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Molecular mechanism of polyketide shortening in anthraquinone biosynthesis of *Photobrhabdus luminescens*†

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Anthraquinones, a widely distributed class of aromatic natural products, are produced by a type II polyketide synthase system in the Gram-negative bacterium *Photobrhabdus luminescens*. Heterologous expression of the *antABCDEFGHI* anthraquinone biosynthetic gene cluster in *Escherichia coli* identified AntI as an unusual lyase, catalysing terminal polyketide shortening prior to formation of the third aromatic ring. Functional *in vitro* and *in vivo* analysis of AntI using X-ray crystallography, structure-based mutagenesis, and molecular simulations revealed that AntI converts a defined octaketide to the tricyclic anthraquinone ring *via* retro-Claisen and Dieckmann reactions. Thus, AntI catalyses a so far unobserved multistep reaction in this PKS system.

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Introduction

Polyketide natural products attract significant attention due to their remarkable biological and pharmacological activities. Their structural diversity derives from only a few building blocks assembled by different polyketide synthase systems (PKS):^{1,2} type I PKS represent elaborate multifunctional enzyme complexes responsible for the synthesis of macrolides including erythromycin and candicin,³ whereas type III PKS catalyse the formation of stilbenes, chalcones, and pyrones.⁴ On the other hand, type II PKS constitute discrete proteins that act in an iterative manner to synthesise polycyclic aromatic compounds such as anthraquinones (AQs), tetracyclines and doxorubicins.^{5,6} First, a specific acyl carrier protein (ACP) is loaded with an α -carboxylated precursor, and the activated unit is then transferred onto the corresponding ketosynthase. Next, iterative elongation cycles with malonyl-coenzyme A building blocks form the octaketide framework.¹ Subsequently, ketoreductases, cyclases and aromatases convert the unstable and highly reactive octaketide into the corresponding target compounds. Notably, the variability of the natural products can be further increased by tailoring enzymes, which are not part of

the gene cluster and therefore can be difficult to identify. Yet, these catalysts are specific and perform only minor modifications such as alkylations, hydroxylations, or oxidations, while preserving the octaketide skeleton.

Type I and type III PKS subcllasses are well-known from bacteria, fungi, and plants.¹ In contrast, type II PKS have only been characterised in Gram-positive *Streptomyces* and other actinomycetes.⁵ We identified a biosynthetic gene cluster encoding a type II PKS in the Gram-negative entomopathogenic bacterium *Photobrhabdus luminescens*, an enzyme system that enables the biosynthesis of AQs.⁷ Notably, along with the antimicrobial aurachins⁸ and pyxidicyclines⁹ from myxobacteria, AQs represent another fascinating example of the type II PKS machinery. Here, we have introduced the anthraquinone gene cluster *antABCDEFGHI* from *P. luminescens* in *E. coli*. Surprisingly, the formed AQ-256 product exhibits a heptaketide framework, which is unusual since polyphenolic polyketides derived from bacterial type II PKS systems are generally octaketides.¹ Our present study reveals that AntI shortens the octaketide intermediate 1 to the final AQ heptaketide. Although this type of reaction is uncommon, a related mechanism has been described for Ayg1p¹⁰ from *Aspergillus fumigatus* and WdYg1p¹¹ from *Wangiella dermatitidis*. Both fungal proteins are part of PKS systems involved in the biosynthesis of melanin pigments.

We provide insights into the molecular mechanism of AntI *via* biochemical, crystallographic, computational, and functional characterisations, and show how this enzyme catalyses an elimination reaction, followed by cyclisation to the anthraquinone ring. Taken together, our combined findings elucidate an interesting type of polyketide shortening mechanism.

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Results and discussion

The gene cluster encoding the anthraquinone biosynthetic machinery (*antABCDEFGHI*) in *P. luminescens* was heterologously expressed in *E. coli* (strain EC^{AQ}). This approach allowed detection of the central metabolite 1,3,7-trihydroxyanthracene-9,10-dione (AQ-256), a tricyclic aromatic compound not found in wild-type *E. coli* (Fig. 1). Yet, methylated derivatives such as AQ-270a (one methyl group) and AQ-284a (two methyl groups) were identified in the natural producer *P. luminescens* (Fig. 2). As these molecules were not detected in our engineered EC^{AQ} strain, this suggests that the responsible methyltransferases are absent in the *ant* cluster, and therefore must be located elsewhere in the *Photobacterium* genome. Notably, the EC^{AQ}-system is accompanied by several shunt products (SP) such as **SP1**, **SP^{EC}1** and **SP^{EC}2** (Fig. 2). One reason for this observation might be that the expression level of the individual genes in the heterologous host is not balanced. Nevertheless, all shunt products could be traced back to the common precursor molecule 1 harboring an octaketide skeleton as its basic structural element, which is covalently bound to the acyl carrier protein AntF *via* a phosphopantetheine (PPT) prosthetic group (see below). To elucidate the formation of these distinct compounds, we selectively removed individual genes from our EC^{AQ} operon. We focused on classical type II PKS enzymes such as the ketoreductase AntA (C9 carbonyl reduction), aromatase AntH (first cyclisation, C7/C12), and cyclase AntC (second cyclisation, C5/C14). The resulting intermediates were identified by UV-VIS spectroscopy and mass spectrometry (ESI Fig. 1–3, Tables 1–2†). Mutants constructed in *P. luminescens* (ESI Fig. 4†) and labelling experiments in *E. coli* (ESI Fig. 3†) allowed assignment of the reaction cascade based on known biosynthetic pathways^{1,2} (ESI Fig. 5†).

