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# A synthetic cyclitol-nucleoside conjugate polyphosphate is a highly potent second messenger mimic†

Wolfgang Dohle,‡<sup>a</sup> Xiangdong Su,‡<sup>a</sup> Stephen J. Mills,‡<sup>a</sup> Ana M. Rossi,<sup>b</sup> Colin W. Taylor<sup>b</sup> and Barry V. L. Potter • \*\*

Reactions that form sec-sec ethers are well known, but few lead to compounds with dense functionality around the O-linkage. Replacement of the  $\alpha$ -glucopyranosyl unit of adenophostin A, a potent D-myoinositol 1,4,5-trisphosphate (IP<sub>3</sub>R) agonist, with a D-chiro-inositol surrogate acting substantially as a pseudosugar, leads to "p-chiro-inositol adenophostin". At its core, this cyclitol-nucleoside trisphosphate comprises an ether linkage between the axial 1-hydroxyl position of p-chiro-inositol and the 3'-hydroxyl group of an adenosine ribose sugar. A divergent synthesis of p-chiro-inositol adenophostin has been achieved. Key features of the synthetic strategy to produce a triol for phosphorylation include a new selective mono-tosylation of racemic 1,2:4,5-di-O-isopropylidene-myoinositol using tosyl imidazole; subsequent conversion of the product into separable camphanate ester derivatives, one leading to a chiral myo-inositol triflate used as a synthetic building block and the other to L-1-O-methyl-myo-inositol [L-(+)-bornesitol] to assign the absolute configuration; the nucleophilic coupling of an alkoxide of a ribose pent-4-ene orthoester unit with a structurally rigid chiral myoinositol triflate derivative, representing the first sec-sec ether formation between a cyclitol and ribose. Reaction of the coupled product with a silvlated nucleobase completes the assembly of the core structure. Further protecting group manipulation, mixed O- and N-phosphorylation, and subsequent removal of all protecting groups in a single step achieves the final product, avoiding a separate N6 protection/deprotection strategy. D-chiro-Inositol adenophostin evoked Ca<sup>2+</sup> release through IP<sub>3</sub>Rs at lower concentrations than adenophostin A, hitherto the most potent known agonist of IP₃Rs.

#### Introduction

Inositols are present in cells from every domain of life (archaea, bacteria and eukaryotes), with *myo*-inositol the most abundant of the nine inositol stereoisomers. Polyphosphorylated derivatives of *myo*-inositol are common in biology and have many functions, including roles in regulating ion-channels, hosphate levels, metabolic flux, transcription, mRNA export and translation, insulin signalling, embryonic development and stress responses. D-myo-Inositol 1,4,5-trisphosphate (IP<sub>3</sub>, 1, Fig. 1), which is an intracellular messenger that evokes Ca<sup>2+</sup> signals after binding to its own receptor (IP<sub>3</sub>R)<sup>1a,9</sup> and higher *myo*-inositol polyphosphates are subjects of intense interest. Ia,10

oxygen at C1 of p-chiro-inositol forms an axial/axial configured

glycosidic bond (see compound 6, Fig. 1).

Such compounds are implicated in diverse areas of biology and

disease.11 myo-Inositol is also a component of glycosyl-phos-

phatidylinositol (GPIs, see compound 2, Fig. 1), <sup>12</sup> which anchor proteins to the cell surface and are present in all eukaryotic cells

and some bacteria.13 Other stereoisomers of inositol also have biological activities,1a but these are less common or have not been widely investigated.14 Beyond GPIs, derivatives of myoinositol can make simple conjugated derivatives with sugars; these are much less common and only a few natural products are known. In general, an equatorial hydroxyl group of an inositol is linked to the glycosidic C1 of a sugar. Examples of the rarer inositol derivatives are fagopyritol (3 & 4) and pinitol derivatives (5 & 6) which contain a p-chiro-inositol-galactose conjugate, and are isolated from various plant sources (Fig. 1).15 Most of the conjugates have a galactose unit connected via its glycosidic C1 to an equatorially configured oxygen at either C3 (3), C2 (4 & 5) or C5 (not shown) of the D-chiroinositol unit, with further conjugates sometimes added at the primary C6 hydroxyl group of galactose (see compounds 3 & 4, Fig. 1). Only one example was found where the axial configured

<sup>&</sup>lt;sup>a</sup>Medicinal Chemistry & Drug Discovery, Department of Pharmacology, University of Oxford, Mansfield Road, Oxford, OX1 3QT, UK. E-mail: barry.potter@pharm.ox.ac. uk; Tel: +44-1865-271945

<sup>&</sup>lt;sup>b</sup>Department of Pharmacology, University of Cambridge, Tennis Court Road, Cambridge, CB2 1PD, UK

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<sup>‡</sup> Equal contribution.

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OH 
$$(HO)_2PO$$
  $OP(OH)_2$   $OP(OH)$ 

Fig. 1 D-myo-Inositol 1,4,5-trisphosphate (IP<sub>3</sub>) 1, CD52 GPI anchor 2 and natural products 3-6 containing D-myo- and D-chiro-inositol sugar conjugates (inositol component shown in blue).

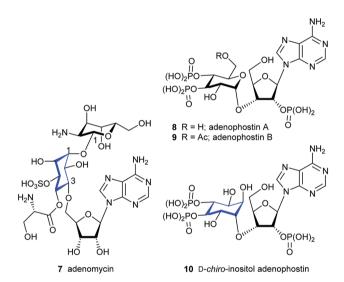


Fig. 2 Adenomycin 7, adenophostin A 8, adenophostin B 9 and D-chiro-inositol adenophostin 10.

