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NosL is a dedicated copper chaperone for assembly of the Cu_Z center of nitrous oxide reductase†

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Nitrous oxide reductase (N₂OR) is the terminal enzyme of the denitrification pathway of soil bacteria that reduces the greenhouse gas nitrous oxide (N₂O) to dinitrogen. In addition to a binuclear Cu_A site that functions in electron transfer, the active site of N₂OR features a unique tetranuclear copper cluster bridged by inorganic sulfide, termed Cu_Z. In copper-limited environments, N₂OR fails to function, resulting in truncation of denitrification and rising levels of N₂O released by cells to the atmosphere, presenting a major environmental challenge. Here we report studies of *nosL* from *Paracoccus denitrificans*, which is part of the *nos* gene cluster, and encodes a putative copper binding protein. A *Paracoccus denitrificans* $\Delta nosL$ mutant strain had no denitrification phenotype under copper-sufficient conditions but failed to reduce N₂O under copper-limited conditions. N₂OR isolated from $\Delta nosL$ cells was found to be deficient in copper and to exhibit attenuated activity. UV-visible absorbance spectroscopy revealed that bands due to the Cu_A center were unaffected, while those corresponding to the Cu_Z center were significantly reduced in intensity. *In vitro* studies of a soluble form of NosL without its predicted membrane anchor showed that it binds one Cu(I) ion per protein with attomolar affinity, but does not bind Cu(II). Together, the data demonstrate that NosL is a copper-binding protein specifically required for assembly of the Cu_Z center of N₂OR, and thus represents the first characterised assembly factor for the Cu_Z active site of this key environmental enzyme, which is globally responsible for the destruction of a potent greenhouse gas.

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Introduction

Nitrous oxide (N₂O) is a significant greenhouse gas with a ~300 fold greater global warming potential than CO₂ and an ability to deplete stratospheric ozone.¹ Agriculture produces 65% (6.8 Tg N-N₂O per year) of the total N₂O emitted each year. The main contributor within this sector is the soil microbial community, which produces 40% of these emissions.² The surge in atmospheric N₂O from 270 ppb to 324 ppb over the last 100 years correlates strongly with the use of anthropogenic nitrogen-based fertilisers in farming to improve crop yield.³ The doubling of available nitrogen in the environment has enriched a class of soil dwelling microorganisms called denitrifiers,

which respire anaerobically by reducing nitrate to dinitrogen (N₂) gas *via* the free intermediates nitrite, nitric oxide and N₂O using different metallo-enzymes. Environmental factors such as soil pH, moisture, carbon to nitrogen ratio, temperature and a lack of copper⁴⁻⁶ have all been identified as factors leading to increased N₂O emissions from these microbes.

Encoded by the *nosZ* gene, the cupro-enzyme nitrous oxide reductase (N₂OR or NosZ) catalyses the 2-electron reduction of N₂O to N₂. Two distinct and approximately equally abundant clades of N₂OR-containing bacteria and archaea have been identified.⁷ Importantly, clade II members act as an N₂O sink, while members of clade I, such as α -, β - and γ -proteobacteria, are able to produce and remove N₂O under optimum conditions. In order to assist with future strategies for the control of emissions from soil ecosystems, a key task is to explore how the enzyme is produced and matured in N₂O emitting bacteria.³

N₂OR in the denitrifying α -proteobacterium *Paracoccus denitrificans* (PdN₂OR) is exported to the periplasm *via* the twin-arginine translocation (TAT) pathway,^{8,9} where its two copper centers, Cu_A and Cu_Z, are assembled. Several crystal structures are now available for N₂OR,¹⁰⁻¹³ revealing that the Cu_A site is housed in a C-terminal cupredoxin domain, while the Cu_Z site lies within the N-terminal seven bladed β -propeller domain.

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The Cu_A center contains two copper ions that are bridged by two conserved Cys residues and further coordinated by Met, His and Trp ligands to form a site that closely resembles the electron transfer Cu_A site present in subunit II of cytochrome *c* oxidase.¹⁴ In N₂OR the Cu_A site also acts as an electron shuttle, accepting electrons from small electron donors such as cytochrome *c*₅₅₀ (ref. 15) or pseudoazurin, in *P. denitrificans*,¹⁶ for the reduction of N₂O at the Cu_Z site. This center comprises four copper atoms coordinated by seven conserved His residues and bridged by one ([4Cu:S]) or two ([4Cu:2S]) sulfides, depending on the presence or absence of O₂, respectively, during purification.^{10,12}

A major challenge that is particularly important for addressing N₂O emissions from soil is to understand how the cofactor sites of N₂OR are assembled and, in doing so, identify assembly/chaperone systems that are involved. For clade I organisms, including *P. denitrificans*, any such systems must be periplasmic, as mutations in the TAT leader sequence of N₂OR results in the protein remaining in the cytoplasm, in a folded but copper-free state.¹⁷ Studies of the biogenesis of copper cofactor sites of eukaryotic cytochrome *c* oxidase have identified the proteins Cox17 and ScoB, which are involved in the assembly of the Cu_A site,¹⁸ and prokaryotic homologues PCu_AC and SenC have been implicated in the maturation of the Cu_A site of the aa₃-type cytochrome *c* oxidase from *Rhodobacter sphaeroides*.¹⁹

In comparison, little is known about the assembly of the Cu_Z site. The *nos* gene cluster of clade I N₂O reducing bacteria vary between denitrifying phyla but contain, in addition to *nosZ*, five other core genes (*nosRZDFYL*), while predominantly α - and β -proteobacteria members also contain a further two genes (*nosC* and *nosX*). *nosDFY* are predicted to encode an ABC-type transporter homologous to ATM1 from eukaryotes, which transports a sulfur-containing species out of the mitochondrion for Fe-S cluster assembly in the cytoplasm,²⁰ suggesting that these proteins are likely involved in supplying sulfide for assembly of the Cu_Z center. In *P. denitrificans*, downstream of *nosDFY* are two further accessory genes that are part of the same operon, *nosL* and *nosX*. The only mutation analysis of *nosL* has been in *P. stutzeri*⁹ and of *nosX* in *P. denitrificans*,²¹ both of which led to the conclusion they alone are not important for whole-cell N₂O reduction. However, *nosL* is part of the denitrification core gene cluster, its expression is responsive to cellular copper status,²² and the NosL protein has been shown to bind copper,²³ suggesting that it plays a role in maturation or activation of N₂OR.

Here, we report genetic and biochemical studies of *P. denitrificans nosL/NosL*. The data demonstrate that NosL is a Cu(i)-binding protein that is required for efficient assembly of the N₂OR Cu_Z center, and thus represents the first characterised assembly factor for this unique metal center in biology.