Most polyphenolic polyketides derived from bacterial type II PKS systems are octaketides.¹ In this respect, AQ-256 with its heptaketide framework is exceptional. Moreover, the local alignment search in the NCBI databank revealed that AntI (Uniprot: Q7MZT8) is predominantly found in *Photobacterium*

species. Thus, AntI is a promising candidate for polyketide shortening, followed by a unique cyclisation of the third aromatic ring in AQ biosynthesis (Fig. 1). In order to confirm our hypothesis, we generated a deletion mutant of *antI* in *P. luminescens* (PL^{ΔantI}). As expected, the modified strain lost the ability to produce AQ-256 and its methylated derivatives (Fig. 2), and only the nitrogen-containing octaketide derivative **SP1** was formed. Next, we removed *antI* from our engineered *E. coli* strain (EC^{AQΔantI}) and analysed the resulting product pattern. As expected, AQ-256 is no longer present. However, in contrast to the *Photobacterium* ΔantI mutant, which produced only the **SP1** metabolite, we now observed several compounds. *Via* detailed characterisation of the EC^{AQΔantI} lysate, we identified **SP1–5**, **SP^{EC}1**, and **SP^{EC}2**, all of which could unambiguously be shown to derive from molecule 1 (Fig. 2, ESI Fig. 5†). In summary, (i) the dihydropyridine moiety of **SP1** results from 1 by assimilation of intracellular NH₃; (ii) **SP4** is formed from **SP1** after dehydration and decarboxylation; (iii) utahmycin A¹² (**SP5**) is an oxidation product of **SP4**; (iv) 3,8-dihydroxy-1-methyl-anthraquinone-2-carboxylic acid (DMAC, **SP^{EC}1**) is the oxidized product of an aldol-condensation of 1; and (v) aloesaponarin II (**SP^{EC}2**) results from the spontaneous decarboxylation of **SP^{EC}1**. The assignment of these shunt products strongly indicates that AntI orchestrates the cyclisation of the third ring system in the AQ biosynthesis. In this reaction, 4-(3-acetyl-5-hydroxy-4-oxo-1,4-dihydronaphthalen-2-yl)-3-oxobutanoic acid (1) bound to AntF *via* the PPT-prosthetic group possibly acts as the AntI substrate that is converted to 8-hydroxy-1,10-dihydroanthracene-2,3,9(4H)-trione (2, Fig. 1). We thus propose that AntI catalyses the last step in this reaction cascade, whereas formation of the aromatic quinone moiety of AQ-256 occurs in an enzyme-independent manner by keto–enol-tautomerism and spontaneous oxidation.

The rapid turnover of the precursor molecule 1 in combination with the resulting large variety of different shunt products achieved in our modified *E. coli* system, encouraged us to additionally validate octaketide intermediate 1 as the true AntI substrate. To this end, the biosynthesis of the well-known blue

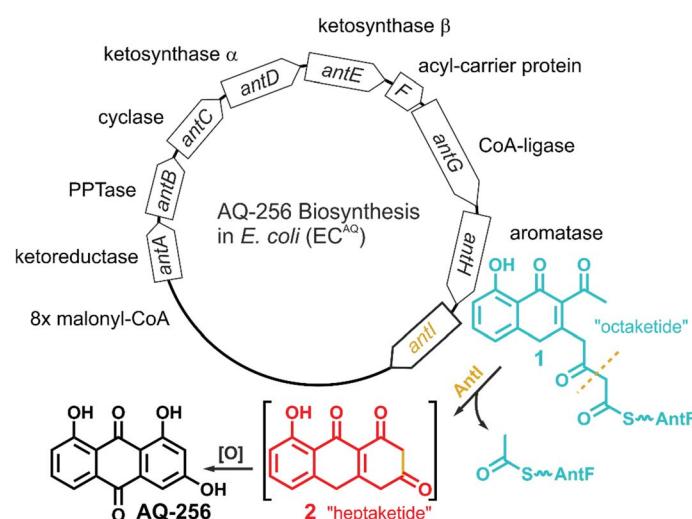


Fig. 1 Overview of the *ant* biosynthetic gene cluster from *Photobacterium luminescens*. The AQ biosynthesis is highlighted with a focus on AntI catalyzed polyketide shortening.



