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Regioselectivity Switch in Pd-Catalyzed Hydroallylation of Alkynes

Ding-Wei Ji,^{a,b} Yan-Cheng Hu,^a Hao Zheng,^{a,b} Chao-Yang Zhao,^{a,b} Qing-An Chen^{*a} and Vy M. Dong^cReceived 00th January 20xx,
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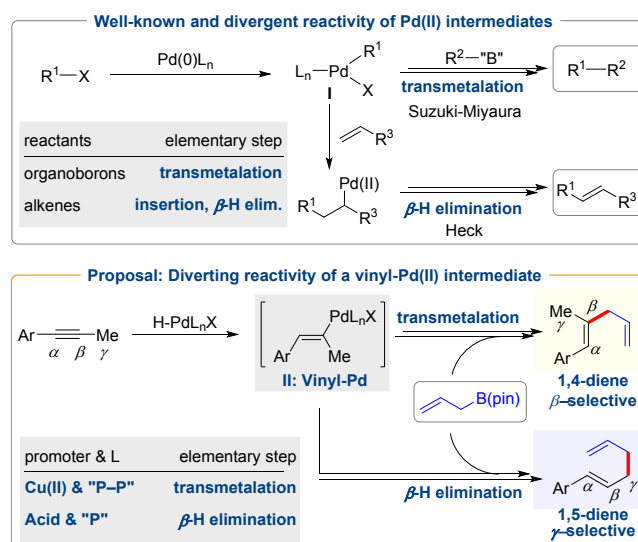
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By exploiting the reactivity of a vinyl-Pd species, we control the regioselectivity in hydroallylation of alkynes under Pd-hydride catalysis. A monophosphine ligand and carboxylic acid combination promotes 1,5-dienes by a pathway involving isomerization of alkynes to allenes. In contrast, a bisphosphine ligand and copper cocatalyst favor 1,4-dienes *via* a mechanism that involves transmetalation. Our study highlights how to access different isomers by diverting a common organometallic intermediate.

Introduction

Inventing novel cross-couplings has relied upon our ability to divert readily accessible and common organometallic intermediates.¹ For example, Pd(II) intermediates I (generated from organic halides or their analogues) can transmetallate with organoborons in a Suzuki-Miyaura cross-coupling² or undergo insertion to olefins and subsequent β -hydride elimination in the Heck reaction (Scheme 1, top).³ Inspired by the power of this concept, we set out to divert the reactivity of a vinyl-Pd(II) intermediate II (generated from alkynes, Scheme 1, bottom) to achieve useful skipped dienes.

The hydroallylation of alkynes has attracted attentions due to occurrence of skipped dienes in bioactive compounds and natural products.⁴ While various catalysts have been developed to generate the 1,4-diene motif,⁵ access to the 1,5-diene isomer *via* hydroallylation of alkynes has been elusive. In considering this challenge, we were inspired by the work of Trost,⁶ Yamamoto,⁷ and Breit⁸ who have used alkynes as redox-neutral allyl precursors.^{9–12} Early studies established that Pd–H can add to an alkyne to generate the vinyl-Pd species II. With this in mind, we set out to manipulate the reactivity of this Pd(II) species II towards transmetalation or β -H elimination to enable selective access to the 1,4 or 1,5-dienes, respectively (Scheme 1, bottom). Herein, we report the use of ligands and promoters to enable a regiodivergent synthesis of skipped dienes. Our study contributes to the art of diverging catalytic intermediates to access different constitutional isomers.¹³



Scheme 1. Divergent reactivity of Pd(II) species

Results and discussion

We began our study with 1-phenyl-1-propyne **1a** and allyl-B(pin) **2a** as the model substrates. After examining various combinations of ligands and additives, we obtained compelling results (Chart 1). In the presence of $Pd_2(dba)_3$ (2.5 mol%) and a proton source $nBuOH$ (2.0 equiv.), we found that monophosphine ligands such as $PPhCy_2$ and PCy_3 could give 1,5-diene (**3a**) as major product. With the aid of PCy_3 , 1,5-diene **3a** could be obtained in 18% yield accompanied by trace amounts of 1,4-dienes **4a** and **4a'** mixture. To improve the yield of 1,5-diene **3a**, we chose Brønsted acid to facilitate the formation of active Pd(II)–H catalyst.⁷ By adding 10 mol% 1-adamantanecarboxylic acid, we observed **3a** in 83% yield with excellent selectivity (Chart 1B). Bisphosphine ligands gave only trace amount of 1,5-diene **3a** because these ligands occupy the otherwise vacant sites needed for β -hydride elimination.^{1,14}