Complex, but feasible, synthetic routes to such compounds and GPIs require conjugation at the sugar C1 glycosidic centre. GPIs are sugar conjugates, where a p-myo-inositol derivative is usually oxygen-linked from its C6 position to a p-glucosamine unit at its glycosidic C1 position. A phospholipid moiety at C1 of the D-myo-inositol unit enables GPIs to insert into the plasma membrane. The difficulty of isolating and purifying GPIs from biological sources lead to the development of methodologies for the synthesis of GPIs, GPI-anchored proteins and glycoproteins.16-19 To our knowledge, sec-sec inositol-hexose sugar conjugates are rare, apart from those at the glycosidic centre. Only one example of a pseudo-disaccharide could be found<sup>20</sup> and there were no examples of inositol-pentose sugar conjugates. However, one example of a prim-sec ether linkage is known, for the antibiotic adenomycin 7 (Fig. 2).21 This compound contains one adenosine unit linked through its

primary C5' hydroxyl group to the C3 position of an *L-chiro*-inositol derivative to form a *prim-sec* ether bond. Additionally, a hemi-acetal is located at the glycosidic C1 of an *L*-gulosamine linked to the *L-chiro*-inositol C1. Interestingly, contrary to glycosidic connections in GPIs and in 3–6, the latter linkage in 7 is reported to be β-glycosidic.<sup>21</sup> Given the extensive biology of *myo*-inositol sugar conjugates and growing interest in the other eight inositol isomers, <sup>1α</sup> there is a need to generate analogues to construct *prim-sec* or more particularly *sec-sec* ether linkages. In the latter case, an example of particular interest can be drawn from the Ca<sup>2+</sup>-signalling field in the form of the adenophostins 8 & 9 (Fig. 2).<sup>22</sup> These are naturally occurring fungal metabolites with a phosphorylated glucose component that resembles part of the IP<sub>3</sub> pharmacophore and which are more potent than IP<sub>3</sub> 1 in evoking Ca<sup>2+</sup> release through IP<sub>3</sub> receptors (IP<sub>3</sub>Rs).

IP<sub>3</sub>Rs are intracellular Ca<sup>2+</sup> channels that allow IP<sub>3</sub>, produced when extracellular stimuli promote hydrolysis of phosphatidylinositol 4,5-bisphosphate, to release Ca<sup>2+</sup> from intracellular stores.23 The resulting Ca2+ signals regulate diverse cellular activities.24 Opening of the Ca2+-permeable channel of the IP<sub>3</sub>R is initiated when IP<sub>3</sub> binds to the IP<sub>3</sub>-binding core (IBC, residues 224-604) of each of the four subunits of the tetrameric IP<sub>3</sub>R (Fig. 3A and B). 25,26 Structure-activity analyses, 27 structures of N-terminal fragments of IP<sub>3</sub>R<sup>25,28,29</sup> and of complete IP<sub>3</sub>R,9b alongside mutagenesis studies have established that IP3 initiates IP3R activation by causing partial closure of the clam-like IBC. The 1- and 5-phosphates of IP3 interact primarily with residues in the α-domain of the IBC, while the 4-phosphate interacts with IBC-β (Fig. 3A). Adenophostin A (8) and adenophostin B (9) bind to IP<sub>3</sub>R with greater affinity than IP<sub>3</sub> and they are  $\sim$ 6-10-fold more potent than IP<sub>3</sub> in evoking Ca<sup>2+</sup> release.<sup>30,31</sup> Hitherto, adenophostin A is the most potent known agonist of IP<sub>3</sub>R.<sup>31</sup> Our model for the interaction of adenophostin A with the IBC,32 which is supported by mutagenesis33 and SAR analyses, 34,35 suggests that the glucose 3",4"-bisphosphate motif of adenophostin A mimics the 4,5-bisphosphate motif of IP3, while

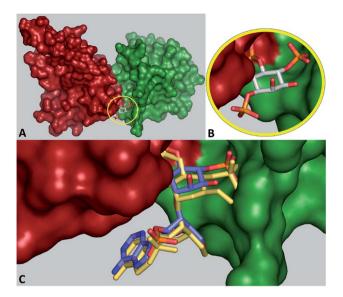


Fig. 3 (A) IP<sub>3</sub>-binding core (IBC) of IP<sub>3</sub> receptors with IP<sub>3</sub> 1 bound (PDB 1N4K). The IBC- $\alpha$  is shown in red, the IBC- $\beta$  in green and IP<sub>3</sub> 1 in stick representation (carbon atoms in grey). (B) Close-up view of (A). (C) Model of D-chiro-inositol adenophostin 10 (carbons in light blue stick representation) with adenophostin A 8 (all atoms in yellow stick representation) binding to the IBC.

the 2"-hydroxyl and 2'-phosphate of adenophostin A mimic the non-essential 6-hydroxyl and 1-phosphate groups of  $\mathrm{IP}_3$ .  $^{25,36}$  A recently published cryo-EM structure of  $\mathrm{IP}_3\mathrm{R1}$  with and without adenophostin A bound  $^{37}$  hints at substantially different binding modes for adenophostin A. However, the resolution of ligands in such structures is far from optimal and the ligand structure here is chemically incorrect.