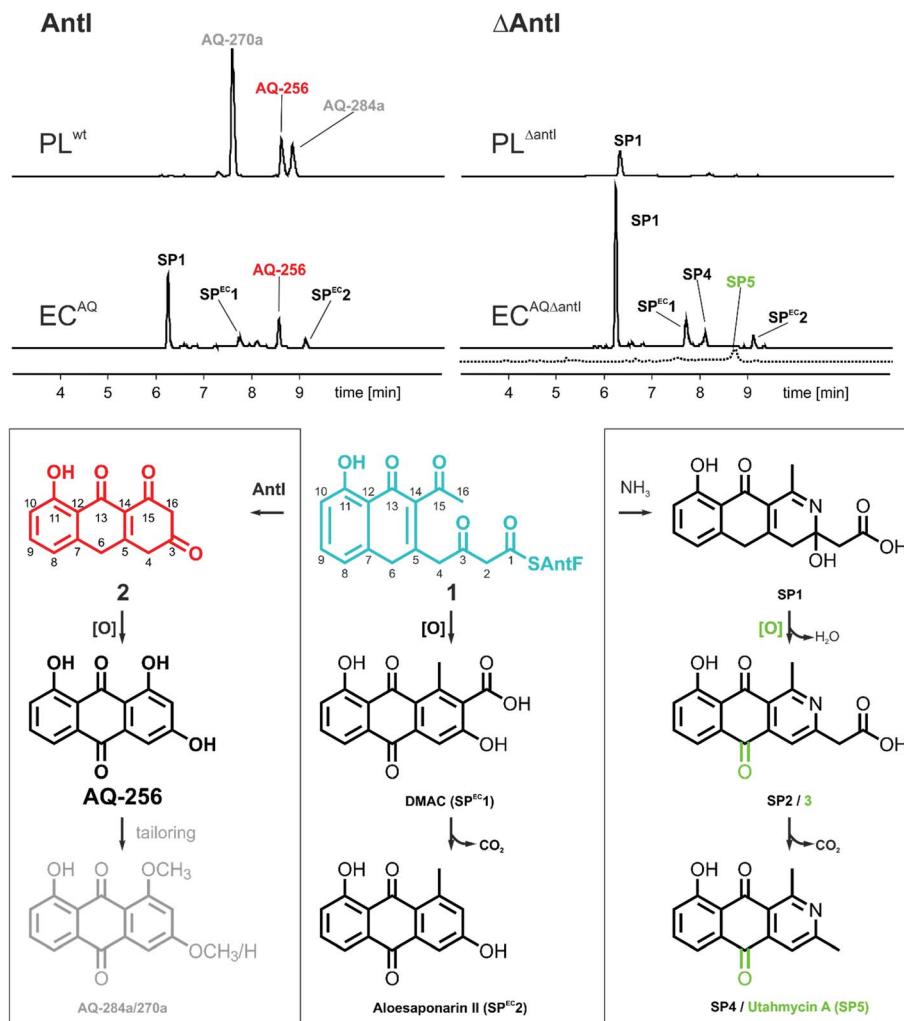


Fig. 2 HPLC/UV (420 nm, solid line) and HPLC/MS (EIC m/z 254 for SP5, dashed line) analysis for polyketide production in the presence and absence of AntI in *E. coli* (EC) and *Photobacterium luminescens* (PL). Identified key metabolites (top) and suggested biosynthesis pathways to these compounds (bottom) are shown.

pigment actinorhodin from *Streptomyces coelicolor* appeared to be ideal, as it transforms 1 into the octaketide 4-dihydro-9-hydroxy-1-methyl-10-oxo-3-H-naphtho-[2,3-c]-pyran-3-(S)-acetate (**S-DNPA**). In this case, the stereospecific ketoreductase (Red1) acts as a tailoring enzyme by recognising the metabolite and reducing its 3β -keto group to a secondary alcohol.¹³ This peculiarity led us to extend the $EC^{AQ\Delta AntI}$ construct *via* a codon-optimized *actVI ORF1* gene, encoding Red1, which resulted in the $EC^{AQ\Delta AntI+red1}$ strain. In this variant, **AQ-256** was no longer produced, but we indeed detected the compound **S-DNPA** (Fig. 3a), confirming the activity of Red1 in our model system. These results further prove that octaketide 1 is formed, and then subsequently shortened and cyclised by AntI.

With the identification of AntI's substrate (1), we wanted to obtain a deeper mechanistic insight into the catalytic properties of this fascinating enzyme. AntI was therefore heterologously expressed in *E. coli*, purified, and crystallised with one subunit in the asymmetric unit. Native diffraction data were collected to 1.85 Å resolution, and the structure was phased by single-wavelength anomalous dispersion using selenomethionine

labelling (PDB ID: 6HXA, $R_{free} = 0.209$, ESI Table 3†). The structure reveals that AntI functions as a homodimer, with each subunit comprising two domains (Fig. 4a) that are connected by a short linker sequence (residues 130–134). The N-terminal domain (residues 7–130) comprises an α -helix bundle ($\alpha1$ – $\alpha5$) that orchestrates the arrangement of the two subunits in a back-to-back orientation with a contact area of 2055 Å². The C-terminal domain (residues 134–381) contains a twisted eight-stranded β -sheet ($\beta1$ – $\beta8$), which is flanked by 7 α -helices ($\alpha6$ – $\alpha12$) exhibiting an α / β -hydrolase fold (Fig. 4c).¹⁴

