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Enantioselective and site-specific copper-catalyzed reductive allyl–allyl cross-coupling of allenes†

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A copper-catalyzed asymmetric reductive allyl–allyl cross-coupling reaction of allenes with allylic phosphates wherein allenes were used as allylmetal surrogates has been achieved for the first time. This protocol provides an efficient and straightforward route to optically active 1,5-dienes in a highly enantioselective and site-specific fashion. Furthermore, all-carbon quaternary stereogenic centers could also be constructed with this protocol. The versatility of the products is demonstrated through a diverse array of further transformations of the enantioenriched 1,5-dienes.

Introduction

The development of highly efficient and straightforward methodologies for creation of configurationally well-defined stereocenters during the course of C–C bond formation is of paramount importance in organic synthesis.¹ Among such developments, metal-catalyzed asymmetric allyl–allyl cross-coupling reactions between allylic electrophiles and allylmetal reagents have been regarded as particularly valuable methods, due to not only the ability to establish a new stereogenic center with concomitant introduction of two differential flexible allyl functional groups in one synthetic operation, but also providing a convenient route to access highly important chiral 1,5-diene structures, which serve as highly versatile building blocks in organic synthesis and are also found in many important biologically active molecules, such as plakortide E, FK-506, and chaetoglobosin (Fig. 1a).² In this respect, several transition metal complexes of Pd, Au, Cu, and Ni had been successfully employed to catalyze such racemic transformations over the past few years;³ nevertheless, the development of an asymmetric version to expedite access to enantioenriched 1,5-dienes remains a long-standing challenge.

Recently, Morken and co-workers described a Pd-catalyzed allyl–allyl cross-coupling of allylic electrophiles with allylboron reagents in a highly regio- and enantioselective manner through the use of a key chiral small bite-angle bidentate phosphine

ligand.⁴ Subsequently, the group of Carreira also developed phosphoramidite-ligated Ir-catalyzed asymmetric coupling reactions between secondary allylic alcohols and allylsilanes.⁵ More recently, by employing Earth-abundant base metals as the catalysts, Feringa, Ohmiya and Sawamura independently reported Cu-catalyzed regio- and enantioselective allyl–allyl cross-coupling of allylic bromides or phosphates with corresponding allylic Grignard reagents or allylic boronates (Fig. 1b).⁶ In addition, Hoveyda and co-workers also elegantly disclosed an approach for the synthesis of chiral boron-containing 1,5-dienes bearing a tertiary stereogenic centre through Cu-catalyzed boron allylation of monoalkyl-substituted allenes.⁷ Notwithstanding these advances, stoichiometric preformed

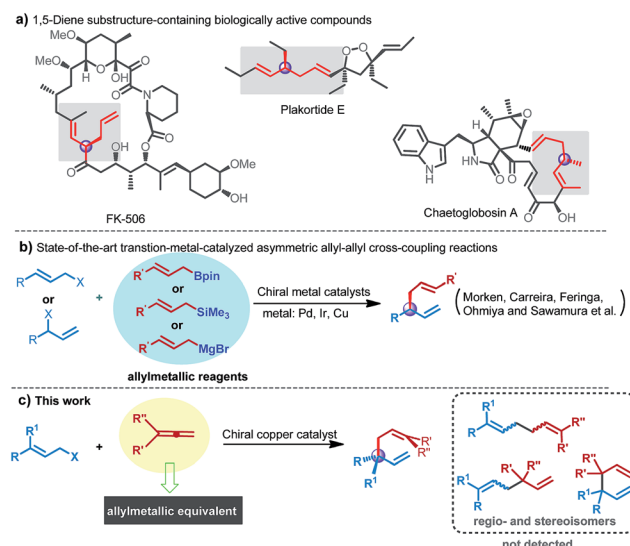


Fig. 1 Representative compounds and synthetic methods of 1,5-dienes.

