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Facile synthesis of novel dithioacetal–naphthalene derivatives as potential activators for plant resistance induction†

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In this paper, a series of novel dithioacetal–naphthalenes were designed and synthesized for plant immunity. Their antiviral activities were evaluated against tobacco mosaic virus (TMV) and cucumber mosaic virus (CMV). The results indicated that most compounds exhibited better activity against CMV than against TMV. These dithioacetal derivatives also displayed good bacterial activity against rice bacterial leaf blight. Among them, compound **S16** exhibited relatively good anti-CMV, anti-TMV, and antibacterial activity. Structure–activity relationships indicated that introducing the naphthalene moiety enhanced their activities for plant resistance induction. Therefore, the basic motif of compound **S16** could be the most promising candidate for further structural optimization to develop a potential activator for plant resistance induction.

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Introduction

Plant viruses are pathogenic to plants, infecting many plants, especially vegetables, including pepper, tomato, cucumber, and so on.¹ Owing to the diversity of virus species, the different transmission mechanisms and the mutability of viruses under field conditions, it is extremely difficult to control viral infection,² which results in massive economic losses each year.^{3–5} Although numerous chemical and biological controls have been applied, there have been few efficient compounds that can protect plants completely from virus infection.⁶ In the field, the reported antiviral agents,⁷ such as ninominomycin, dufulin, ribavirin, lentinan polysaccharide, emodinmethyl ether, morinanidine hydrochloride, chlorbromo-isocyanurate, DHT, DADHT, and chitooligosaccharide, are often effective only against one virus while disabled against the others. Furthermore, the agents with a preventive effect are less than 60% among the antiviral agents.⁸ Therefore the development of new antiviral agents with an efficient broad-spectrum is extremely urgent.

In fact, plants have self-defend system and can resist the infection of bacteria, mold and viruses.⁹ Such self-defensive ability is commonly induced by some external or internal

elicitors. Based on plant immune resistance, these elicitors are developed into antiviral agents and used for the prevention and treatment of plant virus.^{10–14} In the development of the chemical antiviral, people mainly consider the virus itself. Many of these reported drugs suffer from low effectiveness and narrow-spectrum antiviral property. In addition, the absolute parasite of the virus on the host is closely related to the plant. In general, two factors of virus inhibition and plant activation must be considered together to develop efficient broad-spectrum antiviral agents.

Dithioacetal and its derivatives have extensive biological activities, such as antibacterial,¹⁵ antileishmanial,¹⁶ antiviral,¹⁷ and antifungal.^{18,19} Song *et al.* found that dithioacetal derivative **6f** (Fig. 1) exerted markedly curative and protective activities against PVY and CMV.²⁰ Further research results by this group showed that dithioacetal **C14** (Fig. 1) elicited excellent curative and protective activities against PVY, CMV and TMV.²¹ Their mechanism associated with the change of SOD, CAT, and POD was also demonstrated. Based on above, introduction of plant immune elicitors, dithioacetal compounds could be developed into novel antiviral agents with effectiveness and spectrum width.

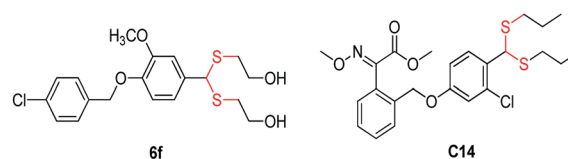


Fig. 1 The structures of the reported dithioacetal compounds with high activity.

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Naproxen acetic acid and naproxen sodium derivatives are important plant regulators for agriculture.²² These regulators can promote plant growth and chlorophyll synthesis. Especially it can promote the formation of adventitious roots and roots with fruit expansion and anti-falling function. We then introduced the plant regulated naphthalene group to dithioacetal skeleton (Fig. 2). Moreover, their bioactivities against cucumber mosaic virus and tobacco mosaic virus were subsequently evaluated and preliminary structure–activity relationship was concluded.

Results and discussion

The condensation reaction of aromatic aldehyde bearing naphthalene ring (**M1**) with allyl mercaptan was chosen as a model reaction to optimize the reaction conditions and the results are summarized in Table 1. It was observed that the reaction failed to give the product without the catalyst (Table 1, entry 1). In initial studies, different catalysts were screened in dichloromethane as the solvent (1 mL) at 40 °C (Table 1, entry 2 and 3). It was found that the acid ionic liquid (BIL) could effectively catalyze the condensation reaction. The effects of solvents were then evaluated by using ionic liquid (BIL) as the catalyst (Table 1, entry 3–6). Dichloroethane (Table 1, entry 5) was found to be an optimal solvent, giving **S1** in a good yield (up to 97%). Almost no change in reactivity was observed when the loading of BIL-HSO₄ was reduced from 10 mol% to 5 mol% (Table 1, entries 5 and 7). However, the product decreased from 97% to 69% when the catalyst load was reduced from 5 mol% to 1 mol% (Table 1, entries 7 and 8).

The reusability of the catalyst (BIL-HSO₄) was investigated and the results are described in Fig. 3. The reaction mixture was concentrated under reduced pressure to remove the dichloromethane. 20 mL water was added to the mixture. Then the catalyst was recovered by using simple filtration technique. The filtrate containing ionic liquid was dried over vacuum to remove excess water. This catalyst was directly subjected to the condensation reaction using the model reaction with our optimized reaction conditions. It is important to note that the recycled catalysts produced excellent yields of dithioacetal (**S1**)

Table 1 Optimizations of reaction conditions for the synthesis of **S1** catalyzed by ionic liquid (BIL)

Entry ^a	Catalyst (mol%)	Solvent	Temp/°C	Time	Yield ^b /%
1	—	DCM	40	6	0
2	ZrCl ₄ (10)	DCM	40	6	71
3	BIL-HSO ₄ (10)	DCM	40	6	81
4	BIL-HSO ₄ (10)	CH ₃ CN	80	6	67
5	BIL-HSO ₄ (10)	DCE	80	6	97
6	BIL-HSO ₄ (10)	DOX	110	6	87
7	BIL-HSO ₄ (5)	DCE	80	12	96
8	BIL-HSO ₄ (1)	DCE	80	12	69

^a Reaction conditions: aldehyde (1.0 mmol), thiol (1.0 mmol), solvent (1.0 mL). ^b Isolated yield.

in 91–94%, respectively (Fig. 3). It was observed that the yields were consistent without significant loss in its catalytic activity.