AntI forms a rod-shaped particle (Fig. 4b) with its N- and C-terminal domains enclosing a conspicuous intramolecular substrate binding cavity with approximate dimensions of 8 × 8 × 20 Å³. A DALI search¹⁵ demonstrated similarity of AntI with 2,6-dihydroxy-pseudo-oxy-nicotine hydrolase (DHPON; PDB ID: 2JBW, with a sequence identity of 23%, N-domain: Z-score = 10.6, RMSD = 2.5 Å; C-domain: Z-score = 30.5, RMSD = 1.9 Å),¹⁶ a C–C bond cleaving α / β -hydrolase from the Gram-positive bacterium *Arthrobacter nicotinovorans* that is involved in nicotine degradation (Fig. 4c). The structural comparison of AntI with DHPON

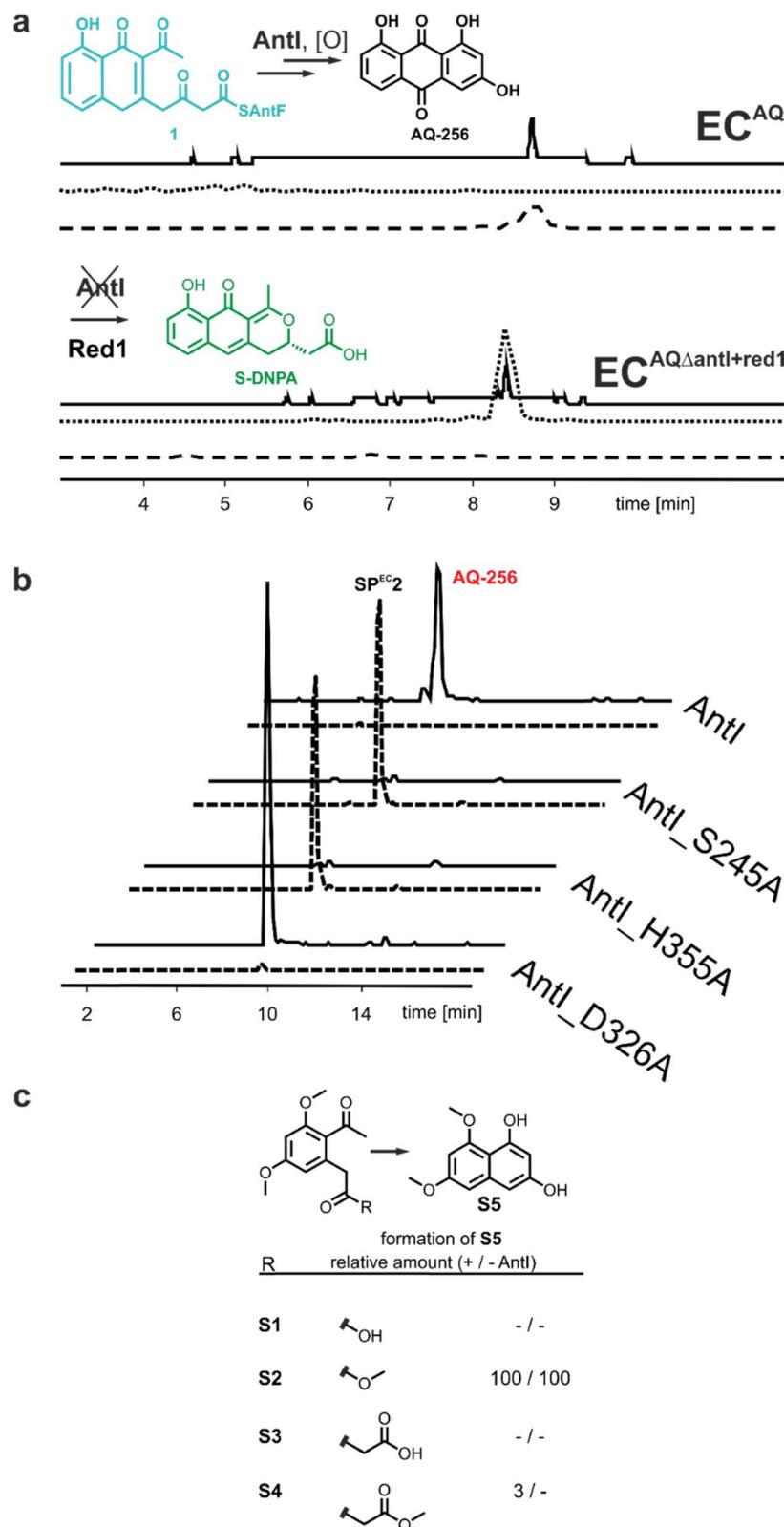


Fig. 3 (a) HPLC analyses of *E. coli* expressing *antABCDEFGH* and *antABCDEFGH* + *actVI-ORF1* (encoding Red1). Besides UV traces (420 nm, solid line), EIC for AQ-256 (m/z 255 $[M - H]^-$, dashed line) and S-DNPA (m/z 287 $[M + H]^+$, dotted line) are shown. (b) In HPLC/MS analyses of *E. coli* expressing *antABCDEFGH* with the Antl-mutants S245A or H355A, metabolite 1 is no longer detected, but compound SP^{EC2} (m/z 253 $[M - H]^-$, dashed line) is formed. Mutation of D326 to alanine did not alter catalysis. (c) Conversion of model compounds S1–S4 into S5.

