Chiral phosphoric acid-catalyzed enantioselective construction of structurally diverse benzothiazolopyrimidines†

Lucie Jarrige, a Danijel Glavač, b Guillaume Levitre, a Pascal Retailleau, a Guillaume Bernardat, c Luc Neuville d and Géraldine Masson e *

A highly efficient catalytic enantioselective [4+2] cycloaddition was developed between 2-benzothiazolimines and enecarbamates. A wide range of benzothiazolopyrimidines bearing three contiguous stereogenic centers was obtained in high to excellent yields and with excellent diastereo- and enantioselectivities (d.r. > 98:2 and up to >99% ee). Furthermore, this chiral phosphoric acid-catalyzed strategy was scalable and enabled access to a new class of optically pure Lewis base isothiourea derivatives.

Introduction

Isothioureas constitute an important class of organocatalysts due to their highly nucleophilic nature.1 Recent efforts have been directed to the development of chiral derivatives capable of promoting enantioselective transformations, which are difficult to control using traditional nucleophilic catalysts. Among the several promising examples, benzothiazolopyrimidine, better known as homobenzotetramisole (HBTM), has assumed a leading position as a chiral catalyst. Chiral HBTMs have been successfully used in a number of asymmetric transformations including the kinetic resolution of secondary alcohols and carboxylic acids, desymmetrization, acyl transfer, annulation processes, and lactonisation, among others. Nevertheless, despite the interest in optically active HBTMs, there have been only limited reports on their enantioselective synthesis. In 2008, Birman et al. disclosed the first asymmetric synthesis of HBTMs following a seven-step sequence involving a key SNAr reaction between 2-chlorobenzothiazole and amino-alcohols prepared from the chiral pool of amino acids (Scheme 1a). Chiral amino-alcohols, produced by catalytic enantioselective procedures, were later engaged in similar routes as reported by the same group as well as by Smith et al. and Okamoto et al. (Scheme 1a). More recently, Enders et al. established an efficient asymmetric synthesis of precursors of HBTMs by the N-heterocyclic carbene-catalyzed Mannich/lactamization domino reaction of 2-benzothiazolimines and α-chloroaldehydes (Scheme 1b). Despite these examples, no catalytic asymmetric approach has been reported so far to directly construct these heterocycles. An attractive direct access relying on theaza-Diels–Alder approach has been reported by Mellor et al. for the synthesis of homobenzotetramisole in an achiral stoichiometric context. The requirement of a full equivalent of the acid to perform the reaction is self-evident due to the presence of the basic isothiourea that might capture the acid during the formation of HBTMs. Turning this process into a catalytic enantioselective approach would be highly

Scheme 1  Previous studies and our strategy.
appealing, not only to access new potent chiral Lewis bases but also for the generation of novel biologically active molecules.13

In the past few years, chiral phosphoric acids have been widely recognized as powerful catalysts for the synthesis of various heterocycles containing nitrogen atoms.14 In this context, we have recently documented phosphoric acid-catalyzed enantioselective aza-Diels–Alder reactions14,15 of various azadienes with enecarbamates as dienophiles.16 Based on these studies, we envisioned that this Brønsted acid 1 may be capable of promoting the [4+2] cycloaddition between 2-benzothiazolimines 2 and enecarbamates 3 (Scheme 1c) allowing the development of an enantioselective synthesis of benzothiazolopyrimidines 4.

Because of the formation of the basic isothiourea during the process, catalyst poisoning might lead to difficulties in developing a catalytic version. Nevertheless, we reasoned that tuning the catalyst or using additives to induce a hydrogen shuttle could help in overcoming the problem.17

