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Catalyst-free visible-light-induced condensation to synthesize bis(indolyl)methanes and biological activity evaluation of them as potent human carboxylesterase 2 inhibitors†

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A mild strategy for visible-light-induced synthesis of bis(indolyl)methanes was developed using aromatic aldehydes and indole as substrates. This reaction could be performed at room temperature under catalyst- and additive-free conditions to synthesize a series of bis(indolyl)methanes in good to excellent yields. In addition, all synthesized bis(indolyl)methanes together with β -substituted indole derivatives synthesized according to our previous work, were evaluated for their inhibitory effect against human carboxylesterase (CES1 and CES2). Primary structure-activity relationship analysis of all tested compounds showed that the modifications of β -substituted indole at the β -site with another indolyl group led to a significant enhancement of the inhibitory effect on CES2, and the bisindolyl structure is essential for CES2 inhibition. These results demonstrated that these bis(indolyl)methanes are potent and selective CES2 inhibitors, which might be helpful for medicinal chemists to design and develop more potent and selective CES2 inhibitors for biomedical applications.

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1. Introduction

Mammalian carboxylesterases (CES, E.C. 3.1.1.1) are key members of the serine hydrolase superfamily, and play important roles in the hydrolytic metabolism of ester-, thioester-, amide-, and carbamate-bond containing endogenous and xenobiotic compounds.^{1,2} In the human body, there are two predominant carboxylesterases, human carboxylesterase 1 (CES1) and human carboxylesterase 2 (CES2).^{3,4} As the major CES distributed in the human intestines and colon, CES2 is mainly known as the enzyme involved in the metabolism of some ester-containing drugs and environmental toxicants.^{5,6} For example, CPT-11, a carbamate prodrug prescribed for the treatment of colorectal cancer, could trigger severe delayed diarrhea due to the overproduction of SN-38 (the hydrolytic product of CPT-11) in the small intestine.⁷⁻⁹ Therefore, it is hypothesized that co-administration with potent CES2 inhibitors may ameliorate CPT-11 caused severe diarrhea in patients and thus improve the therapeutic effect.^{10,11} In addition to the

well-known roles of CES2 in xenobiotic metabolism, the enzyme also participates in endogenous metabolism.^{12,13} More recently, CES2 has been reported with triacylglycerol and diacylglycerol hydrolase activity, its expression and function in liver are strongly related with several metabolic diseases, such as obesity and non-alcoholic steatohepatitis.^{13,14} Thus, CES2 inhibitors could be used as potential tools for exploring the biological functions of CES2 in complex biological systems.

Indole derivatives are widely distributed in nature and possess beneficial biological activities, including anticancer,^{15,16} antifungal,¹⁷ anti-inflammatory,^{18,19} and antimicrobial activities.^{20,21} Among them, bis(indolyl)methanes and β -substituted indole are distinguished skeletons which are important intermediates for chemical synthesis and important pharmacophores.²² In view of this, there is already a considerable demand in medicinal and organic chemistry for developing versatile synthetic approaches to these two kinds of compounds and careful study their biological activities.²³⁻²⁸ The interest for developing mild and environment-friendly synthetic method of indole derivatives is one thread in our group.^{29,30} Meanwhile, we also focus on the further applications of heterocyclic compounds synthesized in our laboratory.

In recently years, biochemists have made significant breakthrough on the design and development of several highly specific optical substrates for sensing the real activities of CES1 or CES2 in complex biological systems.³¹⁻³⁷ These newly developed optical substrates have been successfully used for high

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throughput screening of natural or unnatural inhibitors against CES.^{38–43} However, indole and its derivative inhibitors against CES2 with high potency and enhanced selectivity are rarely reported.^{44,45} Based on the wide and beneficial activity of bis(indolyl)methanes and β -substituted indole derivatives, we were interested in whether these two kinds of compounds have effects against human CES1 and CES2. We have developed some environmentally benign synthetic methodologies for the preparation of β -substituted indole derivatives.^{29,30} Herein, we provide a relatively simple and environmental friendly reaction condition to synthesized bis(indolyl)methanes. More important, these two types of synthetic compounds have been meticulously screened for inhibitory effects against human CES1 and CES2.

