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## Introduction

Biomolecular templates define the precise outcomes of enzyme-catalyzed reactions in essential biological processes, such as DNA replication, translation and transcription. Inspired by these, and other template-driven processes in biology, chemists have exploited templates to control the chemical synthesis of complex architectures, such as macrocycles,<sup>1</sup> catenanes,<sup>2</sup> cages<sup>3</sup> and knots.<sup>4</sup> Dynamic combinatorial chemistry (DCC), in particular, has emerged as a powerful tool to direct the molecular self-assembly of oligomers whereby monomers are linked together using reversible reactions, and template molecules bind to, stabilize, and amplify specific targeted oligomers.<sup>1,4,5</sup> To date, DCC has remained primarily in the realm of synthetic organic chemistry with only a few examples reported of the use of reversible enzyme-mediated reactions to generate relatively small dynamic systems.<sup>6</sup> We present here an enzyme-enabled dynamic mixture of interconverting cyclodextrins and show for the first time the use of a series of templates to selectively obtain different products from an enzyme-mediated dynamic

system (Fig. 1). We show how molecular recognition of artificial templates can be exploited to obtain alternative products to those selected by Nature.

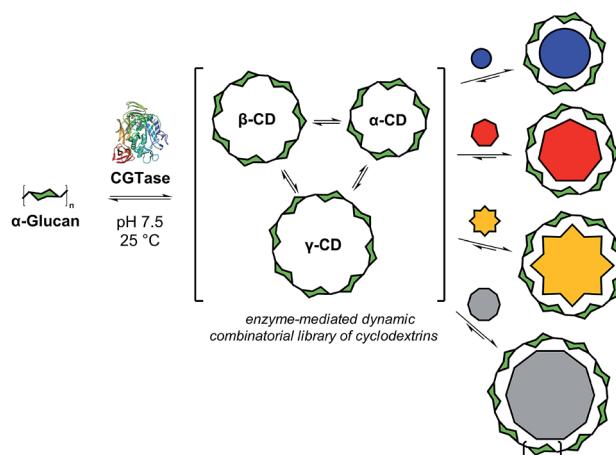


Fig. 1 Concept of enzyme-mediated dynamic combinatorial chemistry for selective cyclodextrin synthesis. Cyclodextrin glucanotransferase (CGTase) is used to generate a dynamic mixture of cyclodextrins starting from various  $\alpha$ -1,4-D-glucans. Addition of suitable templates induces a change in cyclodextrin distribution to selectively produce the 'native' cyclodextrins (either  $\alpha$ -,  $\beta$ -, or  $\gamma$ -CD) or access 'large-ring' cyclodextrins formed from 9 and 10 glucose units.

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In recent decades, chemists have developed more and more complex artificial dynamic chemical systems, exploring different reversible reactions to enable self-assembly 'at equilibrium' and more recently 'out-of-equilibrium'.<sup>7</sup> In principle, reversible enzyme-catalyzed reactions are ideal for use in dynamic systems, as they work under mild, biologically relevant conditions. Enzymes are easily denatured, rendering the library static and the products isolable. The majority of enzymes, however, have evolved to efficiently catalyze unidirectional reactions, making many of them, at first glance, unsuitable for use in dynamic chemical systems. Using optimized conditions, however, dynamic mixtures of short peptides have been reported, generated using peptidase-catalyzed reversible amide formation<sup>6a,c,8</sup> and an aldolase has been employed to form small DCLs of sialic acid analogues.<sup>6b,9</sup>

Cyclodextrin glucanotransferase (CGTase, EC 2.4.1.19) is a glycosyltransferase that catalyses the scrambling of  $\alpha(1 \rightarrow 4)$  glycosidic linkages between the D-glucopyranose monomers of naturally occurring  $\alpha$ -1,4-D-glucans (e.g. maltooligosaccharides or amylose). Importantly, CGTase can cyclize linear  $\alpha$ -1,4-glucans to form cyclodextrins (CDs), in particular,  $\alpha$ -CD,  $\beta$ -CD and  $\gamma$ -CD, which are macrocycles formed from 6, 7, and 8 glucose monomers, respectively.<sup>10</sup> Because of their ability to complex hydrophobic guests, good aqueous solubilities, high stabilities and low toxicities, these so-called 'native' cyclodextrins have found industrial application in many fields, and their supramolecular host-guest chemistries are well-established.<sup>11</sup> The industrial production of cyclodextrins is frequently performed in the presence of different complexing agents, which precipitate either  $\alpha$ ,  $\beta$ , or  $\gamma$ -CD preferentially in order to alter the cyclodextrin product ratio.<sup>12</sup> While  $\alpha$ -CD,  $\beta$ -CD and  $\gamma$ -CD are the major cyclic products of the action of CGTase on

