Synthesis of unsymmetrical B$_2$E$_2$ and B$_2$E$_3$

described by borylene insertion into boradichalcogeniranes†

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We report the selective insertion of a range of borylene fragments into the E=E bonds (E = S, Se, Te) of cyclic boron dichalcogenides. This method provides facile synthetic access to a variety of symmetrical and unsymmetrical four- and five-membered rings.

Introduction

1,3-Dithietanes$^4$ – four-membered [C$_2$S$_2$] cyclic molecules – comprise a scarce class of compounds. This, however, is at odds with the fact that the cyclic 1,3-dithietyl moiety is found in bioactive compounds$^3$ such as antibacterial$^1$b and heptaprotective agents$^{12}$ and pesticides.$^2$ Even more rare are the heavier analogues of 1,3-dithietanes, 1,3-disenetane and 1,3-ditelluretane, which typically arise from the dimerization of a handful of selenolones and tellurolones.$^3$ In the main group, Group 13 analogues of these heterocycles are known but remain limited in scope due to a dearth of options for their selective synthesis.$^4$

Molecular compounds of aluminium and gallium that feature a central planar M$_2$E$_2$ ring surrounded by organic substituents may be accessed by reaction of dihydride or dimethyl precursors with elemental chalcogens$^{13}$ and, by the controlled dimerization of a platinum oxoboryl complex.$^9$ Interestingly, while a theoretical investigation led by Marder identified crucial factors for the rarity of B$_2$O$_2$ rings,$^9$ such cyclic compounds can be formed by the hydrolysis of haloboron complexes,$^9$ and the dimerization of diboroxanes$^9$ and, by the controlled dimerization of a platinum oxoboryl complex.$^9$ Indeed, the chemistry of B-O heterocycles is predominantly the field of six-membered boroxine (B$_3$O$_3$) rings.

More recently, our group reported the convergent reactivity of low-valent boron species $^[[\text{CAAC}](\text{NC})\text{B(EETe)}]_{2}$ and $^[[\text{CAAC}](\text{NC})\text{B(ETeTe)}]_{2}$ and cyclic compounds of aluminium and gallium that feature a central planar M$_2$E$_2$ ring surrounded by organic substituents may be accessed by reaction of dihydride or dimethyl precursors with elemental chalcogens$^{13}$ and, by the controlled dimerization of a platinum oxoboryl complex.$^9$ Indeed, the chemistry of B-O heterocycles is predominantly the field of six-membered boroxine (B$_3$O$_3$) rings.

Recently Moya-Cabrera and coworkers succeeded in synthesizing a unique mixed-chalcogen Al$_2$OTe ring by step-wise reaction of stoichiometric H$_2$O and Te with a β-diketiminate aluminium dihydride.$^5$ In the rather less well-explored area of low-valent Group 13 chemistry Schnöckel$^{et al.}$ synthesized the cyclic compound $[[\mu-\text{S}](\text{NETe})]_{2}$ by oxidizing an Al(i) iodide precursor with S$_8$, while Power$^{et al.}$ activated N$_2$O and S$_8$ with a β-diketiminate-stabilized Ga(i) compound to yield the corresponding gallium oxide and sulfide dimers.$^7$ For boron, strained diamino-B$_2$E$_2$ rings were obtained by Forstner and Muetterties from the reaction of B(NEt$_3$)$_2$ with H$_2$S,$^8$ and later by Nöth through thermolysis or photolysis of the [2 + 2] cycloaddition products of E=C=E (E = O, S, Se) with a bulky iminoborane.$^9$ Interestingly, while a theoretical investigation led by Marder identified crucial factors for the rarity of B$_2$O$_2$ rings,$^9$ such cyclic compounds can be formed by the hydrolysis of haloboron complexes,$^9$ and the dimerization of diboroxanes$^9$ and, by the controlled dimerization of a platinum oxoboryl complex.$^9$ Indeed, the chemistry of B-O heterocycles is predominantly the field of six-membered boroxine (B$_3$O$_3$) rings.