Results

NosL is required for N₂OR activity under copper-limited conditions

Wild type PD1222 (WT), $\Delta nosZ$ (lacking the gene encoding N₂OR) and $\Delta nosL$ strains (Table S1†) were cultured under both Cu-sufficient and Cu-limited anaerobic conditions and their

growth characteristics investigated. For the WT strain, growth yield was affected under Cu-limited conditions (Fig. 1A). For the $\Delta nosZ$ strain, growth rate and yield were affected under both conditions, consistent with the importance of N₂OR for optimal growth under anaerobic nitrate-sufficient conditions (Fig. 1B). Growth of the $\Delta nosL$ strain was similar to WT under Cu-sufficient conditions, but was attenuated under Cu-limited conditions, exhibiting a maximum OD_{600 nm} similar to the $\Delta nosZ$ strain (Fig. 1C).

The impact of *nosL* deletion on N₂OR activity *in vivo* was assessed by measuring N₂O levels in the headspace of cultures. Consistent with previous studies, no N₂O was accumulated by the WT strain under Cu-sufficient conditions and only a transient low level of N₂O (<1 mM at ~16 h) was observed under Cu-limited conditions (Fig. 1D), where the latter is likely due to lower transcription of *nosZ* under these conditions.²² In contrast, the $\Delta nosL$ mutant exhibited a marked Nos-negative phenotype (Nos⁻) regardless of copper levels, with N₂O emitted from cultures to >4 mM after 24 h (Fig. 1E). Like the WT, essentially no N₂O was emitted from $\Delta nosL$ cultures under Cu-sufficient conditions. Importantly, however, after 24 h of growth and once the cells had reached stationary phase, up to 4 mM N₂O was detected in the headspace of $\Delta nosL$ cultures under Cu-limited conditions, similar to the $\Delta nosZ$ strain (Fig. 1F).

The Nos⁻ phenotype of the $\Delta nosL$ strain under Cu-limited conditions was almost fully complemented by expression of a plasmid-borne functional *nosL* gene copy from a taurine inducible promoter (Fig. S1†). In particular, the extent of N₂O release from the complemented $\Delta nosL$ mutant closely resembled the WT strain in the transient accumulation of N₂O at ~16–20 h.

Absence of NosL results in a copper- and catalytically-deficient N₂OR

To investigate the functional properties of NosL in relation to N₂OR biogenesis, N₂OR (NosZ) with a C-terminal Strep-II tag was overproduced in *P. denitrificans* strains grown under Cu-sufficient conditions and purified. N₂OR from $\Delta nosL$ cells contained on average 4 Cu atoms per monomer, compared to ~6 for the enzyme from $\Delta nosZ$ cells (Table 1). N₂OR from a $\Delta nosZL$ double mutant contained ~3.5 copper atoms per monomer, demonstrating that the presence of the chromosomal *nosZ* gene in the $\Delta nosL$ cells had little effect on the copper content of the tagged N₂OR (NosZ) (Table 1).

As purification of N₂OR was carried out aerobically, the enzyme from $\Delta nosZ$ cells contained the Cu_Z^{*} (pink, form II) form of the active site, which is catalytically inactive.²⁴ To activate the isolated N₂OR to the fully reduced form, the enzyme was incubated with excess reduced methyl viologen at room temperature^{25,26} for 150 min, at which point N₂O reductase activity had reached a maximum. N₂OR from $\Delta nosL$ and $\Delta nosZL$ cells exhibited significantly lower maximum activities than N₂OR from $\Delta nosZ$, see Table 1.

Absence of NosL results in N₂OR deficient in the Cu_Z center

The UV-visible absorbance spectrum of as isolated air-oxidized (pink, form II) strep-tagged N₂OR purified from $\Delta nosZ$ *P.*



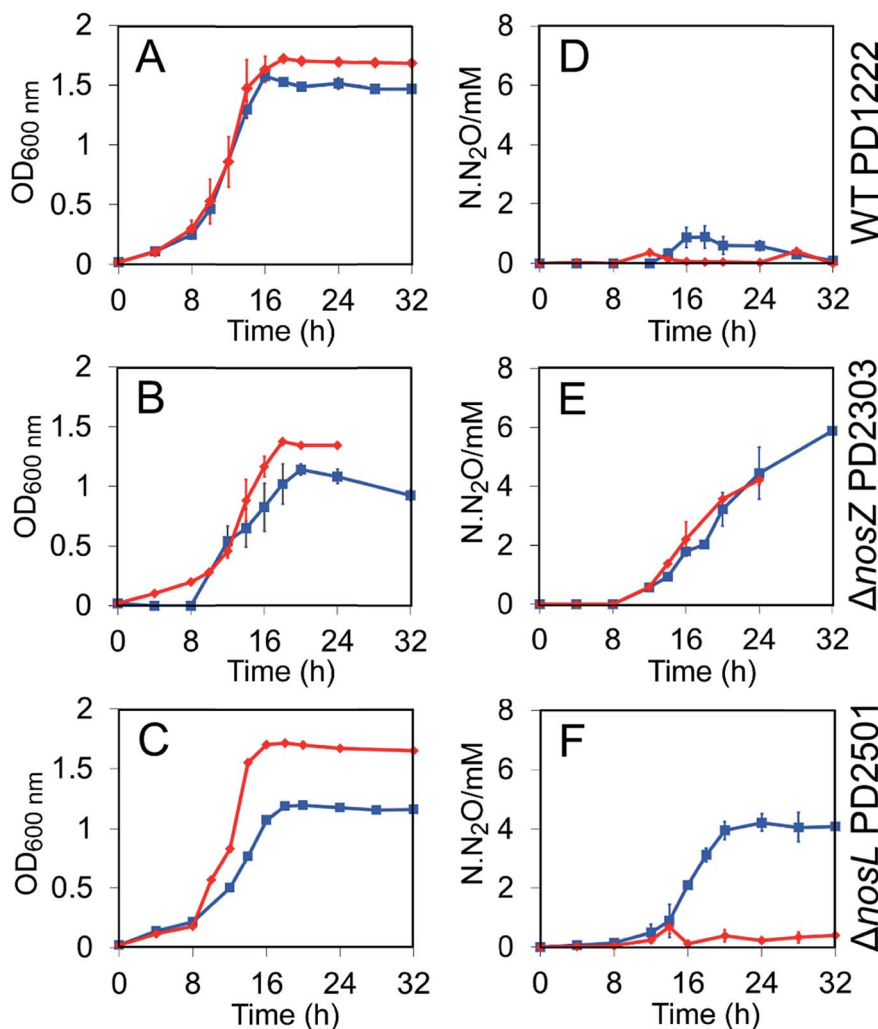


Fig. 1 Growth and N_2O emission profiles for *P. denitrificans* WT and mutant strain cultures. (A)–(C) $OD_{600\text{ nm}}$ as a function of time for WT PD1222 (A), $\Delta nosZ$ deletion mutant PD2303 (B), and $\Delta nosL$ deletion mutant PD2501 (C). (D)–(F) N_2O emissions as $N \cdot N_2O$ (millimolar N in the form of N_2O) for PD1222 (D), PD2303 (E), and PD2501 (F). Strains were grown in anaerobic batch culture conditions in Cu-sufficient media (red diamonds) and Cu-limited media (blue squares). Bars represent SE.

