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A versatile catalyst system for enantioselective synthesis of 2-substituted 1,4-benzodioxanes†

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We report the synthesis of enantiomerically enriched 1,4-benzodioxanes containing alkyl, aryl, heteroaryl, and/or carbonyl substituents at the 2-position. The starting 1,4-benzodioxines were readily synthesized *via* ring closing metathesis using an efficient nitro-Grela catalyst at ppm levels. Excellent enantioselectivities of up to 99:1 er were obtained by using the versatile catalyst system [Ir(cod)Cl]₂/BIDIME-dimer in the asymmetric hydrogenation of 2-substituted 1,4-benzodioxines. Furthermore, DFT calculations reveal that the selectivity of the process is controlled by the protonation step; and coordinating groups on the substrate may alter the interaction with the catalyst, resulting in a change in the facial selectivity.

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Enantiomerically pure 1,4-benzodioxane derivatives are widely found in biologically active compounds.¹ They possess interesting biological activities such as α -adrenergic blocking,² antigastric,³ spasmolytic,⁴ antipsychotic,⁵ anxiolytic,⁶ and hepatoprotective properties⁷ (see Fig. 1 for selected examples). As

a result, their enantioselective syntheses have attracted great attention in the past decades.

Whereas early efforts to synthesize enantiomerically enriched 2,3-dihydro-1,4-benzodioxine containing compounds focused primarily on the use of chiral building blocks⁸ or enzymatic kinetic resolution of carboxylic acids and derivatives,⁹ recent investigations have increasingly focused on exploiting chiral catalysts. Highly enantioselective processes have been developed including Pd-catalyzed asymmetric intramolecular *O*-arylation coupling reactions¹⁰ and asymmetric intramolecular alkene aryloxyarylation reactions.¹¹ Furthermore, direct asymmetric hydrogenation of the 1,4-benzodioxine derivatives has been realized recently *via* Ir-catalyzed reduction of 2-aryl-1,4-benzodioxines¹² and Rh-catalyzed reduction of 2-carboxyl-1,4-benzodioxines,¹³ each of which has limited substrate scope. We aimed to discover a catalyst system that is effective and enantioselective for valuable alkyl and heteroaryl substituted products. Herein, we report a versatile [Ir(cod)Cl]₂/BIDIME-dimer catalyst system for enantioselective synthesis of a wide range of 2-substituted 1,4-benzodioxanes containing polar directing carbonyl groups as well as largely unfunctionalized groups including alkyls, aryls and heteroaryls. Furthermore, a mechanistic rationale for the reversal of facial selectivity between different substrates is provided based on DFT calculations.

We commenced our studies by preparing the requisite starting materials, 2-substituted 1,4-benzodioxines. We proposed to construct the key 1,4-benzodioxines fragment using a ring closing metathesis (RCM) strategy after alkylation of catechol derivatives (Scheme 1). RCM on substrates containing

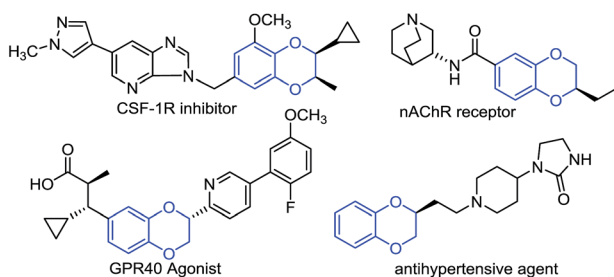


Fig. 1 Examples of chiral 1,4-benzodioxane containing molecules.

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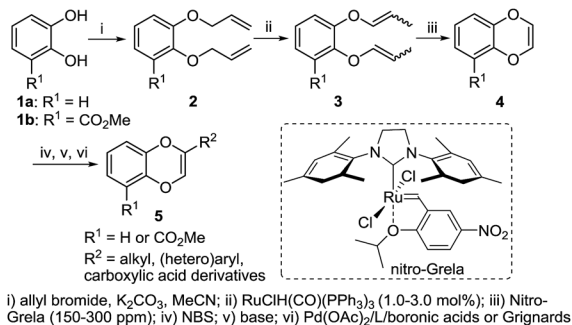
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vinyl ethers is known to be problematic.¹⁴ High catalyst loadings of Grubb's second-generation catalyst (5–8 mol%) were reported for the synthesis of 1,4-benzodioxines,¹⁵ which hindered the practicality of this otherwise powerful transformation. Furthermore, any coordinating group on the substrate would negatively impact the efficiency of the metathesis reaction. In searching for an effective catalyst, we hypothesized that Grela's nitro-aryl RCM catalysts would facilitate the initial metathesis event due to the labile and less Lewis basic nature of the isopropoxy donor. Gratifyingly, the bismesityl-substituted dichloro nitro-Grela catalyst is very reactive even for *ortho*-methyl ester substrate **3b**. As a result, compounds **4a** and **4b** were prepared using 150 or 300 ppm of the catalyst in >80% yields (Scheme 1).

