Dynamic covalent chemistry (DCC) is a widely used method for the self-assembly of three-dimensional molecular architectures. The orthogonality of dynamic reactions is emerging as a versatile strategy for controlling product distributions in DCC, yet the application of this approach to the synthesis of 3D organic molecular cages is limited. We report the first system which employs the orthogonality of alkyne metathesis and dynamic imine exchange to prepare a molecular cage with a reversibly removable vertex. This study demonstrates the rational and controlled application of chemical orthogonality in DCC to prepare organic cages of expanded functionality which respond to chemical stimuli.

Introduction

Recent improvements in alkyne metathesis (AM) have led to expanded applications in both total synthesis and materials chemistry. Alkyne metathesis is one among a number of reversible reactions utilized in dynamic covalent chemistry (DCC), and is particularly useful for the synthesis of 3D organic molecular cages. DCC employs reversible covalent bonds to self-assemble complex structures from simpler building blocks. The DCC toolbox comprises a number of orthogonal reactions, including alkyne/olefin metathesis and imine/hydrazone reactions. Orthogonality in DCC is frequently leveraged in dynamic combinatorial library synthesis, where modification of reaction conditions can alter product distributions through selective activation of dynamic bonds. The ability to ‘turn-on’ a subset of dynamic functionalities affords a synthetic handle to reversibly shift product distributions and generate modified structures through post-synthetic transformations. The utility of orthogonal DCC has also been applied to the synthesis of organic cages and macrocycles through both simultaneous and sequential activation of orthogonal bonds to generate complex architectures from simple precursors. Recent success by our lab and others in the synthesis of organic cages via alkyne metathesis has led us to pursue both synthetic modifications and materials applications of these shape-persistent structures. After surveying the literature, we were motivated by the apparent lack of examples combining alkyne metathesis with orthogonal dynamic chemistries in multitopic DCC. The development of highly active and functional group tolerant AM catalysts further demonstrates that this methodology is poised to expand into new chemical space. We have therefore chosen to pursue synthetic strategies which leverage orthogonal dynamic chemistries to prepare novel molecular cages. Herein, we demonstrate the preparation of a molecular cage incorporating two orthogonal dynamic bonds, each of which are selectively activated (Fig. 1B).

Inspired by the established utility of dynamic imine chemistry, we chose to investigate the feasibility of combining this orthogonal bond with alkyne metathesis. Two key considerations stood out at the onset of this study; namely, whether imine functionality is chemically compatible with AM and geometrically compatible with the alkyne bond to afford

![Fig. 1](image_url)
a reasonable 3D cage structure. Pathway complexity in multi-
topic DCC means that seemingly small variations in precursor
structure can lead to large changes in product distribution.\textsuperscript{6,37}
As such, substituting the trigonal imine bond for one or more
linear alkyne bonds could alter the precursor structure such
that no stable product configuration is available. Drawing
inspiration from Zhang et al. who demonstrated the synthesis
of macrocycles and cages through tandem imine exchange and
olefin metathesis,\textsuperscript{56–58} we hypothesized that a sufficiently pre-
organized alkyl-L metathesis precursor could be generated via
tongling of alkyne subunits via imine condensation.

Results and discussion

To begin, we prepared functional precursors 1 and 2 to
demonstrate the compatibility of AM with the imine functional
group (Scheme 1). These precursors are analogous to those
which afford tetrahedral cages via alkyl-L metathesis, where the
1,3,5-arrangement of a hexasubstituted arene aids in pre-
oragizing the structure towards cage formation.\textsuperscript{59–61} We next
coupled 1 and 2 via imine condensation to afford a single hexa-
tropic alkyl-L metathesis precursor 3 which we hypothesized is
sufficiently preorganized to undergo cyclization to a cage
(Scheme 1). The coupling of 1 and 2 proceeds in good yield and
precursor 3 isolated by precipitation from methanol. This
precursor also exhibits good solubility in organic solvents (i.e.
CHCl\(_3\) and CCl\(_4\)) which are well-suited for alkyl-L metathesis
reactions.

