PERSPECTIVE
Yang Yang and Ji-Woong Lee
Toward ideal carbon dioxide functionalization
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This Perspective recapitulates recent developments of carbon dioxide utilization in carbon–carbon bond formation reactions, with an intention of paving a way toward sustainable CO₂-functionalization and its tangible applications in synthetic chemistry. CO₂ functionalization reactions possess intrinsic drawbacks: the high kinetic inertness and thermodynamic stability of CO₂. Numerous procedures for CO₂ utilization depend on energy-intensive processes (i.e. high pressure and/or temperature), often solely relying on reactive substrates, hampering its general applications. Recent efforts thus have been dedicated to catalytic CO₂-utilization under ambient reaction conditions, however, it is still limited to a few activation modes and the use of reactive substrates. Herein, ideal CO₂-functionalization with particular emphasis on sustainability will be discussed based on the following sub-categories; (1) metal-catalyzed 'reductive' carboxylation reaction of halides, olefins and allyl alcohols, (2) photochemical CO₂-utilization, (3) redox-neutral CO₂-functionalization, and (4) enantioselective catalysis incorporating CO₂ to form C–CO₂ bonds (excluding strain mediated reactions with epoxide- and aziridine-based substrates). Recent progress in these fields will be discussed with the proposed reaction mechanisms and selected examples, highlighting redox-neutral, umpolung, and asymmetric carboxylation to postulate ideal CO₂ functionalization reactions to be developed in the near future.

1. Introduction

Carbon is an essential element for all living organisms, and is present in carbohydrates, amino acids, proteins, and lipids. These biomolecules are synthesized with specific selectivities controlled by the natural molecular foundry – enzymes – to sustain forms of life. The sustainability of bio- and chemical networks in living organisms is powered by the seemingly unlimited solar energy. Owing to the evolution of cyanobacteria and their photosynthesis, our planet became a unique biosphere where water was split into oxygen and hydrogen, while consuming (or fixating) CO₂ to generate reduced organic matter.

Photosynthesis and CO₂ fixation operate under ambient conditions; artificial photosynthesis is yet to be realized, and can ensure sustainable growth of the human civilization. The challenge lies in overcoming the thermodynamic stability and kinetic inertness of CO₂, which possesses the highest oxidation state of carbon. Therefore, it is inevitable to employ reducing
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reagents (reactive metals, H₂, electricity, and highly reducing chemicals) to overcome the intrinsic reaction barrier of CO₂ activation, particularly to enable the reactions to be operative under mild reaction conditions.

Recently, the global society has raised concerns related to excessive energy consumption and uncontrollable anthropogenic CO₂ emission. Although CO₂ functionalization can provide ideal solutions, chemical reactions with CO₂ currently suffer from low efficiency, making it impossible to mitigate the overwhelmingly large quantity of accumulated CO₂ in the atmosphere at low concentrations. Yet, chemical recycling of carbon dioxide has been recognized as a promising supplement to the natural carbon cycle, while producing value-added fine chemicals. In this context, CO₂ can serve as an inexpensive and non-toxic renewable C₁-building block. For example, light hydrocarbons and C₁- or C₂-units (i.e. carbon monoxide, formic acid, formaldehyde, methanol, and oxalic acid) are accessible from CO₂, mostly catalyzed by heterogeneous materials (semiconductors, zeolites, COFs, MOFs, and g-C₃N₄ (ref. 12)). On the other hand, homogeneous catalysis has shown remarkable potential in C–C bond formation reactions, via formal insertion of CO₂ at C–H bonds. The utility of carboxylic acids and their derivatives is certainly applicable with broad interest in organic synthesis and pharmaceutical chemistry.

As categorized in Table 1, catalytic CO₂-functionalization reactions have been reviewed, particularly transition-metal catalyzed C–C bond formation reactions, carboxylation reactions catalyzed by palladium, silver, copper or copper–NHC (N-heterocyclic carbene) complexes, and nickel/iron catalysts, asymmetric CO₂-functionalization reactions and photocatalytic CO₂-functionalization. Other types of reactions are addressed in this review. It is worth noting that although CO₂ reduction is a very attractive strategy, there are too many difficulties and challenges, such as high energy consumption, the need for a reducing agent and the use of transition metal catalysts.

Table 1: A summary of recent reviews cited regarding CO₂-utilization related subjects