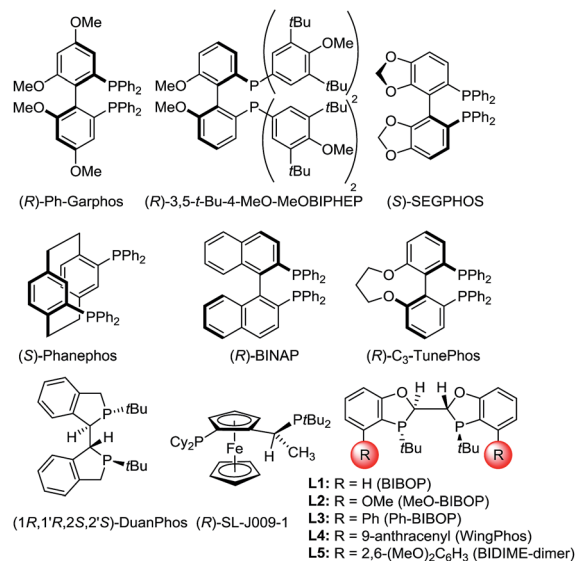
With a highly efficient synthesis of the 2-substituted 1,4-benzodioxines in hand, enantioselective reduction was tested. The initial focus was on the asymmetric hydrogenation of 2-cyclopropyl-substituted substrate **5a** (Table 1). Given that a general, highly enantioselective method for 2-alkyl-substituted 1,4-benzodioxanes has not been reported to date, we theorized that the small size of the cyclopropyl substituent would be a good test case for this purpose. More than 20 commercially available bidentate phosphine ligands were screened¹⁶ (representative examples are shown in Table 1, entries 1–8); however, all of the screened ligands provided moderate enantioselectivities (<90:10 er), and the reaction did not reach full conversion even with addition of acetic acid (AcOH), which was reported to increase the reactivity of the catalyst for iridium-catalyzed hydrogenation.¹⁷ We hypothesized that the chiral bis-dihydrobenzooxaphosphole (BIBOP) type ligands (**L1**–**L5**) could be used to advantage for this transformation, as they have the potential to provide a deep chiral pocket as well as steric tunability for enantioinduction.¹⁹ By increasing the size of the substituent on the BIBOP ligands from hydrogen to aryl, a general increase in enantioselectivity and conversion was observed (Table 1, entries 9–13). Notably, 9-anthracenyl-substituted ligand **L4** (WingPhos) provided 95:5 er; and 2,6-dimethoxyphenyl-substituted ligand **L5** (BIDIME-dimer) provided the highest enantioselectivity (97:3 er). Importantly, side products from cyclopropyl ring opening or ester hydrolysis were not observed under the reaction conditions and product **6a** was isolated in 98% yield using the BIDIME-dimer ligand.



Scheme 1 Synthesis of 1,4-benzodioxines.

Table 1 Ligand screening of Ir-catalyzed asymmetric hydrogenation of cyclopropyl-substituted 1,4-benzodioxine **5a**^a

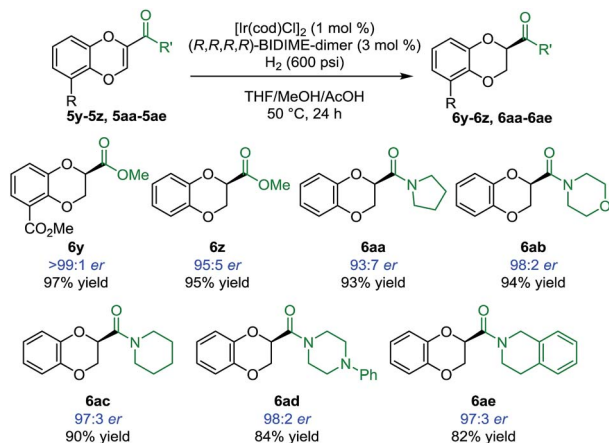
Entry	Ligand	Conv ^b (%)	er ^c
1	(<i>R</i>)-Ph-Garphos	79	76:24
2	(<i>R</i>)-3,5- <i>t</i> Bu-MeOBIPHEP	93	81:19
3	(<i>S</i>)-SEGPHOS	67	16:84
4	(<i>R</i>)-BINAP	92	81:19
5	(<i>R</i>)-C ₃ -TunePhos	83	77:23
6	(<i>S</i>)-Phanephos	91	88:12
7	(1 <i>R</i> ,1' <i>R</i> ,2 <i>S</i> ,2' <i>S</i>)-DuanPhos	63	41:59
8	(<i>R</i>)-Josiphos SL-J009-1	75	56:44
9	L1	68	47:53
10	L2	94	59:41
11	L3	>98	66:34
12	L4	>98	95:5
13	L5	>98 (98) ^d	97:3