Having prepared imine-linked precursor 3 we investigated its
reactivity under alkyl-L metathesis conditions. Given the lack of
examples of imine functionality in AM,\textsuperscript{42} a key consideration
was the selection of an appropriate metathesis catalyst. Zhang
and co-workers have demonstrated that tridentate, phenol-
based supporting ligands generate highly active and function-
ally tolerant molybdenum metathesis catalysts.\textsuperscript{44,45,51} We began
our investigations using ligand B from which the active catalyst
is generated \textit{in situ} upon stirring with molybdenum precatalyst
A.\textsuperscript{44,51} In our preliminary experiments we were delighted to find
that upon exposure of precursor 3 to this catalyst system at mild
temperature (40 °C) a product putatively assigned to the imine-
labeled cage 4 was present as determined by Gel Permeation
Chromatography (GPC) and MALDI-MS analysis. Upon further
optimization, we found that using 10 mol% molybdenum
catalyst A/ligand B in CCl\(_4\) (3 mM) at 40 °C afforded mixed
imine/alkyne cage 4 in 80% yield and approximately 96% purity
(Scheme 1). As is standard in alkyl-L metathesis, 5 Å molecular
sieves were added to remove 2-butyne and drive the reaction
forward.\textsuperscript{61} The metathesis reaction proceeds well at useful
scales (e.g. 100 mg of precursor) and the product is easily iso-
lated by precipitation and filtration. GPC analysis supports
conversion of 3 to 4, with higher-molecular weight species (ca.
4%) also present in the isolated material (see Fig. S2†). MALDI-
MS analysis of the isolated product suggests 4 and the presence
of a dimeric structure, which may be a catenated cage-structure
or a dimer of precursor 3 (Fig. S1†). Despite numerous attempts
to obtain an X-ray crystal structure of 4, including several poorly
diffracting single crystals, we were unable to obtain a data set
suitable for structure refinement. Cage 4 was further charac-
terized by \(^1\)H and \(^{13}\)C NMR (see ESI†). \(^1\)H NMR analysis of iso-
lated 4 shows the loss of the propynyl signal in precursor 3 and
a downfield shift indicating successful metathesis. Two char-
acteristic doublets for the alkyne-linked ‘base’ of 4 are nearly
identical in chemical shift to the tetrahedral cage we have re-
ported previously \((δ\textsubscript{H} 7.45/7.02 \text{ ppm and } δ\textsubscript{C} 44/56.99 \text{ ppm respec-
tively})\), and the presence of four additional doublets are
consistent with the differentiated imine-linked vertex. \(^1\)H–\(^1\)H
COSY analysis allowed for assignment of the aromatic proton
signals (see ESI†). \(^{13}\)C NMR analysis also shows the loss of the
propynyl signal of 3. Cage 4 was further characterized by
a number of 2D NMR experiments (i.e. HSQC, HMBC, and
NOESY, see ESI†).

A common characteristic of DCC reactions is the ability for
these systems to ‘self-correct’ to a discrete product.\textsuperscript{56,53,63–65} In
many systems employing alkyl-L metathesis, high molecular
weight oligomeric/polymeric products are often formed at early
reaction times and gradually convert to a single product.\textsuperscript{6,9} In
previous studies on the synthesis of \(T\textsubscript{d}\) symmetric cages from
tritopic precursors, we observed the formation of oligomeric
intermediates which proceed on a pathway to a discrete cage.\textsuperscript{8}
Given that 4 results from a single hexatomic precursor, we
wondered if its formation bypasses any oligomeric intermediates
and only proceeds \textit{via} sequential intramolecular metath-
esis steps.\textsuperscript{66–68} We monitored the progress of the AM reaction of
3 to 4 by GPC and observed that, even at the dilute 3 mM
reaction concentration, oligomeric products are formed which
reversibly correct to a discrete product (Fig. 2). In addition to
higher molecular weight species (16–17 min retention time),
early time points display a ‘shouldering’ which may be intra-
molecular metathesis products as precursor 3 converts to cage.
The dimeric product observed in the isolated 4 may account for
the peak observed at ca. 17 minutes retention time. The relative
proportion of higher molecular weight products also appears to
be lower than we have observed in the metathesis reaction of

