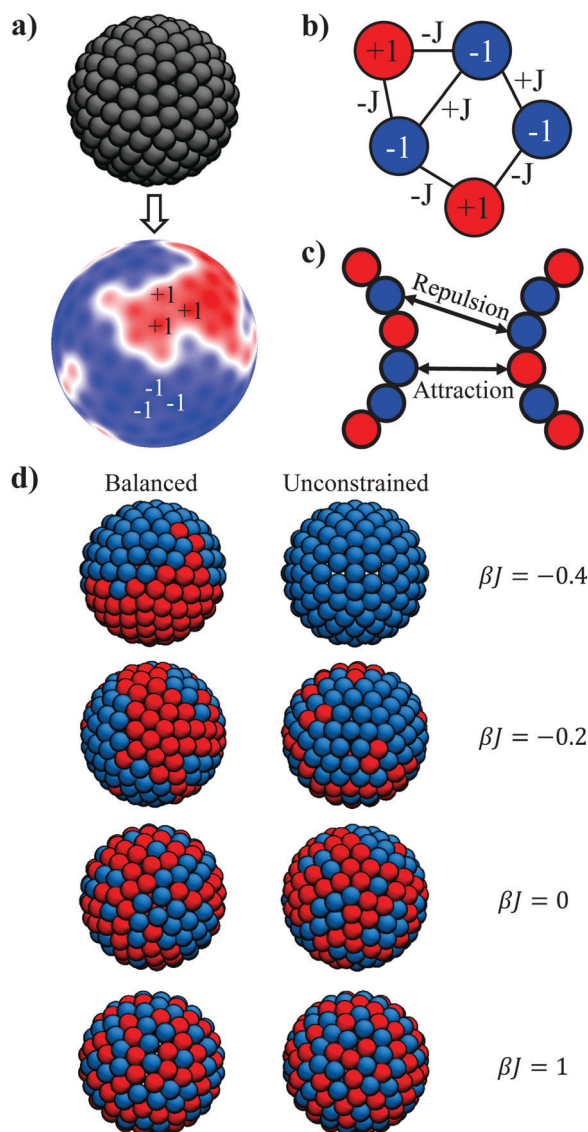


Fig. 2 The random patchy sphere model. (a) Spherical surface is tessellated with beads that are assigned one of two states, +1 (red) or -1 (blue), to generate patchiness. (b) A schematic of the Ising model on sphere surface. A Monte Carlo simulation of the Ising model is applied to generate controllable random assignment. (c) Different interactions between different types of sites. Sites of the same type repel while sites of opposite type attract. (d) Tunable patch sizes by tuning the βJ value in the Ising model. A constraint forcing the number of each type of site to be equal can be imposed (“balanced”) or not (“unconstrained”).

observed in the Ising model, and the patchy particle appears as a Janus particle (“balanced”) or isotropic particle (“unconstrained”). We choose particles with 42 beads representing distinct sites on the surface of each spherical particle for the remainder of this work.

The force and torque acting on these particles can be obtained by summing up the site–site interactions. In addition to a stiff quadratic repulsive interaction representing excluded volume effects, beads interact *via* pairwise attractive or repulsive interactions, depending on the site types in the pair, to model heterogeneous surface effects. Throughout the main text, sites of the same type repel while sites of opposite type attract, as shown

in (Fig. 2c). The appendix considers the implications of an inverted set of interactions. The Yukawa potential is used for the functional form of the site–site interactions because of its simplicity and applicability to modeling screened electrostatics:

$$E(r) = \pm E_Y \frac{a_B}{r} \exp(-\kappa r), \quad (2)$$

where r is the center-to-center distance between two bead sites, E_Y is the interaction strength of the Yukawa potential, a_B is the radius of individual bead sites, and κ sets interaction range. In the present work, we choose $\kappa a_B = 1$ so that the interaction is short-ranged relative to the length scale of the patchy particle. The sign of this potential is either positive (same type) or negative (opposite type) depending on the types of the pair of sites.

2.2 Brownian Dynamics of rigid composite-bead particles

2.2.1 Governing equations. Brownian dynamics (BD) simulation with hydrodynamic interactions is used to simulate the suspension dynamics of composite-bead particles with heterogeneous surfaces. For any rigid composite-bead particle in a quiescent suspension, its trajectory satisfies the Langevin equation:

$$\frac{d}{dt} \begin{bmatrix} \mathbf{p} \\ \mathbf{L} \end{bmatrix} = \begin{bmatrix} \mathbf{F}^H \\ \mathbf{T}^H \end{bmatrix} + \begin{bmatrix} \mathbf{F}^B \\ \mathbf{T}^B \end{bmatrix} + \begin{bmatrix} \mathbf{F}^P \\ \mathbf{T}^P \end{bmatrix}, \quad (3)$$

where \mathbf{p} and \mathbf{L} are the linear momentum and angular momentum of all the particles, respectively. The force (\mathbf{F}) and torque (\mathbf{T}) vectors consist of three contributions: (i) a hydrodynamic contribution (superscript H) caused by drag of the solvent on the particle; (ii) a stochastic contribution (superscript B) responsible for Brownian motion of the particles; and (iii) a deterministic non-hydrodynamic inter-particle contribution (superscript P) defined as the negative gradient of the interaction potential, $-\nabla E$, including effects from excluded volume and heterogeneous site–site interactions.

For a micrometer or nanometer sized particle in a viscous fluid, the characteristic time scale for momentum relaxation, or the inertial time scale is approximately $\tau^I = \rho R_H^2 / \eta_s$, where ρ is the density of the particle, R_H is the hydrodynamic radius of the particle, and η_s is the viscosity of the solvent. This time scale about 10^{-6} seconds for micron-sized colloids in water. However, the diffusion time scale $\tau^D = 6\pi\eta_s R_H^3 / k_B T$ which describes the characteristic time for Brownian motion, is approximately 1 second in the same scenario. Thus, changes in the momenta occur much faster than all other dynamic processes, and the left hand side of the Langevin eqn (3) is approximately zero in this over-damped regime.

Hydrodynamic forces and torques (\mathbf{F}^H and \mathbf{T}^H) are long-ranged interactions due to the flow field perturbation caused by particle motion in a viscous fluid. They greatly affect the dynamics of colloidal dispersions, and including them in computational models is crucial to accurately explain kinetic phenomena such as gelation,^{13,14} and quantify transport properties such as diffusivity and viscosity.^{36,45} The hydrodynamic force \mathbf{F}^H and torque \mathbf{T}^H in the over-damped regime are linear in



the particles' translational velocity \mathbf{U} and angular velocity $\boldsymbol{\Omega}$:⁴⁶

$$\begin{bmatrix} \mathbf{F}^{\text{H}} \\ \mathbf{T}^{\text{H}} \end{bmatrix} = - \begin{bmatrix} \mathbf{R}_{\text{FU}} & \mathbf{R}_{\text{F}\boldsymbol{\Omega}} \\ \mathbf{R}_{\text{TU}} & \mathbf{R}_{\text{T}\boldsymbol{\Omega}} \end{bmatrix} \cdot \begin{bmatrix} \mathbf{U} \\ \boldsymbol{\Omega} \end{bmatrix}, \quad (4)$$

where \mathbf{R} matrices with different subscripts represent force/torque–velocity/angular velocity couplings and are called resistance tensors. Therefore, the governing equation of particle motion can be converted to:

$$\begin{bmatrix} \mathbf{U} \\ \boldsymbol{\Omega} \end{bmatrix} = \begin{bmatrix} \mathbf{R}_{\text{FU}} & \mathbf{R}_{\text{F}\boldsymbol{\Omega}} \\ \mathbf{R}_{\text{TU}} & \mathbf{R}_{\text{T}\boldsymbol{\Omega}} \end{bmatrix}^{-1} \cdot \left(\begin{bmatrix} \mathbf{F}^{\text{B}} \\ \mathbf{T}^{\text{B}} \end{bmatrix} + \begin{bmatrix} \mathbf{F}^{\text{P}} \\ \mathbf{T}^{\text{P}} \end{bmatrix} \right). \quad (5)$$

Fig. 3 shows a schematic plot of this system of linear equations as we apply them to random patchy particles.

