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High throughput analysis enables high throughput experimentation in pharmaceutical process research

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High throughput experimentation has become widely used in the discovery and development of new medicines. Progress in the field relies on an interplay between high throughput experimentation and high throughput analysis technologies. In this perspective, a personal account of the evolution of high throughput analysis and the role that measurement science has played in enabling the success of high throughput experimentation is presented, along with an assessment of current challenges and future prospects.

Introduction

Measurement tools have played a key role in the field of synthetic organic chemistry, from early applications of UV and IR spectroscopy to the rise of NMR and the various chromatographic and mass spectrometry techniques that underpin contemporary chemical research. Measurement science has also played a central role in enabling the rise of high throughput experimentation, which has become an important feature of modern pharmaceutical discovery and development.^{1–5} High throughput experimentation is now a large, complex and rapidly evolving branch of chemistry and an important focus for academic research.^{6–14} In addition, high throughput experimentation is now routinely employed by a number of other industries beyond pharma.^{15–18} As a complete review of this vast subject is beyond the scope of this perspective article, I will instead provide a personal account of my own experiences with the development of high throughput analysis based largely on research carried out during my career in pharmaceutical process research and development.

Teamwork is important in many areas of science, but especially so in the pharmaceutical industry, where virtually all inventions and discoveries are made jointly by teams. I am grateful to have been a member of some very exciting and interesting teams during my time in pharma, and to have had the good fortune of leading some teams of hardworking and extremely capable researchers. All of the discoveries and developments chronicled here are the result of teamwork, and I am forever grateful to my many collaborators and co-workers for making these discoveries possible.

Pharmaceutical process research at the dawn of the new millennium

When I began at Merck & Co., Inc. in the late 1990s, the company was undergoing a shift in research operating model, largely brought about by the lessons learned from the commercialization of the anti-HIV drug, Crixivan (indinavir; Fig. 1). The process chemistry group had historically attacked each new clinical candidate with characteristic competence and completeness, developing a scalable and economically viable synthesis even for those candidates that later went on to fail in early clinical testing. Crixivan tested the limits of this development strategy in three significant ways: first, the timeline for development and commercialization of the drug was escalated to warp speed, owing to the growing HIV epidemic and the lack of suitable therapeutics to address this unmet need.¹⁹ Second, owing to a high dose of nearly a gram per patient per day, the sheer volume of required material seriously taxed available resources. Third, Crixivan was a very complex molecule, especially from a stereochemical perspective. Not only did the molecule contain 5 stereogenic centers, but the relative isolation of these centers meant that many traditional strategies for diastereocontrol were of limited value. Our company was well versed in dealing with complex synthetic problems, with significant expertise in the development of enantiopure drugs, as evidenced by the remarkable manufacturing route to enantiopure Aldomet developed in the early 1960's.^{20–22} However, the scale, stereochemical challenges and accelerated development pace of the Crixivan project was a significant challenge to the chemists tasked with accessing numerous potential intermediates in enantiopure form, and endeavoring to assemble these fragments according to several different synthetic schemes.²³ Influenced by these events, we launched an initiative to explore the emerging field of laboratory automation and robotics to speed and streamline the development of these challenging, complex drug candidates.

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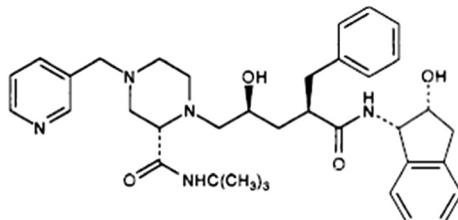


Fig. 1 Increasing chemical and stereochemical complexity of new drugs, such as the HIV protease inhibitor, Crixivan, precipitated changes in the strategy for development of new pharmaceutical candidates.

I joined the Process Research department as part of a cross-departmental effort termed the selective synthesis venture, with the mission to explore the use of robotics and laboratory automation to speed and facilitate process research investigations into the preparation of enantiopure pharmaceuticals and synthetic intermediates. Prior to graduate school I had worked at Velsicol Chemical Company on the discovery synthesis of new herbicides, which taught me the value of optimized workflows and rapid compound purification. I also developed new reagents for enzyme immunoassays at the Diagnostics Division of Abbott Laboratories, which familiarized me with liquid handlers, microplate experimentation and high throughput spectroscopic assays. My research at Abbott also taught me the value of convenient, prepackaged reagent kits and the importance of working closely with grass roots researchers to drive the implementation and adoption of new research technologies. After receiving my Ph.D., I joined Regis Technologies, a small firm in the Chicago area, in order to bring chromatography technologies invented during graduate school to the marketplace. As Director of Research we commercialized new chromatography stationary phases and enantioselective catalysts, created a laboratory for custom GMP preparative HPLC purification, developed high throughput experimentation techniques for multiparallel synthesis and evaluation of new stationary phases and helped launch GMP contract pharmaceutical synthesis capabilities for the company, all experiences that would prove useful in the pharmaceutical arena.

I learned many important lessons upon becoming immersed in pharmaceutical process research. I was initially surprised by the urgency of problem solving, with new synthetic routes being considered and abandoned at breakneck speed. There was a narrow window of opportunity in which a solution to a problem could be provided, with elegant solutions offered after the window had closed being judged nice, but of little value.

Access to instruments

The difficulty in gaining access to analytical instrumentation was a perennial problem during my foray in small chemical business. I recall spending an enormous amount of time and energy acquiring an NMR spectrometer with somewhat marginal performance. Meanwhile, it was becoming very clear

that mass spectrometers were going to be critically important, and these instruments were well beyond the price range of our small-scale operations. After moving to the pharmaceutical industry, I had the fortune to join a team which realized the need to invest in state-of-the-art tools, and was given the opportunity to lead an effort that coordinated the identification, acquisition and evaluation of new enabling technologies across the entire company. In this capacity, we carried out more than 300 new technology evaluations over 15 years, ranging from new analytical instrumentation to larval zebrafish assay platforms, next generation sequencers and specialized computer software and databases. We put together a highly successful company-wide annual new technologies symposium and created a mechanism for carrying out funded collaborations with academia, a successful program that resulted in many important research collaborations.²⁴

Building the new paradigm for high throughput process research

At the outset, the implementation plan for the selective synthesis venture was to simply evaluate automation and robotics equipment and see what works. Early visions of how automation and robotics would impact process research imagined individual process chemists integrating these tools into everyday research workflows. Not surprisingly, uptake was limited, due in part to challenges with complex and difficult to operate equipment and in part to unfamiliarity of process chemists with this new technology.