2. Experiment

2.1. Chemicals and reagents

Fluorescein diacetate (FD), Loperamide (LPA) and bis-*p*-nitrophenyl phosphate (BNPP) were purchased from TCI (Tokyo, Japan) and their purity was greater than 98% by HPLC-UV measurement. β -Substituted indole derivatives **1a–p** were synthesized according to our previous work.^{29,30} Compounds **1a–p**, **4a–p**, LPA and BNPP were prepared in DMSO (Tedia, USA) to prepare a stock solution and stored in a refrigerator at 4 °C until use. The specific fluorescent probes D-luciferin methyl ester (DME) was synthesized in our laboratory.³⁶ Ultrapure water (Millipore, Bedford, USA). Human liver microsomes (HLM) purchased from Research Institute for Liver Disease (Shanghai) Co., Ltd. Luciferin detection reagent (LDR) was obtained from Promega Biotech (Madison, USA). Phosphate buffer (PBS) is prepared by ultrapure water and stored in a 4 °C refrigerator, divided into pH 6.5 and pH 7.4. All ¹H NMR (500 MHz) were recorded on a Bruker Avance spectrometer with chemical shifts reported as ppm (in DMSO-*d*₆, TMS as the internal standard).

2.2. General procedure for the synthesis of indole derivatives (**4a–p**)

A mixture of the appropriate aldehyde (0.5 mmol), 2-methylindole (1.0 mmol) and 1.5 mL solvent (EtOH–H₂O 1 : 1, v/v) were added in a 10 mL Schlenk tube. This solution was stirred under irradiation of blue LEDs (15 W LED light band was used). The reaction was monitored by TLC. After completion, 1 mL EtOH was added and vigorously stirred for 10 min to remove the impurity. Corresponding purified product was obtained after filtered and washed with cold aqueous ethanol. If necessary, product was purified by column chromatography using silica gel (200–300 mesh). (Please note: low power blue LED light band was used in the experiment. Exothermic effect was not obvious.)

The detailed structural characterization of indole derivatives (**4a–p**) are described in the ESI.†

2.3. Fluorescence-based CES inhibition assays

2.3.1. Inhibition of indole derivatives on CES1-mediated DME hydrolysis. The inhibitory effects towards CES1 was investigated with DME used as the specific probe substrates for

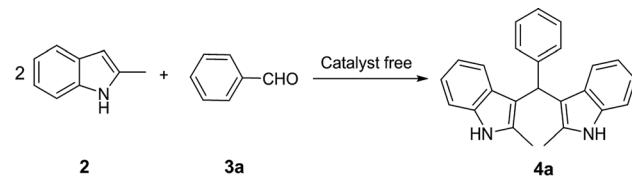
CES1.³⁶ In brief, each compound (2 μ L, different concentrations according to the initial screening settings) and CES1 enzyme source (HLM, 2 μ L, 0.02 mg mL^{−1}, initial concentration) were placed in phosphate buffer (94 μ L, pH 6.5), add the positive inhibitor BNPP in parallel and incubate for 5 min in a 37 °C incubator. The reaction was then initiated by the addition of probe DME (2 μ L, 0.15 mM, approximately equal to the K_m value of CES1 hydrolyzed DME). After shaking the reaction for 10 min in a 37 °C incubator, 50 μ L of the reaction solution was pipetted and mixed with LDR (50 μ L) to terminate the reaction. Detection was performed in a multi-mode microplate reader (SpectraMax® iD3, Molecular Devices, Austria). All compounds were formulated in chromatographic grade DMSO with a final volume ratio of DMSO of less than 1% (v/v).

2.3.2. Inhibition of indole derivatives on CES2-mediated FD hydrolysis.

The inhibitory effects toward CES2 were investigated with FD as the specific probe substrate. The reaction procedure mainly includes the following parts: firstly, phosphate buffer (194 μ L, pH 7.4), CES2 enzyme source (HLM, 2 μ L, 0.2 mg mL^{−1}, initial concentration), indole derivatives (2 μ L), and add the positive inhibitor LPA in parallel. The mixture was mixed and placed in a 37 °C incubator for 10 min, then added with FD (2 μ L, 0.2 mM, initial concentration) for 20 min, and an equal volume of acetonitrile (chromatographic grade) was added to the reaction solution to terminate the reaction. Then, 200 μ L of the mixture was pipetted into the microtiter plate. Place it in the microplate reader for testing. All compounds were formulated in chromatographic grade DMSO with a final volume ratio of DMSO of less than 2% (v/v).