<table>
<thead>
<tr>
<th>Year (ref.)</th>
<th>Title</th>
<th>Keywords</th>
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<tbody>
<tr>
<td>2014 (ref. 3a)</td>
<td>Catalysis for the valorization of exhaust carbon: from CO₂ to chemicals, materials, and fuels.</td>
<td>CO₂ emission and utilization</td>
</tr>
<tr>
<td>2014 (ref. 3b)</td>
<td>Technological use of CO₂</td>
<td>CO₂ capture</td>
</tr>
<tr>
<td>2001 (ref. 3c)</td>
<td>Porous inorganic membranes for CO₂ capture: present and prospects</td>
<td>CO₂ emission and utilization</td>
</tr>
<tr>
<td>2007 (ref. 4)</td>
<td>Catalysis research of relevance to carbon management: progress, challenges, and opportunities</td>
<td>CO₂ conversion</td>
</tr>
<tr>
<td>2018 (ref. 7)</td>
<td>Transformation of carbon dioxide</td>
<td>Catalysis, carbon life cycle assessment</td>
</tr>
<tr>
<td>2018 (ref. 8a)</td>
<td>Sustainable conversion of carbon dioxide: an integrated review of catalysis and life cycle assessment</td>
<td>Photocatalytic CO₂ reduction</td>
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<tr>
<td>2013 (ref. 8b)</td>
<td>Cocatalysts in semiconductor-based photocatalytic CO₂ reduction: achievements, challenges, and opportunities</td>
<td></td>
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<tr>
<td>2014 (ref. 8c)</td>
<td>Photocatalytic conversion of CO₂ into renewable hydrocarbon fuels: state-of-the-art accomplishment, challenges, and prospects</td>
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<tr>
<td>2017 (ref. 11a)</td>
<td>The chemistry of metal–organic frameworks for CO₂ capture, regeneration and conversion</td>
<td>MOFs in CO₂</td>
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<tr>
<td>2017 (ref. 11b)</td>
<td>Metal organic framework based catalysts for CO₂ conversion</td>
<td>g-C₃N₄ in CO₂ utilization</td>
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<td>2015 (ref. 12a)</td>
<td>A review on g-C₃N₄ for photocatalytic water splitting and CO₂ reduction</td>
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<td>2018 (ref. 15a)</td>
<td>Transition metal-catalyzed carboxylation reactions with carbon dioxide</td>
<td>Metal-catalyzed carboxylation</td>
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<td>2016 (ref. 15b)</td>
<td>Metal-catalyzed carboxylation of organic (pseudo)halides with CO₂</td>
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<td>2018 (ref. 15c)</td>
<td>Transition metal-catalyzed carboxylation of unsaturated substrates with CO₂</td>
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<td>2016 (ref. 15d)</td>
<td>Recent advances in palladium-catalyzed carboxylation with CO₂</td>
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<td>2016 (ref. 17)</td>
<td>Silver-catalyzed carboxylation</td>
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<td>2016 (ref. 18)</td>
<td>Copper-catalyzed carboxylation reactions using carbon dioxide</td>
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<td>2013 (ref. 19)</td>
<td>N-heterocyclic carbene (NHC)–copper-catalysed transformations of carbon dioxide</td>
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<td>2016 (ref. 20)</td>
<td>Ni- and Fe-catalyzed carboxylation of unsaturated hydrocarbons with CO₂</td>
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<td>2015 (ref. 23a)</td>
<td>Recent advances in the catalytic preparation of cyclic organic carbonates</td>
<td>Cyclic organic carbonates</td>
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<tr>
<td>2018 (ref. 23b)</td>
<td>Catalytic strategies for the cycloaddition of pure, diluted, and waste CO₂ to epoxides under ambient conditions</td>
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<tr>
<td>2015 (ref. 23c)</td>
<td>Synthesis of cyclic carbonates from epoxides and carbon dioxide by using organocatalysts</td>
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<tr>
<td>2018 (ref. 24a)</td>
<td>Catalytic reductive N-alkylations using CO₂ and carboxylic acid derivatives: recent progress and developments</td>
<td>Catalytic alkylation</td>
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<td>2017 (ref. 24b)</td>
<td>Utilization of CO₂ as a C₁ building block for catalytic methylation reactions</td>
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<td>2017 (ref. 21a)</td>
<td>Enantioselective incorporation of CO₂: status and potential</td>
<td>Asymmetric functionalization</td>
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<td>2016 (ref. 21b)</td>
<td>CO₂-mediated formation of chiral fine chemicals</td>
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<tr>
<td>2018 (ref. 22a)</td>
<td>Photoredox catalysis as a strategy for CO₂ incorporation: direct access to carboxylic acids from a renewable feedstock</td>
<td>Photocatalytic carboxylation using CO₂</td>
</tr>
<tr>
<td>2017 (ref. 22b)</td>
<td>Photocatalytic carboxylation of activated C(sp³)–H bonds with CO₂</td>
<td>Formic acid and methanol derivatives</td>
</tr>
<tr>
<td>2017 (ref. 83a)</td>
<td>Reversible hydrogenation of carbon dioxide to formic acid and methanol: Lewis acid enhancement of base metal catalysts</td>
<td></td>
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<tr>
<td>2015 (ref. 83b)</td>
<td>CO₂ hydrogenation to formate and methanol as an alternative to photo- and electrochemical CO₂ reduction</td>
<td></td>
</tr>
<tr>
<td>2014 (ref. 83c)</td>
<td>Recycling of carbon dioxide to methanol and derived products-closing the loop</td>
<td></td>
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</table>
also tabulated to guide the readers for further reading in specific topics of interest. For example, carbonate formation reactions with epoxides and ring-strain mediated reactions, catalytic alkylation with CO$_2$, etc., will not be discussed in this Perspective.

The purpose of this Perspective is the following: providing a general concept of catalytic CO$_2$-functionalization by exemplifying recent progress (up to 2018). Section 2 will discuss transition-metal catalysis with a hint of sustainability. Sections 3 and 4 will explore recently reported photochemical redox catalysis by utilizing synthetic dyes with the aid of pre-established transition metal catalysis, and single-electron reduction of CO$_2$ via a redox-neutral mechanism. The future perspective on ideal CO$_2$-functionalization will also be discussed in the context of umpolung carboxylation, redox-neutral photochemistry and asymmetric CO$_2$-activation to reduce the prevailing energy input or highly reactive species. This discussion will lead to an alternative platform for sustainable CO$_2$ recycling, to mimic the natural carbon cycle by utilizing the combined knowledge in organic, inorganic, photo- and materials chemistry, and enzymatic engineering for improved carbon fixation as well.

2. Metal-catalyzed reductive carboxylation with halides, olefins and allyl alcohols

The catalytic application of transition metals for carboxylation with CO$_2$ was triggered by the seminal work by Nobile (Scheme 1, eqn (1)) and Osakada (Scheme 1, eqn (2)), where stoichiometric Ph-Ni(L)-Br (L = 2,2'-bipyridine (bpy))

![Scheme 1](image1.png)

**Scheme 1** Stoichiometric CO$_2$-functionalization using Ni(0).

![Scheme 2](image2.png)

**Scheme 2** Ni-catalyzed reductive CO$_2$-functionalization reactions.
participated in CO$_2$ insertion at the Ph-Ni bond, affording benzoic acid as the final product.

The Martin group employed a Pd(II)–Pd(0) cycle in the catalytic carboxylation reaction of aryl bromides using ZnEt$_2$ as a terminal reducing reagent.$^{27}$ This methodology was further expanded to abundant Ni(II) catalysis by the Tsuji group,$^{28}$ realizing carboxylation of aryl chloride with Mn powder as a reducing reagent. New reductive carboxylation reactions were developed later by the Martin group with a broad range of substrate scope, including organic halides,$^{29}$ sulfonylates,$^{30}$ amines,$^{31}$ esters,$^{32}$ allyl acetates,$^{33}$ allyl alcohols$^{34}$ (Scheme 2a), allyl amine salts$^{35}$ (Scheme 2b), and unsaturated hydrocarbons (Scheme 2c and d).$^{36}$ The facile insertion of CO$_2$ into R-Ni was tested with olefin substrates, enabling olefin activation without an apparent hydride donor (Scheme 2). These protocols provided a broad substrate scope and high functional group tolerance. However, it is necessary to use (over)stoichiometric amounts of reducing reagents (i.e. Mn, Zn, ZnR$_2$, and etc.) to complete the catalytic cycle.