My previous experience convinced me that small, specialized groups focusing on particular problem areas were the right initial strategy for bringing high throughput experimentation to process chemists, and we set about creating capability clusters to provide solutions in several different areas. We evolved a general approach employing platform solutions based on ready to go kits of reagents that could be used to rapidly solve frequently encountered classes of problems (Fig. 2). We also found it important to employ a 'sales force' of workers who could gain timely access to information about emerging problems in development projects, and who could work with project chemists on the implementation of solutions. Providing project chemists with timely solutions helped us to develop a reputation for taking care of difficult problems quickly while allowing our groups to develop a culture of relentlessly attacking problems to provide rapid solutions. To do this well required that experimental workflows be streamlined and optimized, including not just experimentation and analytical measurement but also the analysis, interpretation and reporting of experimental data.

Chiral chromatography

Today, high throughput analysis in the pharmaceutical industry is closely associated with catalyst and enzyme screening for synthetic chemistry,²⁵ but this was actually one of the last elements of the high throughput experimentation capability build to come into place. Since a major focus of our effort was



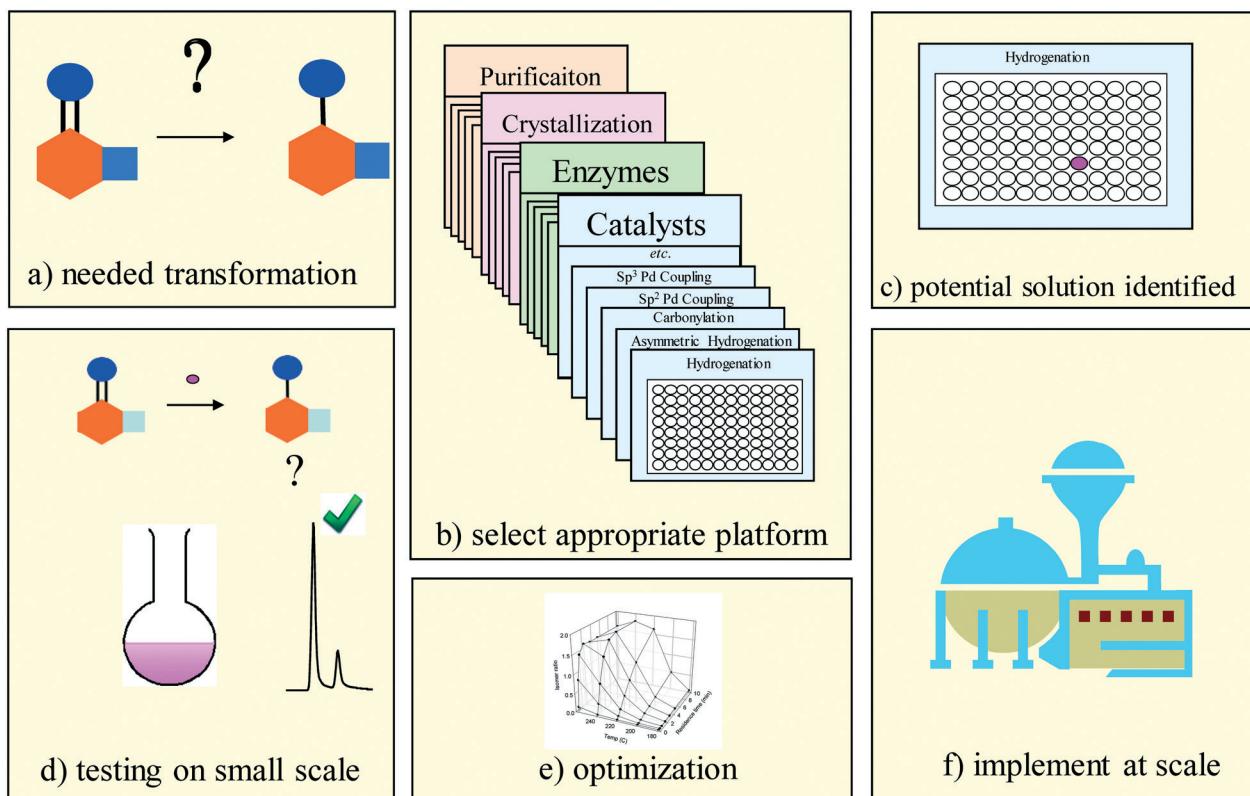


Fig. 2 Use of high throughput experimentation facilitates modern pharmaceutical process research. Needed transformations (a) are matched against available platform tools (b) to identify a potential solution (c) which is then tested on small scale (d) before being optimized (e) and implemented at scale (f).

developing rapid solutions to problems involving chirality and stereochemistry, one of the first platforms created was a rapid analytical method development capability for the chromatographic separation of enantiomers using SFC, which afforded researchers a 'same day' or at worst 'next day' ability to measure the enantiopurity of newly synthesized compounds. This platform has evolved both in terms of speed and universality over the ensuing years,^{26,27} but still provides an important foundation for many other experimental platforms (Fig. 3).

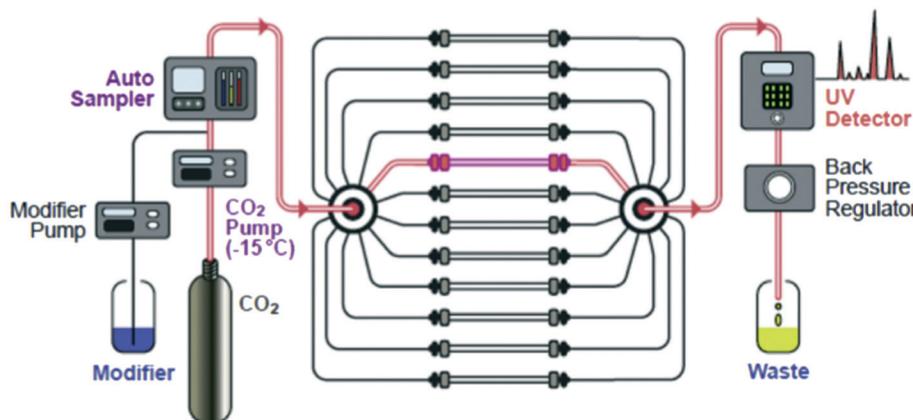
Crystallization

We next leveraged our ability to rapidly measure enantiopurity for unknown compounds to construct a microplate-based screening platform for classical resolution *via* diastereomeric crystallization. We scoured the compendia of literature reports, obtained valuable advice from Andre Collet at University of Lyon and other noted experts, then created freezer-stored microplate screening kits with pre-dispensed solutions of chiral acid and base resolving agents. Again, these kits allowed same day or next day identification of conditions for carrying out classical resolutions, enabling near real time solutions to emerging development problems. The platform was soon extended to screening kits for final salt form, using the counterions commonly in use for pharmaceuticals (Fig. 4). The technique was successfully used in many

development projects,²⁸ ultimately becoming incorporated into an automated salt form screening robot co-developed with Symyx.²⁹ In addition to rapid chromatographic screening tools for analyzing the composition of crystals and mother liquors, we acquired and utilized other tools for high throughput analysis, including microplate based techniques for X-ray powder diffraction, differential scanning calorimetry and kinetic profiling *via* microplate nephelometry. Subsequently, a host of analytical tools involving various forms of microscopy, spectroscopy and thermal analysis have been incorporated into these microscale crystallization screening platforms, which have become the standard approach for research on new solid forms in the pharmaceutical industry.³⁰