denitrificans displayed bands at 480, 535 and 645 nm (Fig. 2A and $S2^\dagger$), in agreement with literature for N_2OR enzymes from a range of bacteria.^{11,26–29} These features arise from S^{2-} to Cu(II) charge-transfer transitions and transitions due to interactions between Cu(I) and Cu(II) ions.²⁷ The bands at 480 and 550 nm

correspond to the Cu_A centre, while that at 645 nm is characteristic of the Cu_Z centre (in its Cu_Z^* form).²⁹

The spectra of N_2OR enzymes purified from $\Delta nosL$ and $\Delta nosZL$ strains were very similar in the 450–550 nm region, but exhibited significantly reduced intensity beyond 550 nm

Table 1 Copper content and activities of N_2OR enzymes from different *P. denitrificans* strains

Strain	Cu-sufficient		Cu-limited
	Cu/monomer ^a	Maximum specific activity ^b ($\mu\text{mol } N_2O \text{ min}^{-1} \text{ mg}^{-1}$)	Cu/monomer
$\Delta nosL/pMSL002$ (StrepII tagged- N_2OR)	4.1 ± 0.43	95.7 ± 3.9	N.D.
$\Delta nosZ/pMSL002$	5.9 ± 0.56	196.2 ± 9	4.8 ± 0.42
$\Delta nosZL/pMSL002$	3.4 ± 0.26	93.6 ± 0.8	N.D.

^a Total copper contents per monomer were determined using a BCS copper assay. ^b N_2O reductase activity was determined using a reduced methyl viologen assay ($\mu\text{mol } N_2O \text{ min}^{-1} \text{ mg}^{-1}$ enzyme). Proteins were pre-incubated with a 500-fold excess reduced methyl viologen for 150 min prior to activity assay. All reactions were carried out in triplicate and SD is shown. N.D., not detectable.



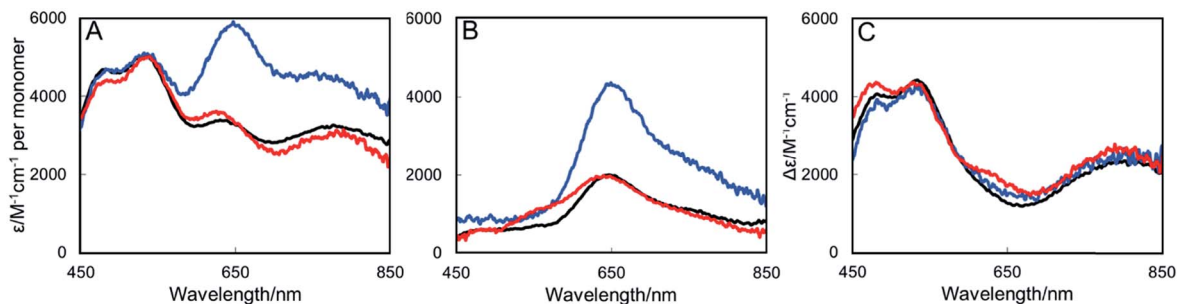


Fig. 2 UV-Visible absorbance spectra of N_2OR purified from Cu-sufficient *P. denitrificans* mutant strains. (A) N_2OR isolated from $\Delta nosZ$ PD2303 (blue line), $\Delta nosL$ PD2501 (black line) and $\Delta nosZL$ PD2505 (red line) following oxidation using ferricyanide. (B) As in (A) but following reduction using sodium dithionite. (C) Oxidized minus reduced difference spectrum. Oxidized spectra were normalised to $\epsilon_{580\text{ nm}} 5000\text{ M}^{-1}\text{ cm}^{-1}$ per monomer as described by Rasmussen *et al.*²⁷ Sample buffer contained 20 mM HEPES, 150 mM NaCl, pH 7.2.

(Fig. 2A). The apparent absorbance maximum was shifted to $\sim 635\text{ nm}$, consistent with the enhanced relative influence of the underlying absorbance due to Cu_A (maximum at 535 nm). Reduction with sodium dithionite resulted in the spectra shown in Fig. 2B. Bands at 480 and 540 nm were lost, consistent with the reduction of the Cu_A center to its colorless diamagnetic $Cu(I)/Cu(I)$ state. The remaining band is characteristic of the Cu_Z center following addition of dithionite.²⁷ N_2OR from both $\Delta nosL$ and $\Delta nosZL$ strains exhibited a much less intense Cu_Z absorbance than that from $\Delta nosZ$, indicating diverse occupancies of the center. Ferricyanide-oxidized minus dithionite-reduced difference spectra (Fig. 2C) closely overlay, particularly in the $450\text{--}550\text{ nm}$ region, demonstrating that the N_2OR Cu_A centers of the different enzymes are essentially identical, and close to fully populated, as estimated by the measured extinction coefficients.²⁷ Together, these data demonstrate that N_2OR isolated from a $\Delta nosL$ background is specifically deficient in its Cu_Z center.

Absence of NosL under copper-limited growth conditions results in complete absence of Cu_A and Cu_Z centers in N_2OR

The copper determinations, activity assays and spectroscopic characterizations of N_2OR described above were performed with samples isolated from cultures grown in copper-sufficient media, in which only a minor growth phenotype for the $\Delta nosL$ strain was observed (Fig. 1). Thus, N_2OR characteristics under low copper, where N_2O is generated from cultures, were investigated. N_2OR from $\Delta nosZ$ cells contained ~ 5 Cu per monomer (Table 1) with an absorbance spectrum indicative of complete, or near complete, Cu_A center population, but a less than stoichiometric population of Cu_Z (Fig. 3). In contrast, N_2OR from $\Delta nosL$ and $\Delta nosZL$ cells contained no detectable copper (Table 1), and gave UV-visible absorbance spectra with no absorbance in the visible region (Fig. 3), indicating complete failure to assemble either of the copper centers of N_2OR in the absence of NosL.

NosL binds Cu(I) with attomolar affinity

NosL contains a Type-II signal peptidase recognition sequence that, when cleaved, produces a protein with an N-terminal Cys

residue that is predicted to bind lipid and anchor NosL into the outer membrane.^{23,30} The NMR solution structure of NosL, lacking its membrane anchor sequence, from the β -proteobacterium *Achromobacter cycloclastes* revealed two independent homologous domains with an unusual $\beta\beta\alpha\beta$ topology.³¹ The same authors showed that the protein binds Cu(I) specifically and XAFS data were consistent with a Cu(I) coordination consisting of S and N/O ligands.²³ To determine the biochemical/biophysical properties of *P. denitrificans* NosL, the protein lacking its periplasmic export signal sequence and its predicted N-terminal Cys residue²³ was purified resulting in a metal-free form of the protein, which gave a mass of $18\,890\text{ Da}$ (predicted mass $18\,891\text{ Da}$) by LC-ESI-MS (Fig. S3[†]). The final, gel filtration step of purification resulted in a broad elution band that suggested a mixture of monomer/dimer association states for NosL, a result confirmed by native PAGE (Fig. S4[†]), which showed two species of NosL. Similar observations were made for apo-NosL from *A. cycloclastes*.²³

Titration of apo-NosL with Cu(I) resulted in the series of spectra shown in Fig. 4A, in which broad absorbance in the near-UV region of the spectrum was observed to gradually increase and saturate at a level of 1 Cu(I) per NosL (Fig. 4B). The absorbance is characteristic of charge transfer transitions involving Cu(I) coordinated to a cysteine thiolate.³² Cu(I)-binding was also investigated by CD spectroscopy, which confirmed the tight association of one Cu(I) per protein but also suggested that further Cu(I) can associate with NosL, albeit weakly (Fig. S5[†]). Gel filtration and native PAGE analysis of Cu(I)-NosL (Fig. S4[†]) also demonstrated that Cu(I) binding does not significantly affect the association state equilibrium.