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allylic metals have been generally indispensable in both racemic and asymmetric transformations to date,^{3–6} which significantly impedes further applications because of the intrinsic limitations associated with these reagents, such as the need for multistep operations of unsaturated hydrocarbons or organic halides, uncontrollable reactivity profiles, sensitivity toward water and poor functional group compatibility. Moreover, the more daunting issue of the enantioselective generation of all-carbon quaternary stereocenters during these processes remains an unmet synthetic challenge, owing to increased energy barriers as a result of the highly congested nature and the diminished steric bias compared with those of the corresponding chiral tertiary centers leading to more difficulty in discriminating enantiotopic faces.⁸ So far, to our knowledge, the sole reported example was demonstrated by Morken *et al.* through utilization of allylboron as the nucleophile with a precious Pd catalyst.^{4b} Thus, the development of an alternative method for enantioselective allyl–allyl cross-coupling to furnish versatile 1,5-dienes that have the ability to create an all-carbon quaternary stereocenter with concomitant avoidance of the use of preformed allylic metals (or metalloids) and precious metal catalysts is highly desirable.

Recently, an interesting protocol for utilizing readily available and stable unsaturated hydrocarbons as latent carbon nucleophiles *in lieu* of preformed organometallic reagents for coupling with various electrophiles in transition-metal catalysis has attracted tremendous attention.⁹ In this regard, Buchwald,¹⁰ Yun,¹¹ Hoveyda¹² and our group¹³ have disclosed copper hydride (CuH)-catalyzed regio- and enantioselective reductive allylation (or hydroallylation) reactions using olefins or alkynes as latent alkyl- or vinylmetal reagents. Inspired by these achievements, we envisioned whether the readily available allenes could serve as latent effective allylic metal equivalents to couple with allylic electrophiles leading to synthetically important 1,5-diene-containing molecules. In fact, allene-derived allylic nucleophiles that react with ketone, imine, alkyl triflates and CO₂ under reductive conditions have been described by Tsuji, Buchwald and Lalic in the past few years.¹⁴ In comparison, the more intricate reductive allyl–allyl cross-coupling reaction remains unexploited to date, as exploiting such a reliable catalytic method that could well balance the reactivity profiles (the chemoselectivity of CuH species to allene *versus* allylic electrophile)^{13,15} regio- and stereocontrol, as well as having the power to establish an all-carbon quaternary stereocenter. Herein, we report the first Cu-catalyzed site-specific and enantioselective reductive allyl–allyl cross-coupling of allenes with allylic phosphates to afford enantioenriched 1,5-dienes without the use of any pre-formed allylic metals,¹⁶ highlighting the ability to construct more hindered all-carbon quaternary stereocenters by this approach (Fig. 1c).

Results and discussion

We initiated our study by choosing allene **1a** as the latent allylic metal reagent to couple with allylic phosphate **2a** under our previously reported hydroallylation conditions (Table 1).¹³ Pleasingly, the expected S_N2'-type reductive allyl–allyl cross-

Table 1 Optimization of the reaction conditions^a

Entry	Ligand	Solvent	Yield ^b (%)	ee ^c (%)
1	L1	THF	50	33
2	L2	THF	0	0
3	L3	THF	50	89
4	L4	THF	0	0
5	L5	THF	0	0
6	L6	THF	0	0
7	L3	Toluene	70	76
8	L3	Dioxane	60	93
9	L3	Cyclohexane	66	77
10 ^d	L3	Dioxane	73	93
11 ^e	L3	THF	0	0