In 2017, a breakthrough CO$_2$-functionalization was reported by the Martin group proposing a ‘chain-walking’ mechanism with catalytic Ni–H species (Scheme 3).$^{37}$ Although the β-hydride elimination is undesired in transition metal-catalyzed coupling reactions,$^{38}$ in the proposed reaction mechanism, a chain-walking process was key to generate thermodynamically more stable species, thus contributing to the high regio- and chemoselectivity of the targeted insertion reactions.$^{39}$ For carboxylation reactions with CO$_2$, the Martin group showed temperature-controlled site-selectivity affording linear and branched carboxylated products (l : b ratios). The authors suggested a Curtin–Hammett scenario, where the reaction proceeded through common intermediates or transition states under fast equilibrium (Scheme 3a). More strikingly, the chain-walking mechanism was translated to a useful method starting from a mixture of alkyl bromides – expanding the utility of the protocol significantly. Regardless of regioisomers, linear alkanes were smoothly converted to carboxylated products under a bromination/carboxylation reaction sequence (1 atm of CO$_2$). The iterative reversible β-hydride elimination/insertion reactions occurred, converging regioisomers of alkyl bromides into a single carboxylated product (Scheme 3b).

The proposed chain-walking process with high site-selectivity represents a significant potential toward fatty-acid syntheses from bulk petroleum raw materials. In this context,
the same group extended the methodology with olefin substrates, enabling carboxylation reactions in the presence of water as a proton source.\textsuperscript{1,4b} In the case of alkenes, water served as a way to access metal-hydride species,\textsuperscript{38} namely Ni-H species, which in turn can participate in the above-mentioned chain-walking mechanism. Indeed, a linear carboxylic acid was the main product with high selectivity (b : l = 1 : 99) even from an unre- fined mixture of olefin isomers (Scheme 4a). As for alkynes, however, only a branched carboxylation product was obtained (Scheme 4b). The authors proposed that the Ni-L\textsubscript{2} complex favored the formation of a thermodynamically more stable \(\alpha,\beta\)-unsaturated nickelalactone (Ni-1) with internal alkynes in a CO\(_2\) environment. Therefore, a branched carboxylic acid was obtained with high selectivity (b : l = 99 : 1) after reduction with H\(_2\) and Pd/C. The ‘uni-directional’ chain-walking mechanism highlights the potential application of this process in producing added value chemicals from CO\(_2\) and crude industrial feedstock.

It is noteworthy that the variation of the ligand is critical in Ni-catalyzed reactions. The substituent adjacent to the nitrogen atoms in bidentate ligands (L\textsubscript{1} and L\textsubscript{2}), such as bipyridine and phenanthroline, differentiates the site-selectivity of the carboxylation reaction. High site-selectivity is a pre-requisite for many organic transformations, for example in allylic substitution reactions. Catalytic metal–ligand complexes govern chemo-, regio- and even enantioselectivity.\textsuperscript{39} Allyl alcohol is a substrate class with high accessibility yet low chemical utility for allylation reactions due to the apparently low leaving group ability of the hydroxide. It has been proved that in \textit{in situ} activation of allylic alcohol with ‘activating reagents’ can mediate various types of transformation,\textsuperscript{40} shortening the synthetic steps avoiding the preparation of activated substrates\textsuperscript{41} (like amines,\textsuperscript{41a} ammonium salts,\textsuperscript{41b} carbamates,\textsuperscript{41c} carbonates,\textsuperscript{41d} esters,\textsuperscript{41e} ethers,\textsuperscript{41f} nitro compounds,\textsuperscript{41g} phosphates,\textsuperscript{41h} and sulfones\textsuperscript{41i})). For example, CO\(_2\) was involved in the asymmetric Pd-catalyzed direct \(\alpha\)-allylation of ketones.\textsuperscript{42} The use of CO\(_2\) as a catalyst is noticeable although only a ‘catalytic’-amount of it would be necessary for the process.

The Martin group employed CO\(_2\) as an activating reagent as well as a C1 source for the carboxylation of allylic alcohols to afford \(\beta,\gamma\)-unsaturated carboxylic acids (Scheme 5a).\textsuperscript{43} Once again, ligand-controlled selectivity was observed starting from linear or branched allylic alcohols affording high yields of linear and branched carboxylation products (Scheme 5b). The former resulted from CO\(_2\) insertion between the \(\alpha\)-carbon and Ni(0) center. Alternatively, \(\alpha\)-branched acids were obtained when the tridentate ligand L\textsubscript{3} was employed. The critical role of the ligands was rationalized by stoichiometric studies of active Ni-L\textsubscript{2} or Ni-L\textsubscript{3} species in the absence of Zn metal (yields: linear, 0%; branched 73%). The transition-state, Ni-2, was proposed for the nucleophilic attack from the \(\gamma\)-carbon of an \(\eta^1\)-allyl Ni(0) intermediate to CO\(_2\). Also, a six-membered cyclic conformation can be suggested, similar to the reported nucleophilic addition of Pd-(\(\pi\)) allyl intermediates to CO\(_2\) or carbonyl substrates.\textsuperscript{44,45} The utility of the reaction was further verified by producing useful intermediates for the synthesis of \(\gamma\)-lactone-based bioactive compounds.\textsuperscript{44}

Dienes, abundant and accessible chemical feedstocks, have the same oxidation states as allylic alcohols. However, activation of dienes and conjugated olefins poses a great challenge. Recently, Ni-based catalysts were evaluated for a catalytic carboxylation reaction of dienes toward carboxylated or dicarboxylated products in stoichiometric amounts of a Ni(0) complex.\textsuperscript{45} Although limited only to activated substrates, alkynes\textsuperscript{46} and silylallenes\textsuperscript{47} were transformed to the desired dicarboxylated products. The Martin group successfully implemented a catalytic dicarboxylation reaction for 1,3-dienes with high site-selectivity (up to 90%), to furnish diesters (Scheme 6a).\textsuperscript{34} Various functional groups were tolerated including heterocycles, organotin, nitrile, and esters. Structurally simple dienes such as butadiene, isoprene, and piperylene – major byproducts of steam cracking in ethylene production plants – were converted to the corresponding terminal diacids with excellent site-selectivity in moderate yields (up to 65% yield, 99 : 1 selectivity, Scheme 6b). Single crystal structure analysis determined the formation of monocarboxylated \(\eta^1\)-Ni nickelalactone (Ni-3). The corresponding dicarboxylation product could be obtained when Ni-3 was treated with CO\(_2\) under optimized reaction conditions (also see Mori group’s work\textsuperscript{42d}), shedding some light on the reaction mechanism (Scheme 6c).