Our previous works^{36,45} efficiently evaluated the dynamics of such assemblies by describing bead–bead hydrodynamic interactions with the Rotne–Prager–Yamakawa (RPY) mobility tensor⁴⁷ $\mathcal{M}_{\alpha\beta}^{\text{RPY}}$, which is a far-field approximation for velocity–force couplings between spherical beads:

$$\mathbf{u}_\alpha = \sum_{\beta} \mathcal{M}_{\alpha\beta}^{\text{RPY}} \cdot \mathbf{f}_\beta, \quad (6)$$

where \mathbf{u}_α and \mathbf{f}_β represent velocity of bead α and force on bead β , respectively. $\mathcal{M}_{\alpha\beta}^{\text{RPY}}$ is the block at row α , column β of the grand mobility matrix \mathcal{M} :

$$\mathcal{M}_{\alpha\beta}^{\text{RPY}} = \begin{cases} \frac{1}{6\pi\eta_s a} \mathbf{I} & \text{if } \alpha = \beta, \\ \left(1 + \frac{a^2}{3} \nabla^2\right) \frac{1}{8\pi\eta_s r} (\mathbf{I} + \hat{\mathbf{r}}\hat{\mathbf{r}}) & \text{if } \alpha \neq \beta \text{ and } r \geq 2a, \\ \frac{1}{6\pi\eta_s a} \left[\left(1 - \frac{9r}{32a}\right) \mathbf{I} + \frac{3r}{32a} \hat{\mathbf{r}}\hat{\mathbf{r}} \right] & \text{if } \alpha \neq \beta \text{ and } r < 2a, \end{cases} \quad (7)$$

where a is the hydrodynamic radius of each bead, and $\hat{\mathbf{r}}$ and r are the unit vector and distance between the centers of beads α and β , respectively. We choose to set $a = a_{\text{B}}$ for this work.

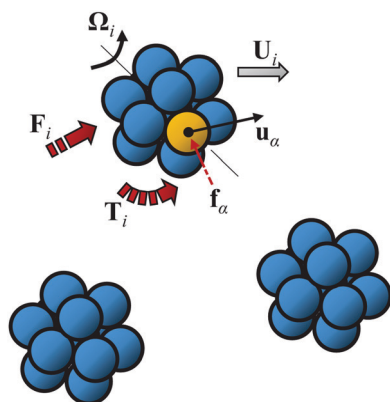


Fig. 3 BD simulation with hydrodynamic interactions for rigid composite-bead particles. Beads making up a particle are constrained to a rigidly moving manifold, and forces on each bead are transformed into the total force and torque acting the composite-bead particle.

Other choices yield similar hydrodynamic interactions among the particles.

The force on any bead $\mathbf{f}_\alpha = \mathbf{f}_\alpha^{\text{E}} + \mathbf{f}_\alpha^{\text{C}}$ can be divided into an external force $\mathbf{f}_\alpha^{\text{E}}$, representing the force from other particles or external fields, and a constraint force $\mathbf{f}_\alpha^{\text{C}}$, which is the internal force necessary to enforce rigid body motion of the particles. The external force is known, but the constraint force cannot be directly expressed. However, there are relationships that $\mathbf{f}_\alpha^{\text{E}}$ and $\mathbf{f}_\alpha^{\text{C}}$ must satisfy: (i) the summation and total moment of $\mathbf{f}_\alpha^{\text{E}}$ on a composite-bead particle i must be equal to the total force and torque on that particle; and (ii) the summation and total moment of $\mathbf{f}_\alpha^{\text{C}}$ on any particle i must be zero. Therefore, bead force \mathbf{f}_α satisfies:

$$\begin{aligned} \mathbf{F}_i^{\text{B}} + \mathbf{F}_i^{\text{P}} &= \sum_{\alpha \in i} \mathbf{f}_\alpha, \\ \mathbf{T}_i^{\text{B}} + \mathbf{T}_i^{\text{P}} &= \sum_{\alpha \in i} \mathbf{H}_{i,\alpha} \cdot \mathbf{f}_\alpha, \end{aligned} \quad (8)$$

where $\mathbf{H}_{i,\alpha}$ is:

$$\mathbf{H}_{i,\alpha} = \begin{bmatrix} 0 & -(z_\alpha - z_i) & y_\alpha - y_i \\ z_\alpha - z_i & 0 & -(x_\alpha - x_i) \\ -(y_\alpha - y_i) & x_\alpha - x_i & 0 \end{bmatrix}, \quad (9)$$

with x_i, y_i, z_i the geometric center of particle i and $x_\alpha, y_\alpha, z_\alpha$ the position of bead α in particle i . Because the composite-bead particle moves rigidly, the velocity of any bead \mathbf{u}_α can be calculated from the velocity and angular velocity of the rigid composite:

$$\mathbf{u}_\alpha = \mathbf{U}_i + \mathbf{H}_{i,\alpha}^{\text{T}} \cdot \boldsymbol{\Omega}_i, \quad (10)$$

where \mathbf{U}_i and $\boldsymbol{\Omega}_i$ are the translational and angular velocity of particle i , respectively. Because the transformations in (8) and (10) are linear, we can define a grand matrix $\boldsymbol{\Sigma}$ that correlates force, torque, translational velocity, and angular velocity on all composite-bead particles to bead forces \mathbf{f} and velocities \mathbf{u} :

$$\begin{bmatrix} \mathbf{F}^{\text{B}} \\ \mathbf{T}^{\text{B}} \end{bmatrix} + \begin{bmatrix} \mathbf{F}^{\text{P}} \\ \mathbf{T}^{\text{P}} \end{bmatrix} = \boldsymbol{\Sigma} \cdot \mathbf{f} \quad \text{and} \quad \mathbf{u} = \boldsymbol{\Sigma}^{\text{T}} \cdot \begin{bmatrix} \mathbf{U} \\ \boldsymbol{\Omega} \end{bmatrix}. \quad (11)$$

If we combine eqn (6) and (11), a resistance relation between the total force/torque and translational/angular velocity of hydrodynamically interacting, rigid composite-bead particles can be obtained:

$$\begin{bmatrix} \mathbf{F}^{\text{B}} \\ \mathbf{T}^{\text{B}} \end{bmatrix} + \begin{bmatrix} \mathbf{F}^{\text{P}} \\ \mathbf{T}^{\text{P}} \end{bmatrix} = \boldsymbol{\Sigma} \cdot \mathcal{M}^{-1} \cdot \boldsymbol{\Sigma}^{\text{T}} \cdot \begin{bmatrix} \mathbf{U} \\ \boldsymbol{\Omega} \end{bmatrix}, \quad (12)$$

with the matrix product $\boldsymbol{\Sigma} \cdot \mathcal{M}^{-1} \cdot \boldsymbol{\Sigma}^{\text{T}}$ acting as the resistance tensor in (5).

According to the fluctuation–dissipation theorem, the mean value and the autocorrelation function of the Brownian stochastic force on the beads \mathbf{f}^{B} can be characterized by:⁴⁶

$$\begin{aligned} \langle \mathbf{f}^{\text{B}} \rangle &= 0, \\ \langle \mathbf{f}^{\text{B}}(0) \mathbf{f}^{\text{B}}(t) \rangle &= 2k_{\text{B}} T \cdot \mathcal{M}^{-1} \delta(t), \end{aligned} \quad (13)$$



where $\delta(t)$ is the Dirac delta function. The stochastic force \mathbf{F}^B and torque \mathbf{T}^B for the entire composite-bead particle can be obtained by summing the \mathbf{f}^B on each bead in it.

An efficient implementation of this algorithm is described in further detail in the ESI,[†] as well as other works on immersed boundary simulations.^{52,53} It is important to note that when samples fully equilibrate, thermodynamic properties can be computed using other, simple simulation methods such as Markov-chain Monte Carlo, molecular dynamics, or freely-draining Brownian dynamics, which neglect the hydrodynamic interactions among the particles. For transport properties such as diffusion and viscoelasticity, the hydrodynamic interactions are critical to quantitative modeling.

2.2.2 Simulation of sheared dispersions and calculation of the stress. When a linear flow field is imposed on a dispersion, additional hydrodynamic forces \mathbf{F}_e^H , torques \mathbf{T}_e^H , and stresslets \mathbf{S}_e^H of composite-bead particles can be expressed as:

$$\begin{aligned}\mathbf{F}_e^H &= \mathbf{R}_{FE} \cdot \mathbf{e}, \\ \mathbf{T}_e^H &= \mathbf{R}_{TE} \cdot \mathbf{e}, \\ \mathbf{S}_e^H &= \mathbf{R}_{SE} \cdot \mathbf{e},\end{aligned}\quad (14)$$

where \mathbf{R}_{FE} , \mathbf{R}_{TE} , and \mathbf{R}_{SE} are resistance tensors representing force, torque, and stresslet couplings with strain field, and \mathbf{e} is the imposed rate-of-strain tensor. The force and torque on the composite-bead particles give rise to translational and rotational motion, as shown in (4), so the extra linear velocity and angular velocity caused by strain rate \mathbf{e} are:

$$\begin{bmatrix} \mathbf{U}_e \\ \mathbf{\Omega}_e \end{bmatrix} = \begin{bmatrix} \mathbf{R}_{FU} & \mathbf{R}_{F\Omega} \\ \mathbf{R}_{TU} & \mathbf{R}_{T\Omega} \end{bmatrix}^{-1} \cdot \begin{bmatrix} \mathbf{R}_{FE} \\ \mathbf{R}_{TE} \end{bmatrix} \cdot \mathbf{e} + \begin{bmatrix} \mathbf{U}_\infty \\ \mathbf{\Omega}_\infty \end{bmatrix}, \quad (15)$$

where \mathbf{U}_∞ and $\mathbf{\Omega}_\infty$ represent the linear velocity and angular velocity of the imposed flow. For a simple shear flow with velocity gradient $\dot{\gamma}_{xy}$,

$$\mathbf{U}_\infty = \begin{bmatrix} \dot{\gamma}_{xy}y \\ 0 \\ 0 \end{bmatrix} \quad \text{and} \quad \mathbf{\Omega}_\infty = \begin{bmatrix} 0 \\ 0 \\ -\dot{\gamma}_{xy}/2 \end{bmatrix}, \quad (16)$$

and the symmetric rate-of-strain tensor is:

$$\mathbf{e} = \begin{bmatrix} 0 & \dot{\gamma}_{xy}/2 & 0 \\ \dot{\gamma}_{xy}/2 & 0 & 0 \\ 0 & 0 & 0 \end{bmatrix}. \quad (17)$$

Combining the velocity/angular velocity in (5) and (15) gives the total velocity of hydrodynamically interacting composite-bead particles.