L1 R = 2,4,6-(Me)₃

L2 R = 2,4,6-(*i*-Pr)₃

L3 R = 3,5-[2,4,6-(*i*-Pr)₃C₆H₂]₂

L4 Ar = 2,4,6-Me₃C₆H₂

L5

L6

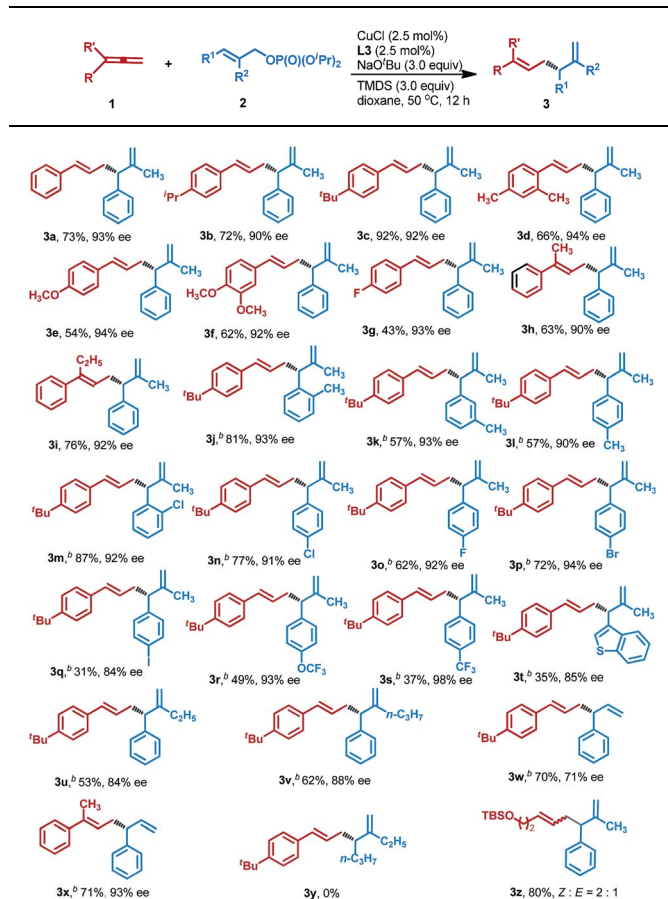
^a Reaction conditions: **1a** (0.2 mmol), **2a** (0.3 mmol), CuCl (2.5 mol%), ligand (2.5 mol%), NaOtBu (0.6 mmol) and TMDS (0.6 mmol) in 2 mL solvent at 50 °C in a N₂ atmosphere unless otherwise stated.

^b Isolated yield. ^c ee determined by HPLC. ^d **1a** (0.3 mmol) and **2a** (0.2 mmol). TMDS = (Me₂HSi)₂O. ^e Using –Br or –OC(O)O^tBu as the leaving group in the allyl electrophile.

coupling product **3a** was obtained in a moderate yield with exclusive regioselectivity, albeit low enantioselectivity, under these catalytic conditions (entry 1). In view of the unique ability of chiral N-heterocyclic carbene (NHC) to facilitate Cu-catalyzed S_N2'-type allylation,^{12,13,17} we therefore briefly surveyed various chiral NHCs, such as **L2**–**L4**, and **L3** (ref. 17c and d) showed acceptable reactivity profiles and excellent enantiocontrol (entries 2–4). Likewise, chiral bisphosphine ligands **L5** and **L6**, which exhibited excellent reactivity and enantioselectivity for the reported allyl–allyl cross-coupling reactions,^{4–6} however, were ineffective in this catalytic system (entries 5 and 6). To our delight, after subsequent solvent screening and switching the ratio of allene **1a** and allylic phosphate **2a** (entries 7–10), the desired product **3a** could be isolated in 73% yield with 93% ee (entry 10). In addition, the other allylic electrophiles, including analogous allyl bromide and carbonate, were found to be completely ineffective under the present catalytic conditions (entry 11). Remarkably, under all conditions only regioisomer **3a** was obtained, suggesting the exclusive regioselectivity. Further evaluation of various copper catalysts, ligands, bases, allylic phosphates and hydrosilanes is discussed in the ESI.†

Having identified the optimal conditions, we explored the substrate scope with respect to various allenes **1** and allylic phosphates **2** (Table 2). The allenes bearing either aliphatic substituents, alkoxyl or halogen on the aryl rings, generally displayed high levels of reactivities and excellent enantioselectivities. A relatively lower efficiency was observed for the allene **1g**, which is presumably ascribed to its relatively electron