Results and discussion

With this proposal in mind, we examined the aza-Diels–Alder reaction of 2-benzothiazolimine18 2a with (E)-N-benzyl propenylcarbamate 3a in the presence of chiral phosphoric acids 1. Gratifyingly, in dichloromethane at room temperature, the reaction promoted by 1a afforded benzothiazolopyrimidine 4a as a single diastereomer, with an enantiomeric excess of 95% albeit with a low yield (Table 1, entry 1). Obviously, while yields remained low, we were delighted to observe that no full catalyst inhibition was occurring. Encouraged by this result, we screened various reaction conditions to search for the optimal reaction conditions. Performing the reaction at 50 °C in DCE slightly improved the kinetics (16 h instead of 24 h) even if overall yields were not superior and significant erosion of enantioselectivity was observed.19 Then, a series of phosphoric acids were screened in DCE at 50 °C (entries 2–4), and we found that the catalyst structure had a marked impact on the reaction efficiency. For instance, the use of chiral phosphoric acid 1c having two bulky 3,3′-(2,4,6-trisopropylphenyl) groups improved the yield, presumably by facilitating the catalyst turnover. A further survey of reaction conditions revealed that the optimal conditions consisted in performing the reaction with 3 equivalents of 3a in the presence of 10 mol% of 1c, delivering 4a in 90% yield with almost perfect diastereo- and enantioselectivity (entry 5).

With optimized reaction conditions in hand, the scope of this chiral phosphoric acid-catalyzed enantioselective aza-Diels–Alder reaction was investigated and the results are summarized in Scheme 2. First, 2-benzothiazolimines bearing various substituents on the aromatic rings were evaluated. Pleasingly, the reaction was insensitive to the electronic nature of the substituents on the benzothiazole rings. For instance, the benzothiazole bearing either electron-donating or electron-withdrawing groups, such as methyl, halogen and trifluoromethyl groups, provided the corresponding benzothiazolopyrimidines 4b–f in high yields and excellent enantioselectivities (up to 99% ee). An unsubstituted benzothiazole ring proceeded uneventfully to give cycloadduct 4g with almost perfect enantioselectivity (up to 99% ee). It is worth noting that the reactions performed on a 1 mmol scale displayed the same results as the ones on a 0.10 mmol scale, further indicating the practicality of this protocol. On the other hand, the electronic properties of aromatic rings that came from aldehydes had no significant impact on the enantiomeric excess, though the reaction with the electron-rich imine gave a slightly lower yield (4h vs. 4a). When the reaction was conducted with imine, which had bromine at the meta-position, the desired cycloadduct 4k was obtained in 80% yield and 98% ee. Notably, 4k can be enriched to >99% ee by single recrystallization, which is very interesting in view of their use as organocatalysts. Moreover, a relative and absolute configuration was unambiguously determined by single crystal X-ray analysis (see the ESIF). Finally, the reaction with imine 2l derived from 2-thiophenecarboxaldehyde proceeded smoothly to give the cycloadduct 4l in almost quantitative yields with 98% ee. This novel catalytic enantioselective cycloaddition was also applied to representative enecarbamates (Scheme 2). Notably, a variety of protecting groups bonded to the nitrogen atom of 3 (including Boc, Alloc and 4-(trifluoromethyl)benzylcarbamate) were well tolerated and the corresponding benzothiazolopyrimidines (4m–q) were, in all cases, isolated in excellent yields and excellent stereoselectivities. It is noteworthy that the catalyst loading was successfully reduced to 5.0 mol% (4q, entry 10). A comparable enantioselectivity (97%) was still achieved with a slightly lower chemical yield (70%).

In addition, β-substituted enecarbamates bearing different alkyl groups reacted efficiently to deliver expected trisubstituted-cycloadducts 4r–u in good to excellent yields with high selectivities (94 to 98% ee). The reaction with 3c or 3d

Table 1 Optimization of the enantioselective [4+2] cycloadditiona

<table>
<thead>
<tr>
<th>Entry</th>
<th>Ar</th>
<th>T (°C)</th>
<th>Solvent</th>
<th>Yield (%)</th>
<th>ee (%)</th>
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</thead>
<tbody>
<tr>
<td>1</td>
<td>Ph (1a)</td>
<td>RT</td>
<td>DCM</td>
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<td>95</td>
</tr>
<tr>
<td>2</td>
<td>1a</td>
<td>50</td>
<td>DCE</td>
<td>41</td>
<td>80</td>
</tr>
<tr>
<td>2</td>
<td>9-Phenanyl (1b)</td>
<td>50</td>
<td>DCE</td>
<td>29</td>
<td>85</td>
</tr>
<tr>
<td>4</td>
<td>2,4,6-TRIP (1c)</td>
<td>50</td>
<td>DCE</td>
<td>62</td>
<td>98</td>
</tr>
<tr>
<td>5</td>
<td>1c</td>
<td>50</td>
<td>DCE</td>
<td>90 (68)</td>
<td>99</td>
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a Reaction conditions: 2a (0.10 mmol), 3a (0.20 mmol) and 1 (0.01 mmol) in 1.0 mL of solvent for 16 h. Yields refer to the chromatographically pure diastereomer 4a determined to be higher than 98 : 2 by 1H NMR. ee values were determined by HPLC with a chiral stationary phase. For 24 h. With 3 equiv. of 3a. With 5 mol% of 1c.
containing a cyclopropyl or silyl ether group formed products 4s or 4t with a slight decrease in the yield without affecting asymmetric induction.