The bulk stress, $\boldsymbol{\sigma}$ in a dispersion deformed at rate of strain \mathbf{e} is given by:⁴⁸

$$\boldsymbol{\sigma} = -p_f \mathbf{I} + 2\eta_s \mathbf{e} - nk_B T \mathbf{I} + \boldsymbol{\sigma}^H + \boldsymbol{\sigma}^P + \boldsymbol{\sigma}^B, \quad (18)$$

where p_f is the pressure of solvent, $2\eta_s \mathbf{e}$ is the contribution from the bulk solvent under shear, and $-nk_B T \mathbf{I}$ is the isotropic

pressure contributed by thermal energy. The last three terms are the particle phase contributions to the stress: $\boldsymbol{\sigma}^H$ is the hydrodynamic contribution; $\boldsymbol{\sigma}^P$ is the contribution from deterministic inter-particle interactions; and $\boldsymbol{\sigma}^B$ is the contribution arising from Brownian relaxation of the dispersion.

In the dilute limit, the hydrodynamic contribution is given by the familiar Einstein's expression $\boldsymbol{\sigma}^H = 5\eta_s \phi \mathbf{e}$. However, in solutions with higher concentrations, the stress coupling with particle motion and strain field gives higher order terms and cannot be simplified by this expression. In addition to the stresslet \mathbf{S}_e^H contribution, the relative motion to the fluid also contributes to the hydrodynamic stress $\boldsymbol{\sigma}^H$ linearly through the stresslet coupling resistance tensors \mathbf{R}_{SU} and $\mathbf{R}_{S\Omega}$. The total stress contributed hydrodynamically is:⁴⁹

$$\boldsymbol{\sigma}^H = -\frac{1}{V} \sum_{ij} \left(\begin{bmatrix} \mathbf{R}_{SU} & \mathbf{R}_{S\Omega} \end{bmatrix} \cdot \begin{bmatrix} \mathbf{R}_{FU} & \mathbf{R}_{F\Omega} \\ \mathbf{R}_{TU} & \mathbf{R}_{T\Omega} \end{bmatrix}^{-1} \cdot \begin{bmatrix} \mathbf{R}_{FE} \\ \mathbf{R}_{TE} \end{bmatrix} - \mathbf{R}_{SE} \right)_{ij} \cdot \mathbf{e}, \quad (19)$$

where V is the volume of the simulation box, and the summation is performed on all pairs of N composite-particles, and the subscript ij refers to the i th- j th block of the operator.

The 5 new resistance tensors \mathbf{R}_{FE} , \mathbf{R}_{TE} , \mathbf{R}_{SE} , \mathbf{R}_{SU} , and $\mathbf{R}_{S\Omega}$ can be evaluated using the same approach to derive (12). For a rigid particle i consisting of beads α , its stresslet \mathbf{S} is:

$$\mathbf{S}_i = \frac{1}{2} \sum_{\alpha \in i} \left[(\mathbf{x}_\alpha - \mathbf{x}_i) \mathbf{f}_\alpha^T + \mathbf{f}_\alpha (\mathbf{x}_\alpha - \mathbf{x}_i)^T \right]. \quad (20)$$

Similarly, an extra term for affine motion must be added to the velocity expression of each bead (10) in an imposed linear flow with strain rate tensor \mathbf{e} :

$$\mathbf{u}_\alpha = \mathbf{U}_i + \mathbf{H}_{i,\alpha}^T \cdot \mathbf{\Omega}_i + \mathbf{e} \cdot (\mathbf{x}_\alpha - \mathbf{x}_i). \quad (21)$$

Both of these relations are linear, so we can use tensors \mathbf{K} and \mathbf{K}' to express compactly relations (20) and (21) for all the particles:

$$\mathbf{S} = \mathbf{K} \cdot \mathbf{f} \quad \text{and} \quad \mathbf{u} = \boldsymbol{\Sigma}^T \cdot \begin{bmatrix} \mathbf{U} \\ \mathbf{\Omega} \end{bmatrix} + \mathbf{K}' \cdot \mathbf{e}, \quad (22)$$

and obtain the resistance relations:

$$\begin{aligned} \begin{bmatrix} \mathbf{R}_{FE} \\ \mathbf{R}_{TE} \end{bmatrix} &= \boldsymbol{\Sigma} \cdot \mathcal{M}^{-1} \cdot \mathbf{K}', \\ \mathbf{R}_{SE} &= \mathbf{K} \cdot \mathcal{M}^{-1} \cdot \mathbf{K}', \\ \begin{bmatrix} \mathbf{R}_{SU} & \mathbf{R}_{S\Omega} \end{bmatrix} &= \mathbf{K} \cdot \mathcal{M}^{-1} \cdot \boldsymbol{\Sigma}^T. \end{aligned} \quad (23)$$

The contribution from deterministic inter-particle interactions $\boldsymbol{\sigma}^P$ can be expressed as:⁴⁹

$$\boldsymbol{\sigma}^P = -\frac{1}{V} \sum_i \left(\mathbf{x} \mathbf{F}^P + \begin{bmatrix} \mathbf{R}_{SU} & \mathbf{R}_{S\Omega} \end{bmatrix} \cdot \begin{bmatrix} \mathbf{R}_{FU} & \mathbf{R}_{F\Omega} \\ \mathbf{R}_{TU} & \mathbf{R}_{T\Omega} \end{bmatrix}^{-1} \cdot \begin{bmatrix} \mathbf{F}^P \\ \mathbf{T}^P \end{bmatrix} \right)_i, \quad (24)$$

where the first term is the virial contribution, and the second term is the hydrodynamic stress due to relative motion of particles in a



fluid driven by inter-particle force and torque. The subscript i here refers to i th component or diagonal block of the operator.

The Brownian contribution to the stress σ^B is:⁴⁹

$$\sigma^B = -\frac{k_B T}{V} \sum_i \left[\nabla \cdot \left([\mathbf{R}_{SU} \quad \mathbf{R}_{S\Omega}] \cdot \begin{bmatrix} \mathbf{R}_{FU} & \mathbf{R}_{F\Omega} \\ \mathbf{R}_{TU} & \mathbf{R}_{T\Omega} \end{bmatrix}^{-1} \right) \right]_i, \quad (25)$$

which can either be evaluated directly⁵⁰ or estimated by various random finite difference schemes.⁵¹ An efficient numerical scheme of this algorithm is described in detail in the ESI.†

A high-performance algorithm for dynamic simulation of rigid composite-bead particles has been developed that combines a newly developed integration scheme^{52,53} with the computation of bead mobility tensor is powered by a fast algorithm for evaluating the RPY mobility, named the Positively Split Ewald (PSE) method.⁵⁴ The implementation is built as a plugin to HOOMD-blue,^{55,56} a molecular dynamics suite optimized for massively parallel processing on GPUs, and can be downloaded from a publicly accessible domain: <http://web.mit.edu/swangroup/software.shtml>.

2.3 Ramped-frequency sweep method

To obtain linear viscoelastic properties $G'(\omega)$ and $G''(\omega)$ under small amplitude oscillatory shear (SAOS), a ramped-frequency sweep (“chirp”) method^{57,58} is used to investigate the rheology at a range of frequencies with a single simulation run. A wide bandwidth oscillatory strain γ_{xy} signal is imposed:

$$\gamma_{xy} = \gamma_{\max} \sin \left[\frac{t_f \omega_0}{\ln(\omega_f/\omega_0)} \left[\left(\frac{\omega_f}{\omega_0} \right)^{t/t_f} - 1 \right] \right], \quad (26)$$

where γ_{\max} is the maximum strain amplitude, t_f is the total duration of the sweep, and ω_0 and ω_f are the lowest and highest pulsation of the sweep controlling the range of probed frequencies. From the shear stress response $\sigma_{xy}(t)$, the complex modulus can be calculated from:

$$G^*(\omega) = \frac{\tilde{\sigma}_{xy}(\omega)}{\tilde{\gamma}_{xy}(\omega)}, \quad (27)$$

where $\tilde{\sigma}_{xy}(\omega)$ and $\tilde{\gamma}_{xy}(\omega)$ are Fourier transform of shear stress and strain, respectively.