Equivalent titrations with Cu(II) followed by absorbance spectroscopy resulted in spectra very similar to those for Cu(I) (Fig. S6[†]), suggesting that Cu(II) may undergo auto-reduction upon binding to NosL. This possibility was investigated using EPR (Fig. S7 and Table S2[†]). Addition of Cu(II) to NosL resulted in a characteristic $S = 1/2$ Cu(II) signal that, by comparison with a Cu(II) standard, corresponded to only $\sim 8\%$ of the Cu(II) initially added. The same sample in the presence of EDTA (a Cu(II) chelator), which prevents Cu(II) ions in solution dimerizing to form EPR-inactive species, resulted in a Cu(II) concentration of $\sim 11\%$. Similar experiments but with the addition of



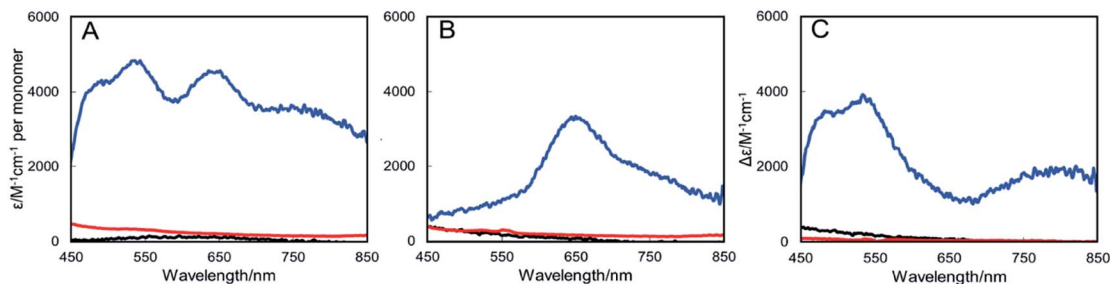


Fig. 3 UV-Visible absorbance spectra of N_2OR purified from Cu-limited *P. denitrificans* mutant strains. (A) N_2OR isolated from $\Delta nosZ$ PD2303 (blue line), $\Delta nosL$ PD2501 (black line) and $\Delta nosZL$ PD2505 (red line) following oxidation using ferricyanide. (B) As in (A) but following reduction with sodium dithionite. (C). Oxidised minus reduced difference spectra. Spectra were normalised to the protein concentration. Sample buffer contained 20 mM HEPES, 150 mM NaCl, pH 7.2.

Cu(I) in place of Cu(II) resulted in 4% and 8% in the absence and presence of EDTA, respectively (Table S2[†]). As isolated NosL contained <1% Cu(II). Together, these data indicate that Cu(II) undergoes auto-reduction to EPR silent Cu(I) upon binding to NosL.

To further characterise Cu(I)-binding to NosL, ESI-MS under non-denaturing conditions, where non-covalent interactions are preserved, was also employed. Fig. 5 shows the deconvoluted spectrum of apo-NosL with the major peak at 18 890 Da (as observed by LC-ESI-MS, Fig. S3[†]), along with a number of lower intensity peaks to the higher mass side, due to non-covalent sodium and ammonium adducts. Attempts to remove these adducts from the non-denaturing MS, *via* buffer exchange, changes in pH and ionic strength, were unsuccessful. NosL was loaded with a single Cu(I) ion and the peak envelope of the resulting deconvoluted mass spectrum was at +63 Da relative to that of the apo-NosL spectrum (Fig. 5), consistent with the binding of a single Cu(I) ion.

Competition binding experiments using the high-affinity chelator BCS were used to determine the dissociation constant for Cu(I)-binding to NosL. Cu(I)-NosL was titrated with BCS, and the partition of Cu(I) between NosL and BCS determined from measured $A_{483\text{ nm}}$, due to $[Cu(BCS_2)_2]^{3-}$, together with the well-characterised formation constant for $Cu(BCS_2)_2^{3-}$, $\beta_2 = 10^{19.8}$ (Fig. S8[†])³³. From these, an average K_d value of $\sim 4 \times 10^{-18}$ M was determined³⁴ (Table S3[†]), demonstrating very tight binding of Cu(I) to NosL.

Discussion

Despite *nosL* being a core component of the *nos* gene clusters in a range of microorganisms, mutational studies⁹ have so far failed to reveal a function in N_2O reduction. A *nosL* mutant of *Pseudomonas stutzeri* (containing an insertion towards the 3' end of the gene) exhibited a slightly lower growth rate but produced active, holo- N_2OR .⁹ The presence of a CXXC motif in the encoded protein prompted the suggestion that NosL might be a protein disulfide isomerase, but analysis of the sequence of NosL from *Sinorhizobium meliloti*³⁵ showed that one of the Cys residues of the *P. stutzeri* protein is not conserved.

Here, a *nosL* deletion mutant of *P. denitrificans* was generated and compared to the WT and *nosZ* deletion strains, as growth and N_2O benchmarks, under both Cu-limited and Cu-sufficient conditions. The *nosL* mutant exhibited a Cu-limited growth phenotype relative to WT, associated with a deficiency in the activity of N_2OR , leading to accumulation of N_2O in the culture headspace. The phenotype was complemented *in trans* by *nosL* and also under copper replete conditions, strongly suggesting that NosL functions in an aspect of copper metabolism. The reason why the previously reported *P. stutzeri nosL* mutant did not present a phenotype is unclear; one possibility is that there was sufficient copper in the growth medium to mask the phenotype, but growth conditions were not reported.⁹

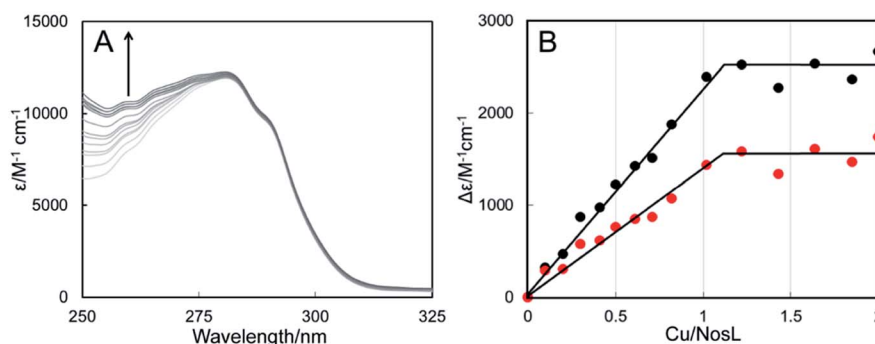


Fig. 4 Copper-binding to NosL. (A) UV-visible absorbance titration of NosL (21 μM) with Cu(I). (B) Plots of $\Delta\epsilon_{260\text{ nm}}$ (black circles) and $\Delta\epsilon_{265\text{ nm}}$ (red circles) as a function of Cu(I) ions per protein molecule. NosL was in 100 mM MOPS, 100 mM NaCl, pH 7.5.