amylose, these products build up over a period of time and large-ring cyclodextrins (CD9 to CD60, with degrees of polymerization of 9 to 60) have been observed briefly in the early stages of CGTase action on amylose.<sup>13</sup> We hypothesized that since cyclodextrin formation appears to be dynamic, it might be possible to target specific products from this reaction, including large-ring cyclodextrins, using a thermodynamic templating effect.

Herein we present the selective templated synthesis of cyclodextrins, exploiting CGTase to establish a dynamic chemical system of linear and cyclic oligosaccharides (Fig. 1). CGTase catalyzes not only reversible transglycosylation but also irreversible hydrolysis.<sup>13,14</sup> Nevertheless, we show that a kinetically-trapped yet dynamic mixture of interconverting cyclodextrins can form, which operates transiently under *pseudo*-thermodynamic control. By addition of carefully chosen templates, we can either shift the library composition to produce almost exclusively  $\alpha$ -,  $\beta$ -, or  $\gamma$ -CD, or we can entirely alter the outcome of this enzymatic transformation to obtain larger ring cyclodextrins,  $\delta$ -CD (CD9) and  $\varepsilon$ -CD (CD10). The distribution of the different cyclodextrins generated in the presence of a template can be predicted using knowledge of the formation and binding constants, and controlled sequential interconversion of cyclodextrins can be achieved in a single reaction vessel.

## Results and discussion

### *pseudo*-Thermodynamic control in an out-of-equilibrium system

The enzyme-promoted reversible formation and interconversion of cyclodextrins was investigated by examining the action