We focused on developing a synthetic route to "p-chiro-inositol adenophostin" **10**. The aim was to replace the  $\alpha$ -glucopyranosyl unit of **8** with the most similarly-configured inositol

unit (D-chiro-inositol, acting substantially as a pseudo-sugar). The resulting ligand would allow us to study any potentially novel effects of this modification on Ca2+ release. Since the phosphorylated inositol-nucleoside hybrid closely resembles adenophostin A, it would establish whether nature evolved a phosphorylated glucose moiety as the best mimic of the inositol bisphosphate motif or whether the activity of adenophostin A can be still further improved by replacing glucose with a motif more similar to the cyclitol in IP3. Finally, and importantly, if high adenophostin A-like potency were achieved in 10, the hybrid would offer an axial hydroxyl group at the Dchiro-inositol C6 position (C2 position in myo-inositol), not present in adenophostin A. This position could be modified (e.g. via extended substituents) to target the cleft between the IBC domains (Fig. 3B) and might facilitate the design of highaffinity ligands as potential antagonists by virtue of their ability to prevent closure of the IBC clam. Such a strategy cannot be envisaged for adenophostin A.

Examining all the bonds that need to be formed to assemble 10, it became clear that its total synthesis would be challenging, especially the formation of the sec-sec ether bond between the inositol ring and the ribose sugar. Linkages are usually made between a reactive sugar glycosyl acceptor at the 1-position and inositol, both in protected form, to give inositol-sugar conjugates after deprotection (Fig. 1). Non-glycosidic conjugates of this type are quite rare. However, some years ago in the synthesis of a pseudo-disaccharide a cyclitol derivative was used as the nucleophile and a triflate on a rigid sugar unit as the electrophile.20 This strategy would not easily work in our case because a p-chiro-inositol derivative would need to react as a nucleophile ideally with a triflate at the 3-position of a xylose derivative. S<sub>N</sub>2 reactions using 3-trifluoromethanesulfonyl- and 3-tosyl-xylose derivatives are described, but coupled products are only achieved when using non-alkaline nucleophiles like azide or thiophenolate.38,39 With the hydrogen at C4 in xylose placed almost anti-periplanar to the leaving group, the less

Scheme 1 Retrosynthetic analysis of p-chiro-inositol adenophostin 10.

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Fig. 4 Triflates that undergo elimination.

Scheme 2 Synthesis of the D-ribose orthoester derivative 15.

reactive 3-tosyl derivative undergoes mainly elimination to form an olefin when reacted with alkoxides, but cleavage of the tosyl group is also observed.<sup>39</sup>

We report here a divergent strategy for the synthesis of **10** that includes the first *sec–sec* ether formation between a cyclitol and a ribose sugar as one key feature. Others are a selective mono-tosylation of a racemic inositol diol derivative and subsequent conversion of the product into diastereoisomeric camphanate esters separable by column chromatography. One diastereoisomer provides the chiral inositol triflate to be used as a building block, and the other provides a route to L-(+)-bornesitol in order to confirm absolute configurations. Coupling of the inositol triflate with the ribose alkoxide of

a pent-4-ene orthoester derivative and subsequent reaction with a silylated nucleobase assembles the core structure for phosphorylation. Pan-phosphorylation of all hydroxyl and amino groups and removal of all protecting groups inventively in a single step affords the final product **10** without recourse to an adenine *N6*-protection strategy.

#### Results and discussion

#### Chemistry

The retrosynthetic analysis of *p-chiro*-inositol adenophostin **10** (Scheme 1), leads to **14** as the first compound containing all three components, *p-chiro*-inositol, ribose and a purine base as derivatives. An initial analysis for the formation of the *sec-sec* ether suggested that *myo*-inositol triflates may be used. Three such potential triflates were synthesised in turn, with increasing rigidity of the inositol ring (**18–20**, Fig. 4).

Initially, although triflate 18 in the chair conformation did not have an anti-periplanar H-Ins-OTf motif, the triflate eliminated to form a double bond in the presence of an alkoxide, demonstrating that the inositol framework needed to be more rigid. <sup>13</sup>C NMR assignment indicated a quaternary carbon derived from proton abstraction located at  $\delta_{\rm C} = 154.1$  and an olefinic C-H at  $\delta_{\rm C}=97.0$ . The rigidity of the inositol ring was further increased to give derivative 19. However, elimination in the presence of base was still observed and similar olefinic fingerprint signals in the <sup>13</sup>C NMR were seen as above. A still more constrained structure was an obvious further choice, so no elimination should occur. A third compound 20, provided a more rigid inositol triflate and the crude product from alkoxide treatment was purified by column chromatography. However, elimination still occurred and two products were observed in the <sup>1</sup>H NMR spectrum. After a number of inositol triflates eliminated to form cyclic vinyl ethers, triflate 16 (and the most difficult to manipulate to the final product) was prepared (Scheme 5) and successfully used as a functionalised chiral myo-inositol intermediate for the preparation of the Dchiro-inositol derivative 14. An inositol triflate with a similar protecting group pattern has been previously reported in S<sub>N</sub>2

Scheme 3 Synthesis of protected L-myo-inositol 3-O-camphanate 29 and D-myo-inositol 3-O-camphanate 30.

Scheme 4 Synthesis of protected L-myo-inositol 3-O-triflate derivative 16 and L-(+)-bornesitol 33.

reactions. However, only non-basic nucleophiles such as fluoride, azide and thioacetate were used, but no alkoxides.<sup>40</sup>

Although the formation of sec-sec ethers is well known, only a few examples lead to densely functionalised compounds.  $^{20,41,42}$  The main problem in the required  $S_N2$  type reaction is the use of highly substituted, sterically hindered starting materials. A limited number of electrophiles, such as a triflate within a rigid organic framework have been used successfully.20,41 Compound 14 could be assembled in two steps from the ribose orthoester derivative 15. Note that although 15 is depicted as one isomer at the orthoester moiety, which is indeed the case, in fact the exact stereochemistry at the quaternary carbon has not been proven. Diverse attempts to derivatise 15 and 23 to give a crystalline product for X-ray studies were made, but all of these failed. Compound 15 was coupled to the L-myo-inositol triflate derivative 16. Subsequent anomeric coupling of the orthoester group with silylated 6-chloropurine 17 gave compound 14.