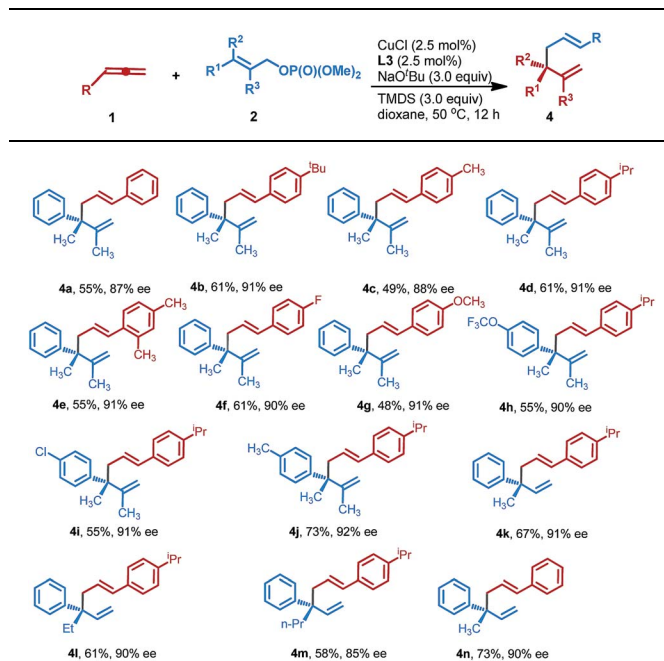
Table 2 Substrate scope for the synthesis of tertiary stereogenic center-containing 1,5-dienes^a

^a Reaction conditions: **1** (0.3 mmol), **2** (0.2 mmol), CuCl (2.5 mol%), L3 (2.5 mol%), NaOtBu (0.6 mmol) and TMDS (0.6 mmol) in 2 mL dry dioxane at 50 °C in a N₂ atmosphere unless otherwise stated. Isolated yield. ^b -OP(O)(OMe)₂ as the leaving group.

deficient nature, and this result is in accordance with a recent report by Hoveyda and co-workers.¹⁸ 1,1-disubstituted alkenes **1h** and **1i** could also be efficiently converted into corresponding 1,5-dienes **3h** and **3i** in good yields with excellent enantioselectivities. We next assessed the scope of allylic phosphate components under these conditions. We found that a variety of functional groups were well tolerated, including alkyl, -F, -Cl, -Br, -I, -OCF₃ and -CF₃, providing the expected chiral 1,5-dienes **3j**–**3s** in moderate to good yields with generally excellent enantiocontrol. We noted that the substituents at either *ortho*-, *meta*- or *para*-positions of the aromatic rings in the allylic phosphates, such as **3j**–**3l** and **3m**–**3n**, had a negligible effect on enantiocontrol, while *ortho*-substituted allylic phosphates showed higher reactivities. Furthermore, the introduction of halogens in the aryls in **3m**–**3q** could provide more opportunities for further elaboration, particularly for the high reactivity of aryl-I, which is rarely tolerated in CuH-catalyzed transformations. Heteroaromatic ring-containing allylic phosphates and β-ethyl-, propyl- or H-substituted allylic phosphates also enabled the reaction to proceed smoothly, furnishing chiral 1,5-

dienes **3t**–**3w** in acceptable yields with good enantioselectivities. Additionally, 1,1-disubstituted allene was also a valid substrate for this transformation and the corresponding product **3x** was obtained with excellent enantioselectivity, but alkyl substituted allylic phosphate was not suitable for this reductive cross-coupling reaction. By comparison, in the case of alkyl substituted allene, the corresponding product 1,5-diene **3z** could be efficiently formed, although a mixture of alkene isomers was observed, and this result is in line with the more recent findings of Tsuji.¹⁶