Preparative chromatography

During this time, we developed a variety of different platforms for preparative chromatographic purification of pharmaceutical intermediates and final products.^{31–33} Leveraging my experience at Regis enabled us to rapidly transplant successful strategies and workflows to the pharmaceutical development environment. Initially, we focused on the small-scale purifications that can powerfully enable synthetic route investigations, especially the chromatographic separation of enantiomers. Again, our analytical chromatography method development platform provided a convenient entrée, allowing



Conditions	1990s	→ 2005	→ Current
Column size (mm)	250 x 4.6	250 x 4.6	150 x 4.6
Particle size (μm)	10	5	3
Flow rate (mL/min)	1.5	2	3
Starting % MeOH	4	4	1
Approx. run time (min)	30	16	6

Fig. 3 Method development platform for analytical chiral SFC and how it has evolved over time.

us to develop methods and carry out enantioseparations on gram scale with only 1–2 days turnaround. Key to our success in this effort was our collaboration with Terry Berger and Berger Instruments on developing and beta testing the first commercial semi-preparative SFC instrument introduced to the marketplace. This close working relationship with cutting edge instrument developers became a key component of our strategy for success, allowing us to shape and guide commercial product development in ways that maximized research productivity. The net result of the introduction of the prep chiral SFC resolution platform was that various strategies for enantioselective synthesis,³⁴ diastereocontrol, kinetic resolution³⁵ or checking propensity for racemization^{36–39} could be quickly explored by project chemists, allowing competing synthetic routes to be rapidly evaluated and differentiated. This strategy also proved useful for researchers in drug metabolism,⁴⁰ and especially for medicinal chemists, where the work on a candidate for treatment of Alzheimer's disease nicely illustrates the value of the approach.⁴¹

Preparative chromatographic enantioseparation helps to speed and simplify synthetic route investigation, especially in cases where traditional synthetic methods offer few clear avenues to access enantiopure reaction intermediates. In the gamma secretase medicinal chemistry program, enantioselective synthetic routes to these subtly chiral spiro compounds, illustrated in Fig. 5, were not readily apparent. Despite their 'near meso' character, which renders

enantioselective synthesis of these compounds challenging, gram scale chromatographic enantioseparations on chiral stationary phases could be readily accomplished in a few hours. The investigation of more than 65 different structural analogs was facilitated by such milligram to gram scale resolutions, greatly streamlining medicinal chemistry lead optimization and candidate selection. In addition, chromatographic resolution on hundreds of grams scale was used to support this program during early clinical development.

Key to the success of this new prep SFC purification capability was the ability of the instruments to run unattended using automated sample injection and fraction collection. Indeed, we soon found that our instruments, which were acquired for resolving a few grams at a time, were being increasingly used in prolonged, round-the-clock, multi-day campaigns at the hundreds of grams scale. In the race against time to prepare drug candidates for preclinical testing, knowing that chromatographic resolution works at small scale can provide a workable alternative for accessing enantiopure material at larger scale. I was experienced with carrying out larger scale chromatographic resolutions, and familiar with the use of industrial scale simulated moving bed chromatography (SMB) for manufacturing scale separations, however, process chemists were initially reluctant to consider the use of chromatography in a kilogram scale delivery, let alone a commercial process. To persuade skeptics, we shared the example of how R.B. Woodward, one of the legends of 20th century synthetic chemistry, used the

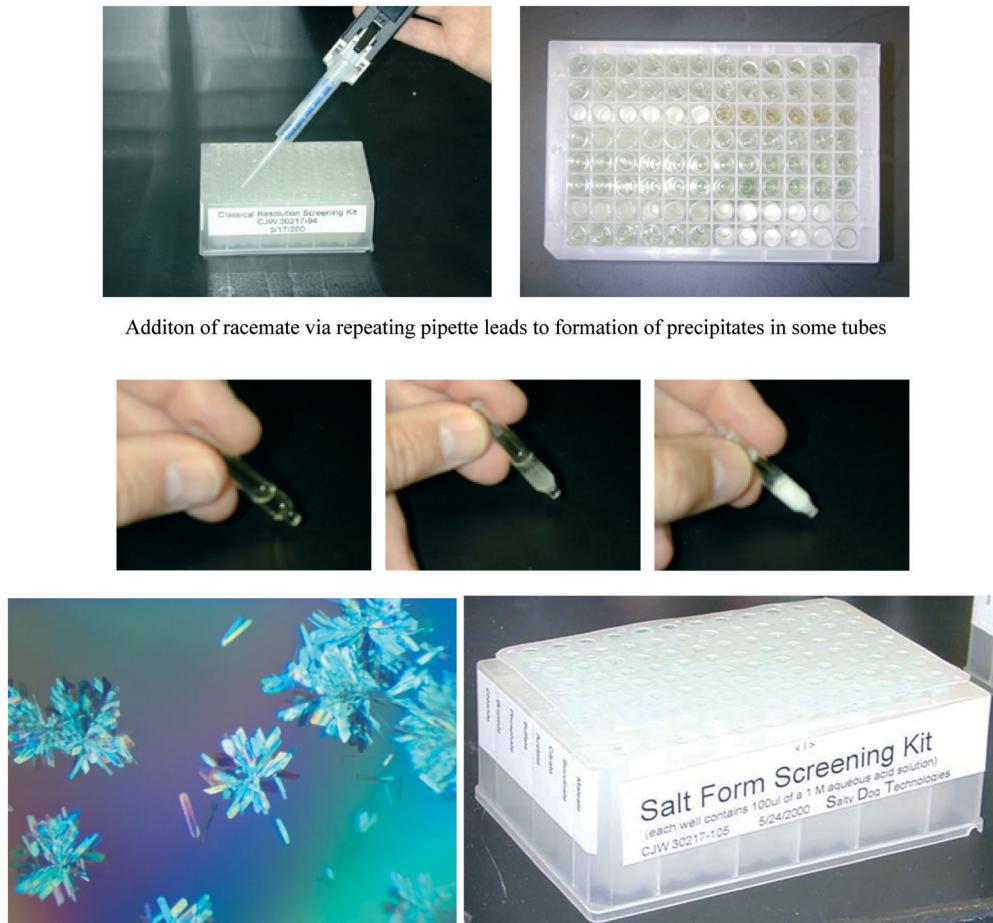


Fig. 4 Microscale screening kits for investigating crystallizations (classical resolution and pharmaceutical salt form) were early high throughput experimentation platforms that enabled rapid problem solving in pharmaceutical process research.