Under these optimized reaction conditions (Table 1, entry 7), substituted aldehydes bearing naphthalene ring (**M1–M6**), using acid ionic liquid (BIL),²³ reacted with different thiols to generate novel dithioacetals **S1–S16** with excellent yields of 83–97%.

First, compounds **S1–S16** were measured for their phytotoxic activity against tobacco.²⁴ The data of phytotoxic activity at 500 μg mL⁻¹ indicated that compounds **S1–S16** showed no toxicity to the tested plant.

The *in vivo* anti-TMV activities of target compounds **S1–S16** at the concentration of 500 μg mL⁻¹ were evaluated through the half leaf method.^{25–29} The results were shown in Table 1. Ningnanmycin and dufulin were used as the controls. Most of the title compounds exhibited good antiviral activities against CMV. Compounds **S5**, **S8**, **S12** and **S16** possessed excellent curative and protective activities from 61.5% to 71.3%, which was significantly greater than those of the controls with the vicinity of 50%. Compounds **S1**, **S4**, **S9** and **S13** exhibited good curative and protective activities with 55% or so, which were slightly exceeded than those of controls. Other compounds maintained moderate antiviral activities. The results of the activity against TMV showed that the target compounds displayed common

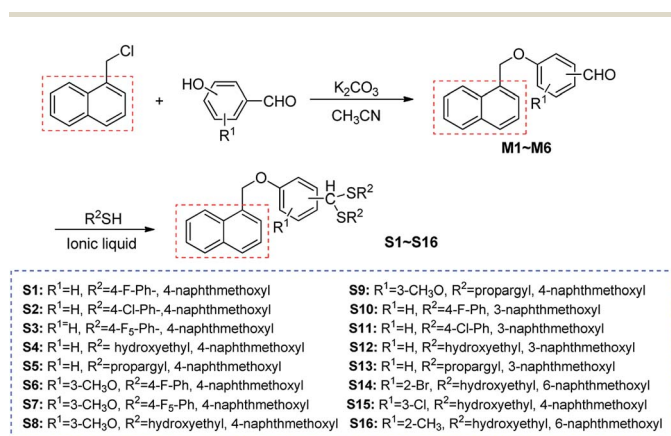


Fig. 2 The synthetic routes of novel dithioacetal derivatives.

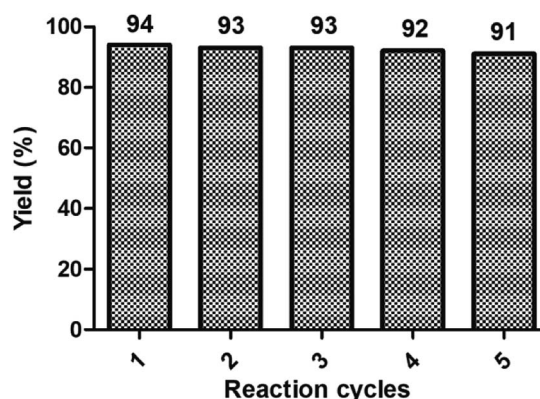


Fig. 3 Reusability of ionic liquid (BIL-HSO₄) for the synthesis of compound **S1**.



Table 2 Antiviral activity of the title compounds (S1–S16) against TMV and CMV at 500 $\mu\text{g mL}^{-1a}$

Compd	Anti-TMV		Anti-CMV	
	Curative activity (%)	Protective activity (%)	Curative activity (%)	Protective activity (%)
S1	45.3 \pm 2.5%	37.8 \pm 2.2%	53.2 \pm 1.6%	57.8 \pm 2.3%
S2	31.7 \pm 2.4%	16.2 \pm 3.1%	36.6 \pm 2.1%	31.2 \pm 1.9%
S3	35.8 \pm 2.5%	26.7 \pm 2.0%	47.5 \pm 2.2%	46.9 \pm 1.7%
S4	32.5 \pm 1.9%	28.2 \pm 2.1%	57.4 \pm 1.3%	59.3 \pm 2.2%
S5	45.9 \pm 2.4%	38.6 \pm 5.3%	66.1 \pm 2.8%	68.3 \pm 2.4%
S6	34.1 \pm 2.1%	34.9 \pm 1.9%	44.2 \pm 2.3%	41.6 \pm 1.6%
S7	39.6 \pm 3.0%	37.6 \pm 2.7%	47.1 \pm 2.6%	44.9 \pm 2.1%
S8	40.6 \pm 3.4%	35.3 \pm 2.8%	61.5 \pm 3.1%	64.4 \pm 2.0%
S9	41.2 \pm 2.4%	42.6 \pm 3.8%	56.4 \pm 2.3%	54.8 \pm 1.6%
S10	34.8 \pm 1.8%	29.0 \pm 3.3%	46.1 \pm 1.9%	48.7 \pm 3.5%
S11	50.0 \pm 2.7%	47.4 \pm 2.4%	41.7 \pm 2.2%	43.8 \pm 2.6%
S12	32.4 \pm 3.6%	18.4 \pm 3.1%	62.9 \pm 1.5%	68.3 \pm 2.0%
S13	40.1 \pm 2.7%	36.4 \pm 2.3%	59.6 \pm 2.1%	56.3 \pm 1.5%
S14	39.3 \pm 2.5%	34.5 \pm 2.4%	49.1 \pm 2.8%	46.4 \pm 2.2%
S15	45.1 \pm 2.6%	44.7 \pm 2.1%	65.3 \pm 1.9%	63.1 \pm 1.7%
S16	49.6 \pm 2.4%	48.1 \pm 1.7%	71.5 \pm 1.4%	69.1 \pm 2.1%
Control ^b	46.3 \pm 2.1%	49.4 \pm 2.6%	51.3 \pm 1.8%	53.1 \pm 2.1%
Control ^c	53.1 \pm 1.7%	63.4 \pm 2.4%	48.7 \pm 2.1%	49.4 \pm 2.6%

^a The reaction was conducted in anoxic conditions. ^b Dufulin was used as the control. ^c Ningnanmycin was used as the control.

inhibitory effects (Table 2). The curative activities of **S1** (45.3%), **S5** (45.9%), **S11** (50.0%) and **S16** (49.6%) against TMV were similar to those of dufulin (46.3%), slightly lower than ningnanmycin (53.1%). The protective activities against TMV of **S11** (47.4%) and **S16** (49.6%) were similar to those of dufulin (49.4%), much less than ningnanmycin (63.4%). Moreover, the activity of the title compound **S8** is excess than those of the accordingly intermediates referring to the literature.²⁰