We considered that 1,5-dienes bearing all-carbon quaternary stereocenters widely distributed in many important biologically active molecules have significant value in organic synthesis,^{8,19} while the synthesis of such highly congested stereocenter-containing molecules remains an unmet challenge. Thus, we were particularly interested in whether this protocol would be applicable for the construction of an all-carbon quaternary stereogenic center. After briefly screening the reaction conditions, to our delight, the anticipated quaternary carbon center-containing chiral 1,5-diene **4** could be readily obtained with exclusive regioselectivity and excellent enantioselectivity. As depicted in Table 3, various aliphatic substituents, -OCH₃ and -F, on the aromatic rings of the allenes could be efficiently transformed into the corresponding 1,5-dienes **4a**–**4g** in moderate to good yields with generally excellent enantiocontrol. Additionally, a number of allylic electrophiles, including different substituents on the aryl ring, and β- or γ-position of allylic phosphates, could also be cross-coupled efficiently,

Table 3 Substrate scope for the synthesis of quaternary stereogenic center-containing 1,5-dienes^a

^a Reaction conditions: **1** (0.3 mmol), **2** (0.2 mmol), CuCl (2.5 mol%), L3 (2.5 mol%), NaOtBu (0.6 mmol) and TMDS (0.6 mmol) in 2 mL dry dioxane at 50 °C under a N₂ atmosphere unless otherwise stated. Isolated yield.



affording optically active 1,5-dienes **4h–4n** with high enantioselectivities. Finally, the absolute configuration of the major enantiomer was determined as (*R*) by comparison of the optical rotation of **4n** with a reported value.^{4b}

To showcase the synthetic utility of chiral 1,5-dienes, the reductive allyl–allyl cross-coupling product **3x** could be expediently converted into a series of synthetically valuable optically active compounds **5–8** with exclusive chemoselectivity (Fig. 2). For example, chiral 1,5-diene **3x** was subjected to olefin cross-metathesis conditions with vinylB(pin), providing the versatile building block chiral vinyl borate **5** in good yields. In addition, **3x** could also be efficiently transformed into α , β -unsaturated ester **6** in good yields under the same reaction conditions. Furthermore, chiral alcohol **7** and multi-aromatic compound **8** could also be efficiently obtained *via* the hydroboration/oxidation or hydroboration/Suzuki reaction.

To gain some insight into the mechanism, we assessed the impact of different allylic phosphate isomers on the reactivity, and regio- and enantioselectivity. As depicted in Fig. 3a, besides (*E*)-allylic phosphate, (*Z*)-allylic phosphate and racemic secondary allylic phosphate were also valid substrates and exhibited identical reactivity and exclusive regioselectivity, despite the relatively lower enantiocontrol of (*Z*)-allylic phosphate (Fig. 3a, eqn (1)–(3)). Moreover, according to previous reports⁶ and regioconvergent outcomes, which suggested that these transformations might produce the same allylcopper(III) intermediate **III** prior to reductive elimination to form the branched chiral 1,5-diene **3w** (*vide infra*). Based on this experimental result and the previously proposed mechanism for Cu-

catalyzed allylic substitutions,²⁰ a possible mechanism was described in Fig. 3b. Initially, chemoselective insertion of the terminal allene into catalytic ligated L^{*}CuH species **I**, which is derived from the *in situ* generated L^{*}CuOR with hydrosilane through a direct metathesis process, could catalytically form allylcopper species **II**. Subsequently, oxidative addition of intermediate **II** with an allylic electrophile would furnish 16-electron allylcopper(III) intermediates **III** and **IV**, which probably equilibrate *via* σ – π – σ isomerization. Finally, the chiral regioisomer 1,5-diene **3** was generated *via* reductive elimination and simultaneous release of the Cu(I) species, which then reacts with a hydridic silane *via* a σ -bond metathesis pathway, regenerating the L^{*}CuH catalyst.