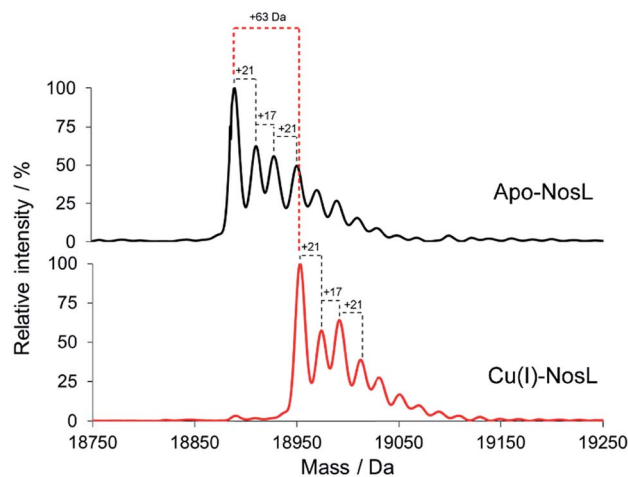


Fig. 5 Deconvoluted ESI-MS of NosL (20 μ M) in the apo-form and containing 1 Cu(I) per protein, as indicated, in 50 mM ammonium acetate, pH 7.8.

Strep-tagged N₂OR enzymes purified from $\Delta nosZ$, $\Delta nosL$ and $\Delta nosZL$ mutants exhibited clear differences. Under Cu-sufficient conditions, N₂OR from $\Delta nosL$ cells contained only ~ 4 Cu per N₂OR monomer and was substantially less active (though it was apparently sufficiently active *in vivo* to mask any obvious *in vivo* phenotype). Importantly, spectroscopic characterisation revealed that the N₂OR protein from $\Delta nosL$ was specifically deficient in its Cu_Z center, demonstrating that NosL functions in the assembly of this unique biological metal center required for N₂O destruction.

Under Cu-limited conditions, tagged N₂OR purified from the $\Delta nosZ$ mutant contained nearly 5 Cu/monomer, while those from $\Delta nosL$ or $\Delta nosZL$ mutant strains lacked copper entirely, indicating complete failure to assemble either N₂OR copper center in the absence of NosL. Thus, although the Cu_A can be reconstituted using copper alone,³⁶ when copper is limited, NosL may also supply Cu for incorporation into Cu_A.

The N-terminal sequence of NosL contains a periplasmic export signal and a lipobox that is predicted to be processed by Lgt, Lsp and Lnt enzymes, resulting in a mature protein that is membrane-anchored *via* a triacylated N-terminal Cys residue (indicated by a yellow arrowhead in Fig. S9†).³⁷ This is likely a substrate of the Lol system,³⁷ such that it is located in the outer membrane, with its soluble domains facing the periplasm. Studies of a soluble form of NosL, lacking its periplasmic targeting sequence and the N-terminal Cys residue that is acylated, revealed that it binds Cu(I) with attomolar affinity, a characteristic of many copper chaperones.

Our data are consistent with Cys coordination, in agreement with previous studies of *A. cycloclastes* NosL using thiol-specific reagents and EXAFS, with data from the latter consistent with a three or four coordinate Cu(I) center. The best fit was obtained for three coordinate Cu(I) with (O/N)₂S₂ ligands.²³ Alignment of NosL sequences from the two clades of N₂O reducing bacteria identifies two conserved Cys residues: one through which the protein is believed to be anchored to the membrane (see, Fig. S9,†

yellow arrowhead); and, one within a CXM motif that is likely to participate in Cu(I)-binding.²³ The Met residue of this motif is also strictly conserved and was proposed to provide the second sulfur ligand identified by EXAFS.²³ There is one other absolutely conserved Met residue but, as acknowledged by McGuirl *et al.*, there is no absolutely conserved His residue (Fig. S9†).

Separation of NosL proteins according to the clade to which the organism belongs may shed some further light on this, as it reveals residues that are conserved within, but not between, clades. Clade I NosL proteins, which include *P. denitrificans* and *A. cycloclastes*, contain a well-conserved His residue close to the CXM motif, resulting in a CXMX₃H motif. Clade II NosL proteins do not contain this His but instead contain a conserved Cys residue at the N-terminal side of the CXM motif (resulting in a CXXCXM motif). *P. stutzeri* NosL is the exception to this, as it comes from a Clade I organism but is more similar to Clade II NosL proteins.

In summary, we identify here the first Cu chaperone with a specific role in Cu_Z center assembly. Our data indicate that NosL, which binds Cu(I) with attomolar affinity, is a key part of a high-affinity Cu trafficking pathway that enables assembly of the Cu_Z center of N₂OR. The pathway functions under Cu-sufficient conditions, but becomes essential under Cu-limited conditions. During copper limitation, the NosL pathway may also play a key role in supplying Cu for the Cu_A center. How NosL delivers Cu to N₂OR is unknown; it is likely that NosL functions directly in the transfer of Cu(I) to N₂OR, and is thus a copper chaperone, but further studies are now needed to investigate this. The identification of NosL as a key component of a Cu-trafficking pathway for the assembly of the active holo-reductase N₂OR, the sole pivotal enzyme for N₂O destruction, is a significant advance towards the long-term aim of mitigating microbial emissions of N₂O into the atmosphere.

Materials and methods

Construction of mutant *nosL* and *nosZL* deficient strains of *P. denitrificans*

Unmarked deletions of *nosL* in *P. denitrificans* wild type or $\Delta nosZ$ backgrounds (Table S1†) were produced by the method of allelic replacement.²² Briefly, regions flanking *nosL* (Table S4†) were cloned into pK18*mobsacB* using *EcoRI* and *PstI* sites to generate a suicide plasmid (pSPBN1, Table S1†) that was subsequently conjugated into *P. denitrificans* PD1222 or $\Delta nosZ$ PD2303 using the helper *E. coli* pRK2013 strain. Single cross-over recombination events were screened using spectinomycin (25 μ g ml⁻¹) and kanamycin (50 μ g ml⁻¹). Primary Spec^R/Km^R trans-conjugants were grown to stationary phase in Luria Bertani broth (LB) with no antibiotic. Double cross over events were selected for using a high-salt modified LB agar supplemented with 6% (w/v) sucrose. Sucrose resistant colonies with Spec^R were screened using colony PCR and gene deletion confirmed by sequencing. Deletion strains were named PD2501 ($\Delta nosL$) and PD2505 ($\Delta nosZL$). For complementation of $\Delta nosL$ cells the coding sequence of Pden_4215 was synthesised (Genscript) with flanking 5' *NdeI* and 3' *EcoRI* restriction sites and sub-cloned into a taurine-inducible modified pLMB509 plasmid with



gentamicin resistance ($20 \mu\text{g ml}^{-1}$) to generate pSPBN2. The complementation plasmid was conjugated into the mutant strain using the helper *E. coli* pRK2013 strain and successful conjugants were Spec^R/Gm^R. Expression of *nosL* from the plasmid was induced by adding 1 mM taurine to the medium at the start of the growth experiment.