Abrupt start-up and cessation of the shear rate often introduce noise into the Fourier transform. To resolve this issue, we apply the Optimally Windowed Chirp (OWCh)⁵⁹ method using a Tukey window function w to modulate the strain ($\gamma_{xy}^w = \gamma_{xy} w$) and smooth the initial and final stages of the oscillatory shear:

$$w = \begin{cases} \frac{1}{2} + \frac{1}{2} \cos \left(\frac{2\pi}{\delta} \left[\frac{t}{t_f} - \frac{\delta}{2} \right] \right) & \text{if } \frac{t}{t_f} \in \left[0, \frac{\delta}{2} \right], \\ 1 & \text{if } \frac{t}{t_f} \in \left(\frac{\delta}{2}, 1 - \frac{\delta}{2} \right), \\ \frac{1}{2} + \frac{1}{2} \cos \left(\frac{2\pi}{\delta} \left[\frac{t}{t_f} - 1 + \frac{\delta}{2} \right] \right) & \text{if } \frac{t}{t_f} \in \left(1 - \frac{\delta}{2}, 1 \right], \end{cases} \quad (28)$$

where δ is the duration of the smoothing window, and is set to 0.1 through this work. The maximum strain amplitude γ_{\max}

is set to 3% so that the suspension is within the linear viscoelasticity regime but the signal to noise ratio is high.

2.4 Adhesive hard sphere model

Baxter's adhesive hard sphere (AHS) model is the most widely used theory describing colloidal dispersions with short-ranged attractions⁶⁰ and serves as the standard reference for comparison of experiments and simulations with different types of particles. In this model, the range of attraction potential is assumed to be infinitesimally small, and the attraction strength is characterized by a single parameter, the Baxter temperature τ , which can be obtained from the second virial coefficient B_{22} :⁶⁴

$$\tau = \frac{1}{4(1 - B_{22}/B_{22}^{\text{HS}})}, \quad (29)$$

where B_{22}^{HS} is the second virial coefficient of the equivalent hard-sphere. B_{22} measures the mean attraction between two particles and is the dominant contribution to thermodynamic properties in the dilute limit. A lower Baxter temperature τ indicates a stronger mean attraction. The phase diagram for AHSs in terms of volume fraction, ϕ , and Baxter temperature, τ , is known from previous works^{61–63} and is shown in Fig. 4.

To compare suspensions of patchy particles and isotropic particles, we compute Baxter temperature τ as a measurement of the averaged interaction strength between particles. This also allows us to directly compare our patchy particles to Baxter's AHS model. The Baxter temperature can be computed from the second virial coefficient B_{22} by (29). For particles with isotropic interactions, the pair potential, E , is only a function of distance r , so B_{22} is obtained by a simple integration:

$$B_{22} = -2\pi \int_0^\infty [\exp(-E/k_B T) - 1] r^2 dr. \quad (30)$$

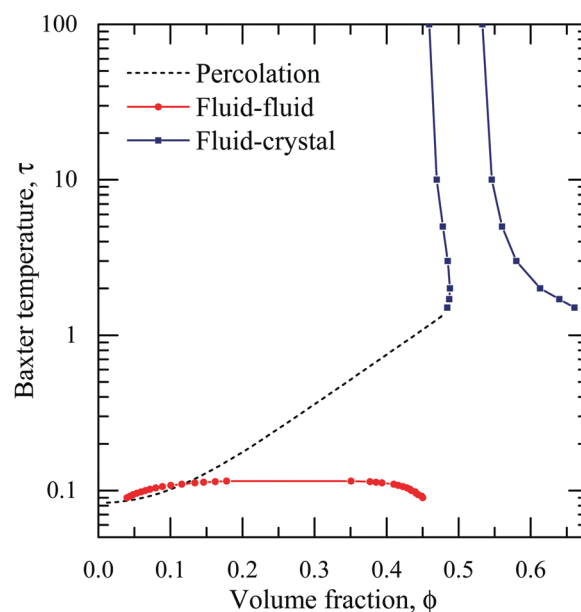


Fig. 4 Phase diagram for Baxter's adhesive hard sphere (AHS) model.^{60–63}



For the patchy particles, however, the pair potential, E , is a function of both particles' orientations as well, so the integration must be over the 6 degrees of freedom of the pair of particles:

$$B_{22} = -\frac{1}{16\pi^2} \int_0^\infty r^2 dr \int_0^\pi \sin \theta d\theta \int_0^{2\pi} d\phi \quad (31)$$

$$\times \int_{-\pi/2}^{\pi/2} \cos \beta d\beta \int_0^{2\pi} d\alpha \int_0^{2\pi} [\exp(-E/k_B T) - 1] d\gamma,$$

where r , θ , and ϕ are the distance, polar and azimuthal angles of the second particle relative to the first particle, and α , β , and γ are the Euler angles of the second particle relative the orientation of the first particle. For any of the patchy particles, this integral is evaluated using umbrella sampling Monte Carlo integration and B_{22} is used to determine the equivalent Baxter temperature.

3 Results and discussion

Brownian dynamics simulations with hydrodynamic interactions have been performed on suspensions of random patchy particles with different βJ value from -0.3 to 1.0 (unconstrained and balanced), isotropically attractive particles (particles composed of the same type of attractive sites), and Janus particles (particles with two types of sites occupying the two hemispheres) within a volume fraction range of 0.10 to 0.30 . For each βJ value, 5 unconstrained and 5 balanced random patchiness patterns are generated from the Monte Carlo method, and separate simulations are performed on single particle types. 500 identical composite-bead particles are included in the simulation box, and each particle is composed of 42 beads.

For each different Baxter temperature τ , a random hard-sphere configuration of composite-bead particles is generated first, and the suspension is quenched to the corresponding target interaction strength immediately. Solution micro-structure, percolation transition, thermodynamics, and rheology are then investigated. It should be noted that the volume fraction used in this work is based on the hydrodynamic radius of composite-bead particles. The hydrodynamic radius is found by computing the translational drag coefficient of a single composite bead particle and dividing by $6\pi\eta_s$. As shown in the ESI,[†] the equivalent volume fraction obtained by fitting the osmotic pressure to the equation of state for hard-sphere suspensions is approximately 20% larger. This represents a 6% difference between the hydrodynamic radius and the "thermodynamic" radius of the particles. Any micro-structural models depending on the hard sphere volume fraction are appropriately scaled to represent the model in terms of the hydrodynamic volume fraction.

3.1 Micro-structure and percolation transition

At high Baxter temperature or low interaction strength, Brownian motion dominates over the site-site interactions, and the micro-structure of any particle type is similar to that of hard-sphere suspensions. As the Baxter temperature decreases, both isotropic particles and patchy particles form bonds due to the attractive Yukawa potential and aggregate to form clusters. However, the structure of the clusters formed from isotropic particles is

different from the structure of the clusters formed from patchy particles. Because of the anisotropic inter-particle interactions due to heterogeneity, the energy profile between a pair of approaching patchy particles is rough and has strong angular dependency. The roughness of the energy profile creates a barrier preventing rotations that would allow a patchy particle pair traverse between states with local energy minima. This anisotropic effect limits the number of potential configurations accessible when anisotropic particles approach each other, and arrests them from further relaxation. Isotropic particles, on the other hand, do not have the same rough energy landscape and rotate freely within clusters.

Simulation snapshots of different types of particles quenched well below their percolation transition ($\tau = 0.05$) at $\phi = 0.30$ are shown in Fig. 5. All of the suspensions percolate, but the micro-structure is different despite having the same Baxter temperature. Because isotropically attractive particles can form bonds with each other from any orientation, and neighbors can freely rotate among different configurations, the isotropic particles form compact clusters. Patchy particles form more clusters because only limited orientations are allowed for bonding, and the constraint due to heterogeneity creates a barrier to further structural relaxation. Close packed clusters are seldom observed in patchy particle solutions.