Bioassay results indicated that the introduction of naphthalene moiety to the dithioacetals can effectively improve the antiviral activity. In addition, among these compounds **S1–S5**, the R² groups affect the antiviral activity (**S1**, **S5** > **S2**, **S3**, **S4**), with 1-naphthalene-methyl moiety at the *para*-position of dithioacetals in anti-TMV activity. The effect of the R² groups in anti-CMV activity is more obvious (**S5** > **S4** > **S1** > **S3** > **S2**). Compound **S5** or **S4** exhibited high activities when R² is hydroxyethyl or propargyl moiety. To study the influence of the R¹ group, among these compounds **S6–S9**, its effect on activity is quite small with the CH₃O group at the *meta*-position in anti-TMV activity. Nevertheless, compound **S8** with the CH₃O group at the R¹ position and the hydroxyethyl group showed good activity against CMV, further illustrate that the hydroxyethyl group is favorable for activity (**S4**, **S8** and **S12**) regardless of naphthalene and R¹ position on dithioacetals. To screen out high active compounds, the hydroxyethyl group was choosed to change the R¹ group and naphthalene position of the title compounds, the compound **S16** exhibited best activity against CMV and TMV. The results indicate that R₁ and naphthalene moiety is important for antiviral activities. Meanwhile, the hydroxyethyl group is indispensable for the antiviral activities of these compounds.

The antibacterial activities of the title compounds (**S1–S16**) against rice bacterial leaf blight were evaluated by turbidimeter tests.³⁰ Bismertiazol was used as the positive control.³¹ The results showed that the title compounds displayed moderate to good antibacterial activities (Table 3). Among them, compounds **S4**, **S7**, **S8** exhibited good *in vitro* antibacterial activity at concentrations of 200 and 100 $\mu\text{g mL}^{-1}$, which were slightly superior to bismertiazol (82.3 and 57.8%, respectively). Compounds **S2**, **S6**, **S15** and **S16** displayed moderate antibacterial activity, compared to that of control. Other compounds maintain low antibacterial activities. From the Table 3, SAR is easier to summarize. When R¹ = H and 1-naphthalene-methoxy at the 4-position, the order of activities (**S1–S5**) from high to low activity at 100 $\mu\text{g mL}^{-1}$ was hydroxyethyl (71.1%), 4-Cl-Ph- (57.0%), 4-F-Ph- (42.4%), 4-F₅-Ph- (33.0%), propargyl (19.9%). Different dithioacetal groups obviously affected the activities. When R¹ was 3-CH₃O and naphthalene-methoxy at 4-position, the order of activities (**S6–S9**) at 100 $\mu\text{g mL}^{-1}$ was hydroxyethyl (71.7%), 4-F₅-Ph- (70.6%), 4-F-Ph- (63.1%), propargyl (46.8%). Among them, 3-CH₃O group could increase the activity. When R¹ was H-group and naphthalene-methoxy at 3-position, the order of activities (**S10–S13**) was hydroxyethyl (51.8%), 4-F-Ph- (30.3%), 4-Cl-Ph- (25.6%), propargyl (22.8%). When R² was selected for the hydroxyethyl group, the sequence activities of compounds were **S15** (63.2%), **S16** (57.1%), **S14** (39.6%). The above results further suggested that the dithioacetal hydroxyethyl group of the title compounds is critical for the activity and the R¹ substituted and 1-naphthalene-methoxy of the derivatives influenced the antibacterial activity.

To investigate the stability property of aromatic dithioacetals, compound **S1** was selected to investigate in different conditions. First, compound **S1** was stable under the condition



Table 3 Activities of the title compounds (S1–S16) against rice bacterial leaf blight

Compd	Inhibition ^a (%)		Compd	Inhibition ^a (%)	
	200 µg mL ⁻¹	100 µg mL ⁻¹		200 µg mL ⁻¹	100 µg mL ⁻¹
S1	68.1 ± 1.8	42.4 ± 2.2	S9	66.2 ± 3.5	46.8 ± 5.8
S2	98.9 ± 2.2	57.0 ± 4.5	S10	79.2 ± 2.6	30.3 ± 5.2
S3	74.5 ± 3.8	33.0 ± 2.4	S11	49.8 ± 1.3	25.6 ± 5.4
S4	98.7 ± 3.9	71.1 ± 1.9	S12	90.0 ± 3.7	51.8 ± 3.8
S5	51.8 ± 1.9	19.9 ± 1.7	S13	50.0 ± 0.5	22.8 ± 6.6
S6	84.7 ± 3.0	63.1 ± 0.8	S14	69.1 ± 4.6	39.6 ± 5.0
S7	93.1 ± 3.4	70.6 ± 3.7	S15	98.9 ± 3.9	63.2 ± 4.5
S8	100.0 ± 1.6	71.7 ± 3.2	S16	93.3 ± 3.4	57.1 ± 1.1
Control ^b	82.3 ± 4.3	57.9 ± 3.1			

^a Average of three replicates. ^b Bismertiazol was used as the control.

of 0.1 M hydrochloric acid solution. Subsequently, compound S1 was also stable in 0.1 M NaOH. The results indicated that aromatic dithioacetals acetals displayed much better stability than acetals.

Conclusions

In summary, a series of novel dithioacetals–naphthalene, total sixteen compounds have been designed and synthesized with moderate yields. Their structures have been fully confirmed. Bioassay results showed that some compounds exhibited good anti-CMV activities, antibacterial activity and moderate anti-TMV activities *in vivo*. Among them, compound S16 in particular showed the potent activity against CMV, TMV and good antibacterial activity. Therefore, the basic motif of S16 can be used as lead compound for further development.

Experimental

General information

All of the reagents were purchased from commercial suppliers and used without further purification. All of the solvents were used without further purification and drying before use. Thin-layer chromatography with UV detection was conducted on silica gel GF254. The melting points of the products were determined with a WRX-4 microscopic melting point meter (Shanghai Yice Apparatus & Equipment Co., Ltd., China) with an uncorrected thermometer. ¹H, and ¹³C nuclear magnetic resonance (NMR) spectra were recorded on a Bruker Ascend-400 spectrometer (Bruker, Germany) in CDCl₃ solution with tetramethylsilane as internal standard. High resolution mass spectral (HRMS) data were determined with Thermo Scientific Q Exactive (Thermo).