Growth and phenotypic analysis of cultures

Anaerobic minimal media batch cultures (400 ml) were grown in sealed Duran flasks (500 ml total volume), fitted with a septum to allow for gas-tight sample extraction. Minimal media consisted of: 30 mM succinate, 20 mM nitrate, 11 mM dihydrogen orthophosphate, 29 mM di-sodium orthophosphate, 0.4 mM magnesium sulfate, 1 mM ammonium chloride, pH 7.5. The minimal media was supplemented with a 2 ml l^{-1} Vishniac and Santer trace element solution³⁸ where copper sulfate was present (Cu-sufficient, $12.8 \mu\text{M}$) or excluded from the original recipe (Cu-limited, $<0.5 \mu\text{M}$), as previously described.²² Media were inoculated using a 1% inoculum from a starter culture to give a starting $\text{OD}_{600 \text{ nm}}$ of 0.02 and incubated at 30 °C. Samples of the liquid culture were taken in 1 ml aliquots and $\text{OD}_{600 \text{ nm}}$ measured. 3 ml gas samples were removed from the headspace of the cultures and stored in pre-evacuated 3 ml Exetainer® vials. A 50 μl gas sample was injected into a Clarus 500 gas chromatograph (PerkinElmer) with an Elite-PLOT Q (30 m \times 0.53 mm internal diameter) and an electron capture detector. Carrier gas was N₂, make-up gas was 95% (v/v) argon, 5% (v/v) methane. Standards containing N₂O at 0.4, 5, 100, 1000, 5000, and 10 000 ppm (Scientific and Technical Gases) were measured and total N₂O was determined as previously described.²²

Purification and characterisation of affinity-tagged N₂OR from *P. denitrificans* strains

N₂OR (NosZ) was expressed *in trans* in *P. denitrificans* using pLMB511, a derivative plasmid of the taurine-inducible expression vector pLMB509 for α -proteobacteria (Table S1†).³⁹ The EcoRI site at position 1107 bps in pLMB509 was removed by PCR-based site-directed mutagenesis to generate pMSL001 (see Table S4† for primers). Subsequently, pMSL001 was modified by cloning of an NdeI–EcoRI fragment (Table S1† for sequence) to yield pLMB511, which has a unique NdeI–BamHI–XmaI–EcoRI multiple cloning site that also contains the Strep-II tag sequence. As high-GC content precluded PCR gene amplification, the coding sequence of Pden_4219 (*nosZ*) was synthesised (GenScript) and cloned into pLMB511 as a NdeI–XmaI fragment, yielding pMSL002 from which N₂OR (NosZ) with a C-terminal Strep-tag II was overproduced. The pMSL002 plasmid with Gen^R ($20 \mu\text{l ml}^{-1}$) was conjugated into *Pd* Δ nosZ (PD2303), *Pd* Δ nosL (PD2501) and *Pd* Δ nosZL (PD2505) using the *E. coli* pRK2013 helper strain. Conjugants were screened for both Gen^R/Spec^R and first cultured in LB and subsequently in 4 L minimal media supplemented with 2 ml l^{-1} trace element solution, at 30 °C. Overproduction of strep-tagged N₂OR was initiated by the addition of 10 mM taurine when the culture reached $\text{OD}_{600 \text{ nm}} \sim 0.6$ and cultures were incubated at 30 °C for

24 h. Cells were harvested by centrifugation at $5000 \times g$ and resuspended in binding buffer (20 mM HEPES, 150 mM NaCl, pH 7.2) with a protease inhibitor (cOmplete™ from Roche, 1 tablet per 50 ml resuspended cells) and lysed using a French pressure cell at 1000 psi. The cell lysate was centrifuged at $205\,000 \times g$ for 1 h at 4 °C and the supernatant applied to a Hi-Trap HP Strep II affinity column (5 ml, GE Healthcare). N₂OR-Strep-tag II was eluted using elution buffer (20 mM HEPES, 150 mM NaCl and 2.5 mM desthiobiotin, pH 7.2) and exchanged back into binding buffer using a 30 kDa MWCO Centricon filter unit. Purity of the sample was confirmed using SDS-PAGE analysis and LC-MS. Protein concentrations were determined using the Bradford assay (BioRad)⁴⁰ and bovine serum albumin as a protein standard.

UV-visible absorbance spectra of N₂OR-Strep-tag II were recorded on a Jasco V-550 spectrophotometer. Circular dichroism spectra were recorded using Jasco J-810 Spectropolarimeter. Samples were made anaerobic by sparging with nitrogen gas for 5 min and oxidised or reduced with 5 mg ml⁻¹ stocks of potassium ferricyanide and sodium dithionite, respectively, in 20 mM HEPES, 150 mM NaCl, pH 7.2, by titrating concentration equivalents. Total copper content of the protein was determined using a colorimetric bathocuproinedisulfonic acid (BCS) assay. A 100 μl protein sample was heated to 95 °C for 1 h with an equal volume of 20% (v/v) HNO₃. The reaction was cooled and neutralised using 0.6 ml saturated ammonium sulfate solution. Copper was reduced using 100 μl hydroxylamine (100 mM) and 100 μl BCS (10 mM) added. The absorbance at 483 nm was recorded after 30 min. A standard curve was produced using a standard copper sulfate solutions (Sigma).

Activities of N₂OR-Strep-tag II enzymes was determined using an adapted methyl viologen assay^{25,41} in which samples were incubated with a 500-fold excess of reduced methyl viologen. Reaction was initiated by adding N₂O saturated buffer and the oxidation of blue (reduced) methyl viologen to its oxidised colourless form was followed at 600 nm as a function of time and data converted to specific activity using $\epsilon_{600 \text{ nm}} = 13\,600 \text{ M}^{-1} \text{ cm}^{-1}$ for the reduced methyl viologen cation radical.⁴¹