2.3.3. Statistical analysis. The half maximal inhibitory concentration (IC₅₀) of each compound was determined using various inhibitor concentrations under the same incubation conditions as mentioned above. All values obtained from experiments were expressed as mean \pm SD. The IC₅₀ values of the compounds were evaluated using GraphPad Prism 7.0 (GraphPad Software, Inc., La Jolla, USA) nonlinear regression.

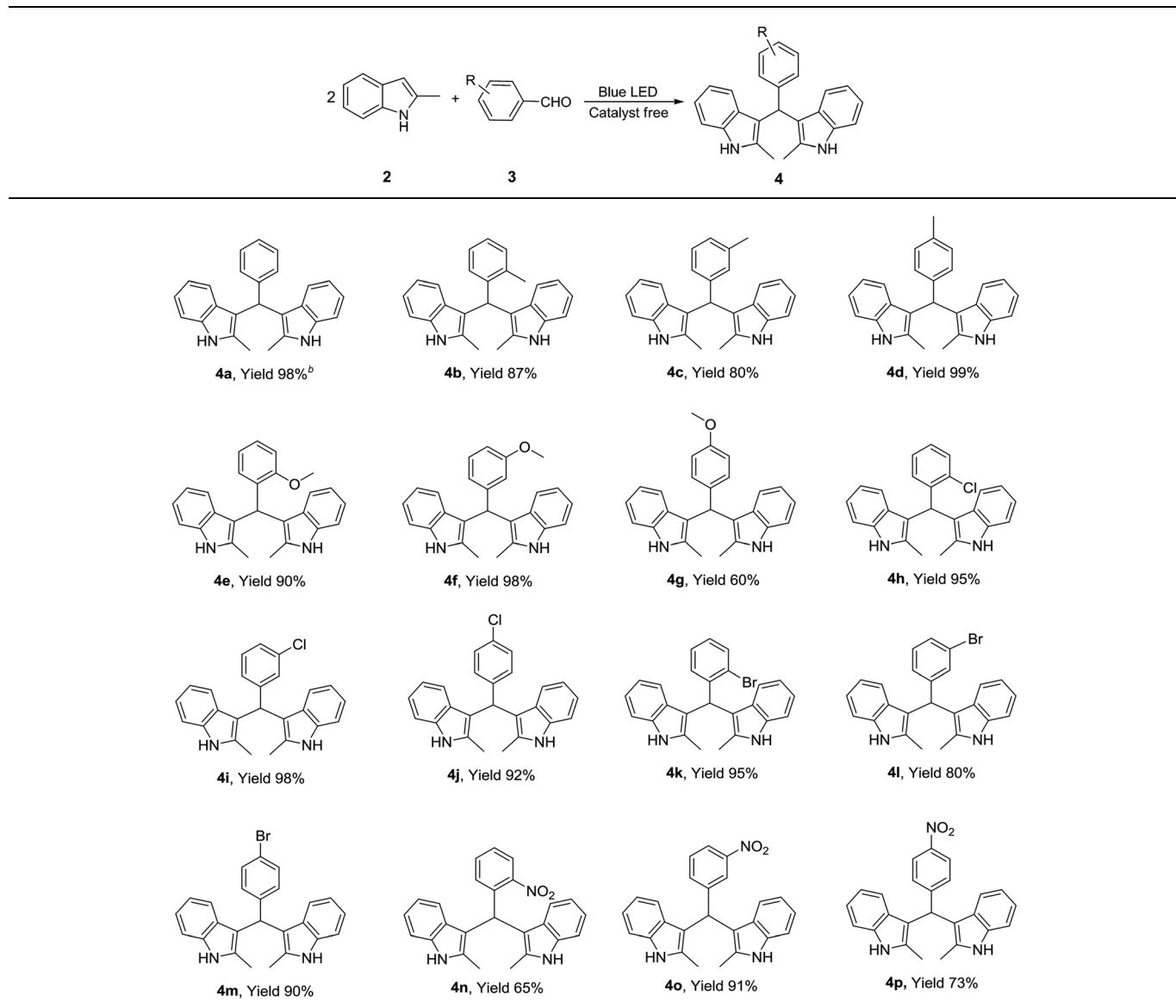
Table 1 Optimization of the reaction conditions for the synthesis of **4a**