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Results and discussion

Borylene complexes of the group VI metals figure among the most studied metal-stabilized sources of borylene fragments.$^14$
In recent years, borylene complexes of chromium, molybdenum and tungsten have been shown to engage in metal-to-metal, metal-to-carbon\(^{1-3}\) and metal-to-nitrogen\(^{4-6}\) borylene transfer, in C–H activation\(^7\) and in \([2 + 1]\) addition reactions.\(^8\) Towards our study of the reactivity of boradichalcogeniranes with low-valent boron fragments, the sterically protected borylene complex \([\text{OC}]_3\text{Cr}–\text{BTp}\) (3) (Tp = 2,6-bis(2,4,6-trimethylphenyl)phenyl) was deemed a good starting point. Treatment of boraditellurirane complex 1 with one equivalent of 3 in toluene allowed us to isolate cyclo-[\{\text{Bu}\}{\text{IMe}}\text{B}–\text{Te}–\text{B}(\text{Tp})–\text{Te}\] (4) after recrystallization from a toluene/pentane mixture (yield: 24%). High-resolution mass spectrometry (HRMS) and single-crystal X-ray diffraction analysis allowed us to identify 4 as an NHC-stabilized 1,3-ditellurirane, which was only observed as minor products, which we were unable to isolate, however we were able to obtain a few crystals of the selenium-containing analog of 4, which allowed us to determine a solid-state structure and clarity. Selected bond distances (Å) and angles (\(^\circ\)):

- \(\text{B}–\text{Te}1 = 2.153(2)\) and 2.139(2)
- \(\text{B}–\text{Se}1 = 2.115(4)\)
- \(\text{B}–\text{C(Tp)} = 1.630(7)\)
- \(\text{B}–\text{Se}2 = 1.921(4)\)
- \(\text{Te}1–\text{Te}2 = 2.331(2)\)
- \(\text{Te}1–\text{B}–\text{Te}2 = 102.7(1)\)
- \(\text{Se}1–\text{B}–\text{Se}2 = 105.4(2)\)
- \(\text{Te}1–\text{B}–\text{Se}2 = 120.2(3)\)
- Bond valence sums (BVS) for Te1: 4.56, Te2: 4.58, B1: 3.06, B2: 3.04

Interestingly, in these cases the Se and S analogs of the four-membered \(\text{B}_2\text{Te}_2\) compound 4 were only observed as minor products, which we were unable to isolate, however we were able to obtain a few crystals of the selenium-containing analog of 4, which allowed us to determine a solid-state structure and

Fig. 1 Synthesis of boradichalcogeniranes (and a manganese adduct thereof) from a manganese borylene complex, and reaction of these chalcogenides with chromium borylene complex 3.

Fig. 2 POV-Ray depictions of the crystallographically determined structures of compounds 4 (top), 5a (bottom left) and 5b (bottom right). Atomic displacement ellipsoids depicted at the 50% probability level. Hydrogen atoms and ellipsoids of peripheral groups omitted for clarity. Selected bond distances (Å) and angles (°): 4: B1–C(Tp) 1.569(3), B1–Te1 2.153(2), B1–Te2 2.139(2), B2–C(Bu) 1.637(4), B2–C(IMe) 1.624(5), B2–Te1 2.358(2), B2–Te2 2.331(2), Te1–B1–Te2 102.7(1), B1–