The transition metal-catalyzed carboxylation reactions of the above-mentioned recent examples showed unprecedented catalytic performances with a variety of substrates, yet they require stoichiometric reducing reagents to sustain the catalytic cycle.

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\textbf{Scheme 4} Site-selective carboxylation dictated by the degree of unsaturation.
Certain improvements have been attempted by utilizing insoluble reducing reagents (Mn, and Zn powder) replacing highly reactive RMgX, Et₂Zn, or AlMe₃. A na ni dea l CO₂ functionalization process, a redox-neutral mechanism would be more desirable, where no additional oxidants or reductants are required. In this context, the next two sections will describe reactions utilizing photocatalysts, demonstrating sustainable light-induced chemical reduction reactions, mimicking photosynthesis.

3. Photocatalytic carboxylation with CO₂

Photosynthesis is the master process in the realm of CO₂-functionalization, as it is called CO₂-fixation. This ideal process operates via multi-step electron transfer and chemical transformation reactions, resulting in somewhat limited CO₂-fixation efficiency, constraining the capacity of nature’s carbon cycle (Fig. 1 highlighted in green). Recent efforts in enzymatic engineering in chemical biology for in vitro CO₂ fixation would potentially lead to enhanced photosynthesis.

Solar energy obviously represents one of the most promising and limitless energy sources, which can be harnessed using a photosensitizer. In the late 1970s, seminal studies were reported regarding photocatalytic CO₂ reduction by the Tazuke, Fujishima, Honda, and Lehn groups, which formed the basis of the modern photoredox activation of CO₂. Further developments in photocatalysts played a significant role in CO₂ reduction reactions mainly targeting industrial feedstock.
Scheme 6  Ni-catalyzed dicarboxylation of 1,3-dienes and a mechanistic study.

**a) Scope of dicarboxylation**

\[
\begin{align*}
    &\text{NiBr}_2(\text{TBA})_2 (5 \text{ mol\%}), \text{L4} (5 \text{ mol\%}) \\
    &\text{Mn}, \text{CO}_2 (1 \text{ atm}), \text{DMA}, 50 ^\circ\text{C}, \text{then}, \text{TMSCHN}_2, \text{B}_2(\text{OH})_4\cdot\text{H}_2\text{O}
\end{align*}
\]

- **R** = H; 74%
- **R** = CF\textsubscript{3}; 89%
- **R** = F; 70%
- **R** = i-Bu; 64%
- **R** = OMe; 67%

\[
\begin{align*}
    &\text{NiBr}_2(\text{TBA})_2 (5 \text{ mol\%}), \text{L4} (5 \text{ mol\%}) \\
    &\text{Mn}, \text{CO}_2 (1 \text{ atm}), \text{DMA}, 50 ^\circ\text{C}, \text{then}, \text{TMSCHN}_2, \text{B}_2(\text{OH})_4\cdot\text{H}_2\text{O}
\end{align*}
\]

- **R** = Me; 90%
- **R** = CO\textsubscript{2}Me; 81%

- **R** = SnBu\textsubscript{3}; 54%
- **R** = CN; 51%

- **R** = Me; **R** = H; 75%
- **R** = H; **R** = Me; 68%

**b) Dicarboxylation of structurally simple dienes**

\[
\begin{align*}
    &\text{NiBr}_2(\text{TBA})_2 (5 \text{ mol\%}), \text{L4} (5 \text{ mol\%}) \\
    &\text{Mn}, \text{CO}_2 (1 \text{ atm}), \text{DMA}, 50 ^\circ\text{C}, \text{then} \text{H}_2
\end{align*}
\]

- **R**\textsuperscript{1} = **R**\textsuperscript{2} = H; 61%, 93.7
- **R**\textsuperscript{1} = Me; **R**\textsuperscript{2} = H; 63%, 99.1
- **R**\textsuperscript{1} = H; **R**\textsuperscript{2} = Me; 65%, 77.23

**c) Stoichiometric studies with Ni-3**

\[
\begin{align*}
    &\text{TBAB (equiv), L4 (equiv)} \\
    &\text{Mn (1 equiv), 50 ^\circ\text{C}, CO}_2 (1 \text{ atm})
\end{align*}
\]

- 37% (**E**/*Z* = 36:64)

Fig. 1  Natural photosynthesis and an example of artificial photosynthesis.
molecules, such as carbon monoxide, methanol, methane and formic acid. In this regard, artificial CO$_2$ functionalization reactions have shown elegant modes of action in C–C bond formation reactions. For example, a photochemical CO$_2$ fixation provided ε-amino acid derivatives in a one-step reaction (Fig. 1 bottom right). The key to the success of this field will be...
to maintain mild reaction conditions to conserve complex molecular structures of products, while providing appropriate reduction potential for the reductive CO₂-functionalization.

The Iwasawa group demonstrated a dual catalytic system with a Pd/Ir-photocouple for carboxylation reactions of aryl halides in the absence of metallic reducing reagents (Scheme 7).²⁵ Hünig’s base (3 eq.) served as a sacrificial electron donor in photoredox cycles, generating Pd(0)-complexes in the proposed catalytic carboxylation cycles (Scheme 10b). Although Ar–Pd(ii)–Br(XPhos) possesses a high reduction potential (−2.28 V, vs. Fc/Fc⁺), a new peak at −1.4 V was observed from cyclic voltammetry (CV) measurements. The coordination of CO₂ on Pd might influence the redox chemistry of the metal complex, therefore reducing the required reduction potential. In addition to the common insertion of CO₂ into the active Pd(ii)–C bond, the authors suggested the formation of two intermediates, a Pd(I)- or Pd(ii)–CO₂ complex (Scheme 7b, path b). After methylation with TMSCHN₂, various carboxylic acid esters were obtained including a sterically hindered acid (i.e. 2,4,5-trisopropyl carboxylic acid methyl ester).