Entry ^a	Solvent	Vol./mL	Time/h	Yield ^b /%
1	EtOH–H ₂ O	0.75/0.75	4	98
2	EtOH–H ₂ O	0.9/0.6	4	93
3	EtOH–H ₂ O	0.5/1.0	4	95
4	EtOH–H ₂ O	0.75/0.75	5	98
5	EtOH–H ₂ O	0.75/0.75	3	94
6 ^c	EtOH–H ₂ O	0.75/0.75	4	90
7 ^d	EtOH–H ₂ O	0.75/0.75	4	96

^a The reactions were carried out with 0.5 mmol benzaldehyde and 1.0 mmol indole. ^b Isolated yield. ^c At 45 °C without irradiation of blue LEDs. ^d Irradiation of white LEDs.



Table 2 Substrate scope for the synthesis of 4^a

^a The reactions were carried out with 0.5 mmol benzaldehyde and 1.0 mmol indole in 1.5 mL EtOH–H₂O (1 : 1) under irradiation of blue LEDs.

^b Isolated yield.

3. Results and discussion

3.1. Design synthetic protocols and optimize reaction conditions

Over the years, a plethora of synthetic protocols has been developed due to the wide range of applications in biomedicine and agriculture of bis(indolyl)methanes (BIMs).^{46–48} Among them, using indoles and aldehydes as substrates is one of the most effective synthesis methods that received more attention.^{49–52} Therefore, the first purpose of this study is to develop facile pathway for the green synthesis of BIMs *via* the reaction of indole with aldehydes. Recently, visible light as a rich, readily available and renewable clean energy source,^{53,54} has been

widely utilized to promote various organic synthetic transformations.^{55–57} In this context, visible light assisted reactions without catalyst meet the requirements of mild reaction conditions and environmental friendliness, which should be the best choice to implement the reaction of indole and aldehydes. Based on the references and our experiences, aqueous ethanol was good solvent for the reaction containing indole and aldehyde, therefore, it was used throughout our study. In view of the inhibitory property of β -substituted indole derivatives 1l–p toward CES2 in Section 3.3 and to realize our envisaged protocol, model reaction of 2-methylindole with benzaldehyde was performed with irradiation by blue light-emitting diodes (LEDs; 15 W) at room temperature under catalyst-free



Table 3 The IC₅₀ values of indole derivatives toward CES2^a

	CES2 IC ₅₀ (μM)		CES2 IC ₅₀ (μM)
1a	24.47 ± 4.55	4a	0.64 ± 0.13
1b	11.08 ± 2.45	4b	0.56 ± 0.16
1c	3.29 ± 0.28	4c	0.34 ± 0.091
1d	>100	4d	0.28 ± 0.027
1e	>100	4e	0.20 ± 0.044
1f	4.37 ± 0.66	4f	1.28 ± 0.24
1g	9.01 ± 2.49	4g	0.15 ± 0.032
1h	20.27 ± 5.19	4h	0.57 ± 0.086
1i	7.79 ± 1.24	4i	1.23 ± 0.37
1j	4.28 ± 0.36	4j	0.84 ± 0.23
1k	>100	4k	0.35 ± 0.091
1l	2.98 ± 0.53	4l	3.17 ± 0.81
1m	2.64 ± 0.35	4m	1.21 ± 0.28
1n	2.89 ± 0.30	4n	0.19 ± 0.045
1o	4.12 ± 0.84	4o	0.35 ± 0.11
1p	7.47 ± 0.77	4p	1.22 ± 0.20

^a All data presented are averages of at least three separate experiments.

condition. Delightedly, the desired product **4a** was obtained in 98% yield (Table 1, entry 1). The heat would generate due to the use of LED band and the temperature was about 45 °C. Considered this, a control experiment without light irradiation was conducted at 45 °C, but not at room temperature. Based on this result, visible light was proved essential for the reaction, because in the absence it, the yield was reduced (entries 1 and 6). When replace blue LED with white LED, a slight lower score is obtained, but still better than conventional heat method. The solvent was also finely tuned by employing ethanol with water in different volumes and proportions. After some experiments, we found this conversion took place with maximum isolated yield in the mixture of 0.75 mL ethanol and 0.75 mL water. Optimization result indicated that reacted 4 hours were adequate for this reaction. Target molecule could form and isolate in 98% yield after filtration and washing with 50% ethanol.