Purification and characterisation of NosL

A codon-optimised gene encoding an N-terminally truncated version of *Pd*NosL (Pden_4215) was synthesised (GenScript) and sub-cloned into pET-21a(+) using 5' NdeI and 3' EcoRI restriction sites to generate pSPBN3. The truncation, which resulted in the replacement of the first 16 residues with a Met, was designed to simplify the expression and maturation of the protein in *E. coli* as it yielded a soluble protein located in the cytoplasmic fraction. The pSPBN3 plasmid was used to transform *E. coli* BL21 (DE3) to ampicillin ($100 \mu\text{g ml}^{-1}$) resistance. Typically, 2 L flasks containing 500 ml LB supplemented with $100 \mu\text{g ml}^{-1}$ ampicillin were inoculated with 1% (v/v) of an overnight culture and grown for 2 h, 180 rpm, 37 °C until $\text{OD}_{600 \text{ nm}}$ reached ~ 0.6 . Expression of the NosL-encoding gene was induced by addition of 500 μM IPTG and cultures were



subsequently incubated at 37 °C, 180 rpm for 5 h. Cells were harvested by centrifugation at 4000 × *g* for 15 min at 4 °C, resuspended with buffer A (50 mM MES, pH 6.5) and lysed by three rounds of sonication, each for 8 min 20 s (0.2 s intervals, 50% power), on ice. The cell lysate was centrifuged at 40 000 × *g* for 45 min at 4 °C and the supernatant applied to a DEAE column (HiPrep DEAE FF 16/10; GE Healthcare) equilibrated in buffer A. NosL was eluted using a 0–50% gradient of buffer B (50 mM MES, 1 M NaCl, pH 6.5). Fractions containing NosL were buffer exchanged using a 10 kDa MWCO Centricon into buffer A and applied to a Q-sepharose column (HiPrep Q FF 16/10; GE Healthcare) and eluted using a 20–50% gradient of buffer B. NosL-containing fractions (as determined by SDS-PAGE) were pooled and subsequently applied to an S-100 gel filtration column (120 ml, GE Healthcare) equilibrated in 100 mM MOPS, 100 mM NaCl, pH 7.5 (buffer C) and eluted in the same buffer. Fractions containing pure NosL were combined and dialysed overnight against buffer C containing 1 mM EDTA at 4 °C, and subsequently back into buffer C.

An extinction coefficient at 280 nm of $11\,923 \pm 5.2\text{ M}^{-1}\text{ cm}^{-1}$, determined using a guanidine hydrochloride assay,⁴² was used to quantify the NosL protein concentration. Copper analysis revealed that the protein was purified in a copper-free form. Copper titrations were carried out using a 1 M stock solution of Cu(I)Cl (in 1 M NaCl and 0.1 M HCl) or CuCl₂ dissolved in water. The protein was titrated with copper to give increases in the ratio of Cu : NosL of 0.1 per addition and spectra were recorded between 240 and 600 nm after each addition. Competition assays between Cu(I)-NosL and BCS were carried out to measure the dissociation constant, K_d , for Cu(I)-binding to NosL, using the extinction coefficient $\epsilon_{483\text{nm}} = 13\,300\text{ M}^{-1}\text{ cm}^{-1}$ to determine the concentration of [CuBCS₂]³⁻, as previously described,³³ with 10 min incubation after each addition. Absorbance spectra were recorded using a Jasco V550 spectrophotometer.

LC-MS was conducted using a Bruker microQToF-QIII electrospray ionisation time of flight (TOF) mass spectrometer calibrated in the *m/z* range 300–2000 using ESI-L Low Concentration Tuning Mix (Agilent Technologies). Samples were prepared by ten-fold dilution of 50 μM protein solution with 2% (v/v) acetonitrile and 0.1% (v/v) formic acid to 0.5 ml. Samples were loaded into the LC-MS via an autosampler using an UltiMate 3000 HPLC system (Dionex). A 20 μl injection volume of the protein was applied to a ProSwift reversed phase RP-1S column (4.6 × 50 mm; Dionex) at 25 °C. A gradient elution was performed at a flow rate of 200 ml min⁻¹ using solvents A (0.1% formic acid) and B (acetonitrile, 0.1% formic acid). Once loaded the following chromatographic method was used: isocratic wash (2% B, 0–2 min), linear gradient from 2–100% B (2–12 min), followed by an isocratic wash (100% B, 12–14 min) and column re-equilibration (2% B, 14–15 min). Mass spectra were acquired throughout using the following parameters: dry gas flow 8 l min⁻¹, nebuliser gas pressure 0.8 bar, dry gas 240 °C, capillary voltage 4500 V, offset 500 V, collision RF 650 Vpp. Mass spectra from manually chosen elution volumes were averaged and deconvoluted using a maximum entropy deconvolution algorithm in Compass DataAnalysis version 4.1 (Bruker Daltonik).

Samples for non-denaturing ESI-MS were prepared in volatile buffer, ammonium acetate, 50 mM, pH 7.8. Lines were washed with anaerobic buffer prior to sample loading to ensure all O₂ was removed and protein samples were loaded into a Hamilton syringe and directly infused into the ESI source at a rate of 300 μl h⁻¹. Data was acquired in 5 min increments with ion scans between 500 and 3000 *m/z*. NosL mass spectra (*m/z* 1000–3000) were recorded with acquisition controlled by Bruker qTOF Control software, with parameters as follows: dry gas flow 4 l min⁻¹, nebulise gas pressure 0.8 Bar, dry gas 180 °C, capillary voltage 4000 V, offset 500 V, quadrupole voltage 5 V, collision RF 1000 Vpp, collision cell voltage 20 V. Spectra were deconvoluted as above. Exact masses are reported from peak centroids representing the isotope average neutral mass. Predicted masses are given as the isotope average of the neutral protein or protein complex, in which Cu(I)-binding is expected to be charge compensated.⁴³

Continuous wave X-band electronic paramagnetic resonance (EPR) measurements were recorded using a Bruker EMX EPR Spectrometer equipped with an ESR-900 liquid helium flow cryostat (Oxford Instruments). Spectra were recorded at 10 K with the following instrumental settings: microwave frequency, 9.4652 GHz; microwave power, 3.18 mW; modulation frequency, 100 kHz; modulation amplitude, 5 G; time constant, 82 ms; scan rate, 22.6 G s⁻¹; single scan per spectrum. A 98 μM Cu(II)-EDTA standard was used to estimate Cu(II) concentrations for protein samples by spin integration of signal area for spectra versus that of the standard, where all spectra were recorded under non-saturating power conditions.

Author contributions

S. P. B., D. J. R., A. J. G. and N. L. B. conceived the study. S. P. B., M. J. S. L., J. M. B. and D. A. S. performed the experiments. S. P. B., D. A. S., A. J. G. and N. L. B. analysed the data. S. P. B., D. J. R., A. J. G. and N. L. B. wrote the manuscript.

Conflicts of interest

The authors declare no competing financial interests.

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References

- 1 A. R. Ravishankara, J. S. Daniel and R. W. Portmann, *Science*, 2009, **326**, 123–125.
- 2 H. W. Hu, D. Chen and J. Z. He, *FEMS Microbiol. Rev.*, 2015, **39**, 729–749.