Preparation of aromatic aldehydes 1. Chloromethyl naphthalene (I, 10 mmol) was added to a vial containing acetonitrile (40 mL), and hydroxyl substituted benzaldehyde (II, 10 mmol) and potassium carbonate (10 mmol) were then added. After the mixture was stirred and refluxed for 6 h, the solvent was removed *in vacuo*. The residue was purified by column

chromatography on silica gel (EtOAc/hexane as eluent = 1 : 20) affording the aromatic aldehydes (M1–M6).

4-(Naphthalen-1-ylmethoxy)benzaldehyde (M1). White solid, mp 122–123 °C, yield 95%; ¹H NMR (400 MHz, CDCl₃) δ 9.91 (s, 1H), 8.07–7.99 (m, 1H), 7.96–7.83 (m, 4H), 7.65–7.43 (m, 4H), 7.16 (d, *J* = 8.7 Hz, 2H), 5.59 (s, 2H); ¹³C NMR (101 MHz, CDCl₃) δ 190.77 (s), 163.80 (s), 133.83 (s), 132.04 (s), 131.40 (s), 131.27 (s), 130.27 (s), 129.42 (s), 128.84 (s), 126.77 (s), 126.68 (s), 126.09 (s), 125.30 (s), 123.43 (s), 115.21 (s), 68.95 (s).

3-Methoxy-4-(naphthalen-1-ylmethoxy)benzaldehyde (M2). White solid, mp 111–112 °C, yield 94%; ¹H NMR (400 MHz, CDCl₃) δ 9.84 (s, 1H), 8.04 (d, *J* = 8.0 Hz, 1H), 7.92–7.81 (m, 2H), 7.62–7.48 (m, 3H), 7.48–7.36 (m, 3H), 7.08 (d, *J* = 8.2 Hz, 1H), 5.65 (s, 2H), 3.90 (s, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 190.91 (s), 153.73 (s), 150.30 (s), 133.79 (s), 131.28 (s), 131.22 (s), 130.50 (s), 129.14 (s), 128.82 (s), 126.57 (s), 126.54 (s), 126.27 (s), 126.00 (s), 125.35 (s), 123.31 (s), 112.74 (s), 109.60 (s), 69.51 (s), 56.07 (s).

3-(Naphthalen-1-ylmethoxy)benzaldehyde (M3). White solid, mp 95–96 °C, yield 97%; ¹H NMR (400 MHz, CDCl₃) δ 9.99 (s, 1H), 8.09–8.00 (m, 1H), 7.94–7.83 (m, 2H), 7.62–7.44 (m, 7H), 7.32–7.26 (m, 1H), 5.54 (s, 2H); ¹³C NMR (101 MHz, CDCl₃) δ 192.07 (s), 159.41 (s), 137.91 (s), 133.83 (s), 131.67 (s), 131.51 (s), 130.20 (s), 129.30 (s), 128.78 (s), 126.78 (s), 126.60 (s), 126.03 (s), 125.32 (s), 123.82 (s), 123.60 (s), 122.30 (s), 113.32 (s), 68.94 (s).

3-Chloro-4-(naphthalen-1-ylmethoxy)benzaldehyde (M4). Sandy solid, mp 119–121 °C, yield 96%; ¹H NMR (400 MHz, CDCl₃) δ 9.86 (s, 1H), 8.07 (d, *J* = 8.1 Hz, 1H), 8.00–7.83 (m, 3H), 7.77 (dd, *J* = 8.5, 2.0 Hz, 1H), 7.66 (d, *J* = 6.8 Hz, 1H), 7.63–7.53 (m, 2H), 7.53–7.46 (m, 1H), 7.23 (d, *J* = 8.5 Hz, 1H), 5.70 (s, 2H). ¹³C NMR (101 MHz, CDCl₃) δ 189.67 (s), 159.00 (s), 133.77 (s), 131.45 (s), 131.10 (s), 130.70 (s), 130.55 (s), 130.29 (s), 129.36 (s), 128.85 (s), 126.67 (s), 126.22 (s), 126.11 (s), 125.28 (s), 124.47 (s), 123.22 (s), 113.32 (s), 69.73 (s).

2-Bromo-5-(naphthalen-1-ylmethoxy)benzaldehyde (M5). Red solid, mp 129–131 °C, yield 94%; ¹H NMR (400 MHz, CDCl₃) δ 10.35 (s, 1H), 8.06–7.99 (m, 1H), 7.95–7.86 (m, 2H), 7.65 (d, *J* = 3.2 Hz, 1H), 7.63–7.52 (m, 4H), 7.49 (dd, *J* = 8.1, 7.2 Hz, 1H), 7.15 (dd, *J* = 8.8, 3.2 Hz, 1H), 5.54 (s, 2H). ¹³C NMR (101 MHz, CDCl₃)



δ 191.68 (s), 158.46 (s), 134.72 (s), 134.09 (s), 133.82 (s), 131.46 (s), 131.29 (s), 129.41 (s), 128.79 (s), 126.86 (s), 126.64 (s), 126.05 (s), 125.26 (s), 123.83 (s), 123.50 (s), 118.30 (s), 113.94 (s), 69.20 (s).

5-Methyl-2-(naphthalen-1-ylmethoxy)benzaldehyde (M6). White solid, mp 114–116 °C, yield 93%; ^1H NMR (400 MHz, CDCl_3) δ 10.43 (s, 1H), 8.06–8.02 (m, 1H), 7.96–7.82 (m, 2H), 7.67 (d, J = 2.1 Hz, 1H), 7.63–7.42 (m, 4H), 7.38 (dd, J = 8.5, 2.3 Hz, 1H), 7.10 (d, J = 8.5 Hz, 1H), 5.60 (s, 2H), 2.33 (s, 3H); ^{13}C NMR (101 MHz, CDCl_3) δ 189.88 (s), 159.26 (s), 136.52 (s), 133.81 (s), 131.60 (s), 131.40 (s), 130.62 (s), 129.30 (s), 128.84 (s), 128.47 (s), 126.67 (s), 126.53 (s), 126.06 (s), 125.26 (s), 125.12 (s), 123.36 (s), 113.24 (s), 69.38 (s), 20.30 (s).