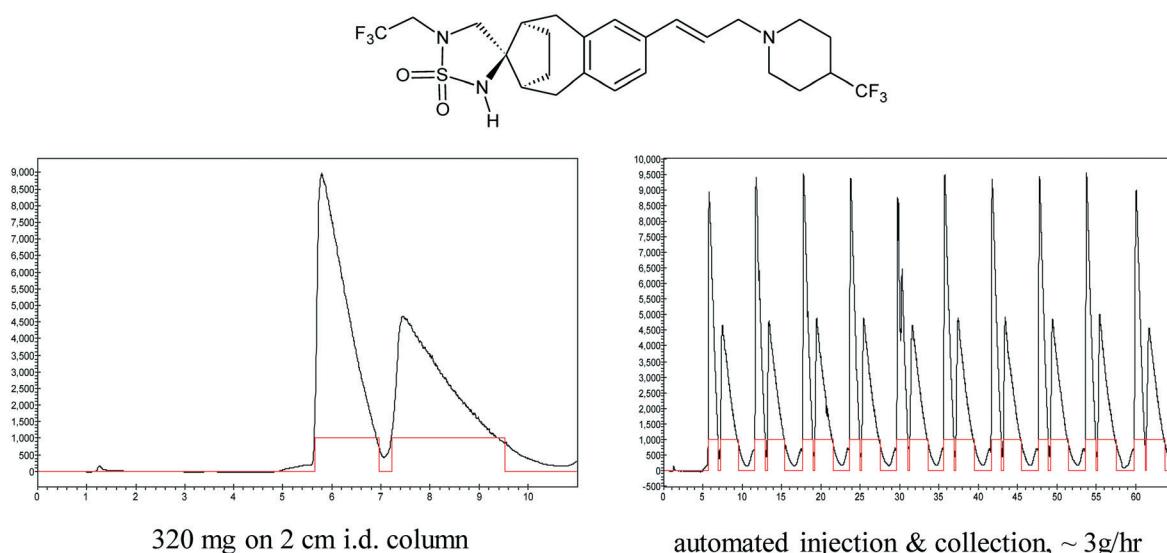


Fig. 5 One of more than 65 different racemates that were resolved to support early lead optimization for a γ secretase program involving 'cryptochiral' structures not easily accessed via enantioselective synthesis.

emerging capability of preparative HPLC to separate isomeric corrins in his vitamin B12 synthesis, leading him to observe

that, "The power of these high pressure liquid chromatographic methods hardly can be imaged by the chemist who has not had

experience with them; they represent relatively simple instrumentation and I am certain that they will be indispensable in the laboratory of every organic chemist in the near future".⁴² Gradually, we made inroads in the use of preparative chromatography in the syntheses of new clinical candidates, carrying out multikilogram separations using not only HPLC,^{43–46} but also novel large scale preparative SFC equipment that we developed in partnership with Thar Technologies and Novasep.⁴⁷

Our analytical HPLC and SFC method development workflows enabled the rapid development of methods for preparative separations of newly prepared compounds, however, analytical chromatography experiments alone are insufficient for predicting preparative chromatographic performance at scale. Loading studies, in which increasing amounts of the mixture of interest are injected onto a column, are needed to estimate time, solvent and material needs for larger scale separations. While such studies require less than a gram of material under normal conditions, we found that at the early stages of synthetic route investigations, this amount of sample was often unavailable. In response, we developed a platform approach for carrying out loading studies using very small capillary columns, allowing accurate prediction of preparative performance with only a milligram or two of sample.⁴⁸ These microscale loadings studies allowed accurate estimates to be made upon scaling up by as much as millionfold, as depicted in Fig. 6.

Batch adsorbents and reactive resins

At the same time that these early preparative chromatographic purification platforms were being assembled, we were also creating platforms for rapid screening of batch adsorbents. Process chemists have long used batch adsorbents such as activated charcoal to remove process impurities, but developing treatment methods often involved slow, iterative empirical testing. At Regis, we had invented a technique for microplate synthesis and screening of new chiral stationary phases that relied on HPLC assessment of the ratios of

adsorbed and free analytes in the supernatant fluid in contact with a few milligrams of stationary phase (Fig. 7).^{49–52} The technique worked well, but had to be performed under highly dilute conditions in order to avoid saturation of the binding sites on the stationary phase particles, which often brought us to the limits of UV detection. In the pharmaceutical development environment, we found that the outstanding sensitivity of MS detection afforded a much greater dynamic range. We also discovered that enantioseparations could be studied using an isotopically labeled 'pseudoenantiomer' strategy⁵³ similar to approaches used by Reetz,⁵⁴ Sawada⁵⁵ and others.

We developed microplate screening kits that facilitated the rapid selection of appropriate adsorbents and optimization of treatment protocols for all manner of impurities in drug substances and intermediates,⁵⁶ initially applying this approach to the removal of discreet, structurally defined small molecule impurities contaminating our process streams, using HPLC, SFC or MS assays to measure the relative ratios of desired product and contaminating impurity. We soon found that analysis using UV-vis plate readers also allowed us to track the removal of colored impurities, even when the exact structure of the impurity was uncertain or polymeric in nature.⁵⁷ By expanding the screening kits to include reactive resins, we were able to develop methods for highly selective removal of reactive impurities, allowing, for example, the effective purging of residual aldehyde starting material from a Wittig-type reaction stream using only 20 g of hydrazine resin per kilogram of desired product.^{58,59} As regulatory rules changed to mandate ultralow levels of inherently reactive potential mutagenic impurities, or PMIs, this purification platform became increasingly important.⁶⁰

The application of adsorbent screening to the removal of metals from pharmaceutical candidates and commercial products became perhaps the most important application for this platform technology.⁶¹ As transition metal catalysts began to be more widely used in pharmaceutical development and manufacturing, we found that the problem of removing residual metal contaminants became more of a concern.