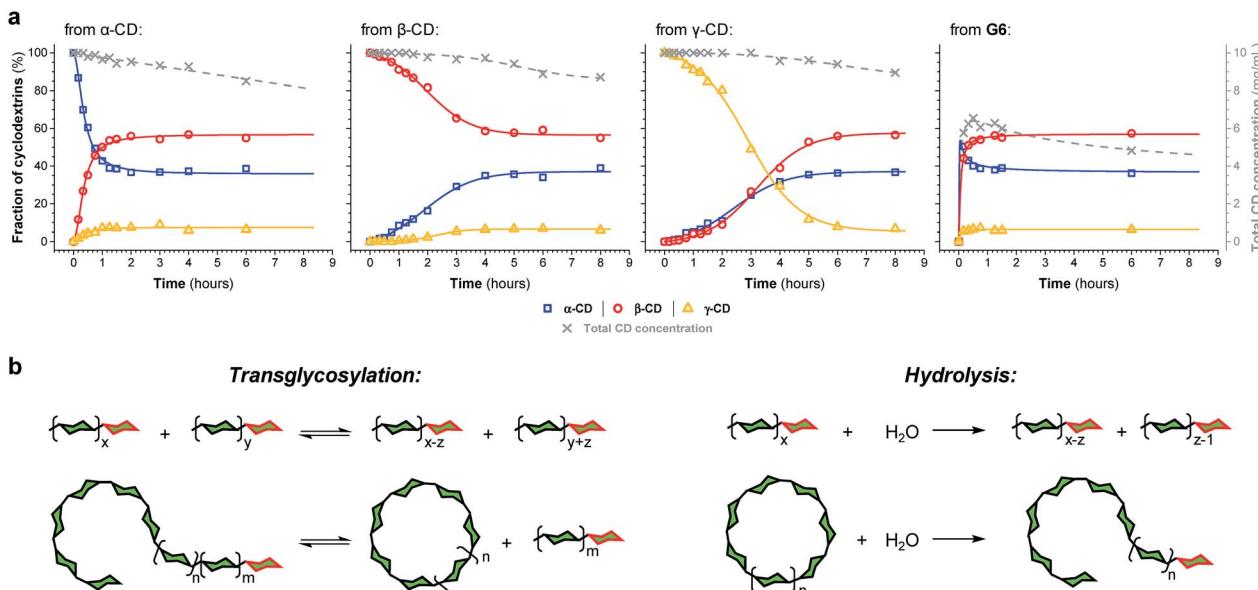


Fig. 2 Thermodynamic control over dynamic cyclodextrin systems. (a) Cyclodextrin distribution (left axis, solid lines) and total cyclodextrin concentration by weight (right axis, dashed lines) as a function of time, determined by HPLC-ELS analysis, when CGTase acts on different  $\alpha$ -1,4-glucans, as indicated ( $10 \text{ mg mL}^{-1}$ ), in  $50 \text{ mM}$  sodium phosphate buffer at pH 7.5. Data points are connected by lines to guide the eye. (b) CGTase catalyzes both reversible inter- and intramolecular glycosyl transfer of  $\alpha$ -1,4-glucans and irreversible hydrolysis of linear and cyclic  $\alpha$ -1,4-glucans ( $n \geq 0$ ;  $m \geq 0$ ;  $x \geq 1$ ;  $y \geq 0$ ;  $x \geq z \geq 1$ ). Green hexagons represent  $\alpha(1 \rightarrow 4)$ -linked D-glucopyranose units. The reducing-end (hemiacetal) glucose units are highlighted in red.



of CGTase from *B. macerans* on  $\alpha$ -CD,  $\beta$ -CD,  $\gamma$ -CD and maltohexaose (**G6**) (Fig. 2a). The reactions were carried out at 25 °C in sodium phosphate buffer at pH 7.5, which are mild reaction conditions optimised to favour thermodynamic control over the system and ensure that the enzyme remained active in solution for at least four weeks. The reactions were monitored using high performance liquid chromatography with evaporative light-scattering detection (HPLC-ELS), which enables detection of the otherwise chromophore- and fluorophore-less oligosaccharides. The reactions each evolved to produce a mixture of cyclodextrins ( $\alpha$ -,  $\beta$ - and  $\gamma$ -CD) and linear  $\alpha$ -1,4-glucans (**G1**–**G8** were detectable). Within a matter of 1–7 hours, depending on the starting material, a steady distribution amongst the three cyclodextrins was reached. In each case, an equilibrium distribution of *ca.* 55–57%  $\beta$ -CD, 37–39%  $\alpha$ -CD, and 6–7%  $\gamma$ -CD was obtained, which indicated that the production of cyclodextrins was not only dynamic, but that the cyclodextrin distribution was determined by their relative stabilities—*i.e.*, that their production takes place under thermodynamic control. Essentially, an enzyme-mediated DCL of cyclodextrins appeared to form.

Unlike in traditional synthetic DCLs, where the covalent linkages between all building blocks can in principle be formed/ cleaved or exchanged, the pathways for interconversion of oligomers in enzyme-mediated DCC is complicated by multi-reactivity and substrate specificity.<sup>10c,e,f,13</sup> CGTase catalyzes both the inter- and intramolecular reversible glycosyl transfer of  $\alpha$ -1,4-glucans, as well as the irreversible hydrolysis of  $\alpha$ -1,4-glucans (Fig. 2b). Furthermore, CGTase can catalyze the trans-glycosylation of some, but not all,  $\alpha$ -1,4-glucans (*e.g.* glucose can act as a glycosyl acceptor but not donor, and linear  $\alpha$ -1,4-glucans with less than 7 glucose residues do not undergo macrocycle-forming intramolecular transglycosylation). Interconversion of cyclodextrins requires a sequence of specific reactions involving the formation of linear oligosaccharide intermediates. In our experiments, the total concentration of cyclodextrins decreased over time (Fig. 2a grey dashed line,  $\times$ ) due to a steady rise in the concentration of linear  $\alpha$ -1,4-glucans. The DCL of cyclodextrins exists only transiently (approx. 1 day). Over a period of several weeks we observed the eventual irreversible hydrolysis of all glycosidic linkages (Fig. 3a and ESI Fig. 1†).