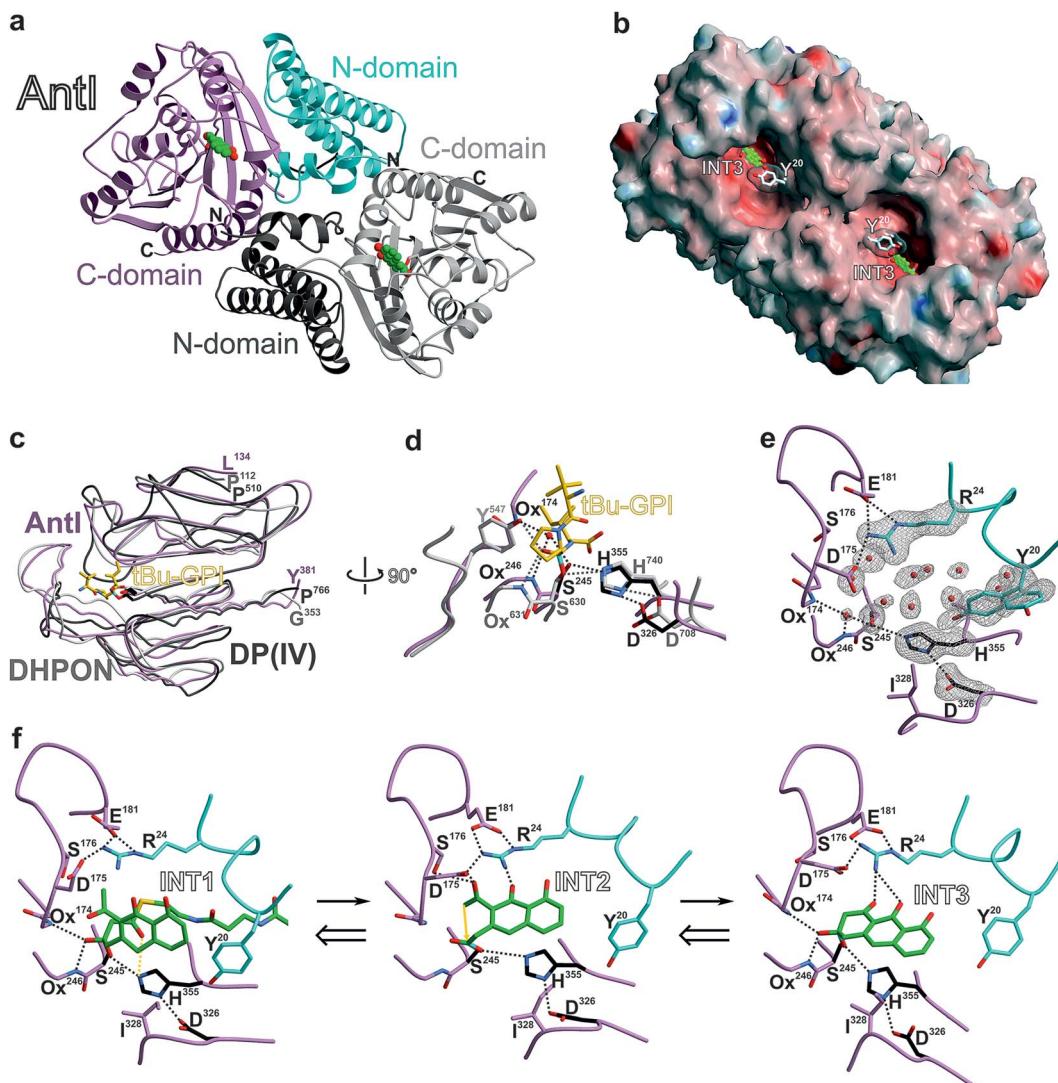


Fig. 4 Crystal structure of Antl. (a) The homodimer (ribbon plot) is depicted with a modelled intermediate (INT3, balls-and sticks, carbon atoms in green). N- and C-domains of subunit A are coloured in pink and cyan, respectively; subunit B is shown in grey. (b) Surface representation of Antl. The active site of the lyase is solvent exposed. Colours highlight negative and positive electrostatic potentials contoured from $-40k_B T/e$ (red) to $+40k_B T/e$ (blue). (c) Superposition of the α/β -hydrolase fold of Antl (residues 174–381, pink, PDB ID: 6HXA) with domains of 2,6-dihydroxy-pseudo-oxynicotine hydrolase (DHPON, light grey, PDB ID: 2JBW)¹⁶ and the human dipeptidyl peptidase DP(IV)–tBu-GPI complex (grey, PDB ID: 2AJB).¹⁷ (d) Structural overlay of Antl with the active site of DP(IV) (C-atoms in grey) covalently bound to the tBu-GPI surrogate. Notably, the tetrahedral coordination fits into the active site of Antl, and identifies its putative oxyanion hole (O_x), which is occupied by a water molecule. In Antl, the main chain atoms of Leu174NH and Phe246NH might form the oxyanion hole, while in DP(IV) Tyr631N and Tyr547OH take over this function. Hydrogen bonds are drawn as dashed lines. (e) Close up view of the active site of Antl showing the $2F_o - F_c$ electron density map (grey meshes contoured to 1.0σ) for the catalytic triad, Tyr20 (alternative conformation which might function as the gate keeper) and Arg24. The abundance of well-defined water molecules (red spheres) is a clear indication for the possible substrate binding site. (f) Proposed multistep reaction sequence of Antl including a retro-Claisen reaction and a Dieckmann condensation. The ligand-bound Antl structures remains dynamically stable during the MD simulations (see ESI†). The calculations were performed in the reverse direction from INT3 to INT1.