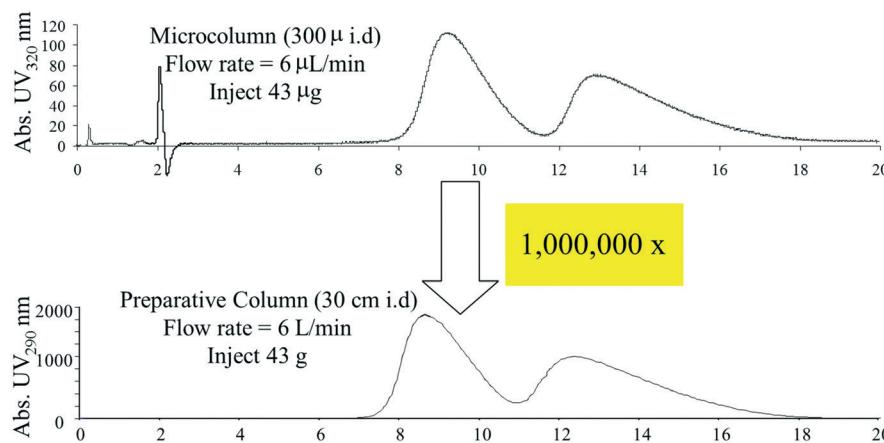


Fig. 6 Loading studies carried out using microcapillary columns can be used to accurately predict preparative chromatographic performance at large scale.



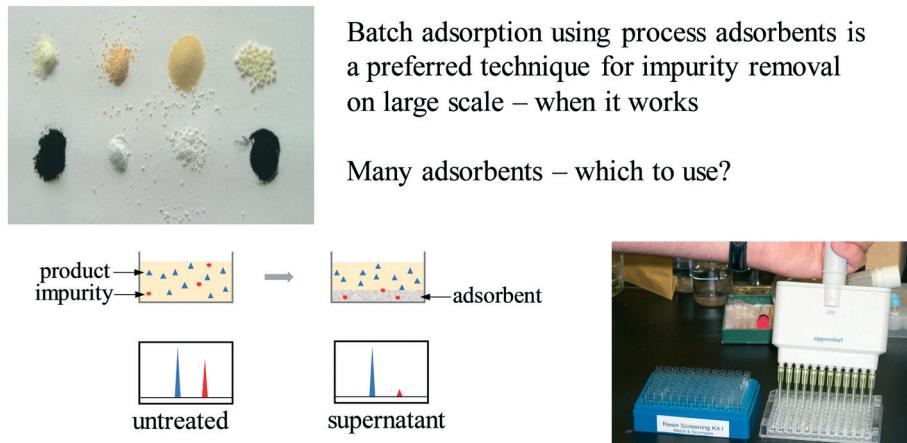


Fig. 7 Analysis of the supernatant solution of an equilibrium partition experiment enables high throughput screening of process adsorbents.

While simple approaches such as recrystallization or liquid–liquid extraction can sometimes be effective for removing metal impurities, many drug candidates are peppered with basic nitrogens and other binding sites, rendering them excellent metal binders. Thus, removal of metal impurities from these compounds often turned into a sort of ‘tug-of-war’. By expanding our adsorbent screening kits to include specific materials for metal remediation, we found that we could often identify highly selective and effective treatments to purge metal impurities from pharmaceutical APIs or advanced intermediates. A notable landmark was a treatment for removal of residual rhodium that was implemented in the commercial manufacturing process for the diabetes drug, Januvia.⁶² Not only did this treatment reduce rhodium to acceptable levels, but it also concentrated the valuable material in a carbon-based adsorbent that could be burned for metal recovery. We partnered with ES Industries in the commercialization of screening kits and developed a variety of improved and cost-effective treatment protocols for metal removal.⁶³ Throughout this research, inductively coupled plasma mass spectrometry (ICP-MS) provided a workhorse technique for metals analysis. We developed flow injection analysis (FIA) techniques that allowed rapid analysis of the samples coming from microplate screening,⁶⁴ and HPLC-ICP-MS and other techniques that enabled study of discrete organometallic species and kinetic profiling of organometallic reactions.^{65–67} While effective, these analytical techniques were very expensive and somewhat cumbersome. We worked closely with Professor Kaz Koide at University of Pittsburgh to develop a series of highly sensitive catalysis-based fluorogenic⁶⁸ and chromogenic^{69,70} reagents and assay protocols for ‘point of use’ assessment of the concentration of palladium, our most frequently encountered metal impurity.

High throughput analysis for high throughput reaction screening

Most of the high throughput analysis methods described thus far required resolution of only two components – either prod-

uct and impurity or a mixture of two enantiomers. As we began to develop analysis methods for supporting high throughput experimentation in the screening of catalysts and reaction conditions, we faced additional analytical challenges. The presence of various ligands, bases, additives and solvents in these samples can lead to interference in mass spectrometry-based methods for assessing the relative amounts of starting material, product and impurities. Consequently, caution must be exercised to ensure either that such interference is not taking place or that potential interfering compounds are chromatographically resolved from the components of interest. We found that fast chromatographic methods coupled with mass spectrometry detection were often the best choices for such situations, although our definition of ‘fast chromatography’ changed dramatically over the course of this research.⁷¹

Developing high throughput analytical methods for enzymatic transformations is generally easier than for more complex catalytic reactions containing a variety of ligands, additives and solvents. We found chromatography combined with MS detection to be a powerful tool for screening enzymatic reactions. A very early example involving high throughput screening of enantioselective yeast catalyzed ketone reductions is illustrated in Fig. 8, where chromatographic resolution of the enantiomers of the product alcohol was relatively straightforward, but complicated by peak overlap from the strongly UV-absorbing starting ketone, requiring analysis times of more than 45 minutes for complete resolution. We found that by using mass specific detection we could deconvolute the signal of the lower molecular weight ketone from that of the product alcohol, allowing a shortened separation time of only 4 minutes to be used, while avoiding problems with signal suppression.⁷²

Avoiding the analysis bottleneck in high throughput experimentation: parallel HPLC

While an analysis time of 4 minutes per sample was suitable when only a few dozen enzyme-producing yeast clones were being screened, we were soon routinely screening entire 96



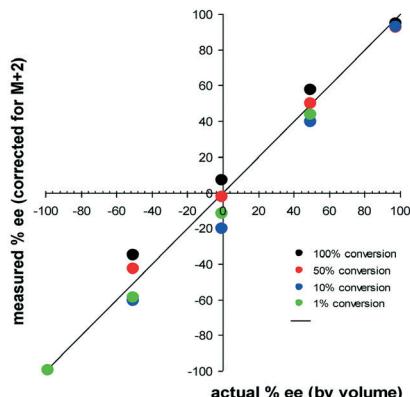
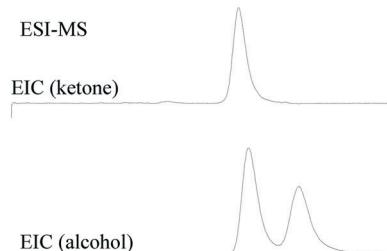
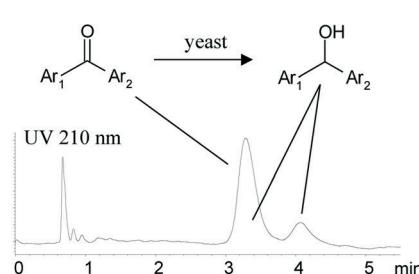


Fig. 8 Early use of MS detection to deconvolute overlapping peaks. Extracted ion chromatograms for species of different molecular weight (ketone and alcohol) allow rapid quantitation acceptable for reaction screening.