\[\text{Te}–\text{Te} = 102.7(1)\]; 5a: B1–C(Tp) 1.585(5), B1–Se1 1.919(5), B1–Se2 1.921(4), B2–C(Bu) 1.641(7), B2–C(IMe) 1.630(7), B2–Se1 2.115(4), B2–Se2 2.106(4), Se1–B1–Se2 119.4(3), C(Tp)–B1–Se1 120.3(3), C(Tp)–B1–Se2 120.2(3), C(Bu)–B1–C(IMe) 112.7(3), Se1–B1–Se3 105.4(2), B2–Se3–Se2 102.7(1), Se3–Se2–B1 100.7(2).
confirm its connectivity (see ESIF). In these reactions, after crystallization from a THF/pentane mixture, the five-membered rings cyclo-[(tBu)(IMe)B–E–B(Tp)–E] (E = Se (5a) and S (5b)) were obtained (yield 22% in both cases). In the 1,2,4-triselena-3,5-diborolane 5a, the $^{11}$B NMR signals for the two boron atoms were found at 6.2 and 72.1 ppm. Similarly, in 5b, they are found at 7.7 and 66.2 ppm. The five-membered rings 5a and 5b are likely to arise from the initial formation of Se and S analogues of 4, which then undergo ring expansion using the starting materials or reaction intermediates as sacrificial chalcogen sources. While $^{11}$B NMR monitoring of the reaction showed the formation of other boron chalcogenide species – as is predicted from the stoichiometry of the reaction – these byproducts could not be identified. 5a and 5b can however be reliably crystallized from the reaction mixture and isolated.

In the solid state, 5a is structurally comparable to the three previously reported examples of crystallographically characterized $\text{SeSeSeSe}$ non-cluster five-membered rings.$^{15}$ Interestingly, it represents, to our knowledge, the first example of such a cycle with an unsymmetrical substitution pattern. The planar sp$^2$-hybridized TpB site features B$a$ cycle with an unsymmetrical substitution pattern. The obtained a reaction with other borylene sources. Thus, the Group 6 aminoborylene cogenadiboretane rings, we elected to react these precursors allowing us to selectively synthesize unsymmetrical B$a$ rings.

Since the insertion of TpB fragments into 2a and 2b did not allow us to selectively synthesize unsymmetrical B–Se dichalcogenadiboretane rings, we elected to react these precursors with other borylene sources. Thus, the Group 6 aminoborylene complex [N(SiMe$_3$)$_2$]$_3$B(N(SiMe$_3$)$_2$) (6) was reacted with 2a and 2b in toluene. In both cases, four-membered cyclic products, cyclo-[(tBu)(IMe)B–E–B[N(SiMe$_3$)$_2$]–E] (E = Se (7a) and S (7b)), were obtained after recrystallization from a toluene/pentane mixture (yield: 52% (7a) and 38% (7b)). The possibility of ring expansion processes (vide supra), combined with the difficulties in quantitatively crystallizing these products, explains the moderate yields of the reactions.

In 7a, $^{11}$B NMR signals at −13.4 and 46.7 ppm can be assigned to the sp$^2$ and sp$^3$ boron centers, respectively. In the solid state, the 1,3-diselena-2,4-diboretane cycle is nearly planar, as in the case of 4. A slight pyramidization of the N(SiMe$_3$)$_2$ group, as well as dihedral angles of ca. 35° between this group and the B$_2$Se$_2$ and a B–N bond distance of 1.442(4) Å (longer than 1.3549(18) Å in 6), suggest only a small amount of N-to-B π-donation. Consistent with this observation, the (Me$_3$-Si)$_2$NB–Se bond distances (1.957(5) and 1.958(5) Å), while being longer than in 5a, are short enough to suggest some Se-to-B π-donation competing with the N–B multiple bonding. On the other side of the molecule, the (tBu)(IMe)B–Se distances (2.112(5) and 2.116(5) Å) are considerably longer, consistent with the presence of an sp$^3$ boron center. In the case of 7b, $^{11}$B NMR signals were found at −6.1 and 49.8 ppm. The solid-state structure of 7b is comparable to that of 7a, with (Me$_3$Si)$_2$NB–Se bond distances of 1.844(4) and 1.855(3) Å, tBu(Ime)B–Se distances of 1.990(3) and 1.998(4) Å and a B–N bond measuring 1.440(5) Å (Fig. 3).