Fig. 6 shows the average number of bonds (or number of nearest neighbors) of each particle ($\langle N_b \rangle$) at $\phi = 0.30$ plotted against the Baxter temperature τ . The number of bonds for all types of particle is approximately the same at τ values larger than 10. However, for τ below 0.01, isotropically attractive particles have more than 6 nearest neighbors, indicating the formation of closely packed clusters, while all the patchy particles have about 4 nearest neighbors regardless of patchiness

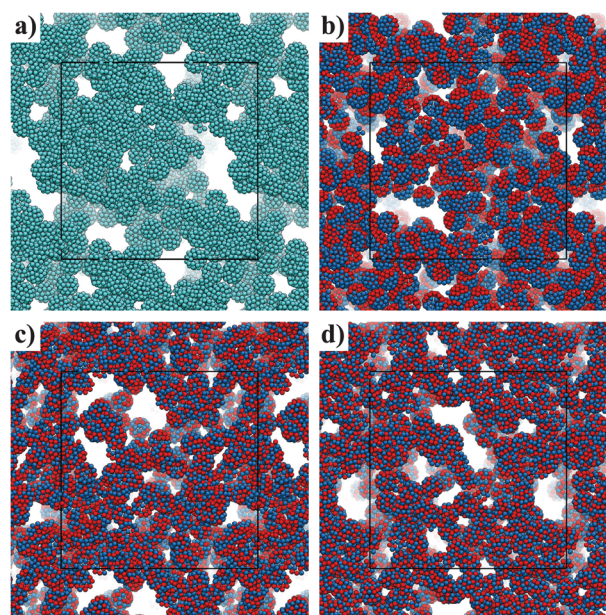


Fig. 5 Simulation snapshots of various types of particles at volume fraction $\phi = 0.30$ and Baxter temperature $\tau = 0.05$: (a) isotropically attractive particles, (b) Janus particles, (c) random patchy spheres with $\beta J = -0.2$, (d) random patchy spheres with $\beta J = 1$.



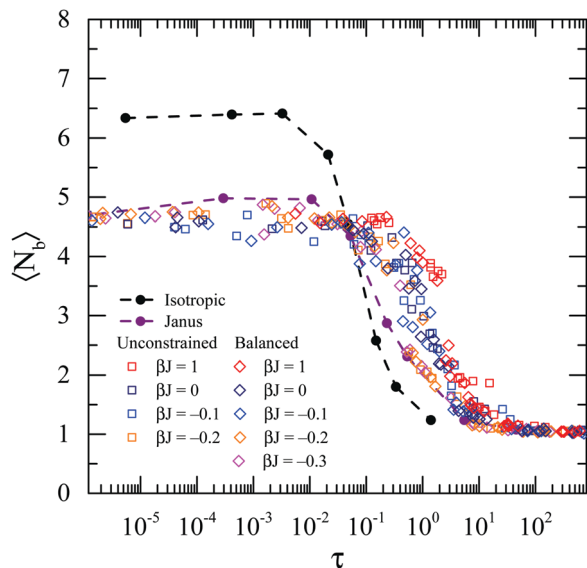


Fig. 6 Average number of bonds for a particle, $\langle N_b \rangle$, as a function of Baxter temperature, τ , at volume fraction $\phi = 0.30$. Different symbols represent different types of particles, as shown in the legend. The 5 balanced/unconstrained patchiness patterns sampled at the same βJ value are not distinguished in symbol (the same convention is applied for the following figures).

type within the same τ regime. The number of bonds in suspensions of patchy particles with finer patches (higher βJ values) starts to increase at a higher τ value than particles with coarser patches (lower βJ values), Janus particles, and isotropically attractive particles. This indicates that particles with the same second virial coefficient exhibit different structures at higher concentrations depending on the surface patchiness. With finer patchiness, patchy particles form clusters at higher τ values where isotropic particles are still well dispersed. The number of bonds is determined by identifying all particle pairs within a cut-off set by the range of the inter-bead interaction, which decays exponentially. Alternative nearest neighbor schemes varying the cut-off by fifteen percent or using the first nearest neighbor shell in the radial distribution function gave quantitatively indistinguishable results.

Because the number of nearest neighbors N_b is an indicator of percolation,^{61,65} the shift of the $\langle N_b \rangle$ - τ curves implies that surface heterogeneity will shift the percolation transition. Typically, the percolation transition is defined as the appearance of an infinitely spanning network in the suspension.⁶⁶ Our simulations define a suspension to be percolated when more than 50% of particle configurations sampled over time have a cluster that connects with its periodic images in all three directions of the simulation box. The percolation transition for different types of particles with volume fractions from 0.10 to 0.30 is shown in Fig. 7. The theoretical prediction from Percus–Yevick theory for adhesive hard-spheres⁶¹ is also included in the plot as a reference. For every particle type, the percolation transition τ increases with volume fraction ϕ , which qualitatively agrees with the theoretical prediction. With finer patchiness, the suspension percolates at higher τ values, and the percolation boundary moves away from the theoretical prediction. For example,

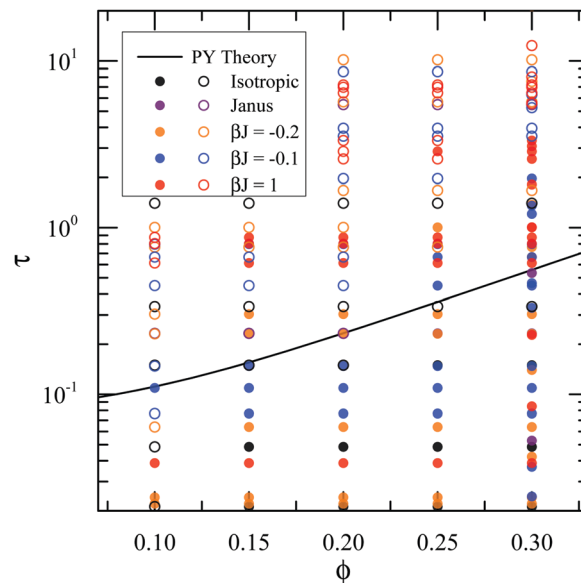


Fig. 7 State diagram showing the percolation transition, τ , of different types of particles as a function of volume fraction. Closed circles represent percolated suspensions, and open circles represent un-percolated suspensions. The solid line is the percolation line predicted from Percus–Yevick theory.⁶¹

the finest patchy particles with the highest $\beta J = 1$ percolate at around $\tau = 4$ with $\phi = 0.30$, while the theory predicts the percolation Baxter temperature to be approximately $\tau = 0.6$.

For isotropic particles with short-ranged attractions in the dilute limit, the Baxter temperature is the only determining factor for structure and percolation. For patchy particles, even at low concentrations, there are differences arising from many-body effects. The highly localized attractive interactions of patchy particles bind them together and form a network, which prevents the particles from exploring all phase space. This rough energy landscape explains the variation in averaged number of bonds $\langle N_b \rangle$ and Baxter temperature τ beyond the percolation transition.

3.2 Osmotic pressure

Surface heterogeneity greatly affects the thermodynamic properties of the dispersions. Fig. 8 shows the osmotic pressure P , normalized by nk_bT where n is the particle number density, plotted as a function of Baxter temperature τ . As τ decreases, the attraction between particles reduces the osmotic pressure of the suspension. The theoretical prediction from Percus–Yevick theory by Baxter⁶⁰ is also included to compare with our simulation results.

The isotropically attractive particle suspensions agree with the theoretically predicted osmotic pressure. As the surface patchiness becomes finer, however, the pressure deviates from the theory. At the same τ value, suspensions of particles with finer patches have lower osmotic pressure than those of particles with coarser patches. This is another example where the Baxter temperature, or averaged pair interaction strength cannot give a successful prediction of the thermodynamic properties of heterogeneous particles.



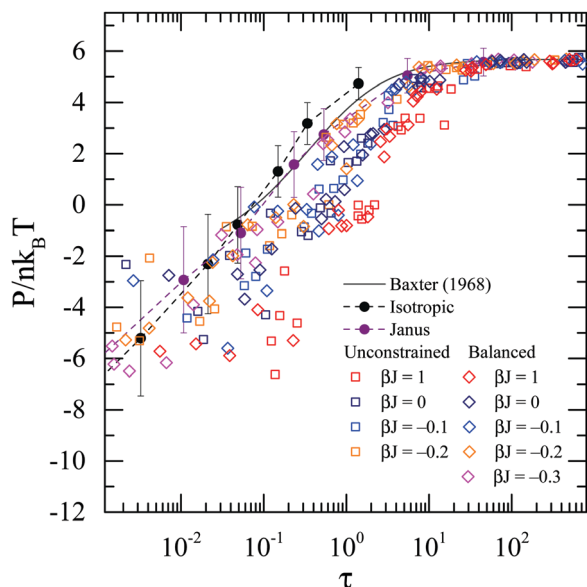


Fig. 8 Osmotic pressure P normalized by $nk_B T$, where n is the particle number density, as a function of Baxter temperature τ . The prediction from Percus–Yevick theory by Baxter⁶⁰ is shown as the black curve.

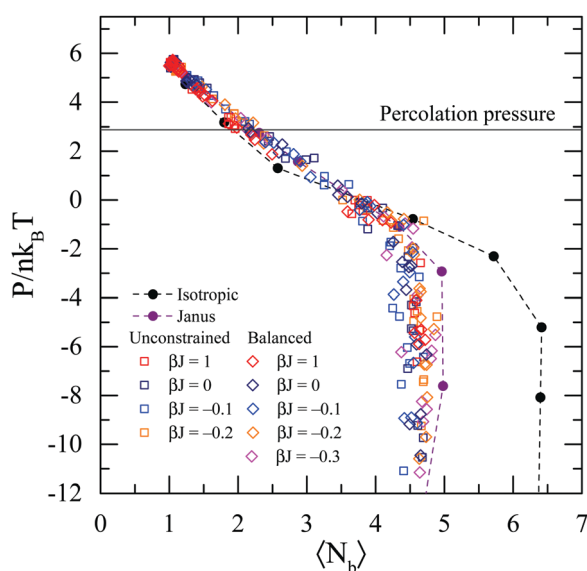


Fig. 9 Normalized osmotic pressure $P/nk_B T$ as a function of average number of bonds $\langle N_b \rangle$. The solid line is the theoretical percolation osmotic pressure using Percus–Yevick theory.