Starting from simple feedstocks, Ar–Br and alkyl–Br, the König group reported visible light-induced carboxylation mediated by nickel catalysts (Scheme 8).⁵⁶ The plausible reaction mechanism could be divided into two distinct catalytic cycles. The first one involved a one electron delivery to a Ni(0) or Ni(i) complex from the anion radical [4CzIPN⁻]. Hantzsch ester (HEH, 2 equiv. required) was used as a terminal reducing agent in the presence of a reducing excited sensitizer [4CzIPN*] and light (left circle, Scheme 8b). Second, the oxidative addition to a Ni(0) complex was suggested, which undergoes reduction and then an insertion reaction with CO₂ (right circle, Scheme 8b). The catalytically active Ni(0) species can be regenerated from Ni(i) with electron sources produced from the left circle.

The same group expanded the dual catalysis strategy to the carboxylation of styrenes,⁵⁷ affording Markovnikov (branched) or anti-Markovnikov (linear) products selectively controlled by the choice of the ligand (Scheme 9). The suggested reaction mechanism explained that the observed chemoselectivity (branched/linear) was controlled by the different ligands (L₆, neocuproine and L₇, dppb). According to DFT calculations, Ni(0) species with the more sterically demanding ligand dppb (L₇) tend to coordinate with CO₂, forming a 5-membered nickelalactone (Ni-4) with styrene. With the less hindered neocuproine ligand (L₆), the reaction proceeds via hydrometalation of styrene to afford Ni(II), which is subsequently reduced by the catalytic action of the photosensitizer (4CzIPN, Scheme 9b). The

![Scheme 9 Site-selective photocatalytic carboxylation controlled by ligands.](image-url)
electrons generated from the photocatalytic cycle are used to reduce Ni(n) or Ni(i) to Ni(0), which can diverge to the hydro-
metalation step (left) and CO₂ activation step (right) to generate branched and linear products respectively while completing the catalytic cycles.

The Jamison group reported styrene functionalization reactions in a CO₂ atmosphere to generate β-aryl carboxylic acids (Scheme 10). In this case PMP (1,2,2,6,6-pentamethylpiperidine) was employed as a sacrificial organic electron donor, while utilizing water as an additive under modified reaction conditions. Although it is unclear, the addition of water induced high selectivity toward the mono-carboxylated product compared to other tested hydride or proton donors. The suggested reaction mechanism shows that the carboxylation with CO₂ results in the formation of a stabilized benzylic radical (E° = from −1.82 to −0.71 V vs. SCE). Therefore, further reduction is feasible leading to the generation of carboxylated benzylic anion species, which could be protonated upon addition of water.

Photoactivation of organic substrates has been a successful transformation with high chemoselectivity to produce Markov-
nikov (branched) or anti-Markovnikov (linear) carboxylic acids. Also, the Murakami group reported a carboxylation reaction with α-alkyl ketones and CO₂ via a Norrish type II activation mechanism. The carboxylation reactions at toluenyl carbon were also conducted in natural sunlight at ambient temperature with good isolated yields of the desired products. The authors suggested an energetically feasible [4 + 2]cycloaddition reaction by DFT calculations, which was determined by the thermal reaction of benzocyclobutenol to generate an o-quino
dimethane intermediate.

Recently, Hou et al. reported carboxylation reactions of internal and terminal alkynes promoted by Co/Ir dual catalysis (Scheme 11). The authors proposed that the reaction proceeded via functionalization of alkynes to generate an (E)-Co-CO₂ complex which is an intermediate for various products – carboxylic acids, pyrones, α,β-unsaturated γ-lactones, coumarins, and 2-quinolones, by sharing a common intermediate, (E)-int (Scheme 11a). Pyrones were formed through a formal [2 + 2+ 2] cycloaddition with terminal alkynes (R²=H). In the case of internal alkynes, pharmaceutically vital heterocycles such as coumarins and 2-quinolones were obtained with high selectivity. The suggested mechanism proceeded via intramolecular cyclization of acrylic acid intermediates. The E/Z isomerization of acrylic acid was confirmed by control experiments with (or without) the Ir-photoredox catalyst under irradiation (or in the dark). Also, this newly developed carboxylation/acyl-migration cascade reaction is feasible for alkyne difunctionalization, highlighting its utility in the field of light-driven CO₂-fixation.

The Yu group reported the photocatalytic hydro-
carboxylation of enamides and imines to afford ω-amino acids with excellent chemo- and regio-selectivity (Scheme 12). The pre-equilibrium of enamides and imines was combined with photocatalytic reduction. Despite the inherent nucleophilicity of enamides, kinetic studies indicated that the imines underwent the desired hydrocarboxylation faster than the competitive β-carboxylation reaction. The authors proposed an umpolung reaction of the ω-amino carbanion under metal-free conditions. The carboxylated products were obtained with a broad substrate scope regardless of the electronic and steric properties of substituents. In addition, the enamide and imine starting materials were equally effective, confirming the fast pre-equilibrium before the reduction/ carboxylation steps.

Very recently, the Walsh group presented photocatalytic carboxylation of benzophenone-derived ketimines by employing an Ir-complex ([Ir-I] under mild conditions (Scheme 13)). The radical anion was generated by single electron transfer (SET) from [Ir-I] to ketimines, which was facilitated by the coordination between the imine and Cy₂MeN⁺. Spin density calculation was carried out to evaluate the radical anions ([A, B] suggesting that the carbon atom was more negatively charged than the nitrogen atom (spin density, radical probability on C: 0.05–0.18 and N: 0.37). Subsequently, the more reactive N-centered radical species abstracts a hydrogen atom from Cy₂MeN⁺ to form an ω-amino carbanion and an iminium cation.