- 3 A. J. Thomson, G. Giannopoulos, J. Pretty, E. M. Baggs and D. J. Richardson, *Philos. Trans. R. Soc., B*, 2012, **367**, 1157–1168.
- 4 L. A. Domeignoz-Horta, L. Philippot, C. Peyrard, D. Bru, M.-C. Breuil, F. Bizouard, E. Justes, B. Mary, J. Léonard and A. Spor, *Global Change Biology*, 2018, **24**, 360–370.
- 5 B. Liu, A. Frostegard and L. R. Bakken, *mBio*, 2014, **5**, e01383.
- 6 T. J. Griffis, Z. Chen, J. M. Baker, J. D. Wood, D. B. Millet, X. Lee, R. T. Venterea and P. A. Turner, *Proc. Natl. Acad. Sci. U. S. A.*, 2017, **114**, 12081–12085.
- 7 C. M. Jones, D. R. Graf, D. Bru, L. Philippot and S. Hallin, *ISME J.*, 2013, **7**, 417–426.
- 8 M. P. Heikkila, U. Honisch, P. Wunsch and W. G. Zumft, *J. Bacteriol.*, 2001, **183**, 1663–1671.
- 9 A. Dreusch, J. Riester, P. M. Kroneck and W. G. Zumft, *Eur. J. Biochem.*, 1996, **237**, 447–453.
- 10 T. Haltia, K. Brown, M. Tegoni, C. Cambillau, M. Saraste, K. Mattila and K. Djinovic-Carugo, *Biochem. J.*, 2003, **369**, 77–88.
- 11 K. Paraskevopoulos, S. V. Antonyuk, R. G. Sawers, R. R. Eady and S. S. Hasnain, *J. Mol. Biol.*, 2006, **362**, 55–65.
- 12 A. Pomowski, W. G. Zumft, P. M. Kroneck and O. Einsle, *Acta Crystallogr., Sect. F: Struct. Biol. Cryst. Commun.*, 2010, **66**, 1541–1543.
- 13 L. K. Schneider and O. Einsle, *Biochemistry*, 2016, **55**, 1433–1440.
- 14 J. A. Farrar, F. Neese, P. Lappalainen, P. M. H. Kroneck, M. Saraste, W. G. Zumft and A. J. Thomson, *J. Am. Chem. Soc.*, 1996, **118**, 11501–11514.
- 15 D. J. Richardson, L. C. Bell, A. G. McEwan, J. B. Jackson and S. J. Ferguson, *Eur. J. Biochem.*, 1991, **199**, 677–683.
- 16 R. J. M. van Spanning, C. Wansell, N. Harms, L. F. Oltmann and A. H. Stouthamer, *J. Bacteriol.*, 1990, **172**, 986–996.
- 17 A. Dreusch, D. M. Burgisser, C. W. Heizmann and W. G. Zumft, *Biochim. Biophys. Acta*, 1997, **1319**, 311–318.
- 18 Y. C. Horng, P. A. Cobine, A. B. Maxfield, H. S. Carr and D. R. Winge, *J. Biol. Chem.*, 2004, **279**, 35334–35340.
- 19 A. K. Thompson, J. Gray, A. Liu and J. P. Hosler, *Biochim. Biophys. Acta*, 2012, **1817**, 955–964.
- 20 V. Srinivasan, A. J. Pierik and R. Lill, *Science*, 2014, **343**, 1137–1140.
- 21 N. F. Saunders, J. J. Hornberg, W. N. Reijnders, H. V. Westerhoff, S. de Vries and R. J. van Spanning, *J. Bacteriol.*, 2000, **182**, 5211–5217.
- 22 M. J. Sullivan, A. J. Gates, C. Appia-Ayme, G. Rowley and D. J. Richardson, *Proc. Natl. Acad. Sci. U. S. A.*, 2013, **110**, 19926–19931.
- 23 M. A. McGuirl, J. A. Bollinger, N. Cosper, R. A. Scott and D. M. Dooley, *J. Biol. Inorg. Chem.*, 2001, **6**, 189–195.
- 24 C. Carreira, S. R. Pauleta and I. Moura, *J. Inorg. Biochem.*, 2017, **177**, 423–434.
- 25 S. Ghosh, S. I. Gorelsky, P. Chen, I. Cabrito, J. J. Moura, I. Moura and E. I. Solomon, *J. Am. Chem. Soc.*, 2003, **125**, 15708–15709.
- 26 M. Prudencio, A. S. Pereira, P. Tavares, S. Besson, I. Cabrito, K. Brown, B. Samyn, B. Devreese, J. Van Beeumen, F. Rusnak, G. Fauque, J. J. Moura, M. Tegoni, C. Cambillau and I. Moura, *Biochemistry*, 2000, **39**, 3899–3907.
- 27 T. Rasmussen, B. C. Berks, J. N. Butt and A. J. Thomson, *Biochem. J.*, 2002, **364**, 807–815.
- 28 C. L. Coyle, W. G. Zumft, P. M. Kroneck, H. Korner and W. Jakob, *Eur. J. Biochem.*, 1985, **153**, 459–467.
- 29 S. Dell'Acqua, S. R. Pauleta, J. J. Moura and I. Moura, *Philos. Trans. R. Soc., B*, 2012, **367**, 1204–1212.
- 30 S. Okuda and H. Tokuda, *Annu. Rev. Microbiol.*, 2011, **65**, 239–259.
- 31 L. M. Taubner, M. A. McGuirl, D. M. Dooley and V. Copie, *Biochemistry*, 2006, **45**, 12240–12252.
- 32 D. L. Pountney, I. Schauwecker, J. Zarn and M. Vasak, *Biochemistry*, 1994, **33**, 9699–9705.
- 33 Z. Xiao, F. Loughlin, G. N. George, G. J. Howlett and A. G. Wedd, *J. Am. Chem. Soc.*, 2004, **126**, 3081–3090.
- 34 L. Zhou, C. Singleton and N. E. Le Brun, *Biochem. J.*, 2008, **413**, 459–465.
- 35 Y. K. Chan, W. A. McCormick and R. J. Watson, *Microbiology*, 1997, **143**, 2817–2824.
- 36 M. G. Savellieff, T. D. Wilson, Y. Elias, M. J. Nilges, D. K. Garner and Y. Lu, *Proc. Natl. Acad. Sci. U. S. A.*, 2008, **105**, 7919–7924.
- 37 W. R. Zuckert, *Biochim. Biophys. Acta*, 2014, **1843**, 1509–1516.
- 38 W. Vishniac and M. Santer, *Bacteriol. Rev.*, 1957, **21**, 195–213.
- 39 A. J. Tett, S. J. Rudder, A. Bourdès, R. Karunakaran and P. S. Poole, *Appl. Environ. Microbiol.*, 2012, **78**, 7137–7140.
- 40 M. M. Bradford, *Anal. Biochem.*, 1976, **72**, 248–254.
- 41 J. K. Kristjansson and T. C. Hollocher, *J. Biol. Chem.*, 1980, **255**, 704–707.
- 42 C. N. Pace, F. Vajdos, L. Fee, G. Grimsley and T. Gray, *Prot. Sci.*, 1995, **4**, 2411–2423.
- 43 K. L. Kay, C. J. Hamilton and N. E. Le Brun, *Metallomics*, 2016, **8**, 709–719.