Osmotic pressure P is plotted as a function of averaged number of nearest neighbors $\langle N_b \rangle$ as shown in Fig. 9. The data for all types of particles collapse into a single master curve, which seems to indicate there exists a common, coarse-grained thermodynamic model from which to analyze the structure and percolation of diverse sets of patchy particles. In Fig. 9, the osmotic pressure decreases with increasing attraction strength, while the number of bonds increases. The only perceived difference between particle types in the P – $\langle N_b \rangle$ curves occurs at very negative osmotic pressures, where isotropic particles

have six bonds on average while patchy particles have four. These bond numbers agree with the isostatic contact numbers of frictionless spheres ($z_{\text{iso}} = 6$) and perfect frictional spheres ($z_{\text{iso}} = 4$), respectively. This indicates the constraint due to surface heterogeneity is similar to that from surface friction. Negative osmotic pressure indicates that the dispersion would undergo syneresis if it were not attached to the boundaries. Such a condition is common in dispersions of attractive particles.

Because of the universal correlation between micro-structure and osmotic pressure regardless of particle type, osmotic pressure P may be more useful than τ to describe the percolation transition. The percolation transition is shown on the ϕ – P plane instead of the ϕ – τ plane in Fig. 10(a). The solid curve is the percolation pressure predicted by Percus–Yevick theory, which is obtained by calculating the percolation transition τ value⁶¹ and using Baxter's equation of state⁶⁰ to compute the corresponding pressure. Unlike the ϕ – τ diagram in Fig. 7, the theoretical percolation curve clearly separates percolated and unpercolated samples for all types of particles. The only exception is at $\phi = 0.10$, where fluid–fluid phase separation competes with the percolation transition. The agreement between theory and simulation when adopting osmotic pressure P instead of Baxter temperature τ indicates that heterogeneous particles at the same pressure have the same micro-structure. This is in contrast to Fig. 5 where the influence of heterogeneity on packing structure at the same τ is evident. At the percolation transition point, there appears to be a universal critical mechanical load, or pressure, for any particle patchiness.

Fig. 10(b) shows a schematic of the percolation transition in P – τ – ϕ space for different patchy particles. For a given particle type, the osmotic pressure P is a single-valued function of volume fraction ϕ and Baxter temperature τ . Therefore, the equations of state for different particle types can be represented as different two-dimensional surfaces (“particle surfaces”) in P – τ – ϕ space. The percolation transition can also be represented by a surface. Because of the universal percolation pressure which is only a function of ϕ , this surface is perpendicular to the P – ϕ plane. The intersecting curve between a particle surface and the percolation surface can be used to compute the percolation transition as a function of τ for that particle type. Because different particle surfaces intersect the percolation surface in different places, the projections of the intersections into the τ – ϕ plane are different. Therefore, heterogeneity strongly affects the values of τ at which percolation occurs, as shown in Fig. 7. Finally, the osmotic pressure and number of nearest neighbors at lower volume fractions are depicted in the ESI† and are consistent with the observations at $\phi = 0.3$. This collapse in terms of the osmotic pressure is reminiscent of the use of the compressibility factor to collapse dynamics in hard sphere systems.⁶⁷

3.3 Gelation

Multiple definitions of gelation are commonly used in literature. In this work, gelation is defined as described in the seminal work by Chambon and Winter (1987).²⁶ A colloidal dispersion is at the critical gel point if the tangent of the loss angle, $\tan \delta$, is



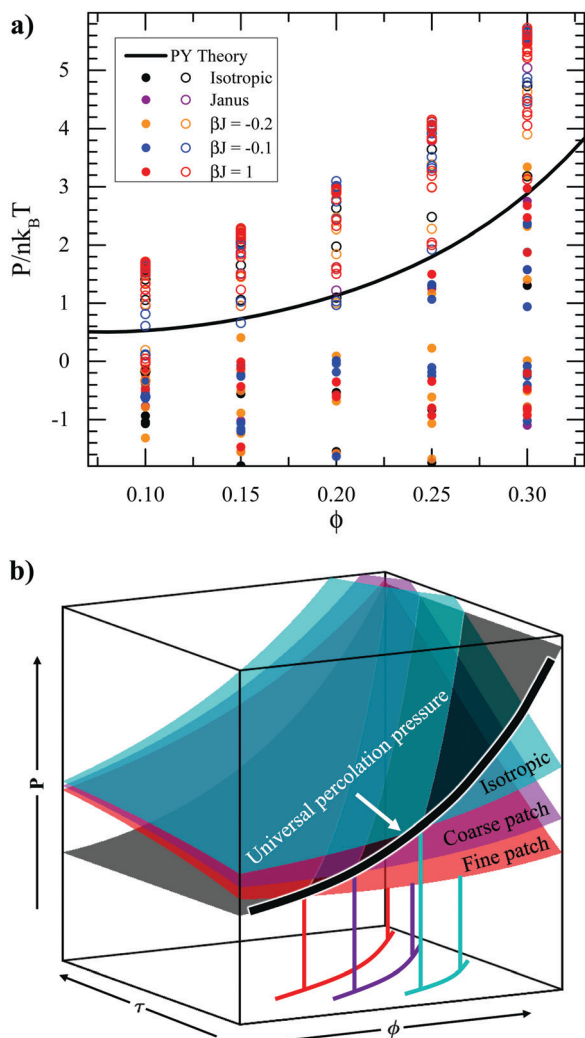


Fig. 10 Osmotic pressure, determines the percolation transition regardless of surface heterogeneity. (a) Phase diagram in the pressure–volume fraction (P – ϕ) plane, where closed circles represent percolated suspensions, and open circles represent unpercolated suspensions. The percolation transition has a universal osmotic pressure regardless of particle surface heterogeneity that agrees with Percus–Yevick theory. (b) P – τ – ϕ surfaces for different particle types. The percolation transition can also be represented by a surface perpendicular to the P – ϕ plane, because of the universal percolation pressure. The projections of the intersections between equation of state surfaces and the percolation surface into the τ – ϕ plane are the different percolation τ values as shown in Fig. 7.

independent of the oscillation angular frequency ω :

$$\tan \delta(\omega) = \frac{G''(\omega)}{G'(\omega)} = \tan(n\pi/2), \quad (32)$$

where n is called relaxation exponent and both $G'(\omega)$ and $G''(\omega)$ scale with ω^n . When the Baxter temperature τ is higher than the gel point value, G' is lower than G'' . The rheology of the dispersion is similar to that of a liquid, and the scaling with ω can be predicted from theory if in the hard-sphere limit.⁶⁸ When τ is lower than the gel point, G' is higher than G'' , and the dispersion behaves like a solid. For the composite-bead particle suspensions

in this work, at the critical gel point, the relaxation exponent n is approximately 0.5, and G' overlaps with G'' .²⁷

Fig. 11 shows G' and G'' of a suspension at $\phi = 0.30$ in simulations of SAOS with a ramped-frequency sweep (26). The complex modulus is normalized by the pressure scale of the system $k_B T/R_H^3$, and angular frequency is normalized by the diffusion time scale of the system $\tau^D = 6\pi\eta_s R_H^3/k_B T$. Linear viscoelastic responses of 4 types of particles are shown at Baxter temperatures above, below, and at the critical gel point.

The Baxter temperature τ at the critical gel point differs among suspensions with different particle types. Similar to observations of τ at the percolation transition, the gel point τ increases for finer patchiness. A similar trend with respect to the percolation transition volume fraction is also observed at the gel transition: τ increases with the volume fraction ϕ . The critical gel point for different particle types and volume fractions ϕ can be found in Table 1. Some patchy particles gel above the AHS percolation transition, τ , which again confirms that τ cannot be used to predict phase behavior of heterogeneous particle suspensions at moderate concentrations.

The solid black lines in Fig. 11 show the modulus:

$$G = \lambda \frac{k_B T}{R_H^3} \left(\frac{6\pi\eta_s R_H^3 \omega}{k_B T} \right)^{0.5}, \quad (33)$$

where the prefactor $\lambda \approx 1.0$ for all particle types at $\phi = 0.30$, which means the complex modulus G^* has the same value at the gel point even for heterogeneous particles. The complex modulus G^* at the gel point is only determined by two physical scales: the pressure scale $k_B T/R_H^3$ and diffusion time scale $\tau^D = 6\pi\eta_s R_H^3/k_B T$. We observe this universal linear viscoelastic behavior at the gel point for all volume fractions between 0.10 and 0.30 though the λ values decrease with decreasing ϕ : $\lambda(\phi = 0.20) \approx 0.4$ and $\lambda(\phi = 0.10) \approx 0.1$.