![Scheme 10](image)

Scheme 10 Photocatalytic direct β-selective hydrocarboxylation of styrenes.
[Cy2N=CH2]⁺ via an umpolung reactivity (Scheme 13b). The carbanion then undergoes nucleophilic addition to CO₂ affording the desired carboxylation product. The obtained α,α-disubstituted α-amino acid shows potential application of the protocol in asymmetric synthesis to generate quaternary stereogenic centers, which are often difficult to control.⁶⁵

Direct carboxylation of imines and amines with CO₂ represents a very promising pathway to afford α-amino acids, especially those promoted by photoredox catalysts as shown above (Schemes 12, 13 and 17). Compared to tertiary amines, however, α-functionalization (i.e. α-carboxylation) of primary amines still remains a great challenge due to the lower reactivity of the α-C–H bond. Besides carboxylation reactions, CO₂ has been used as an activating group,³³,a directing group⁶⁶ and a protecting group⁶⁷ in organic synthesis. Ye et al. recently reported the photocatalytic α-alkylation of primary amines to yield γ-lactams with CO₂ as a temporary activator and as a protecting group (Scheme 14).⁶⁸ Various α,β-unsaturated esters were tolerated in the presence of an Ir-II photosensitizer. Quinuclidine was employed as a sacrificial electron donor. According to the suggested reaction mechanism, CO₂ was regenerated releasing lactam products via an intramolecular cyclization reaction. The in situ carbamate formation reaction suppressed the reactivity of primary amines while increasing the reactivity of α-C–H bonds according to the computational studies. The generation of the α-radical of the substrate is highly intriguing due to the potential applications toward various electrophiles and radical–radical coupling reactions. Furthermore, the use of
Scheme 12 Photocatalytic hydrocarboxylation of enamides and imines.

a) Reaction conditions

Conditions: C8H4CO2, 4CzlPN (2 mol%), Pr2NEt, DMF, rt., blue LEDs, 4 h, then 2 M HCl

b) Plausible mechanism

R = H, 83%; p-Me, 74%; p-Cl, 82%; p-CF3, 72%; m-OMe, 69%; m-CF3, 74%; m-Cl, 80%; o-Cl, 87%; o-F, 93%.

84% 76%

Scheme 13 Photocatalytic carboxylation of ketimines.

a) Reaction conditions

i) Ir-1 (0.5 mol%), Cy2NMe (2 eq.), CO2 (balloon), MeCN, Blue LEDs, rt.

ii) TMSCHN2, MeOH/Et2O

b) Plausible mechanism

c) Selected examples

scope of imines

R = H, 87%; p-OMe, 95%; p-CF3, 86%; m-F, 91%.

86% 65%
tertiary amines as a base will enable a potential asymmetric catalysis to afford enantioenriched products.

This section summarizes recent progress in photo-CO$_2$-functionalization without strong metallic reducing agents. Instead, an organic sacrificial electron source or a reducing reagent was employed (i.e. triethylamine, piperidine, Hünig’s base and Hantzsch esters) in the presence of photocatalysts with an appropriate reduction potential to complete the catalytic cycles. Various types of substrates underwent C–CO$_2$ bond formation reactions to provide unique molecular structures under ambient photosynthetic conditions (low CO$_2$ pressure, and accessible light sources). However, there is still plenty of room to develop more elegant methodologies in terms of sustainability. The next section will discuss redox-neutral carboxylation without external reductants.

4. Recent developments in redox-neutral CO$_2$-functionalization

It is thought that catalytic carboxylation of non-activated organic substrates would be an ideal approach to CO$_2$-utilization, avoiding reactive organometallic reagents (RMgX, RLi, R$_2$Zn, R$_4$Sn, etc.). For example, solar energy provides chemical reduction potential to enable CO$_2$ conversion in the Calvin cycle, where actual CO$_2$-fixation and C–CO$_2$ bond formation reaction occur under mild conditions via an $\alpha$-ketol
This “enantioselective” CO$_2$-fixation process generates a new C–C bond while creating additional stereogenic center(s) via a redox-neutral pathway. Accordingly, recent progress in photo-redox catalysis offers a promising platform to develop sustainable CO$_2$ utilization reactions under mild conditions in the absence of additional reducing reagents.$^{54-78}$ When combined with practicability and scalability, redox-neutral CO$_2$-functionalization strategies will provide a tangible scenario of sustainable artificial carbon fixation.

The following examples in this section represent their redox-neutral reaction profile in terms of the proposed reaction mechanisms – no terminal reducing or oxidizing reagents. Despite the fact that these reactions require activated substrates or radical initiators or a strong base, the generation of C–CO$_2$ bonds with CO$_2$ is a remarkable step toward truly ideal CO$_2$-functionalization. Keeping in mind that solar energy might be the only and truly sustainable energy source, a few examples of redox-neutral photocatalytic CO$_2$-functionalization reactions are also highlighted in this section.

The Sato group recently reported a direct carboxylation reaction at the allylic C(sp$^3$)–H bond (Scheme 15).$^{79}$ The use of the AlMe$_3$ – non-nucleophilic base – was ascribed to the initial generation of catalytically active Co(0) species, therefore the catalytic cycle is free from an external reducing reagent. The carboxylation reaction of allylarenes and 1,4-dienes was proven to be effective with a nucleophilic η$^1$-allyl-Co(0) catalyst after intensive screening of transition metal catalysts such as Cr(0), Mn(0), Fe(0), Rh(0), Ir(0), Ni(0) and Cu(0). The role of the ligand was critical; Xantphos (L9) showed high selectivity without the formation of isomerization or methylation byproducts by the use of AlMe$_3$. Various terminal alkenes were smoothly converted to β,γ-unsaturated acids with excellent functional group tolerance, including amides, esters, and ketones. The authors suggested that the presence of the low-valent Co(0)-complex was the key to the successful carboxylation reaction with high selectivity. This protocol expands upon the scope of carboxylation to C(sp$^3$)–H bonds, which represents atom- and step-economic approaches to construct molecular complexity by incorporating CO$_2$.

Very recently, the Yu group reported photocatalytic carboxylation of tetraalkyl ammonium salts via C–N bond cleavage (Scheme 16).$^{72}$ Trimethylamine was generated in situ by single-electron transfer (SET) from the excited Ir I to the substrates. In turn, the resulting active Ir I could be reduced by the tertiary amine. Afterwards, carbanions undergo a carboxylation reaction after another SET step between the excited photoredox catalyst and the alkyl radical. The authors suggested that the oxidized trimethylamine was transformed to amine species, like α-radical [Me$_2$NCH$_2$], or dimethylamine after hydrolysis. As electron donors, trimethylamine and dimethylamine accounted for 2 equivalents of reducing reagents required to complete the catalytic cycle. This built-in reductant was generated and demonstrated carboxylation reactions without additional reducing reagents, compared to Ni-catalyzed reductive carboxylation of benzylic C–N bonds.$^{31}$