^a Reaction conditions: **5a** (0.1 mmol), $[Ir(cod)Cl]_2$ (0.001 mmol), ligand (0.003 mmol), THF (0.2 mL), MeOH (0.2 mL), AcOH (4 mmol), H_2 (600 psi), 50 °C for 24 h. ^b Conversion was determined by HPLC. ^c Enantiomeric ratio was determined by chiral SFC. ^d Isolated yield.

Having identified BIDIME-dimer as the optimal ligand, other 2-alkyl-substituted 1,4-benzodioxines were examined in the Ir-catalyzed asymmetric hydrogenation reaction (Scheme 2). The cyclopropyl analogue without the methyl ester substituent at the 5-position was hydrogenated to afford **6b** with comparable enantioselectivity (94:6 er). Other cycloalkyl substituents ranging from 4- to 7-membered-rings also performed well in the reaction to give the corresponding 1,4-benzodioxanes (**6c**–**6e**) with very good enantioselectivities. Additionally, 2-aryl-substituted 1,4-benzodioxanes **6f** and **6g** were synthesized with excellent enantioselectivities of 98:2 er and >99:1 er, respectively. The synthesis of **6f** (98:2 er) is





Scheme 4 Scope of Ir-catalyzed asymmetric hydrogenation of 1,4-benzodioxines^a. ^a Reaction conditions: **5** (0.2 mol), $[Ir(cod)Cl]_2$ (0.002 mmol), (R,R,R,R) -BIDIME-dimer **L5** (0.006 mmol), THF (0.3 mL), MeOH (0.3 mL), AcOH (8 mmol), H_2 (600 psi), 50 °C for 24 h. Isolated yield. Enantiomeric ratio was determined by chiral SFC or HPLC.

Based on the previous studies of related Ir-catalyzed hydrogenations, reduction was expected to occur by sequential protonation-hydride transfer mechanism (Fig. 2).²²

In such a mechanism, protonation or hydride transfer can be enantiodetermining. Calculations of this system indicate that the energy of the separated ion pair is much higher than that of the protonation transition state (by ~ 25 kcal mol⁻¹) or the subsequent hydride transfer transition state. Since transfer of the iridium hydride to the opposite face *via* the separated ion pair is not favorable energetically, the calculations support the Ir catalyst establishing a facial preference during the protonation stage. To understand the origin of this facile preference and the enantioselectivity observed, we further analyzed the protonation step computationally. We found that protonation of both alkyl and carboxymethyl substituted benzodioxines proceeds as a Markovnikov process (C2 protonation on Fig. 2. See ESI† for details). We confirmed this regioselectivity for the

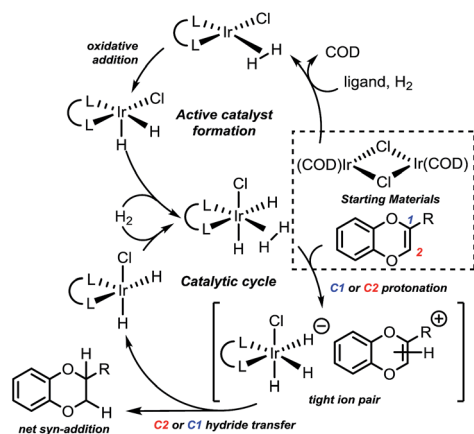


Fig. 2 Proposed hydrogenation mechanism. See ESI† for the computational analysis of the active catalyst formation. Computational study performed for R = Me and R = CO₂Me.

reaction on both faces of the benzodioxine. This feature of the mechanism allows us to use the same model for both electron-poor and electron-rich R substituents.