reveals the catalytic centre of a serine-type protease composed of Ser245, Asp326, and His355. In the active site, Ser245 is positioned at the tip of a sharp turn between sheet β 5 and helix α 8, called the nucleophilic elbow,¹⁴ and points towards the helix dipole generated at the N-terminus of helix α 8 (Fig. 4d). The mutation of Ser245 or His355 to alanine confirmed the catalytic importance of these residues (Fig. 3b). In contrast, the Asp326Ala mutant remains active. Thus, Asp326 polarises His355, but

proton transfer from Ser245 to His355 is possible also without the aspartate residue. Inspection of the $2F_o - F_c$ -electron density map illustrates that the catalytic centre is enclosed by a cluster of bulk solvent molecules, the release of which might facilitate substrate binding (Fig. 4e). Consistent with the asymmetric hydrophilic rim of 1, which is composed of three oxygen atoms, the specificity pocket exhibits an amphiphilic character. The hydrophobic half of this pocket comprises Phe27, Ile244, Ile326,

Val356, Leu358, and Ile361, whereas the polar part includes Arg24, Asp175, Ser176, Glu181, and Asp287. The central opening into the chamber is formed by Tyr20, Pro281, and Asp327. Interestingly, the sidechain of Tyr20 is observed in two alternative conformations (Fig. 4e), suggesting a regulatory role for this residue in allowing the substrate to access the active centre.

We aimed to obtain detailed insights into the reaction cascade of AntI. While the elimination reaction could clearly be assigned to AntI, the enzyme-dependency of the cyclisation resulting in the anthracene ring system still required experimental validation. We therefore synthesised surrogates **S1** and **S2** representing non-cyclised derivatives of the heptaketide intermediate (Fig. 3c, ESI Fig. 6†): **S1** harbours a free carboxylate moiety, whereas the methoxy group of **S2** mimics the sidechain of Ser245. Interestingly, **S1** is stable in solution and in the presence of AntI, whereas **S2** undergoes rapid ring closure to **S5** even without AntI (Fig. 3c). These findings indicate that an oxo-ester with AntI has to be formed, to enable intermolecular cyclization. Hence, AntI is involved in the multistep biosynthetic route by catalyzing polyketide shortening and subsequent ring formation.

The next goal was to confirm the retro-Claisen reaction for shortening the octaketide skeleton to a heptaketide. In contrast to the Dieckmann condensation, this requires a strong nucleophile, which might represent the rate-limiting step in AntI catalysis. To this end, we determine whether AntI works like a tailoring enzyme that uses intermediate 1 as its substrate after hydrolysis from the PPT arm (containing a free carboxylate), or whether 1 is still covalently bound to its ACP AntF. We synthesised 2-carboxymethyl-benzoic acid (surrogate **S3**), as well as its corresponding methyl ester **S4** which mimics metabolite 1. Both molecules remain stable in solution and in the presence of AntI. This observation is in contrast to the spontaneous enzyme-independent aromatisation of ester surrogate **S2** (see above). Thus, compared to the Dieckmann condensation, we assume that AntF is mandatory for the retro-Claisen reaction.

To obtain mechanistic insights into the structure and dynamics of these reaction intermediates, we performed molecular simulations based on our experimentally solved structure of AntI. As a starting point for these computations, we used a surrogate pose of the hydrolytic domain of the homologous porcine dipeptidyl peptidase IV (DP(IV), residues 511–766, PDB ID: 2AJB, RMSD of 145 C α -atoms = 2.7 Å, sequence identity 17%). DP(IV) was solved with the tripeptide *tert*-butyl-Gly-Pro-Ile (*t*Bu-GPI), a surrogate that covalently binds in a sp³-configuration to the catalytic Ser630O^Y by forming an orthoester.¹⁷ Hereby, the alkoxide moiety of the inhibitor is negatively charged due to its strong interaction with the oxyanion hole. The tetrahedral coordination of *t*Bu-GPI resembles the expected transition state structure with the alcoholate stabilised by hydrogen bonds to the backbone of Tyr631NH (3.1 Å) and the sidechain of Tyr547OH (2.9 Å). Intriguingly, superimposing the α/β -hydrolase fold of DP(IV) with AntI reveals an excellent structural overlap of the catalytic residues (Fig. 4c). We observe that in AntI the putative oxyanion hole is occupied by a well-defined water molecule (Fig. 4d) that forms hydrogen-bonds to the backbone of Leu174NH (2.7 Å) and Phe246NH (3.2 Å).