In the above-mentioned cases, organic amines act as sacrificial electron donors, where the resulting radical cation trialkyl amines have dramatically reduced pK$_a$ at the α-protons.$^{27}$ In the presence of a base, a deprotonation reaction would generate an amine with an α-radical, which can couple with other reactive species. The single-electron reduction of CO$_2$ to CO$_2^-$ is in general a rate-determining step due to the high reduction potential ($\sim 2.21$ V vs. SCE (saturated calomel electrode) in DMF (N,N-dimethylformamide)).$^{24}$ A viable C–C bond formation

**Scheme 15** Cobalt-catalyzed direct carboxylation of allylic C(sp$^3$)–H bonds.
reaction with $\text{CO}_2$ and amine based $\alpha$-radicals would afford $\alpha$-amino acids as the product. This was realized by the Jamison group demonstrating a metal-free photoredox conversion of $\text{CO}_2$ (Scheme 17). An organic sensitizer, $p$-terphenyl, mediated single electron transfer reactions (reduction potential: $-2.63 \text{ V vs. SCE}$ in DMF) to perform the suggested one-electron reduction of $\text{CO}_2$, providing $\alpha$-amino acids in the absence of additional reducing reagents. Various aryl-substituted $\alpha$-amino acids were prepared in good to excellent yields. The convenience of continuous flow chemistry was an added benefit of the photocatalysis to provide essential synthetic building blocks from carbon dioxide. The generation of $\text{CO}_2$-radical anion is highly attractive, considering its vast application potential in organic synthesis for carboxylation reactions. This photo-catalysis mediated by $p$-terphenyl showed promise toward metal-free $\text{CO}_2$-functionalization via a single-electron reduction mechanism in terms of atom-economy (redox-neutral), feasibility (continuous flow setups), and utility of the final products ($\alpha$-amino acids) containing stereogenic centers.

Owing to the recent developments in organic photosynthesis and photosensitizers, unprecedented reactivity patterns were achieved with $\text{CO}_2$ as a C1 source. For example, the Martin group showed photocatalytic dicarbofunctionalization of styrene derivatives initiated by radicals under mild reaction conditions, where stabilized benzyl carbanions react with $\text{CO}_2$ (Scheme 18). Various radical initiators, such as trifluoro- and difluorosulfonates, and trifluoroborate salts, were proven to be effective under photochemical reaction conditions. The photocatalytic redox cycle was mediated by an Ir-complex ($\text{Ir-II}$). This protocol provides two new C–C bonds with a stereogenic center in the absence of additional stoichiometric reducing reagents. Trisubstituted alkenes were also employed to afford carboxylic acids with a quaternary stereogenic center. The convenient introduction of the (di)trifluoromethyl group highlights potential applications of radical carboxylation reactions in drug discovery and pharmaceutical industry.

The Yu group developed the first thiocarboxylation of styrenes by using an Fe/S complex as the photosensitizer.

Scheme 16 Photocatalytic carboxylation with a built-in reductant as the electron donor.

Scheme 17 Photoredox $\text{CO}_2$-activation to access $\alpha$-amino acids using a $p$-terphenyl photosensitizer.
Various β-thioacids were synthesized selectively with different regioselectivities from the previous protocol (Scheme 18). Mechanistic studies revealed that single-electron reduction of CO₂ can be initiated by the excited Fe/S complex, yielding the CO₂ radical anion (CO₂⁻). This radical intermediate was trapped subsequently by an alkene substrate to generate a stabilized alkyl radical, which led to anti-Markovnikov regioselectivity. Thiolation of alkyl radicals was mediated by the [Fe/S] radical cation, highlighting the application potential of the methodology in the synthesis of β-thioacids – an intermediate for the antidepressant drug thiazensim. Also, considering the Fe- and S-rich environment in the prebiotic era, the presented reaction could help us to rethink the CO₂ chemistry in the primordial soup, potentially affording complicated photoredox reactions with CO₂ to furnish chiral molecules.

(Scheme 19). Various β-thioacids were synthesized selectively with different regioselectivities from the previous protocol (Scheme 18). Mechanistic studies revealed that single-electron reduction of CO₂ can be initiated by the excited Fe/S complex, yielding the CO₂ radical anion (CO₂⁻).
The progress in redox-neutral CO2 functionalization showed elegant reaction mechanisms operating under mild conditions, for example, via CO2 insertion into metal–carbon bonds or CO2⁺⁻ captured by activated substrates. This represents a promising and ideal mode of action, whereby no additional sacrificial redox agents were applied to construct multiple C–C, and C–X bonds. Thus, high atom economy and step-efficiency are expected in constructing molecules with CO2 as a non-toxic C1 source, boosting research in CO2-utilization from recently developed dicarbfofunctionalization. Meanwhile, the structural diversity of recent CO2-functionalization reactions shows the significant potential of CO2 in asymmetric synthesis and catalysis. Further investigations on the asymmetric activation of CO2 and its utilization in CO2-functionalization will allow us to achieve higher values of products while recycling CO2.

5. Asymmetric catalytic carboxylation with CO2

Besides the asymmetric synthesis of cyclic carbonates or poly-carbonates with epoxides or diols,⁷⁻¹² the construction of enantioselective C–CO2 bonds using CO2 has been a formidable challenge under the influence of chiral catalysts or chiral environments. This is due to the high stability of CO2, limiting the scope of reaction partners; highly reactive organometallic species and/or harsh reaction conditions are necessary thus low stereoselectivity is in general expected. In 2004, the Mori group reported the carboxylative cyclization reaction of bis-1,3-dienes catalyzed by a Ni catalyst (Scheme 20). The authors performed facile 5-membered ring formation reactions in the presence of excess amounts of dialkyl zinc (4.5 equiv.). The obtained products possess three consecutive stereogenic centers with absolute diastereoselectivity with good yield and excellent enantioselectivity.

In 2017, the Marek group developed an enantioselective Cu-catalyzed carboxamidation reaction of cyclopentanes, which could be selectively carboxylated with CO2 as an electrophile (Scheme 21). High diastereoselectivity was observed which is not fully understood yet based on the control experiment without the copper catalyst (racemic but moderate diasteroselectivity, 9 : 1 dr). Other electrophiles such as iodine, bromine and allyl bromide were smoothly incorporated to furnish the desired products. Although Grignard reagents are reactive nucleophiles, the sequential addition of the alkene and CO2 prevented direct attack of these nucleophiles on CO2 at low reaction temperature (0 °C) in the presence of a copper catalyst. The observed stereoselectivity was attributed to the stability of the stereogenic center at the carbon–Cu moiety, explaining the cis geometry between the nucleophile and electrophilic CO2.