As the reaction takes place in a polar medium, ionization of or loss of dihydrogen from the Ir complexes is highly likely²³ (Fig. 3). For the non-coordinating substrates, we propose protonation *via* a ‘mode A’ mechanism, where substrate reacts with a neutral or cationic Ir complex. Both neutral and cationic mode A mechanisms provide very similar ligand–substrate interactions that determine the stereochemical outcome of the reaction, allowing us to consider only one of these mechanisms in our study.

For the coordinating substrate we envision an intramolecular process where the substrate coordinates the metal center prior to protonation (mode B). Such an arrangement will cause the subsequent reaction with dihydrogen to be more entropically favorable than in mode A. We propose that different arrangements of the mode A and mode B transition states alter the selectivity determining ligand–substrate interactions, thereby affecting the facial preference of the protonation. To test this hypothesis, the reactions of methyl and carboxymethyl substituted benzodioxines were analyzed. Computational analysis of the mode A-protonation of the methyl substrate indicated that reaction takes place preferentially on the face A of the starting material (see Fig. 4).

In contrast, the protonation of the methoxycarbonyl substituted benzodioxine *via* mode B favors proton delivery to face B, resulting in the opposite selectivity (Fig. 5).

The selectivity of the methyl substrate appears to be dictated by an unfavorable steric interaction between the substrate and the aryl portion of the ligand (Fig. 6, mode A). The computationally predicted selectivity is in good agreement with experimental results (Fig. 4, computed 80:20 *vs.* experimental 90:10). In contrast, control of the stereoselectivity with the methyl ester is provided by the stabilizing electronic interactions (Fig. 6, mode B). Computationally predicted values for enantiomeric ratio of this substrate are also in very good agreement with

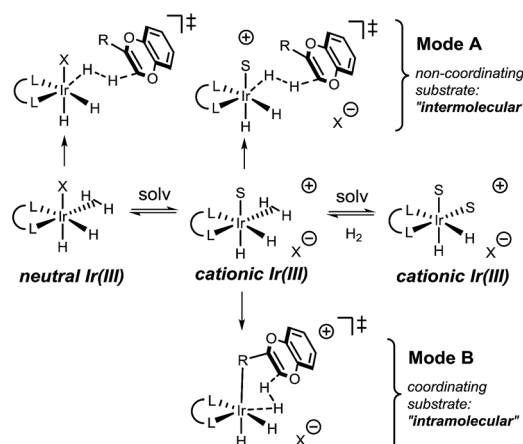


Fig. 3 Possible solvation equilibrium and proposed protonation transition states for coordinating and non-coordinating substrates.



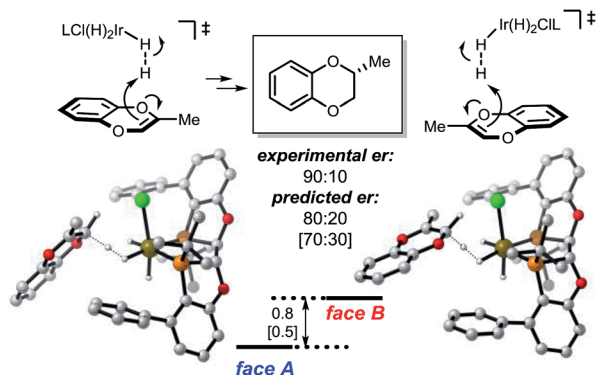


Fig. 4 Energetics of the mode A protonation of the Me-substituted benzodioxine. Selectivity-determining conformations are shown. Free energy and enthalpy (in brackets) gaps are calculated using PBE-D2/6-311+G(d,p), Ir:LANL2DZ(f), IEFPCM-MeOH//PBE/6-31G(d), Ir:LANL2DZ.²⁴ Values are in kcal mol⁻¹. Ligand L3 was used for the calculation to minimize conformational isomers; the experimental value was taken from 6h.

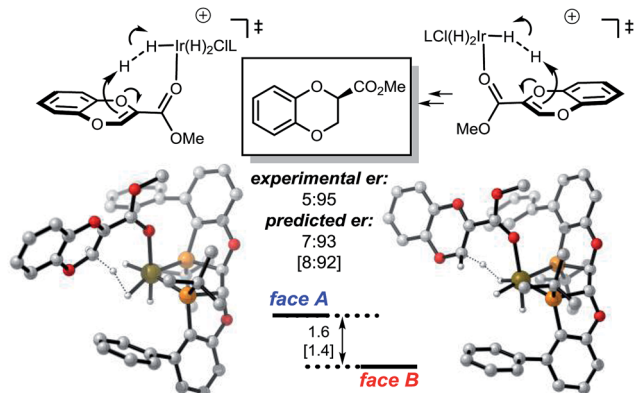


Fig. 5 Energetics of the mode B protonation of the CO₂Me-substituted benzodioxine. Selectivity-determining conformations are shown. Free energy and enthalpy (in brackets) gaps are calculated using PBE-D2/6-311+G(d,p), Ir:LANL2DZ(f), IEFPCM-MeOH//PBE/6-31G(d), Ir:LANL2DZ. Values are in kcal mol⁻¹. Ligand L3 was used for the calculation to minimize conformational isomers; the experimental value was taken from 6z.