The evolution of the dynamic systems, and the total concentration of cyclodextrins present, when a steady distribution was achieved, varied with the starting material. The reaction with **G6** produced significantly lower concentrations of cyclodextrins (Fig. 2a and ESI Fig. 1d†). We therefore chose to investigate the rate of irreversible hydrolysis when CGTase acts upon  $\alpha$ -CD,  $\beta$ -CD,  $\gamma$ -CD and **G6** by monitoring the build-up of reducing-end (hemiacetal) glucose residues. While hydrolysis generates a range of linear products that are fed back into the dynamic system to produce cyclodextrins, each hydrolytic cleavage results in a new reducing-end glucose, which cannot be reincorporated into a cyclodextrin (Fig. 2b).<sup>15</sup> The hydrolysis reaction was found to follow second order reaction kinetics with respect to the decreasing concentration of hydrolysable glycosidic linkages, and almost identical rate constants ( $k \approx 1.1 \times 10^{-3} \text{ M}^{-1} \text{ s}^{-1}$ ) were measured for the four different starting

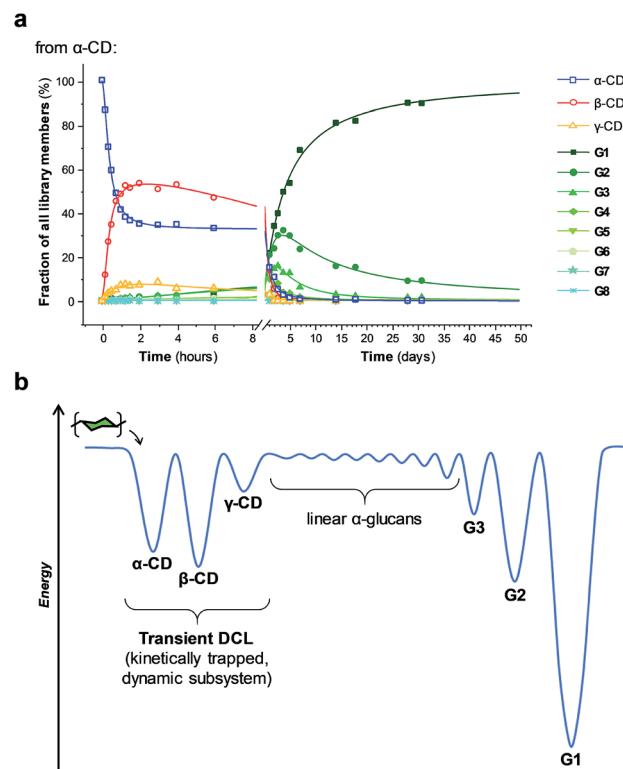


Fig. 3 Cyclodextrins are kinetically trapped products of CGTase action on  $\alpha$ -1,4-glucans, forming a transient DCL in the initial phases of the reaction. (a) Distribution of both linear (**G1**–**G8**) and cyclic ( $\alpha$ -,  $\beta$ -,  $\gamma$ -CD)  $\alpha$ -1,4-glucans produced over time when CGTase acts on  $\alpha$ -CD. (b) Energy diagram visualizing the suggested energy landscape of the dynamic system.  $\alpha$ -,  $\beta$ -, and  $\gamma$ -CD are kinetically trapped in local minima with surpassable energy barriers for their interconversion, but **G1** is the thermodynamic product and located in the deepest well.

materials ( $\alpha$ -CD,  $\beta$ -CD,  $\gamma$ -CD and **G6**) (ESI Fig. 2a†). The observed lower conversion of **G6** to cyclodextrins is then not a consequence of faster hydrolysis. It is rather related to the fact that 1/6th of the glucose residues in **G6** (namely the reducing-end glucose) cannot be utilized to form cyclodextrins (so the highest possible theoretical yield of cyclodextrins from **G6** would be  $5/6 \approx 83\%$ ). Furthermore, for **G6**, hydrolysis immediately generates very short maltooligosaccharides, from which there is a low probability that intermolecular glycosyl transfer will lead to linear  $\alpha$ -1,4-glucans long enough to then cyclize and form cyclodextrins.

We thus reached the conclusion that although glucose is the true thermodynamic product of the action of CGTase on  $\alpha$ -1,4-glucans,  $\alpha$ -CD,  $\beta$ -CD and  $\gamma$ -CD must exist within a kinetically trapped sub-system over which there is *pseudo*-thermodynamic control. Fig. 3b suggests a picture of the overall energy landscape in the system. The energy barrier for the interconversion of cyclodextrins is readily overcome, so that thermodynamic control exists transiently over this sub-system. As with all kinetically trapped systems, the starting point from which equilibrium is approached (*i.e.* the particular  $\alpha$ -1,4-glucan supplied to the system) affects the degree to which the system becomes kinetically trapped (*i.e.* the overall cyclodextrin



concentration and the lifetime of the transient DCL). So long as thermodynamic control exists transiently, however, there is the opportunity to employ templates to stabilize and amplify specific oligomeric products within the system.