^a Dalian Institute of Chemical Physics, Chinese Academy of Sciences, Dalian 116023 P. R. China, E-mail: qachen@dicp.ac.cn

^b University of Chinese Academy of Sciences, Beijing 100049, P. R. China

^c Department of Chemistry, University of California, Irvine, California, 92697-2025, USA.

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Chart 1. Ligand and promoter effects on hydroallylation^[a]

		PPh ₃	PPhCy ₂	Xphos	PCy ₃	P ^t Bu ₃
A	Mono P:	PPh ₃	PPhCy ₂	Xphos	PCy ₃	P ^t Bu ₃
	Yield ^[b] 3a :	n.d.	9%	n.d.	18%	n.d.
	Yield ^[b] 4a/4a' :	n.d.	trace	n.d.	trace/2%	n.d.
	Bis P-P:	dppe	dppf	dpp-benz	dCype	dppp
B	Mono P:	PPh ₃	PPhCy ₂	Xphos	PCy ₃	P ^t Bu ₃
	Yield ^[b] 3a :	34%	80%	21%	83%	12%
	Yield ^[b] 4a/4a' :	2%/2%	trace	n.d.	trace	n.d.
	Bis P-P:	dppe	dppf	dpp-benz	dCype	dppp
C	Mono P:	PPh ₃	PPhCy ₂	Xphos	PCy ₃	P ^t Bu ₃
	Yield ^[b] 3a :	3%	9%	10%	trace	9%
	Yield ^[b] 4a/4a' :	trace	trace	n.d.	n.d.	n.d.
	Bis P-P:	dppe	dppe	dpp-benz	dCype	dppp
Yield ^[b] 3a :	n.d.	n.d.	12%	n.d.	n.d.	
Yield ^[b] 4a/4a' :	trace	53%/5%	34%/4%	5%/trace	17%/1%	
				79%/7% ^[c]		

[a] **1a** (0.20 mmol), **2a** (2.0 equiv), Pd₂(dba)₃ (2.5 mol%), monophosphine (10 mol%) or bisphosphine (5 mol%), promoter (10 mol%), ^tBuOH (2.0 equiv), dioxane (1.0 mL), 90 °C, 6 h. [b] Determined by ¹H NMR or GC-FID with 1,3,5-trimethoxybenzene as the internal standard. [c] MeOH (3.0 equiv) instead of ^tBuOH, dioxane (0.5 mL), 70 °C, 24 h.

Table 1. Regioselective hydroallylation of alkynes^[a]

A) 1,5-Dienes Synthesis Promoted By βH Elimination				B) 1,4-Dienes Synthesis Promoted By Transmetalation			
3a : 82%	3b : 78%	3c : 71%	3d : 75% ^b	4a/4a' : 71% (11.9:1)	4b/4b' : 62% (7.9:1)	4c/4c' : 70% (8.7:1)	4d/4d' : 70% (5.6:1) ^b
3e : 74%	3f : 78%	3g : 84%	3h : 84%	4e/4e' : 71% (12.0:1)	4f/4f' : 75% (18.5:1)	4g/4g' : 86%	4h/4h' : 88% (17.9:1)
3i : 81%	3j : 90%	3k : 72% ^b	3l : 78%	4i : 78%	4j : 99%	4k/4k' : 87% (9.6:1)	4l/4l' : 93% (20.0:1)
3m : 84%	3n : 52% ^b	3o/3o' : 78% (1:1.7)	3p : n.d.	4m/4m' : 65% (11.8:1)	4n/4n' : trace	4o/4o' : 80% (18.4:1)	4p/4p' : 49% (13.3:1) ^c
3q : n.d.	3r : 56%	3s/3s' : 57% (11.5:1)	3t : 38%	4q/4q' : 60% (12.8:1)	4r : 23%	4s : 65%	4t/4t' : 25% (13.6:1)

^aIsolated Yield of the all isomers. Unless otherwise noted, selectivity > 20:1; ^bAccompanied by small amount of inseparable alkyne **1**, the yield of product has been adjusted accordingly.