The ribose fragment of **10** was synthesised in four steps from D-(-)-ribose **21** to give the pent-4-ene orthoester **22**.<sup>43</sup> Reaction of **22** with sodium methoxide in methanol furnished diol **23** in

high yield.<sup>43</sup> Selective benzylation of the primary hydroxyl group using silver carbonate and benzyl bromide in toluene gave compound **15** in good yield (Scheme 2).<sup>44</sup>

Synthesis of the chiral L-myo-inositol triflate derivative 16 started from racemic 1,2:4,5-di-O-isopropylidene-myo-inositol 24<sup>45</sup> (Schemes 3 and 4). In order to convert an L-myo-inositol derivative into the D-chiro-inositol derivative via an S<sub>N</sub>2 reaction a triflate was required as a leaving group at C3. The hydroxyl group at C6 was protected as a p-methoxybenzyl (PMB) ether, but selective protection at the more reactive 3hydroxyl group was required. Selective mono-tosylation of the hydroxyl group at C3 of 24 using tosyl chloride in pyridine lead to low yields of compound 25.46 However, a new method was devised for this conversion using tosyl imidazole<sup>47</sup> in the presence of cesium fluoride in DMF at room temperature. Compound 25 was obtained in good yield along with a minor amount of the bis-tosylated product 26. Reaction of 25 with sodium hydride and p-methoxybenzyl chloride in DMF gave compound 27 in high yield. Subsequent removal of the tosyl group was achieved by dissolving 27 in dichloromethane followed by the addition of methanol and magnesium turnings.48 Compound 28 was treated with (1S)-(-)-camphanic

Scheme 5 Synthesis of mono-benzylated compound 36.

P(OBn)<sub>2</sub> 0

38. 64%

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MeOH/H<sub>2</sub>O 9:1, 70 °C, 18 h.

Scheme 6 Synthesis of D-chiro-inositol adenophostin 10

**10**. up to 37%

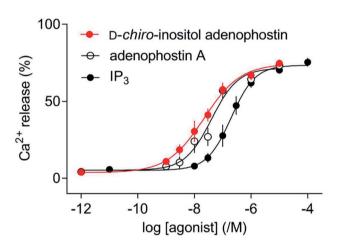


Fig. 5 D-chiro-Inositol adenophostin 10 is more potent than adenophostin A  $\bf 8$  in evoking Ca<sup>2+</sup> release through IP<sub>3</sub>R. Results (%, means  $\pm$  SEM, n=6) show the Ca<sup>2+</sup> release evoked by the indicated concentrations of IP<sub>3</sub> 1, adenophostin A 8 or p-chiro-inositol adenophostin 10 from the intracellular stores of permeabilized HEK cells expressing IP<sub>3</sub>R1.

chloride49 to afford diastereoisomers 29 and 30, which were separated by column chromatography and both were isolated in good yield (Scheme 3).

The absolute configuration was determined using diastereoisomer 30, which was converted in several steps via ester hydrolysis, methylation and protecting group removal via 31 and 32 into the known natural product L-(+)-bornesitol 33; the configuration of this product was confirmed by comparison to previously reported <sup>1</sup>H NMR and optical rotation data.<sup>50</sup> The chiral myo-inositol triflate 16 was then synthesised in good overall yield from 29 by ester hydrolysis and reaction of chiral alcohol 34 with triflic anhydride and pyridine in dichloromethane (Scheme 4).

Compound 15 was first dissolved in anhydrous THF/HMPA before adding sodium hydride to generate the alkoxide. The chiral triflate 16 was then added to the alkoxide to give 35 in 86% yield, transforming the L-myo-inositol unit into the desired D-chiro-inositol derivative by inversion of configuration at C3. DMPU could also be used to replace HMPA as a co-solvent, but gave overall slightly lower yields (about 10% less). However, no reaction was observed when DMF alone was used as a solvent. Subsequent coupling of 35 via its pent-4-ene orthoester group

Table 1 Effects of IP<sub>3</sub>, adenophostin A 8 and p-chiro-inositol adenophostin 10 on  $Ca^{2+}$  release from the intracellular stores of permeabilized HEK-IP<sub>3</sub>R1 cells<sup>a</sup>

Compound	Ca <sup>2+</sup> release			
	$\mathrm{pEC}_{50}$	$EC_{50}$ $(nM)$	(%)	h
IP <sub>3</sub> (1)	$6.73\pm0.13$	186	$73\pm2$	$1.2\pm0.1$
Adenophostin A (8)	$7.45 \pm 0.16$	35	$71\pm3$	$1.5\pm0.4$
D-chiro-Inositol adenophostin (10)	$\textbf{7.67} \pm \textbf{0.14}$	21	$72\pm3$	$1.4\pm0.6$

<sup>&</sup>lt;sup>a</sup> Results are means ± SEM (Ca<sup>2+</sup> release (%), Hill coefficient (h) and pEC<sub>50</sub>) or means (EC<sub>50</sub>, half maximal effective concentration) from six independent experiments each performed in duplicate. There were no significant differences between the ligands in the values for Ca<sup>2+</sup> release (%) or h. The pEC<sub>50</sub> values were significantly different (P < 0.05), for IP<sub>3</sub> vs. adenophostin A, IP<sub>3</sub> vs. p-chiro-inositol adenophostin and p-chiro-inositol adenophostin adenophostin and p-chiro-inositol adenophostin a inositol adenophostin vs. adenophostin A.