### Templated selective cyclodextrin synthesis

By utilizing templates known to bind strongly and selectively to cyclodextrins in aqueous solution we could successfully promote the synthesis of  $\alpha$ -,  $\beta$ - or  $\gamma$ -CD with high selectivity. We found that when  $\alpha$ -CD (10 mg  $\text{mL}^{-1}$ ) was treated with CGTase in the presence of sodium dodecyl sulfate (SDS) (1) (10 mM),<sup>15</sup>  $\alpha$ -CD made up 83% of the total cyclodextrin concentration at *pseudo*-equilibrium (Fig. 4b), compared with 38% in the absence of template (Fig. 4a). The use of 1-adamantane carboxylic acid (2)<sup>16</sup> as a template resulted in the formation of almost exclusively  $\beta$ -CD (>99% of total CDs, Fig. 4c). Templating with  $\text{NaBPh}_4$  (3)<sup>17</sup> selectively yielded  $\gamma$ -CD (>99% of total CDs, Fig. 4d). Identical distributions of cyclodextrins in the presence of templates were obtained irrespective of which  $\alpha$ -1,4-glucan was used as the starting material ( $\alpha$ -,  $\beta$ -,  $\gamma$ -CD or G6) (ESI Fig. 3–5†), which is further evidence of *pseudo*-thermodynamic control.

The supramolecular chemistry of large-ring cyclodextrins is relatively unexplored. They are accessible only after extensive chromatographic separation from complex mixtures containing numerous cyclodextrins of different sizes formed when amylose is briefly exposed to CGTase.<sup>13,18</sup> It was recently reported that

CD9–CD11 can bind dodecaborate cluster ions with high affinity and some selectivity.<sup>19</sup> We therefore treated  $\alpha$ -CD with CGTase in the presence of  $\text{Cs}_2\text{B}_{12}\text{I}_{12}$  (4) and observed a remarkable shift in product distribution (Fig. 4e and ESI Fig. 6†). At *pseudo*-equilibrium, two new peaks were observed in the HPLC-ELS chromatogram and identified as CD9 and CD10 (by MALDI-TOF-MS analysis following fractional collection of the HPLC-separated compounds). These new peaks from the larger cyclodextrins CD9 and CD10 made up 13% and 4% of the total area of all cyclodextrin peaks, respectively, after 6 hours of reaction, whereas they were practically unobservable in the absence of the template. A similarly high yield of CD10 has recently been reported, but using a genetically engineered CGTase.<sup>20</sup> CD9 has been isolated in only 0.04% yield using a native CGTase.<sup>18a</sup> We found that despite the higher reported affinities of CD9 ( $K_a = 6.8 \times 10^5 \text{ M}^{-1}$ ) and CD10 ( $K_a = 2.1 \times 10^6 \text{ M}^{-1}$ ) for the  $\text{B}_{12}\text{I}_{12}^{2-}$  dianion compared with  $\gamma$ -CD ( $K_a = 6.7 \times 10^4 \text{ M}^{-1}$ ),<sup>19,21</sup>  $\gamma$ -CD was the major cyclodextrin product generated in this reaction. Product distribution in a DCL is determined not only by the relative affinities of the library members for a given template, but also the intrinsic stability of the different library members.<sup>22</sup> That CD9 and CD10 are not observed in the absence of a template is indicative of their much smaller formation constants. Their templated synthesis is therefore significantly more challenging and requires particularly strong interactions with a template. At such high affinities it also becomes increasingly likely that library members can