^cIsolated product together with 1-phenyl-1,3-butadiene in a 4.0:1 ratio, the yield of product has been adjusted accordingly.

Next, we aimed to selectively prepare the 1,4-diene **4a**, which is the minor isomer that results from direct coupling of alkyne **1a** and allylboron **2a**. To alter regioselectivity, we pursued a co-catalyst that would accelerate transmetalation between the allyl species and the vinyl-Pd intermediate, therefore enable the synthesis of 1,4-dienes. An evaluation of co-catalysts revealed that Cu(OAc)₂ promoted the formation of **4a/4a'** (Chart 1C). Moreover, in contrast to the monophosphine ligands that promotes the formation of **3a**, bisphosphine ligands such as dppe and dpp-benz gave higher yield of **4a/4a'**. When dppe was used as ligand, **4a/4a'** was delivered in 58% combined yield as a 10:1 mixture of regioisomers. Using methanol instead of butanol as the proton source improved the yield of **4a** to 79%. In the absence of palladium, no products were detected, which indicates this transformation is not catalyzed by copper alone.

As shown in Table 1A, we obtained various 1,5-dienes in moderate to good yields and high selectivities (>20:1, **3** vs **4**). Substrates with electron-donating groups proceed smoothly to deliver 1,5-dienes **3b-3d**. Fluoro and chloro groups are also well tolerated, yielding products **3e** and **3f** in 74% and 78%. Aryl alkynes bearing electron-withdrawing substituents (CF₃, Ph, Ac, and CO₂Me) show slightly higher reactivities, providing the linear products (**3g-3j**) in high yields (81-90%). High selectivities are still obtained for *meta*-substituted 1-aryllalkyne **1k-1m**. The steric hindrance of alkyne influences the reactivity and offers **3n** with 52% yield. On replacing the phenyl group



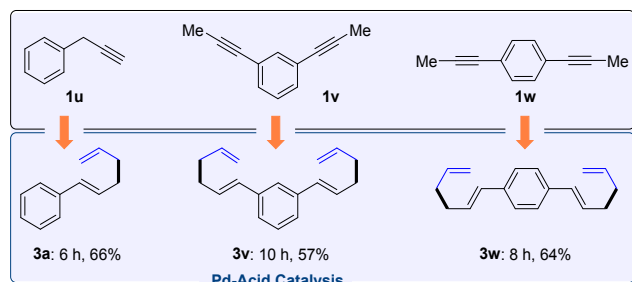
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with 2-naphthalenyl groups, the substrate transforms into a mixture of linear product **3o** and branched product **3o'** in a 1:1.7 ratio. The branched product **3o'** probably originates from the formation η^3 - π -benzyl-palladium intermediate.¹⁵ No desired 1,5-diene product is obtained when alkyl-substituted alkyne **1p** or **1q** is subjected to standard conditions. Instead, isomerization product 1-phenyl-1,3-butadiene is observed for the conversion of **1p**. Notably, substrates bearing pyridine rings, which were incompatible in previously reported palladium catalysis,¹⁶ also lead to **3r** and **3s** in moderate yields. Finally, the late-stage modification of the estrone derivative **1t** affords **3t** in 38% yield. This 1,5-diene synthesis complements known allyl-allyl couplings that require pre-functionalized allyl precursors such as allyl chlorides or carbonates.¹⁶⁻¹⁷