Procedure for synthesis of dithioacetal derivatives (S1–S16). Aromatic aldehyde (M, 1.0 mmol), thiols (1.0 mmol), 2.5 mol% Brønsted acidic ionic liquid were added to 1.0 mL of dichloromethane. The mixture was stirring at 40 °C for 3 h. The resulting mixture was concentrated under reduced pressure to give crude product. 5 mL water was added to the crude product and continues to stirring for 0.5 h. Then the water with ionic liquid was removed. Finally, the crude product was purified by column chromatography using hexane/EtOAc (1 : 2, v/v) or recrystallization with alcohol.

Bis(4-fluorophenyl)-4-(naphthalen-1-ylmethoxy)-phenyl-dithioacetal (S1). White solid, mp 103–105 °C, yield 96%; ^1H NMR (400 MHz, CDCl_3) δ 8.05 (d, J = 7.4 Hz, 1H), 7.95–7.86 (m, 2H), 7.61–7.47 (m, 4H), 7.37–7.29 (m, 4H), 7.24 (d, J = 8.6 Hz, 2H), 7.01–6.91 (m, 6H), 5.49 (s, 2H), 5.26 (s, 1H); ^{13}C NMR (101 MHz, CDCl_3) δ 164.08 (s), 161.61 (s), 158.62 (s), 135.71 (d, J = 8.4 Hz), 133.80 (s), 132.05 (s), 131.72 (s), 131.49 (s), 129.30 (s), 129.11 (s), 128.75 (s), 126.57 (d, J = 12.3 Hz), 125.97 (s), 125.30 (s), 123.64 (s), 116.07 (s), 115.85 (s), 114.87 (s), 68.72 (s), 61.55 (s); IR (KBr, cm^{-1}) ν 1604.8 (s), 1584.1 (s), 1509.1 (s), 1487.6 (s), 1244.6 (s), 1222.7 (s); HRMS (ES) m/z for $\text{C}_{30}\text{H}_{22}\text{F}_2\text{OS}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 523.0972, found 523.0978.

Bis(4-chlorophenyl)-4-(naphthalen-1-ylmethoxy)-phenyl-dithioacetal (S2). White solid, mp 113–114 °C, yield 92%; ^1H NMR (400 MHz, CDCl_3) δ 8.09–8.02 (m, 1H), 7.96–7.85 (m, 2H), 7.62–7.47 (m, 4H), 7.34–7.20 (m, 10H), 6.98 (d, J = 8.7 Hz, 2H), 5.49 (s, 2H), 5.36 (s, 1H); ^{13}C NMR (101 MHz, CDCl_3) δ 158.75 (s), 134.21 (s), 134.07 (s), 133.80 (s), 132.72 (s), 132.02 (s), 131.50 (s), 131.28 (s), 129.15 (s), 129.04 (s), 128.75 (s), 126.66 (s), 126.52 (s), 125.98 (s), 125.31 (s), 123.64 (s), 114.99 (s), 68.75 (s), 60.21 (s); IR (KBr, cm^{-1}) ν 1505.4 (s), 1471.6 (s), 1233.5 (s), 1091.4 (s), 1005.4 (s); HRMS (ES) m/z for $\text{C}_{30}\text{H}_{22}\text{Cl}_2\text{OS}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 555.0381, found 555.0384.

Bis(pentafluorophenyl)-4-(naphthalen-1-ylmethoxy)-phenyl-dithioacetal (S3). White solid, mp 137–139 °C, yield 83%; ^1H NMR (400 MHz, CDCl_3) δ 8.05–7.98 (m, 1H), 7.95–7.83 (m, 2H), 7.62–7.51 (m, 3H), 7.50–7.44 (m, 1H), 7.41 (d, J = 8.7 Hz, 2H), 6.98 (d, J = 8.7 Hz, 2H), 5.66 (s, 1H), 5.48 (s, 2H); ^{13}C NMR (101 MHz, CDCl_3) δ 159.67 (s), 133.79 (s), 131.73 (s), 131.43 (s), 129.20 (s), 128.76 (s), 129.11 (s), 128.46 (s), 126.64 (s), 126.55 (s), 125.99 (s), 125.27 (s), 123.52 (s), 115.24 (s), 68.76 (s), 57.63 (s); IR (KBr, cm^{-1}) ν 1513.8 (s), 1487.9 (s), 1230.5 (s), 1098.2 (s); HRMS (ES) m/z for $\text{C}_{30}\text{H}_{14}\text{F}_{10}\text{OS}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 667.0219, found 667.0216.

Bis(2-hydroxyethyl)-4-(naphthalen-1-ylmethoxy)-phenyl-dithioacetal (S4). White solid, mp 105–107 °C, yield 94%; ^1H NMR (400 MHz, CDCl_3) δ 8.08–8.01 (m, 1H), 7.93–7.84 (m, 2H), 7.61–7.45 (m, 4H), 7.44–7.37 (m, 2H), 7.07–6.99 (m, 2H), 5.49 (s, 2H), 5.07 (s, 1H), 3.74 (t, J = 5.5 Hz, 4H), 2.86 (dt, J = 14.0, 5.8 Hz, 2H), 2.73 (dt, J = 14.0, 5.9 Hz, 2H), 2.17 (s, 2H); ^{13}C NMR (101 MHz, CDCl_3) δ 158.77 (s), 133.81 (s), 132.34 (s), 132.07 (s), 131.52 (s), 129.12 (s), 128.94 (s), 128.72 (s), 126.65 (s), 126.50 (s), 125.95 (s), 125.30 (s), 123.65 (s), 115.13 (s), 68.80 (s), 61.31 (s), 52.68 (s), 35.66 (s); IR (KBr, cm^{-1}) ν 3477.8 (s), 3414.4 (s), 1617.8 (s), 1510.5 (s), 1238.2 (s); HRMS (ES) m/z for $\text{C}_{22}\text{H}_{24}\text{O}_3\text{S}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 423.1059, found 423.1064.