2a and 2b also react with [Cp(OC)$_3$Mn=B(Bu)] (Cp = cyclopentadienyl) (8),$^{18a}$ a manganese borylene complex that has a rich reactivity$^{12}$ and that is a precursor of 1, 2a and 2b. While we were not able to obtain high-quality crystals from this reaction, HRMS of the isolated products revealed a formula consistent with cyclo-[(tBu)(IMe)B–E–B(Bu)]–E] (E = Se (9a) and S (9b)), arising once again from the transfer of a borylene fragment (tBuB) from 8 to 2a and 2b. The solution $^1$H and $^{13}$C NMR spectra of 9a and 9b are consistent with the formula assignment from HRMS, revealing a 1 : 1 : 1 ratio of IMe and two inequivalent tBu groups. The $^{11}$B NMR resonances of 9a ($\delta$ = −10.8 and 77.5) and 9b ($\delta$ = −3.6 and 72.9) reveal a similar environment for the sp$^3$ boron atom as that in the corresponding four-membered rings 7a and 7b, which leads us to...
postulate that products 9a and 9b are indeed the cyclic compounds \( \text{cyclo-}[([\text{Bu}][\text{Ime}]B-E-B([\text{Bu}]-E)] \).

Finally, 2a was also found to react with the macrocyclic complex \( \text{cyclo-}[([\text{CAAC}]\text{BCN})_4 \) (10). This metal-free tetrameric borylene was previously shown to be a useful source of the corresponding monomer \([([\text{CAAC}]\text{BCN}) \) in its reaction with Lewis bases and to generate of one of the few reported examples of cyclo-\([\text{BSe}]_2 \) complexes by reaction with elemental selenium. Thus, the reaction of 2 and 10 in toluene led to the isolation of the new 1,3-diseleno-2,4-diborolane cyclo-\([([\text{Bu}][\text{Ime}]B-\text{Se}-\text{B(\text{CAAC})CN}-\text{Se}) \) (11) in a 34% yield. While once again unsymmetrically substituted, this compound is the first of the series reported herein to feature two sp\(^3\)-hybridized boron atoms. The \( ^{11}\text{B} \) NMR spectrum of 11 shows two signals at \(-9.5 \) and \(-32.8 \) ppm. While both signals lie within the typical range for sp\(^3\) boron atoms, we can assign the former to the ([Bu][Ime]B group by comparison to 7a and 9a, while the latter is similar to the signals of both isomers of the previously reported cyclo-\([([\text{CAAC}]\text{CN})B-S-B(\text{CN})(\text{CAAC})=S] \) (\( \delta = -31.8 \) and \(-33.5 \) ppm). \(^{14}\)

Rather surprisingly, and in contrast with other 1,3-diseleno-2,4-diborolanes,\(^{6,9,10} \) 11 adopts a clearly bent butterfly structure in the solid state. This situation is plausibly a consequence of the unsymmetrical substitution pattern, which gives rise to a large difference in steric bulk on both faces of the cycle. The B-Se distances are slightly shorter on the (CAAC)(CN)B side (2.067(1) and 2.079(2) \( \AA \)) than around the ([Bu][Ime]B moiety (2.114(1) and 2.114(2) \( \AA \)) (Fig. 4).

**Conclusions**

In conclusion, we demonstrate that borylene fragments can be inserted into the E-E bonds (\( E = \text{Te}, \text{Se} \) and \( S \)) of boradichalcogeniranes. This reaction gives unprecedented synthetic access to unsymmetrically substituted 1,3-dichalcogena-2,4-diborolates and 1,2,4-dichalcogena-3,5-diborolanes. We have thus reported the synthesis of heterocycles featuring unsymmetrical boron centers in sp\(^3\)-sp\(^3\) and sp\(^3\)-sp\(^2\) hybridization combinations. Given the lack of methods for the synthesis of such unsymmetrical compounds, this approach will provide us with a useful platform to study the properties and applications of these unusual heterocycles.

**Conflicts of interest**

There are no conflicts to declare.

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**Notes and references**