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with silylated 6-chloropurine<sup>51</sup> in the presence of an activator under anhydrous conditions gave compound 14 in good yield. 43b The trans-isopropylidene group was then selectively removed by treatment with ethylene glycol and p-toluenesulfonic acid in dichloromethane for no longer than 20 minutes to give compound 13 in almost quantitative yield. After various attempts, we found that a one-pot reaction procedure strategy was unsuitable to mono-benzylate compound 13, since the harsh reaction conditions resulted in complex mixtures and the desired compound 36 was not observed.<sup>52</sup> Selective monobenzylation of 13 was best achieved in a two-stage procedure.53 First, 13 was reacted with n-Bu<sub>2</sub>SnO in refluxing acetonitrile using a Soxhlet condenser containing activated 3 Å MS. The resulting tin acetal was isolated and directly converted into the desired mono-benzylated compound 36 in 26% yield using benzyl bromide, cesium fluoride and tetra-n-butylammonium iodide in anhydrous DMF. However, no regioselectivity was observed and the other mono-benzylated compound 37 was isolated in a similar yield together with unconverted starting

material 13 (Scheme 5). Small samples of both chromatographically separable regioisomers 36 and 37 were treated with TFA/DCM to remove the p-methoxybenzyl group and the products then analysed by <sup>1</sup>H NMR and COSY without and with the addition of D<sub>2</sub>O. All hydroxyl protons and the corresponding adjacent cyclitol ring protons were identified. The 1,2-diol relationship was indicated by a strong coupling between these two adjacent ring protons. For the 1,3-diol relationship that coupling was not found. Compound 36 led to a 1,2-diol, whereas 37 led to a 1,3-diol. Therefore, 36 was identified as the required 2-O-benzylated derivative that was then treated with ammonia in ethanol at 70 °C to give 12 in 69% yield with concomitant removal of the benzoyl ester (Scheme 6). After trying various methods we found that removal of the PMB protecting group was best achieved using TFA in anhydrous dichloromethane for no longer than 5 minutes.54 Longer reaction times and/or traces of moisture usually lead to increasing amounts of more polar by-products where the cis-isopropylidene protecting group had also been removed. For the final phosphorylation of triol 11 we had originally envisaged use of imidazolium triflate33,35d as a means of achieving O-selective phosphitylation over N-phosphitylation. While this approach has been used well in the past, we found it to be very cumbersome for the phosphorylation of low milligram quantities of triol. It was not possible to achieve complete selective phosphitylation and to completely exclude water, then titrate in the required amount of phosphitylating reagent without achieving substantial amounts of concomitant Nphosphitylation, although products from this could be easily separated. We therefore reasoned that since *N*-phosphates have been reported to be labile under acidic and/or hydrogenation reaction conditions<sup>55</sup> such a contaminant should therefore convert into the free amino group during the final step of our synthesis. This idea thus argued for a pan-phosphitylation approach, whereby 11 was first phosphorylated with an excess of reagent that easily leads to the tris-O-phosphorylated mono-N-phosphorylated product that could be converted to the mixed tris-phosphate/mono-phosphoramidate 38 upon oxidation. To

illustrate this further, the conversion of 11 to the fully deblocked 10 was carried out on a <10 mg scale. Compound 11 was treated with 5-phenyl-1H-tetrazole and dibenzyl diisopropylphosphoramidite<sup>56</sup> in dichloromethane followed by oxidation of the product with t-butyl hydroperoxide to achieve the fully protected compound 38 in good overall yield (Scheme 6). The N-phosphate group was readily distinguished from the Ophosphate by its broad <sup>31</sup>P NMR resonance peak at  $\delta$  –0.9 in comparison to sharp peaks at  $\delta$  –1.2 and –1.5 (two overlapping peaks) respectively. Subsequent treatment of 38 with hydrogen generated from cyclohexene in the presence of palladium hydroxide on carbon in methanol and water at 70 °C overnight led to an acidic reaction mixture (pH 3), as all O- and N-benzylated phosphate groups were converted to their free acids. Subsequently, this also lead to removal of the cis-isopropylidene group and degradation of the in situ generated acid-labile free N-phosphate by nucleophilic cleavage of its P-N bond. 55 As a pleasing result, therefore, all different types of protecting group were removed in a single step to give D-chiro-inositol adenophostin 10 (Scheme 6). Yields of 10 were routinely quite variable and possibly complicated by retention of compound on the hydrogenation catalyst. The best yield achieved so far is 37%. We imagine that this mixed O- and N-phosphitylation methodology, avoiding a sometimes cumbersome extra Nprotection strategy, will find application in many nucleotide synthesis situations where only small amounts of compound are available. For biological evaluation, 10 was purified by semiprep HPLC (purity > 99%) and quantified by UV spectroscopy. Final structural characterisation was additionally confirmed by <sup>1</sup>H-<sup>31</sup>P NMR correlation (see ESI†).