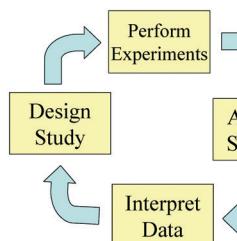
well plates of variants, with similar increases in our high throughput catalysis screening workflows as well. Before long, we were examining multiple 96 well microplates for each catalysis screening problem, with the result that slower analysis methods began to cause significant delays in research activities. Fig. 9 illustrates how analysis cycle time can become a bottleneck for high throughput experimentation. Even fast methods (*e.g.* 1 min cycle time) can result in significant delays when large numbers of samples must be analyzed. It is important to note that as the number of samples increases, the logistics of manually combing through the data to compile and compare results can also become quite onerous, leading many researchers to become frustrated with larger scale experimental designs. In the early days, we used a home-built Microsoft Excel tool for data consolidation prepared by enzyme chemist, Jeff Moore. Subsequently, the use of a data amalgamation and visualization tool from Virscidian has become popular.

There are fundamentally two different strategies for dealing with this bottleneck for high throughput analysis, either dramatically reduce the time for each individual assay or carry out multiple assays in parallel. Attempts at speeding analysis were occasionally successful, but not routinely reliable at the outset. Consequently, we made significant investments in acquiring multiparallel analysis capabilities, with a notable collaboration with Eksigent on the development of a specialized analytical instrument dedicated to this task.^{73,74} The resulting 8 channel multiparallel microfluidic HPLC was

a marvel of instrument engineering, with outstanding performance characteristics that allowed us to routinely deliver a plate time (time for analysis of an entire 96 well microplate) of only half an hour (Fig. 10). In addition, the eight independent channels of the instrument could also be used for multiparallel method development screening, with up to 8 different column-mobile phase combinations being investigated simultaneously.^{75,76} Multiparallel analysis using capillary electrophoresis⁷⁷ and SFC were also used to a lesser extent. It should be noted that a simpler approach to parallel analysis is possible and often utilized, namely splitting a single microplate between two or more instruments. Nevertheless, this strategy requires subdivision of the parent plate and coordinated re-integration of the data once measurements are completed, a tedious and error prone process. In contrast, the Eksigent instrument injects samples from a single plate, with data archiving and display *via* convenient plate view software.

MISER chromatographic analysis

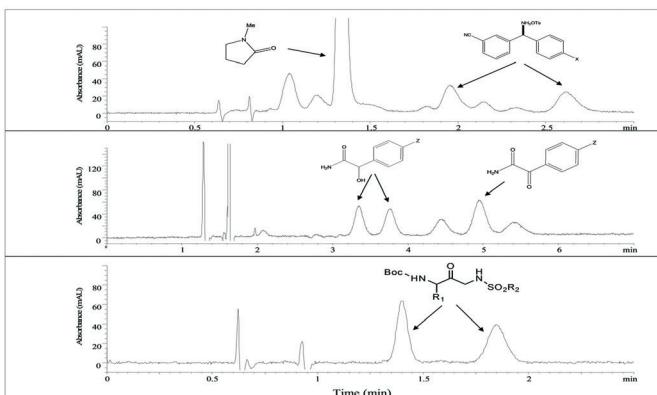
As high throughput experimentation continued to be more widely used, it became inconvenient for all analysis to take place on one or two centralized multiparallel chromatography instruments. In addition, these instruments were expensive, somewhat temperamental and unfamiliar to process chemists. On the other hand, process chemists were by this time becoming quite adept at the use of HPLC and HPLC-MS instrumentation. We found that the use of MISER chromatography (multiple



	Total Analysis Time		
	96 wells	384 wells	1536 wells
Cycle Time	30 min	48 hours	1 month
	1 min	1.6 hours	26 hours

Fig. 9 Avoiding the analysis bottleneck is critical for success in high throughput experimentation.





- Designed for high-throughput analysis
- 8 columns (0.3 mm i.d.) 16 pumps
- low dwell volume, ultrafast gradients
- ~ 0.5 h for analysis of 96 well plate

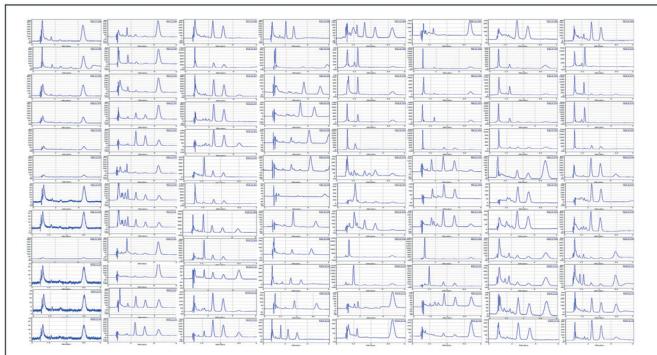
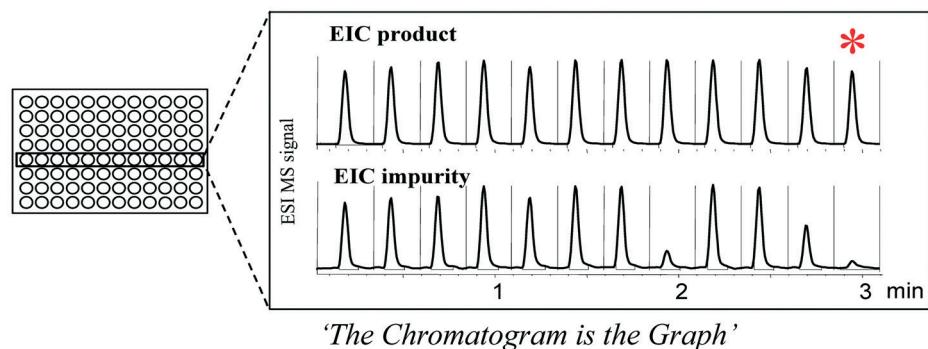


Fig. 10 Multiparallel analysis using the 8 channel Eksigent Express 800 microfluidic HPLC enabled routine high throughput analysis to support high throughput pharmaceutical process research studies.

injections in a single experimental run)⁷⁸ is well suited for high throughput analysis studies using conventional HPLC-MS instrumentation, allowing versatility in method development and high throughput screening as well as a simplified manner of presenting experimental results that bypasses many of the ob-

stacles associated with data collation, graphing and interpretation (Fig. 11). This technique is similar to flow injection analysis, but uses just enough chromatographic retention of the compounds of interest to allow separation from potentially interfering matrix components. Sequential injections of