3.2. Substrate scope

Using the optimized reaction conditions, we examined the generality and scope of the protocol across a range of

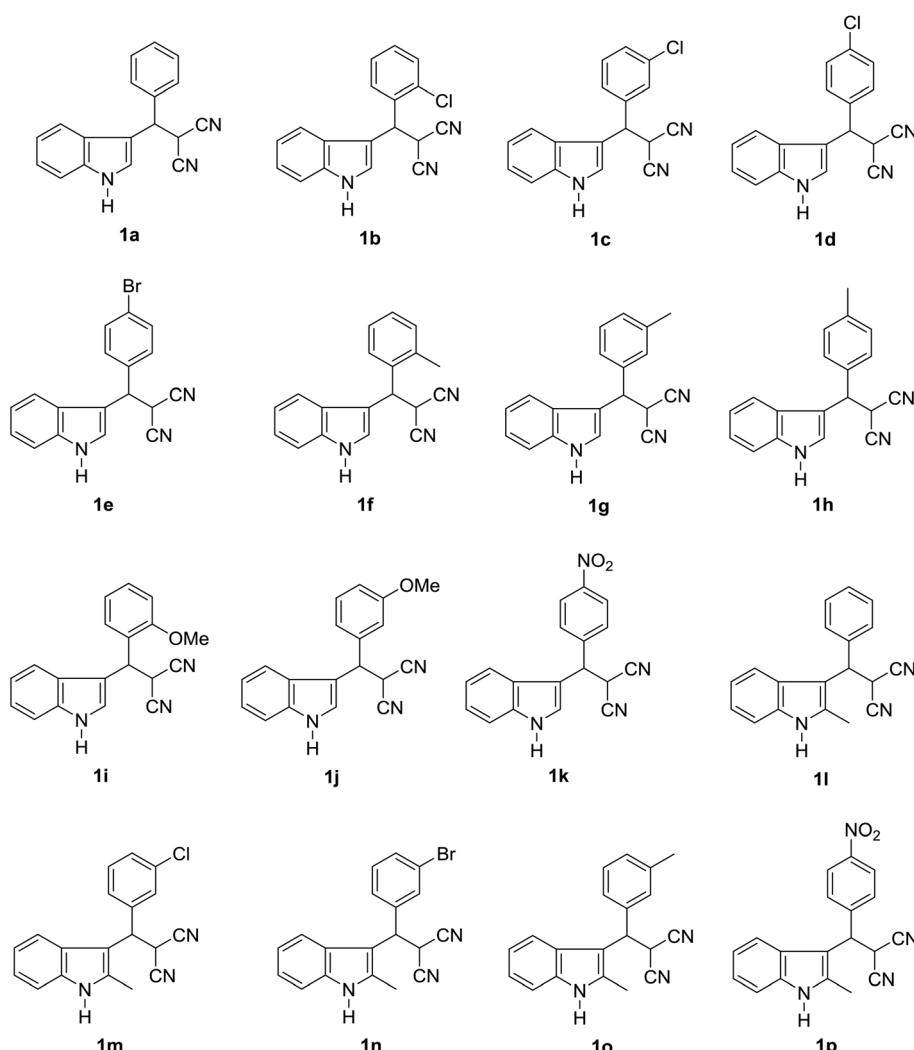


Fig. 1 Chemical structures of tested β -substituted indole derivatives **1a–p**.