#### **Biology**

D-chiro-Inositol adenophostin **10** was investigated as an agonist at the IP<sub>3</sub>R using an intracellular Ca<sup>2+</sup> mobilisation assay in comparison to adenophostin A **8** and IP<sub>3</sub> **1**. We used a low-affinity Ca<sup>2+</sup> indicator within the intracellular stores of permeabilized HEK cells expressing IP<sub>3</sub>R1 to record Ca<sup>2+</sup> release evoked by synthetic ligands. Maximally effective concentrations of IP<sub>3</sub>, adenophostin A **8** or D-chiro-inositol adenophostin **10** caused release of a similar fraction ( $\sim$ 70%) of the Ca<sup>2+</sup> sequestered by intracellular stores. However, adenophostin A **8** and **10** were 5.3 and 8.9-fold more potent than IP<sub>3</sub>, respectively. Moreover, D-chiro-inositol adenophostin **10** was 1.7-fold more potent than adenophostin A **8** (Fig. 5 and Table 1).

#### Conclusion

In summary, a concise synthesis of D-chiro-inositol adenophostin has been achieved. Key features of the strategy include a new selective mono-tosylation reaction of a racemic *myo*inositol diol derivative and subsequent elaboration of the product into separable camphanate derivatives of a fully protected intermediate. One such diastereoisomer is converted to L-(+)-bornesitol to confirm the absolute configuration and the other leads to the required chiral *myo*-inositol triflate that is used as a synthetic building block. Critical formation of a *sec*-

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sec ether used a rigidly structured chiral inositol triflate with an alkoxide of a suitable protected ribose derivative; subsequent reaction of the fully protected coupled product with a silylated nucleobase assembled the core structure and base amination and protecting group manipulations finally afforded a triol for phosphorylation. After phosphitylation of both the adenine amino group and inositol hydroxyl groups, removal of all protecting groups was accomplished in a single step to afford the final product. Thus, replacement of the α-glucopyranosyl unit in adenophostin A with the equivalent p-chiro-inositol motif leads to the trisphosphate D-chiro-inositol adenophostin 10. 10 Was evaluated as an agonist for intracellular Ca2+ release through the IP<sub>3</sub>R and its EC<sub>50</sub> of 21 nM was lower than that of 8 (35 nM). It is the most potent agonist of IP<sub>3</sub>Rs so far identified and may offer new opportunities to develop high-affinity antagonists of IP<sub>3</sub>R.

#### Conflicts of interest

There are no conflicts to declare.

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#### Notes and references

- 1 (a) M. P. Thomas, S. J. Mills and B. V. L. Potter, Angew. Chem., Int. Ed., 2016, 55, 1614; (b) B. V. L. Potter and D. Lampe, Angew. Chem., Int. Ed., 1995, 34, 1933; (c) R. F. Irvine and M. J. Schell, Nat. Rev. Mol. Cell Biol., 2001, 2, 327; (d) M. D. Best, H. Zhang and G. D. Prestwich, Nat. Prod. Rep., 2010, 27, 1403; (e) A. J. Hatch and J. D. York, Cell, 2010, **143**, 1030; (f) A. Chakraborty, S. Kim and S. H. Snyder, Sci. Signaling, 2011, 4, re1; (g) G. E. Gillaspy, New Phytol., 2011, **192**, 823; (h) J.-Y. Lee, Y.-r. Kim, J. Park and S. Kim, Ann. N. Y. Acad. Sci., 2012, 1271, 68; (i) M. L. Croze and C. O. Soulage, Biochimie, 2013, 95, 1811.
- 2 (a) R. H. Michell, Biochem. Soc. Symp., 2007, 74, 223; (b) R. H. Michell, Nat. Rev. Mol. Cell Biol., 2008, 9, 151; (c) R. H. Michell, Adv. Enzyme Regul., 2011, 51, 84.
- 3 (a) J. B. Parys and H. De Smedt, Adv. Exp. Med. Biol., 2012, 740, 255; (b) S. B. Shears, S. B. Ganapathi, N. A. Gokhale, T. M. H. Schenk, H. Wang, J. D. Weaver, A. Zaremba and Y. Zhou, Subcell. Biochem., 2012, 59, 389.
- 4 A. Saiardi, Adv. Biol. Regul., 2012, 52, 351.
- 5 W. F. Boss and Y. J. Im, Annu. Rev. Plant Biol., 2012, 63, 409.
- 6 A. R. Alcázar-Román and S. R. Wente, Chromosoma, 2008,
- 7 M. M. Tsui and J. D. York, Adv. Enzyme Regul., 2010, 50, 324.
- 8 (a) R. Valluru and W. Van den Ende, Plant Sci., 2011, 181, 387; (b) R. Valluru and W. Van den Ende, Plant Sci., 2012, **185-186**, 340.