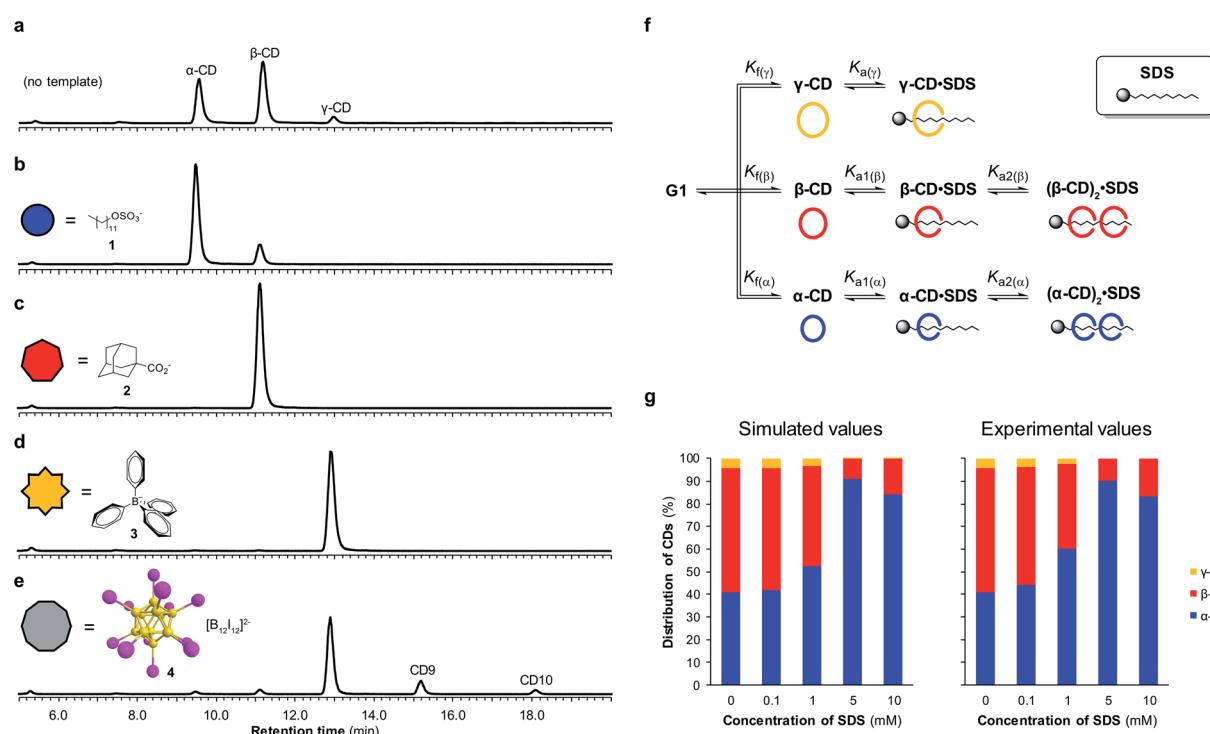


Fig. 4 Template-directed enzyme-mediated dynamic combinatorial synthesis of cyclodextrins. (a–e) Chromatograms (HPLC-ELS) showing the *pseudo*-equilibrium distribution of cyclodextrins produced when CGTase acts on  $\alpha$ -CD in the absence or presence of templates 1–4 (10 mM). (f) Model employed to simulate a DCL of cyclodextrins and predict the concentrations of  $\alpha$ -,  $\beta$ -, and  $\gamma$ -CD in the presence of SDS (1). (g) Comparison between the predicted and experimentally determined cyclodextrin distributions generated in the presence of different concentrations of SDS (1).



become kinetically trapped in the presence of the template. Although the highly selective synthesis achieved for  $\alpha$ -,  $\beta$ - and  $\gamma$ -CD could not be replicated for CD9 and CD10, this result demonstrates the possibility to obtain entirely new products from enzymatic reactions *via* a thermodynamic template effect.

It is noteworthy that in the presence of templates, higher overall yields of cyclodextrins were observed and the build-up of linear products was slower for all starting materials. The rate of CGTase-catalyzed irreversible hydrolysis of the different  $\alpha$ -1,4-glucans in the presence of 1-adamantane carboxylic acid was thus investigated. Significantly smaller apparent second order rate constants (*ca.* 45-fold lower) were determined (ESI Fig. 2b†), compared with those for hydrolysis in the absence of template. It appears that the cyclodextrins are 'protected' against hydrolysis by host–guest complex formation.

### DCL simulation

Unsatisfied with the obtained 83% selectivity for  $\alpha$ -CD achieved using SDS (1) as a template, we sought to improve this result. Screening of a series of different amphiphiles revealed that templates with longer aliphatic chains were more effective at amplifying  $\alpha$ -CD, but  $\beta$ -CD was still consistently obtained as a side product (ESI Fig. 7†). We suspected that the inability to completely switch the cyclodextrin distribution to  $\alpha$ -CD was not because  $\alpha$ -CD bound too weakly to 1, but rather due to a competing interaction between 1 and  $\beta$ -CD. As it has been shown that in DCLs where several oligomers can bind the template, the distribution of the different oligomers formed is influenced by the concentration of the template,<sup>22</sup> we investigated the action of CGTase on  $\alpha$ -CD in the presence of varying

concentrations of SDS (1) (Fig. 4g). An improved 89% selectivity was obtained when the template concentration was lowered from 10 mM to 5 mM.