Table 4 The IC_{50} values of indole derivatives toward CES2 and CES1^a

	CES2 IC_{50} (μM)	CES1 IC_{50} (μM)
4g	0.15 ± 0.032	2.90 ± 0.24
4n	0.19 ± 0.045	3.68 ± 0.43
LPA ^b	1.72 ± 0.28	>100
BNPP ^c	5.66 ± 0.62	0.24 ± 0.02

^a All data presented are averages of at least three separate experiments.

^b Loperamide, a positive inhibitor against CES2. ^c Bis-*p*-nitrophenyl phosphate, a positive inhibitor against CES1.

benzaldehyde incorporating various substituents, such as Me, MeO, Cl, Br, and NO₂. These reactions worked well in most cases, affording the corresponding products in moderate to excellent yields (Table 2). Aromatic ring bearing a weak electron-withdrawing group appeared to afford slightly higher yields compared with benzaldehyde having an electron-donating group. Introduction of strongly electron-withdrawing group on benzene ring such as nitryl was disfavored to the yield of reaction. The presence of substituents at various positions of benzaldehyde appreciably affected the yields too. For instance, methoxyl in meta-position on benzene ring had weak electron-withdrawing inductive effect and no conjugation effect. So, it afforded a better yield than other two methoxyl substituted benzaldehyde (Table 2, **4e–4g**). This outcome also showed that introducing strongly electron-withdrawing group in *ortho* or

para positions of phenyl ring would decrease the yield (Table 2, **4n**, **4p**). However, increasing the electron withdrawing substituents on aromatic ring at the 3-position of aromatic aldehyde gave the desired products in good yields too. In other words, nitro-group on aromatic ring in *ortho* or *para* positions, the reactivity of carbonyl was further increased. The electron deficient benzaldehyde, which containing an overactive carbonyl group was disadvantageous to the reaction under this condition. But satisfactory outputs could be furnished if they reacted without irradiation of blue LED. Generally, the electronic effect played a major role in governing the reactivity of the substrates and steric effect was not the dominant factor in the product formation.

3.3. Evaluation of inhibitory activities against CES1 and CES2

Following the establishment of these indole derivatives and considering the pharmacophore characters of the indole, a further investigation was carried out to evaluate whether these indole compounds possess potential biological activities against CES2 using FD as specific optical substrate. As shown in Table 2, we firstly focused our attention on the variation of the benzyl moiety of β -substituted indole derivatives **1a–q**. The benzyl unit with substitutes (Cl, Br, Me, MeO) at 2-position or 3-position afforded increased inhibitory effect against CES2 compared with compound **1a**. Compound **1h** with 4-methyl on