- 9 (a) M. J. Berridge, P. Lipp and M. D. Bootman, Nat. Rev. Mol. Cell Biol., 2000, 1, 11; (b) N. Paknejad and R. K. Hite, Nat. Struct. Mol. Biol., 2018, 25, 660.
- 10 (a) M. P. Thomas and B. V. L. Potter, FEBS J., 2014, 281, 14; (b) M. S. Wilson, T. M. Livermore and A. Saiardi, Biochem. J., 2013, 452, 369.
- 11 M. J. Berridge, Physiol. Rev., 2016, 96, 1261.
- 12 B. Antonsson, Biochim. Biophys. Acta, Lipids Lipid Metab., 1997, 1348, 179.
- 13 T. Kinoshita, J. Lipid Res., 2016, 57, 4.
- 14 (a) B. L. Turner, M. J. Papházy, P. M. Haygarth and I. D. McKelvie, Philos. Trans. R. Soc., B, 2002, 357, 449; (b) B. L. Turner, A. W. Cheesman, H. Y. Godage, A. M. Riley and B. V. L. Potter, Environ. Sci. Technol., 2012, 46, 4994.
- 15 (a) M. Horbowicz, P. Brenac and R. L. Obendorf, Planta, 1998, 205, 1; (b) R. L. Obendorf, K. J. Steadman, D. J. Fuller, M. Horbowicz and B. A. Lewis, Carbohydr. Res., 2000, 328, 623; (c) K. J. Steadman, M. S. Burgoon, R. L. Schuster, B. A. Lewis, S. E. Edwardson and R. L. Obendorf, J. Agric. Food Chem., 2000, 48, 2843; (d) K. J. Steadman, D. J. Fuller and R. L. Obendorf, Carbohydr. Res., 2001, 331, 19; (e) M. B. Cid, F. Alfonso and M. Martín-Lomas, Carbohydr. Res., 2004, 339, 2303; (f) N. Yang and G. Ren, J. Agric. Food Chem., 2008, 56, 757; (g) N. Yang and G. Ren, J. Agric. Food Chem., 2008, 56, 761; (h) W. Gui, B. A. Lemley, I. Keresztes, A. M. Condo Jr, K. J. Steadman and R. L. Obendorf, Carbohydr. Res., 2013, 380, 130; (i) D. Donati, L. R. Lampariello, R. Pagani, R. Guerranti, G. Cinci and E. Marinello, Phytother. Res., 2005, 19, 1057; (j) L. B. Lahuta and T. Dzik, J. Plant Physiol., 2011, 168, 352.
- 16 (a) Z. Guo, Curr. Org. Synth., 2013, 10, 366; (b) M. G. Paulick and C. R. Bertozzi, Biochemistry, 2008, 47, 6991.
- 17 Z. Guo and L. Bishop, Eur. J. Org. Chem., 2004, 3585.
- 18 A. V. Nikolaev and N. Al-Maharik, Nat. Prod. Rep., 2011, 28,
- 19 (a) J. Xue and Z. Guo, J. Am. Chem. Soc., 2003, 125, 16334; (b) X. Wu and Z. Guo, Org. Lett., 2007, 9, 4311.
- 20 H. Paulsen and B. Mielke, Liebigs Ann. Chem., 1990, 169.
- 21 (a) T. Ogita, N. Otake, Y. Miyazaki and H. Yonehara, Tetrahedron Lett., 1980, 21, 3203; (b) K. Isano, J. Antibiot., 1988, 41, 1711.
- 22 M. Takahashi, T. Kagasaki, T. Hosoya and S. Takahashi, J. Antibiot., 1993, 46, 1643.
- 23 M. J. Berridge, Nature, 1993, 361, 315.
- 24 K. Mikoshiba, Adv. Biol. Regul., 2015, 57, 217.
- 25 I. Bosanac, J.-R. Alattia, T. K. Mal, J. Chan, S. Talarico, F. K. Tong, K. I. Tong, F. Yoshikawa, T. Furuichi, M. Iwai, T. Michikawa, K. Mikoshiba and M. Ikura, Nature, 2002, 420, 696.
- 26 K. J. Alzayady, L. Wang, R. Chandrasekhar, L. E. Wagner II, F. Van Petegem and D. I. Yule, Sci. Signaling, 2016, 9, ra35.
- 27 A. M. Rossi and C. W. Taylor, J. Cell Sci., 2019, 132, jcs222463.
- 28 M.-D. Seo, S. Velamakanni, N. Ishiyama, P. B. Stathopulos, A. M. Rossi, S. A. Khan, P. Dale, C. Li, J. B. Ames, M. Ikura and C. W. Taylor, Nature, 2012, 483, 108.