The Yu group recently reported a highly regio- and enantioselective copper-catalyzed CO2-functionalization reaction of olefins owing to enantioselective Cu–H catalysis⁸⁴ (Scheme 22). Inspired by the CO2 reduction reaction to methanol⁸⁵ and other higher alcohols, the authors developed the sequential enantioselective Cu–H addition, carboxylation and reduction reactions to achieve hydroxymethylation of olefins. A preliminary mechanistic study revealed that the L12 Cu–R (C) species showed no reactivity toward reduced CO2 [R,Si–OC(OH)] (dashed arrow), indicating the direct carboxylation of C in the chiral environment to ensure the obtained high enantioselectivity. Furthermore, the developed methodology was applied to 1,3-dienes, affording (Z)-selective homoallylic alcohols with good enantioselectivities. Further derivatization of the hydroxymethylation products afforded elegant syntheses of enantoenriched (R)-(−)-curcumene⁸⁷ and (S)-(+)–ibuprofen, starting from CO2 as a C1 building block.

Although asymmetric catalytic C–C bond formation has achieved relatively considerable progress, only a few methodologies have been reported with CO2 as a sustainable C1 source while creating stereogenic center(s) with high stereoselectivity.
Considering that the carbon fixation process produces carbohydrates and biomass with absolute enantioselectivity, it is a logical extension to implement asymmetric carboxylation reactions in artificial CO$_2$ fixation. Chemical synthesis offers various synthetic pathways and tools that can be easily tested, potentially providing a playground for facile screening and method development. For example, photochemical reactions with chiral catalysts including a chiral iridium catalyst or Lewis-acid assisted photocatalysis for CO$_2$-functionalization are seemingly feasible methods to be developed. Considering the mode of action of RubisCo enzyme, redox-inactive metals and ligands (e.g. Mg–biotin complex) would be critical to improve the availability of CO$_2$ in organic reactions. On the other hand, it could be inferred that chiral CO$_2$-complexes may play a significant role in CO$_2$-activation via bifunctional asymmetric catalysts. It would be exciting to see the development of CO$_2$-functionalization, with foreseeable sustainability and increased utility of the final products in organic synthesis.

6. Conclusion and outlook: umpolung reactivities towards CO$_2$

It is a formidable challenge to define an “ideal” carbon dioxide functionalization considering that many factors – environmental impact, atom-economy, sustainability, utility of products, and reaction conditions – are involved in designing reaction processes. Harnessing the full capacity of CO$_2$-functionalization can be envisaged with sustainable and accessible chemical feed stocks, catalysts, and reaction conditions.

Scheme 21  Asymmetric carbomagnesiation/carboxylation of cyclopropanes.

Scheme 22  Enantioselective incorporation of CO$_2$ via hydrocarboxylation of styrene derivatives and dienes.
Victor Grignard, in 1912, stated this in his Nobel Lecture – “Willstatter in fact recognized that … organic magnesium compounds must form and that the absorption of CO₂ gas by chlorophyll would in every way be comparable to the Grignard reaction”. The mode of action of magnesium compounds in chlorophyll differs from what Grignard speculated, however, one of the earliest umpolung reactions with CO₂ and Grignard reagents paved a way for modern CO₂ functionalization to date. Considering the formation of Grignard reagents, an umpolung process utilized polarized bonds, C⁺⁻⁻X⁻⁻⁻⁻⁻, by inverting the electronic nature of the carbon to nucleophilic by forming C⁺⁻⁻Mg⁻⁻⁻⁻⁻X.

In this context, recent developments in umpolung carboxylation reactions have shown unprecedented reaction patterns (Scheme 23):⁶¹,⁶²,⁹³ for example, umpolung reactivity has been implemented to functionalize CO₂ for an aldehyde carboxylation reaction through a redox-neutral mechanism (Scheme 23a).⁹³ The obtained product, ω-keto acid, was smoothly converted to the corresponding ω-amino acid under reductive amination reaction conditions mimicking the biosynthesis of various amino acids. The use of nitrogen-containing nucleophiles offers direct synthesis of ω-amino acid derivatives (Scheme 23c–e). By employing cyanohydrin, hydrazone, photocatalysts, and a base, in situ generated umpolung species were transformed to the desired carboxylated products under mild reaction conditions, with or without reducing agents. This is particularly interesting to hypothesize the evolution of ω-amino acids from the CO₂-rich prebiotic environment. It has been postulated that cyanide is an abundant source of a carbon nucleophile in the synthesis of biologically active molecules in the primordial soup.⁹⁴ The use of aldehydes, cyanide, and CO₂ in synthesizing biologically active molecules is a promising step toward answering the important question: what is the origin of life? Was there an involvement of photochemical CO₂-activation? Was it promoted by an optically pure component to induce homochirality? The forthcoming ideal CO₂-functionalization may answer these conundrums.

In summary, this Perspective collects the recent literature in CO₂ functionalization and groups it into four categories: (1) metal catalyzed direct carboxylation, (2) photocatalytic carboxylation reactions, (3) redox-neutral carboxylation, and (4) asymmetric introduction of catalytic CO₂ for C–C bond formation reactions. Even a broad scope of substrates and remote site functionalization were achieved; transition metal-catalyzed reductive carboxylation is mostly limited to CO₂ insertion reaction with (over)stoichiometric amounts of reducing reagents. However, photoredox catalysts present promising access to more diversified CO₂ reactions, like dicarbofunctionalization, single-electron reduction and radical coupling via...
a redox-neutral mechanism. Thanks to these developments of methodologies, as discussed at the end of Section 5, more examples in challenging enantioselective C-CO2 bond formation will be realized in the near future. Although enzymatic CO2 functionalization reactions are not covered in this Perspective, they have shown their very promising application in artificial carbon recycling processes.11–15 Synergetic and interdisciplinary CO2 fixation with biological and chemical catalysts will be particularly interesting in (asymmetric) photocatalytic conversion of CO2, truly mimicking photosynthesis to provide ideal CO2 functionalization reactions.

Conflicts of interest

There are no conflicts to declare.

References

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