Short column, fast injections
Multiple Injections in a Single Experimental Run
(MISER)



Small degree of chromatographic retention to avoid co-elution with interfering compounds

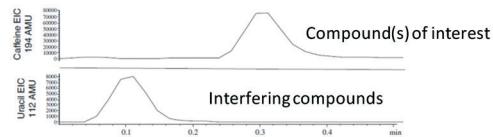


Fig. 11 MISER analysis combines speed of analysis with ease of interpretation (EIC = extracted ion chromatogram).

different samples with ongoing data collection leads to a MISERgram with a graph-like quality that facilitates data interpretation in high throughput screening. While initially used for adsorbent screening studies and studies of enzymatic reactions, we soon found that MISER analysis with MS detection often afforded excellent results in high throughput screening of catalysts and reaction conditions as well,^{79,80} affording typical cycle times for analysis of a 96 well microplate of 30 minutes or less. Using a mass multiplexing approach, where samples of different masses from different wells were combined, allowed us to analyze 1536 different reactions in only 2.4 hours by MISER HPLC-MS.⁸¹ Variants of the technique using GC⁸² or SFC⁸³ analysis were developed, allowing for rapid analysis of volatile samples and high throughput determinations of enantiopurity. We found that the technique was readily grasped by students and educators in studies investigating the high throughput analysis of chili peppers,⁸⁴ soft drinks⁸⁵ and beer.⁸⁶

Ultrafast chromatographic analysis

MISER chromatography grew out of a need to deliver the fastest possible cycle time allowed by an autosampler (~22 s in the early days, ~10 s currently) with the use of MS detection to deconvolute information that was otherwise obscured by insufficient chromatographic resolution. A series of innovations in chromatographic stationary phase particle design and column technology over the last decade has led to remarkable increases in column efficiency, now making possible the complete resolution of species that were previously only partially resolved.⁸⁷⁻⁸⁹ The corresponding increase in analysis speed stemming from this advance means that many sample mixtures are now often completely resolved within the timespan of the fastest possible injection cycle time permitted by commercial autosamplers.⁹⁰ For instance, the chromatographic separation of enantiomers can now be accomplished for almost all compounds in less than a minute, with many separations taking just a few seconds.^{91,92} Consequently, direct ultrafast chromatographic resolution is now becoming increasingly important for high throughput analysis. Interestingly, fast MISER LCMS using these approaches can also be used for kinetic profiling of organic reactions⁹³ or for separations in the second dimension of two-dimensional chromatography.⁹⁴

Analysis of closely related species

High throughput analysis is especially difficult when discrimination between very similar species is required. While the chromatographic separation of mirror image enantiomers was once considered quite challenging, recent developments in chromatographic columns and instrumentation are now making these separations increasingly routine for many analytes. Nevertheless, certain 'cryptochiral' compounds are extremely challenging to resolve using chromatography,⁹⁵ owing to near meso structures (Fig. 12). We recently reported a high throughput screening workflow that expedites the selection of chiral solvating agents for the NMR analysis of

enantiopurity,⁹⁶ which is especially useful for these challenging compounds.

Additional separation challenges for closely related compounds are illustrated in Fig. 13. Complex mixtures of diastereomers^{97,98} and regiosomeric species resulting from non-specific C–H activation chemistry⁹⁹ or drug metabolism¹⁰⁰ are increasingly important in pharmaceutical discovery and development, and can often be chromatographically resolved using specialized columns and conditions.^{101,102} Similarly, dehalogenation impurities, often originating as unwanted side products of transition metal catalysis, are sometimes very difficult to chromatographically resolve from the parent drug molecule. In an in-depth study of this phenomenon, we found that specialized halogenated stationary phases were often effective for resolution of such mixtures.¹⁰³⁻¹⁰⁵ In addition, the excellent shape-selectivity of chiral stationary phases makes them well suited for dealing with many challenging separations of closely related species.^{106,107} Interestingly, a study of the separation problems of academia showed that many academic separation challenges involve the resolution of closely related species, which can often be addressed using modern separation science tools from the pharmaceutical industry.¹⁰⁸

Sensors and other non-chromatographic technologies

Non-chromatographic techniques for high throughput analysis offer speed advantages, although care must be taken to ensure that analyses are free from sample interference. Direct MS measurements using DESI or DART ionization are increasingly being used,¹⁰⁹ and we recently achieved the high throughput analysis of 1536 well microplates of reaction mixtures using MALDI MS in 8 minutes (3 samples per second),¹¹⁰ which compares favorably with the speed of droplet microfluidic MS analysis reported by Kennedy^{111,112} and others. In addition, colorimetric, fluorimetric and CD-based sensors such as those developed by Anslyn,¹¹³ Wolf¹¹⁴ Joyce^{115,116} and Koide⁶⁸⁻⁷⁰ are increasingly being used in high throughput analysis of reaction mixtures.¹¹⁷

Looking forward in high throughput analysis

Continued innovation in the development of high throughput analysis techniques can be expected as the use of high throughput experimentation becomes more widely practiced in chemical research. Chromatographic methods are currently limited by the speed of sample injection (currently ~10 seconds per sample) but the combination of faster autosamplers and recent developments in ultrafast chromatography will likely lead to significant increases in analysis speed, putting chromatographic methods on a par with competing spectroscopic techniques.^{92,118} Linking of platforms for high throughput experimentation and high throughput analysis will further streamline research, eliminating the need for manual intervention between experimentation and analysis. In addition, the use of artificial intelligence and machine learning will lead to faster experimental cycles, reducing the