29 K. Hamada, H. Miyatake, A. Terauchi and K. Mikoshiba, *Proc. Natl. Acad. Sci. U. S. A.*, 2017, 114, 4661.

**Chemical Science** 

- 30 M. Takahashi, K. Tanzawa and S. Takahashi, *J. Biol. Chem.*, 1994, **269**, 369.
- 31 A. M. Rossi, A. M. Riley, B. V. L. Potter and C. W. Taylor, *Current Topics in Membranes Series Elsevier*, ed. I. Serysheva, Academic Press, Burlington, 2010, vol. 66, pp. 209–233.
- 32 H. J. Rosenberg, A. M. Riley, A. J. Laude, C. W. Taylor and B. V. L. Potter, *J. Med. Chem.*, 2003, 46, 4860.
- 33 K. M. Sureshan, A. M. Riley, A. M. Rossi, S. C. Tovey, S. G. Dedos, C. W. Taylor and B. V. L. Potter, *Chem. Commun.*, 2009, 45, 1204.
- 34 E. P. Nerou, A. M. Riley, B. V. L. Potter and C. W. Taylor, *Biochem. J.*, 2001, 355, 59.
- 35 (a) F. Chrètien, N. Moitessier, F. Roussel, J.-P. Mauger and Y. Chapleur, *Curr. Org. Chem.*, 2000, 4, 513; (b) H. Hotoda, K. Murayama, S. Miyamoto, Y. Iwata, M. Takahashi, Y. Kawase, K. Tanzawa and M. Kaneko, *Biochemistry*, 1999, 38, 9234; (c) V. A. Correa, A. M. Riley, S. Shuto, G. Horne, E. P. Nerou, R. D. Marwood, B. V. L. Potter and C. W. Taylor, *Mol. Pharmacol.*, 2001, 59, 1206; (d) K. M. Sureshan, A. M. Riley, M. P. Thomas, S. C. Tovey, C. W. Taylor and B. V. L. Potter, *J. Med. Chem.*, 2012, 55, 1706.
- 36 S. T. Safrany, R. J. H. Wojcikiewicz, J. Strupish, S. R. Nahorski, D. Dubreuil, J. Chleophax, S. D. Gero and B. V. L. Potter, *FEBS Lett.*, 1991, 278, 252.
- 37 G. Fan, M. R. Baker, Z. Wang, A. B. Seryshev, S. J. Ludtke, M. L. Baker and I. I. Serysheva, *Cell Res.*, 2018, 28, 1158.
- 38 L. R. Comstock and J. M. Denu, *Org. Biomol. Chem.*, 2007, 5, 3087.
- 39 (a) A. K. Sanki and T. Pathak, *Tetrahedron*, 2003, 59, 7203; (b)
   A. K. Sanki and T. Pathak, *Synlett*, 2002, 1241.
- 40 F. Tagliaferri, S.-N. Wang, W. K. Berlin, R. A. Outten and T. Y. Shen, *Tetrahedron*, 1991, 47, 1.
- 41 I. Cumpstey, Carbohydr. Res., 2009, 344, 2285.
- 42 (a) H. Paulsen and W. von Deyn, *Liebigs Ann. Chem.*, 1987, 141; (b) T. Akhtar and I. Cumpstey, *Tetrahedron Lett.*, 2007, 48, 8673; (c) T. Akhtar, L. Eriksson and I. Cumpstey, *Carbohydr. Res.*, 2008, 343, 2094.
- 43 (a) C. V. S. Ramamurty, P. Ganney, C. S. Rao and B. Fraser-Reid, *J. Org. Chem.*, 2011, 76, 2245; (b) B. Fraser-Reid, P. Ganney, C. V. S. Ramamurty, A. M. Gómez and J. C. López, *Chem. Commun.*, 2013, 49, 3251.

- 44 (a) W. Koenigs and E. Knorr, Chem. Ber., 1901, 34, 957; (b)
  X. Zhu and R. R. Schmit, Angew. Chem., Int. Ed., 2009, 48, 1900; (c) D. Crich, J. Org. Chem., 2011, 76, 9193; (d)
  Z. Wimmer, L. Pechova and D. Saman, Molecules, 2004, 9, 902.
- 45 (a) J. Gigg, R. Gigg, S. Payne and R. Conant, Carbohydr. Res., 1985, 142, 132; (b) S.-K. Chung and Y. Ryu, Carbohydr. Res., 1994, 258, 145.
- 46 D. R. Gauthier Jr and S. L. Bender, *Tetrahedron Lett.*, 1996, 37, 13.
- 47 G. Reyes-Rangel, J. Vargas-Caporali and E. Juaristi, *Tetrahedron*, 2016, 72, 379.
- 48 M. Sridhar, B. A. Kumar and R. Narender, *Tetrahedron Lett.*, 1998, 39, 2847.
- 49 H. Gerlach, Helv. Chim. Acta, 1985, 68, 1815.
- 50 (a) J. Gigg, R. Gigg, S. Payne and R. Conant, J. Chem. Soc., Perkin Trans. 1, 1987, 1757; (b) A. M. Riley, H. Wang, S. B. Shears and B. V. L. Potter, Chem. Commun., 2015, 51, 12605.
- 51 L. D. Nord, N. K. Dalley, P. A. McKernan and R. K. Robins, J. Med. Chem., 1987, 30, 1044.
- 52 (a) E. A. Mash, L. T. A. Kantor and S. C. Waller, *Synth. Commun.*, 1997, 27, 507; (b) V. Saikam, S. Dara, M. Yadav, P. P. Singh and R. A. Vishwakarma, *J. Org. Chem.*, 2015, 80, 11916; (c) M. Giordano and A. Iadonisi, *J. Org. Chem.*, 2014, 79, 213.
- 53 (a) S. David and S. Hanessian, *Tetrahedron*, 1985, 41, 643; (b)
  D. Wagner, J. Verheyden and J. G. Moffatt, *J. Org. Chem.*, 1974, 39, 24; (c) C. Augé, S. David and A. Veyrières, *Chem. Commun.*, 1976, 375; (d) H. Takaku, K. Kamaike and H. Tsuchiya, *J. Org. Chem.*, 1984, 49, 51; (e) A. Nashed and L. Anderson, *Tetrahedron Lett.*, 1976, 17, 3503; (f)
  M. A. Nashed and L. Anderson, *Carbohydr. Res.*, 1977, 56, 419; (g) M. A. Nashed, *Carbohydr. Res.*, 1978, 60, 200.
- 54 L. Yan and D. Kahne, Synlett, 1995, 523.
- 55 (a) T. Wada, T. Moriguchi and M. Sekine, J. Am. Chem. Soc., 1994, 116, 9901; (b) M. L. Wolfrom, P. J. Conigliaro and E. J. Soltes, J. Org. Chem., 1967, 32, 653.
- 56 (a) A. H. Fauq, A. P. Kozikowski, V. I. Ognyanov, R. A. Wilcox and S. R. Nahorski, *Chem. Commun.*, 1994, 1301; (b)
  P. B. Brady, E. M. Morris, O. S. Fenton and B. R. Sculimbrene, *Tetrahedron Lett.*, 2009, 50, 975; (c)
  C. Moreau, T. Kirchberger, J. M. Swarbrick, S. J. Bartlett, R. Fliegert, T. Yorgan, A. Bauche, A. Harneit, A. H. Guse and B. V. L. Potter, *J. Med. Chem.*, 2013, 56, 10079.