In accordance with the structural similarities of the active site in AntI and DP(IV), we modelled the putative reaction intermediates **INT1–INT3** and refined the protein–ligand complexes by classical molecular dynamics (MD) simulations. The procedure was initiated in reverse order, starting from a fully formed anthracene ring system at the same site as the tetrahedral hemiacetal carbon atom of *t*Bu-GPI is bound. Thus, we first modelled the late product state **INT3** which adopts a configuration having only one chiral centre out of plane with respect to the surrounding sp²-hybridised atoms. We find that **INT3** remains stably positioned at the oxyanion hole, the catalytic centre, and Arg24 during our 100 ns MD simulations. Interestingly, Tyr20, which is located at the entry gate of the substrate binding-pocket, switches between an open and a closed conformation during the MD simulations (ESI Fig. 7†). This finding is consistent with the two alternative positions of this residue in the X-ray structure (Fig. 4e). Furthermore, our simulations suggest that **INT3** forms van der Waals contacts with Leu284 and Ile328, as well as hydrogen bonds between the biaryl hydroxyl and keto groups *via* Arg24. In addition, the aromatic ring system of **INT3** is stabilised from both sides by cation–π interactions with Arg24 and by π-stacking with His355 (Fig. 4f, left panel). These insights allowed us to analyse the Dieckmann condensation in more detail.

INT3 was used as the starting pose for modelling the **INT2** state by following the enzymatic reaction cycle in reverse order. Thus, we opened the third ring of **AQ-256** between C3 and C16, leading to an acetyl group and an ester bond at the nucleophile Ser245O^Y (**INT2**, Fig. 4f, central panel). The **INT2** pose remains stable during the MD simulations, with less than 2 Å fluctuation in the atomic positions of **INT2** relative to **INT3** (ESI Fig. 7†). However, the **INT2** ligand has an increased flexibility that leads to rotation of the acetyl group by 45°. In addition, the carbonyl oxygen coordinates to the sidechain of Ser176, forming close contacts with Asp175 that favour the enol tautomer of the ligand. In accordance with the AntI X-ray structure, the side-chain of Asp175 is in proximity to Asp287 (2.5 Å) and, together with Glu181, forms an ion-pair with Arg24. This conformation facilitates deprotonation of the enol tautomer, *via* proton transfer to His355. Subsequently, the generated alkoxide might act as the nucleophile, initiating the Dieckmann condensation by intramolecular ring closure (Fig. 5a). The reaction takes place *via* a Bürgi–Dunitz angle in which the carbon atom in α -position (C16) is perfectly oriented to form a covalent bond with the carbonyl group (C3) of the acyl–AntI complex leading to **INT3** (Fig. 4f, right panel). Starting from **INT2**, this attack is regioselective from the *re*-face, while the backside is shielded from the catalytically active centre to exclude saponification that would result in shunt products.

We prolonged **INT2** by an additional acetyl group, and converted the ester at the active site Ser245 back to sp³-hybridisation (C3) (**INT1**, Fig. 4f, left panel). In doing this, we expected that the flexible, *ca.* 20 Å long PPT chain, introduces some structural uncertainty into the model. Intriguingly, the MD simulations of this state suggest that the oxyanion hole is rigid and dynamically stable over time, allowing us to properly refine the orientation of the introduced acyl–thioester moiety in **INT1**.