Conclusion

In summary, a copper-catalyzed asymmetric reductive allyl–allyl cross-coupling reaction of allenes with allylic phosphates was achieved for the first time. This protocol exhibited good functional group tolerance, exclusive regioselectivity and highly enantioselective control, and provides a facile route to access chiral 1,5-dienes bearing tertiary or more hindered all-carbon quaternary stereocenters. Studies on further expanding this strategy including utilizing different coupling partners for enantioselective construction of C–C bonds are being carried out in our laboratory.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

- (a) N. Jacobsen, A. Pfaltz and H. Yamamoto, *Comprehensive Asymmetric Catalysis: Suppl. 2*, Springer-Verlag, Berlin, 2004; (b) B. M. Trost, *Proc. Natl. Acad. Sci. U. S. A.*, 2004, **101**, 5348; (c) Z. Lu and S. Ma, *Angew. Chem., Int. Ed.*, 2008, **47**, 258; (d) G. Helmcherm, U. Kazmaier and S. Förster, in *Catalytic Asymmetric Synthesis*, John Wiley & Sons, Hoboken, NJ, 2010; (e) E. Negishi, *Angew. Chem., Int. Ed.*, 2011, **50**, 6738.
- (a) E. Breitmaier, *Terpenes, Flavors, Fragrances, Pharmaca, Pheromones*, Wiley-VCH, Weinheim, 2006; (b) P. M. Dewick, *Medicinal Natural Products: A Biosynthetic Approach*, Wiley, Chichester, 2002.
- For selected examples, see: Pd-catalyzed allylation: (a) H. Nakamura, M. Bao and Y. Yamamoto, *Angew. Chem., Int. Ed.*, 2001, **40**, 3208; (b) E. F. Flegeau, U. Schneider and S. Kobayashi, *Chem.–Eur. J.*, 2009, **15**, 12247 Au-catalyzed allylation: (c) S. Porcel, V. López-Carrillo, C. García-Yebra

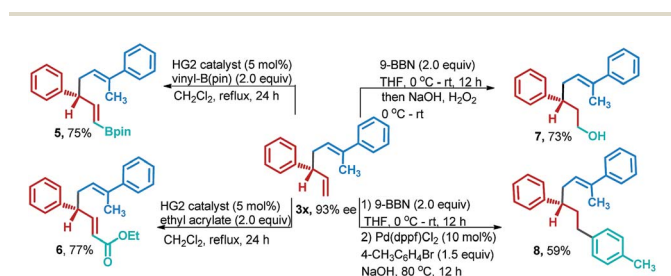


Fig. 2 Further applications of chiral 1,5-diene **3x**.

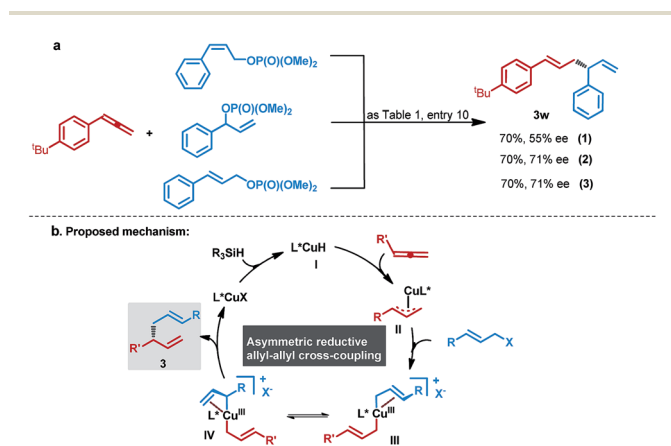


Fig. 3 Control experiments and the proposed mechanism.