To understand this result, the binding interaction between SDS (1) and each cyclodextrin was investigated individually by means of  $^1\text{H}$  NMR spectroscopy titrations (ESI Fig. 8–10†). The resulting binding isotherms clearly showed that SDS (1) can bind two molecules of both  $\alpha$ -CD and  $\beta$ -CD. For  $\alpha$ -CD, two strong binding interactions were observed ( $K_{a1} = (1.9 \pm 0.5) \times 10^4 \text{ M}^{-1}$  and  $K_{a2} = (2.3 \pm 0.4) \times 10^4 \text{ M}^{-1}$ ), while for  $\beta$ -CD a comparably high first binding constant was determined but the second binding event was markedly weaker ( $K_{a1} = (1.6 \pm 0.4) \times 10^4 \text{ M}^{-1}$  and  $K_{a2} = (7 \pm 4) \times 10^2 \text{ M}^{-1}$ ). Using the obtained binding constants, along with relative formation constants for  $\alpha$ -,  $\beta$ -, and  $\gamma$ -CD, determined from the *pseudo*-equilibrium distribution of these cyclodextrins in the absence of template, we simulated a DCL of cyclodextrins according to the model shown in Fig. 4f using the *DCLsim* software previously developed in the Otto group.<sup>22a</sup> The expected distribution of cyclodextrins in the presence of varying concentrations of SDS were calculated and it was predicted that the highest yield of  $\alpha$ -CD would in fact be obtained at a lower template concentration, in line with our experimental results (Fig. 4g). We could rationalize this result on the basis of a 2 : 1 binding mode, as a 5 mM template concentration equates to approximately one molecule of SDS (1) per two molecules of  $\alpha$ -CD. That the *DCLsim* software, developed to predict the outcome of a network of chemical equilibria operating under thermodynamic control, could successfully predict the distribution of cyclodextrins in our system supports the experimental evidence that *pseudo*-