experimentally obtained result (Fig. 5, computed 7:93 vs. experimental 5:95).

The selectivity basis for the mode B approach is outlined in greater detail in Fig. 7. From the transition state geometry, the structural motif contributing to stabilization is a combination of slipped overlapping π - π and $\text{sp}^2\text{CH}-\pi$ interactions,²⁵ which are reported to be relatively strong (~ 2.5 kcal mol⁻¹).²⁶ In our case, imperfect positioning of the CH unit relative to the aromatic system (3.13 Å vs. typical 2.73 Å and 136° vs. 148° for typical interactions)²⁷ results in an overall weaker energetic effect (1.5 kcal mol⁻¹ in this case). This interaction is absent in the transition state arising from the other facial approach (face A in Fig. 5). Even though this electronic interaction is not optimal, the resultant stabilization is sufficient to drive the differentiation of these diastereomeric transition states leading to high selectivity.

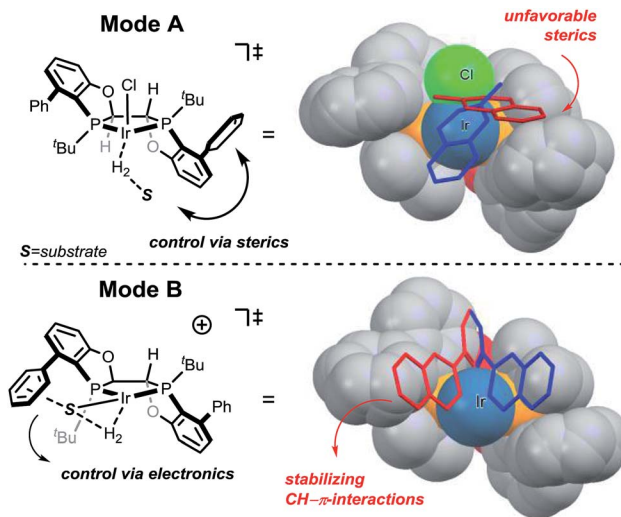


Fig. 6 Stereochemical models for mode A and mode B protonations. Substrate approaches for face A and face B attacks are shown in blue and red respectively.

In summary, we report the discovery of a highly enantioselective (>99:1 er) and versatile IrCl/BIDIME-dimer catalyst system for asymmetric hydrogenation reactions to prepare a wide range of chiral 1,4-benzodioxanes including those with alkyl, aryl, heteroaryl, and carbonyl groups at the 2-position. The route commences with a key RCM step to construct the ring of the requisite 1,4-benzodioxines with ppm level catalyst loadings using the nitro-Grela catalyst. Overall, this route represents a general and atom-economical catalytic approach to construct synthetically challenging chiral benzodioxanes. Computational studies suggest that enantioselectivity of the hydrogenation process is controlled by an outer sphere protonation step. Depending on the nature of the substrate, however, coordination to the metal center can intervene creating an inner sphere approach in the corresponding transition states. This specific coordination is a powerful way of altering the spatial orientation of the substrate during the protonation, providing the means to affect the absolute and relative stereochemical outcome of the reaction.

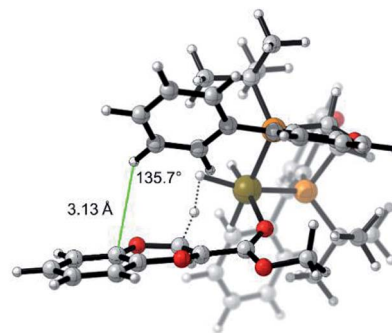


Fig. 7 Model of lowest energy mode B protonation for the 2-methoxycarbonyl-1,4-benzodioxine showing electronic stabilizing interactions (π - π and $\text{sp}^2\text{CH}-\pi$ interactions).²⁵



Conflicts of interest

The authors declare no conflict of interest.

Acknowledgements

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