As shown for percolation, the osmotic pressure P instead of Baxter temperature τ is more useful for predicting the microstructure and thermodynamic properties for heterogeneous particle dispersions. The percolation transition pressure is independent of particle type, and it is interesting to ask whether the osmotic pressure is also predictive of the gel point for different particle types. After all, there is a universal viscoelastic scaling at this point. We performed equilibrium Brownian dynamics simulations at the critical gel point to obtain the osmotic pressures for suspensions of different particle types. Fig. 12 shows the osmotic pressure at the critical gel point with volume fractions from 0.10 to 0.30 overlaid with the percolation transition data in the P – ϕ plane. Compared with the universal percolation transition pressure line, the osmotic pressure P at the critical gel point is not universal. The gel point with respect to pressure increases with patch fineness. Suspensions of particles with finer patchiness have higher osmotic pressure at the critical gel point, while suspensions of Janus particles and isotropically attractive particles only reach critical gel point at very low osmotic pressure, presumably after phase separating.



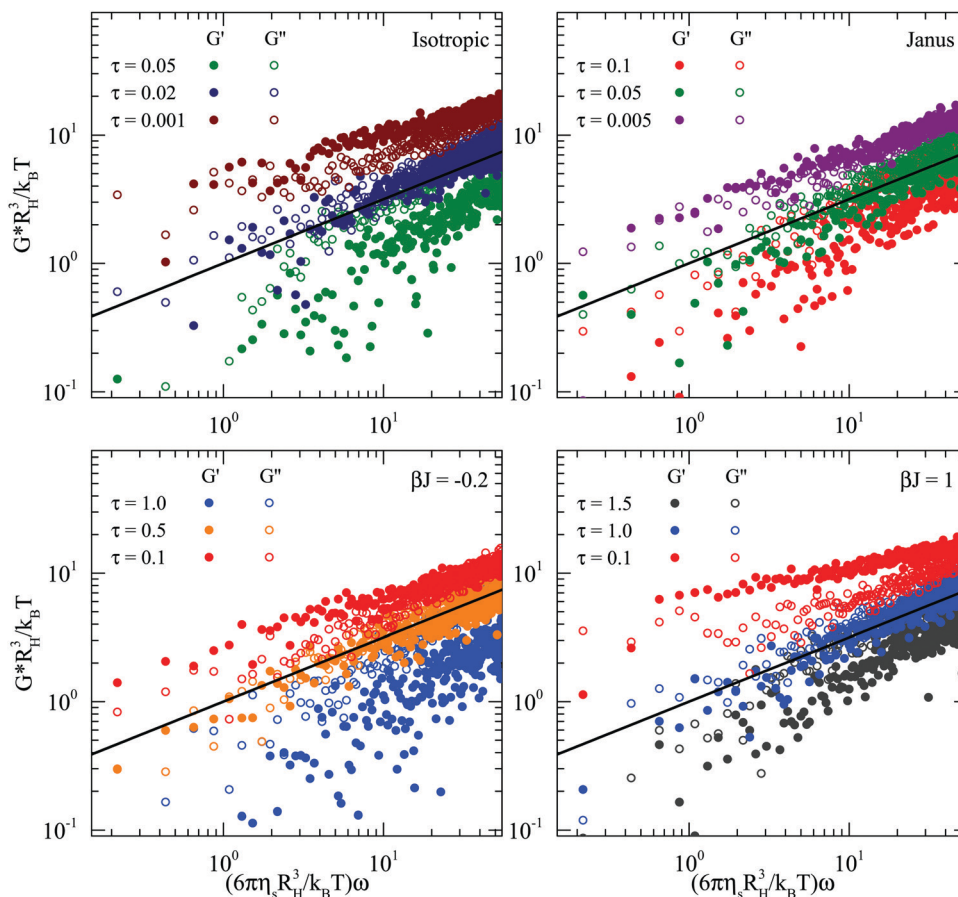


Fig. 11 Storage modulus $G'(\omega)$ and loss modulus $G''(\omega)$ for dispersions with different types of particles (shown in each subplot) at $\phi = 0.30$ and various Baxter temperatures τ above, below, and at the critical gel point. The solid lines are $\omega^{0.5}$ power law scalings: $G = \lambda \frac{k_B T}{R_H^3} \left(\frac{6\pi\eta_s R_H^3 \omega}{k_B T} \right)^{0.5}$, where the prefactor λ is approximately 1.0 in all of the four plots.

Table 1 Baxter temperature τ at the critical gel point for different types particle suspensions and volume fractions ϕ

Type	$\phi = 0.10$	$\phi = 0.20$	$\phi = 0.30$
Isotropic	0.01	0.02	0.02
Janus	0.02	0.03	0.05
$\beta J = -0.2$	0.1	0.3	0.5
$\beta J = 1$	0.3	0.8	1

Below the gel point, particles in the dispersion are dynamically arrested in a network from reaching the thermodynamic equilibrium state. Therefore, it is not surprising that almost all types of particles have an osmotic pressure below that of fluid–fluid phase separation by AHS model⁶⁰ (except random patchy spheres at $\phi = 0.30$), but without phase separating micro-structure. The different osmotic pressure for each type of particles partially explains the previous discrepancies in literature over the relationship among percolation, gelation, and phase separation.^{16,28,31,32}

The percolation transition and the gel point are controlled by osmotic pressure P and complex modulus G^* , respectively, which are mechanical properties with distinct physical origins.

The universality of one does not necessarily ensure universality of the other. The micro-structure and percolation are dependent on the normal mechanical load among particles, which is embodied in the osmotic pressure. Rheology, on the other hand, is also influenced by the bending moments and frictions between neighboring particles, which is still quite sensitive to surface heterogeneity. Therefore, in experiments we may expect to observe critical gelation anywhere from the percolation line to below the spinodal, due to varied surface characteristics of the dispersed particles.

Fig. 13 shows two dynamic properties, mean squared displacement (MSD) $\langle \Delta r^2 \rangle$ and intermediate scattering function (ISF) $F(q, \Delta t)$, for dispersions of different particle types at $\phi = 0.30$.

The mean squared displacement (MSD) is a measurement of the tracer diffusion rate of individual particles. Fig. 13(a) shows the MSD slightly above the percolation transition (“dispersed”) and at the critical gel point (“gel”). For dispersed particles, the MSD curves are identical for all types of particles in both the short-time and long-time limits. Without a percolated network structure, the freely moving particles have similar diffusive motion on all time scales, and the MSD is almost linear in



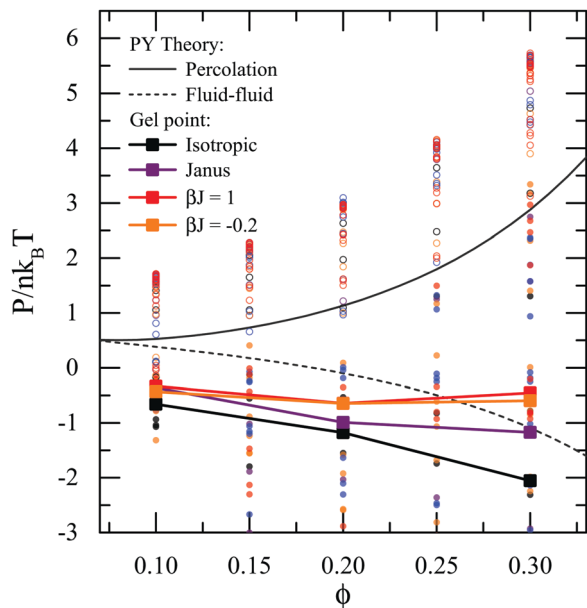


Fig. 12 State diagram in the pressure–volume fraction (P – ϕ) plane comparing percolation transition and gel point. In addition to the comparison of percolation transition between simulation and theory (as in Fig. 10), osmotic pressures at critical gel point (closed squares) of 4 types of particles are also included in the plot. The dashed line represents the osmotic pressure at fluid–fluid phase separation predicted by Baxter's AHS model.⁶⁰ Despite the universal percolation transition pressure, the corresponding pressure at the observed gel point differs due to the surface heterogeneity.

the whole range of lag time Δt . At the critical gel point, however, the MSD curves differ for different types of particles. In the short-time limit, the MSD of patchy particle suspensions are linear in time and lay on the same curve. The isotropic particle suspensions, on the other hand, forms compact clusters and thus has a slightly lower MSD than the patchy particles do. As the lag time increases, the MSD becomes sublinear with lag time and the curves of patchy particles also start to diverge. In the long-time limit, the MSD is linear again. The Janus particle suspension has the highest MSD value in the long lag time regime, while the random patchy particle with $\beta J = 1$ diffuses the slowest.