Bis(propenyl)-4-(naphthalen-1-ylmethoxy)-phenyl-dithioacetal (S5). White solid, mp 114–116 °C, yield 97%; ^1H NMR (400 MHz, CDCl_3) δ 8.07–8.01 (m, 1H), 7.92–7.83 (m, 2H), 7.59 (d, J = 6.8 Hz, 1H), 7.57–7.49 (m, 2H), 7.49–7.43 (m, 1H), 7.38 (t, J = 5.8 Hz, 2H), 7.01 (d, J = 8.7 Hz, 2H), 5.80 (ddt, J = 17.1, 10.0, 7.2 Hz, 2H), 5.48 (s, 2H), 5.18–5.06 (m, 4H), 4.77 (s, 1H), 3.27 (dd, J = 13.7, 7.2 Hz, 2H), 3.06 (dd, J = 13.7, 7.2 Hz, 2H); ^{13}C NMR (101 MHz, CDCl_3) δ 158.54 (s), 133.89 (s), 133.82 (s), 132.33 (s), 132.20 (s), 131.55 (s), 129.30 (s), 129.08 (s), 128.72 (s), 126.65 (s), 126.48 (s), 125.94 (s), 125.32 (s), 123.70 (s), 117.52 (s), 114.95 (s), 68.77 (s), 49.99 (s), 35.28 (s); IR (KBr, cm^{-1}) ν 3351.2 (s), 2123.6 (s), 1537.4 (s), 1432.1 (s), 1218.6 (s); HRMS (ES) m/z for $\text{C}_{24}\text{H}_{24}\text{OS}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 437.1216, found 437.1219.

Bis(4-fluorophenyl)-4-(naphthalen-1-yl-methoxy)-3-methoxyphenyldithioacetal (S6). White solid, mp 117–119 °C, yield 92%; ^1H NMR (400 MHz, CDCl_3) δ 8.09 (d, J = 8.0 Hz, 1H), 7.92–7.79 (m, 2H), 7.60–7.40 (m, 4H), 7.36–7.27 (m, 4H), 6.99–6.89 (m, 4H), 6.87 (d, J = 2.0 Hz, 1H), 6.81 (d, J = 8.3 Hz, 1H), 6.68 (dd, J = 8.2, 2.1 Hz, 1H), 5.55 (s, 2H), 5.19 (s, 1H), 3.80 (s, 3H); ^{13}C NMR (101 MHz, CDCl_3) δ 164.12 (s), 161.65 (s), 149.92 (s), 148.06 (s), 135.80 (d, J = 8.5 Hz), 133.75 (s), 132.54 (s), 132.21 (s), 131.37 (s), 129.16 (d, J = 3.4 Hz), 128.77 (d, J = 13.2 Hz), 126.31 (d, J = 13.6 Hz), 125.86 (s), 125.30 (s), 123.57 (s), 120.29 (s), 116.06 (s), 115.84 (s), 114.15 (s), 111.46 (s), 69.76 (s), 61.82 (s), 56.02 (s); IR (KBr, cm^{-1}) ν 1588.4 (s), 1510.1 (s), 1486.9 (s), 1466.3 (s), 1265.5 (s), 1227.3 (s), 1136.0 (s); HRMS (ES) m/z for $\text{C}_{31}\text{H}_{24}\text{F}_2\text{O}_2\text{S}_2$ [$\text{M} + \text{H}$] $^+$ caclcd 531.1259, found 531.1257.

Bis(pentafluorophenyl)-4-(naphthalen-1-ylmethoxy)-3-methoxyphenyldithioacetal (S7). White solid, mp 118–120 °C, yield 85%; ^1H NMR (400 MHz, CDCl_3) δ 8.05 (d, J = 7.9 Hz, 1H), 7.91–7.82 (m, 2H), 7.58–7.50 (m, 3H), 7.46–7.42 (m, 1H), 7.11 (s, 1H), 6.83 (s, 2H), 5.64 (s, 1H), 5.54 (s, 2H), 3.88 (s, 3H). ^{13}C NMR (101 MHz, CDCl_3) δ 150.23 (s), 149.20 (s), 133.74 (s), 131.84 (s), 131.29 (s), 129.06 (s), 128.92 (s), 128.72 (s), 126.42 (s), 126.21 (s), 125.89 (s), 125.27 (s), 123.43 (s), 120.47 (s), 113.97 (s), 110.90 (s), 69.66 (s), 57.91 (s), 56.10 (s); IR (KBr, cm^{-1}) ν 1514.3 (s), 1489.8 (s), 1236.1 (s), 1088.2 (s); HRMS (ES) m/z for $\text{C}_{31}\text{H}_{16}\text{F}_{10}\text{O}_2\text{S}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 697.0324, found 697.0321.

Bis(2-hydroxyethyl)-4-(naphthalen-1-ylmethoxy)-3-methoxyphenyldithioacetal (S8). White solid, mp 113–114 °C, yield 93%; ^1H NMR (400 MHz, CDCl_3) δ 8.10 (d, J = 8.1 Hz, 1H), 7.91–7.87 (m, 1H), 7.83 (d, J = 8.3 Hz, 1H), 7.59 (d, J = 6.8 Hz, 1H), 7.53 (td, J = 7.7, 1.4 Hz, 2H), 7.47–7.42 (m, 1H), 7.07 (s, 1H), 6.91 (s, 2H), 5.56 (s, 2H), 5.03 (s, 1H), 3.89 (s, 3H), 3.74 (t, J =



5.8 Hz, 4H), 2.84 (dt, $J = 14.0$, 5.8 Hz, 2H), 2.72 (dt, $J = 14.0$, 5.9 Hz, 2H), 2.18 (s, 2H); ^{13}C NMR (101 MHz, CDCl_3) δ 150.33 (s), 148.34 (s), 133.76 (s), 133.18 (s), 132.24 (s), 131.39 (s), 128.83 (s), 128.67 (s), 126.36 (s), 126.28 (s), 125.84 (s), 125.32 (s), 123.58 (s), 120.08 (s), 114.18 (s), 111.34 (s), 69.84 (s), 61.34 (s), 56.15 (s), 53.09 (s), 35.72 (s); IR (KBr, cm^{-1}) ν 3414.5 (s), 32.46.4 (s), 1508.7 (s), 1466.4 (s), 1259.1 (s), 1212.0 (s), 1141.1 (s), 1004.7 (s); HRMS (ES) m/z for $\text{C}_{23}\text{H}_{26}\text{O}_4\text{S}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 453.1165, found 453.1161.