- and A. M. Echavarren, *Angew. Chem., Int. Ed.*, 2008, **47**, 1883; Cu-catalyzed allylation: (d) A. Yanagisawa, N. Nomura and H. Yamamoto, *Tetrahedron*, 1994, **50**, 6017; (e) A. S. E. Karlström and J.-E. Bäckvall, *Chem.-Eur. J.*, 2001, **7**, 1981; Ni-catalyzed allylation: (f) Y. Sumida, S. Hayashi, K. Hirano, H. Yorimitsu and K. Oshima, *Org. Lett.*, 2008, **10**, 1629; (g) R. Matsubara and T. F. Jamison, *J. Am. Chem. Soc.*, 2010, **132**, 6880; (h) A. Jiménez-Aquino, E. F. Flegeau, U. Schneider and S. Kobayashi, *Chem. Commun.*, 2011, **47**, 9456.
- 4 (a) P. Zhang, L. A. Brozek and J. P. Morken, *J. Am. Chem. Soc.*, 2010, **132**, 10686; (b) P. Zhang, H. Le, R. E. Kyne and J. P. Morken, *J. Am. Chem. Soc.*, 2011, **133**, 9716; (c) L. A. Brozek, M. J. Ardolino and J. P. Morken, *J. Am. Chem. Soc.*, 2011, **133**, 16778; (d) H. Le, R. E. Kyne, L. A. Brozek and J. P. Morken, *Org. Lett.*, 2013, **15**, 1432; (e) H. Le, A. Batten and J. P. Morken, *Org. Lett.*, 2014, **16**, 2096; (f) M. J. Arolino and J. P. Morken, *J. Am. Chem. Soc.*, 2014, **136**, 7092.
 - 5 J. Y. Hamilton, N. Hauser, D. Sarlah and E. M. Carreira, *Angew. Chem., Int. Ed.*, 2014, **53**, 10759.
 - 6 (a) V. Hornillos, M. Pérez, M. Fañanás-Mastral and B. L. Feringa, *J. Am. Chem. Soc.*, 2013, **135**, 2140; (b) Y. Yasuda, H. Ohmiya and M. Sawamura, *Angew. Chem., Int. Ed.*, 2016, **55**, 10816.
 - 7 F. Meng, K. McGrath and A. H. Hoveyda, *Nature*, 2014, **513**, 367.
 - 8 K. W. Quasdorf and L. E. Overman, *Nature*, 2014, **516**, 181.
 - 9 For selected reviews, see: (a) H.-Y. Jang and M. J. Krische, *Acc. Chem. Res.*, 2004, **37**, 653; (b) H.-Y. Jang and M. J. Krische, *Eur. J. Org. Chem.*, 2004, 3953; (c) H. Nishiyama and T. Shiomi, *Top. Curr. Chem.*, 2007, **279**, 105; (d) J. F. Bower, I. S. Kim, R. L. Patman and M. J. Krische, *Angew. Chem., Int. Ed.*, 2009, **48**, 34; (e) J. M. Ketcham, I. Shin, T. P. Montgomery and M. J. Krische, *Angew. Chem., Int. Ed.*, 2014, **53**, 9142; (f) S. K. Murphy and V. M. Dong, *Chem. Commun.*, 2014, **50**, 13645; (g) K. D. Nguyen, B. Y. Park, T. Luong, H. Sato, V. J. Garza and M. J. Krische, *Science*, 2016, **354**, aah5133; (h) M. T. Pirnot, Y.-M. Wang and S. L. Buchwald, *Angew. Chem., Int. Ed.*, 2016, **55**, 48.
 - 10 Y.-M. Wang and S. L. Buchwald, *J. Am. Chem. Soc.*, 2016, **138**, 5024.
 - 11 J. T. Han, W. J. Jang, N. Kim and J. Yun, *J. Am. Chem. Soc.*, 2016, **138**, 15146.
 - 12 J. Lee, S. Torker and A. H. Hoveyda, *Angew. Chem., Int. Ed.*, 2017, **56**, 821.
 - 13 G. Xu, H. Zhao, B. Fu, A. Cang, G. Zhang, Q. Zhang, T. Xiong and Q. Zhang, *Angew. Chem., Int. Ed.*, 2017, **56**, 13130.