The stereoselective issue of the reaction (([EF]-enecarbamates to the cis product) raises questions about the mechanism. In order to get insight, we computationally evaluated the free energy of both diastereoisomers of the product 4g (Fig. 1A). The trans form was found favoured by more than 5.2 kcal mol$^{-1}$ over its cis sibling (see the ESI†). This result together with the fact that the cis isomer is actually observed suggests that the cycloaddition occurs under kinetic control. A stepwise process was established by performing a control experiment in the presence of 3 equivalents of ethanol, which evidenced the formation of an N-Cbz-iminium intermediate (5, Fig. 1B).

Based on this, as well as our previous work and other published reports, a possible catalytic mechanism was proposed in Scheme 3. We hypothesized that 1 would form a hydrogen network with the nitrogen Lewis basic sites of 2-benzothiazolimine 2 and the acidic site of enecarbamate 3, leading to complex 6. Mediated by the phosphoric acid double activation, subsequent bond formation would deliver complex 7. The aromatic and R residue would be positioned in a trans situation as found in the product, but the pseudo equatorial arrangement would create substantial strain. Maintained in the chiral cavity by the double hydrogen anchors, rotation of the backbone would furnish a sterically less-congested conformer 8.

Finally the intramolecular nucleophilic addition of benzothiazole to iminium will give the observed cis–trans substituted

**Scheme 2** Scope of the enantioselective synthesis of benzothiazolo-pyrimidines 4. General reaction conditions: 2 (0.1 mmol), 3 (0.3 mmol), 1c (10 mol%), 1,2-DCE (1 mL, 0.1 M), 50 °C, and 24 h. Yields refer to chromatographically purified compounds. The d.r. values were determined by $^1$H NMR analysis to be higher than 98:2 in all cases. The ee values were determined by HPLC on a chiral stationary phase: see the ESI† for details. $^a$ From 1.0 mmol of 2g. $^b$ After recrystallization. $^c$ From 0.3 mmol of 2b. $^d$ With 5 mol% of 1c.

**Scheme 3** Plausible mechanism and transition states.
benzothiazolopyrimidine 4. Heating, rapid formation of hydrogen-bonded complex 6 or use of excess 3 may all facilitate the decomposition of 9 and ensure the catalyst turnover.

As a further illustration of the utility of this novel catalytic cycloaddition, we developed conditions to convert the carbamate group into other functional groups. The N-Boc group of 4q was removed by mild acidic hydrolysis to produce free alcohol 10 (a key intermediate to install additional molecular diversity) with no erosion of enantiopurity (Scheme 4). The absolute configuration was confirmed without ambiguity by X-ray diffraction analysis (see the ESI†).

**Conclusions**

We have developed an efficient enantioselective [4+2] cycloaddition organocatalyzed by chiral phosphoric-acid for the synthesis of benzothiazolopyrimidines bearing three contiguous stereogenic centers, which were unreported before. Involving 2-benzothiazolamines and enecarbamates as substrates, this scalable strategy furnished a wide range of compounds with very high yields and enantioselectivity in a diastereospecific manner. The obtained cycloadducts were successfully converted into corresponding free alcohol heterocycles without any loss of enantioselectivity. Further investigations are in progress concerning the catalytic activity of these new compounds as chiral nucleophilic organocatalysts.

**Conflicts of interest**

There are no conflicts to declare.

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**Notes and references**


19 Addition of AOCb to improve catalyst turnover (see ref. 17) leads to substantial degradation.