Bis(propenyl)-4-(naphthalen-1-ylmethoxy)-3-methoxyphenyldithioacetal (S9). Yellow solid, mp 90–92 °C, yield 95%; ^1H NMR (400 MHz, CDCl_3) δ 8.10 (d, $J = 8.1$ Hz, 1H), 7.92–7.79 (m, 2H), 7.63–7.40 (m, 4H), 7.06 (d, $J = 1.8$ Hz, 1H), 6.94–6.83 (m, 2H), 5.79 (ddt, $J = 17.1$, 10.0, 7.2 Hz, 2H), 5.56 (s, 2H), 5.17–5.04 (m, 4H), 4.73 (s, 1H), 3.88 (s, 3H), 3.27 (dd, $J = 13.7$, 7.1 Hz, 2H), 3.06 (dd, $J = 13.7$, 7.2 Hz, 2H). ^{13}C NMR (101 MHz, CDCl_3) δ 150.17 (s), 148.06 (s), 133.87 (s), 133.76 (s), 133.18 (s), 132.37 (s), 131.40 (s), 128.78 (s), 128.67 (s), 126.34 (s), 126.26 (s), 125.82 (s), 125.35 (s), 123.62 (s), 120.40 (s), 117.54 (s), 114.15 (s), 111.72 (s), 69.84 (s), 56.09 (s), 50.41 (s), 35.34 (s); IR (KBr, cm^{-1}) ν 3331.2 (s), 2118.4 (s), 1557.4 (s), 1462.1 (s), 1208.9 (s); HRMS (ES) m/z for $\text{C}_{25}\text{H}_{26}\text{O}_2\text{S}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 445.1266, found 445.1269.

Bis(4-fluorophenyl)-3-(naphthalen-1-ylmethoxy)-phenyldithioacetal (S10). White solid, mp 105–106 °C, yield 92%; ^1H NMR (400 MHz, CDCl_3) δ 8.03 (d, $J = 7.8$ Hz, 1H), 7.94–7.84 (m, 2H), 7.61–7.51 (m, 3H), 7.49–7.45 (m, 1H), 7.35–7.28 (m, 4H), 7.18 (t, $J = 7.9$ Hz, 1H), 7.00–6.86 (m, 7H), 5.43 (s, 2H), 5.21 (s, 1H); ^{13}C NMR (101 MHz, CDCl_3) δ 164.15 (s), 161.68 (s), 158.82 (s), 140.85 (s), 135.81 (d, $J = 8.2$ Hz), 133.82 (s), 132.10 (s), 131.52 (s), 129.55 (s), 129.08 (s), 128.71 (s), 126.57 (d, $J = 18.1$ Hz), 125.95 (s), 125.28 (s), 123.66 (s), 120.61 (s), 116.05 (s), 115.84 (s), 115.01 (s), 114.26 (s), 68.71 (s), 62.01 (s); IR (KBr, cm^{-1}) ν 1643.2 (s), 1593.7 (s), 1488.9 (s), 1212.1 (s); HRMS (ES) m/z for $\text{C}_{30}\text{H}_{22}\text{F}_2\text{O}_2\text{S}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 523.0972, found 523.0977.

Bis(4-chlorophenyl)-3-(naphthalen-1-ylmethoxy)-phenyldithioacetal (S11). White solid, mp 119–121 °C, yield 91%; ^1H NMR (400 MHz, CDCl_3) δ 8.04 (d, $J = 8.1$ Hz, 1H), 7.97–7.84 (m, 2H), 7.66–7.52 (m, 3H), 7.51–7.44 (m, 1H), 7.36–7.18 (m, 9H), 7.06–7.01 (m, 1H), 7.00–6.90 (m, 2H), 5.45 (s, 2H), 5.32 (s, 1H); ^{13}C NMR (101 MHz, CDCl_3) δ 158.91 (s), 140.47 (s), 134.63 (s), 134.22 (s), 133.81 (s), 132.51 (s), 132.05 (s), 131.51 (s), 129.68 (s), 129.08 (s), 129.04 (s), 128.72 (s), 126.66 (s), 126.51 (s), 125.95 (s), 125.29 (s), 123.65 (s), 120.60 (s), 115.19 (s), 114.26 (s), 68.72 (s), 60.76 (s); IR (KBr, cm^{-1}) ν 1544.6 (s), 1491.6 (s), 1230.5 (s), 1081.4 (s); HRMS (ES) m/z for $\text{C}_{30}\text{H}_{22}\text{Cl}_2\text{O}_2\text{S}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 555.0381, found 555.0376.

Bis(2-hydroxyethyl)-3-(naphthalen-1-ylmethoxy)-phenyldithioacetal (S12). White solid, mp 101–103 °C, yield 94%; ^1H NMR (400 MHz, CDCl_3) δ 8.07 (d, $J = 7.9$ Hz, 1H), 7.97–7.81 (m, 2H), 7.67–7.42 (m, 4H), 7.34–7.26 (m, 1H), 7.21–7.14 (m, 1H), 7.07 (d, $J = 7.7$ Hz, 1H), 6.98 (dd, $J = 8.0$, 2.1 Hz, 1H), 5.51 (s, 2H), 5.04 (s, 1H), 3.73 (s, 4H), 2.84 (dt, $J = 14.0$, 5.8 Hz, 2H), 2.72 (dt, $J = 14.0$, 5.9 Hz, 2H), 2.10 (s, 2H); ^{13}C NMR (101 MHz, CDCl_3) δ 159.11 (s), 141.65 (s), 133.81 (s), 132.10 (s), 131.54 (s), 129.86 (s), 129.08 (s), 128.70 (s), 126.72 (s), 126.50 (s), 125.95 (s), 125.27 (s), 123.70 (s), 120.40 (s), 114.94 (s), 114.24 (s), 68.76 (s), 61.30 (s), 53.17 (s), 35.70 (s); IR (KBr, cm^{-1}) ν 3415.1 (s), 3282.6

(s), 1514.8 (s), 1464.7 (s), 1011.2 (s); HRMS (ES) m/z for $\text{C}_{22}\text{H}_{24}\text{O}_3\text{S}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 423.1059, found 423.1062.