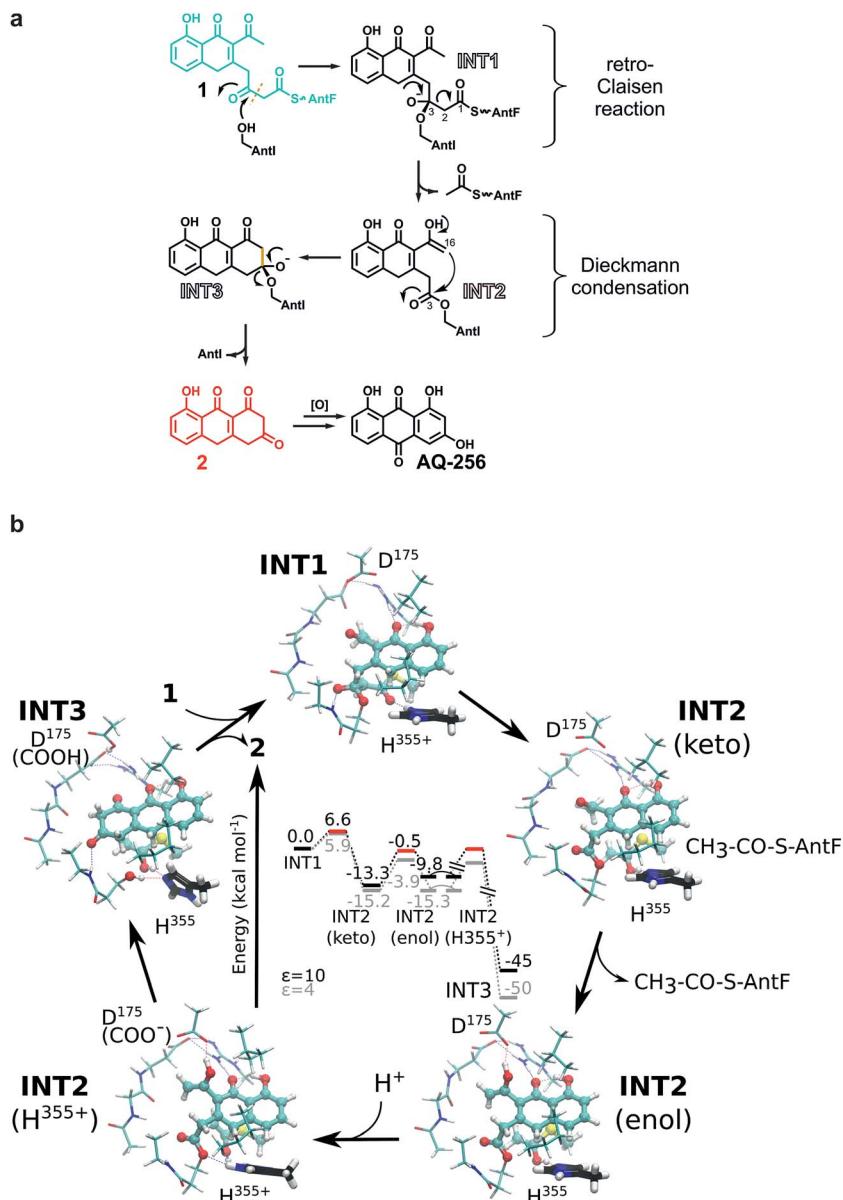


Fig. 5 (a) Proposed reaction scheme for the conversion of 1 into AQ-256 by Antl. (b) Quantum chemical DFT calculations (B3LYP-D3/def2-TZVP/ ε = 4/10) on a putative mechanism starting from INT1. See main text for detailed description.

The steric hindrance from the PPT arm and the adjacent acetyl group at C14 significantly increased in our calculations, relative to the INT2 and INT3 states. The predictions suggest that INT1 may re-orient towards the catalytically active serine residue in a new position induced by the tetrahedral configuration near the oxyanion hole (Fig. 4f, left panel). This conformation directs the carbonyl oxygen of the thioester at C1 towards His355, which might initiate the breakage of the adjacent C2–C3 bond. Due to the anionic charge distribution within the active site, both nitrogens of His355 might be protonated in the late state of INT2. In order to probe the energetics of these putative INT1–INT3 intermediates, we performed quantum chemical density functional theory (DFT) calculations, starting from an active site model system constructed from an MD-relaxed snapshot of INT1 (Fig. 5b). The DFT calculations illustrate that the proton

transfer from His355 to the carbonyl oxygen of the thioester group has a barrier of less than 10 kcal mol⁻¹, and is exergonic by 13–15 kcal mol⁻¹, indicating that the reaction is both kinetically and thermodynamically feasible. Hereby hydrolysis of the ester at the nucleophilic Ser245 is suppressed, and the subsequent Dieckmann condensation favoured. For the keto-enol tautomerisation forming INT2, we obtain a reaction barrier of *ca.* 12 kcal mol⁻¹ leading to an energetically degenerate state at *ca.* –10 kcal mol⁻¹. Proton transfer from the enol-tautomer to the nearby Asp175 increases the nucleophilic character of the enol, hence favouring the ring formation that takes place semi-concertedly. The cyclisation reaction succeeds *via* a Bürgi–Dunitz angle in which the carbon atom in α -position (C16) is perfectly oriented to form a covalent bond with the carbonyl group (C3) of the acyl–Antl complex leading to INT3. Due to

multiple bond formations and cleavage processes, the exact barrier for the **INT2** → **INT3** reaction could not be determined, but the entire trajectory is energetically strongly exergonic by more than 40 kcal mol⁻¹. Additionally, we find that the process is favoured by re-protonation of the Ser245 nucleophile *via* His355, which regenerates the active site residue. Taken together, our sampled configurations and putative DFT models manifest that release of an acetyl-AntF induce shortening of the octaketide backbone and as a consequence, forms the third aromatic ring in **AQ-256**.

Conclusion

Anthraquinone derivatives may contribute to the toxicity of *Photorhabdus* extracts against other organisms such as insects, fungi and bacteria. In most strains of *Photorhabdus*^{18,19} corresponding gene clusters have been identified (ESI Fig. 8†), implying an important function of these natural products from this genus. In such pathways, the lyase AntI is exceptional because it shortens polyketides. The reaction mechanism of the lyase provides a unique route to the formation of the tricyclic aromatic ring (Fig. 5a) – a structural feature typically introduced by dedicated cyclases in type II PKS pathways.^{1,20} Interestingly, AntI and its homologues are also found in type II PKS-encoding biosynthetic gene clusters of cyanobacteria and other Gram-negative bacteria (ESI Fig. 8†), indicating that polycyclic aromatic polyketides are more abundant than previously thought. Furthermore, the amino acids around the proposed active centre of Agy1p from *Aspergillus fumigatus*¹⁰ and WdYg1p from *Wanganella (Exophiala) dermatitidis*¹¹ are identical to AntI (ESI Fig. 9 and 10†), and therefore they can be clearly assigned as lyases. Intriguingly, these two enzymes are produced in pathogenic fungi and play a key role in the biosynthesis of the virulence factor 1,8-dihydroxynaphthalene-melanin. Thus, the structural and functional characterisation of AntI described here opens up a hitherto unexplored mechanism for an exciting class of catalysts.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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