 - 14 For a review on allenes as the allylic nucleophiles in copper catalysis, see: (a) A. P. Pulis, K. Yeung and D. J. Procter, *Chem. Sci.*, 2017, **8**, 5240; For recent examples of CuH catalysis, see: (b) Y. Tani, K. Kuga, T. Fujihara, J. Terao and Y. Tsuji, *Chem. Commun.*, 2015, **51**, 13020; (c) R. Y. Liu, Y. Yang and S. L. Buchwald, *Angew. Chem., Int. Ed.*, 2016, **55**, 14077; (d) M. Lee, M. Nguyen, C. Brandt, W. Kaminsky and G. Lalic, *Angew. Chem., Int. Ed.*, 2017, **56**, 15703; (e) E. Y. Tsai, R. Y. Liu, Y. Yang and S. L. Buchwald, *J. Am. Chem. Soc.*, 2018, **140**, 2007.
 - 15 T. N. Nguyen, N. O. Thiel, F. Pape and J. F. Teichert, *Org. Lett.*, 2016, **18**, 2455.
 - 16 More recently, the racemic version was reported by Tsuji and co-workers, see: T. Fujihara, K. Yokota, J. Terao and Y. Tsuji, *Chem. Commun.*, 2017, **53**, 7898.
 - 17 For selected recent examples, see: (a) R. Shintani, K. Takatsu, M. Takeda and T. Hayashi, *Angew. Chem., Int. Ed.*, 2011, **50**, 8656; (b) A. Harada, Y. Makida, T. Sato, H. Ohmiya and M. Sawamura, *J. Am. Chem. Soc.*, 2014, **136**, 13932; (c) K. Akiyama, F. Gao and A. H. Hoveyda, *Angew. Chem., Int. Ed.*, 2010, **49**, 419; (d) Y. Shi, B. Jung, S. Torker and A. H. Hoveyda, *J. Am. Chem. Soc.*, 2015, **137**, 8948; (e) Z.-Q. Zhang, B. Zhang, X. Lu, J.-H. Liu, X.-Y. Lu, B. Xiao and Y. Fu, *Org. Lett.*, 2016, **18**, 952; (f) S. Guduguntla, J.-B. Gualtierotti, S. S. Goh and B. L. Feringa, *ACS Catal.*, 2016, **6**, 6591.
 - 18 J. Lee, S. Radomkit, S. Torker, J. del Pozo and A. H. Hoveyda, *Nat. Chem.*, 2018, **10**, 99.
 - 19 (a) P. G. Cozzi, R. Hilgraf and N. Zimmermann, *Eur. J. Org. Chem.*, 2007, **36**, 5869; (b) B. M. Trost and C. Jiang, *Synthesis*, 2006, 369; (c) J. P. Das and I. Marek, *Chem. Commun.*, 2011, **47**, 4593; (d) K. W. Quasdorf and L. E. Overman, *Nature*, 2014, **516**, 181; (e) J. Feng, M. Holmes and M. J. Krische, *Chem. Rev.*, 2017, **117**, 12564.
 - 20 For selected recent reviews on Cu-mediated allylic substitutions, see: (a) A. H. Hoveyda, A. W. Hird and M. A. Kacprzynski, *Chem. Commun.*, 2004, 1779; (b) H. Yorimitsu and K. Oshima, *Angew. Chem., Int. Ed.*, 2005, **44**, 4435; (c) A. Alexakis, J. E. Bäckvall, N. Krause, O. Pàmies and M. Diéguez, *Chem. Rev.*, 2008, **108**, 2796; (d) S. R. Harutyunyan, den T. Hartog, K. Geurts, A. J. Minnaard and B. L. Feringa, *Chem. Rev.*, 2008, **108**, 2824; (e) A. H. Cherney, N. T. Kadunce and S. E. Reisman, *Chem. Rev.*, 2015, **115**, 9587; (f) R. Shintani, *Synthesis*, 2016, **48**, 1087.