Bis(propenyl)-3-(naphthalen-1-ylmethoxy)-phenyldithioacetal (S13). White solid, mp 91–93 °C, yield 96%; ^1H NMR (400 MHz, CDCl_3) δ 8.08 (d, $J = 7.8$ Hz, 1H), 7.95–7.81 (m, 2H), 7.65–7.44 (m, 4H), 7.32–7.26 (m, 1H), 7.17 (d, $J = 1.8$ Hz, 1H), 7.05 (d, $J = 7.6$ Hz, 1H), 6.97 (dd, $J = 7.9$, 2.1 Hz, 1H), 5.80 (ddt, $J = 17.1$, 10.0, 7.2 Hz, 2H), 5.51 (s, 2H), 5.21–5.00 (m, 4H), 4.76 (s, 1H), 3.28 (dd, $J = 13.7$, 7.2 Hz, 2H), 3.07 (dd, $J = 13.7$, 7.2 Hz, 2H); ^{13}C NMR (101 MHz, CDCl_3) δ 159.05 (s), 141.61 (s), 133.81 (s), 133.78 (s), 132.21 (s), 131.58 (s), 129.64 (s), 129.07 (s), 128.69 (s), 126.75 (s), 126.47 (s), 125.92 (s), 125.30 (s), 123.78 (s), 120.84 (s), 117.64 (s), 114.62 (s), 114.50 (s), 68.71 (s), 50.48 (s), 35.31 (s); IR (KBr, cm^{-1}) ν 3321.2 (s), 2123.4 (s), 1551.4 (s), 1458.1 (s), 1208.9 (s); HRMS (ES) m/z for $\text{C}_{24}\text{H}_{24}\text{O}_2\text{S}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 415.1161, found 415.1166.

Bis(2-hydroxyethyl)-5-(naphthalen-1-ylmethoxy)-2-bromophenyldithioacetal (S14). Red solid, mp 123–124 °C, yield 93%; ^1H NMR (400 MHz, CDCl_3) δ 8.06 (d, $J = 7.9$ Hz, 1H), 7.88 (dd, $J = 13.8$, 8.2 Hz, 2H), 7.68–7.38 (m, 6H), 6.85 (dd, $J = 8.8$, 2.7 Hz, 1H), 5.51 (s, 3H), 3.76 (t, $J = 5.1$ Hz, 4H), 2.90–2.79 (m, 2H), 2.79–2.60 (m, 2H), 2.16 (s, 2H); ^{13}C NMR (101 MHz, CDCl_3) δ 158.67 (s), 140.15 (s), 133.81 (s), 133.54 (s), 131.65 (s), 131.49 (s), 129.26 (s), 128.75 (s), 126.91 (s), 126.60 (s), 126.02 (s), 125.25 (s), 123.65 (s), 117.12 (s), 115.85 (s), 114.05 (s), 69.06 (s), 61.09 (s), 51.53 (s), 36.01 (s); IR (KBr, cm^{-1}) ν 3415.1 (s), 3291.4 (s), 1517.8 (s), 1467.1 (s), 1016.2 (s); HRMS (ES) m/z for $\text{C}_{22}\text{H}_{23}\text{BrO}_3\text{S}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 501.0164, found 501.0161.

Bis(2-hydroxyethyl)-4-(naphthalen-1-ylmethoxy)-3-chlorophenyldithioacetal (S15). White solid, mp 102–104 °C, yield 94%; ^1H NMR (400 MHz, CDCl_3) δ 8.08 (d, $J = 8.2$ Hz, 1H), 7.94–7.82 (m, 2H), 7.65 (d, $J = 6.9$ Hz, 1H), 7.61–7.43 (m, 4H), 7.31 (dd, $J = 8.5$, 2.3 Hz, 1H), 7.05 (d, $J = 8.5$ Hz, 1H), 5.59 (s, 2H), 5.04 (s, 1H), 3.76 (q, $J = 5.6$ Hz, 4H), 2.85 (dt, $J = 14.0$, 5.8 Hz, 2H), 2.71 (dt, $J = 14.0$, 5.9 Hz, 2H), 2.15 (s, 2H); ^{13}C NMR (101 MHz, CDCl_3) δ 154.10 (s), 133.74 (s), 133.63 (s), 131.55 (s), 131.23 (s), 129.71 (s), 129.05 (s), 128.74 (s), 126.98 (s), 126.49 (s), 126.16 (s), 125.96 (s), 125.31 (s), 123.69 (s), 123.44 (s), 114.08 (s), 69.68 (s), 61.47 (s), 52.25 (s), 35.59 (s); IR (KBr, cm^{-1}) ν 3418.8 (s), 3271.4 (s), 1537.8 (s), 1467.1 (s), 1016.2 (s); HRMS (ES) m/z for $\text{C}_{22}\text{H}_{23}\text{ClO}_3\text{S}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 457.0669, found 457.0663.

Bis(2-hydroxyethyl)-2-(naphthalen-1-ylmethoxy)-5-methylphenyldithioacetal (S16). White solid, mp 139–140 °C, yield 91%; ^1H NMR (400 MHz, CDCl_3) δ 8.11–8.03 (m, 1H), 7.97–7.82 (m, 2H), 7.63–7.40 (m, 5H), 7.12–7.04 (m, 1H), 7.00 (d, $J = 8.3$ Hz, 1H), 5.52 (s, 2H), 5.43 (s, 1H), 3.48 (q, $J = 5.9$ Hz, 4H), 2.67 (dd, $J = 12.8$, 7.0 Hz, 2H), 2.57 (dd, $J = 12.9$, 7.1 Hz, 2H), 2.32 (s, 3H), 1.90 (d, $J = 6.2$ Hz, 2H); ^{13}C NMR (101 MHz, CDCl_3) δ 152.77 (s), 133.83 (s), 132.16 (s), 131.57 (s), 131.06 (s), 129.63 (s), 129.28 (s), 129.14 (s), 128.85 (s), 128.78 (s), 126.91 (s), 126.52 (s), 126.03 (s), 125.37 (s), 123.69 (s), 112.24 (s), 69.38 (s), 60.93 (s), 45.21 (s), 36.05 (s), 20.64 (s); IR (KBr, cm^{-1}) ν 3421.2 (s), 3271.4 (s), 1547.6 (s), 1477.3 (s), 1018.2 (s); HRMS (ES) m/z for $\text{C}_{23}\text{H}_{26}\text{O}_3\text{S}_2$ [$\text{M} + \text{Na}$] $^+$ caclcd 437.1216, found 437.1219.



Biological assays

In vivo antiviral activity. The biological activity of the title compounds against tobacco mosaic virus (TMV) and cucumber mosaic virus (CMV) respectively were evaluated using a half-leaf method according to the previously references.²⁴ Virus purification and activity evaluation of the compounds were performed as previously reported.^{27,28}

In vitro antibacterial activity. The antibacterial activities against rice bacterial leaf blight of title products were evaluated via the turbid meter test according to the reported method.²⁹ Bacterial cultivation and activity test of the title compounds were performed as previously reported.³⁰

Conflicts of interest

The authors confirm that this article content has no conflict of interest.

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