Then, we examined the substrate scope for the synthesis of 1,4-dienes (Table 1B). Various alkynes transform to 1,4-dienes using the Pd/Cu catalyst combo. Although substrates bearing electron-donating groups lead to skipped dienes in moderate to good regioselectivities (**4b-4d**, **4k**, **4m**), electron-withdrawing substrates perform well in terms of yields and selectivities (**4e-4j**, **4l**). Ortho-Substituted alkyne **1n** exhibits no reactivity due to steric hindrance. It is noteworthy that **4p** and **4p'** are successfully acquired in 49% yield, while accompanied by a small amount of isomerization side product (1-phenyl-1,3-butadiene). Comparatively, the cross-coupling between 1-phenyl-1-hexyne **1q** and **2a** provides 1,4-diene products **4q/4q'** without any 1,3-diene side product. Heterocyclic substituted alkynes (**1r** and **1s**) and estrone derivative all successfully deliver the 1,4-diene products.

Besides internal alkynes, terminal alkyne **1u** couples with allylB(pin) **2a** to yield 1,5-diene **3a** (Scheme 2). The bis-allylations of di(prop-1-yn-1-yl)benzene **1v** and **1w** proceed smoothly with high selectivities and moderate yields. These olefin products are potential monomers for polymerization.¹⁸

We also tested the scope of allylborons under our Pd-acid condition (Table 2). Generally, substrates **2b-2d** were less reactive in allyl-allyl couplings. This agrees with previous work



Scheme 2. Hydroallylation of terminal and bis-alkynes

Table 2. The scope of allylborons

1 + 2 (2 equiv.)		product		yield (%)	
				w/o Cu(OAc) ₂	Cu(OAc) ₂
	2b		3x	6	34 ^a
	2c		3y	n.d.	< 5 ^a
	2d		3y	13	38 ^a

^aCu(OAc)₂ (10 mol%)

reported by Morken's group that substituted allylborons were comparatively reluctant in Pd-catalyzed allyl-allyl coupling reactions.^{17g} To improve the reactivity, Cu(OAc)₂ was employed as additional promoter to facilitate transmetalation. The yields of **3x** and **3y** were successfully increased to 34% and 38%, respectively. It should be noted that these reactions all give interesting linear-branched coupling products, this selectivity is rare in Suzuki-type allyl-allyl coupling reactions.¹⁷

While further studies are warranted, we propose the following mechanisms on the basis of literature⁷ and our own observations (Figure 1A, right). First, the oxidative addition of the carboxylic acid with a Pd(0) precursor generates Pd(II)-hydride species **A**. *Syn*-migratory insertion of alkyne **1** into the Pd(II)-H **A** affords vinyl-Pd intermediate **B**. For 1,5-dienes pathway, a vacant coordination site of complex **B** is spared for β -hydride elimination in the presence of monophosphine ligand. Allene **5** is subsequently produced and undergoes reinsertion to Pd(II)-H forming the π -allyl-Pd intermediate **C**. Then, electrophilic intermediate **C** reacts with allylB(pin) **2** to deliver bis(allyl)Pd species **D**.¹⁹ Reductive elimination yields the allyl-allyl coupling product **3a** and turns over Pd(0) catalyst (allyl-allyl coupling cycle).

In the presence of a Cu(II) co-catalyst, we propose that transmetalation of allylboron **2a** to generate allylcopper species **2'** is favored.^{5i,17k} Under this protocol, vinyl-Pd species **B** is also generated by a *syn*-migratory of alkyne **1** into the Pd(II)-H **A** (Figure 1A, left). However, when coordinated with a bidentate ligand, vinyl-Pd intermediate **B** prefers direct transmetalation with allylcopper species **2'** to form **D'** rather than undergoing β -hydride elimination to produce allene **5**. Reductive elimination from **D'** yields product 1,4-diene **4a** (allyl-vinyl coupling cycle).