\(\text{Scheme 1 Synthesis of a } C\textsubscript{6}\text{ symmetric tetrahedral organic cage via a tandem imine condensation/alkyne metathesis strategy.}\)
other tritopic precursors. This may suggest that the predominant pathway for conversion of 3 to 4 consists of intramolecular metathesis steps. Thus, even when alternate reaction pathways are available for 3, the reversible nature of AM allows for these intermediates to return to a pathway that leads to 4. \(^7\) These results also suggest that a tetrahedral cage is the favoured reaction outcome for hexasubstituted tritopic and hexatopic precursors (i.e., 1, 2, and 3) which are geometrically preorganized in a manner that biases the reaction pathway to favour tetrahedral cage formation. \(^7\)

With cage 4 in hand, we turned our attention to identifying conditions for selective activation of the single all-imine connected vertex. In the context of cage synthesis, methods which allow for controlled modification of cage geometry, functionality, and cavity size offer the potential to tailor these properties for specific host-guest interactions. \(^6\) We therefore hypothesized that the presence of orthogonal subsets of dynamic bonds in 4 would prove useful for its selective modification. We were particularly inspired by the efficiency of Sc(III) catalysed transimination reactions. \(^6\) In a seminal publication, Lehn and co-workers demonstrated that transimination is effectively catalysed by Sc(OTf)\(_3\) in the presence of alkyl/aryl amines. \(^7\) The resulting equilibrium distribution has a tendency to favour the imine formed from the most basic amine. \(^7\) Though transimination reactions do proceed in the absence of a Lewis or Brønsted acid catalysts, \(^7\) Lehn has reported rate enhancements of up to five orders of magnitude in the presence of Sc(OTf)\(_3\). \(^7\) The high efficiency of this Lewis acid-catalysed transimination was thus attractive as a method to activate the imine-linked vertex of cage 4.

We began our investigations under the hypothesis that a high concentration of a basic alkyl amine would result in an equilibrium that favours removal of the imine vertex of 4. Methylamine, which is commercially available as a THF solution, was selected for cage disassembly studies. Methylamine is basic and highly volatile which allows for simple removal \(\textit{via}\) high-vacuum (see below). Cage 4 was subjected to 20 equiv. of methylamine in CDCl\(_3\)/MeCN-D\(_3\) with Sc(OTf)\(_3\) as catalyst (Fig. 3). After 5 h at RT, \(^1\)H NMR analysis of the reaction mixture was consistent with disassembly of cage 4 to macrocycle 5 and tris-imine 6 (Fig. 3). NMR analysis shows an upfield shift of the imine proton signal of cage 4 from \(\delta 8.52\) ppm to \(\delta 8.21\) ppm, consistent with transimination from an aryl- to an alkyl-substituted imine (Fig. 3 and S5†). Compound 6 was independently synthesized to confirm its assignment (Fig. S4†). MALDI-MS analysis of the reaction mixture confirms the presence of macrocycle 5 (Fig. S6†). While the NMR spectra suggest that disassembly is nearly complete, detectable levels of the residual cage 4 were also observed in the MALDI spectrum (Fig. S6†). MALDI also suggests detectable levels of partially disassembled cage 4, in which one and two imines, respectively, have reacted with methylamine (Fig. S6†). From the NMR spectra, we conclude the disassembly affords a >20 : 1 ratio of 6 : 4 (see Fig. 3 inset and S5†). Upon addition of Sc(OTf)\(_3\) to the reaction mixture, some cloudiness was observed indicating that formation of insoluble oligomeric products may be a competing side-reaction during the disassembly process. Bronsted acid catalysis also effectively disassembles 4. Subjecting the cage to similar conditions employing catalytic trifluoroacetic acid afforded comparable results to the reaction performed using Sc(OTf)\(_3\) as catalyst (see Fig. S7†). Other alkyl amines (i.e., \(n\)-propylamine) are also capable of disassembling cage 4 under both Sc(OTf)\(_3\) and TFA catalysis (see Fig. S7† for results with \(n\)-propylamine disassembly using TFA).