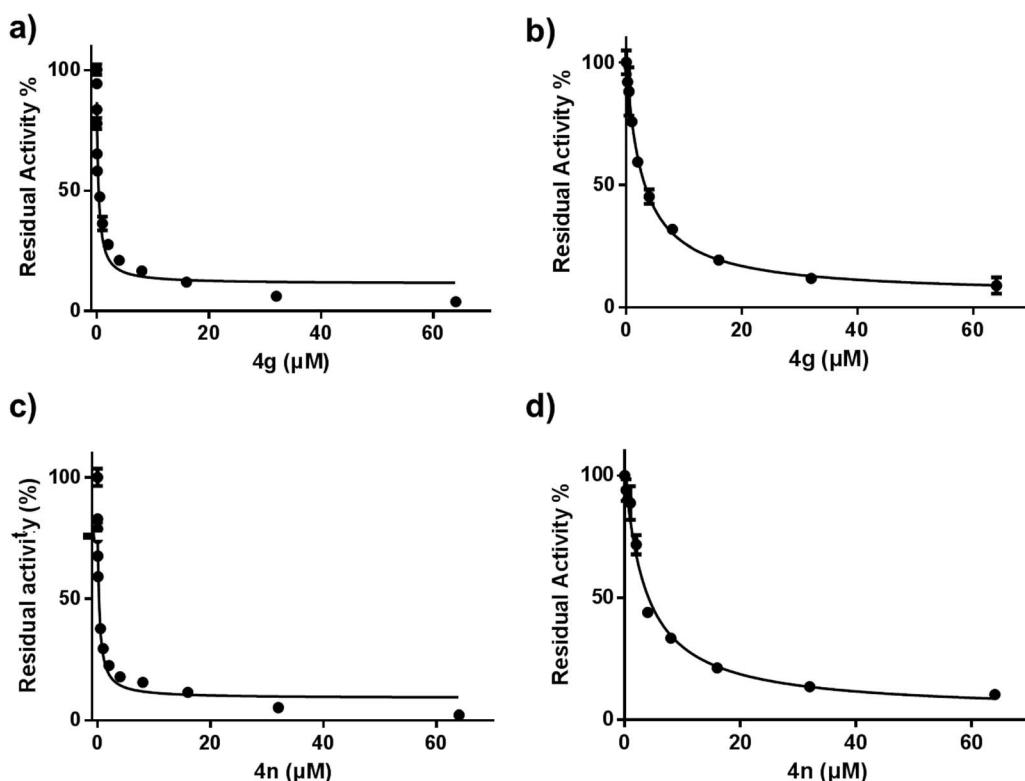


Fig. 2 The dose-dependent inhibition of **4g** and **4n** against CES2 mediated FD hydrolysis and CES1 mediated DME hydrolysis. Left: the dose-dependent inhibition curves of CES2. Right: the dose-dependent inhibition curves of CES1. All data represent the mean of triplicate determinations.



the benzyl unit exhibited similar trends in CES2 inhibition as compound **1a**, but the benzoyl unit with 4-methoxyl (**1d**), 4-chlorine (**1e**) and 4-nitryl (**1k**) substitute displayed poor inhibition toward CES2. Further characterizations of compounds **1l-p** with the variation of indole unit demonstrated that these compound with 2-methyl on indole unit is more beneficial for the inhibitory property of β -substituted indole derivatives toward CES2.

Following the good inhibitory effects of β -substituted indole derivatives on CES2, a survey was next carried out to evaluate the inhibitory effect against CES2 of these bis(indolyl)methanes **4a-p**. To our delight, all tested compounds **4a-p** showed a dramatically increase in the inhibitory effects against CES2 compared with compound **1** (Table 3). In addition, compounds **4g** and **4n** showed excellent inhibitory effect against CES2 with much lower IC_{50} value of 0.15 μ M and 0.19 μ M, respectively. Furthermore, the inhibitory effects of β -substituted indole derivatives **1a-q** and bis(indolyl)methanes **4a-p** on CES1 were routinely screened using one inhibitor concentrations (10 μ M) and using D-luciferin methyl ester (DME) as specific optical substrate. As shown in ESI† Fig. 1, a majority of tested these indole derivatives displayed weak inhibitory effects on CES1 at the concentration (10 μ M), only a few compounds showed relative strong inhibition potentials on CES1-mediated DME hydrolysis. As shown in Table 4, compounds **4g** and **4n** showed reduced inhibitory effect toward CES1 (IC_{50} 2.90 μ M and 3.68 μ M, respectively) in contrast to CES2. In addition, known inhibitors loperamide (LPA) and bis-*p*-nitrophenyl phosphate (BNPP) were tested under identical conditions as positive control for CES2 and CES1, respectively. The result indicated that LPA showed inferior inhibitory effects on CES2 in contrast to compounds **4g** and **4n** (Fig. 2).

4. Conclusions

In summary, a green protocol for catalyst free synthesis of BIMs from 2-methylindole and aryl aldehydes with good yields has been accomplished. We anticipate that blue light assisted reaction would offer efficient and cost effective way to obtain this important class of compounds. The reaction is free from hazardous metal or Lewis acid catalyst and uses environment-friendly solvent. In addition, two kinds of indole derivatives including β -substituted indoles and BIMs were collected and their inhibitory effects against human carboxylesterases were assayed using FD and DME as specific optical substrate for CES2 and CES1, respectively. The primary SAR analysis revealed that the bisindolyl structure of BIMs is a key factor affecting the inhibitory effects of indole derivatives on CES2. Furthermore, BIMs **4g** and **4n** demonstrated strong inhibitory effects on CES2 (IC_{50} , 0.15 μ M and 0.19 μ M, respectively) and good selectivity over CES1. Further studies directed toward expanding the synthetic utility and evaluating the biological activity of BIMs are currently ongoing in our laboratory.

Conflicts of interest

There are no conflicts to declare.

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