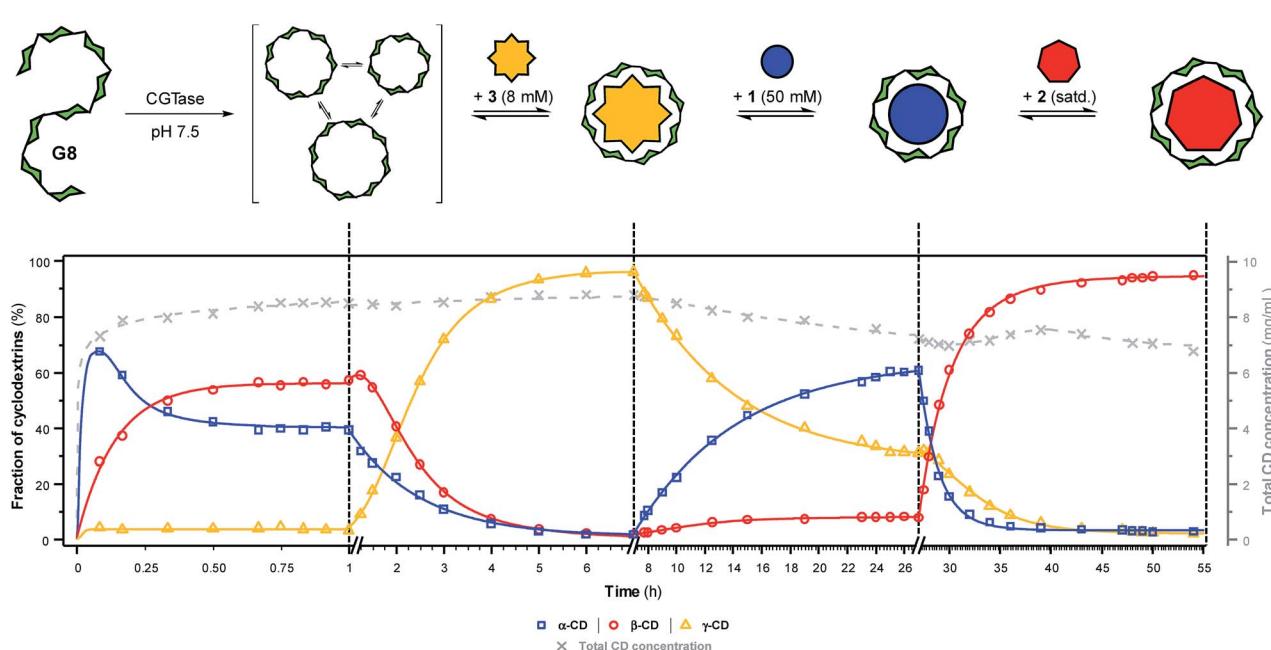


Fig. 5 Sequential interconversion of cyclodextrins. Cyclodextrin distribution (left axis, solid lines) and total cyclodextrin concentration (right axis, dashed line) as a function of time when different templates ( $\text{NaBPh}_4$  (3), SDS (1) and 1-adamantane carboxylic acid (2)) were added sequentially, and in one pot, to a dynamic cyclodextrin system generated by the action of CGTase on maltooctaose (G8). Data points are connected by lines to guide the eye. Conditions and analysis as in Fig. 2.



thermodynamic control exists transiently over this subsystem of kinetically trapped cyclodextrins.

### Sequential cyclodextrin interconversion

To further showcase the extent to which we can use templates to control the outcome of this dynamic cyclodextrin system, we sought to switch sequentially between the three 'native' cyclodextrins in a single reaction vessel (Fig. 5). Maltooctaose (**G8**) was exposed to CGTase (under conditions identical to previous experiments) and allowed to react and reach a *pseudo*-equilibrium cyclodextrin mixture before addition of one equivalent of NaBPh<sub>4</sub> (**3**). Within six hours, the cyclodextrin composition switched to almost exclusively  $\gamma$ -CD (*ca.* 95%). At this time, SDS (**1**) was added in large excess, which, as anticipated, caused a change in the cyclodextrin composition to produce primarily  $\alpha$ -CD (*ca.* 60%) within 20 hours. Due to the lower affinity and selectivity of **1**, compared with the other templates, it was not possible to push the cyclodextrin distribution to a higher  $\alpha$ -CD concentration. 1-Adamantane carboxylic acid (**2**) was then added in excess, and within 20 hours the cyclodextrin composition had switched to nearly exclusively  $\beta$ -CD (*ca.* 95%). Remarkably, CGTase remained active over several days and in the presence of high concentrations of the various templates. Furthermore, cyclodextrins accounted for *ca.* 70% of the glucan material after 54 hours, compared with *ca.* 50% after only 8 hours in the untemplated reaction starting from **G6** (Fig. 2a). As cyclodextrins are presumably not substrates for the enzyme when bound to the templates, the presence of a template reduces the concentration of available substrate in solution. This leads to a longer equilibration time for the interconversion of cyclodextrins (note the time-axis in Fig. 5), but also a reduced rate of hydrolysis, leading to a longer-lasting transient DCL of cyclodextrins operating under *pseudo*-thermodynamic control.

## Conclusions

In summary, we have exploited CGTase to generate dynamic mixtures of interconverting cyclodextrins operating under *pseudo*-thermodynamic control, from which we could template the selective synthesis of  $\alpha$ -CD,  $\beta$ -CD or  $\gamma$ -CD, and access otherwise inaccessible large-ring cyclodextrins. This work shows for the first time the systematic use of artificial templates to selectively access different target products from a system of enzyme-catalyzed reactions, and thus demonstrates unprecedented control over, and the ability to redirect the outcome of enzymatic reactions.

We believe that combining enzyme-mediated reactions with thermodynamic templating represents an important step forward in directed self-assembly. Enzymes offer new opportunities as they can generate dynamic systems from seemingly inert molecules. The approach is not limited to carbohydrate-modifying enzymes. We foresee that a range of enzymes could be employed for templated enzymatic synthesis and also that enzyme-mediated dynamic systems could be generated using sets of complementary enzymes catalysing irreversible reactions

that interconvert products and substrates *via* a series of reaction steps. Employing templates to modulate the fine balance between kinetic and thermodynamic control in enzyme-mediated systems offers a new approach for synthetic (bio) chemists to control enzyme reactivity and selectivity, and obtain new, or difficult-to-access, structures and materials.

## Conflicts of interest

There are no conflicts to declare.

## Acknowledgements

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