Having identified conditions which disassemble 4 we next sought to determine whether the cage could be subsequently reassembled after its conversion to 5 and 6. This would demonstrate that a viable assembly pathway is available from two starting points; \(i.e\) dynamic alkyne metathesis and dynamic imine exchange reactions. In early reassembly experiments using \(n\)-propylamine, we were initially intrigued to observe partial reassembly of cage 4 after removal of excess amine and stirring the reaction mixture overnight. Given that dynamic imine exchange is in equilibrium, we recognized that even after removal of excess amine, an equilibrium would still exist between cage 4 and compounds 5 and 6. \(^7\) We hypothesized that this equilibrium would shift through \textit{in situ} removal of methylamine remaining after transimination of 5 and 6 to cage 4. In dynamic alkyne metathesis, 5-Å molecular sieves (MS) are used to sequester residual 2-butyne and drive the reaction forward. \(^6\) We hypothesized that 5 Å MS would also effectively sequester residual methylamine. In order to probe the feasibility of reassembly, the cage disassembly reaction was performed as described above, using volatile methylamine under scandium catalysis (Fig. 3). After stirring for five hours, the resulting reaction mixture was dried under high vacuum for one hour to remove excess methylamine. After re-dissolving the residue from the disassembly process and allowing the reaction to stir overnight at room temperature in the presence of 5 Å molecular sieves, \(^1\)H NMR analysis of the reaction was consistent with reassembly of cage 4 (Fig. 3). MALDI-MS and \(^1\)H NMR
analysis of the remaining material shows that the cage is reformed efficiently, with no additional products observed by MALDI-MS (Fig. S6†).

The results of this study demonstrate that assembly of 4 is achieved through dynamic alkyne metathesis from a hexatopic precursor 3 and through dynamic imine exchange from 5 and 6. The fact that 4 is accessed from both starting points through orthogonal DCC reactions demonstrates that it possesses a structural stability that is favoured and kinetically viable for both processes. As indicated by the results in Fig. 2, the formation of alternate reaction products does not preclude the formation of cage 4 from either starting point; instead, these intermediates are capable of returning to a pathway to a discrete cage structure. We imagine that the ability to assemble discrete molecular cages via orthogonal dynamic bonds will open new opportunities not only for synthesis of diverse molecular cages, but also for controlled modification of these architectures.

Conclusion

In this report we have demonstrated the synthesis of a C3v symmetric organic molecular cage assembled from two dynamic covalent reactions. By combining orthogonal alkyne and imine bonds the resulting cage possesses an imine-linked vertex which is removed and replaced through Sc(III) catalysed transimination. To the best of our knowledge this is the first example of orthogonal dynamic covalent chemistry being combined with alkyne metathesis for the synthesis of organic molecular cages and the first example of imine functionality being tolerated in an AM reaction. This work not only demonstrates a strategy for preparing cages of altered symmetry but opens new avenues for the preparation of discrete cages which are modified selectively. We believe that this study complements existing strategies for molecular cage synthesis and will allow for the preparation of architectures with greater control over their symmetry and functionality. Further studies into the preparation of complex nanostructures using this method, as well as their applications in selective host–guest chemistry, are ongoing.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

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