The intermediate scattering function characterizes the relaxation processes across many length scales, represented by the various wave vector q values. In Fig. 13(b), the ISF of particle suspensions at $\phi = 0.30$ is shown as a function of lag time Δt . Three q values are selected to probe the dynamics at various length scales. Like the MSD results, the decay of ISF is similar for all types of particles at short-time limit, but differs at longer times. The ISF of the isotropically attractive particle suspension cannot be directly compared with those of the patchy particle suspensions because the micro-structures are quite different. For the heterogeneous particles, the Janus particle suspension decays the most rapidly and thus has the fastest relaxation rate. The random patchy particle suspension with the finest patchiness ($\beta J = 1$) relaxation the slowest. This trend agrees with the MSD results.

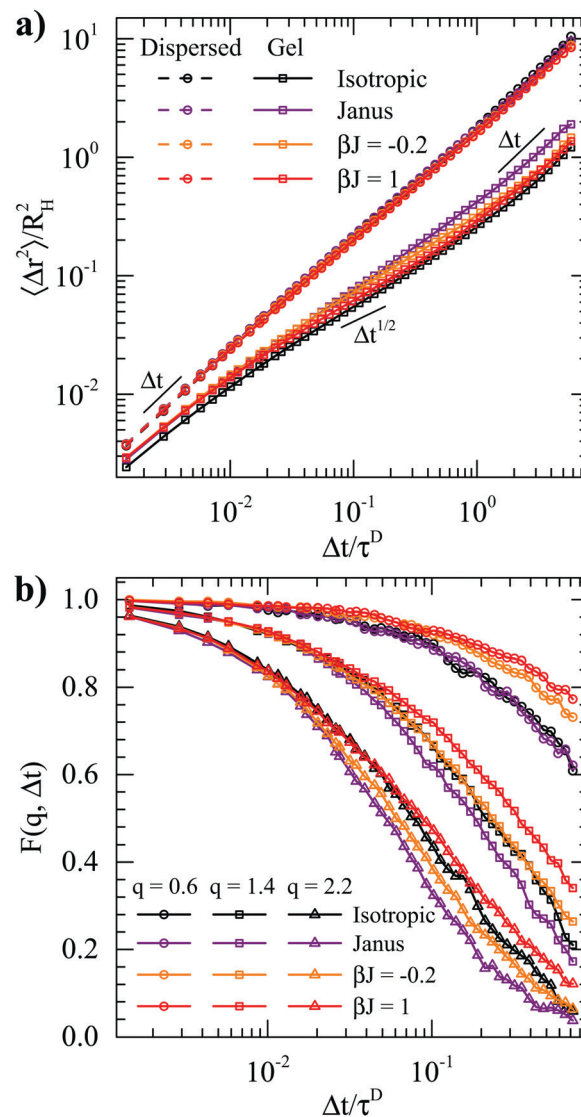


Fig. 13 Dynamics in dispersions with various particle types near the critical gel point for $\phi = 0.30$. (a) Mean squared displacement (MSD) as a function of lag time Δt of 4 types of particles at the critical gel point ("gel", squares) and slightly above the percolation transition point ("dispersed", circles). (b) Intermediate scattering function (ISF) $F(q, \Delta t)$ as a function of lag time at 3 different wave vectors q .

The different long-time dynamics among particles with different heterogeneity demonstrates the constraining effects imposed by patch–patch interactions. Although all the patchy particles form slowly relaxing clusters that give equal averaged bond number, they have different rates of relaxation due to different energy landscape roughness.

4 Conclusions

Using simulations of model dispersions of random patchy spheres, we have elucidated the differences between heterogeneous and isotropic particles. Isotropic particles with short-ranged attractions having the same second virial coefficient B_{22}



are thermodynamically indistinguishable in the dilute limit. We showed that particles having fine patches and the same second virial coefficient can exhibit significant deviations in micro-structure compared to isotropic models, leading to different percolation transition points at modest particle concentrations. Surface heterogeneity can prevent particles from forming compact clusters which shifts the percolation transition away from the theoretical prediction of Baxter's isotropic AHS model.

Baxter temperature τ , which is determined by the second virial coefficient B_{22} , is a measurement of averaged attraction strength between two particles. From measurement of the number of nearest neighbors and the osmotic pressure, we found that τ alone is not suitable for characterizing the state of a dispersion of heterogeneous particles. Different sized patches on the particle surface lead to different τ values at the percolation transition. Instead, the micro-structure is universally controlled by mechanics, and correlated strongly with the osmotic pressure. The percolation transition pressure is the same for all particle types and agrees with the theoretical prediction. In addition to its impact on thermodynamics, heterogeneity on particle surfaces impose extra constraints on relative rotations between neighboring particles, resulting in higher elastic modulus and slower long-time diffusion dynamics. This heterogeneity effect on dynamics and rheology of particles at their critical gel point could account for previous discrepancies in literature over the relationship among percolation, gelation, and phase separation.

To further understand the effect of heterogeneity on colloidal dispersions, future investigation should focus on developing equations of state for random patchy particles as a function of surface patch size. In addition to patchiness, surface roughness, friction, and steric effect due to anisotropic shapes can all constrain particle motion when particles approach each other.

Therefore, further quantitative understanding in the local interactions between particles at contact is required to understand the nature of gelation and the location of gel point. A quantitative description of the strength of rotational hindrance due to heterogeneity, friction, or anisotropic shape is necessary to predict the gel points of different colloidal systems based on individual particle properties. The typical bending moment could be a critical quantity to measure and characterize its contribution to complex modulus.⁶⁹ Lastly, the effects of anisotropic shape are not covered in the present work, but it impacts solution micro-structure and rheology. Understanding the properties of heterogeneous non-spherical particle suspensions, which are exemplified by protein drugs in the biopharmaceutical industry, is critical for their applications.

Conflicts of interest

There are no conflicts to declare.

Appendix: thermodynamics of patchy particles with inverted site–site interactions

Sites interact with each other *via* pairwise attractive or repulsive interactions depending on there types in our random patchy sphere model. Sites of the same type repel while sites of opposite types attract. In this section, we investigate another form of pairwise interactions: sites of the same type attract while sites of opposite types repel. Fig. 14 shows the comparison between the original model and this inverted model of the short-ranged interaction. As shown in Fig. 14(a), regardless of the two different interaction forms, the osmotic pressure only

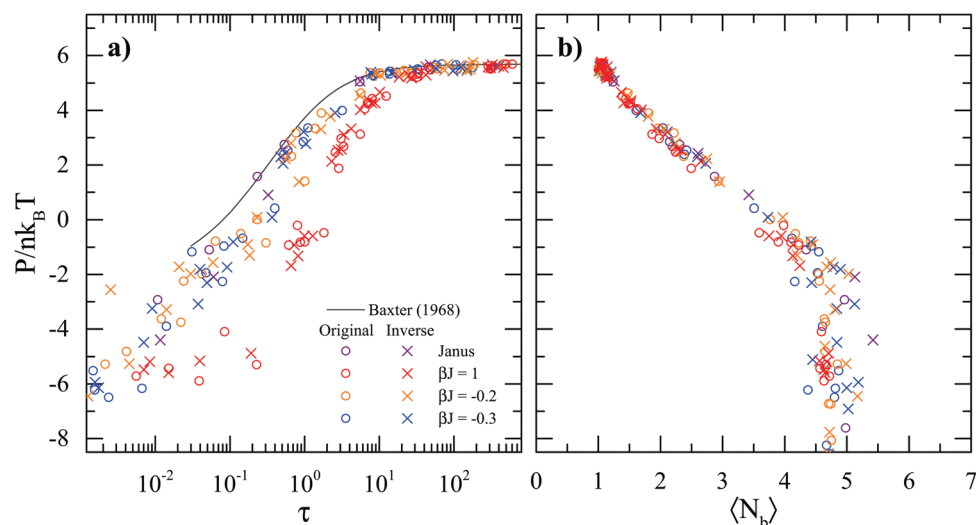


Fig. 14 Comparison of thermodynamic properties between the original interaction model (circles: sites of the same type repel while sites of opposite types attract) and the inverse model (crosses: sites of the same type attract while sites of opposite types repel). Without loss of generality, only balanced patchy spheres are used. (a) Osmotic pressure P normalized by $nk_B T$, where n is the particle number density, as a function of Baxter temperature τ . The prediction from Percus–Yevick theory by Baxter⁶⁰ is shown as the black curve. (b) Normalized osmotic pressure $P/nk_B T$ as a function of average number of bonds $\langle N_b \rangle$.



depends on τ value and βJ value. The original model and the inverse model with the same patch size share the same trend in the plot. Fig. 14(b) shows the osmotic pressure P as a function of averaged number of nearest neighbors $\langle N_b \rangle$. All types of particles collapse into a single curve. The universal correlation between micro-structure and osmotic pressure is not affected by either the patch size or the form of interactions. In conclusion, the thermodynamics of composite-bead patchy particle suspensions with attractive/repulsive site-site interactions is only determined by the patch size and not the site model.

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