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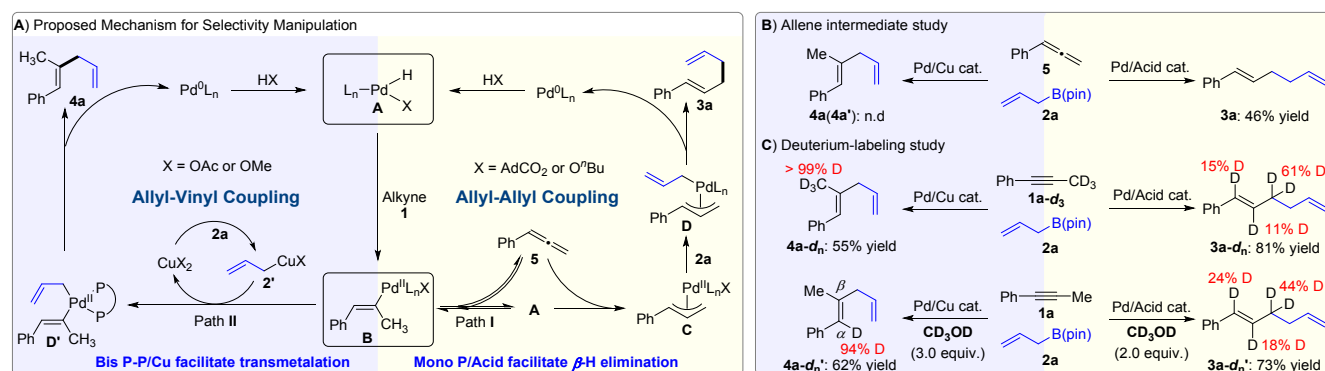


Figure 1. Divergent reactivity of Pd(II) species

To probe the feasibility of an allene intermediate, phenylallene **5** was subjected to couple with allyl-B(pin) **2a** under two standard conditions (Figure 1B, see the SI for details). The allene was transformed into the 1,5-diene **3a** in 46% yield. No formation of **4a** or **4a'** under the Pd-Cu conditions supports that the 1,4-diene products does not arise from the addition of allylB(pin) **2a** to allene **5**. The hydroallylation was also performed with deuterated alkyne **1a-d₃** or methanol (Figure 1C). Under Pd/acid catalysis, we found the deuterium label was scrambled into the α -, β -, and γ -positions of 1,5-diene **3a-d_n** using **1a-d₃**. Similar deuterium scrambling was observed when conducting reaction with deuterated methanol as proton source. This observation supports a reversible hydrometallation of the internal π -system of the allene in the synthesis of 1,5-dienes. When experiments were carried out under Pd-Cu catalysis, the deuterium label remained intact in 1,4-diene **4a-d_n** with deuterated alkyne **1a-d₃**. Only α -deuterated product was achieved using deuterated methanol (Figure 1C). This indicates β -hydride elimination is not involved under Pd/Cu catalysis.

Conclusions

Our work complements other alkyne hydroallylation methods for the synthesis of 1,4-dienes including those developed by Hilt, Hartwig, Lalic and Zhang.⁵ Key to the success of this method is the switchable reactivity of vinyl-Pd intermediates. Acid additive promotes the β -hydride elimination pathway for allyl-allyl coupling with the aid of a monophosphine ligand. Whereas Cu co-catalyst facilitates the direct transmetalation for vinyl-allyl coupling in the presence of a bisphosphine ligand. Transmetalation and β -hydride elimination are two elementary steps featured in many well-known organometallic mechanisms, including the Suzuki-Miyaura and Heck cross-coupling. Insights from this study will guide the future development of related regiodivergent methods in catalysis.

Conflicts of interest

There are no conflicts to declare.

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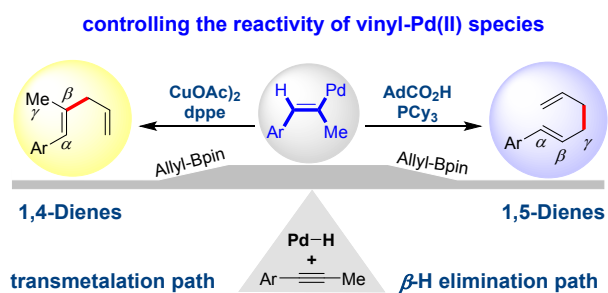
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Graphical Abstracts (Table of Contents Entry)

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Through rational evaluation of ligands and promoters, the reactivity of a key Pd(II) species towards transmetalation or β -H elimination is manipulated.