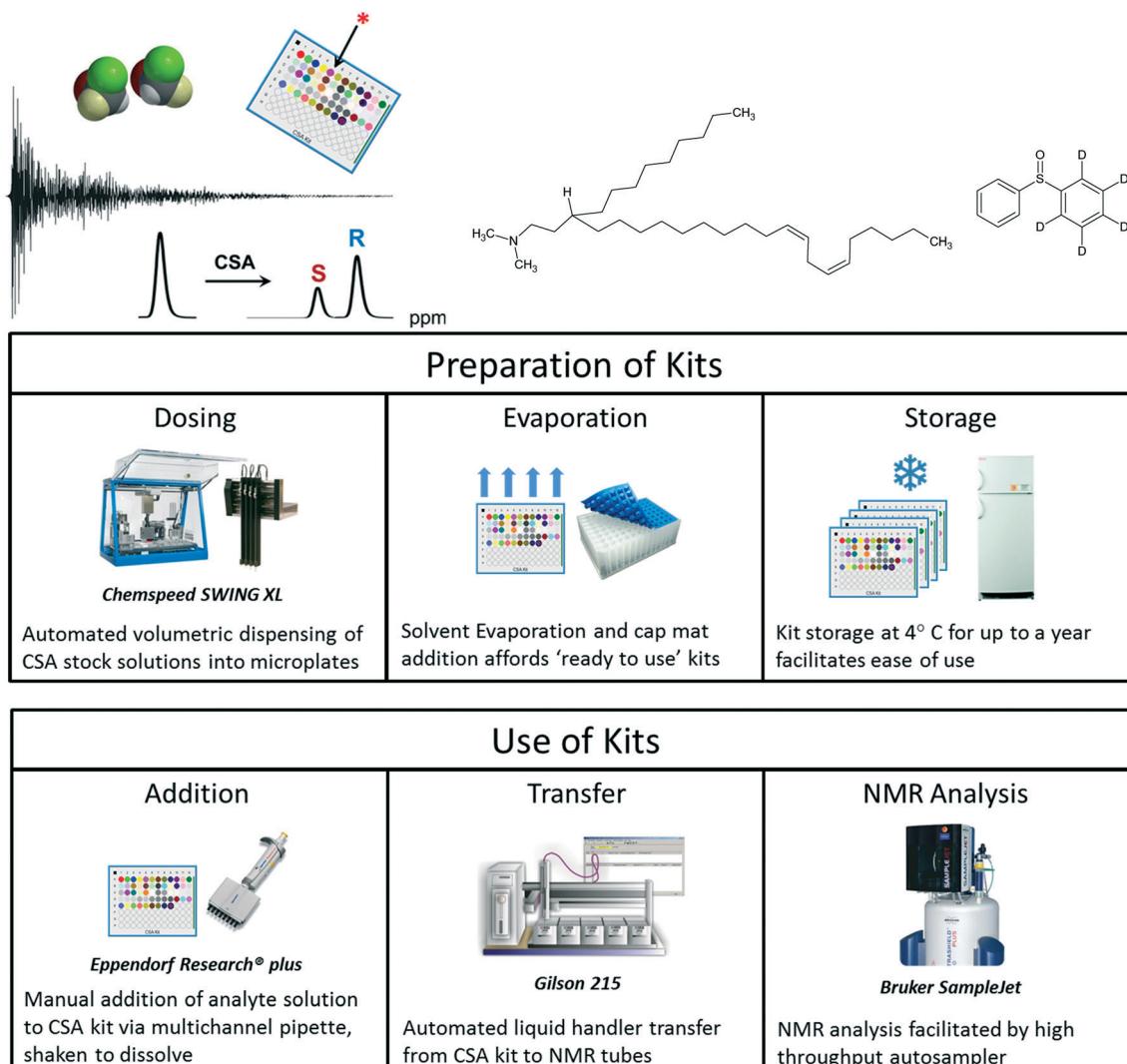


Fig. 12 Automated screening of chiral solvating agents for NMR determination of enantiopurity permits the analysis of challenging compounds not easily resolved using chromatography.

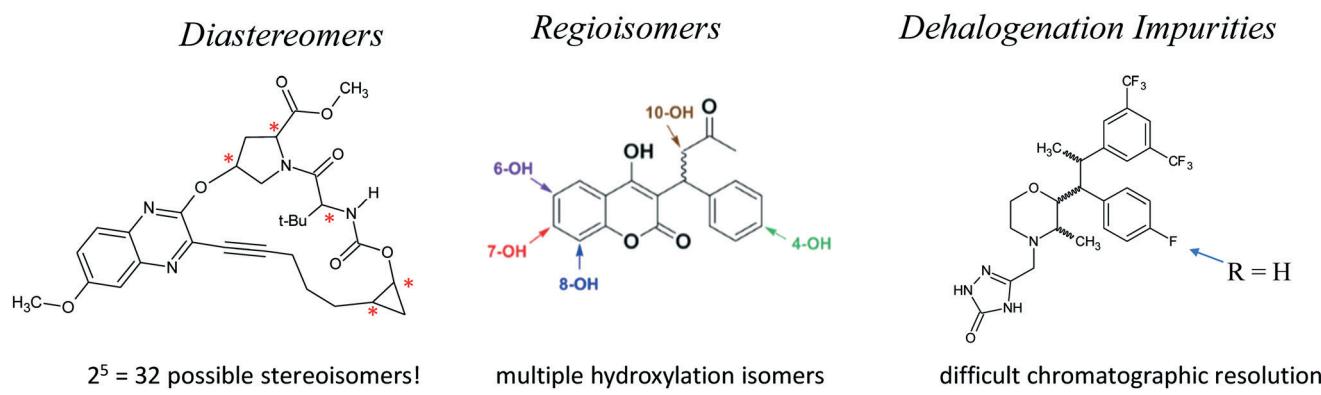


Fig. 13 Closely related species present significant challenges for chromatographic separation.

need for human scoring and intervention between each experimental cycle and permitting automated optimization of reactions and processes. Intelligent systems capable of carrying

out certain routine optimization tasks will lead to a more widespread adoption of these tools beyond the current cadre of technology enthusiasts. Further trends in miniaturization

of experimentation and analysis platforms are likely, permitting screening with reduced material demand and allowing for more convenient, portable tools. In addition, continued reduction in the cost and complexity of experimentation and analysis is likely, leading to expanded accessibility of high throughput analysis tools.

Finally, it should be noted that many of the tools and techniques for high throughput experimentation in pharmaceutical process research have been inspired by high throughput techniques originally developed for clinical diagnostics, drug activity screening or molecular biology applications. In addition, our research direction in this area was strongly influenced by participation in a NASA/JPL project aimed at developing autonomous, miniaturized analysis systems for exploring Saturn's moon, Titan.^{119,120} As the internet of things and the use of miniaturized sensors and wireless communication become more widespread, many other opportunities for technology borrowing will surely become possible.

Conclusions

Innovations in high throughput analysis have enabled the success of high throughput experimentation in pharmaceutical discovery and development. In this perspective, a personal view of some of the developments in this technology over the past quarter century is presented, detailing the challenges of developing new measurement approaches to support high throughput evaluation of crystallization, preparative chromatography and reaction screening workflows within pharmaceutical development. Ongoing efforts to increase analytical speed and throughput have been critically important to the success of this venture, with greater than a thousand-fold acceleration in the rate of analysis being possible in many cases. In addition, improvements in sensitivity, selectivity, user-friendliness and ease of data interpretation have also been important. As high throughput experimentation becomes more firmly established within industry and academia we can look forward in further challenges and innovations relating to the high throughput analysis and interpretation of experimental results.

Conflicts of interest

There are no conflicts to declare.

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